## Week 7

## Bridge to 5.55

## 7.1 Solid-State NMR

3/18: • Do NOESY's always take like 3h30?

- Depends on how much signal you want!
- If you have very little compound, sure. But if you've got 20 mg of a small molecule, you can get a NOESY in 20 minutes.
- So perhaps even less time for my 300 mg sample!
- You can decrease the number of scans for the direct dimension from 32 down to 4, 2, or 1.
  - Will get progressively messier, but if you have a ton of signal, it may not matter.
- You can also get rid of the phase cycling.
- You can also decrease the number of **increments** in the indirect dimension to balance time vs. resolution.
  - Can cut down increments from 256 to 128 or something, too.
- PSet 4: ROE/NOE similar?
  - Yep! No worries.
- PSet 4: No absolute stereochemistry without chiral resolving agent?
  - Yep!
  - Just explain the relative stereochemistry.
- PSet 4: Proton-carbon couplings to interpret?
  - Hard to do because it's a very strained system, so things go in weird ways you might not expect.
  - Some explanation to that effect is good.
- Do you have to run 1D experiments ahead of 2D?
  - You never have to.
  - Sometimes, 1D experiments can capture things that 2D won't. For example, if you're taking a <sup>1</sup>H-<sup>13</sup>C HSQC, the 1D projection of the 2D dots will not encompass any X-H bonds (e.g., exchangeable hydroxyl/amine protons). Thus, if you don't want to miss anything, it might make sense to take a 1D, too.
  - But it doesn't help with phasing or anything.
- <sup>19</sup>F-<sup>13</sup>C HSQC didn't work?

- Check the chemical shift ranges, check the default parameters.
- Let Walt know if I still can't make it work.
- This week:
  - Miscellaneous experiments, transitioning into the biological structural assignment course (5.55).
  - Solids.
- Show and tell: A 3.2 mm solid-state rotor.
  - There's a couple of caps, too. These caps go into the rotor, and that's how we spin it.
  - The fins on the rotor accept a stream of nitrogen and get spun very smoothly.
- Rotors help in solid-state NMR, because you've got to do magic-angle spinning at 54.7°.
  - This angle solves the equation

$$3\cos^2\theta - 1 = 0$$

- This averages out **dipolar couplings**.
- Smaller and smaller rotors can be spun faster and faster. When you get to spinning at 32 kHz, you can just take a normal proton spectrum as if you were in liquid. Requires a 0.6-0.7 mm rotor; trying to cap this is like trying to cap a grain of sand.
- First ultra-small rotors were developed on Albany St. by Robert Guy Griffin!
- **Dipolar coupling**: Two spins that are close enough together to interact.
  - This is an effect that is significant in a solid in a way that it's not in a liquid. Tumbling and diffusion in the liquid phase naturally decouples dipolar interactions.
- Solid-state NMR.
  - Side-bands occur at the rotor frequency (like satellites).
  - Change the spinning speed by a few kilohertz and watch the side bands move to determine what's real and what's fake.
  - You take solid-state NMR because you've got a polymer or something you can't get to go into solution, and you have to get data on it somehow.
  - In the solid state, each molecule has a different orientation with respect to the magnetic field; there is significant *anisotropy*.
- Cross-polarization.
  - Analogous to carbon-proton NOEs.
  - Transfers magnetization from a sensitive nucleus to an insensitive nucleus.
  - Occurs best when you hit the **Hartman-Hahn match**.
  - This means tht <sup>13</sup>C is much more useful than <sup>1</sup>H for solids.
- Hartman-Hahn match: The ratio of the gyromagnetic ratios...
- Solvent-swollen gels can be spun at lower speeds ( $\approx 5\,\mathrm{kHz}$ ) to get spectra.
- Another big thing in solid-state NMR is dynamic nuclear polarization.
  - Can amplify signals by factors of 50-100 instead of the 2-3 you get with cross polarization.
  - Instead of transferring magnetization from one nuclear spin to another nuclear spin, you transfer it from an electron to a nuclear spin.
- Switching gears to special topics in biological NMR.

- Key things to keep in mind.
  - Biopolymers are polymers, but they're **heteropolymers** made of a small number of building blocks.
  - The building blocks are known.
  - Higher-order structure is important; we care less about connectivity and more about this.
- Proton chemical shifts and coupling patterns are often enough to make amino-acid assignments.
  - Intra-residue COSY/TOCSY connectivities extend from the backbone NH through the sidechain.
  - Sequential assignments require NOESY connectivities between adjacent amino acids.
    - Be careful with folded structures! Nearest neighbor may not be next in sequence.
- Larger proteins require uniform <sup>13</sup>C and <sup>15</sup>N labeling.
  - 3D experiments (HNCO, HNCA, HNCOCA) trace the backbone, and CBCACO-type experiments extend from the backbone to the side-chain.
- Amide protons need water in order to be seen.
  - You often have to carefully modulate the pH.