

Surface Adsorption of Chlorine Oxyanion Species to the Air-water Interface

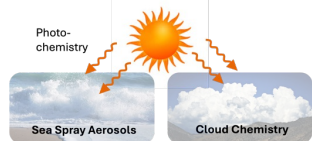
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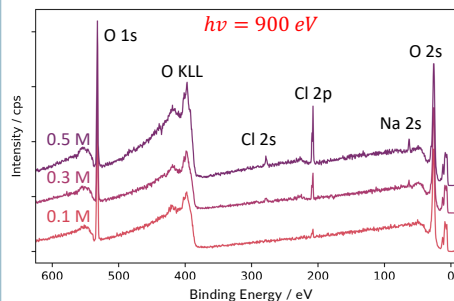
Motivation



The adsorption of select ions to the liquid water interface has profound effects on important environmental processes, such as evaporation rates, aerosol chemistry, accelerated reactions, and more. Despite the overwhelming importance of these molecular details, the underlying physics of ion solvation models used to interpret surface phenomena are still widely debated. Chlorine oxyanions serve as a powerful model system, due to their abundance in seawater and importance in aerosol cloud chemistry. Here, to better understand how size, charge density, and hydration enthalpy, among other factors, influence surface adsorption, we compare the bulk and surface speciation of chlorine oxyanion species (ClO_4^- and ClO_3^-) in liquid water. Using Ambient Pressure X-ray Photoelectron Spectroscopy (APXPS) on planar liquid jets, we are able to track individual ionic species at various depths away from the interface.

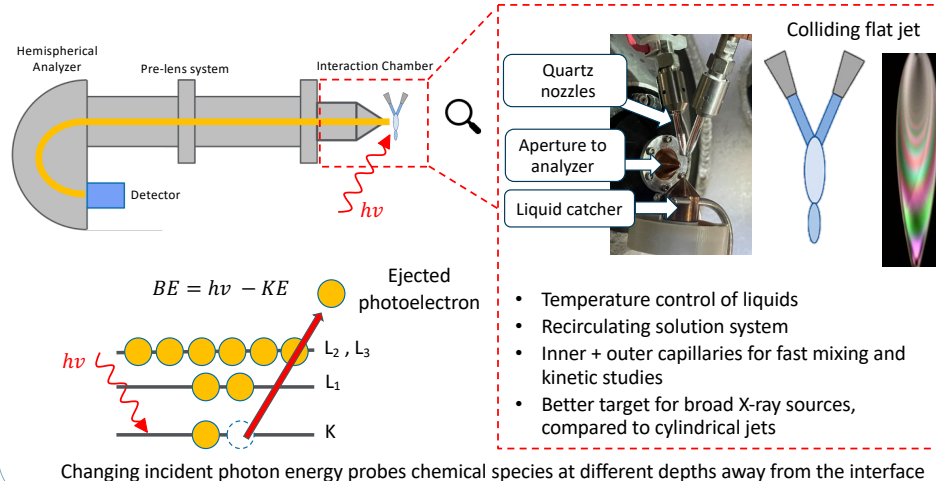
Survey Spectra

NaClO_4 Concentration Series

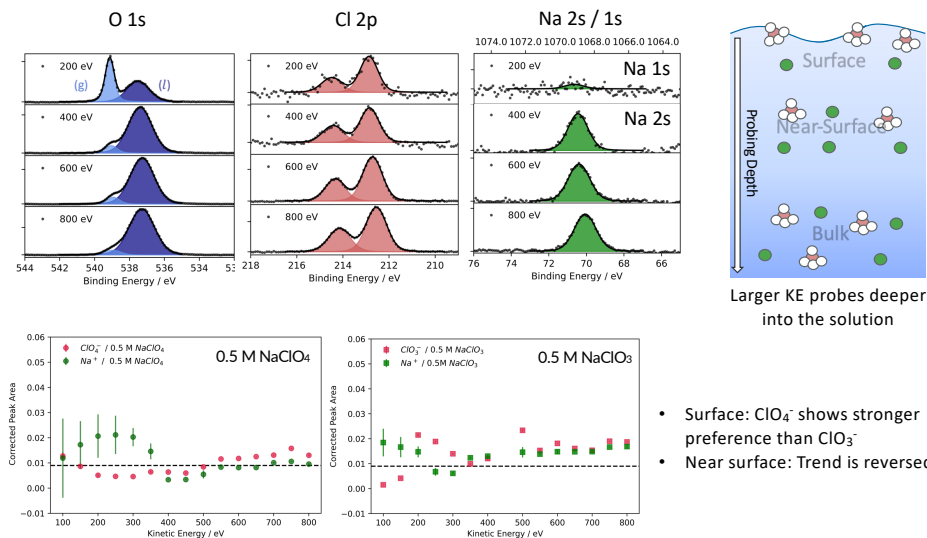


- Contamination free spectra
- All relevant species identified: O, Cl, and Na
- Highly sensitive to concentration, even with short acquisition times

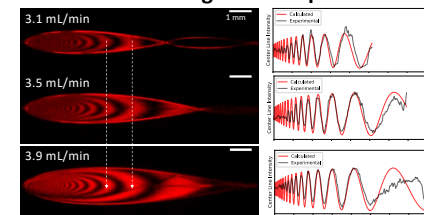
Ambient Pressure X-ray Photoelectron Spectroscopy of Liquid Flat Jets



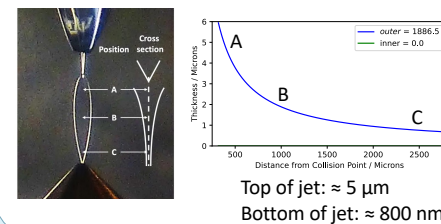
Surface to Bulk: Ion Depth Profiling of Chlorine Oxyanions



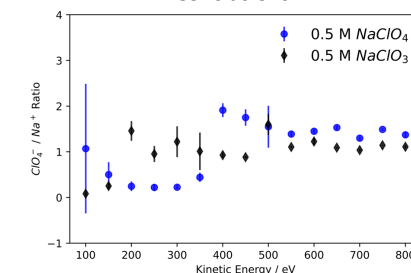
Characterizing Planar Liquid Jets



Thin film interference models are used to reliably measure the flat jet thickness



Conclusions



- 0.5 M NaClO_4 and NaClO_3 solutions show similar surface speciation

On-going collaborative work: Soft X-ray Second Harmonic Generation at X-ray Free Electron Lasers

- Combining surface selective measurements at XFELs with depth profiling at ALS for a "complete" description of ion distributions at the interface