

Quiz: Alcohols, Phenols and Ethers 1

Q1: Which compound is formed when sodium phenoxide reacts with ethyl iodide?

- A) Phenetole
- B) Ethyl phenyl alcohol
- C) Phenol
- D) None of the above

Q2: The reagent that differentiates primary, secondary and tertiary alcohols is Lucas reagent (conc. HCl + ZnCl₂). Which of the following reacts fastest?

- A) 1 deg alcohol
- B) 2 deg alcohol
- C) 3 deg alcohol
- D) Phenol

Q3: Phenol gives violet color with FeCl₃. What is the principal reason?

- A) Hydrogen bonding
- B) Formation of complex
- C) No reaction
- D) Oxidation

Q4: When ethanol is dehydrated with Conc. H₂SO₄, the major product is:

- A) Ethane
- B) Ethene
- C) Ethanol sulfate
- D) Acetaldehyde

Q5: Which product is formed on oxidation of benzyl alcohol with PCC?

- A) Benzoic acid
- B) Benzaldehyde
- C) Benzene
- D) Benzyl chloride

Q6: Which reagent is best for Williamson Ether Synthesis?

- A) Alkoxide + alkyl halide
- B) Alcohol + acid chloride
- C) Ether + acid
- D) Alcohol + base only

Q7: In the reaction C₆H₅OH + Br₂ → which of the following products form?

- A) 2,4,6-tribromophenol
- B) Bromobenzene
- C) 4-bromophenol
- D) No reaction

Q8: Ether cleavage by HI: $R-O-R' + HI \rightarrow$ major products depend on carbocation stability. Which carbon is attacked first?

- A) More substituted
- B) Less substituted
- C) Either equally
- D) No reaction

Q9: Which of the following alcohols will give a positive iodoform test?

- A) Methanol
- B) Ethanol
- C) Tert-butyl alcohol
- D) Phenol

Q10: The acidic strength order among these compounds: water, phenol, ethanol, acetic acid is:

- A) $H_2O < Phenol < C_2H_5OH < CH_3COOH$
- B) $C_2H_5OH < H_2O < Phenol < CH_3COOH$
- C) $C_2H_5OH < H_2O < CH_3COOH < Phenol$
- D) $H_2O < C_2H_5OH < Phenol < CH_3COOH$

Q11: The product of the reaction: $ROH + PCl_5 \rightarrow RCl + HCl + POCl_3$ is:

- A) Alkyl chloride
- B) Alkyl alcohol
- C) Phenol
- D) Ester

Q12: Consider the reaction: $C_2H_5OH \rightarrow C_2H_5Cl$. This is an example of:

- A) Substitution
- B) Elimination
- C) Oxidation
- D) Reduction

Q13: Which compound has the highest boiling point?

- A) Diethyl ether
- B) Ethanol
- C) Ethyl methyl ether
- D) Ethane

Q14: What is the IUPAC name of $CH_3CH(OH)CH_2OH$?

- A) Propane-1,2-diol
- B) Propane-2,3-diol
- C) Butane-2,3-diol
- D) None

Q15: Which reagent converts alcohol to alkyl chloride?

- A) $SOCl_2$
- B) PCl_5
- C) $HCl + ZnCl_2$
- D) All of the above

Q16: The order of decreasing acidic strength: p-nitrophenol, phenol, m-nitrophenol is:

- A) p-nitrophenol > o-nitrophenol > m-nitrophenol
- B) m-nitrophenol > p-nitrophenol > phenol
- C) phenol > m-nitrophenol > p-nitrophenol
- D) o-nitrophenol > p-nitrophenol > m-nitrophenol

Q17: Ether $\text{CH}_3\text{-O-CH}_2\text{CH}_3$ has bond angle C-O-C approximately:

- A) 104.5 deg
- B) 109.5 deg
- C) 120 deg
- D) 90 deg

Q18: Which of the following reactions is of nucleophilic substitution?

- A) $\text{ROH} + \text{SOCl}_2$
- B) $\text{RCH=CH}_2 + \text{H}_2\text{O}$
- C) $\text{C}_6\text{H}_5\text{OH} + \text{Br}_2$
- D) $\text{ROH} \rightarrow \text{Alkene}$

Q19: Phenol reacts with diazonium salt to form:

- A) Azo compound
- B) Ester
- C) Ether
- D) Alkane

Q20: Which of the following does not give a positive iodoform test?

- A) Ethanol
- B) Propanone
- C) 2-propanol
- D) Ethanal

Q21: Which of the following compounds has the highest boiling point?

- A) Propan-1-ol
- B) Butan-1-ol
- C) Butan-2-ol
- D) Pent-1-ol

Q22: Reaction of phenol with zinc dust gives:

- A) Benzene
- B) Toluene
- C) Benzaldehyde
- D) Benzoic acid

Q23: Which enzyme converts glucose to ethanol?

- A) Zymase
- B) Invertase
- C) Maltase
- D) Diastase

Q24: Which ether is used as an anaesthetic?

- A) Diethyl ether
- B) Dimethyl ether
- C) Ethyl methyl ether
- D) Diphenyl ether

Q25: The C-O-H bond angle in alcohols is slightly less than tetrahedral angle due to:

- A) Repulsion between bond pairs
- B) Repulsion between lone pairs
- C) Hydrogen bonding
- D) Large size of oxygen

Q26: When glycerol is treated with excess HI, the major product is:

- A) 2-iodopropane
- B) Propene
- C) 1,2,3-triiodopropane
- D) Allyl iodide

Q27: Arrange the following in increasing acidic strength: phenol, cyclohexanol, 2,4,6-trinitrophenol, ethanol.

- A) Cyclohexanol < ethanol < phenol < 2,4,6-trinitrophenol
- B) Ethanol < cyclohexanol < phenol < 2,4,6-trinitrophenol
- C) Cyclohexanol < ethanol < 2,4,6-trinitrophenol < phenol
- D) Phenol < ethanol < cyclohexanol < 2,4,6-trinitrophenol

Q28: Which of the following undergoes fastest oxidation with dichromate?

- A) 1-propanol
- B) 2-propanol
- C) tert-butyl alcohol
- D) Phenol

Q29: The product of the reaction: $\text{CH}_3\text{CH}(\text{OH})\text{CH}_3 \rightarrow$ is:

- A) Propanol
- B) Acetone
- C) Propanone
- D) No reaction

Q30: Which one of the following gives a positive Tollens' test?

- A) Ethanol
- B) Propanal
- C) 2-propanone
- D) Tert-butyl alcohol

Q31: Ether cleavage by HI for $\text{CH}_3\text{OCH}_2\text{CH}_3$ yields:

- A) Methanol + ethyl iodide
- B) Methyl iodide + ethanol
- C) Ethanol + ethane
- D) Methyl iodide + ethyl iodide

Q32: Which reagent converts phenol to sodium phenoxide?

- A) HCl
- B) NaOH
- C) ZnCl_2
- D) PCl_5

Q33: The IUPAC name of $\text{CH}_3\text{CH}_2\text{OCH}_2\text{CH}_3$ is:

- A) Ethyl ether
- B) Diethyl ether
- C) Methoxyethane
- D) Ethanoxyethane

Q34: Which alcohol would react fastest with Lucas reagent?

- A) 1-butanol
- B) 2-butanol
- C) tert-butanol
- D) Ethanol

Q35: Phenol on reaction with Br_2 in water gives:

- A) Bromobenzene
- B) 2,4,6-tribromophenol
- C) Bromophenol
- D) No reaction

Q36: Which alcohol does not give iodoform test?

- A) Ethanol
- B) Isopropanol
- C) 2-butanol
- D) Methanol

Q37: Product of Williamson ether synthesis from sodium ethoxide and CH_3I is:

- A) Ethyl methyl ether
- B) Dimethyl ether
- C) Diethyl ether
- D) Ethanol

Q38: Which molecule has highest boiling point?

- A) Diethyl ether
- B) Ethanol
- C) Methoxyethane
- D) Ethane

Q39: Which reagent can oxidize primary alcohol to acid?

- A) PCC
- B) KMnO_4
- C) LiAlH_4
- D) Pd/H_2

Q40: Phenol reacts with Na_2CO_3 to give:

- A) Sodium phenoxide
- B) No reaction
- C) Phenol carbonate
- D) Phenyl carbonate

Q41: Dehydration of secondary alcohol results in:

- A) Symmetrical ether
- B) Unsymmetrical ether
- C) Alkene
- D) Ester

Q42: Which test identifies phenols?

- A) Tollens' test
- B) Lucas test
- C) FeCl_3 test
- D) Iodoform test

Q43: Which of these gives an aldehyde on oxidation?

- A) 1-propanol with PCC
- B) 2-propanol with KMnO_4
- C) Tert-butanol with PCC
- D) Methanol with PCC

Q44: Which ether is most easily cleaved by HI?

- A) Methyl tert-butyl ether
- B) Diethyl ether
- C) Ethyl methyl ether
- D) Dipropyl ether

Q45: Product of $\text{C}_6\text{H}_5\text{OH} + \text{CO}_2/\text{NaOH} \rightarrow$ followed by acidification is:

- A) Salicylic acid
- B) Phenol carbonate
- C) Benzoic acid
- D) No reaction

Q46: Order of increasing acidity: ethanol, phenol, acetic acid is:

- A) Ethanol < phenol < acetic acid
- B) Phenol < ethanol < acetic acid
- C) Acetic acid < phenol < ethanol
- D) Phenol < acetic acid < ethanol

Q47: In $\text{CH}_3\text{CH}_2\text{Br} + \text{NaOCH}_3 \rightarrow$ major product is:

- A) Ethyl methyl ether
- B) Methyl bromide
- C) Ethanol
- D) No reaction

Q48: Which compound yields 2,4,6-tribromophenol with bromine water?

- A) Phenol
- B) Aniline
- C) Benzene
- D) Toluene

Q49: The product of dehydration of $\text{CH}_3\text{CH}(\text{OH})\text{CH}_3$ with conc. H_2SO_4 is:

- A) Propane
- B) Propene
- C) Isopropyl sulfate
- D) Acetone

Q50: Reaction of phenol with HNO_3 gives mainly:

- A) o-Nitrophenol
- B) p-Nitrophenol
- C) Mixed o- and p-nitrophenols
- D) m-Nitrophenol