# Chapter – 4: Acids, Bases and Salts

- Daily life lots of substances tamarind, common salt, sugar, vinegar
- Taste lemon juice, orange juice, vinegar, curd, tamarind (*imli*), sugar, common salt, *amla*, baking soda, grapes, unripe mango, cucumber
- Some taste sour some taste bitter some taste sweet some taste salty

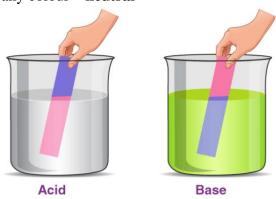
#### **Acids and Bases**

- Curd, lemon juice, vinegar sour taste acidic nature acids
- Acid Latin word *acere* sour taste
- Baking soda does not taste sour no acids bitter taste
- Rub between finger with water feels soapy
- Substances bitter taste feel soapy **basic nature bases**
- Cannot taste everything some harmful
- Special substances test acidic or basic indicators
- Indicators change colour according to solution's nature acidic or basic
- Turmeric, litmus, China rose petals (*Gudhal*)

## **Natural Indicators around Us**

## Litmus: A natural dye

- Most commonly used
- Extracted from lichens
- Mauve (purple) colour in water
- Add to acidic solution turns red add to basic solution turn blue
- Available solution or strip of paper forms
- Strip of paper
  - o Blue
    - Turns red in acid and no change in base
  - o Red
    - Turns blue in base and no change in acid
- Solutions do not affect any colour **neutral**



#### Turmeric is another natural indicator

• Paste of turmeric powder – deposit on filter paper – apply soap solution – basic – changes colour

#### China rose as indicator

- Deep petals in warm water keep them till water is coloured
- Add solutions to it and test them
- Acidic water turns to dark pink (magenta)
- Basic water turns to green
- Neutral water keeps its colour

## **Neutralisation**

- What happens when acid mix with base?
- We will use indicator **phenolphthalein**
- Take a test tube fill with dilute hydrochloric acid note the colour add indicator shake well now add base note the permanent change in colour
- Acidic solution phenolphthalein remain colourless
- Basic solution phenolphthalein changes colour to pink
- Acidic mixed with basic neutralize (cancel out) each other's effect
- New solution neither acidic nor basic
- Test tube gets heated heat produced during neutralization raises temperature
- Reaction between acid and base is neutralization salt and water produced as result along with heat
  - $\circ$  Acid + Base  $\rightarrow$  Salt + Water + heat evolved
  - O Hydrochloric acid (HCl) + Sodium hydroxide (NaOH) → Sodium chloride (NaCl) + Water (H<sub>2</sub>O)

## **Neutralization in Everyday Life**

## **Indigestion**

- Stomach contain hydrochloric acid
- Helps in digestion
- Too much acid indigestion
- Relieve indigestion antacid milk of magnesia magnesium hydroxide neutralizes the effect

#### Ant bite

- Ant bite inject acidic liquid (formic acid) into skin
- Neutralize rubbing moist baking soda (sodium hydrogencarbonate) or calamine solution (zinc carbonate)

#### Soil treatment

- Excessive use chemical fertilizers soil become acidic
- Plants do not grow well soil too acidic or too basic
- Acidic soil add bases quick lime (calcium oxide) or slaked lime (calcium hydroxide)
- Basic soil add organic matter (compost) release acid

#### **Factory wastes**

- Contain acid
- Released into water kills aquatic organisms
- Treat with bases before releasing to water bodies