Week 5

Approximate Solutions of the Schrödinger Equation

5.1 Approximation Methods

10/25:

- The **variational method** and **perturbation theory** are two methods of approximating solutions to Schrödinger equations describing systems more complex than the hydrogen atom.
- To begin our investigation of the variational method, we will look at the particle in a box.
 - Consider a Hamiltonian for an electron in a box of length L=2 a.u. centered around x=0.
 - Note that we take the electron as the fundamental mass, \hbar as the fundamental unit of energy time, and the charge of the electron as the fundamental unit of charge, and the Bohr radius as the fundamental unit of length.
 - Our Hamiltonian is

$$H\psi(x) = -\frac{\hbar^2}{2m} \frac{\mathrm{d}^2}{\mathrm{d}x^2} \psi(x)$$

or, in atomic units,

$$H\psi(x) = -\frac{1}{2}\frac{\mathrm{d}^2}{\mathrm{d}x^2}\psi(x)$$

• Variational theorem: The expectation value of our Hamiltonian with respect to a trial wave function produces an approximate energy. Moreover^[1],

$$E_{\rm approx} \ge E_{\rm gr. st.}$$

- Variational method: Take $\psi_{\text{trial}} = \sum_n a_n |\psi_n\rangle$ where ψ_n is a trial wave function and the a_j 's are parameters of the wave function which we want to optimize to lower E_{trial} .
 - Dirac's ket describes an abstract state of the particle (possibly position, possibly its Fourier transform, momentum).
- Back to the particle in a box:
 - A possible trial wave function (that satisfies the boundary conditions) is

$$\psi_{\rm tr} = (1+x)(1-x) = 1-x^2$$

¹We will prove that the approximate energy is an upper bound on the ground state energy in the homework.

- The energy of $\psi_{\rm tr}$ may be evaluated as follows.

$$E = \frac{\int \psi_{\text{tr}}^*(x) \hat{H} \psi_{\text{tr}}(x) \, dx}{\int \psi_{\text{tr}}^*(x) \psi_{\text{tr}}(x) \, dx}$$

$$= \frac{\int_{-1}^{1} (1 - x^2) \left(-\frac{1}{2} \frac{d^2}{dx^2} \right) (1 - x^2) \, dx}{\int_{-1}^{1} (1 - x^2) (1 - x^2) \, dx}$$

$$= \frac{\int_{-1}^{1} (1 - x^2) \, dx}{\int_{-1}^{1} (1 - x^2) (1 - x^2) \, dx}$$

$$= \frac{4/3}{16/15}$$

$$= \frac{5}{4}$$

$$= 1.25 \text{ a.u.}$$

- From the exact solution to the particle in a box

$$E_1 = 1.23370055 < 1.25 = E_{\text{trial}}$$

so the variational theorem is satisfied.

- Next step: Trial wave function as a linear combination is $\psi_{\rm tr}(x) = c_1 \psi_1(x) + c_2 \psi_2(x)$.
- Plugging this into the SE yields

$$c_1(\hat{H} - E)\psi_1(x) + c_2(\hat{H} - E)\psi_2(x) = 0$$

- $\blacksquare \psi_1, \psi_2$ span the (Hilbert) space of solutions.
- To solve the above equation, multiply by $\psi_1(x)$ and integrate to obtain

$$c_1 \int_{-1}^{1} \psi_1^*(x)(\hat{H} - E)\psi_1(x) dx + c_2 \int_{-1}^{1} \psi_1^*(x)(\hat{H} - E)\psi_2(x) dx = 0$$

and multiply by $\psi_2(x)$ an integrate to obtain

$$c_1 \int_{-1}^{1} \psi_2^*(x)(\hat{H} - E)\psi_1(x) \, dx + c_2 \int_{-1}^{1} \psi_2^*(x)(\hat{H} - E)\psi_2(x) \, dx = 0$$

- Substituting, we have

$$c_1(H_{11} - ES_{11}) + c_2(H_{12} - ES_{12}) = 0$$
 $c_1(H_{21} - ES_{21}) + c_2(H_{22} - ES_{22}) = 0$

- In matrix form, the above two equations become

$$\begin{pmatrix} H_{11} & H_{12} \\ H_{21} & H_{22} \end{pmatrix} \begin{pmatrix} c_1 \\ c_2 \end{pmatrix} - E \begin{pmatrix} S_{11} & S_{12} \\ S_{21} & S_{22} \end{pmatrix} \begin{pmatrix} c_1 \\ c_2 \end{pmatrix} = \begin{pmatrix} 0 \\ 0 \end{pmatrix}$$
$$\mathbb{H}\vec{c} - E \mathbb{S}\vec{c} = 0$$

- We get a matrix that the same dimension as the size of the expansion (in the first case, we had a 1×1 matrix).
- S is the overlap matrix because the wave functions aren't normalized.