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**C20-C/M-104****7019****BOARD DIPLOMA EXAMINATION, (C-20)****SEPTEMBER/OCTOBER—2021****DCE - FIRST YEAR EXAMINATION****ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES***Time : 3 hours ]*

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**PART—A****3×10=30**

- Instructions :**
- (1) Answer **all** questions.
  - (2) Each question carries **three** marks.
  - (3) Answers should be brief and straight to the point and shall not exceed five simple sentences.

- 1. What are fundamental particles? Name them.
- 2. Define solute, solvent and solution.
- 3. Define Arrhenius acid and base. Give one example for each.
- 4. What are non-electrolytes? Give examples.
- 5. Write any three disadvantages of using hard water in industries.
- 6. Write the names of monomers in polythene and teflon.
- 7. Write the composition and uses of water gas.
- 8. Write the composition and applications of vinegar.
- 9. Define COD and BOD.
- 10. Define producers and consumers. Give examples.

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**PART—B**

8×5=40

- Instructions :** (1) Answer **all** questions.  
(2) Each question carries **eight** marks.  
(3) Answers should be comprehensive and criterion for valuation is the content but not the length of the answer.

**11.** (a) State and explain Aufbau's principle and Hund's rule.

**OR**

(b) Define ionic bond. Explain the formation of NaCl.

**12.** (a) Define normality. If 98 grams of  $\text{H}_2\text{SO}_4$  is present in 5 litres of solution. Find the normality of solution.

**OR**

(b) Explain Lewis acid-base theory.

**13.** (a) Define alloy. Write the composition and uses of (i) German silver and (ii) Nichrome.

**OR**

(b) Define galvanic cell. Explain its structure and working.

**14.** (a) Define rusting of iron. Explain the mechanism of rusting of iron.

**OR**

(b) Explain permutit process of removal of hardness of water.

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**15.** (a) Define elastomer. Write preparation and uses of Buna-S.

**OR**

(b) Explain (i) greenhouse effect and (ii) acid rain.

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- 16.** Explain addition polymerisation and condensation polymerisation. Explain with one example each.

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