

# 微纳光电子材料与器件工艺原理

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## Film Deposition Part II: Si Oxidation

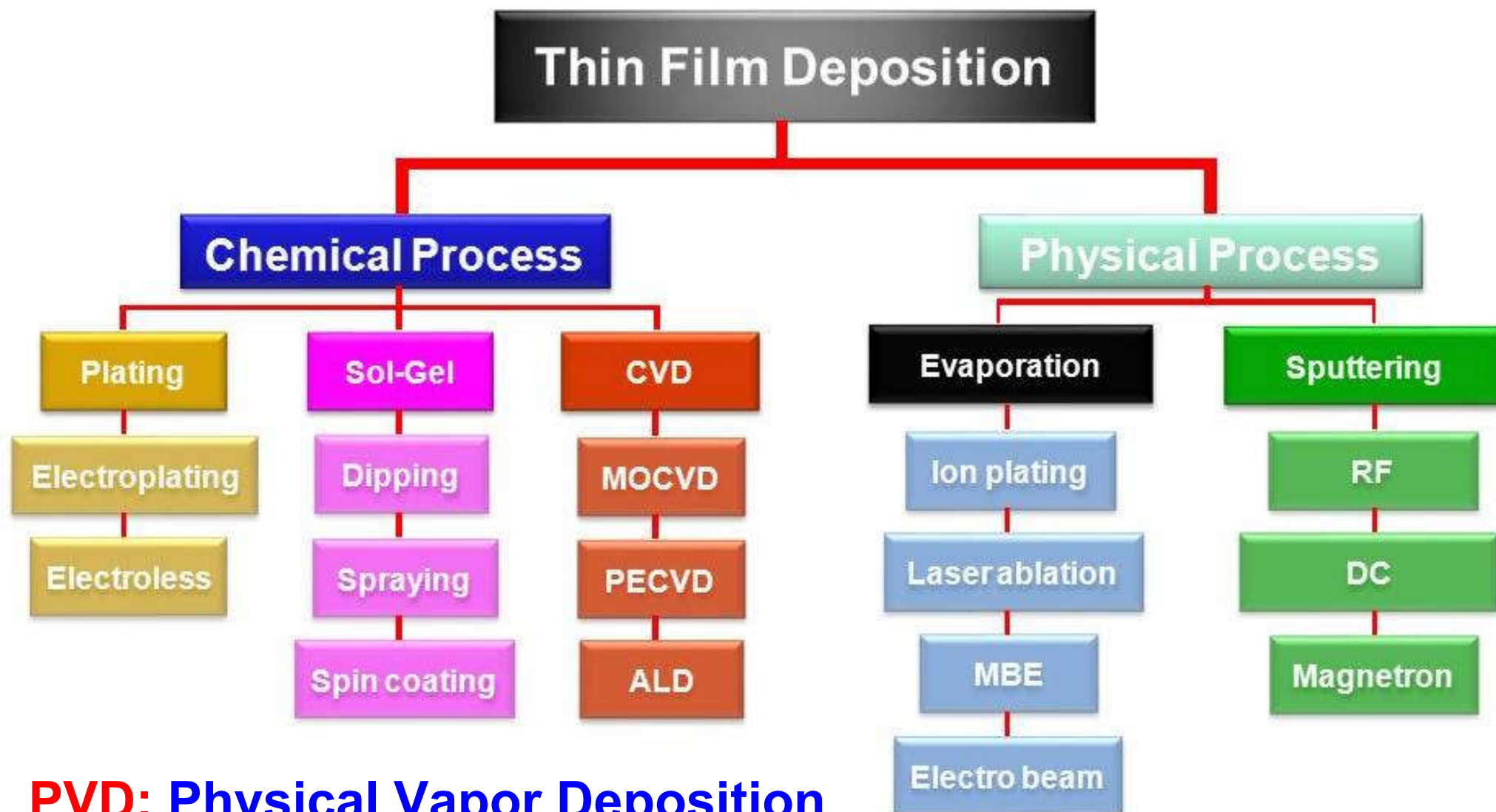
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# Film Deposition



**PVD:** Physical Vapor Deposition

**CVD:** Chemical Vapor Deposition

# Thin Film in CMOS

- CVD

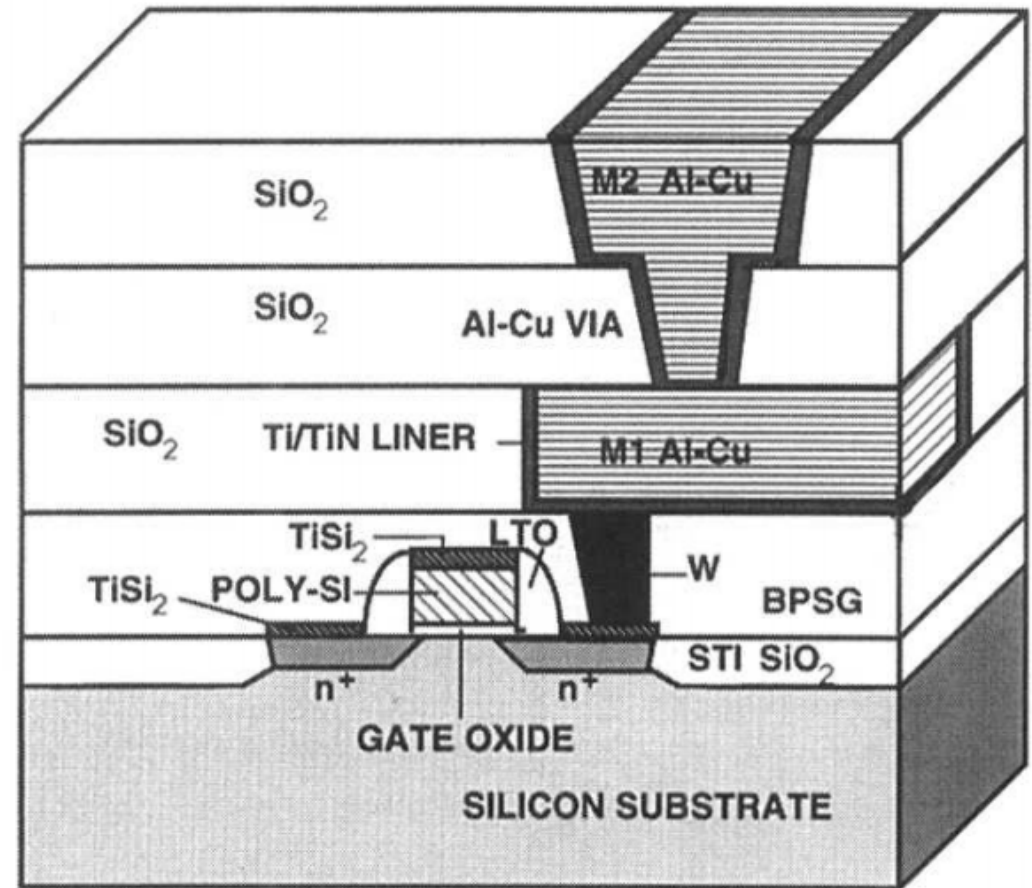
- Si
- poly-Si
- W, SiO<sub>2</sub>, ...

- PVD

- Al, Ti
- ...

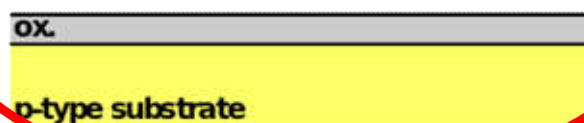
- Electrodeposition

- Cu

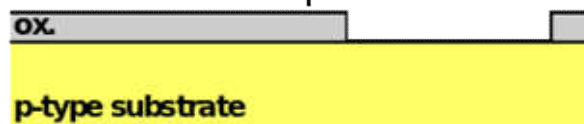


# CMOS Transistors

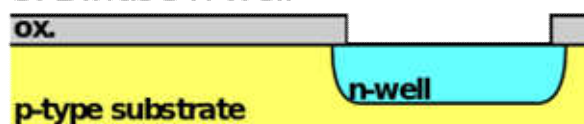
1. Grow field oxide



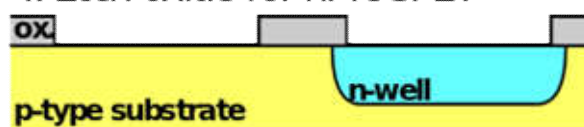
2. Etch oxide for pMOSFET



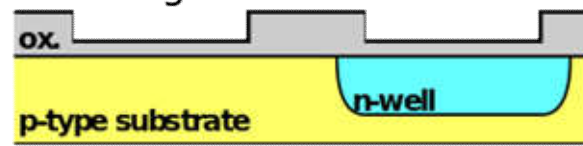
3. Diffuse n-well



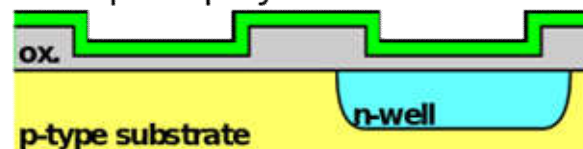
4. Etch oxide for nMOSFET



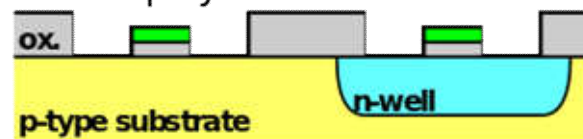
5. Grow gate oxide



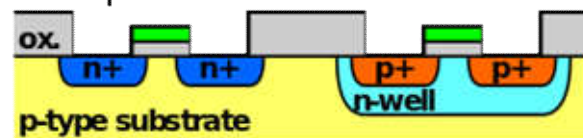
6. Deposit polysilicon



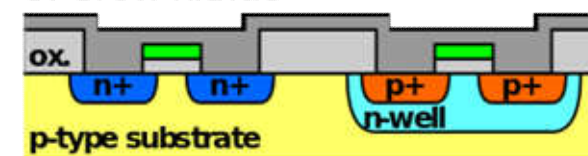
7. Etch polysilicon and oxide



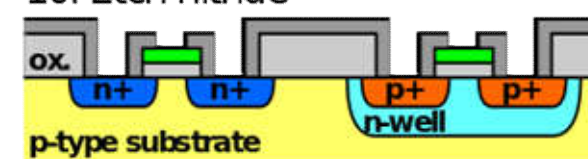
8. Implant sources and drains



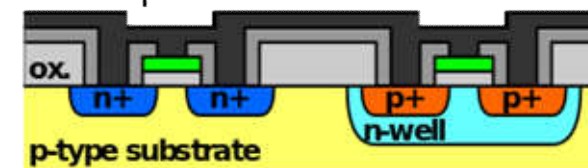
9. Grow nitride



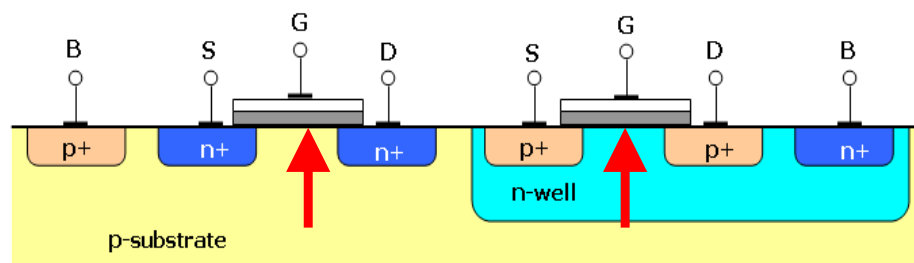
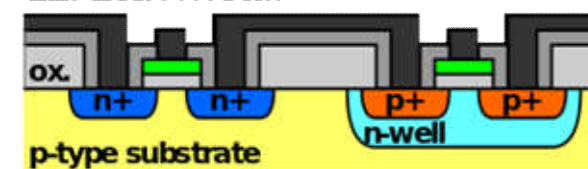
10. Etch nitride



11. Deposit metal



12. Etch metal



# Properties of SiO<sub>2</sub>

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- **Very stable**
  - for Ge, GeO<sub>2</sub> is soluble in water, and decompose at 450 °C
  - for GaAs, GaO<sub>x</sub> and AsO<sub>x</sub> have many defects
- **Easily etched**
  - wet etch (HF solution) or dry etch (F based plasma)
- **Good diffusion barrier (low dopant diffusivity  $D_{ox} \ll D_{Si}$ )**
- **High quality insulator**
  - band gap ~ 8 eV, resistivity > 10<sup>16</sup> Ω\*cm
- **High dielectric strength (> 500 V/μm)**
- **Low interface state / defect density (< 10<sup>10</sup> cm<sup>-2</sup>)**

# Properties of SiO<sub>2</sub>

**TABLE 9.3** Properties of Thermal Silicon Dioxide

DC resistivity ( $\Omega \cdot \text{cm}$ ), 25°C	$10^{14}$ – $10^{16}$	Melting point (°C)	~1700
Density ( $\text{g}/\text{cm}^3$ )	2.27	Molecular weight	60.08
Dielectric constant	3.8–3.9	Molecules/ $\text{cm}^3$	$2.3 \times 10^{22}$
Dielectric strength ( $\text{V}/\text{cm}$ )	$5$ – $10 \times 10^6$	Refractive index	1.46
Energy gap (eV)	~8	Specific heat ( $\text{J}/\text{g} \cdot ^\circ\text{C}$ )	1.0
Etch rate in buffered HF ( $\text{nm}/\text{min}$ ) <sup>a</sup>	100	Stress in film on Si ( $\text{N}/\text{m}^2$ )	$2$ – $4 \times 10^8$ Compression
Infrared absorption peak ( $\mu\text{m}$ )	9.3		
Linear expansion coefficient ( $^\circ\text{C}^{-1}$ )	$5.0 \times 10^{-7}$	Thermal conductivity ( $\text{W}/\text{cm} \cdot ^\circ\text{C}$ )	0.014

Source: After Wolf and Tauber (1986).

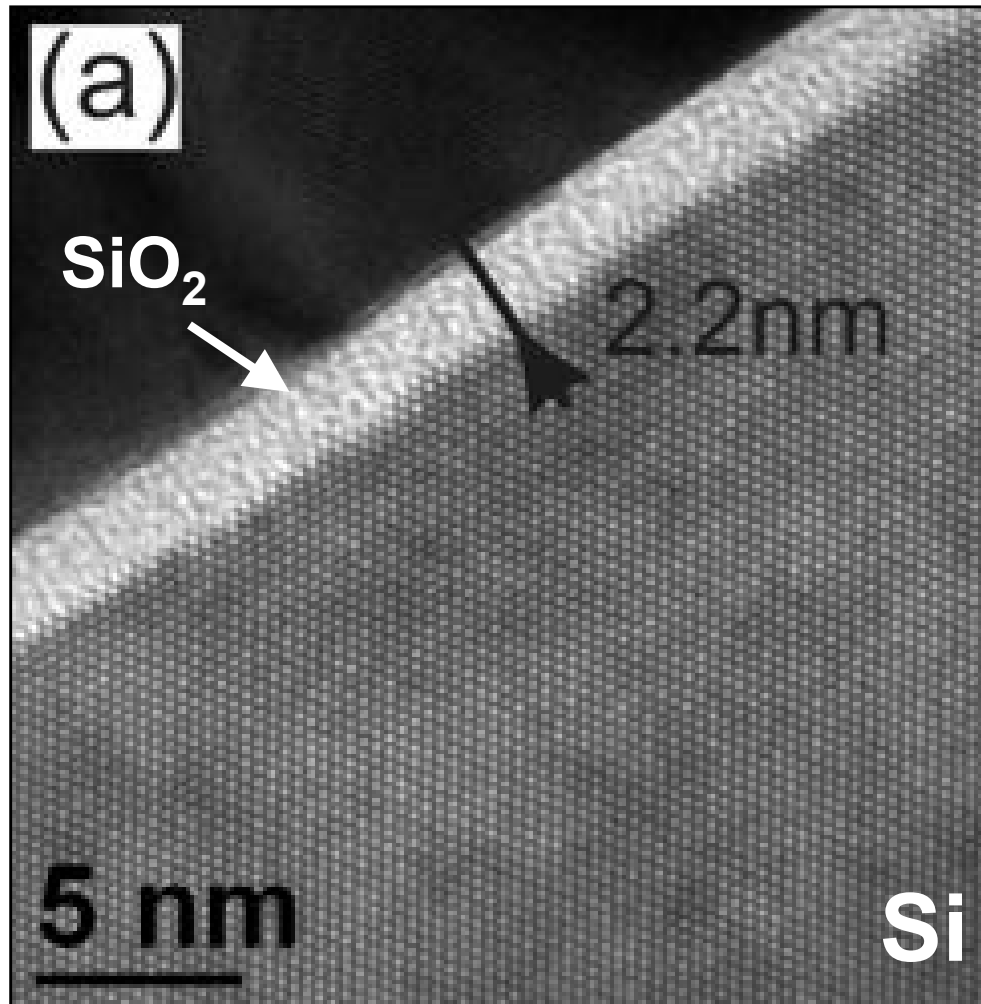
<sup>a</sup>Buffered HF: 28 ml HF, 170 ml H<sub>2</sub>O, 113 g NH<sub>4</sub>F.

**Table 7.2** Diffusivities of Elements in SiO<sub>2</sub><sup>a</sup>

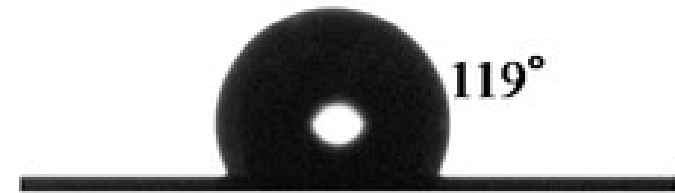
Element	$D$ at 1100°C ( $\text{cm}^2/\text{s}$ )	$D$ at 1200°C ( $\text{cm}^2/\text{s}$ )
B	$3 \times 10^{-17}$ to $2 \times 10^{-14}$	$2 \times 10^{-16}$ to $5 \times 10^{-14}$
Ga	$5.3 \times 10^{-11}$	$5 \times 10^{-8}$
P	$2.9 \times 10^{-16}$ to $2 \times 10^{-13}$	$2 \times 10^{-15}$ to $7.6 \times 10^{-13}$
Sb	$9.9 \times 10^{-17}$	$1.5 \times 10^{-14}$
Ar	$1.2 \times 10^{-16}$ to $3.5 \times 10^{-15}$	$2 \times 10^{-15}$ to $2.4 \times 10^{-14}$



# Native Oxide



clean Si (oxide removed by HF)  
hydrophobic



Si with native oxide  
hydrophilic



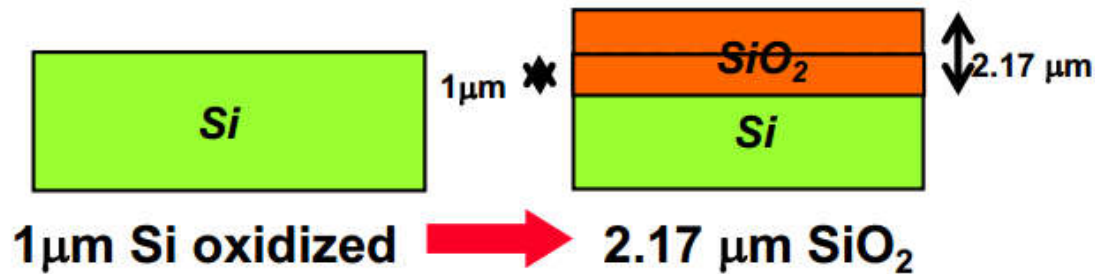
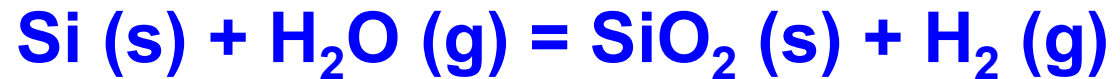
Si forms native oxide in the air (1~2 nm, a few hours)

*Q: amorphous or crystalline  $\text{SiO}_2$  ?*

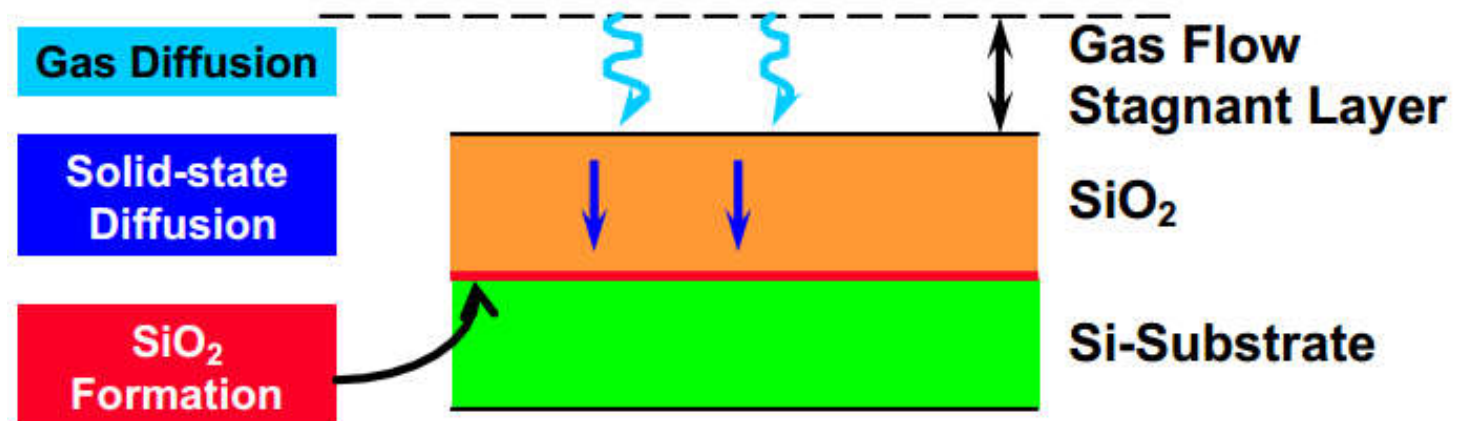
# Thermal Oxide Growth

*dry oxidation*

*wet oxidation*

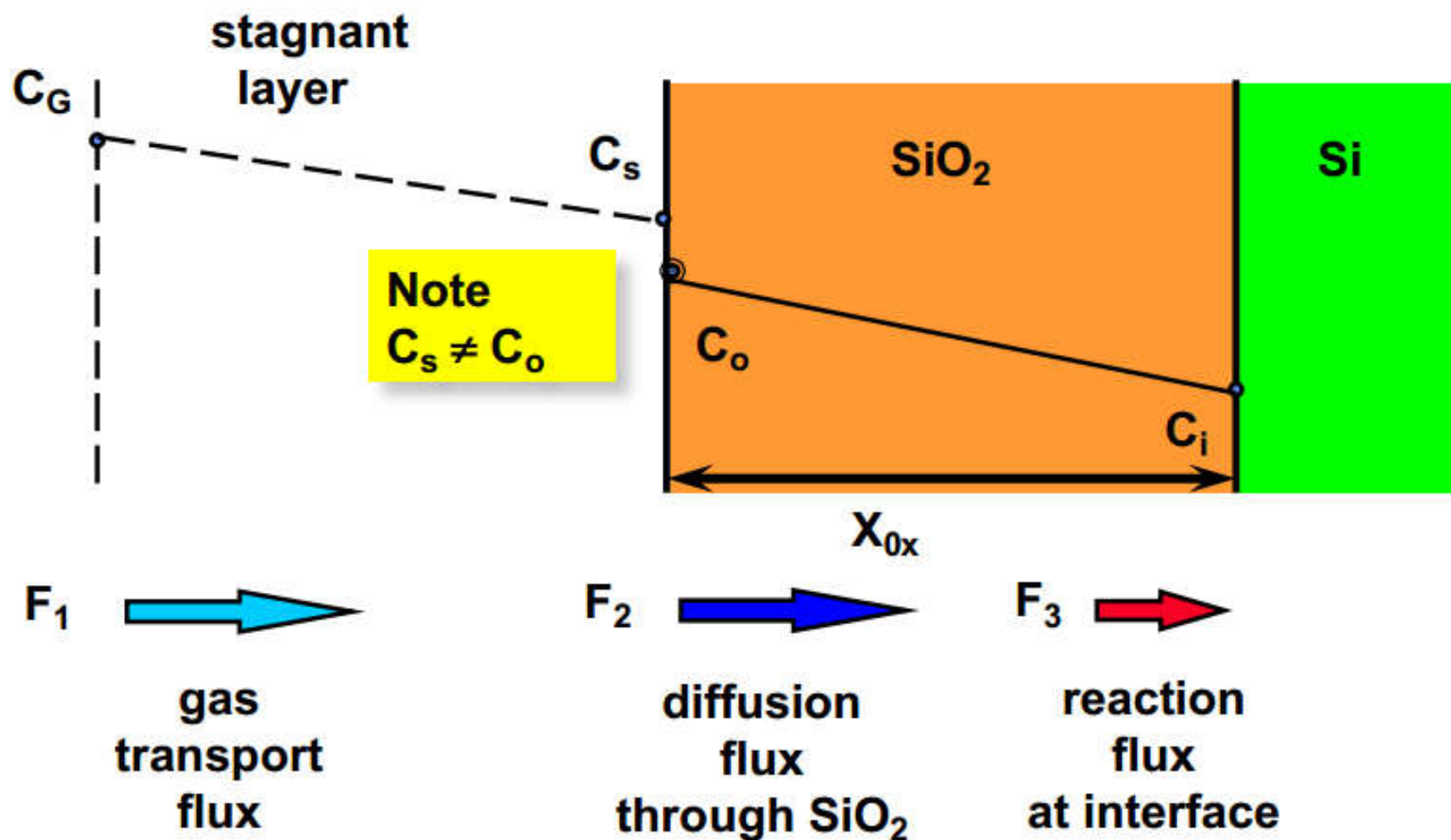


[Video](#)





# The Deal-Grove (D-G) Model



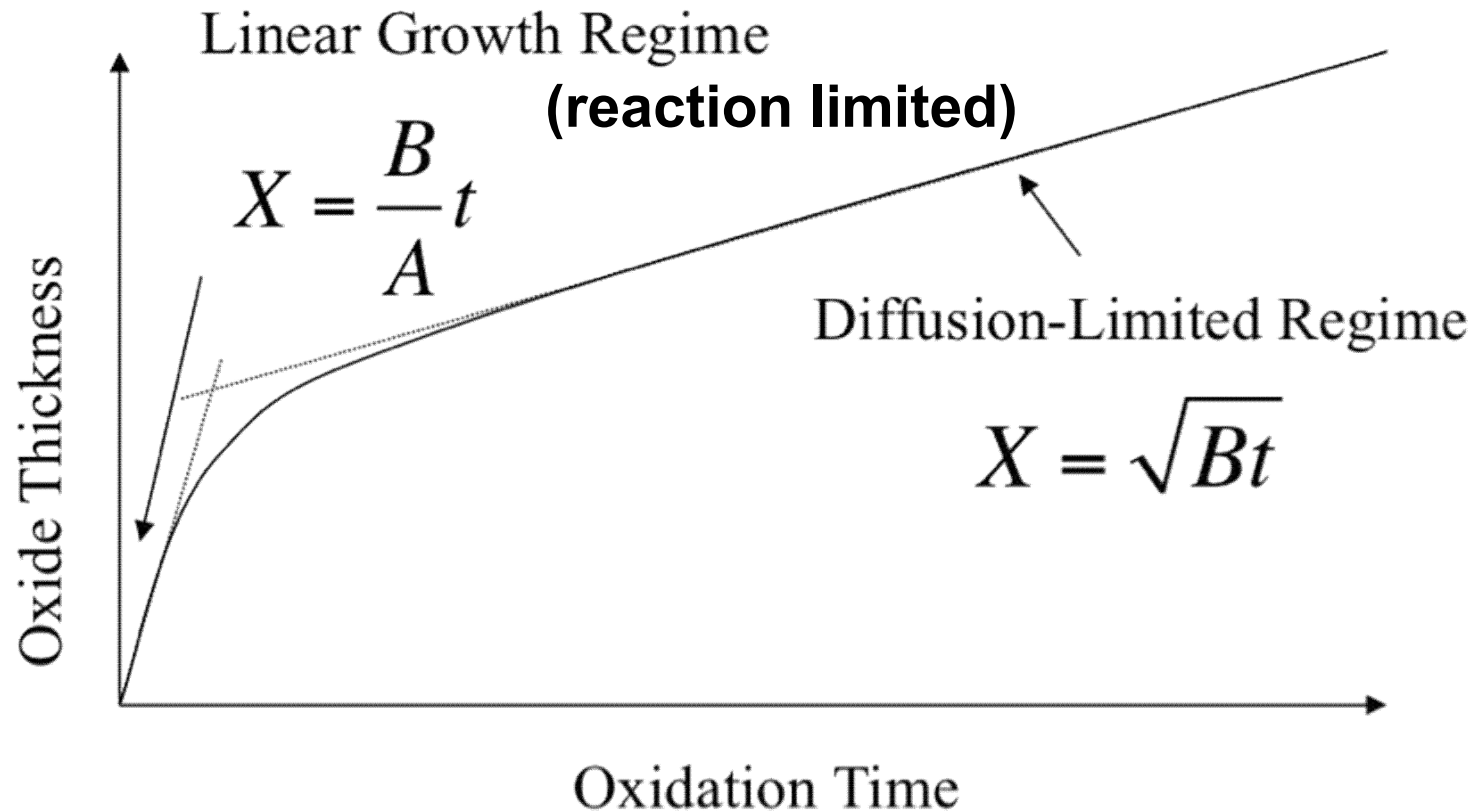
- **F: oxygen flux** – the number of oxygen molecules that crosses a plane per unit area per second

# The Deal-Grove (D-G) Model

$$X_{ox} = \frac{A}{2} \left\{ \sqrt{1 + \left( \frac{t + \tau}{A^2/4B} \right)} - 1 \right\}$$

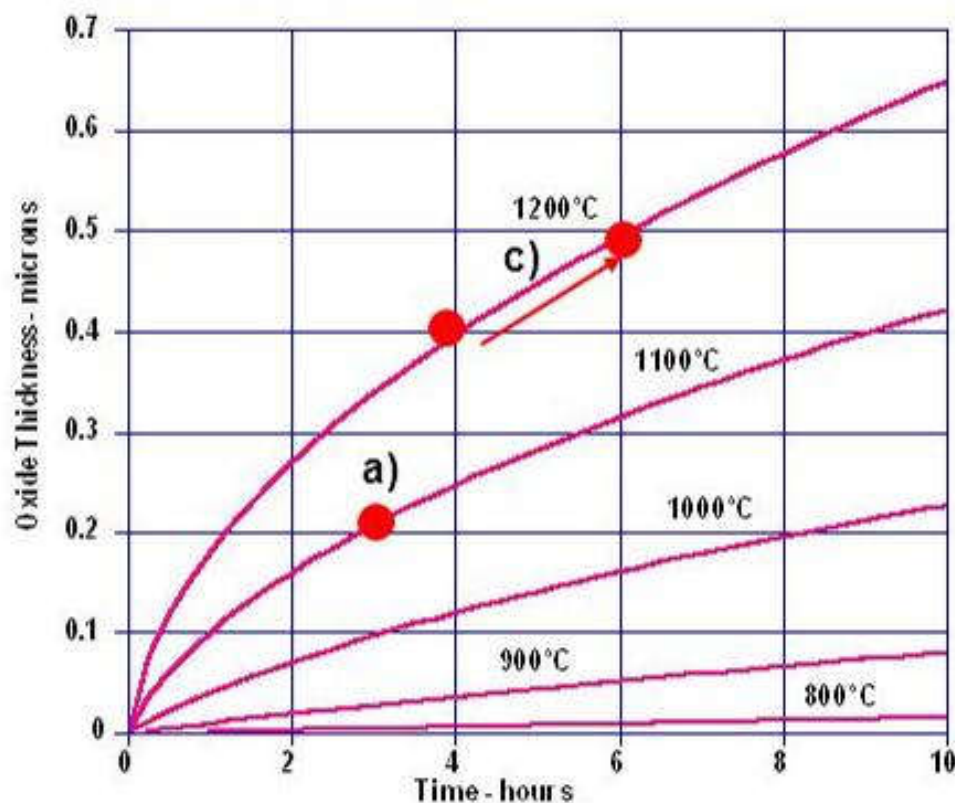
$A$   
 $B$   
 $\tau$

related to reaction  
related to diffusion  
initial native oxide thickness



# Dry vs. Wet Oxidation

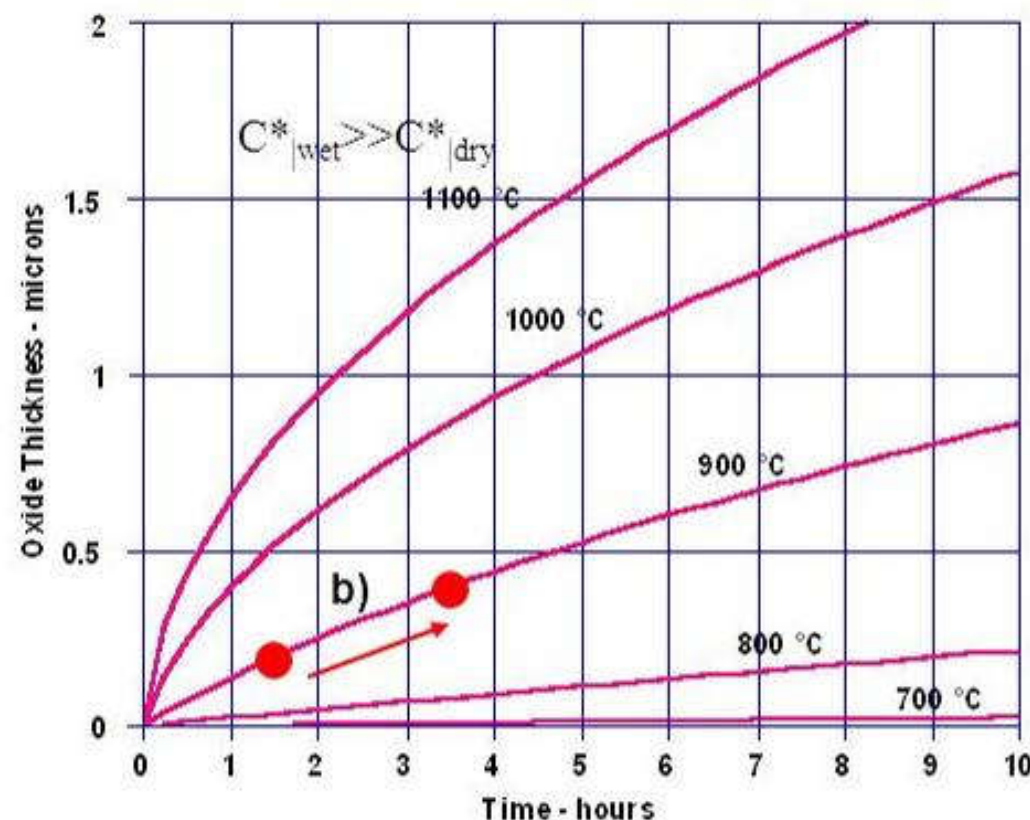
**Dry oxidation** - used up to 100-200 nm



Calculated (100) silicon dry  $O_2$  oxidation rates using Deal Grove.

[Video](#)

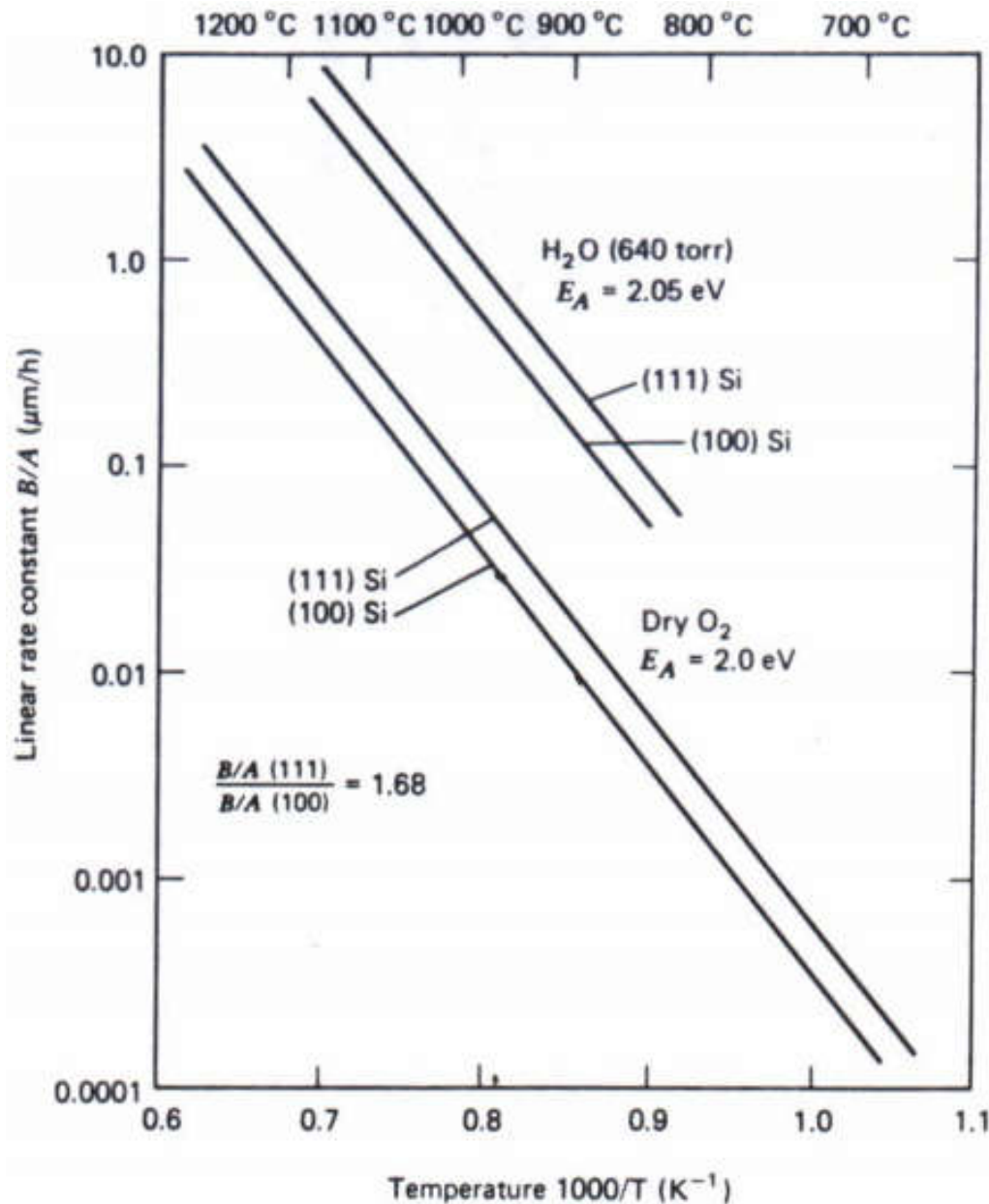
**Wet oxidation** - used for thicker oxides



Calculated (100) silicon  $H_2O$  oxidation rates using Deal Grove.

wet oxidation is 10~100 times faster than dry oxidation because  $H_2O$  has higher solubility/diffusivity in  $SiO_2$

# Crystal Orientation

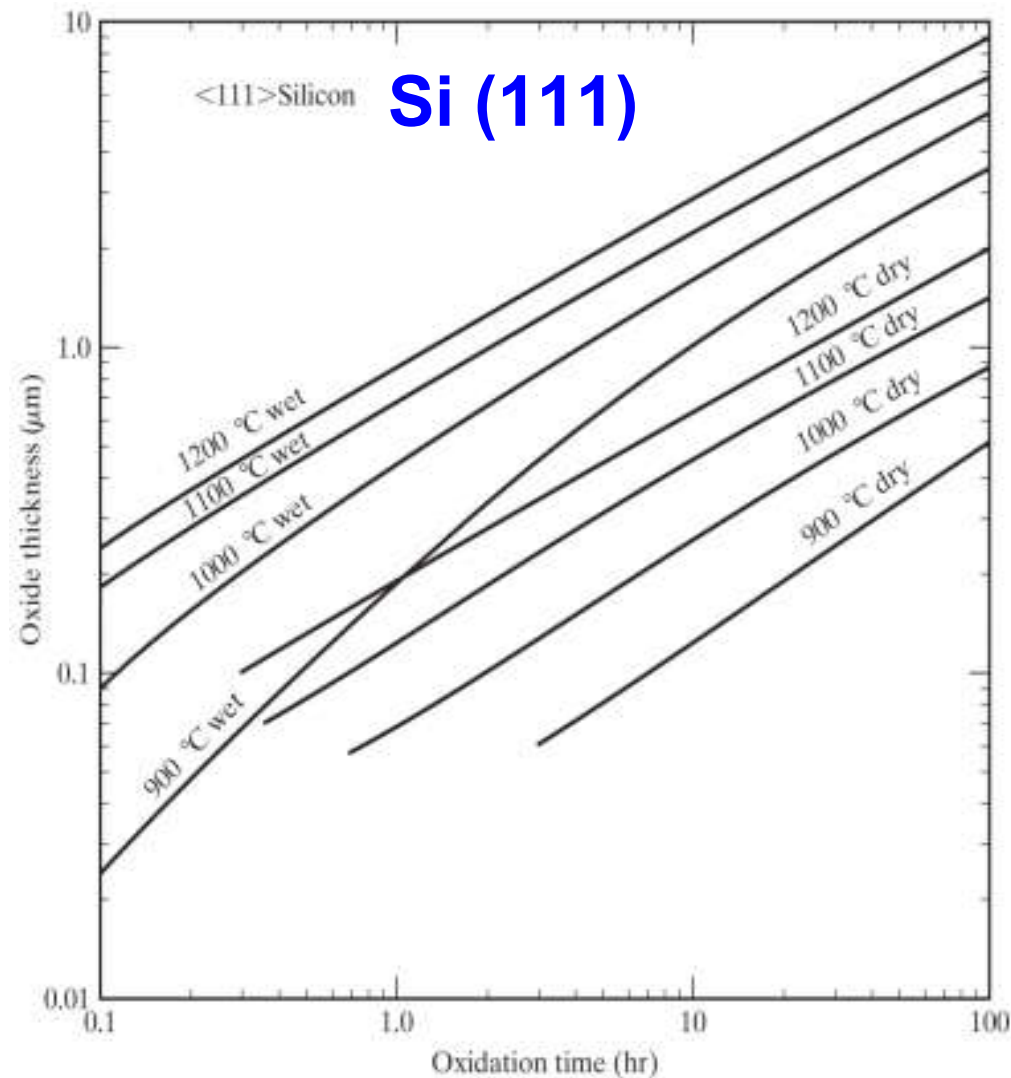
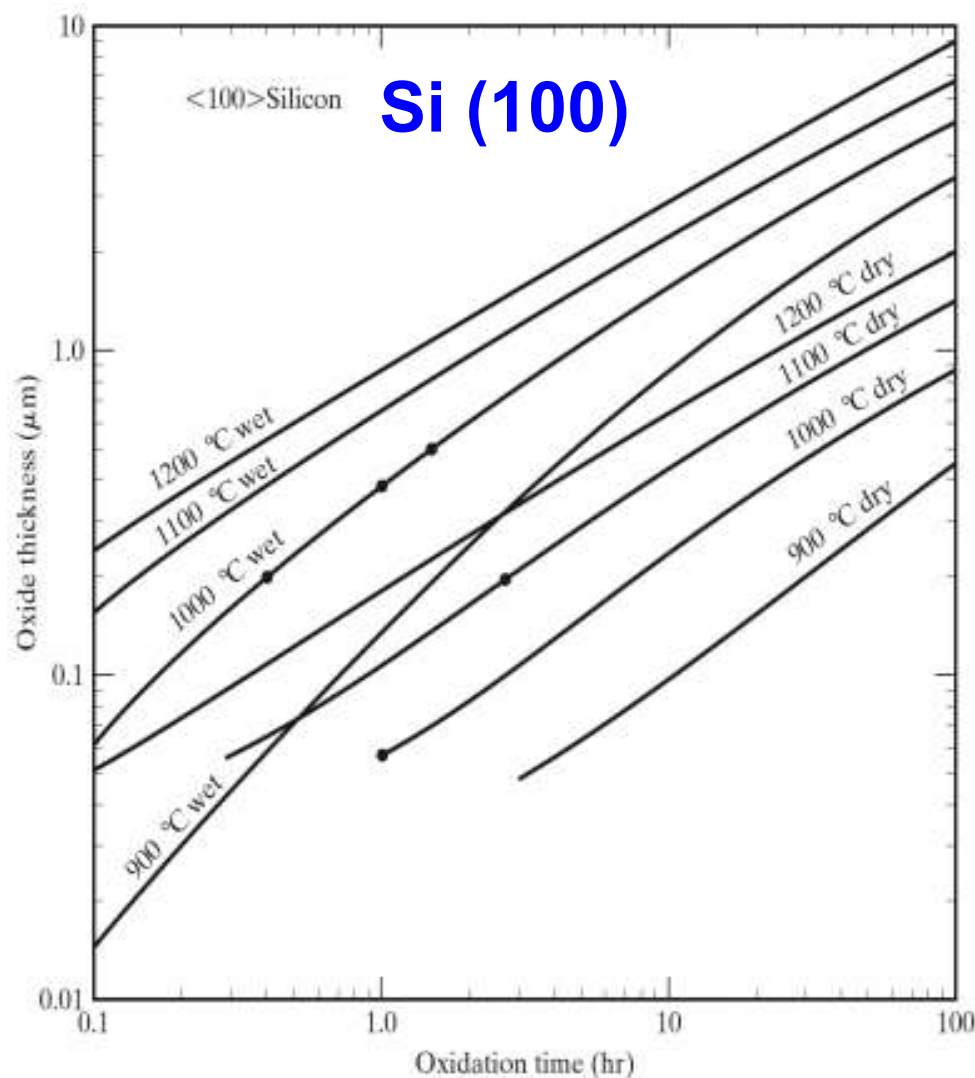


Si (111) has smaller  $A$ ,  
but same  $B$  with Si (100)

higher growth rate at  
initial stage

*why ??*

# Crystal Orientation



**similar rates at long time oxidation (diffusion limited)**

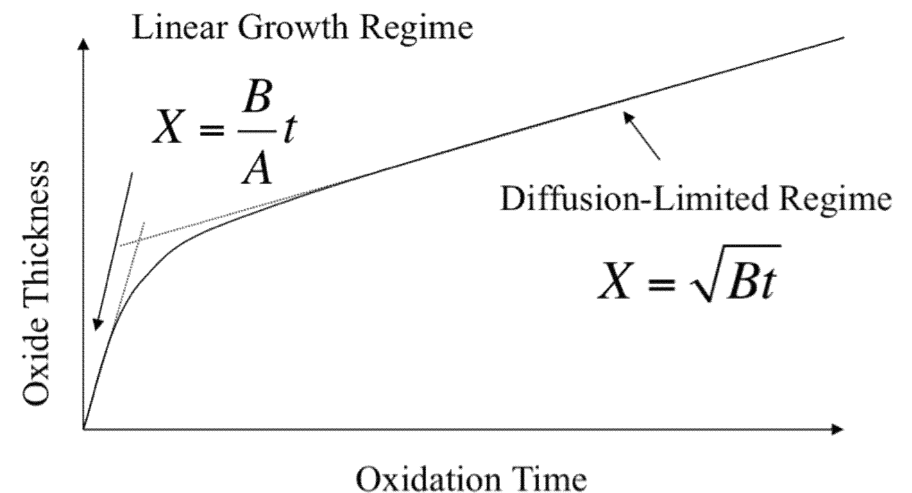


# Thermal Oxidation

## ■ Process Parameters

- ❑ Time
- ❑ Temperature
- ❑ Gas type ( $O_2$ ,  $H_2O$ , ...)
- ❑ Gas pressure
- ❑ Crystal orientation
- ❑ Dopant (B, P, As, ...)

$$X_{ox} = \frac{A}{2} \left\{ \sqrt{1 + \left( \frac{t + \tau}{A^2 / 4B} \right)} - 1 \right\}$$

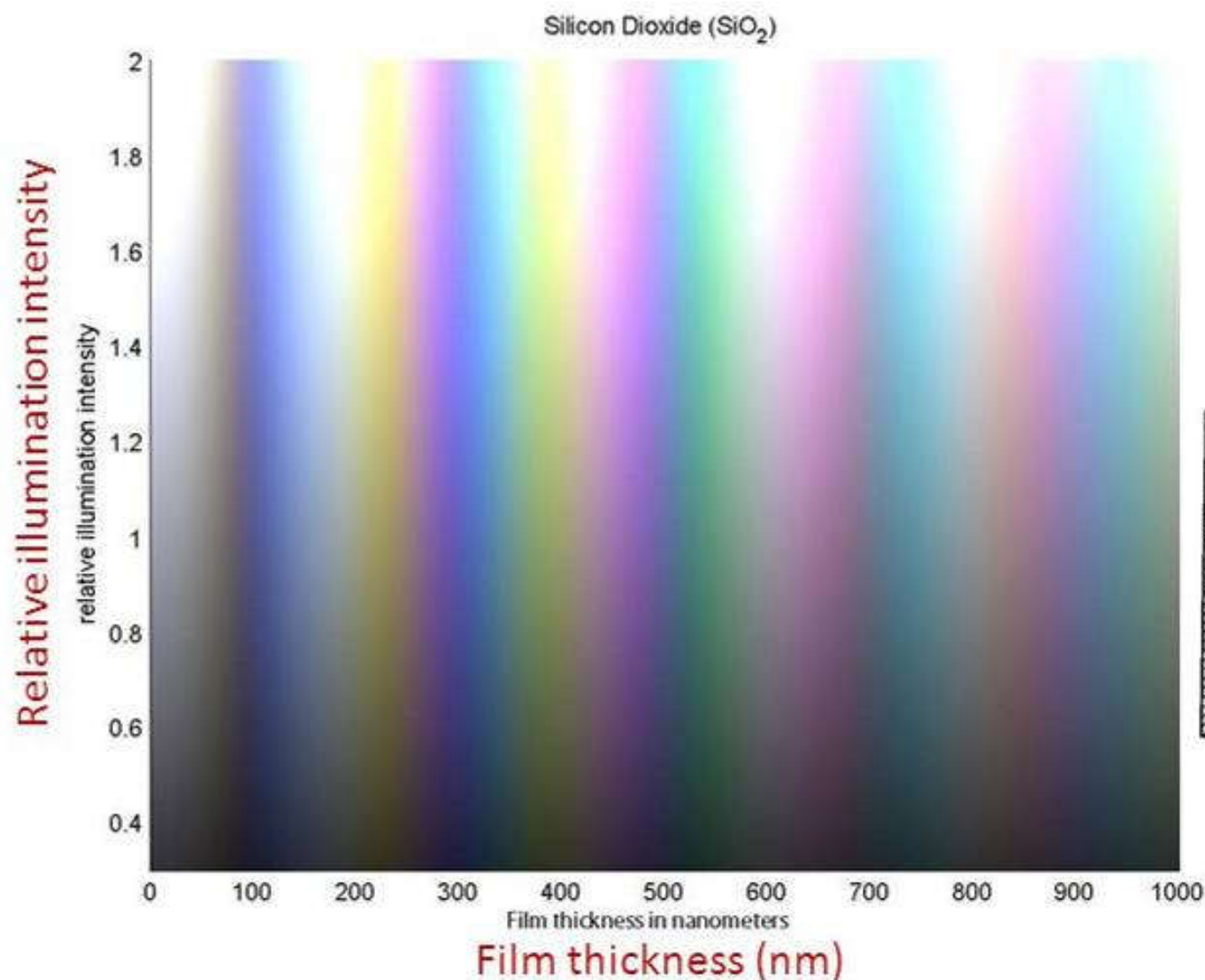


## ■ Control Parameters

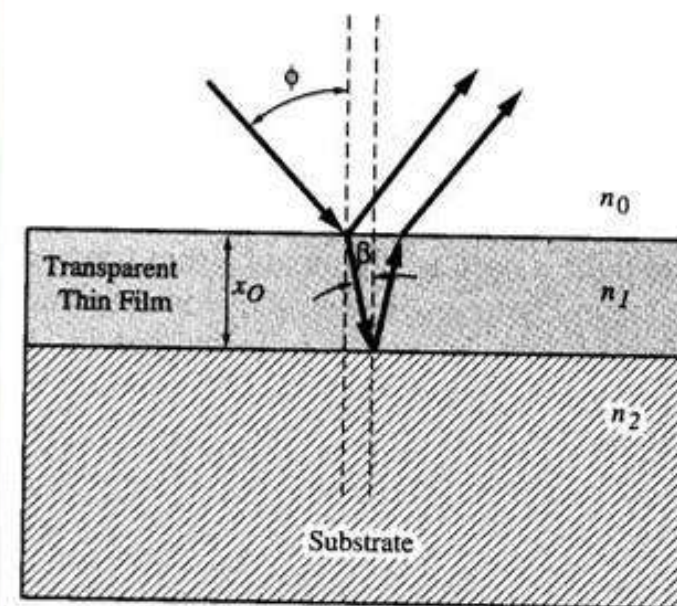
- ❑ Oxide thickness
- ❑ Film quality (defects, dielectric strength, ...)



# SiO<sub>2</sub> Film Thickness Measurement

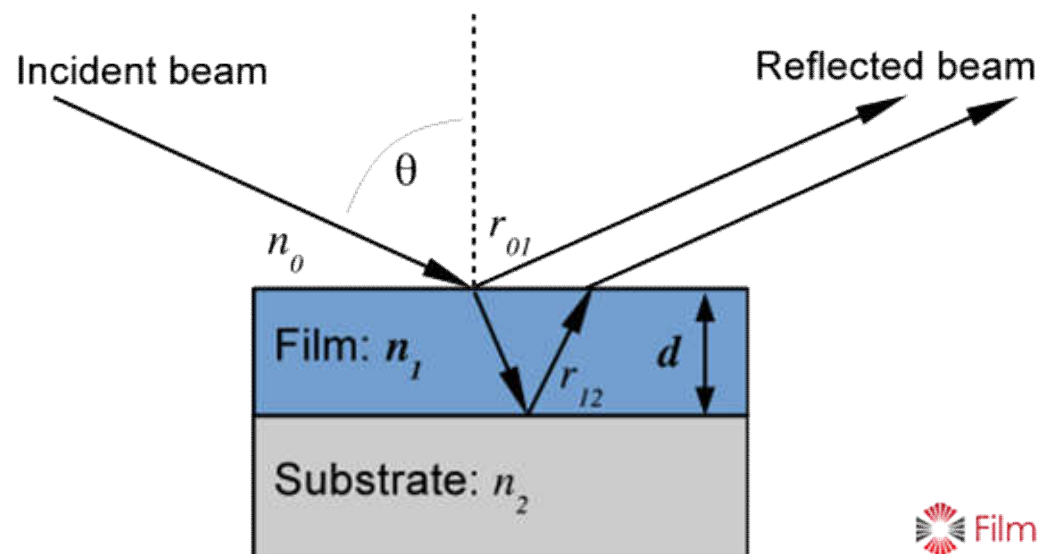


color difference



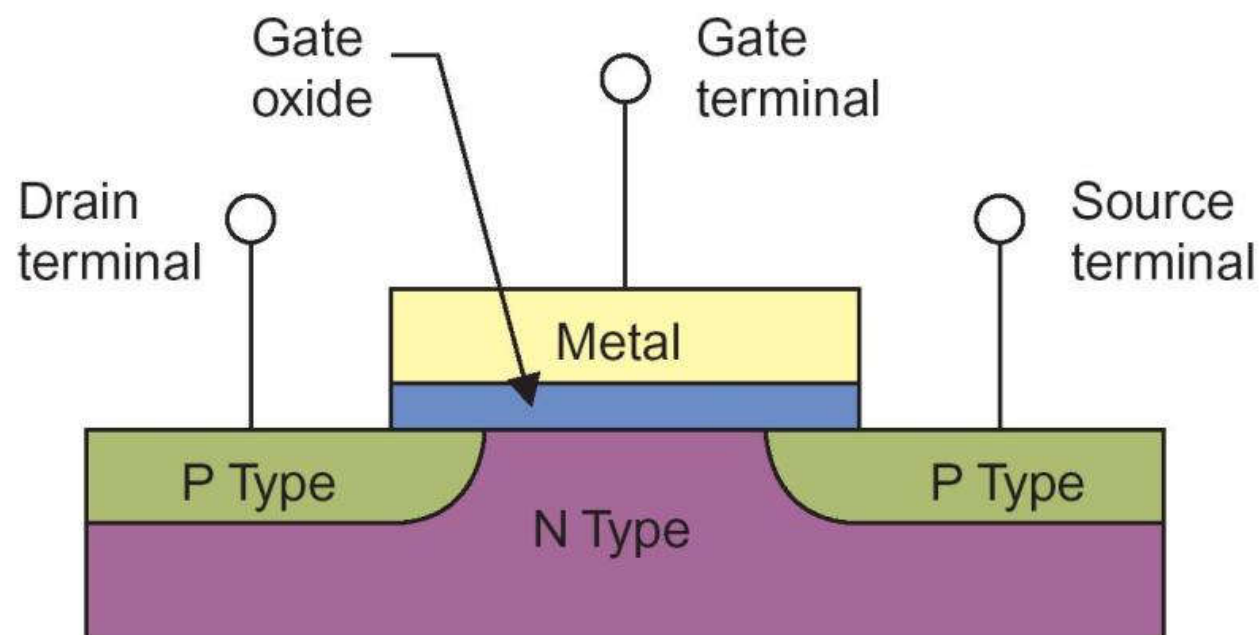
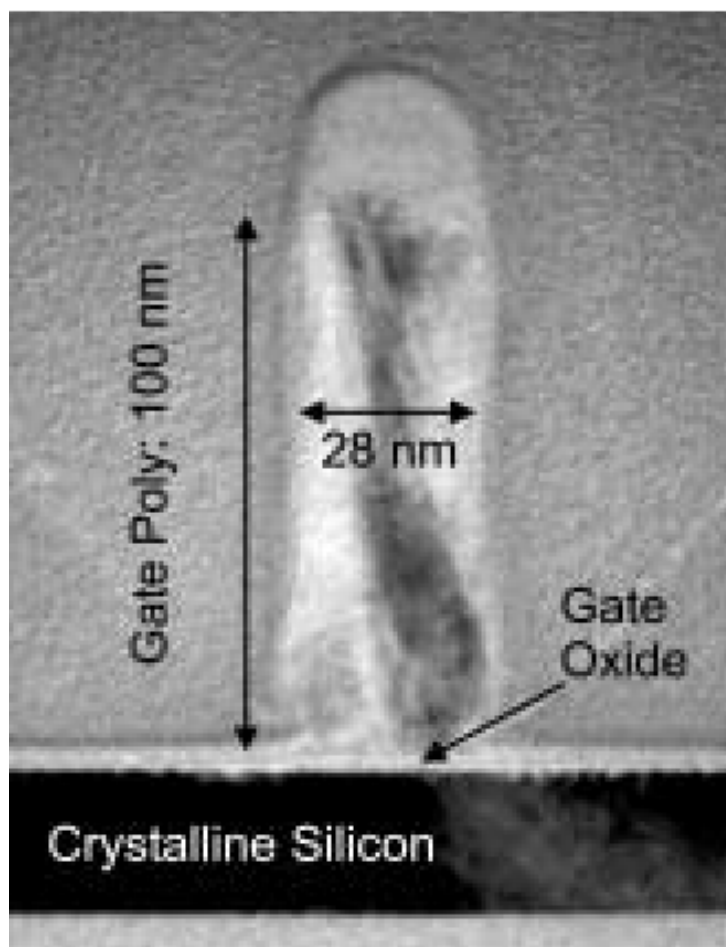
# SiO<sub>2</sub> Film Thickness Measurement

## Spectroscopic Ellipsometer



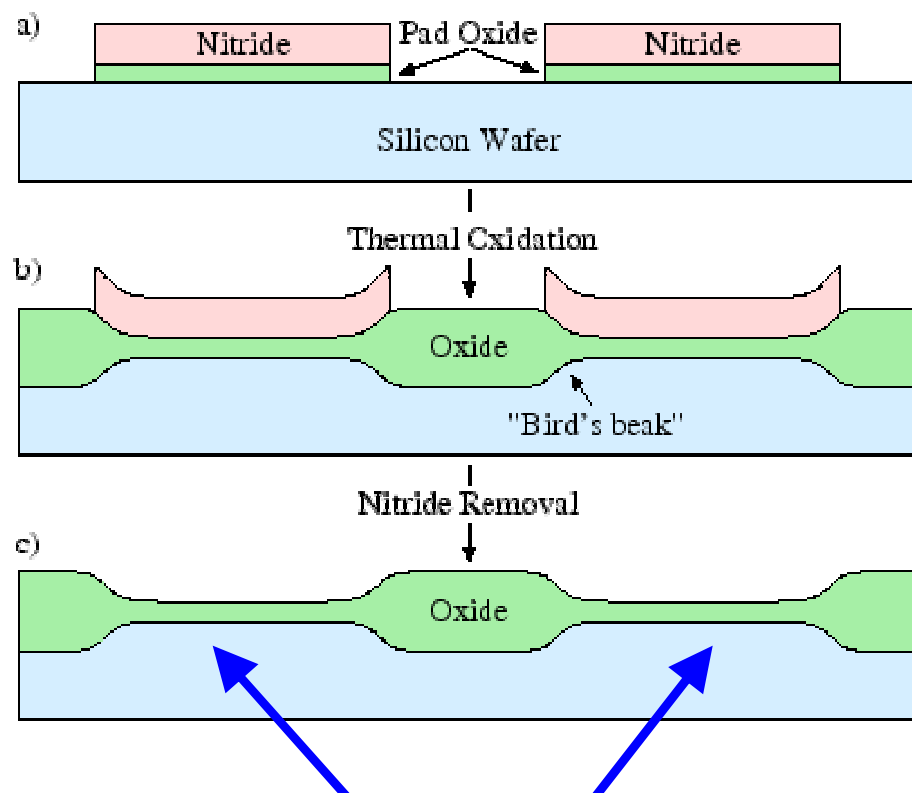
# SiO<sub>2</sub> in CMOS

## gate oxide for transistors

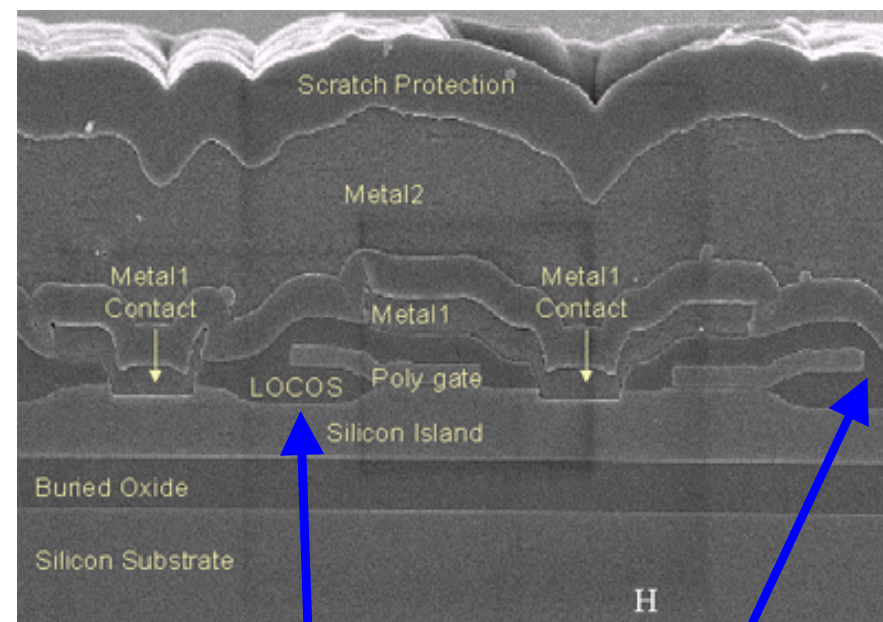


# SiO<sub>2</sub> in CMOS

## Local Oxidation of Si (LOCOS)



transistor region

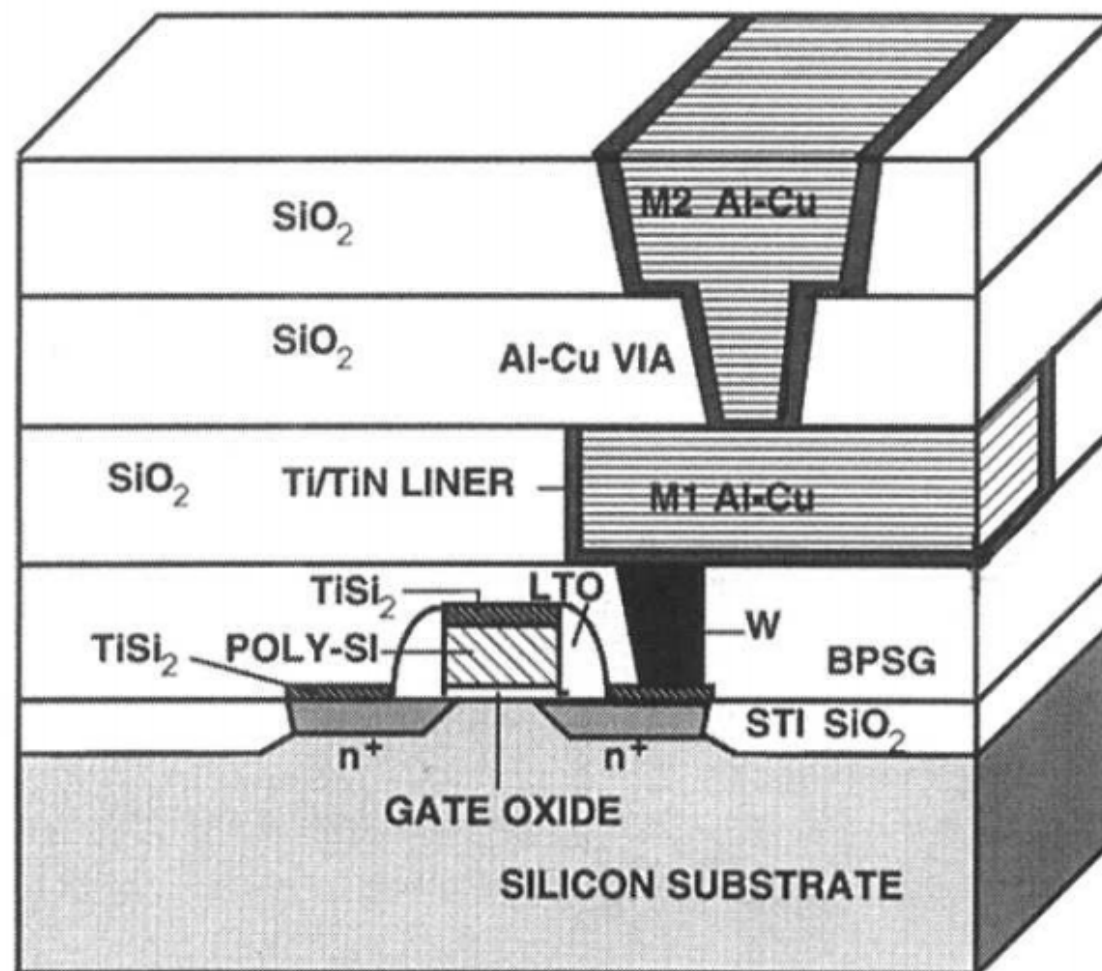


For isolation

# SiO<sub>2</sub> in CMOS

other methods to  
deposit SiO<sub>2</sub>

temperature



*Q: why?*



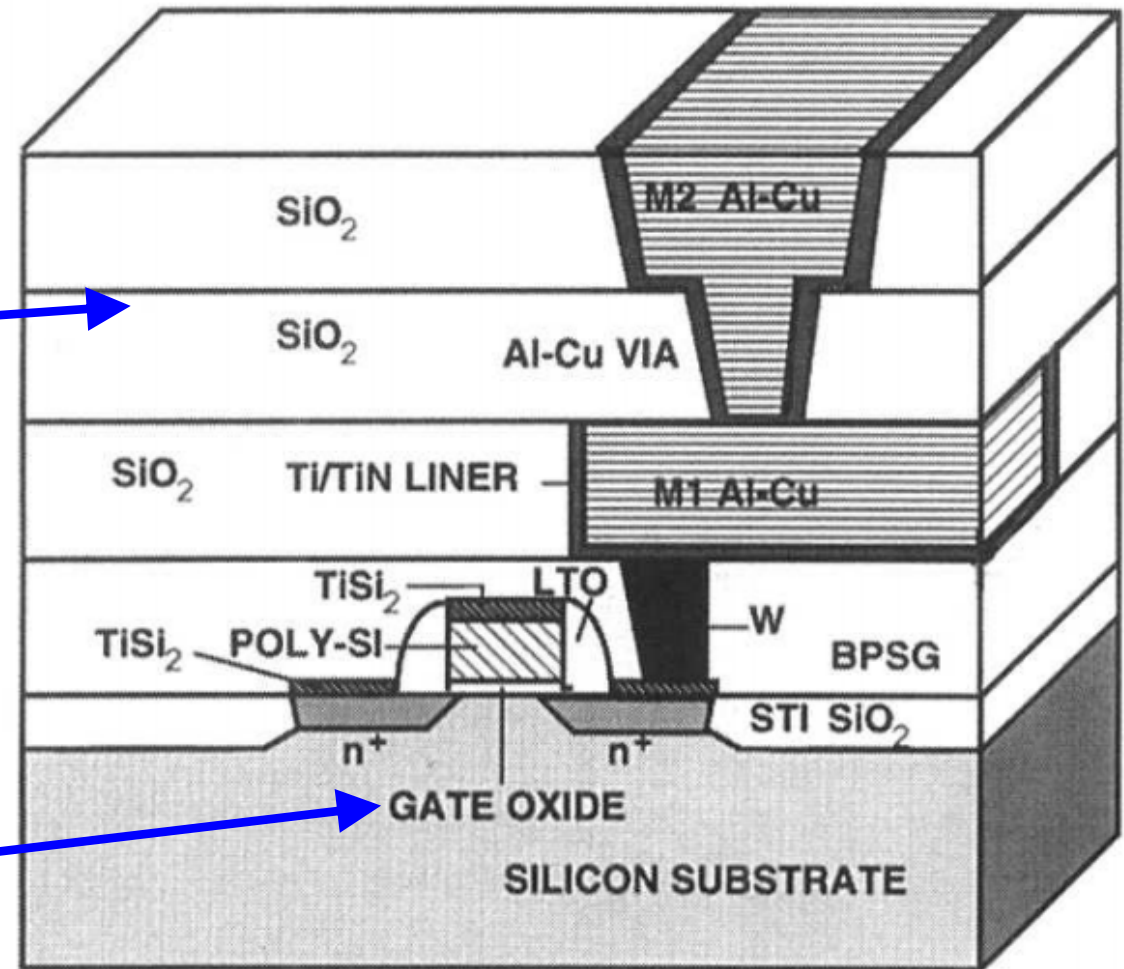
# SiO<sub>2</sub> in CMOS

$$C = \frac{\kappa \epsilon_0 A}{t}$$

low  $\kappa$  dielectric  
for insulating  
reduce *RC* delay

high  $\kappa$  dielectric  
for gate oxide

$$I_{D,Sat} = \frac{W}{L} \mu C \frac{(V_G - V_{th})^2}{2}$$





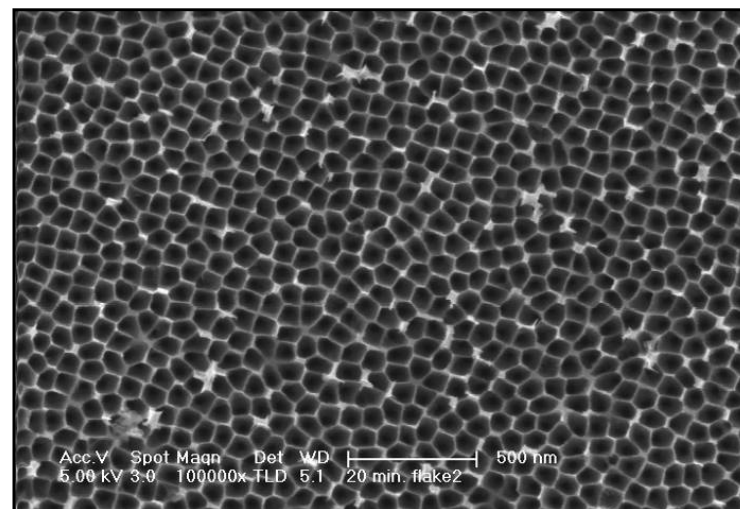
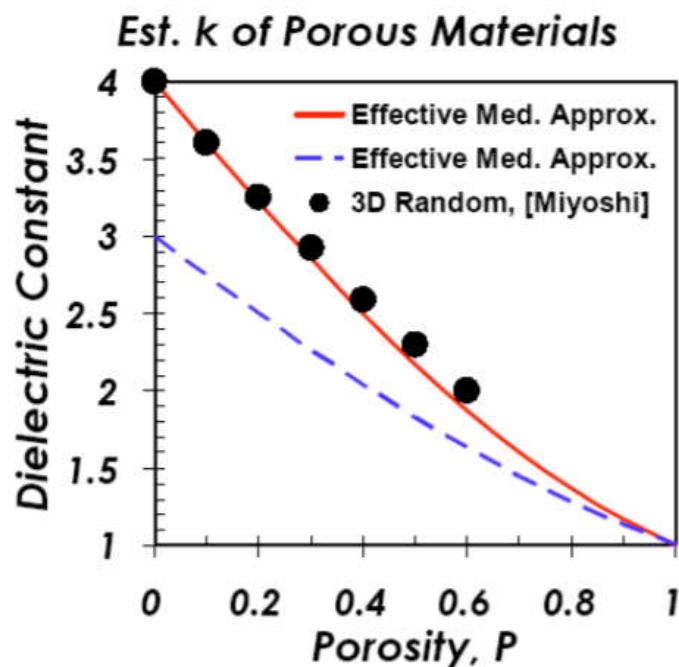
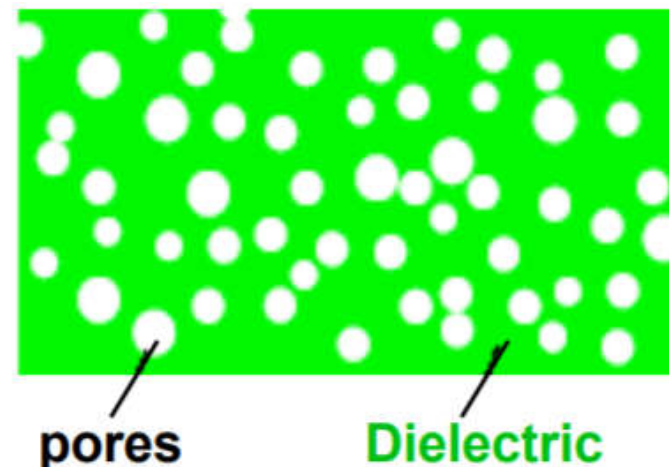
# Porous SiO<sub>2</sub> for Low $\kappa$ Dielectric

SiO<sub>2</sub>

$\kappa = 3.9$

air

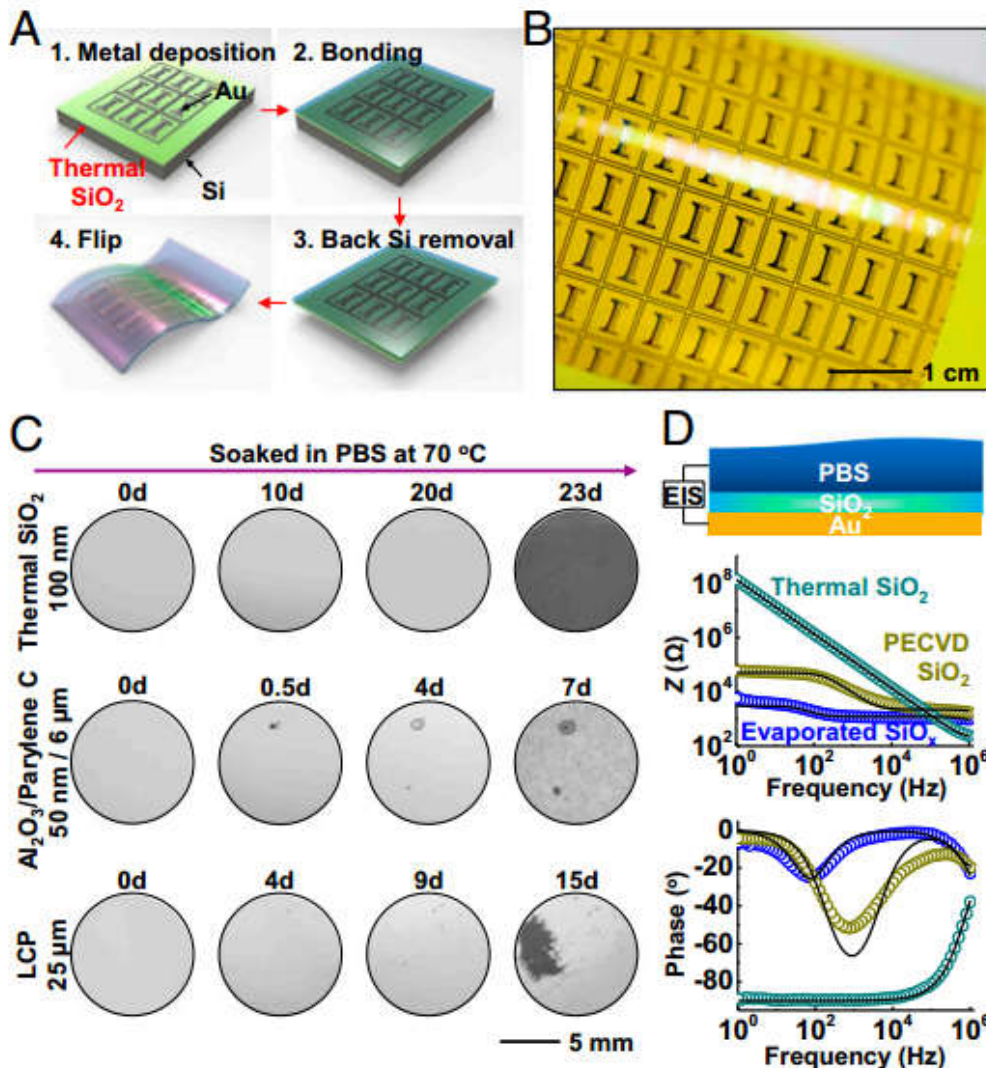
$\kappa = 1.0$



# SiO<sub>2</sub> in Biointegrated Devices

Thermal oxide is the best moisture barrier

useful for implantable devices



At room temperature, it will take **> 100 years** to dissolve **1 μm thermal SiO<sub>2</sub>** in water

***Thank you for your attention***