Fundamentals of Solid State Physics

Electronic Band - Summary

Xing Sheng 盛 兴

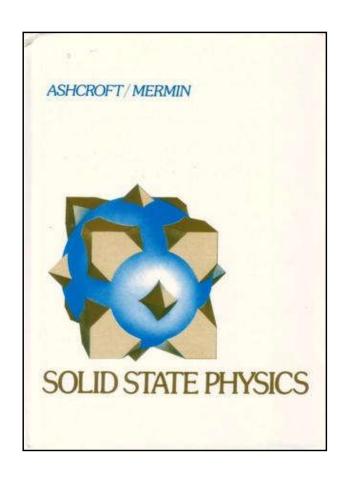


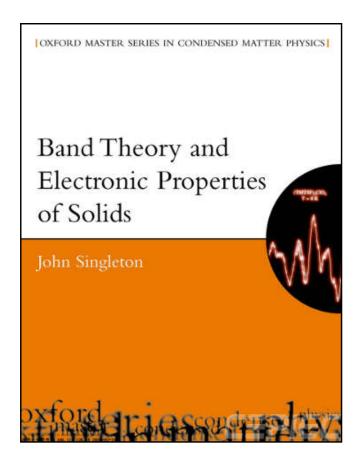
Department of Electronic Engineering Tsinghua University

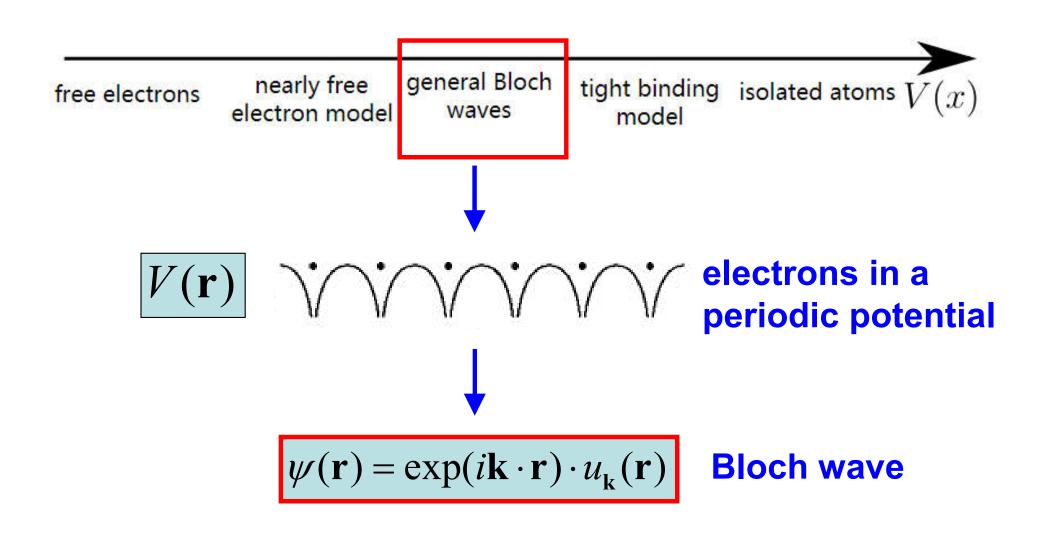
xingsheng@tsinghua.edu.cn

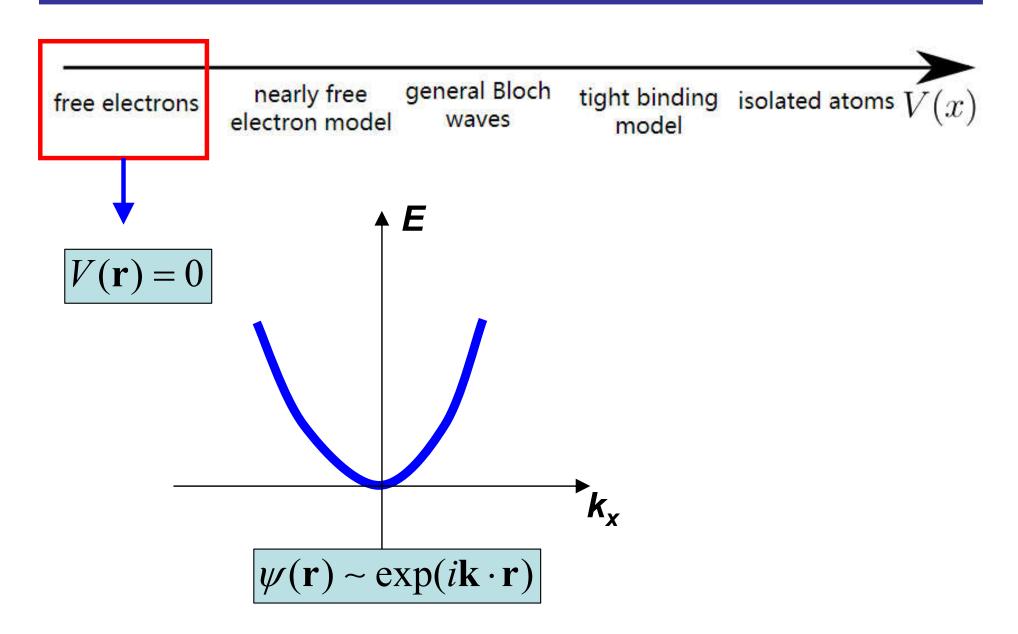
Further Reading

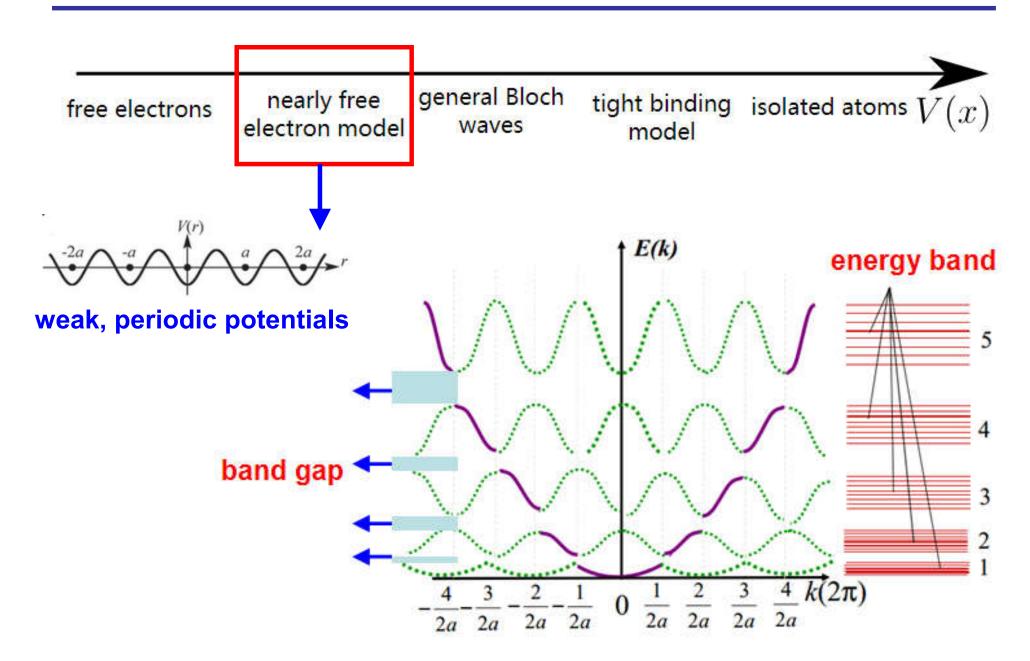
- Singleton, Chapter 5
- Ashcroft & Mermin, Chapter 12, 13

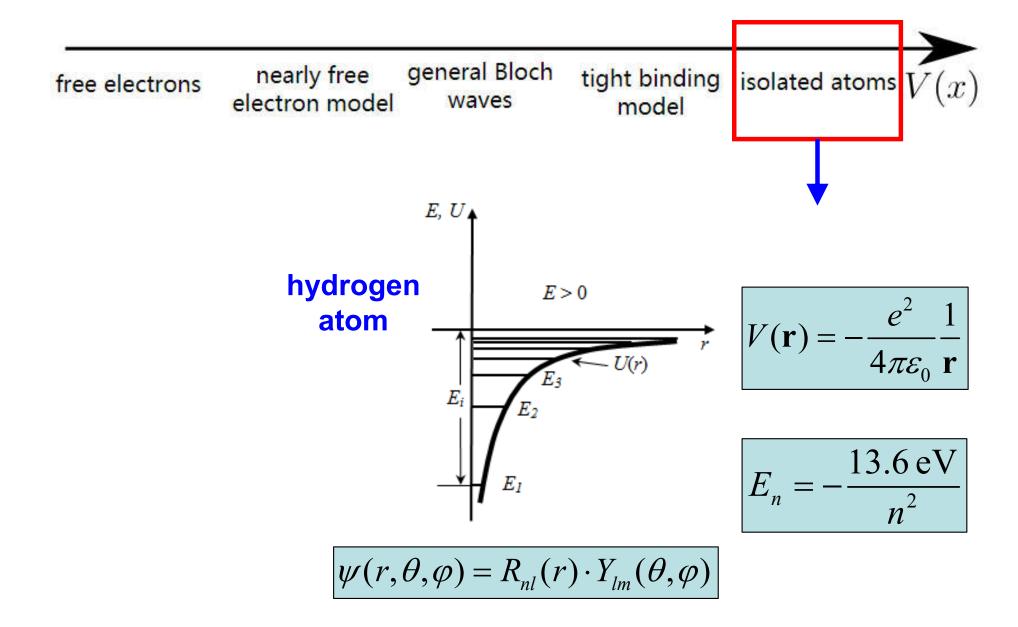


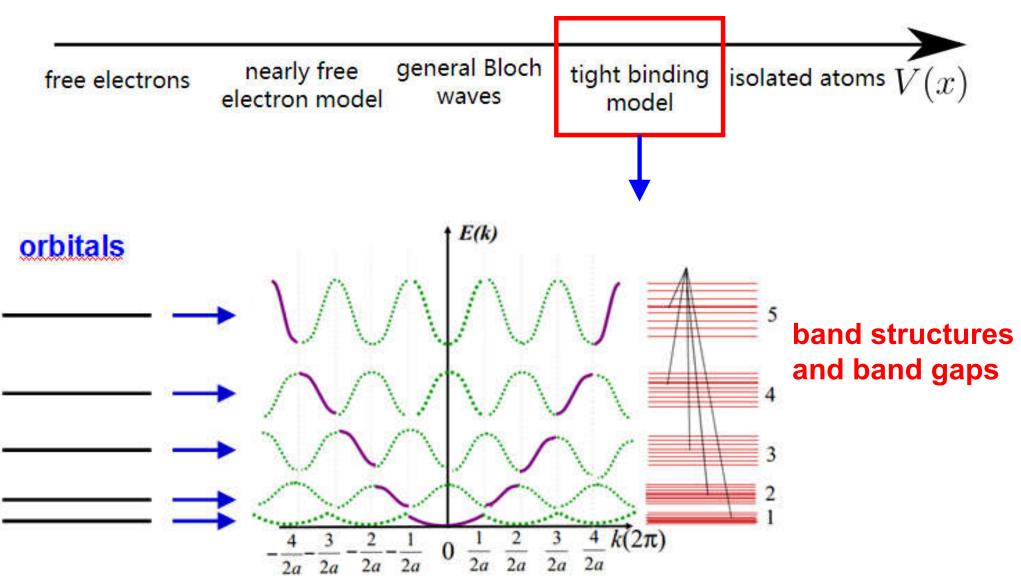










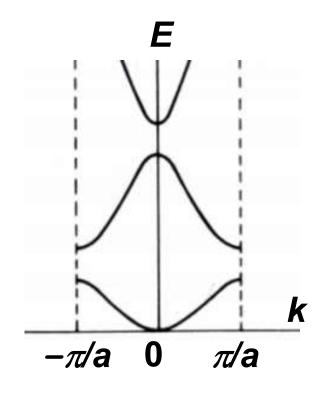


Importance of k

- k is taken from reciprocal space
 - reduced to FBZ

$$k_x = \frac{2\pi n_x}{L_x}$$
 $n_x = 0, \pm 1, \pm 2, \dots$

• $p = \hbar k$ is not electron momentum, is crystal momentum (take the system as a whole)



- group velocity
 - □ velocity of a wave packet (波包)

$$v = \frac{1}{\hbar} \frac{dE(k)}{dk}$$

or $\mathbf{v} = \frac{1}{\hbar} \nabla_{\mathbf{k}} E(\mathbf{k})$

Sommerfeld vs. Bloch

		SOMMERFELD	BLOCH
	QUANTUM NUMBERS (EXCLUDING SPIN)	k (hk is the momentum.)	k, n (h k is the crystal momentum and n is the band index.)
	RANGE OF QUANTUM NUMBERS	k runs through all of k-space consistent with the Born-von Karman periodic boundary condition.	For each n , k runs through all wave vectors in a single primitive cell of the reciprocal lattice consistent with the Born-von Karman periodic boundary condition; n runs through an infinite set of discrete values.
	ENERGY	$\mathcal{E}(\mathbf{k}) = \frac{\hbar^2 k^2}{2m}.$	For a given band index n , $\mathcal{E}_n(\mathbf{k})$ has no simple explicit form. The only general property is periodicity in the reciprocal lattice: $\mathcal{E}_n(\mathbf{k} + \mathbf{K}) = \mathcal{E}_n(\mathbf{k}).$
	VELOCITY	The mean velocity of an electron in a level with wave vector k is:	The mean velocity of an electron in a level with band index n and wave vector k is:
		$\mathbf{v} = \frac{\hbar \mathbf{k}}{m} = \frac{1}{\hbar} \frac{\partial \mathcal{E}}{\partial \mathbf{k}}.$	$\mathbf{v}_n(\mathbf{k}) = \frac{1}{\hbar} \frac{\partial \mathcal{E}_n(\mathbf{k})}{\partial \mathbf{k}}.$
	WAVE FUNCTION	The wave function of an electron with wave vector k is: $\psi_{\mathbf{k}}(\mathbf{r}) = \frac{e^{i\mathbf{k}\cdot\mathbf{r}}}{V^{1/2}}.$	The wave function of an electron with band index n and wave vector k is: $\psi_{nk}(\mathbf{r}) = e^{i\mathbf{k}\cdot\mathbf{r}}u_{nk}(\mathbf{r})$ where the function u_{nk} has no simple
Ashcroft & Mermin p.214		V	explicit form. The only general property is periodicity in the direct lattice: $u_{nk}(\mathbf{r} + \mathbf{R}) = u_{nk}(\mathbf{r}).$

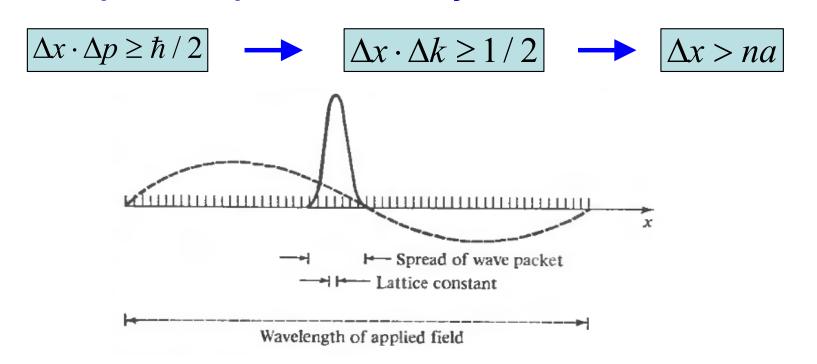
Semi-Classical Model

Electrons are described as wave packets (波包)

$$\psi(\mathbf{r}) = \exp(i\mathbf{k} \cdot \mathbf{r}) \cdot u_{\mathbf{k}}(\mathbf{r})$$

Uncertainty principle

- Ashcroft & Mermin, Chap. 12
- wave packets spread over many cells



Semi-Classical Model

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- Uncertainty principle
 - wave packets spread over many cells

$$\Delta x \cdot \Delta p \ge \hbar / 2 \qquad \longrightarrow \qquad \Delta x \cdot \Delta k \ge 1 / 2 \qquad \longrightarrow \qquad \Delta x > na$$

External electric field varies slowly (DC or AC)

$$\lambda \gg a$$

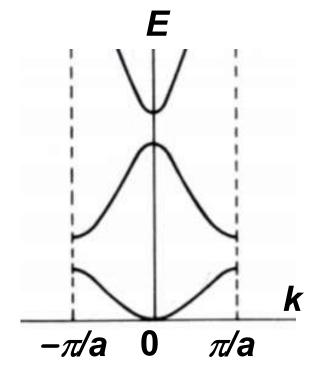
- Collision / relaxation time τ
 - originates from imperfect lattice (vibrations, impurities, defects, etc.)

effective mass

$$\frac{1}{m^*} = \frac{1}{\hbar^2} \frac{d^2 E(k)}{dk^2}$$

For 3D solids, a tensor form

$$\frac{1}{\mathbf{M}^*} = \frac{1}{\hbar^2} \begin{bmatrix} \frac{\partial^2 E}{\partial k_x^2} & \frac{\partial^2 E}{\partial k_x \partial k_y} & \frac{\partial^2 E}{\partial k_x \partial k_z} \\ \frac{\partial^2 E}{\partial k_y \partial k_x} & \frac{\partial^2 E}{\partial k_y^2} & \frac{\partial^2 E}{\partial k_y \partial k_z} \\ \frac{\partial^2 E}{\partial k_z \partial k_x} & \frac{\partial^2 E}{\partial k_z \partial k_y} & \frac{\partial^2 E}{\partial k_z^2} \end{bmatrix}$$



 m^* is a function of k, can be smaller or larger than m_0 , even can be negative

$$m_0 = 9.11*10^{-31} \text{ kg}$$

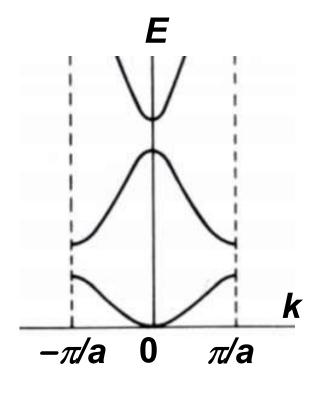
effective mass

$$\frac{1}{m^*} = \frac{1}{\hbar^2} \frac{d^2 E(k)}{dk^2}$$

mobility

$$\mu = \frac{v}{E} = e \frac{\tau}{m^*}$$

conductivity
$$\sigma = ne\mu = ne^2 \frac{\tau}{m^*}$$

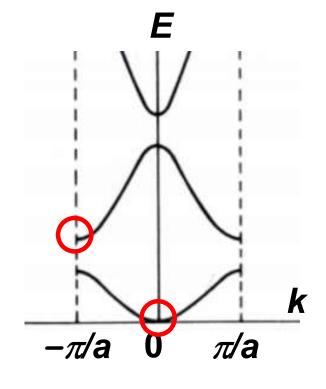


effective mass

$$\frac{1}{m^*} = \frac{1}{\hbar^2} \frac{d^2 E(k)}{dk^2}$$

close to band minimum parabolic approximation

$$E(k) \approx E_0 + \frac{\hbar^2}{2m^*} (k - k_0)^2$$

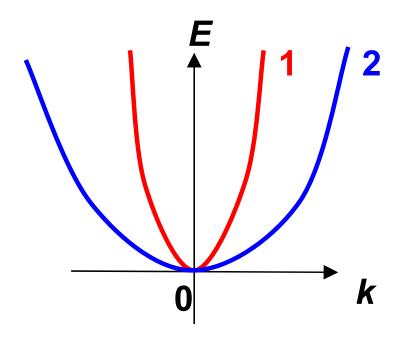


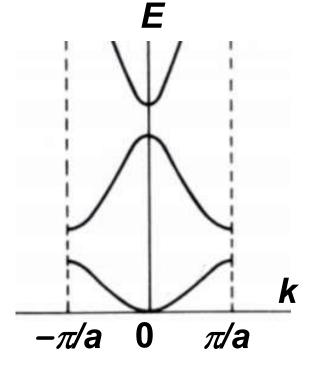
$$g(E) = \frac{dn}{dE} = \frac{1}{2\pi^2} \left(\frac{2m^*}{\hbar^2}\right)^{3/2} (E - E_0)^{1/2}$$

effective mass

$$\left| \frac{1}{m^*} = \frac{1}{\hbar^2} \frac{d^2 E(k)}{dk^2} \right|$$

inverse curvature of the parabolic curve

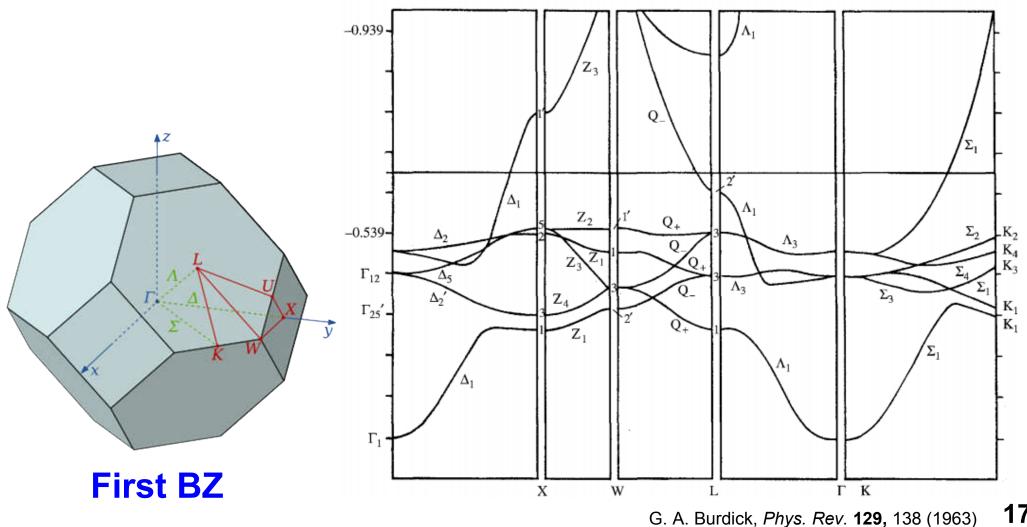




Q:
$$m_1 > m_2$$
 or $m_1 < m_2$?

Band Structure in 3D

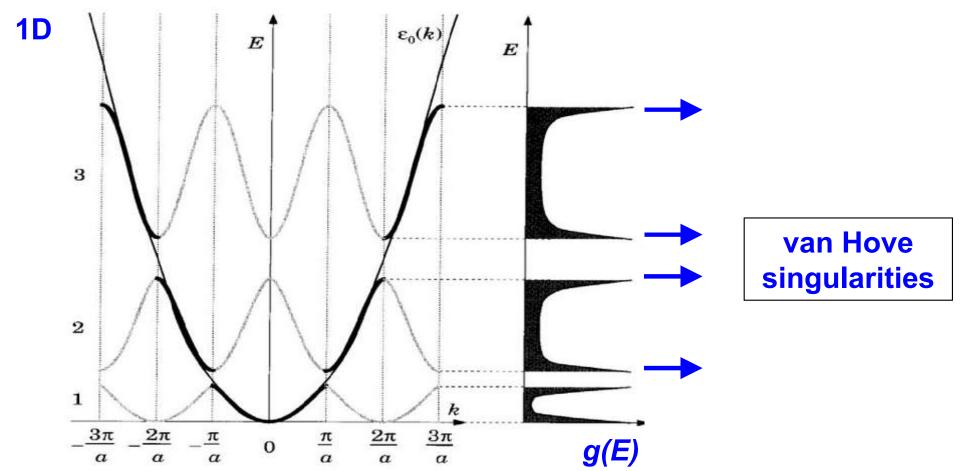
copper (FCC)



Density of States (DOS) 态密度

$$g(E) = \frac{dn}{dE}$$

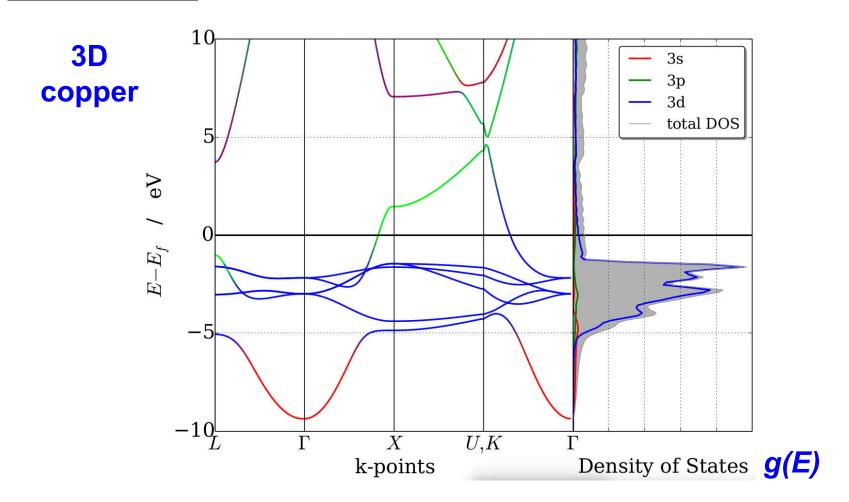
DOS - number of energy states/levels per unit energy in [*E*, *E*+*dE*], per unit volume



Density of States (DOS) 态密度

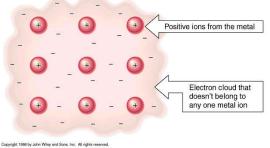
$$g(E) = \frac{dn}{dE}$$

DOS - number of energy states/levels per unit energy in [*E*, *E*+*dE*], per unit volume



Bloch Model - Review

Electrons are in a periodic potential formed by the atomic lattice



- Born-Oppenheimer Approximation
 - The behaviors of electrons and nuclei can be calculated separately.
- Independent Electron Approximation
 - We still assume electrons are independent and do not interact with each other

Real Electrons in Solids is a Nightmare

$$\hat{H}\Psi = E\Psi$$

many-body problem '多体'问题

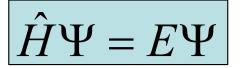
$$\sum_{i=1}^{N} \left(-\frac{\hbar^2}{2m} \nabla_i^2 \Psi - Ze^2 \sum_{\mathbf{R}} \frac{1}{|\mathbf{r}_i - \mathbf{R}|} \Psi \right) + \frac{1}{2} \sum_{i \neq j} \frac{e^2}{|\mathbf{r}_i - \mathbf{r}_j|} \Psi = E \Psi$$

free electron

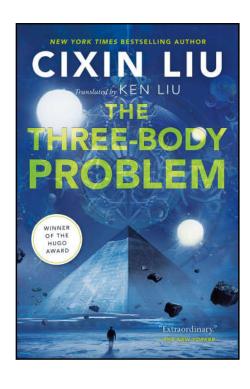
electron-atom interaction

electron-electron interaction

Real Electrons in Solids is a Nightmare

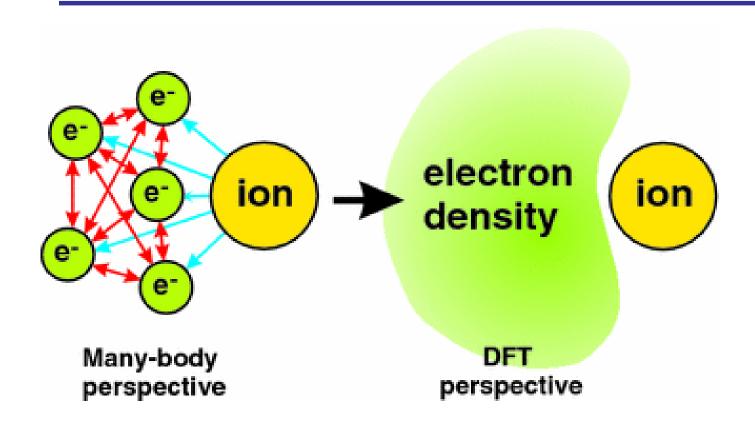


many-body problem '多体'问题





Density Functional Theory 密度泛函理论



wave function → electron density

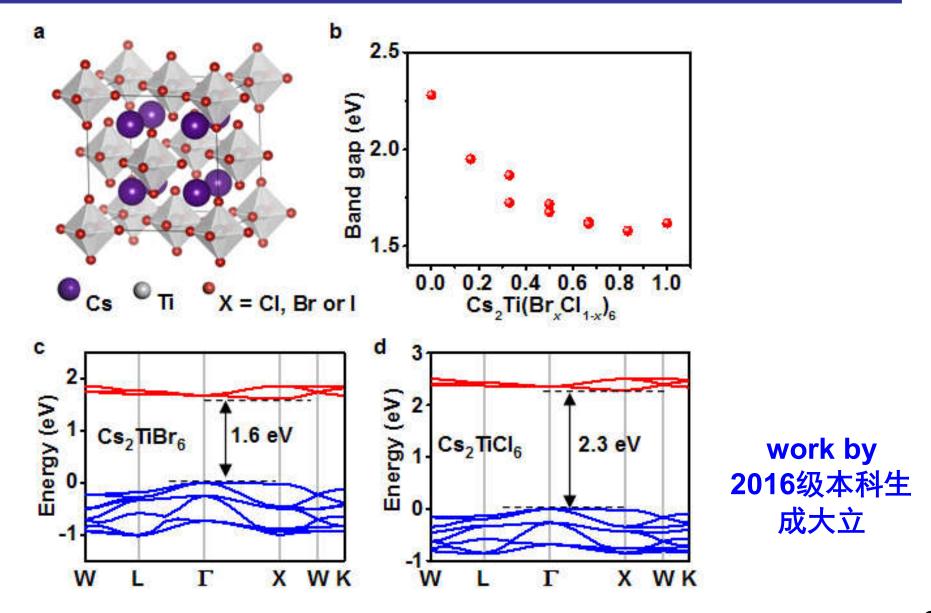


Photo from the Nobel Foundation archive.

Walter Kohn

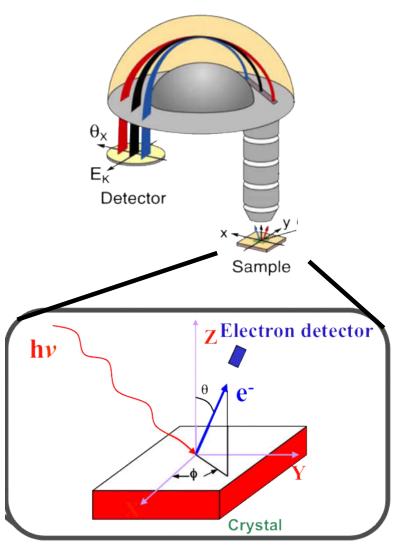
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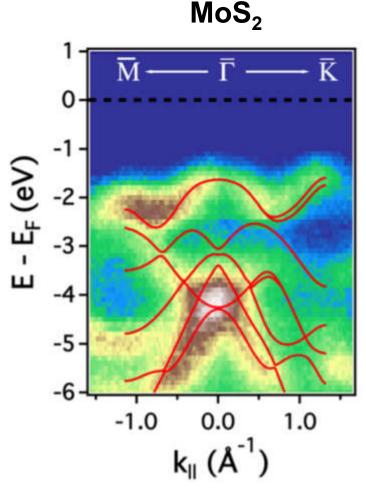
Band Structures by DFT - Example



Measurement of Band Structures

Angle-Resolved Photoemission Spectroscopy (ARPES)





Thank you for your attention