

Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

Sample data

No.	Comment / ID	Start time	Sample size	Corr. f	Density
1/6	Benzoic acid	7/30/2012 5:47:15 PM	0.05649 g	1.0	0 g/mL
2/6	Benzoic acid	7/30/2012 5:52:05 PM	0.07039 g	1.0	0 g/mL
3/6	Benzoic acid	7/30/2012 5:57:11 PM	0.05586 g	1.0	0 g/mL
4/6	Benzoic acid	7/30/2012 6:01:42 PM	0.05597 g	1.0	0 g/mL
5/6	Benzoic acid	7/30/2012 6:06:15 PM	0.06283 g	1.0	0 g/mL
6/6	Benzoic acid	7/30/2012 6:11:16 PM	0.07272 g	1.0	0 g/mL

Results

No.	Comment / ID	Start time	Sample size and results		
1/6	Benzoic acid	7/30/2012 5:47:15 PM	0.05649 g		
			R1 = 0.99551	--	Titer
2/6	Benzoic acid	7/30/2012 5:52:05 PM	0.07039 g		
			R1 = 0.99932	--	Titer
3/6	Benzoic acid	7/30/2012 5:57:11 PM	0.05586 g		
			R1 = 0.99836	--	Titer
4/6	Benzoic acid	7/30/2012 6:01:42 PM	0.05597 g		
			R1 = 0.99774	--	Titer
5/6	Benzoic acid	7/30/2012 6:06:15 PM	0.06283 g		
			R1 = 0.99813	--	Titer
6/6	Benzoic acid	7/30/2012 6:11:16 PM	0.07272 g		
			R1 = 1.00004	--	Titer
-/-			R2 = 0.998	--	Mean Titer

Titer

Titer 0.99818

Statistics

Rx	Name	n	Mean value	Unit	s	srel [%]
R1	Titer	6	0.99818	--	0.00156	0.156
R2	Mean Titer	1	0.998	--	NaN	NaN

Raw data

Sample

No. 1/6
Standard Benzoic acid
Type of standard solid
Comment
Titration stand Rondo60/1A
Weight m = 0.05649 g
Correction factor f = 1.0
Purity p = 100.00 %
Temperature T = 25.0 oC
Sample start 7/30/2012 5:47:15 PM
Sample end 7/30/2012 5:52:05 PM

Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

EQP titration [1]

Titrant TBAOH c = 0.1 mol/L TITER = 0.99600
 Sensor DG113-SC
 Start potential EST = 109.4 mV
 Predispense EPD = -21.0 mV
 VPD = 1.5000 mL
 nEQ = 1
 No. of EQPs and cand. EQP1
 Consumption VEQ1 = 4.646628 mL
 Q1 = 0.462804 mmol
 EEQ1 = -209.3 mV
 EHN1 = -30.5 mV
 Excess VEX = 0.913372 mL
 QEX = 0.090972 mmol
 End VEND = 5.5600 mL
 QEND = 0.553776 mmol
 Termination at EQPs
 Time t = 3:19 min

Calculation

Result R1 = 0.99551 -- Titer
 Formula $R1 = m / (VEQ \cdot c \cdot C)$
 Constant $M / (10 \cdot p \cdot z)$
 C = 0.12212
 Molar mass M[Benzoic acid] = 122.12 g/mol
 Equivalent number z[Benzoic acid] = 1
 Duration tUSE = 04:17 min

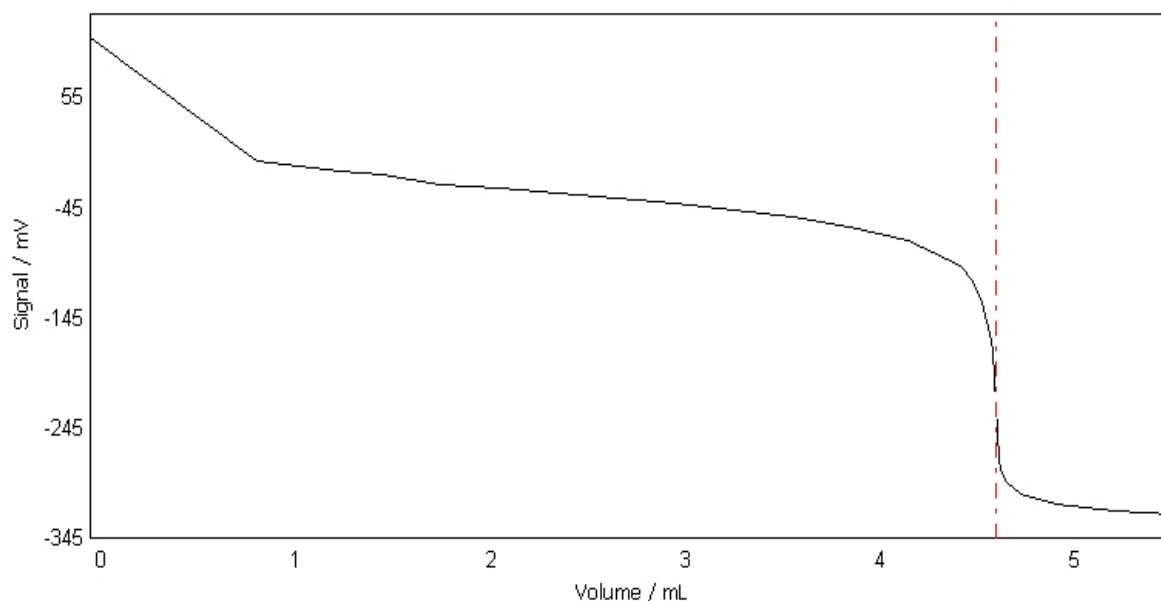
Measured values EQP titration [1]

Titrant TBAOH c = 0.1 mol/L TITER = 0.99600
 Sensor DG113-SC
 Sample 1/6

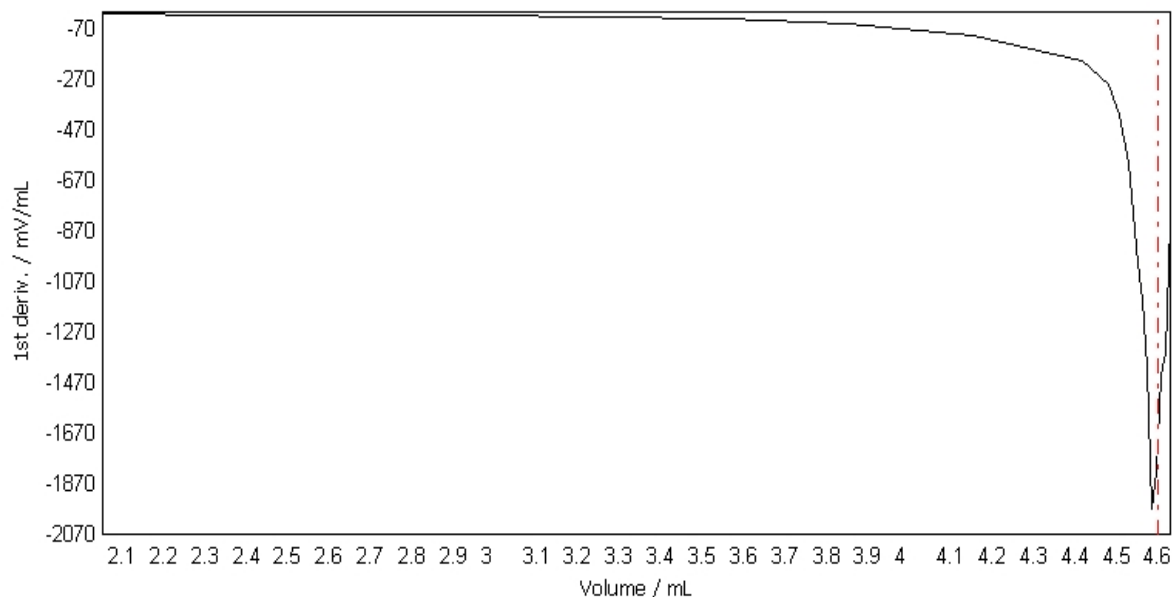
	Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
	0.0000	NaN	109.4	NaN	NaN	0	25.0
	0.8570	0.8570	-2.5	-111.9	NaN	9	25.0
	1.2855	0.4285	-11.4	-8.9	NaN	13	25.0
	1.5000	0.2145	-15.6	-4.2	NaN	17	25.0
	1.8000	0.3000	-23.6	-8.0	NaN	51	25.0
	2.1000	0.3000	-27.7	-4.1	-6.78	54	25.0
	2.4000	0.3000	-31.4	-3.7	-14.97	57	25.0
	2.7000	0.3000	-36.2	-4.8	-14.09	60	25.0
	3.0000	0.3000	-40.8	-4.6	-14.19	64	25.0
	3.3000	0.3000	-46.5	-5.7	-18.60	68	25.0
	3.6000	0.3000	-53.2	-6.7	-27.50	71	25.0
	3.9000	0.3000	-62.1	-8.9	-46.35	76	25.0
	4.2000	0.3000	-74.9	-12.8	-91.67	81	25.0
	4.4635	0.2635	-98.5	-23.6	-192.89	89	25.0
	4.5280	0.0645	-112.0	-13.5	-289.79	95	25.0
	4.5525	0.0245	-120.3	-8.3	-408.58	99	25.0
	4.5745	0.0220	-129.7	-9.4	-601.24	104	25.0
	4.5970	0.0225	-144.6	-14.9	-995.84	111	25.0
	4.6085	0.0115	-153.5	-8.9	-1140.43	115	25.0
	4.6215	0.0130	-165.8	-12.3	-1408.07	120	25.0
	4.6320	0.0105	-175.8	-10.0	-1971.62	124	25.0
	4.6445	0.0125	-198.5	-22.7	-1820.79	134	25.0
EQP1	4.646628	NaN	-209.3	NaN	-1998.47	NaN	NaN
	4.6545	0.0100	-249.3	-50.8	-1442.69	152	25.0
	4.6645	0.0100	-272.0	-22.7	-1346.66	162	25.0
	4.6765	0.0120	-282.9	-10.9	-831.64	167	25.0
	4.7065	0.0300	-294.6	-11.7	NaN	172	25.0
	4.7790	0.0725	-304.9	-10.3	NaN	176	25.0
	4.9600	0.1810	-313.6	-8.7	NaN	179	25.0
	5.2600	0.3000	-319.8	-6.2	NaN	183	25.0
	5.5600	0.3000	-323.5	-3.7	NaN	186	25.0

Method: TBAOH Titer with EQP of 0.1 TBAOH 7/30/2012 2:43:41 PM
Start time: 7/30/2012 5:47:15 PM

E - V curve EQP titration [1]
Sample 1/6



dE/dV - V curve EQP titration [1]
Sample 1/6



Raw data

Sample

No.	2/6
Standard	Benzoic acid
Type of standard	solid
Comment	
Titration stand	Rondo60/1A
Weight	m = 0.07039 g
Correction factor	f = 1.0
Purity	p = 100.00 %

Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

Temperature T = 25.0 oC
Sample start 7/30/2012 5:52:05 PM
Sample end 7/30/2012 5:57:11 PM

EQP titration [1]

Titration TBAOH c = 0.1 mol/L TITER = 0.99600
Sensor DG113-SC
Start potential EST = 110.6 mV
Predispense EPD = -15.1 mV
VPD = 1.5000 mL
nEQ = 1
No. of EQPs and cand. EQP1
Consumption VEQ1 = 5.767914 mL
Q1 = 0.574484 mmol
EEQ1 = -222.9 mV
EHN1 = -28.4 mV
Excess VEX = 0.784586 mL
QEX = 0.078145 mmol
End VEND = 6.5525 mL
QEND = 0.652629 mmol
Termination at EQPs
Time t = 3:30 min

Calculation

Result R1 = 0.99932 -- Titer
Formula $R1 = m / (VEQ \cdot c \cdot C)$
Constant $M / (10 \cdot p \cdot z)$
C = 0.12212
Molar mass M[Benzoic acid] = 122.12 g/mol
Equivalent number z[Benzoic acid] = 1
Duration tUSE = 04:31 min

Measured values EQP titration [1]

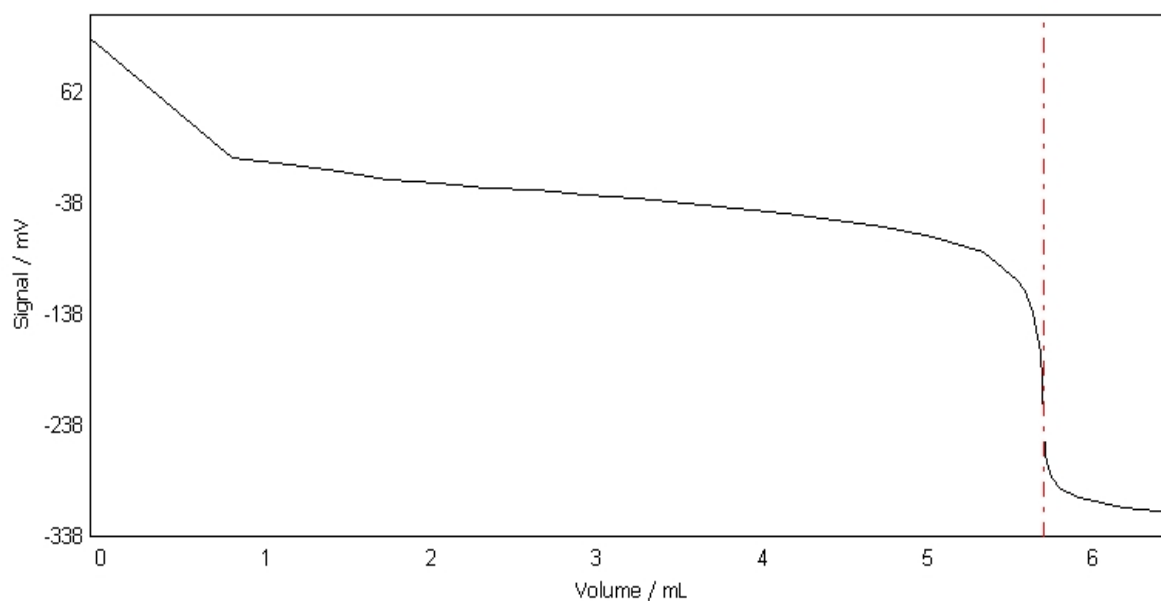
Titration TBAOH c = 0.1 mol/L TITER = 0.99600
Sensor DG113-SC
Sample 2/6

	Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
	0.0000	NaN	110.6	NaN	NaN	0	25.0
	0.8570	0.8570	3.6	-107.0	NaN	6	25.0
	1.2855	0.4285	-4.8	-8.4	NaN	10	25.0
	1.5000	0.2145	-9.7	-4.9	NaN	13	25.0
	1.8000	0.3000	-17.2	-7.5	NaN	46	25.0
	2.1000	0.3000	-20.2	-3.0	-4.84	50	25.0
	2.4000	0.3000	-23.6	-3.4	-11.64	53	25.0
	2.7000	0.3000	-26.3	-2.7	-10.28	56	25.0
	3.0000	0.3000	-29.8	-3.5	-10.89	59	25.0
	3.3000	0.3000	-33.6	-3.8	-12.60	63	25.0
	3.6000	0.3000	-37.8	-4.2	-13.68	66	25.0
	3.9000	0.3000	-42.1	-4.3	-14.67	70	25.0
	4.2000	0.3000	-46.8	-4.7	-14.42	74	25.0
	4.5000	0.3000	-52.5	-5.7	-19.01	78	25.0
	4.8000	0.3000	-59.5	-7.0	-29.59	83	25.0
	5.1000	0.3000	-68.3	-8.8	-54.00	87	25.0
	5.4000	0.3000	-82.5	-14.2	-113.35	93	25.0
	5.6130	0.2130	-107.7	-25.2	-225.97	102	25.0
	5.6535	0.0405	-116.6	-8.9	-318.81	106	25.0
	5.6835	0.0300	-129.0	-12.4	-480.73	113	25.0
	5.6990	0.0155	-136.1	-7.1	-773.24	117	25.0
	5.7225	0.0235	-150.8	-14.7	-1116.05	123	25.0
	5.7365	0.0140	-164.0	-13.2	-1257.44	129	25.0
	5.7465	0.0100	-172.5	-8.5	-1600.75	133	25.0
	5.7620	0.0155	-196.6	-24.1	-1865.17	143	25.0
EQP1	5.767914	NaN	-222.9	NaN	-1878.40	NaN	NaN
	5.7720	0.0100	-241.0	-44.4	-1266.23	158	25.0

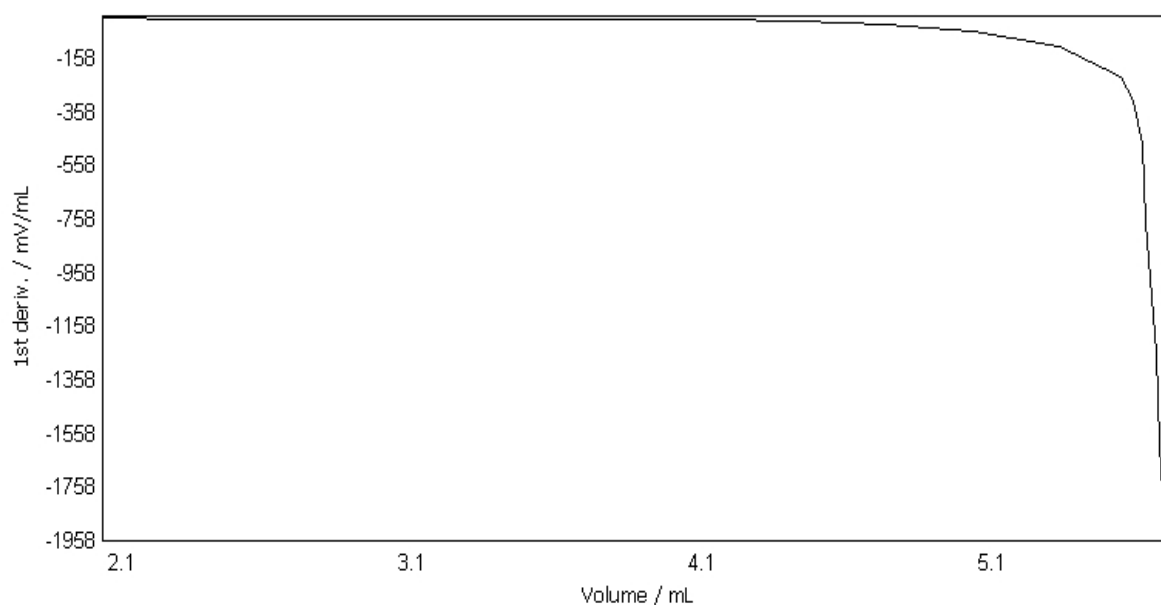
Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
5.7820	0.0100	-263.6	-22.6	-1324.37	167	25.0
5.7925	0.0105	-275.1	-11.5	-923.69	173	25.0
5.8150	0.0225	-284.2	-9.1	NaN	177	25.0
5.8710	0.0560	-296.1	-11.9	NaN	182	25.0
5.9800	0.1090	-303.5	-7.4	NaN	186	25.0
6.2525	0.2725	-312.6	-9.1	NaN	190	25.0
6.5525	0.3000	-316.6	-4.0	NaN	193	25.0

E - V curve EQP titration [1]
Sample 2/6



dE/dV - V curve EQP titration [1]
Sample 2/6



Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

Raw data

Sample

No. 3/6
Standard Benzoic acid
Type of standard solid
Comment
Titration stand Rondo60/1A
Weight m = 0.05586 g
Correction factor f = 1.0
Purity p = 100.00 %
Temperature T = 25.0 oC
Sample start 7/30/2012 5:57:11 PM
Sample end 7/30/2012 6:01:42 PM

EQP titration [1]

Titrant TBAOH c = 0.1 mol/L TITER = 0.99600
Sensor DG113-SC
Start potential EST = 101.2 mV
Predispense EPD = -23.7 mV
VPD = 1.5000 mL
nEQ = 1
No. of EQPs and cand. EQP1 VEQ1 = 4.581688 mL
Consumption Q1 = 0.456336 mmol
EEQ1 = -200.1 mV
EHN1 = -32.0 mV
VEX = 0.900812 mL
QEX = 0.089721 mmol
VEND = 5.4825 mL
QEND = 0.546057 mmol
Excess
End
Termination at EQPs
Time t = 2:53 min

Calculation

Result R1 = 0.99836 -- Titer
Formula $R1 = m / (VEQ * c * C)$
Constant $M / (10 * p * z)$
C = 0.12212
Molar mass M[Benzoic acid] = 122.12 g/mol
Equivalent number z[Benzoic acid] = 1
Duration tUSE = 03:57 min

Measured values EQP titration [1]

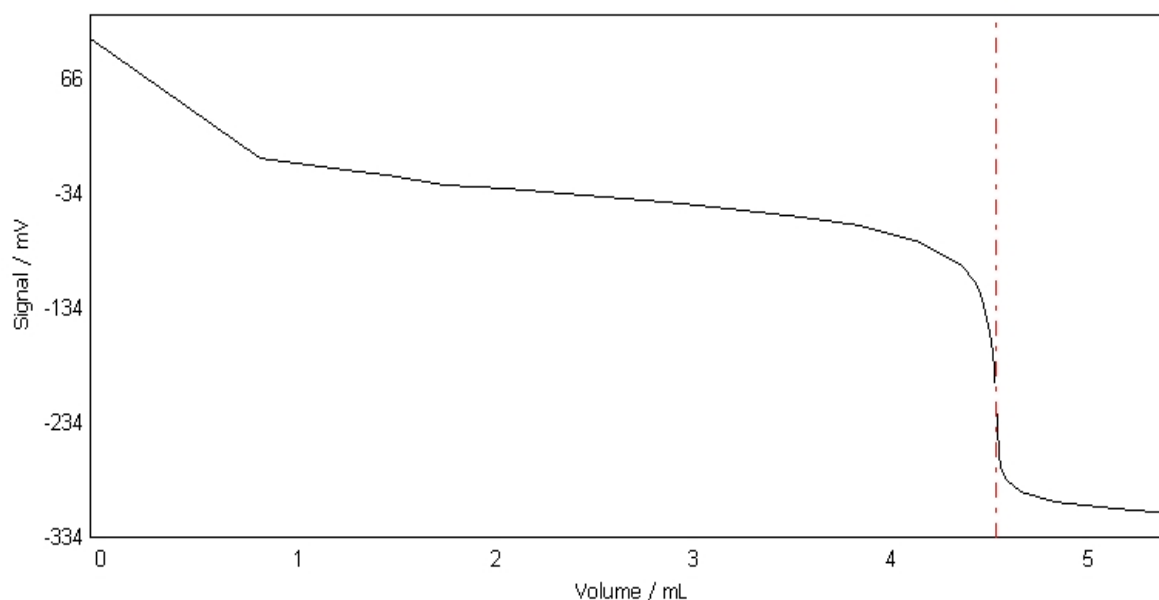
Titrant TBAOH c = 0.1 mol/L TITER = 0.99600
Sensor DG113-SC
Sample 3/6

Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
0.0000	NaN	101.2	NaN	NaN	0	25.0
0.8570	0.8570	-3.3	-104.5	NaN	6	25.0
1.2855	0.4285	-13.3	-10.0	NaN	11	25.0
1.5000	0.2145	-18.1	-4.8	NaN	14	25.0
1.8000	0.3000	-26.1	-8.0	NaN	48	25.0
2.1000	0.3000	-29.6	-3.5	-7.57	51	25.0
2.4000	0.3000	-33.4	-3.8	-14.29	54	25.0
2.7000	0.3000	-37.6	-4.2	-13.00	58	25.0
3.0000	0.3000	-42.3	-4.7	-13.17	61	25.0
3.3000	0.3000	-48.3	-6.0	-17.92	65	25.0
3.6000	0.3000	-54.0	-5.7	-27.52	68	25.0
3.9000	0.3000	-62.1	-8.1	-49.46	71	25.0
4.2000	0.3000	-76.3	-14.2	-105.02	76	25.0

Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

	Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
EQP1	4.4145	0.2145	-97.7	-21.4	-208.73	82	25.0
	4.4715	0.0570	-110.1	-12.4	-304.42	87	25.0
	4.4970	0.0255	-117.9	-7.8	-449.85	91	25.0
	4.5245	0.0275	-130.9	-13.0	-825.79	95	25.0
	4.5410	0.0165	-143.7	-12.8	-1202.51	100	25.0
	4.5510	0.0100	-155.0	-11.3	-1248.72	105	25.0
	4.5610	0.0100	-163.6	-8.6	-1758.45	108	25.0
	4.5790	0.0180	-186.1	-22.5	-1778.90	116	25.0
	4.581688	NaN	-200.1	NaN	-1805.08	NaN	NaN
	4.5890	0.0100	-238.0	-51.9	-1325.63	131	25.0
	4.5990	0.0100	-260.8	-22.8	-1305.37	138	25.0
	4.6110	0.0120	-272.2	-11.4	-840.03	143	25.0
	4.6410	0.0300	-284.4	-12.2	NaN	147	25.0
	4.7100	0.0690	-293.9	-9.5	NaN	150	25.0
	4.8825	0.1725	-303.1	-9.2	NaN	154	25.0
	5.1825	0.3000	-309.5	-6.4	NaN	157	25.0
	5.4825	0.3000	-313.5	-4.0	NaN	160	25.0

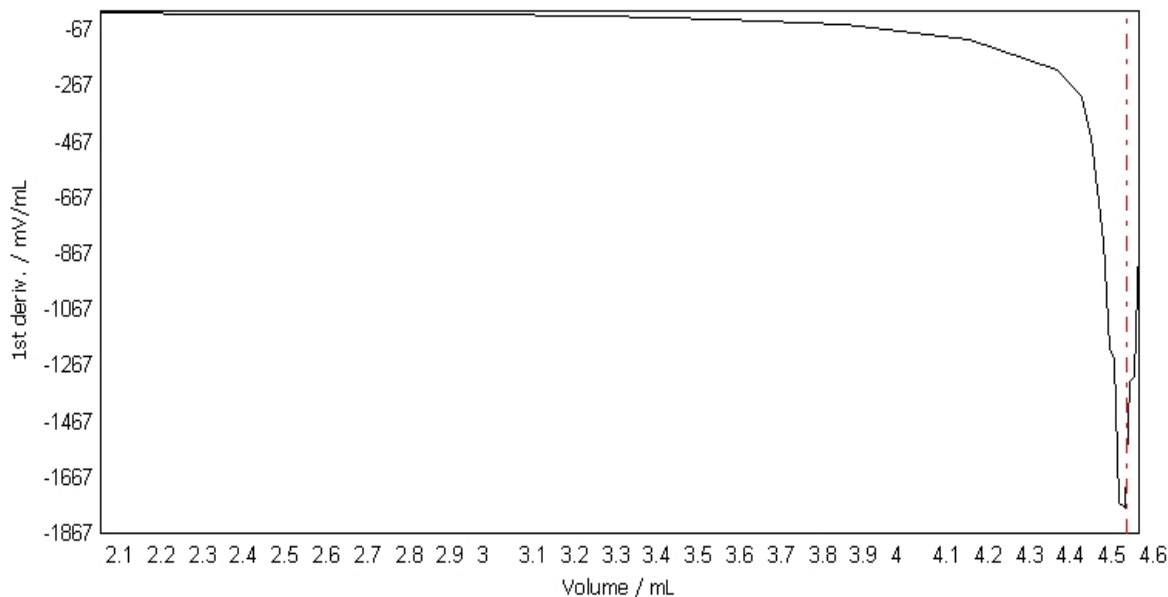
E - V curve **EQP titration [1]**
Sample 3/6



dE/dV - V curve **EQP titration [1]**

Method: TBAOH Titer with EQP of 0.1 TBAOH 7/30/2012 2:43:41 PM
Start time: 7/30/2012 5:47:15 PM

Sample 3/6



Raw data

Sample

No. 4/6
Standard Benzoic acid
Type of standard solid
Comment
Titration stand Rondo60/1A
Weight $m = 0.05597 \text{ g}$
Correction factor $f = 1.0$
Purity $p = 100.00 \%$
Temperature $T = 25.0 \text{ }^{\circ}\text{C}$
Sample start 7/30/2012 6:01:42 PM
Sample end 7/30/2012 6:06:15 PM

EQP titration [1]

Titrant TBAOH $c = 0.1 \text{ mol/L}$ TITER = 0.99600
Sensor DG113-SC
Start potential EST = 99.1 mV
Predispense EPD = -23.7 mV
VPD = 1.5000 mL
No. of EQPs and cand. nEQ = 1
Consumption EQP1 VEQ1 = 4.593586 mL
Q1 = 0.457521 mmol
EEQ1 = -202.4 mV
EHN1 = -31.8 mV
Excess VEX = 0.939914 mL
QEX = 0.093615 mmol
End VEND = 5.5335 mL
QEND = 0.551137 mmol
Termination at EQPs
Time $t = 2:55 \text{ min}$

Calculation

Result $R1 = 0.99774$ -- Titer
Formula $R1 = m / (VEQ \cdot c \cdot C)$
Constant $M / (10 \cdot p \cdot z)$

Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

C = 0.12212
Molar mass M[Benzoic acid] = 122.12 g/mol
Equivalent number z[Benzoic acid] = 1
Duration tUSE = 03:58 min

Measured values EQP titration [1]

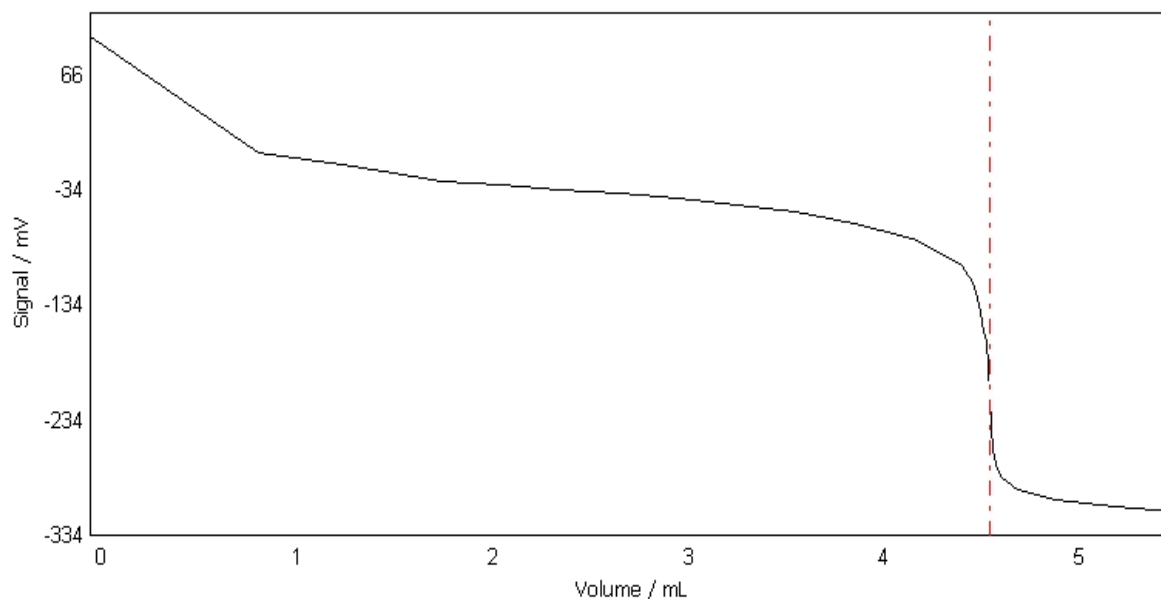
Titrant TBAOH c = 0.1 mol/L TITER = 0.99600
Sensor DG113-SC
Sample 4/6

	Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
	0.0000	NaN	99.1	NaN	NaN	0	25.0
	0.8570	0.8570	-1.1	-100.2	NaN	5	25.0
	1.2855	0.4285	-12.3	-11.2	NaN	10	25.0
	1.5000	0.2145	-17.2	-4.9	NaN	13	25.0
	1.8000	0.3000	-26.0	-8.8	NaN	46	25.0
	2.1000	0.3000	-29.4	-3.4	-8.68	50	25.0
	2.4000	0.3000	-33.1	-3.7	-13.73	53	25.0
	2.7000	0.3000	-37.4	-4.3	-12.66	56	25.0
	3.0000	0.3000	-41.8	-4.4	-12.79	59	25.0
	3.3000	0.3000	-47.2	-5.4	-17.60	63	25.0
	3.6000	0.3000	-53.4	-6.2	-27.43	66	25.0
	3.9000	0.3000	-62.6	-9.2	-49.46	70	25.0
	4.2000	0.3000	-75.6	-13.0	-99.95	75	25.0
	4.4525	0.2525	-99.8	-24.2	-212.01	81	25.0
	4.5095	0.0570	-114.8	-15.0	-323.93	86	25.0
	4.5260	0.0165	-123.0	-8.2	-501.76	90	25.0
	4.5385	0.0125	-133.0	-10.0	-915.77	94	25.0
	4.5485	0.0100	-140.4	-7.4	-1120.63	98	25.0
	4.5660	0.0175	-156.2	-15.8	-1344.00	103	25.0
	4.5770	0.0110	-166.1	-9.9	-1951.51	106	25.0
	4.5905	0.0135	-186.0	-19.9	-1670.27	114	25.0
EQP1	4.593586	NaN	-202.4	NaN	-2008.70	NaN	NaN
	4.6005	0.0100	-239.1	-53.1	-1324.99	129	25.0
	4.6105	0.0100	-259.9	-20.8	-1240.74	137	25.0
	4.6255	0.0150	-273.2	-13.3	-788.17	142	25.0
	4.6570	0.0315	-283.5	-10.3	NaN	146	25.0
	4.7360	0.0790	-293.8	-10.3	NaN	150	25.0
	4.9335	0.1975	-303.0	-9.2	NaN	153	25.0
	5.2335	0.3000	-309.2	-6.2	NaN	157	25.0
	5.5335	0.3000	-313.0	-3.8	NaN	161	25.0

E - V curve EQP titration [1]

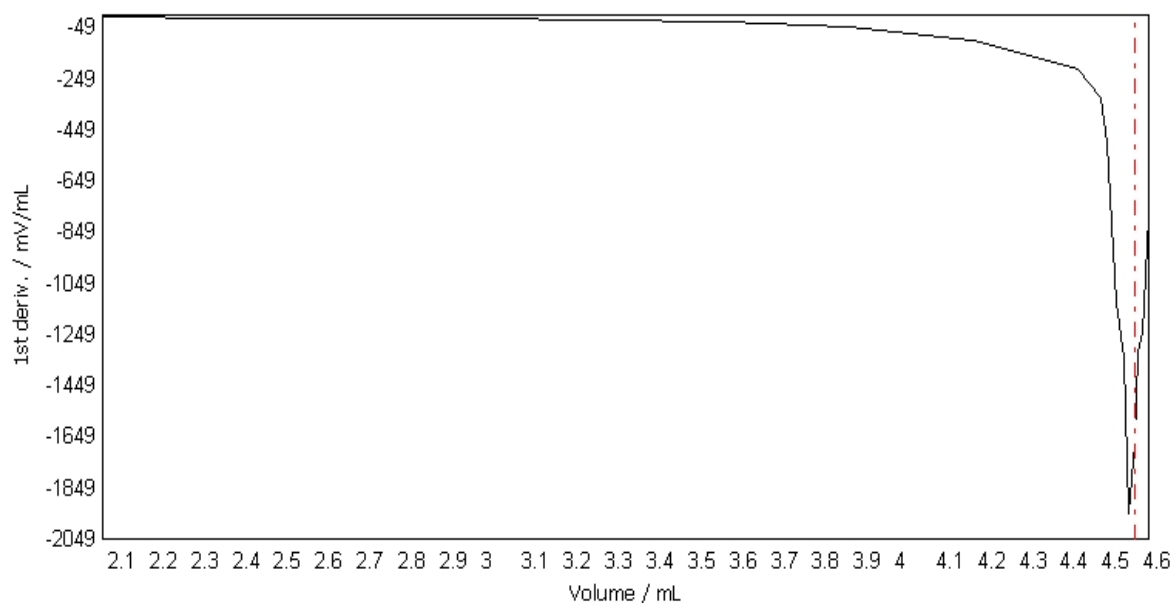
Method: TBAOH Titer with EQP of 0.1 TBAOH 7/30/2012 2:43:41 PM
Start time: 7/30/2012 5:47:15 PM

Sample 4/6



dE/dV - V curve EQP titration [1]

Sample 4/6



Raw data

Sample

No. 5/6
Standard Benzoic acid
Type of standard solid
Comment
Titration stand Rondo60/1A
Weight $m = 0.06283 \text{ g}$
Correction factor $f = 1.0$
Purity $p = 100.00 \%$
Temperature $T = 25.0 \text{ }^{\circ}\text{C}$

Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

Sample start 7/30/2012 6:06:15 PM
Sample end 7/30/2012 6:11:16 PM

EQP titration [1]

Titrant TBAOH c = 0.1 mol/L TITER = 0.99600
 Sensor DG113-SC
 Start potential EST = 93.3 mV
 Predispense EPD = -20.3 mV
 VPD = 1.5000 mL
 nEQ = 1
 No. of EQPs and cand. EQP1
 Consumption VEQ1 = 5.154563 mL
 Q1 = 0.513394 mmol
 EEQ1 = -219.8 mV
 EHN1 = -30.2 mV
 VEX = 0.833437 mL
 QEX = 0.083010 mmol
 VEND = 5.9880 mL
 QEND = 0.596405 mmol
 Excess
 End
 Termination at EQPs
 Time t = 3:22 min

Calculation

Result R1 = 0.99813 -- Titer
 Formula $R1 = m / (VEQ \cdot c \cdot C)$
 Constant $M / (10 \cdot p \cdot z)$
 C = 0.12212
 Molar mass M[Benzoic acid] = 122.12 g/mol
 Equivalent number z[Benzoic acid] = 1
 Duration tUSE = 04:25 min

Measured values EQP titration [1]

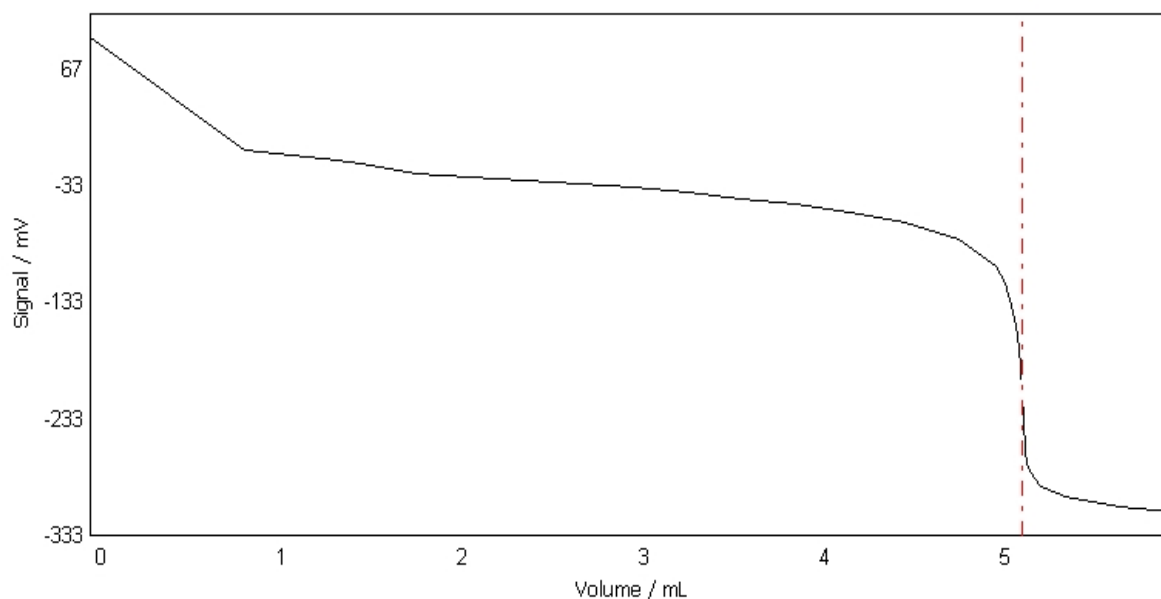
Titrant TBAOH c = 0.1 mol/L TITER = 0.99600
 Sensor DG113-SC
 Sample 5/6

	Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
	0.0000	NaN	93.3	NaN	NaN	0	25.0
	0.8570	0.8570	-3.2	-96.5	NaN	7	25.0
	1.2855	0.4285	-9.5	-6.3	NaN	10	25.0
	1.5000	0.2145	-14.3	-4.8	NaN	13	25.0
	1.8000	0.3000	-22.5	-8.2	NaN	46	25.0
	2.1000	0.3000	-25.2	-2.7	-5.67	50	25.0
	2.4000	0.3000	-28.2	-3.0	-12.37	53	25.0
	2.7000	0.3000	-31.5	-3.3	-10.77	56	25.0
	3.0000	0.3000	-35.2	-3.7	-11.50	59	25.0
	3.3000	0.3000	-39.0	-3.8	-13.63	62	25.0
	3.6000	0.3000	-44.8	-5.8	-14.33	66	25.0
	3.9000	0.3000	-49.2	-4.4	-18.50	69	25.0
	4.2000	0.3000	-55.8	-6.6	-29.51	73	25.0
	4.5000	0.3000	-65.5	-9.7	-54.47	77	25.0
	4.8000	0.3000	-79.9	-14.4	-115.06	83	25.0
	5.0100	0.2100	-102.4	-22.5	-229.05	90	25.0
	5.0600	0.0500	-116.8	-14.4	-346.66	96	25.0
	5.0755	0.0155	-126.2	-9.4	-497.79	102	25.0
	5.0855	0.0100	-131.5	-5.3	-785.20	106	25.0
	5.1105	0.0250	-146.8	-15.3	-1137.87	112	25.0
	5.1275	0.0170	-161.4	-14.6	-1376.65	118	25.0
	5.1375	0.0100	-173.1	-11.7	-1834.03	124	25.0
	5.1475	0.0100	-187.9	-14.8	-1986.23	131	25.0
EQP1	5.154563	NaN	-219.8	NaN	-2016.00	NaN	NaN
	5.1575	0.0100	-233.1	-45.2	-1279.23	148	25.0
	5.1675	0.0100	-259.2	-26.1	-1384.01	159	25.0
	5.1775	0.0100	-271.2	-12.0	-894.73	165	25.0
	5.1990	0.0215	-280.3	-9.1	NaN	170	25.0

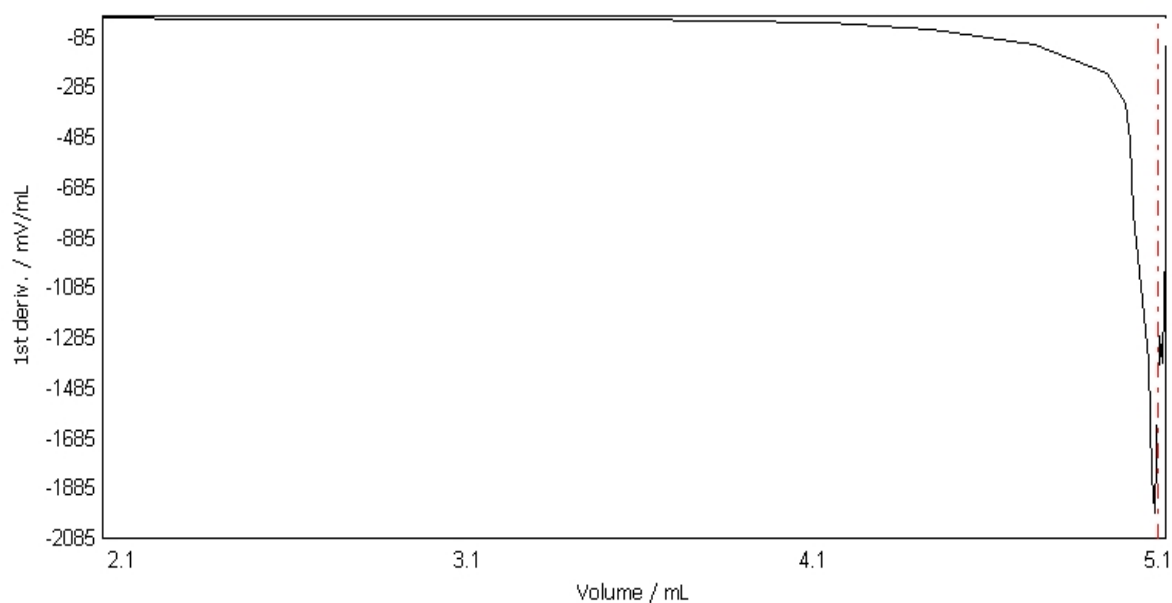
Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
5.2530	0.0540	-290.5	-10.2	NaN	174	25.0
5.3880	0.1350	-300.4	-9.9	NaN	178	25.0
5.6880	0.3000	-308.1	-7.7	NaN	182	25.0
5.9880	0.3000	-312.6	-4.5	NaN	186	25.0

E - V curve **EQP titration [1]**
Sample 5/6



dE/dV - V curve **EQP titration [1]**
Sample 5/6



Raw data

Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

Raw data

Sample

No. 6/6
Standard Benzoic acid
Type of standard solid
Comment
Titration stand Rondo60/1A
Weight m = 0.07272 g
Correction factor f = 1.0
Purity p = 100.00 %
Temperature T = 25.0 oC
Sample start 7/30/2012 6:11:16 PM
Sample end 7/30/2012 6:16:31 PM

EQP titration [1]

Titrant TBAOH c = 0.1 mol/L TITER = 0.99600
Sensor DG113-SC
Start potential EST = 98.1 mV
Predispense EPD = -15.2 mV
VPD = 1.5000 mL
nEQ = 1
No. of EQPs and cand. EQP1
Consumption VEQ1 = 5.954545 mL
Q1 = 0.593073 mmol
EEQ1 = -199.4 mV
EHN1 = -28.1 mV
Excess VEX = 0.934955 mL
QEX = 0.093122 mmol
End VEND = 6.8895 mL
QEND = 0.686194 mmol
Termination at EQPs
Time t = 3:34 min

Calculation

Result R1 = 1.00004 -- Titer
Formula $R1 = m / (VEQ * c * C)$
Constant $M / (10 * p * z)$
C = 0.12212
Molar mass M[Benzoic acid] = 122.12 g/mol
Equivalent number z[Benzoic acid] = 1
Duration tUSE = 04:39 min

Measured values EQP titration [1]

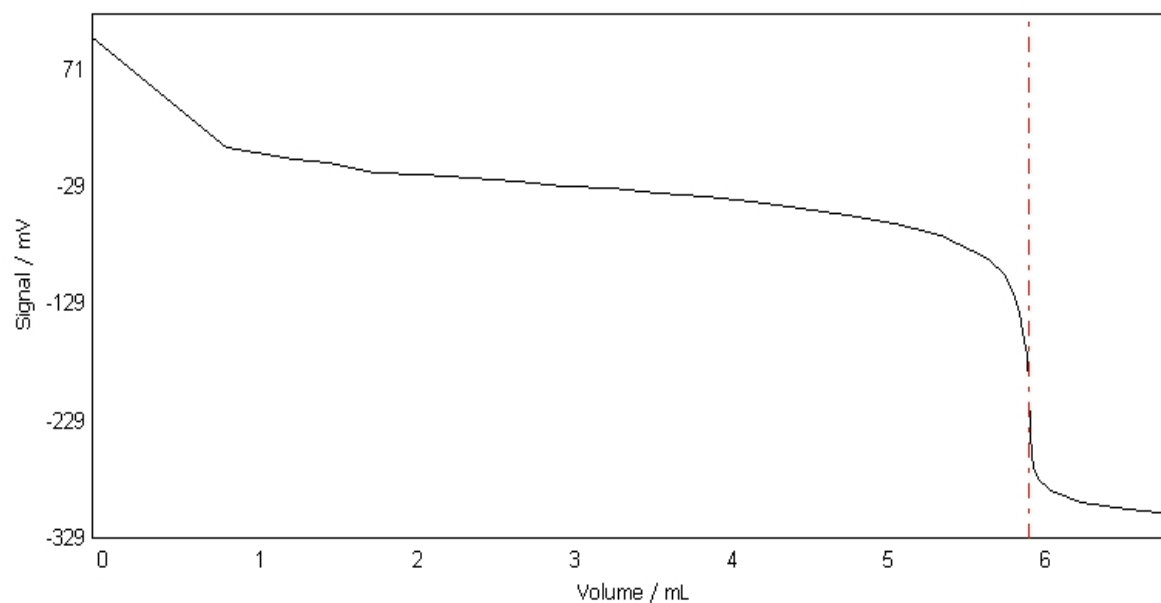
Titrant TBAOH c = 0.1 mol/L TITER = 0.99600
Sensor DG113-SC
Sample 6/6

Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
0.0000	NaN	98.1	NaN	NaN	0	25.0
0.8570	0.8570	4.4	-93.7	NaN	5	25.0
1.2855	0.4285	-6.0	-10.4	NaN	10	25.0
1.5000	0.2145	-9.6	-3.6	NaN	14	25.0
1.8000	0.3000	-17.0	-7.4	NaN	47	25.0
2.1000	0.3000	-19.2	-2.2	-5.09	50	25.0
2.4000	0.3000	-22.2	-3.0	-10.11	54	25.0
2.7000	0.3000	-24.8	-2.6	-9.36	57	25.0
3.0000	0.3000	-28.4	-3.6	-9.37	60	25.0
3.3000	0.3000	-31.2	-2.8	-11.04	63	25.0
3.6000	0.3000	-35.1	-3.9	-11.60	66	25.0
3.9000	0.3000	-38.2	-3.1	-12.89	70	25.0
4.2000	0.3000	-42.4	-4.2	-13.34	73	25.0

Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

	Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s	Temperature oC
	4.5000	0.3000	-47.9	-5.5	-15.87	77	25.0
	4.8000	0.3000	-53.2	-5.3	-21.96	80	25.0
	5.1000	0.3000	-61.2	-8.0	-35.88	85	25.0
	5.4000	0.3000	-70.7	-9.5	-66.98	90	25.0
	5.7000	0.3000	-90.7	-20.0	-151.44	98	25.0
	5.8085	0.1085	-106.1	-15.4	-241.50	105	25.0
	5.8480	0.0395	-116.7	-10.6	-345.21	111	25.0
	5.8715	0.0235	-124.6	-7.9	-495.69	115	25.0
	5.9000	0.0285	-137.5	-12.9	-844.40	121	25.0
	5.9195	0.0195	-152.4	-14.9	-1170.60	128	25.0
	5.9295	0.0100	-160.5	-8.1	-1246.93	132	25.0
	5.9435	0.0140	-173.9	-13.4	-1710.67	138	25.0
	5.9545	0.0110	-199.2	-25.3	-1612.42	150	25.0
EQP1	5.954545	NaN	-199.4	NaN	-1715.35	NaN	NaN
	5.9645	0.0100	-239.2	-40.0	-1237.82	163	25.0
	5.9745	0.0100	-258.8	-19.6	-1189.07	173	25.0
	5.9870	0.0125	-268.7	-9.9	-730.08	178	25.0
	6.0180	0.0310	-279.2	-10.5	NaN	182	25.0
	6.0955	0.0775	-289.2	-10.0	NaN	187	25.0
	6.2895	0.1940	-298.6	-9.4	NaN	191	25.0
	6.5895	0.3000	-304.7	-6.1	NaN	194	25.0
	6.8895	0.3000	-308.3	-3.6	NaN	197	25.0

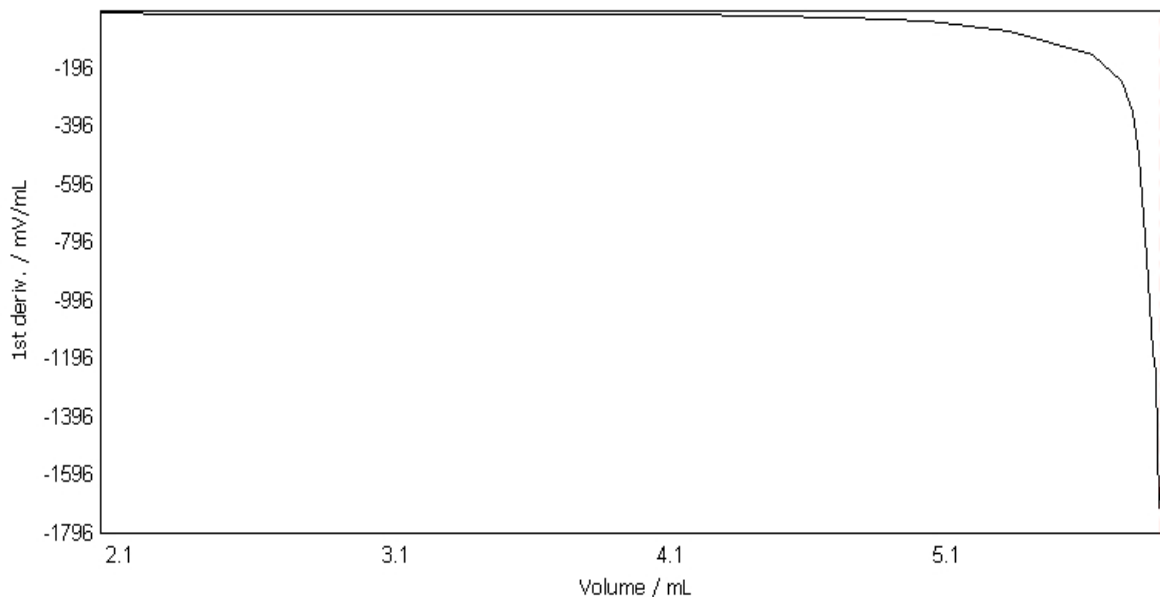
E - V curve **EQP titration [1]**
Sample 6/6



dE/dV - V curve **EQP titration [1]**

Method: TBAOH **Titer with EQP of 0.1 TBAOH** **7/30/2012 2:43:41 PM**
Start time: 7/30/2012 5:47:15 PM

Sample 6/6



Raw data

Calculation

Result R2 = 0.998 -- Mean Titer
 Formula R2=Mean[R1]
 Constant 1
 C = 1
 Molar mass M[None] = 1 g/mol
 Equivalent number z[None] = 1

Titer

Titrant TBAOH c = 0.1 mol/L
 Titer 0.99818

- (1) Modified
- (2) Excluded
- (3) Outside limits
- (4) Resource expired
- (5) srel above max srel
- (6) srel above max srel for multiple determination
- (7) Value outside limits, not saved in setup
- (8) Sample data outside limits
- (9) Standard evaluation used
- (10) Result from buffer

Created: Development Administrator (admin), 7/30/2012 5:52:09 PM