

Principles of Micro- and Nanofabrication for Electronic and Photonic Devices

Etching 刻蚀

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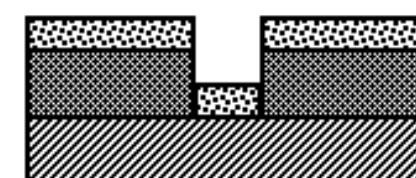
Pattern Formation

Subtractive Process

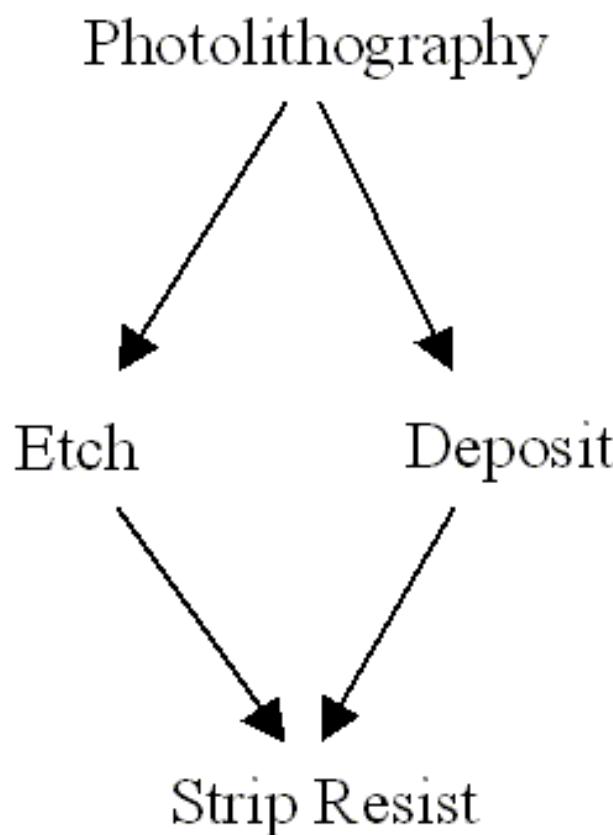


Pattern transfer
by etching

Additive Process

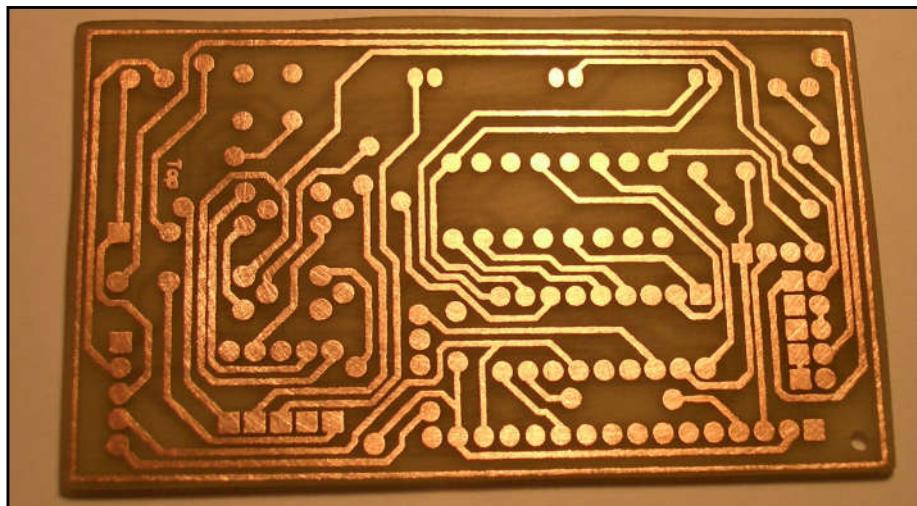


Pattern transfer
by lift off



Etching vs. Corrosion

Etching (刻蚀)



wanted

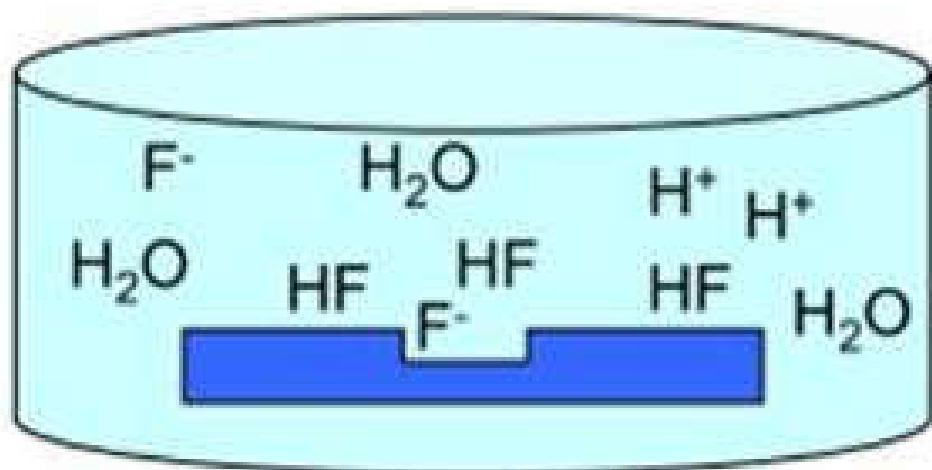
Corrosion (腐蚀)



unwanted

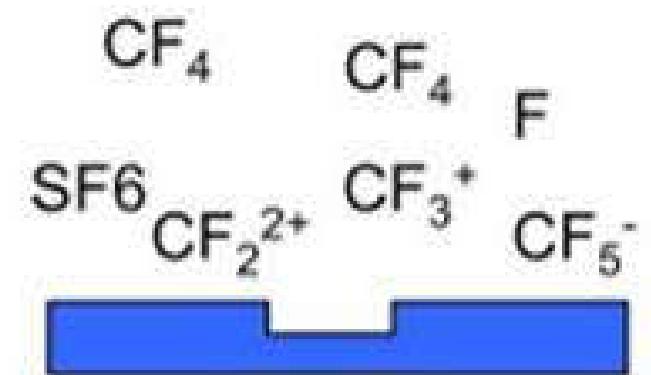
Wet vs. Dry

liquid source



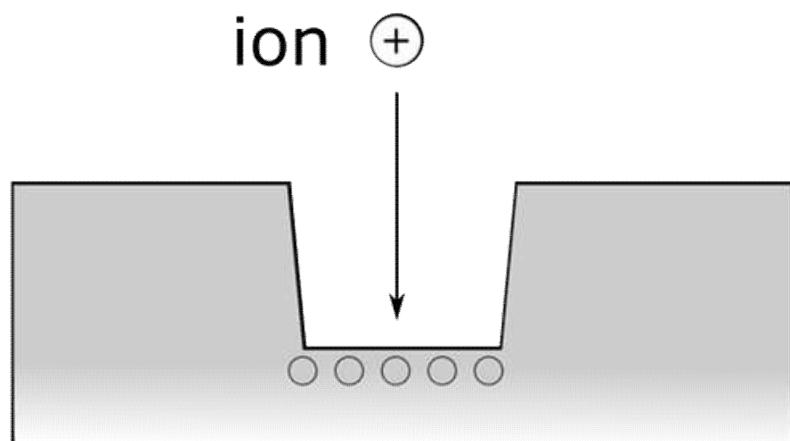
Wet Etch

gas source

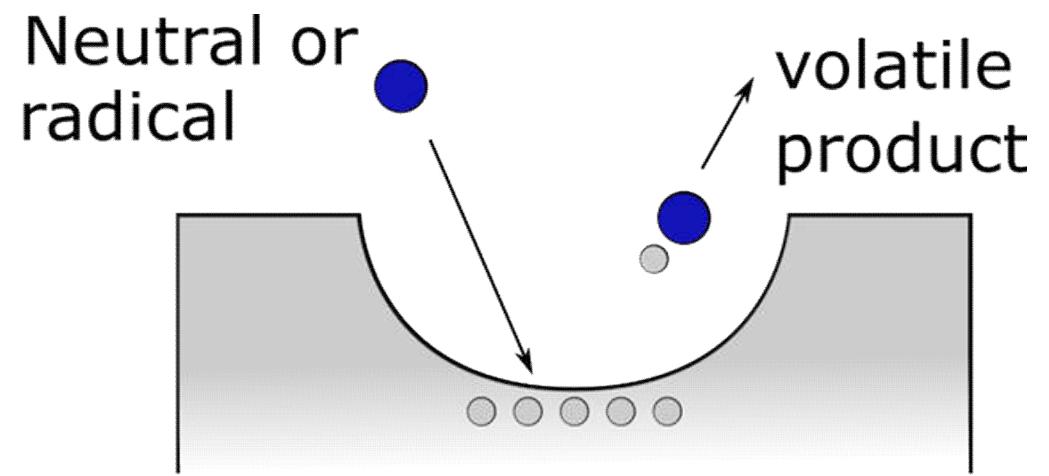


Dry Etch

Chemical vs. Physical



Physical etching
(sputtering)



Chemical Etching

Etching

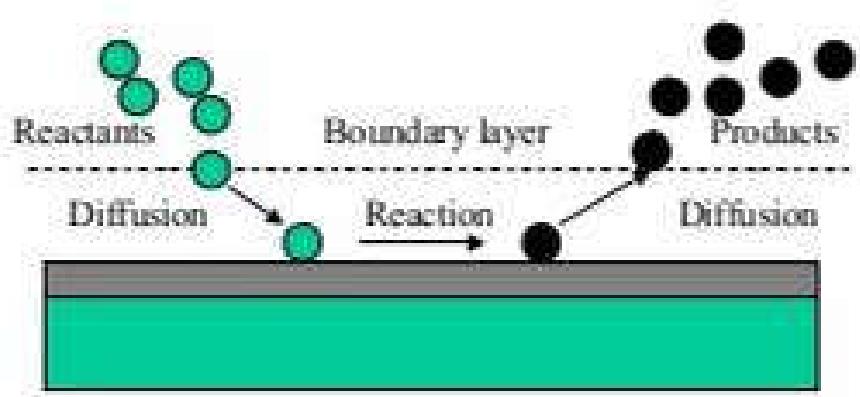
■ Process Parameters

- Time
- Temperature
- Etchant type
- Etchant concentration
- Mask type
- ...

■ Control Parameters

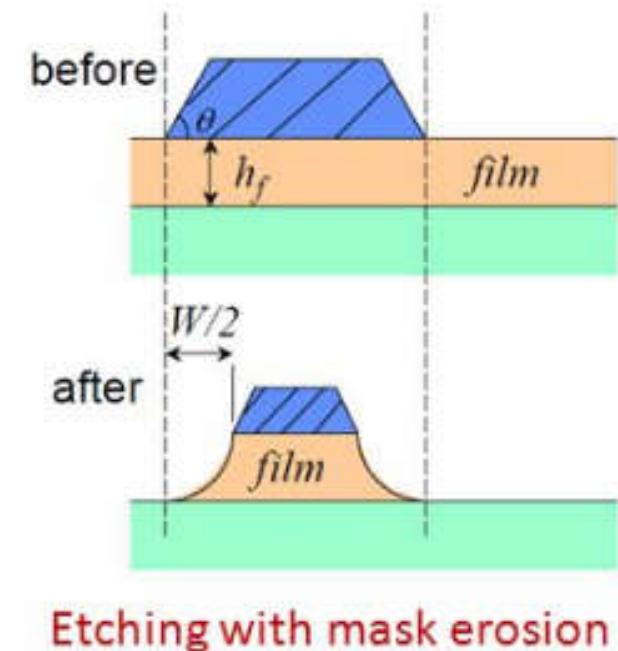
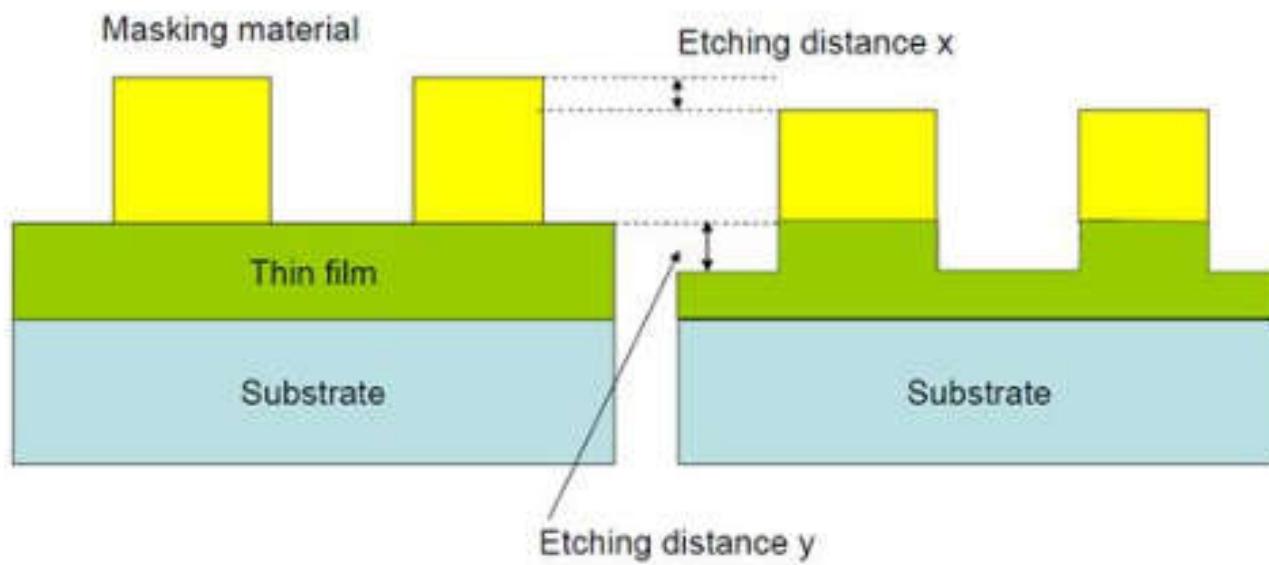
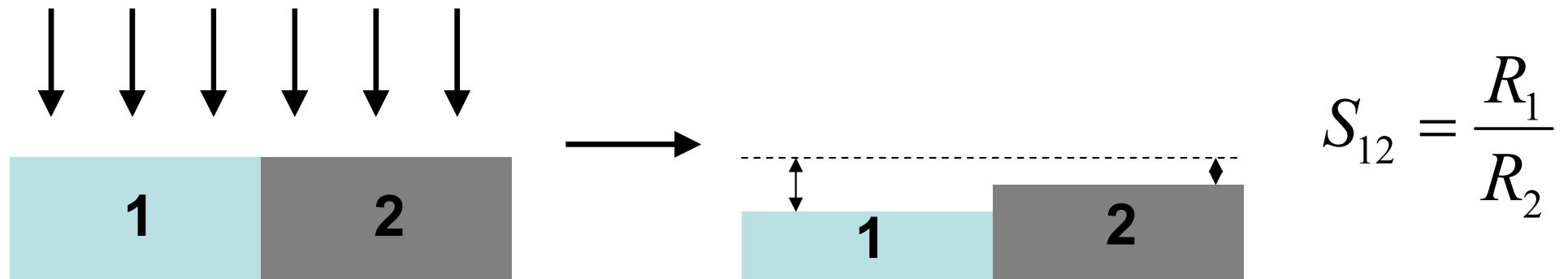
- Etch rate
- Selectivity
- Anisotropy
- Uniformity
- ...

diffusion - reaction - diffusion



- chemical reactions occur
- products should be disposable

Selectivity



Selectivity - Example

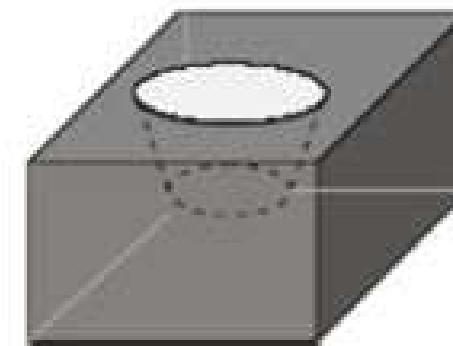
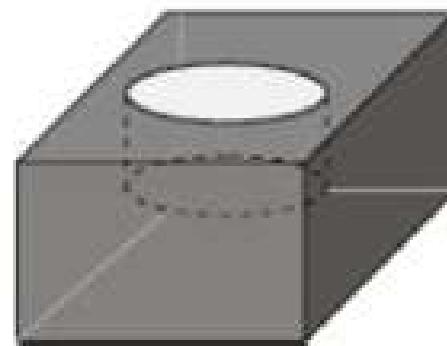
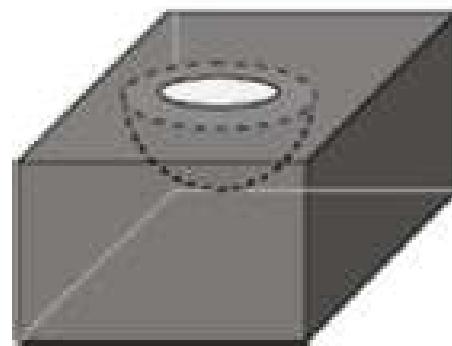
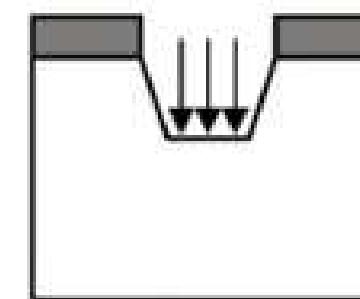
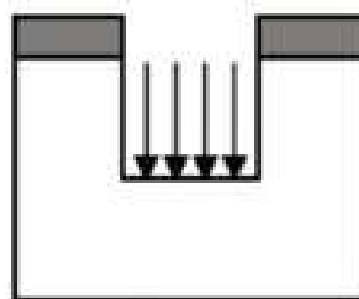
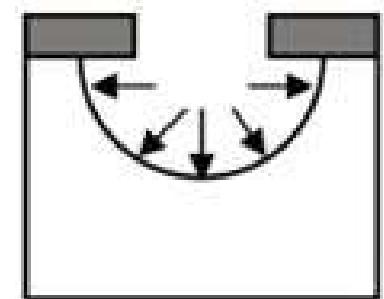


- **SiO_2 / Si wet etch by HF solution**
 - **very large selectivity $S_{\text{SiO}_2/\text{Si}} \sim \text{infinity}$**
- **SiO_2 / Si dry etch by CF_4 plasma**
 - **selectivity $S_{\text{SiO}_2/\text{Si}} \sim 10$**

Anisotropy (各向异性)

degree of anisotropy

$$A = 1 - \frac{R_{lateral}}{R_{vertical}}$$



isotropic
 $A = 0$

fully anisotropic
 $A = 1$

anisotropic
 $0 < A < 1$

Isotropic vs. Anisotropic

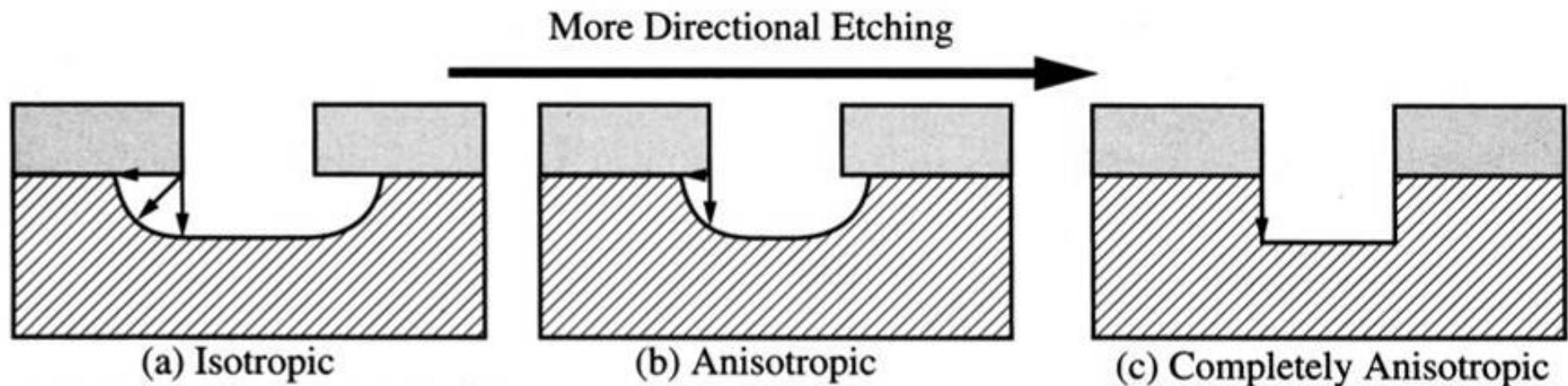


Figure 10–3 Etch profiles for different degrees of anisotropic, or directional, etching: (a) purely isotropic etching; (b) anisotropic etching; (c) completely anisotropic etching.

chemical

physical

Uniformity

(a)

rms roughness = 339A

μm

0.30
0.15
0.00

4

μm

2

0 2 μm

(b)

rms roughness = 202A

μm

0.30
0.15
0.00

4

μm

2

0 μm

(c)

rms roughness = 33.8A

μm

0.30
0.15
0.00

4

μm

2

0 2 μm

(d)

rms roughness = 16.3A

μm

0.30
0.15
0.00

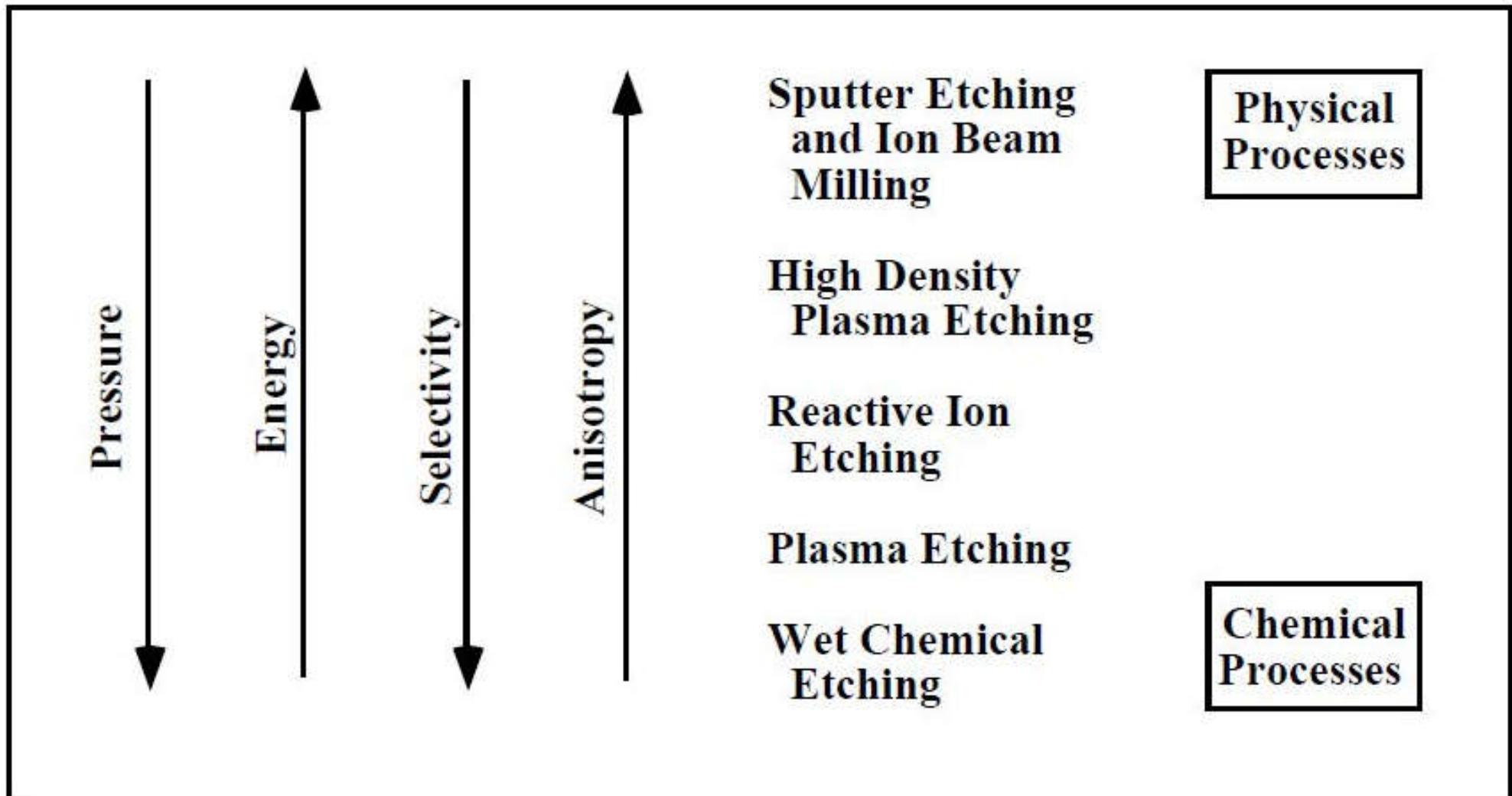
4

μm

2

0 μm

Trends of Etching



Etching Methods

- **Wet Etching 湿法刻蚀**
- **Dry Etching 干法刻蚀**
- **CMP and other methods**

References

- **Wet Etching Recipes**

<http://ieeexplore.ieee.org/iel4/84/11954/00546406.pdf>
<http://ieeexplore.ieee.org/iel4/84/11954/01257354.pdf>

https://cleanroom.byu.edu/chemical_etching.html

- **Guide to references on III-V semiconductor chemical etching**

<http://www.sciencedirect.com/science/article/pii/S0927796X00000279>

Metal Dissolution in Acids

	Element	Oxidation Reaction
React vigorously with cold H ₂ O to form H ₂	Lithium	Li → Li ⁺ + e ⁻
React with steam to form H ₂	Potassium	K → K ⁺ + e ⁻
React with simple acids to form H ₂	Barium	Ba → Ba ²⁺ + 2e ⁻
	Calcium	Ca → Ca ²⁺ + 2e ⁻
	Sodium	Na → Na ⁺ + e ⁻
	Magnesium	Mg → Mg ²⁺ + 2e ⁻
	Aluminum	Al → Al ³⁺ + 3e ⁻
	Manganese	Mn → Mn ²⁺ + 2e ⁻
	Zinc	Zn → Zn ²⁺ + 2e ⁻
	Chromium	Cr → Cr ³⁺ + 3e ⁻
	Iron	Fe → Fe ²⁺ + 2e ⁻
	Cadmium	Cd → Cd ²⁺ + 2e ⁻
	Cobalt	Co → Co ²⁺ + 2e ⁻
	Nickel	Ni → Ni ²⁺ + 2e ⁻
	Tin	Sn → Sn ²⁺ + 2e ⁻
	Lead	Pb → Pb ²⁺ + 2e ⁻
Will not dissolve in simple acids	Hydrogen	H ₂ → 2H ⁺ + 2e ⁻
	Copper	Cu → Cu ²⁺ + 2e ⁻
	Silver	Ag → Ag ⁺ + e ⁻
	Mercury	Hg → Hg ²⁺ + 2e ⁻
	Platinum	Pt → Pt ²⁺ + 2e ⁻
	Gold	Au → Au ⁺ + e ⁻

↑ Increasing ease of oxidation

easy

hydrogen

hard

Metal Dissolution in Acids

Strong Acids + Strong Oxidants

Piranha $\text{H}_2\text{SO}_4 : \text{H}_2\text{O}_2 = 3:1$
dissolves most metals and organics



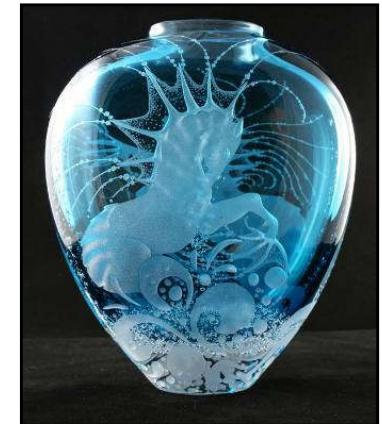
Aqua Regia (王水) $\text{HCl} : \text{HNO}_3 = 3:1$
even dissolves Au, Pt



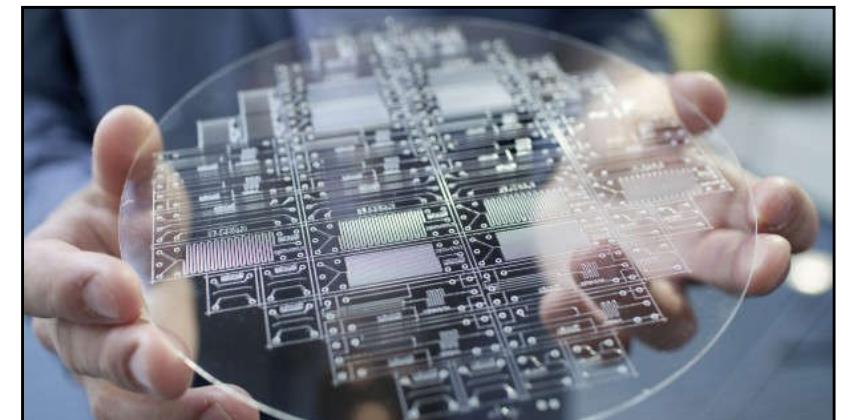
However, difficult to obtain ideal selectivity ...

SiO₂ etching

- Alkali (NaOH, etc) slowly etches SiO₂
 - $\text{SiO}_2 + 2\text{NaOH} = \text{Na}_2\text{SiO}_3 + \text{H}_2\text{O}$
- HF strongly etches SiO₂
 - $\text{SiO}_2 + 6\text{HF} = \text{H}_2\text{SiF}_6 + 2\text{H}_2\text{O}$
- Buffered HF (BHF/BOE)
 - HF + NH₄F
 - lower etch rate
 - safer for use

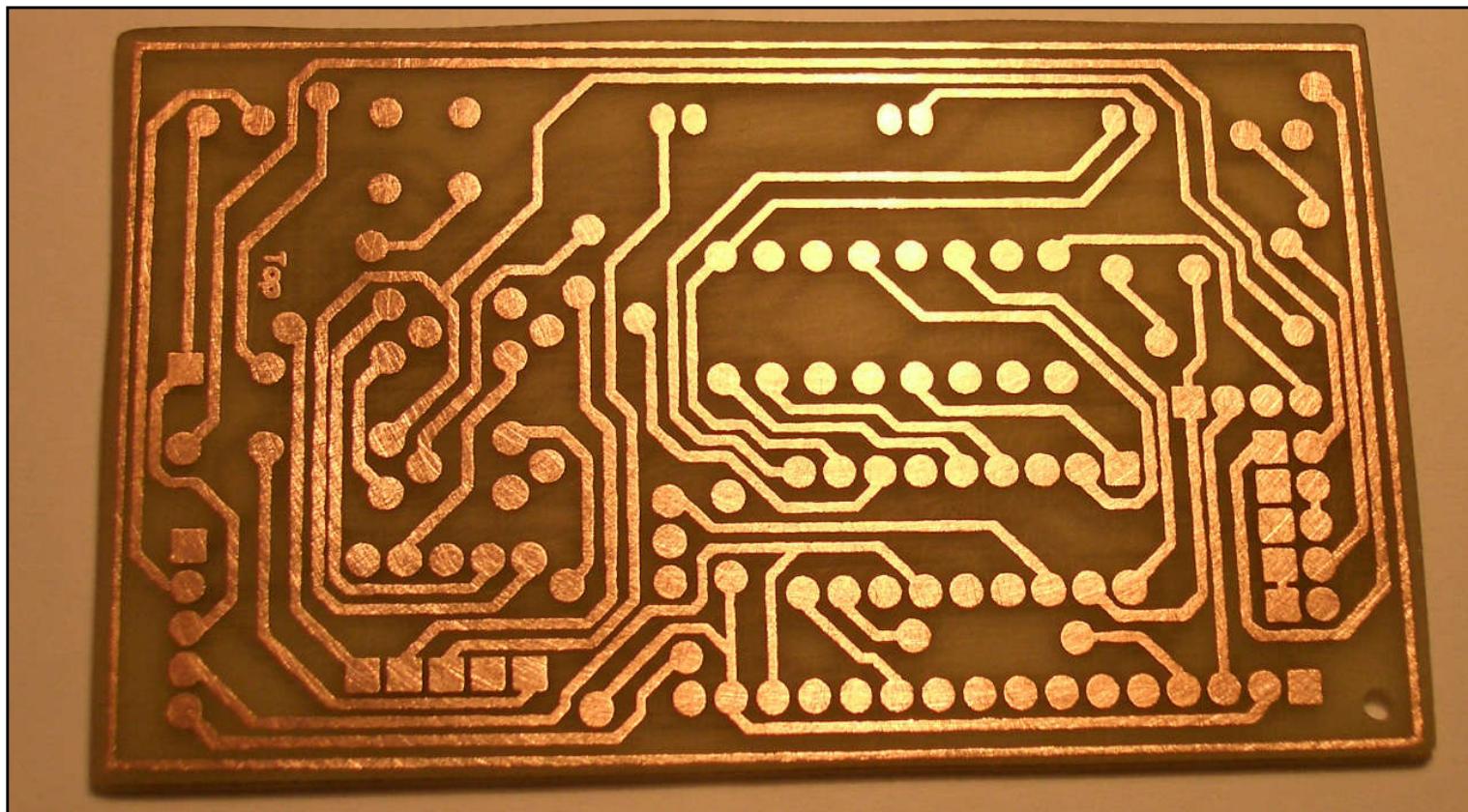


glass art by HF etch

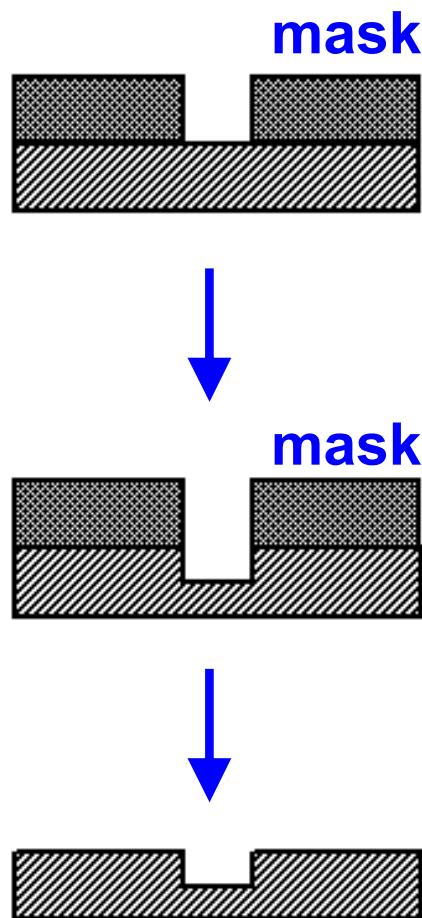


Cu etching

- $\text{Cu} + 2\text{FeCl}_3 = \text{CuCl}_2 + 2\text{FeCl}_2$



Selectivity for Wet Etch

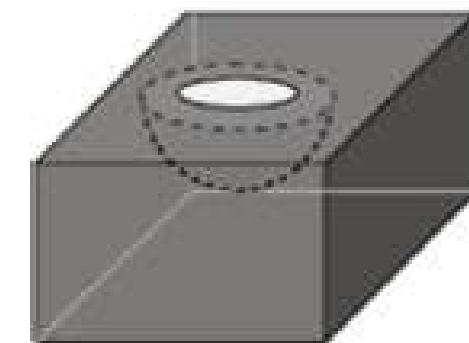
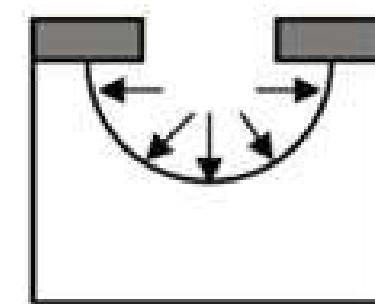


Films	Etchant	Mask
SiO_2	HF	PR
Si	KOH	Si_3N_4
GaAs	$\text{H}_3\text{PO}_4 + \text{H}_2\text{O}_2$	PR
GaP	KOH + $\text{K}_3[\text{Fe}(\text{CN})_6]$	SiO_2
Cu	FeCl_3	PR
Au	$\text{KI} + \text{I}_2$	PR

*most wet etch recipes are isotropic,
except KOH etch for Si*

Isotropy for Wet Etch

- Wet etch is usually isotropic
- Exceptions
 - some etching for single crystals
 - KOH etch Si

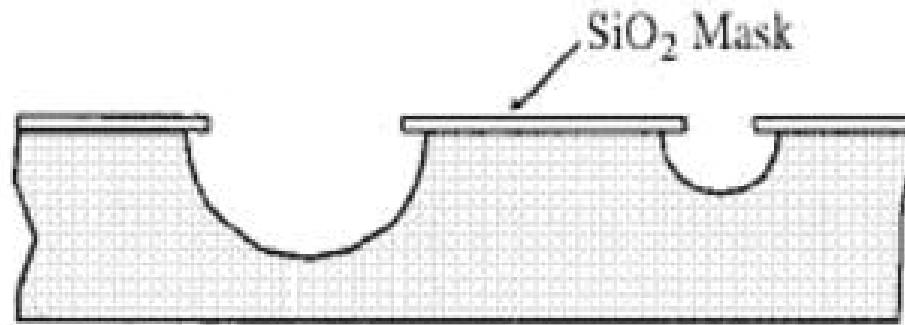


isotropic
 $A = 0$

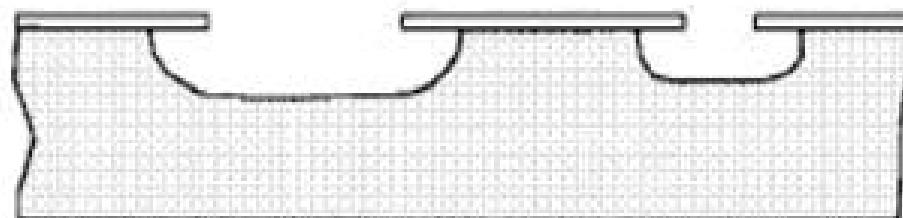
Si etching

- $\text{HNO}_3 + \text{HF}$
 - isotropic etch

Isotropic wet etching: Agitation



Isotropic wet etching: No Agitation



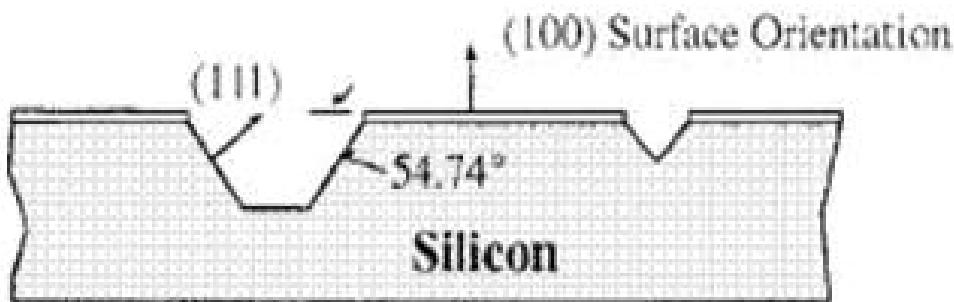
Si etching

- KOH

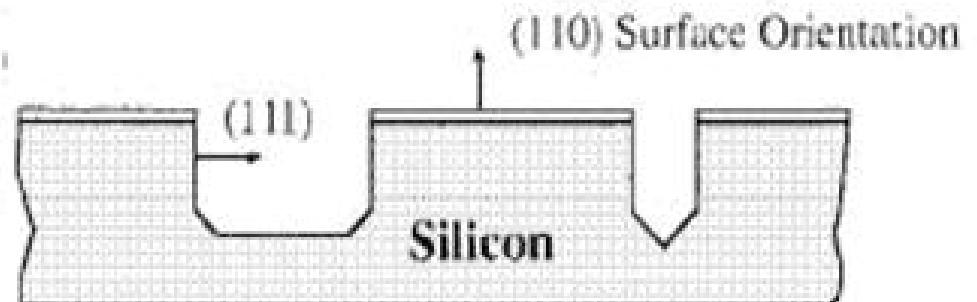
- **anisotropic etch**
- **etch rate (111):(110):(100) ~ 1:600:400**
- **mask: SiO₂, Si₃N₄, Cr/Au, ...**

Q: why?

Anisotropic wet etching: (100)



Anisotropic wet etching: (110)



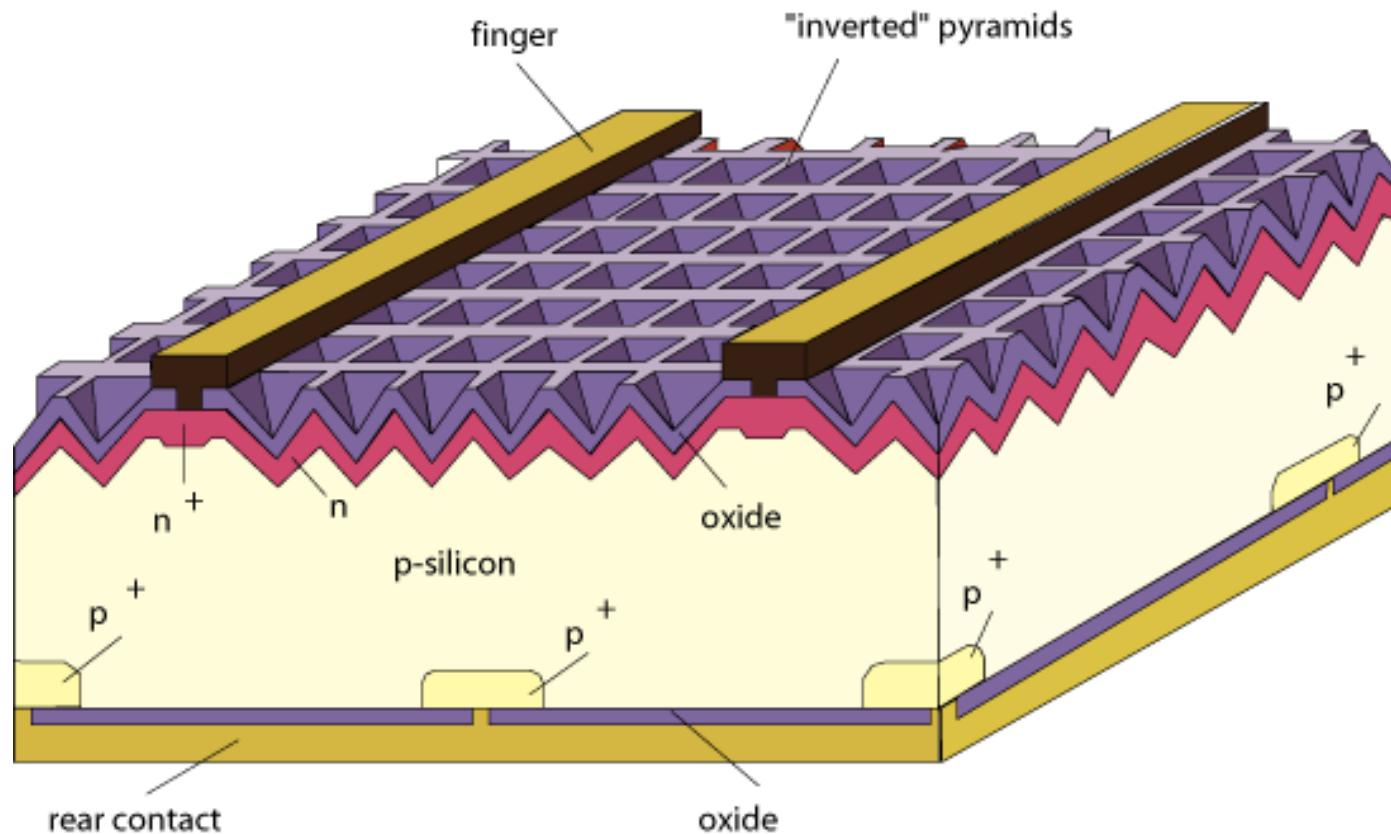
- Other chemistries

- **TMAH: Tetramethyl ammonium hydroxide**
- **EDP: Ethylene diamine pyrochatecol**

Si etching

- Single Crystalline Si Solar Cells
 - KOH anisotropic etch

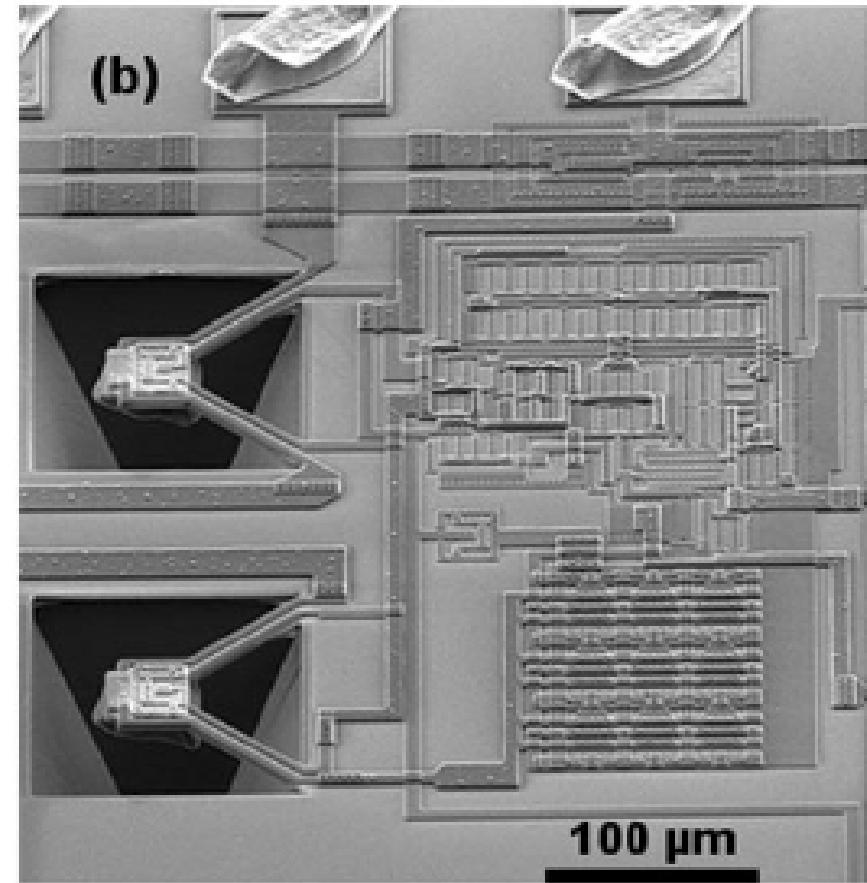
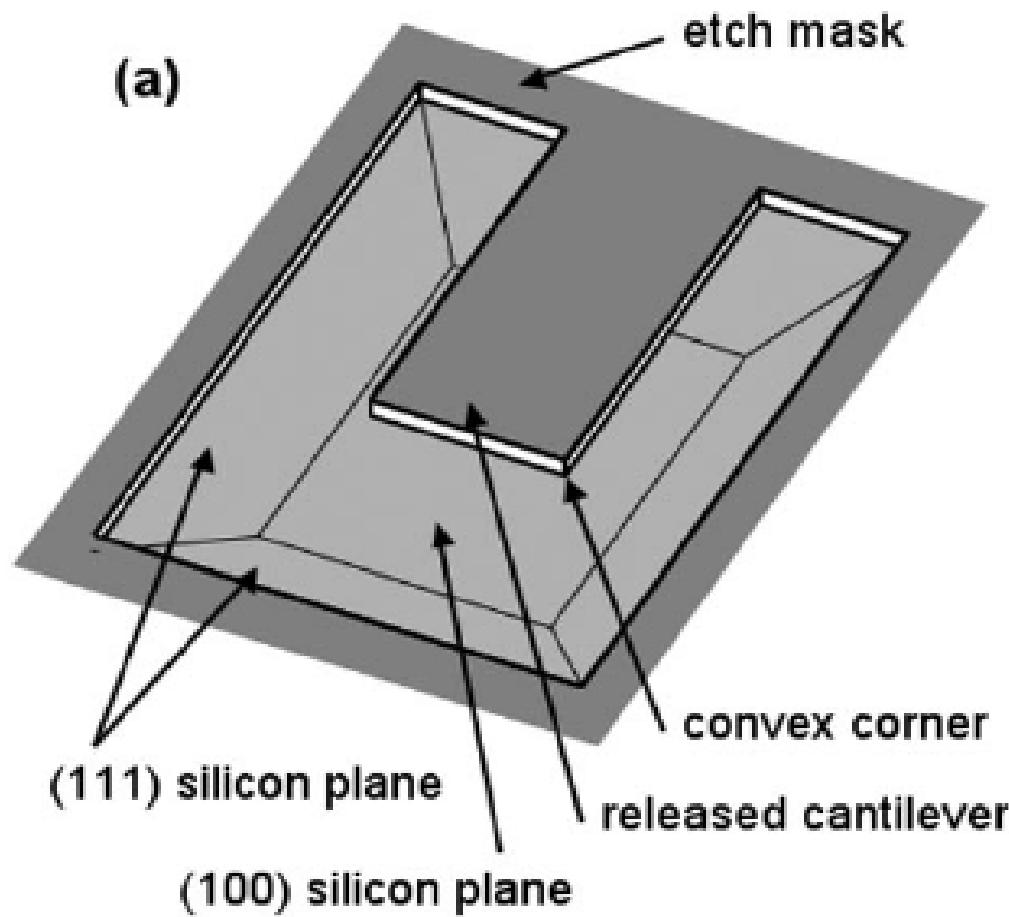
*optical trapping and
antireflection*



world record efficiency: 26%

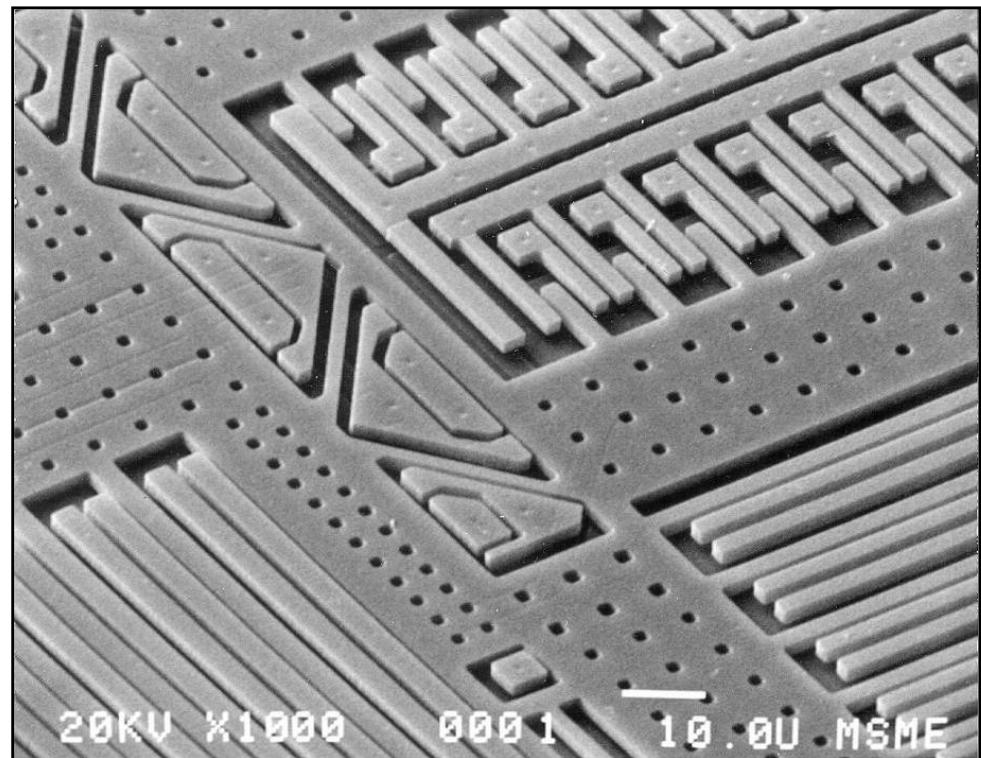
Si etching

- Si cantilever beam
 - KOH anisotropic etch



Si etching

- Micro-Electro-Mechanical Systems (MEMS)



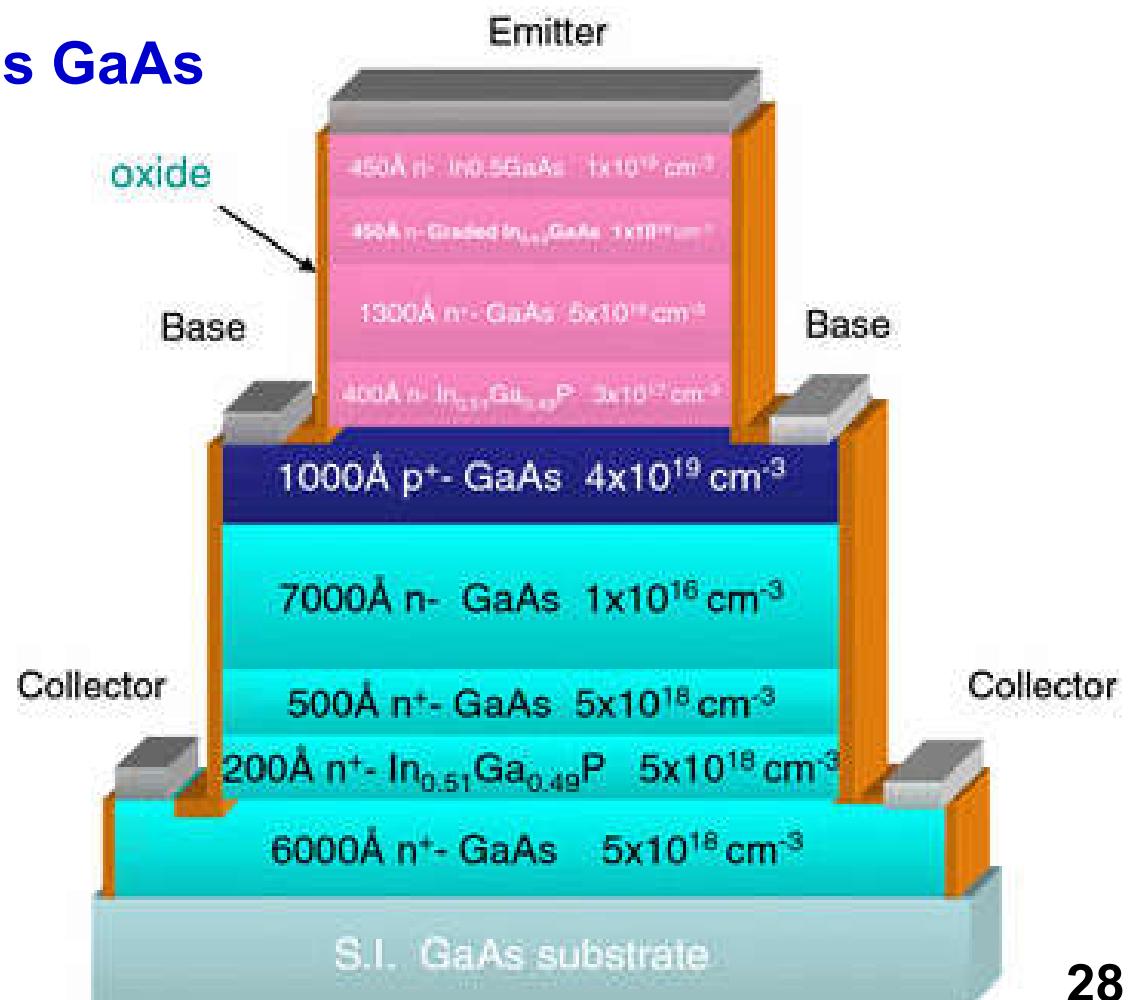
III-V etching

- GaAs, AlGaAs, InGaAs
 - $\text{H}_3\text{PO}_4 + \text{H}_2\text{O}_2$
 - $\text{NH}_4\text{OH} + \text{H}_2\text{O}_2$
- AlGaAs
 - when Al > 70%, HF and HCl etch
- InP, InGaP, InAlP
 - HCl
- GaN, InGaN
 - no reliable wet etchants ...

Etch Stops

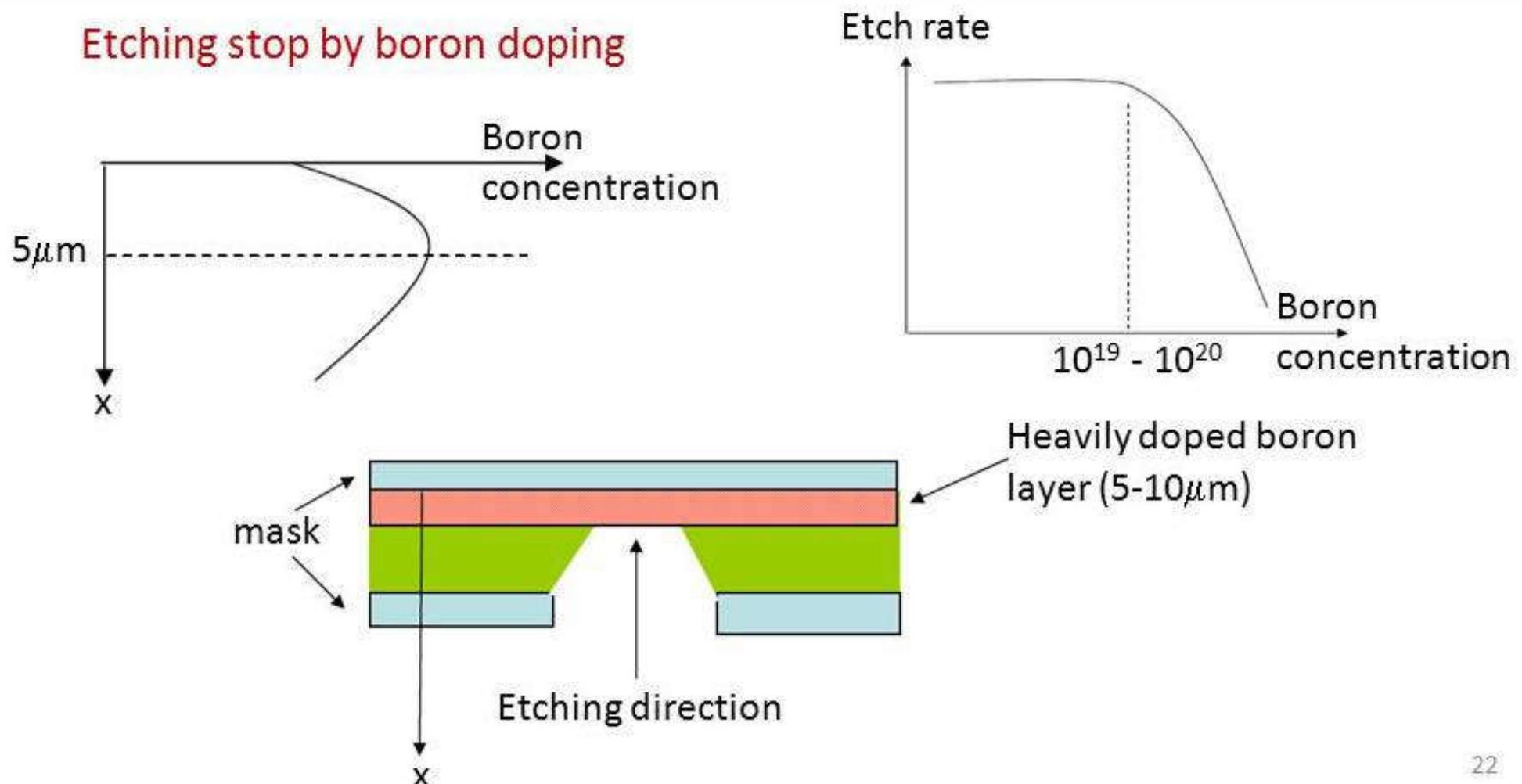
InGaP / GaAs

- lattice matched epitaxy
- $\text{H}_3\text{PO}_4 + \text{H}_2\text{O}_2$ only etches GaAs
- HCl only etches InGaP



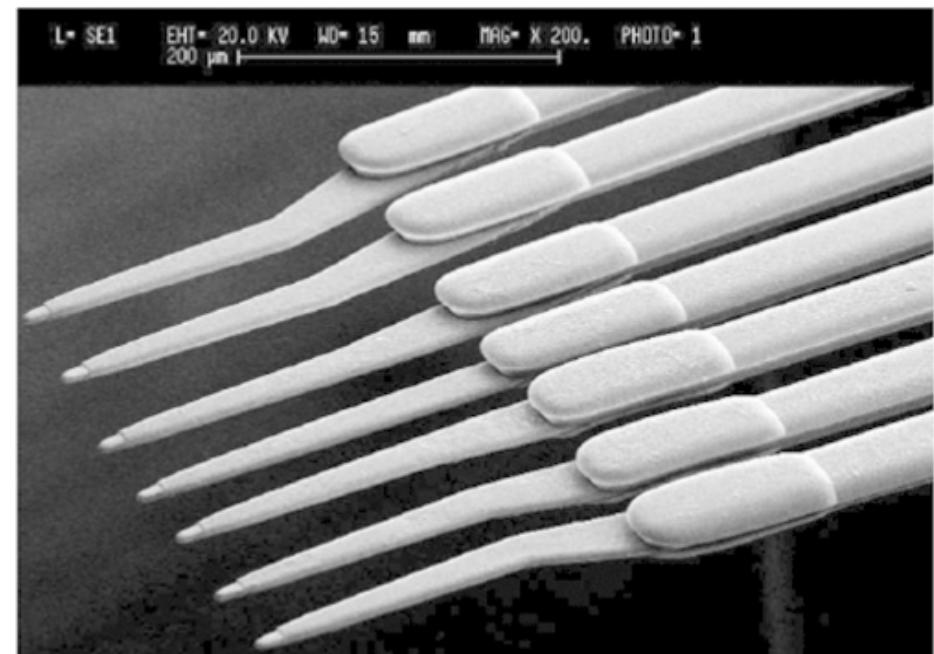
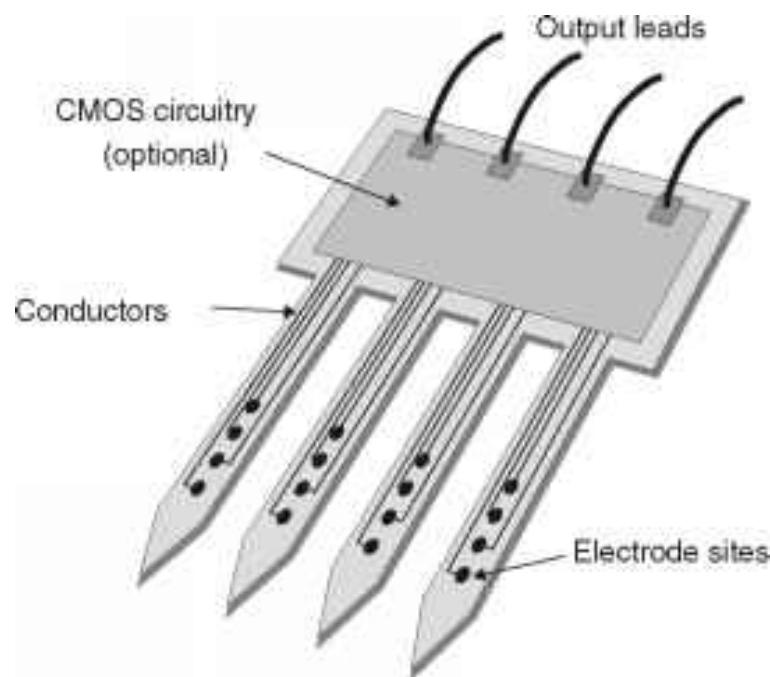
Etch Stops

- highly p-dope Si is resistant to KOH

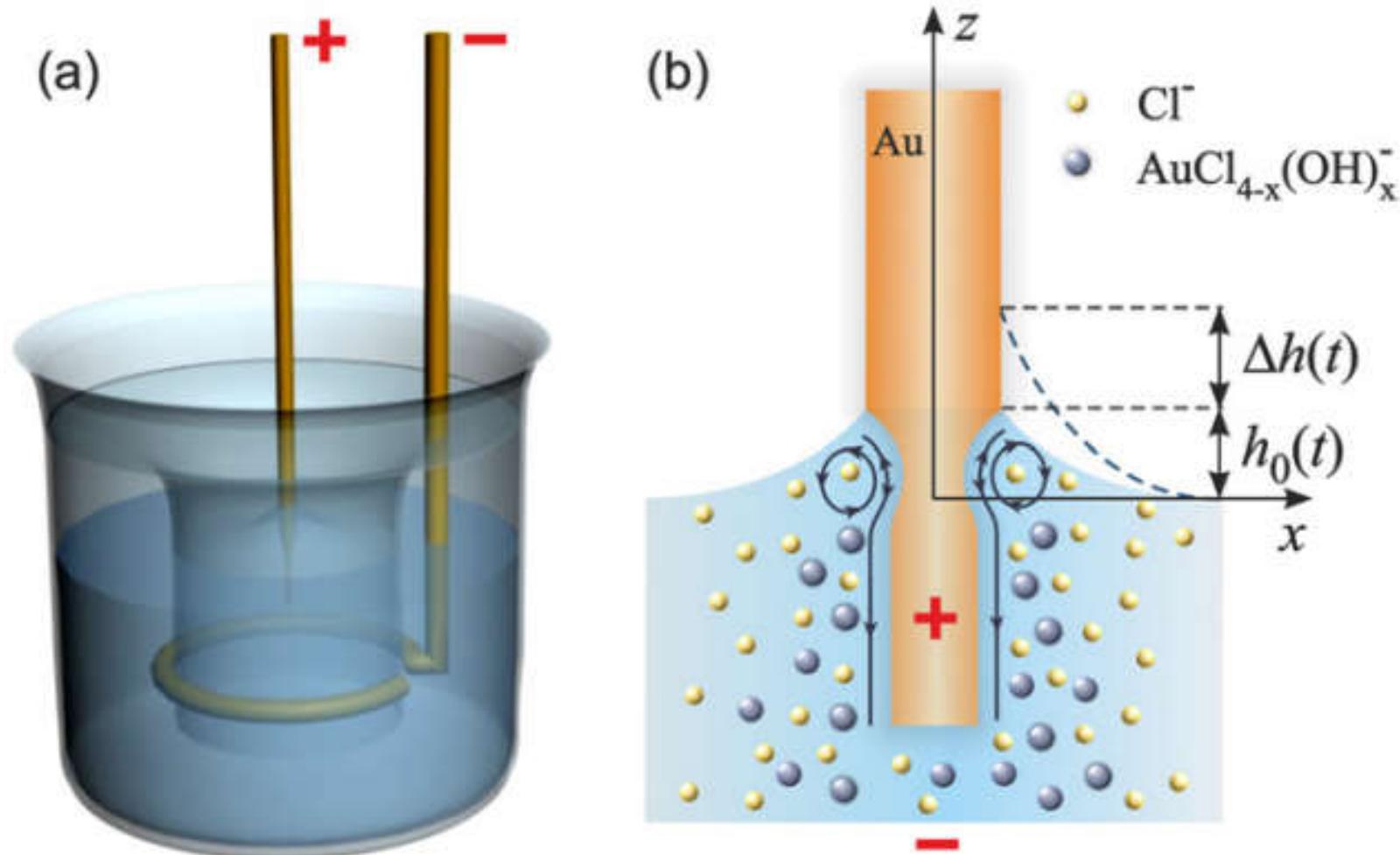


Etch Stops

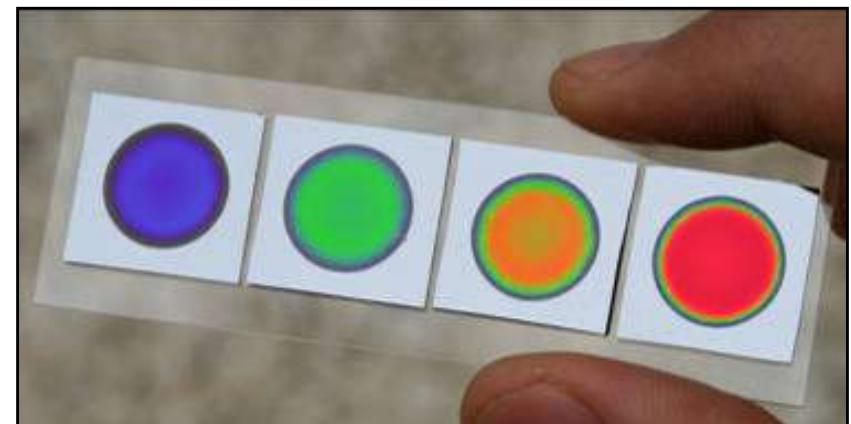
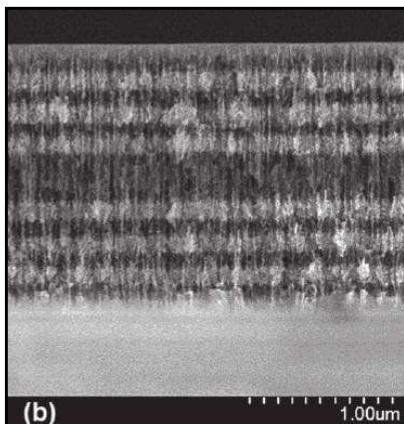
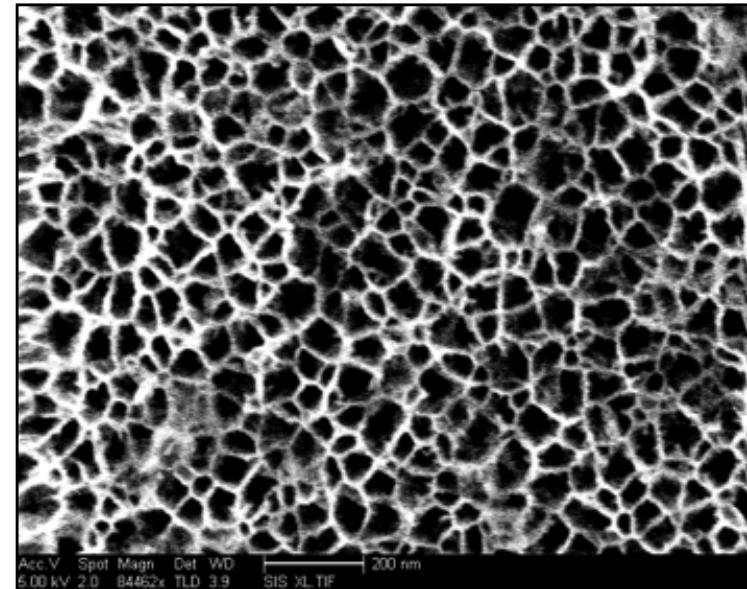
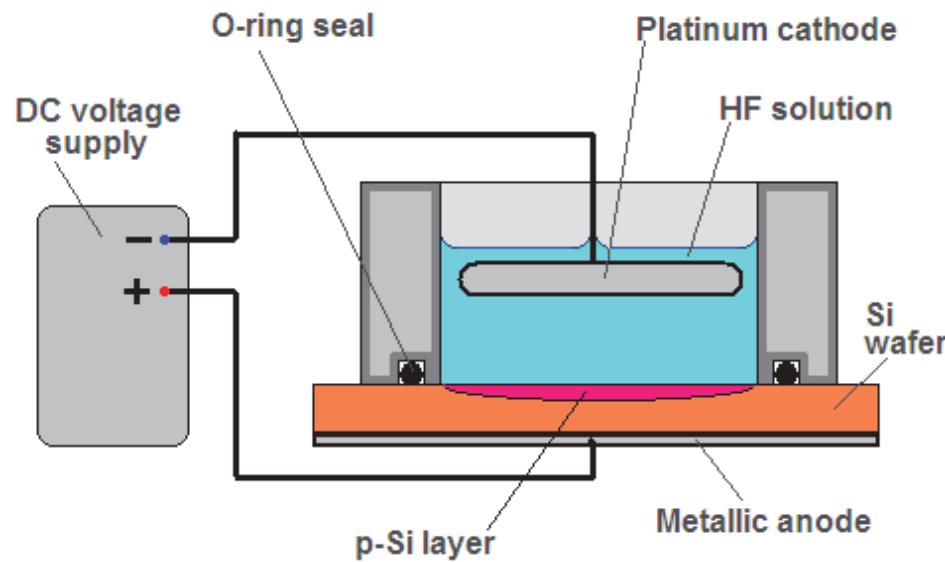
- Silicon based 'Michigan Probe' for neuroscience



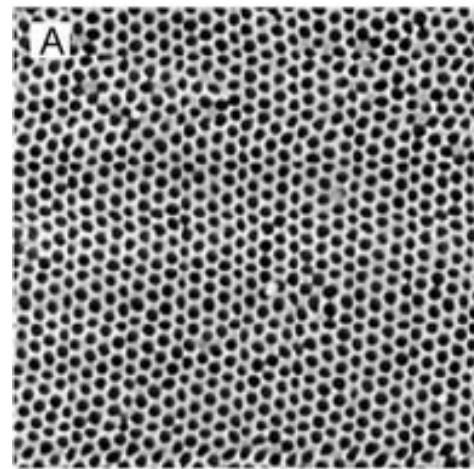
Electrochemical Etch



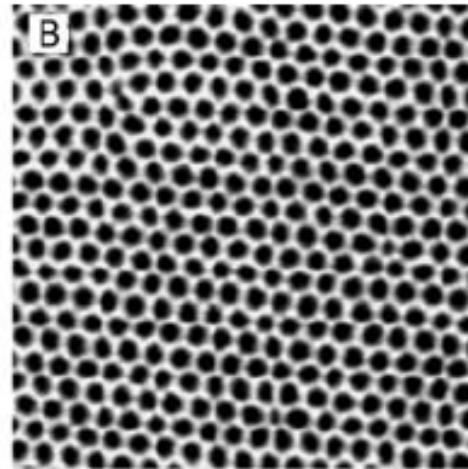
Anodization (阳极氧化) - Porous Si



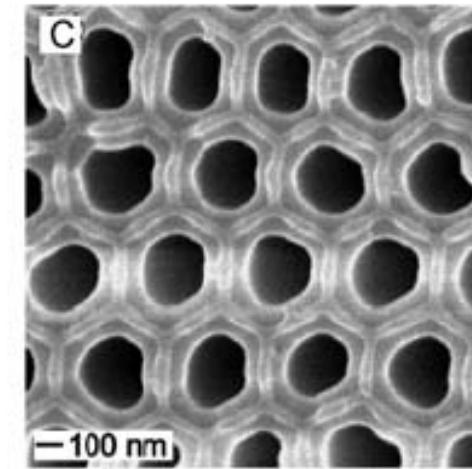
Anodization - Porous Al_2O_3



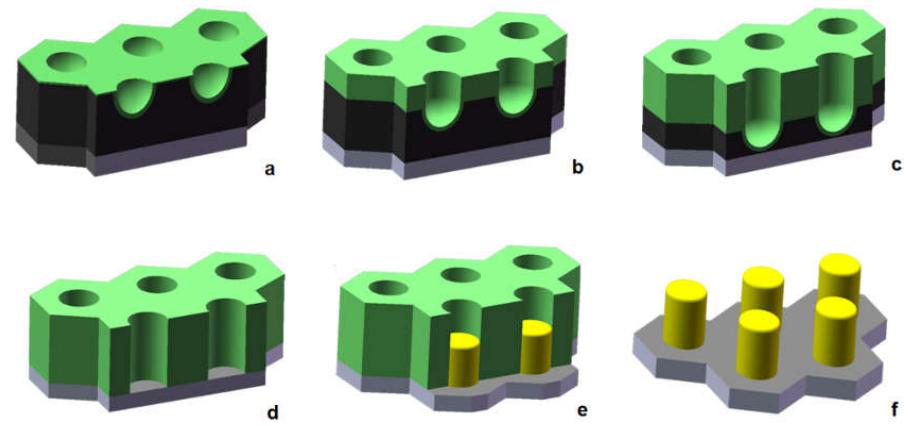
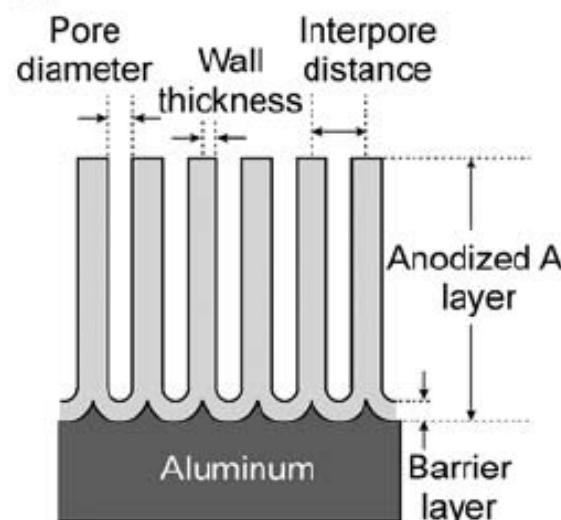
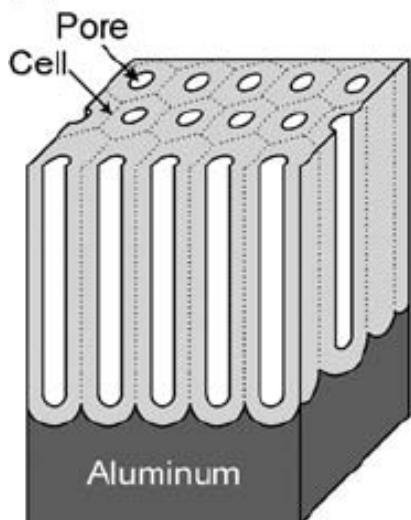
$0.3 \text{ M H}_2\text{SO}_4, 25 \text{ V}$
 $D_p = 60 \text{ nm}$



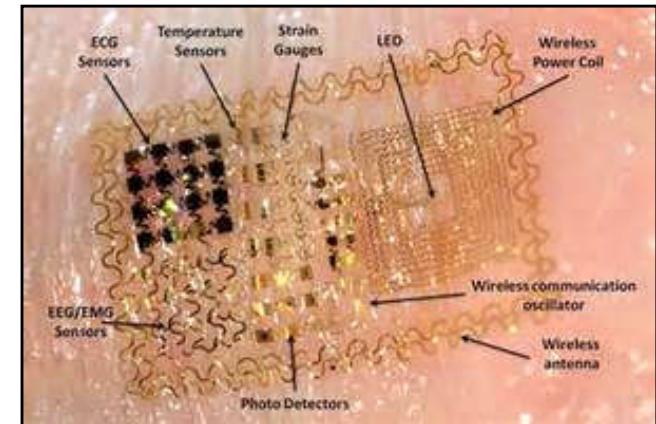
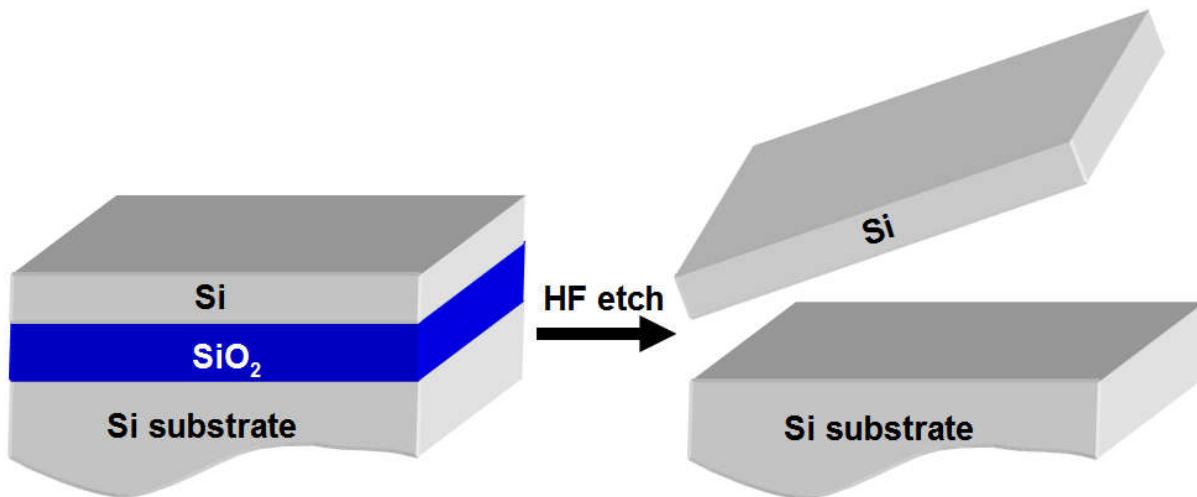
$0.5 \text{ C}_2\text{H}_2\text{O}_4, 40 \text{ V}$
 $D_p = 100 \text{ nm}$



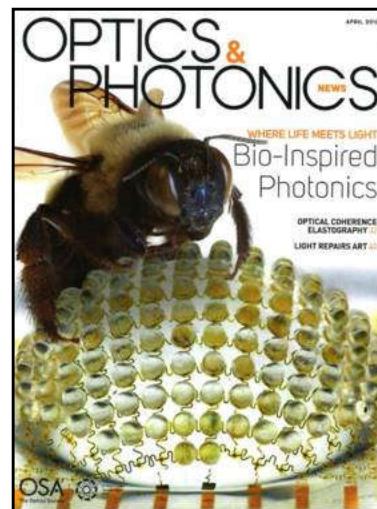
$1.1 \text{ M H}_3\text{PO}_4, 160 \text{ V}$
 $D_p = 420 \text{ nm}$



Thin-Film Si from SOI wafers



human eyes



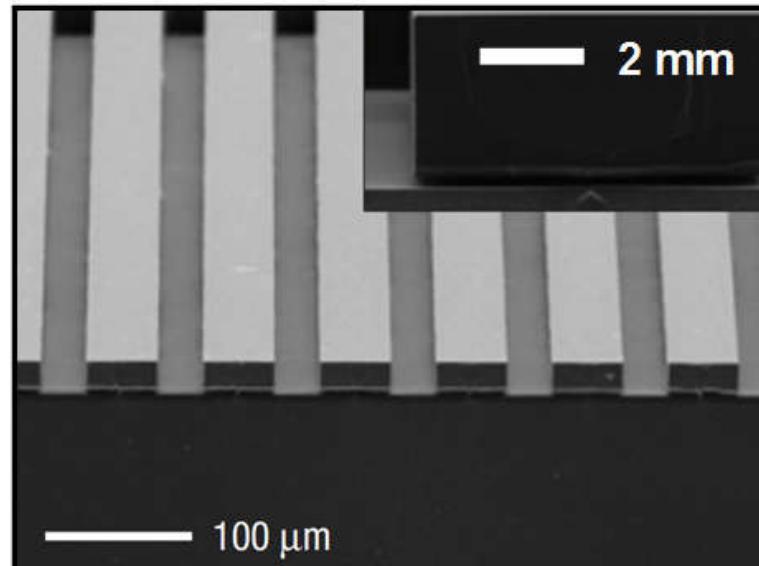
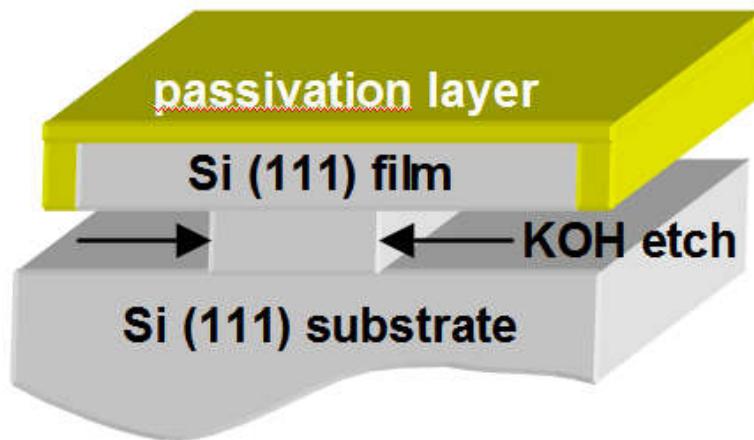
compound eyes



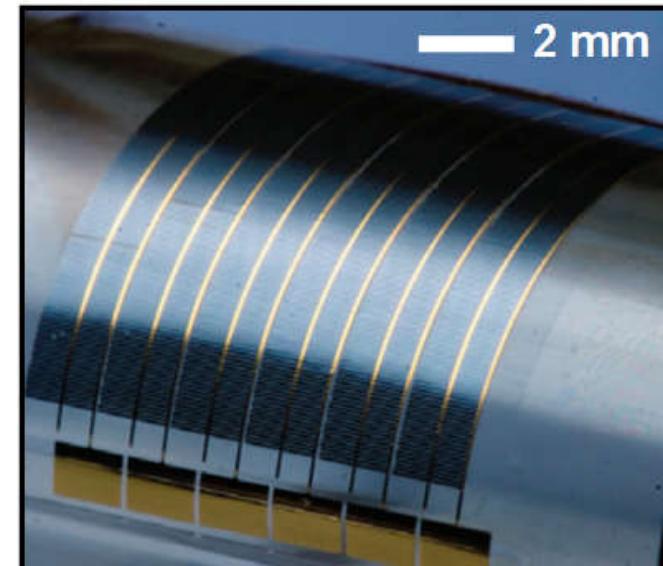
'epidermal' electronics

Thin-Film Si from Si (111)

KOH etches faster for Si (100) than (111)

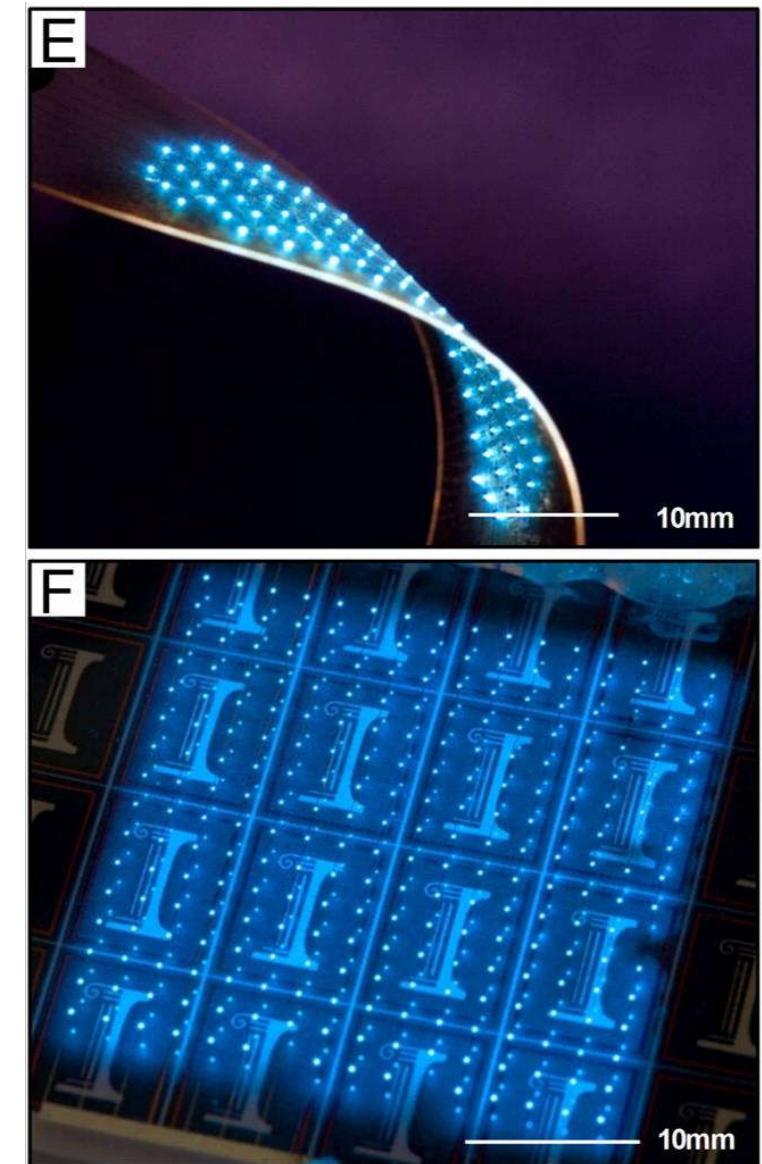
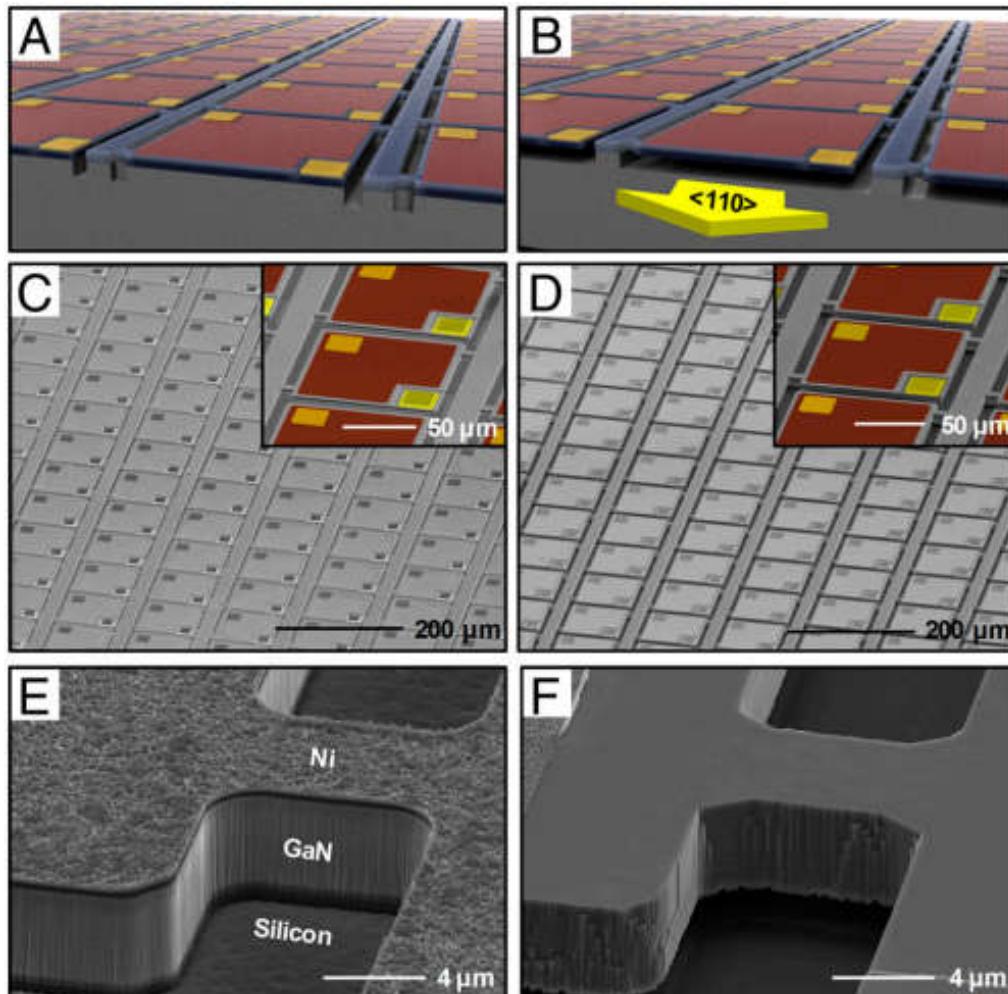


- **Thin-film Si solar cells**
 - **High efficiency (Single Crystal)**
 - **Flexible**
 - **Low cost (wafer reuse)**



GaN on Si (111)

KOH etches faster for Si (100) than (111)

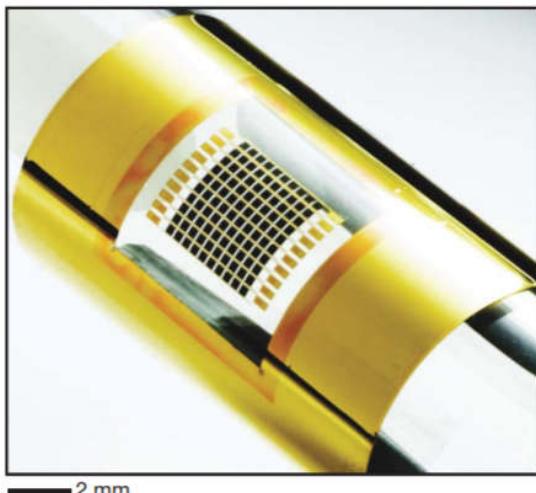
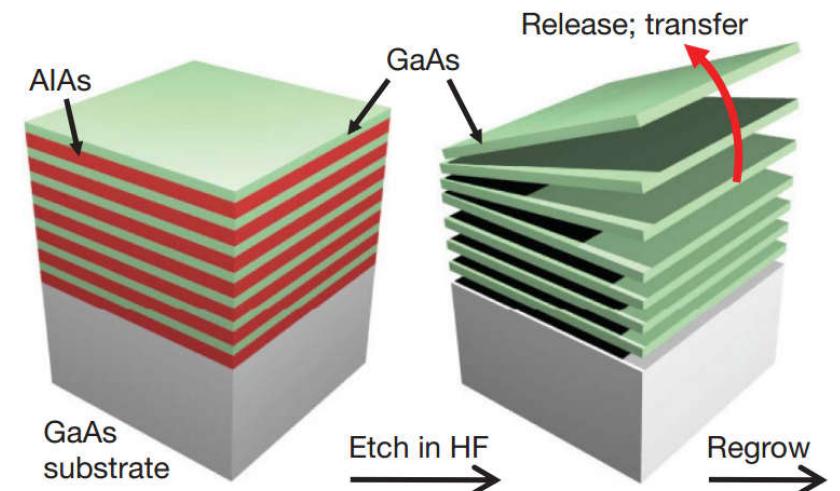


GaAs Device Lift-off

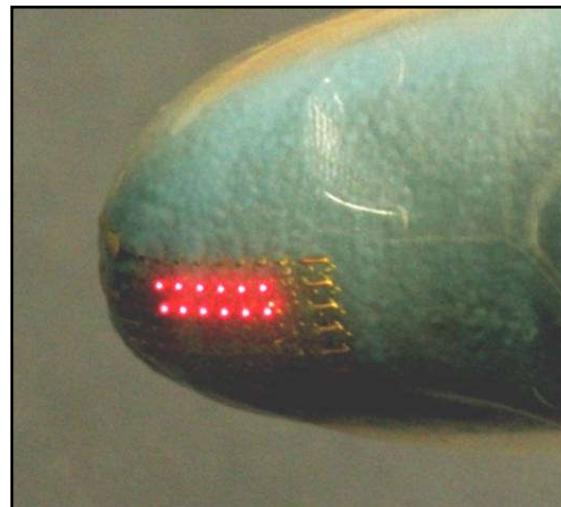
- **GaAs and AlAs**

- lattice matched growth
- AlAs is selectively etched by HF

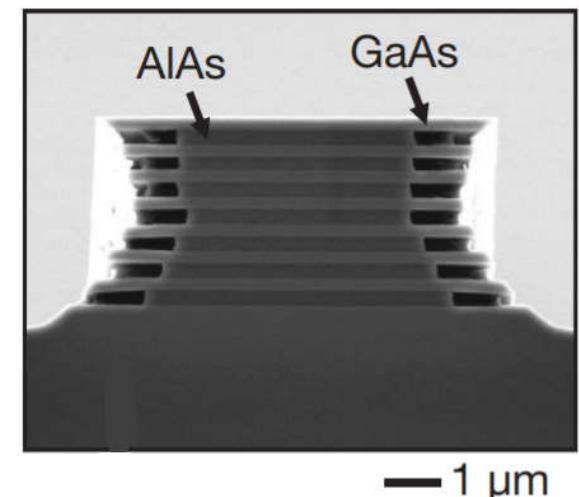
- **flexible III-V devices**



solar cells



LED



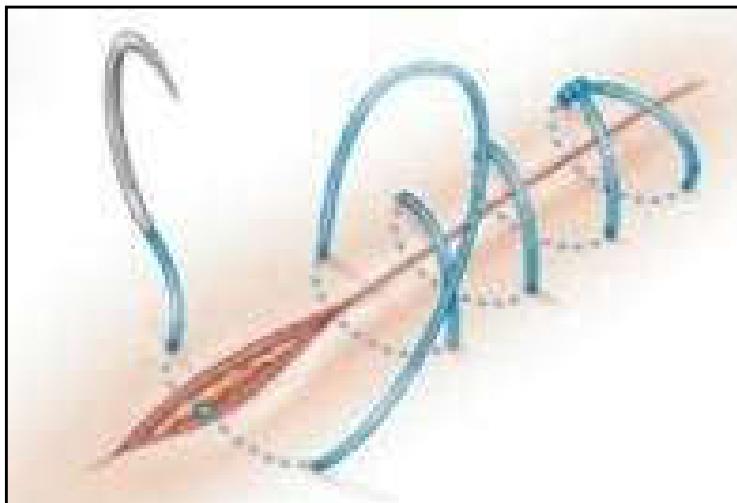
S. I. Park, et al., *Science* **325**, 977 (2009)
J. Yoon, et al., *Nature* **465**, 329 (2010)

Epitaxy Lift-off

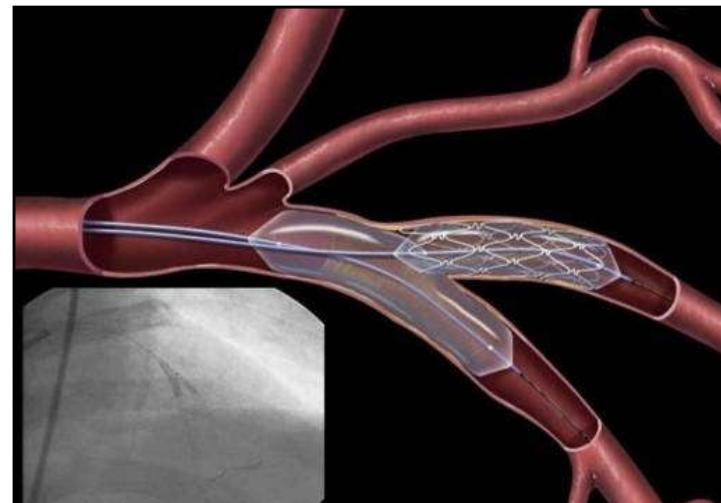
Materials	Sacrificial layers	Substrates	Release methods
Si	SiO ₂	Si	HF wet etch
Si (111)	-	Si (111)	KOH wet etch
Ge	SiO ₂	Si	HF wet etch
SiC	SiO ₂	Si	HF wet etch
GaAs / InGaP	AlAs	GaAs	HF / HCl wet etch
GaAs / InGaP	InAlP	GaAs	HCl wet etch
InGaAs / InP	InGaAs	InP	FeCl ₃ wet etch
GaN	ZnO	sapphire	HCl wet etch
GaN	-	Si (111)	KOH wet etch
InAs	InGaSb	GaSb	NH ₄ OH wet etch

Bio-degradable Materials

Materials that can be dissolved in the body.



Biodegradable Suture



Cardiovascular Stent

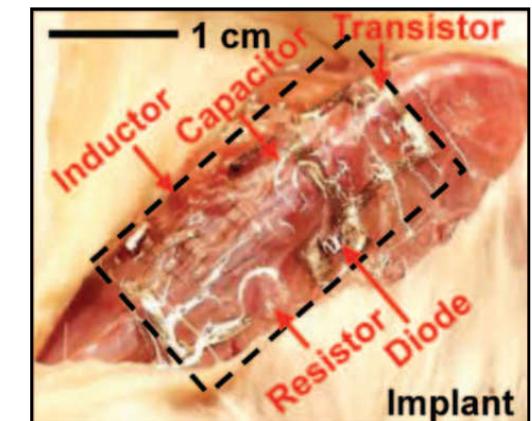
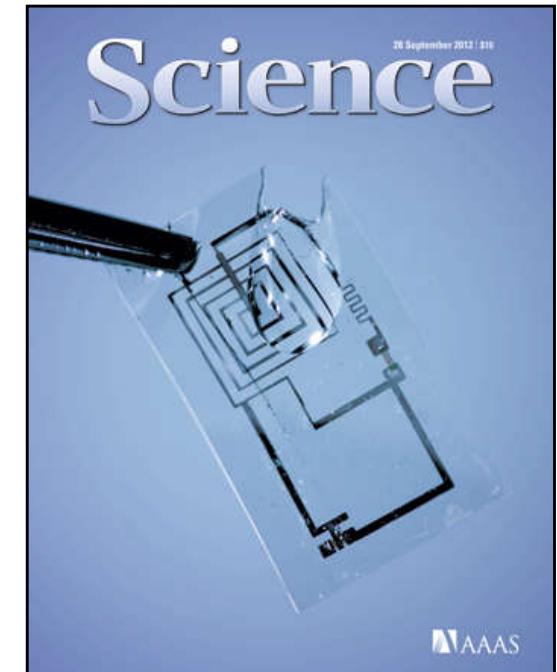
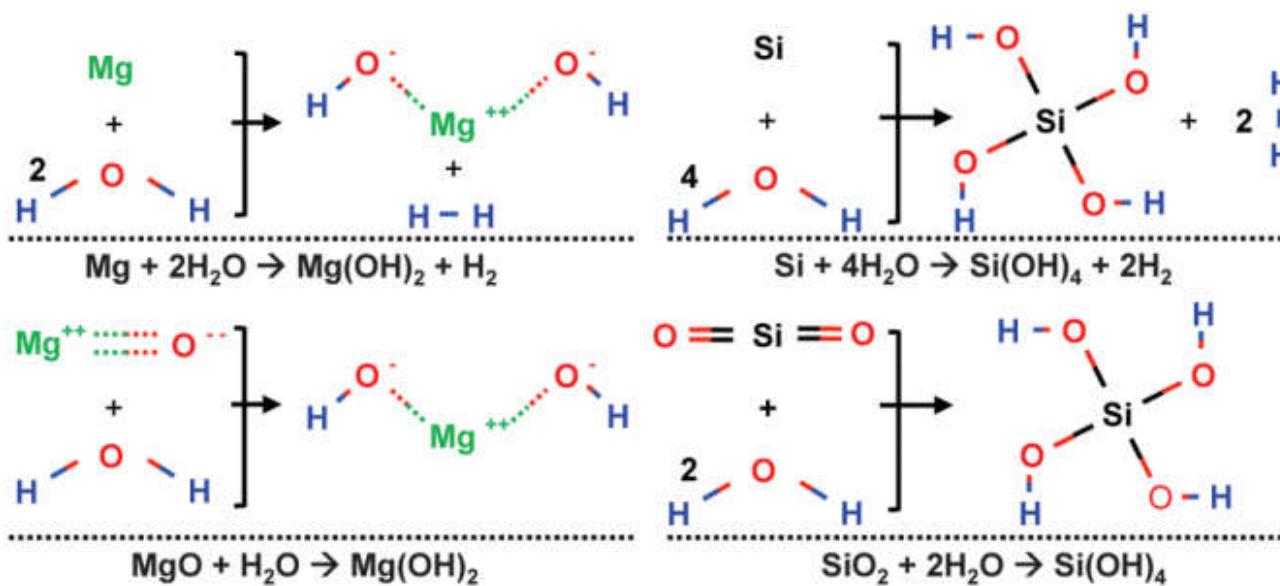
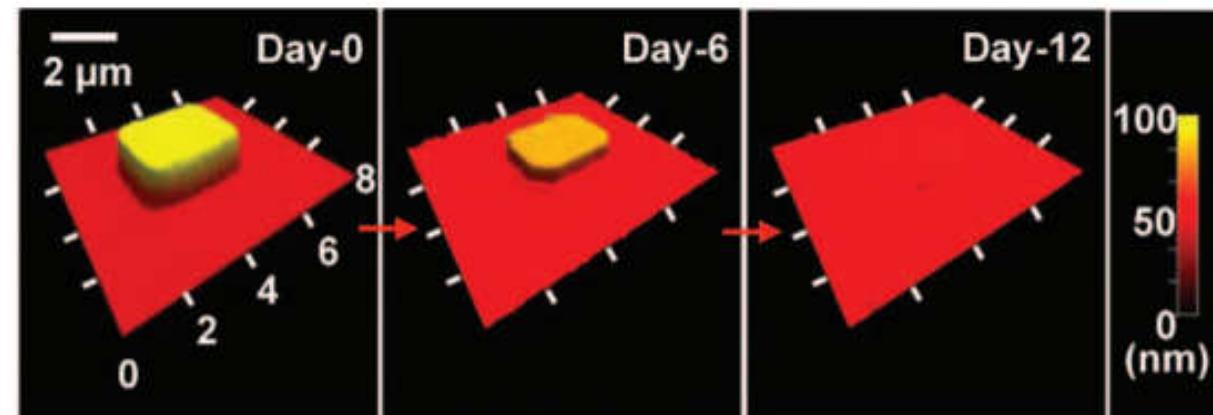


Bone Scaffold

- Biocompatible and Degradable Materials
 - Organic: PLGA, PLA, silk, ...
 - Metals: Mg, Ca, Zn, Fe, ...
 - **Semiconductors:** Si, Ge, ...

Bio-degradable Electronics

Si devices that can be dissolved by body fluids.

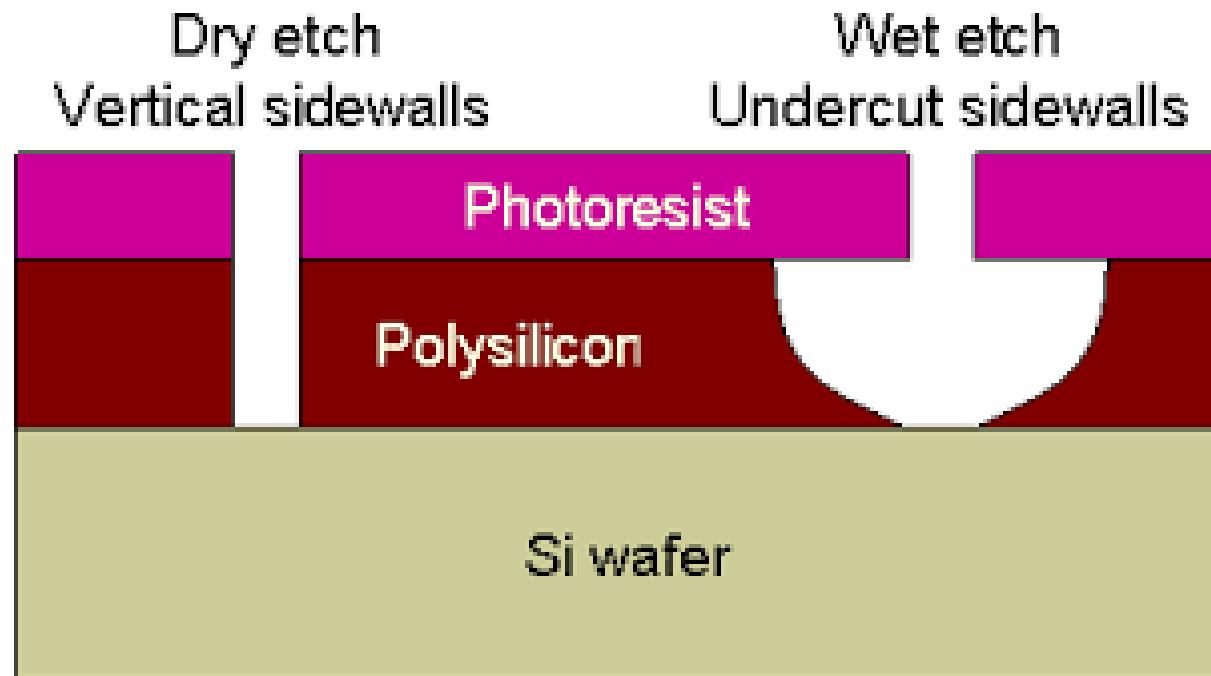


Etching Methods

- **Wet Etching 湿法刻蚀**
- **Dry Etching 干法刻蚀**
- **CMP and other methods**

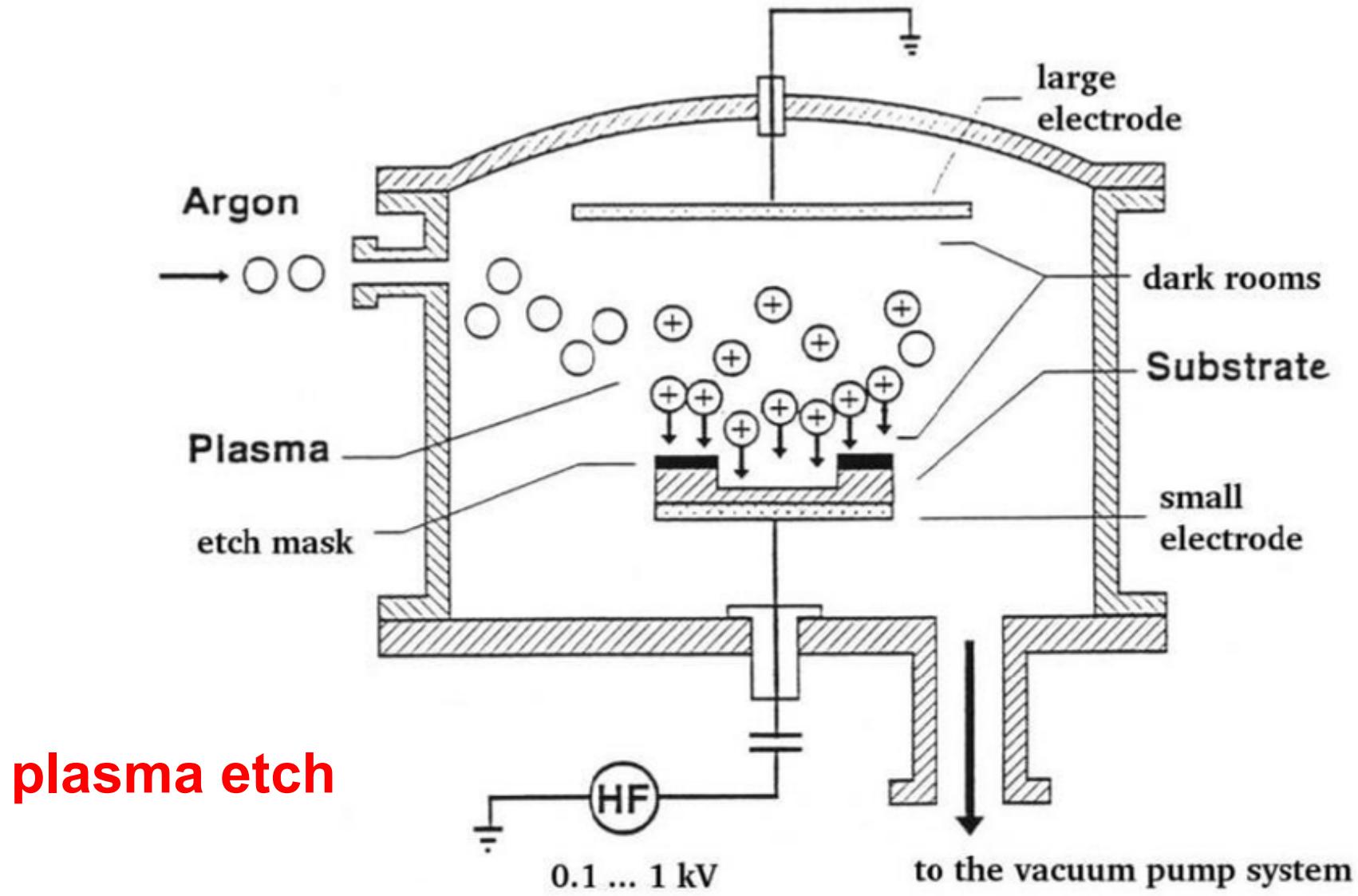
Challenges for Wet Etching

- Most wet etching processes are chemical, isotropic



- For features $< 3 \mu\text{m}$, dry etching has much better resolution

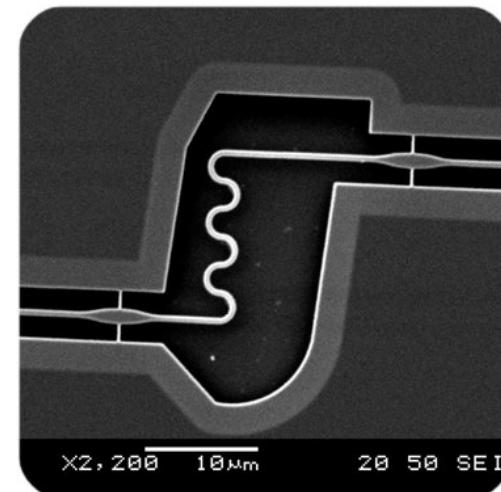
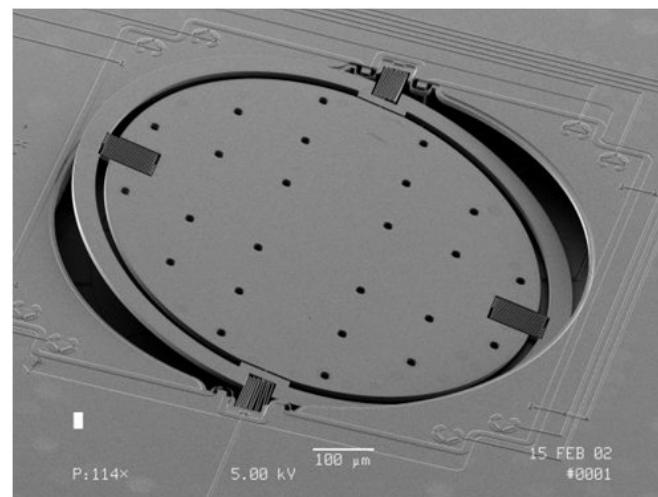
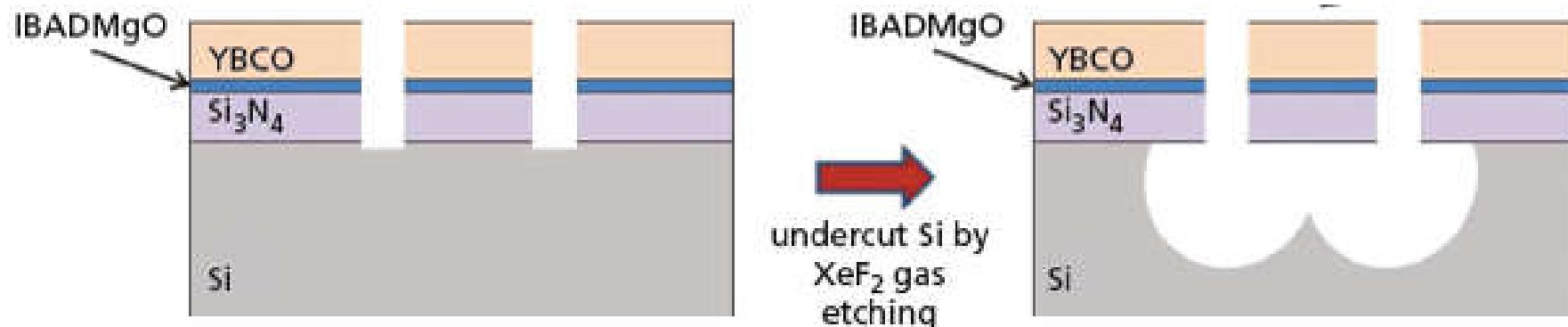
Dry Etching



Dry Etch without Plasma

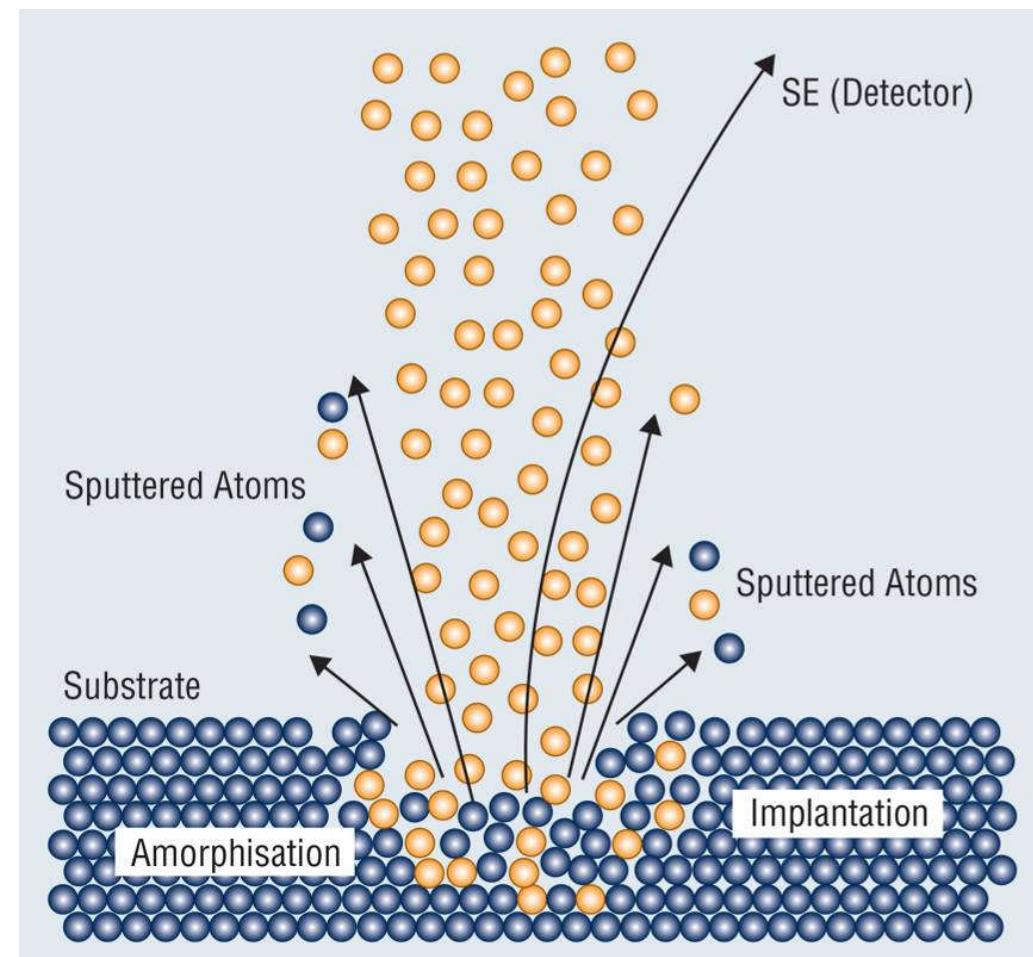
- $2\text{XeF}_2(\text{g}) + \text{Si}(\text{g}) = 2\text{Xe}(\text{g}) + \text{SiF}_4(\text{g})$
 - very isotropic and selective

SiF_4 boiling point 4 °C



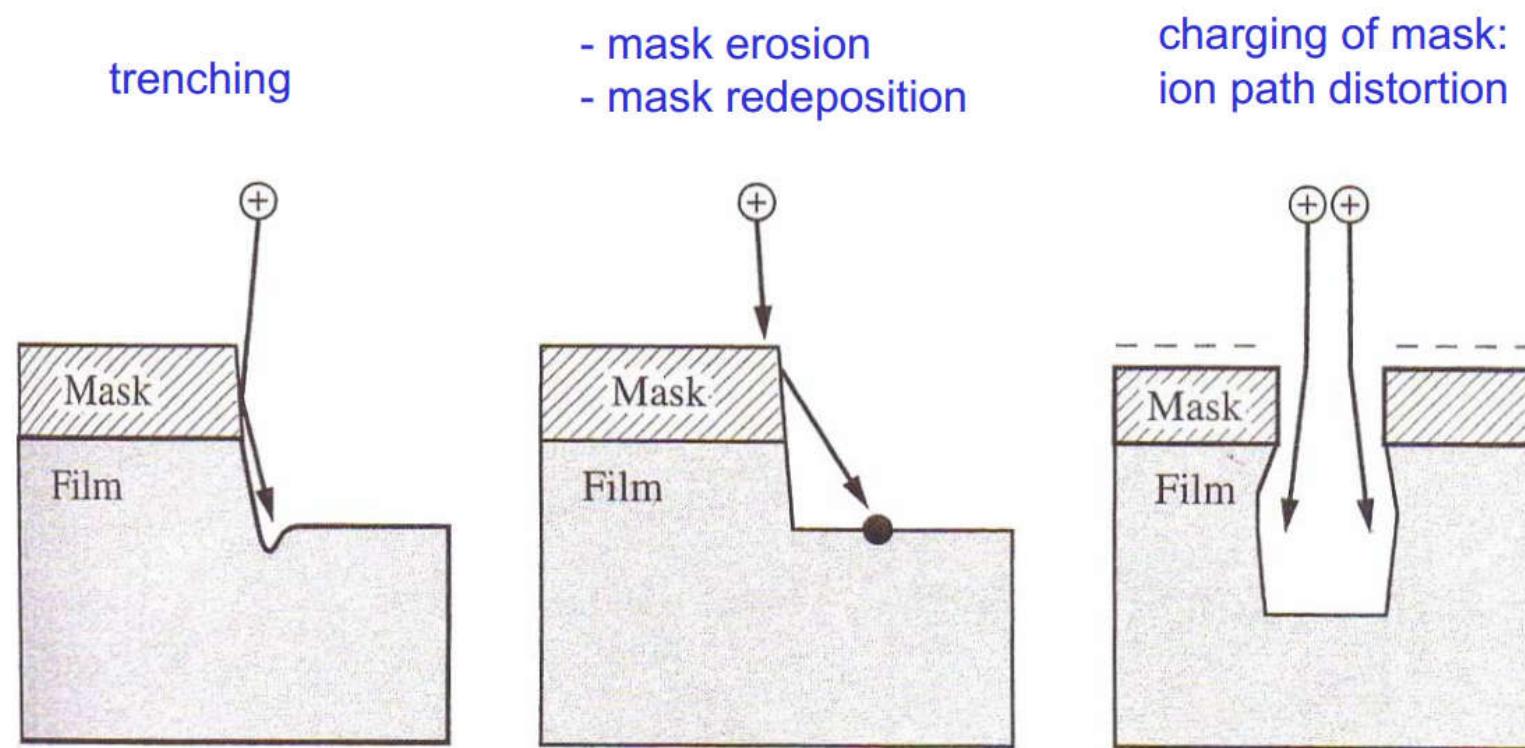
Ion Milling

- Heavy ions (e.g. Ar)
- Highly anisotropic
- Poor selectivity
 - for Au, Pt, ...



Ion Milling

- Heavy ions (e.g. Ar)
- Highly anisotropic
- Poor selectivity



Plasma Etch

- **Chemically reactive ions**

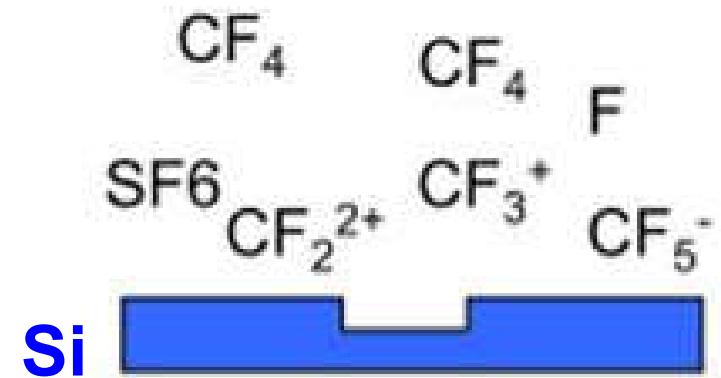
- **improved selectivity**
 - **lower power**

- **Example: Si etch**

- **CF₄ gas does not react with Si**
 - **energized F⁻ plasma can react with Si**
 - **SiF₄ is volatile (boiling point 4 °C)**

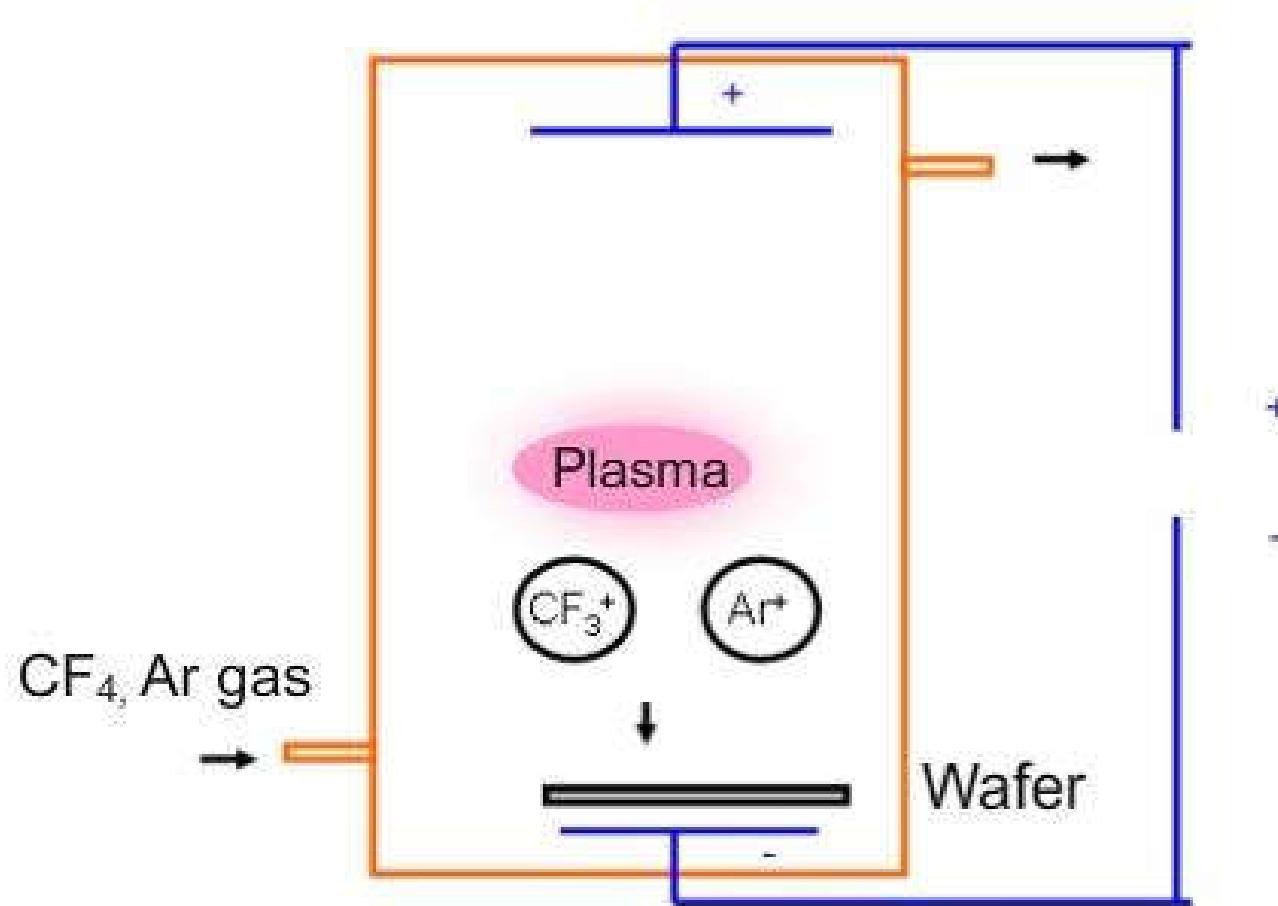
- **Very isotropic**

- **no direction**



Reactive Ion Etching (RIE)

- Improved directionality by applied fields
 - more anisotropic



RIE - Si and SiO₂

- Si
 - SF₆ plasma
- SiO₂
 - CF₄ / CHF₃ plasma
- Photoresists can be used as masks
 - F ions etch PR very slowly

RIE - Si and SiO₂

- Si
 - SF₆ plasma
- SiO₂
 - CF₄ / CHF₃ plasma
- Photoresists can be used as masks
 - F ions etch PR very slowly



SF₆ is heavier than air

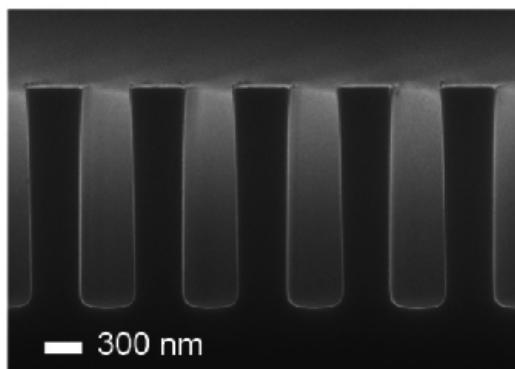
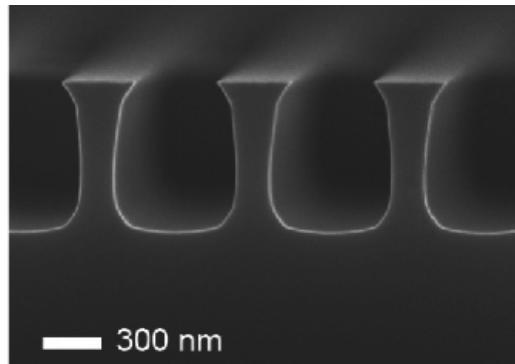
RIE - Organics

- **O₂ plasma**
 - C-H-O + O⁻ = CO₂ + H₂O
- **O₂ plasma does not etch Si, SiO₂, or metals**
 - SiO₂ / metal oxides are non-volatile

RIE - III-Vs

- Cl_2 / BCl_3 / SiCl_4 plasma
 - GaAs/AlGaAs, InP, GaN/InGaN, ...

Q: why?



GaAs trenches

product	boiling point (°C)
GaF_3	1000
AlF_3	> 1000
InF_3	> 1000
GaCl_3	200
AlCl_3	180
AsCl_3	130

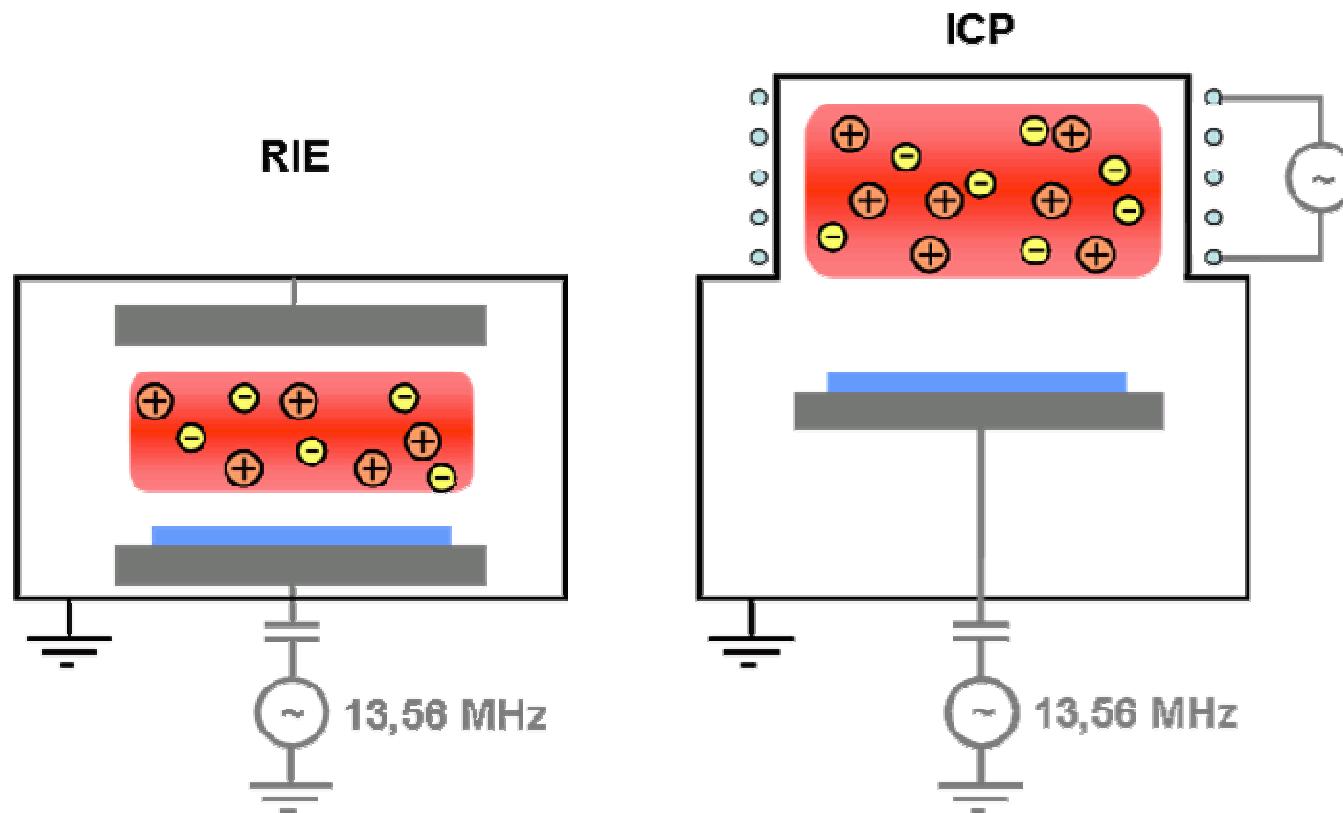
RIE - Recipes

Table 6.2 Materials and corresponding RIE gases

Materials to be etched	Chemical gases (multi choices)
Single-crystal silicon	CF_3Br , HBr/NF_3 , SF_6/O_2
Polysilicon	$\text{SiCl}_4/\text{Cl}_2$, BCl_3/Cl_2 , $\text{HBr}/\text{Cl}_2/\text{O}_2$, HBr/O_2 , Br_2/SF_6
Al	$\text{SiCl}_4/\text{Cl}_2$, BCl_3/Cl_2 , HBr/Cl_2
Al-Si-Cu, Al-Cu	$\text{BCl}_3/\text{Cl}_2 + \text{N}_2$
W	SF_6 , NF_3/Cl_2
TiW	SF_6
WSi_2 , TiSi_2 , CoSi_2	$\text{CCl}_2\text{F}_2/\text{NF}_3$, CF_4/Cl_2
SiO_2	CCl_2F_2 , CHF_3/CF_4 , CHF_3/O_2 , CH_3CHF_2
Si_3N_4	CF_4/O_2 , CF_4/H_2 , CHF_3 , CH_3CHF_2
GaAs	$\text{SiCl}_4/\text{SF}_6$, $\text{SiCl}_4/\text{NF}_3$, $\text{SiCl}_4/\text{CF}_4$
InP	CH_4/H_2
Photoresists	O_2

ICP-RIE

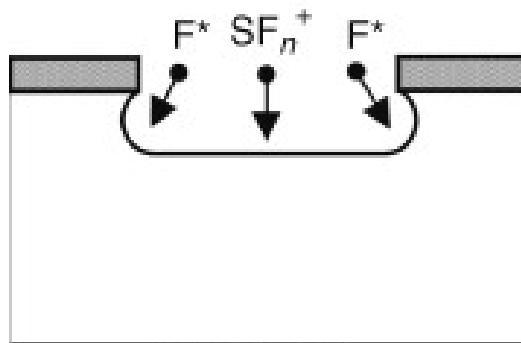
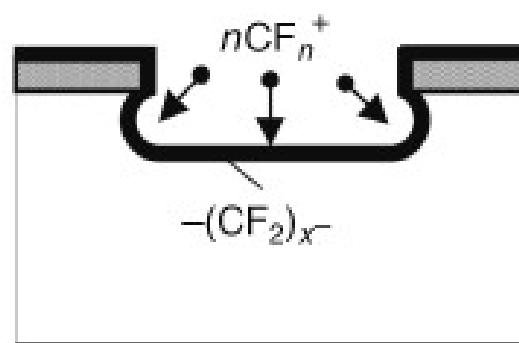
- Inductively Coupled Plasma (ICP)
 - higher power
 - mostly for III-Vs



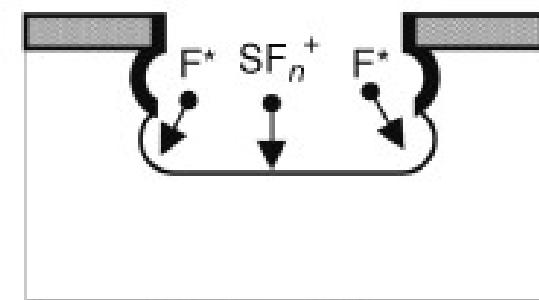
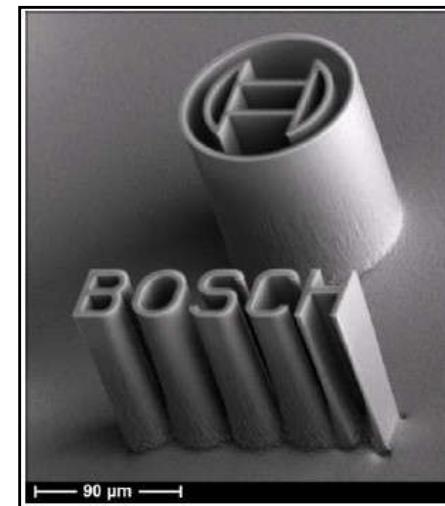
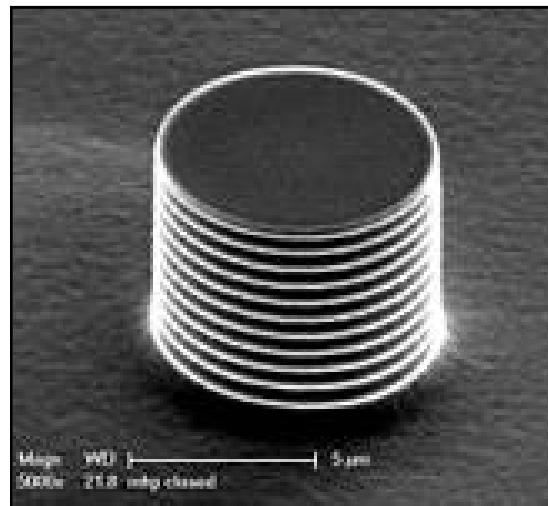
Deep RIE for Si

- alternative etch / passivation

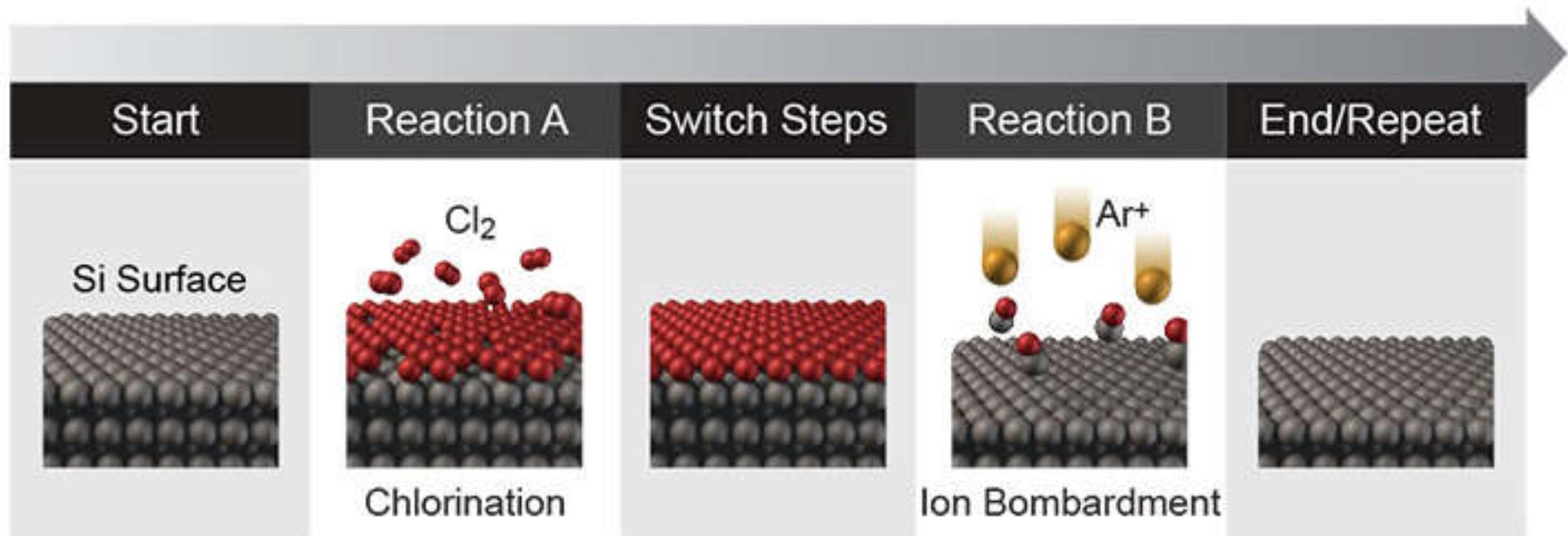
'Bosch process'

(i) SF_6 etching

(ii) Passivation

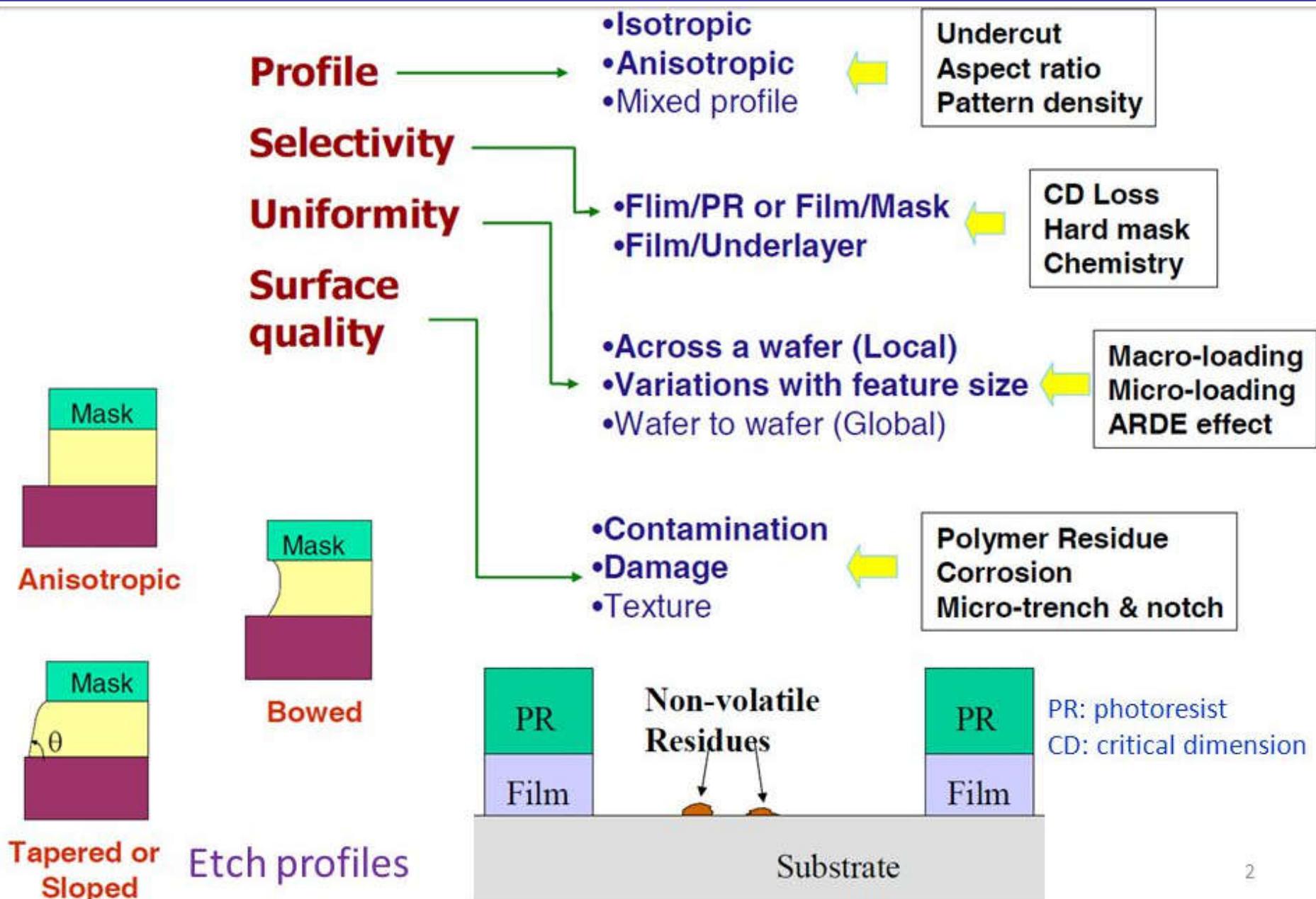
(iii) SF_6 etching

Atomic Layer Etching (ALE)



1. $\text{Si} + 2\text{Cl}_2 = \text{SiCl}_4$
2. SiCl_4 removed by plasma
3. Repeat 1 and 2

Issues in Dry Etch



Summary of Dry Etch

<u>Type of Etching</u>	<u>Excitation Energy</u>	<u>Pressure</u>
Gas/Vapor Etching	none	high (760-1 torr)
- <i>isotropic, chemical, very selective</i> (e.g. XeF_2 gas etch Si even without plasma)		
Plasma Etching	10's to 100's of Watts	Medium (>100 torr)
- <i>isotropic, chemical, selective</i>		
Reactive Ion Etching	100's of Watts	Low
- <i>directional, physical & chemical, fairly selective</i>	(10-100 mtorr)	
Sputter Etching	100's to 1000's of Watts	Low (~10 mtorr)
- <i>directional, physical, low selectivity</i> (e.g. ion beam etching/milling using Ar^+)		

Etching Methods

- **Wet Etching 湿法刻蚀**
- **Dry Etching 干法刻蚀**
- **CMP and other methods**

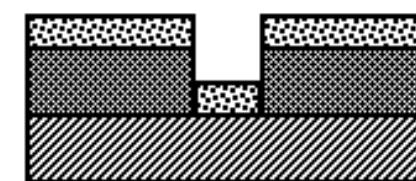
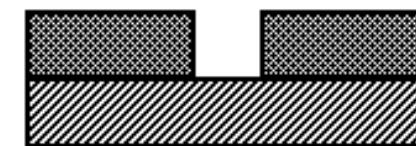
Pattern Formation

Subtractive Process



Pattern transfer
by etching

Additive Process



Pattern transfer
by lift off

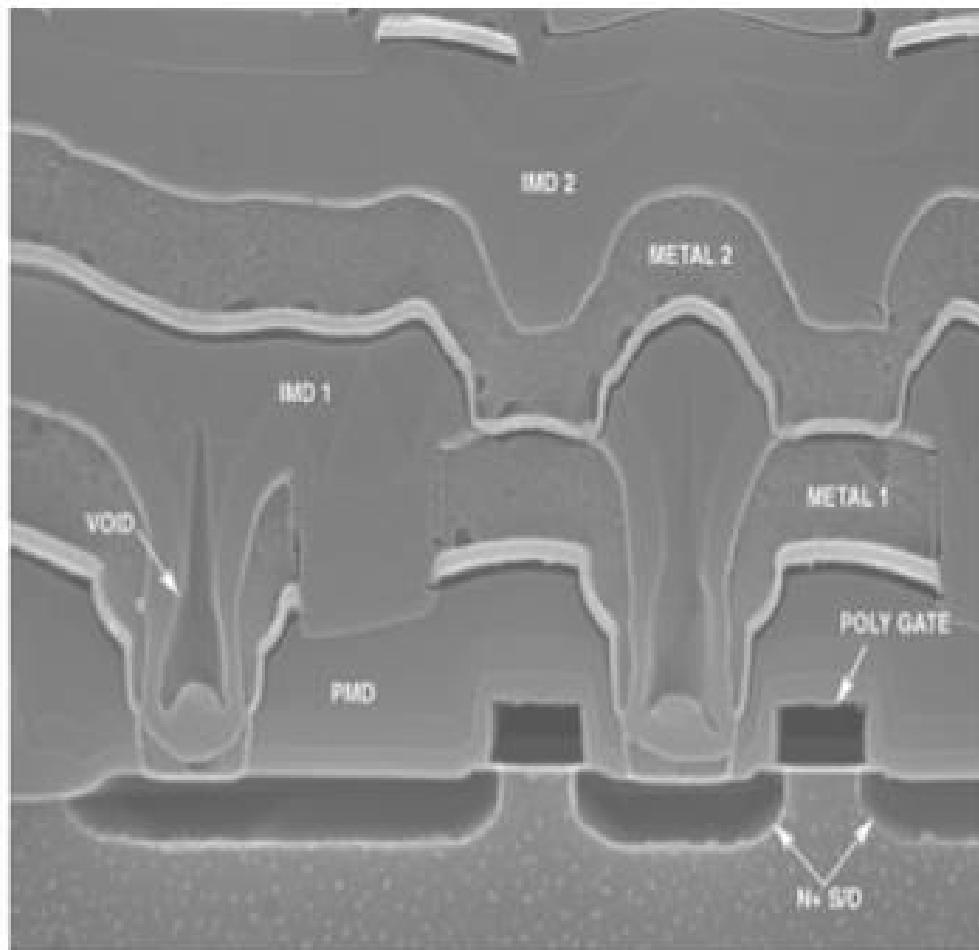
Photolithography

Etch

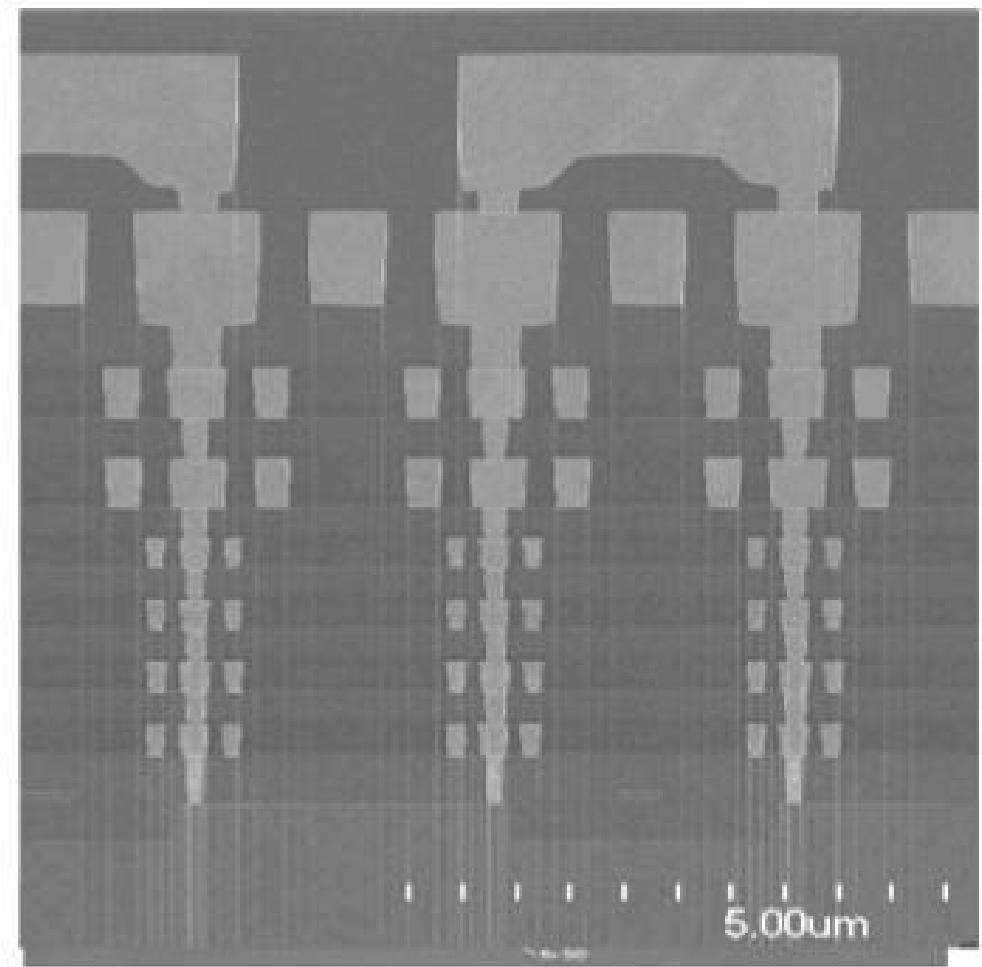
Deposit

Strip Resist

Planar Layers are Desirable

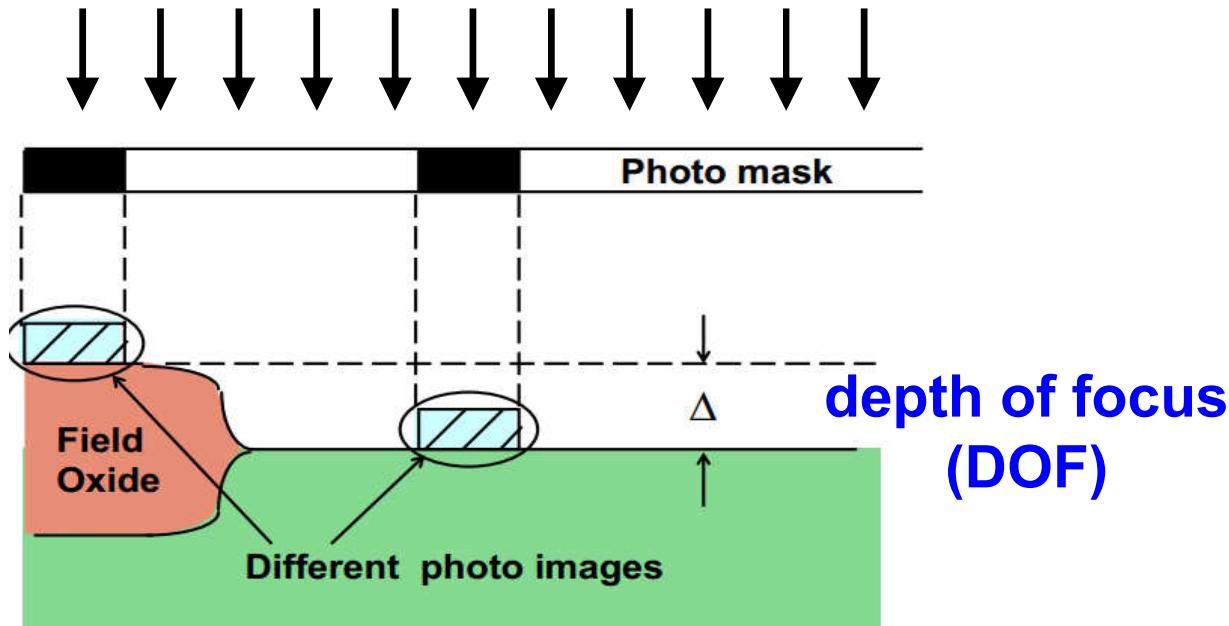


non-planarized IC



planarized IC

Issues with Depth of Focus (DOF)

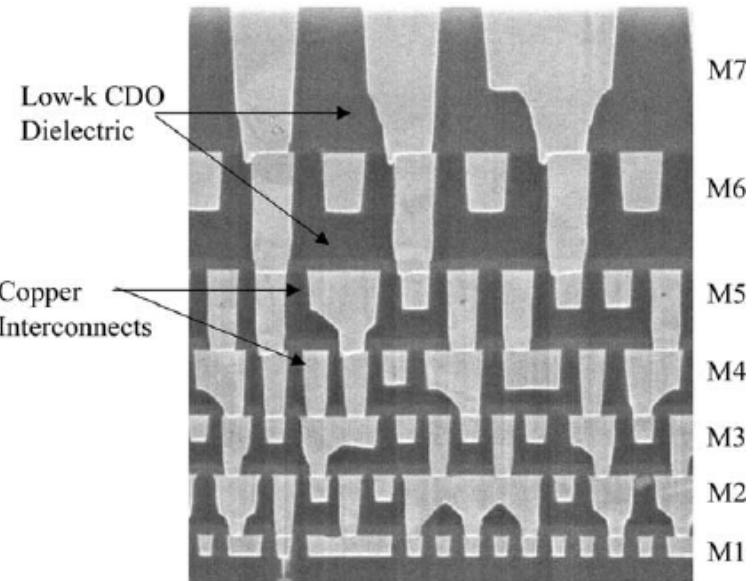
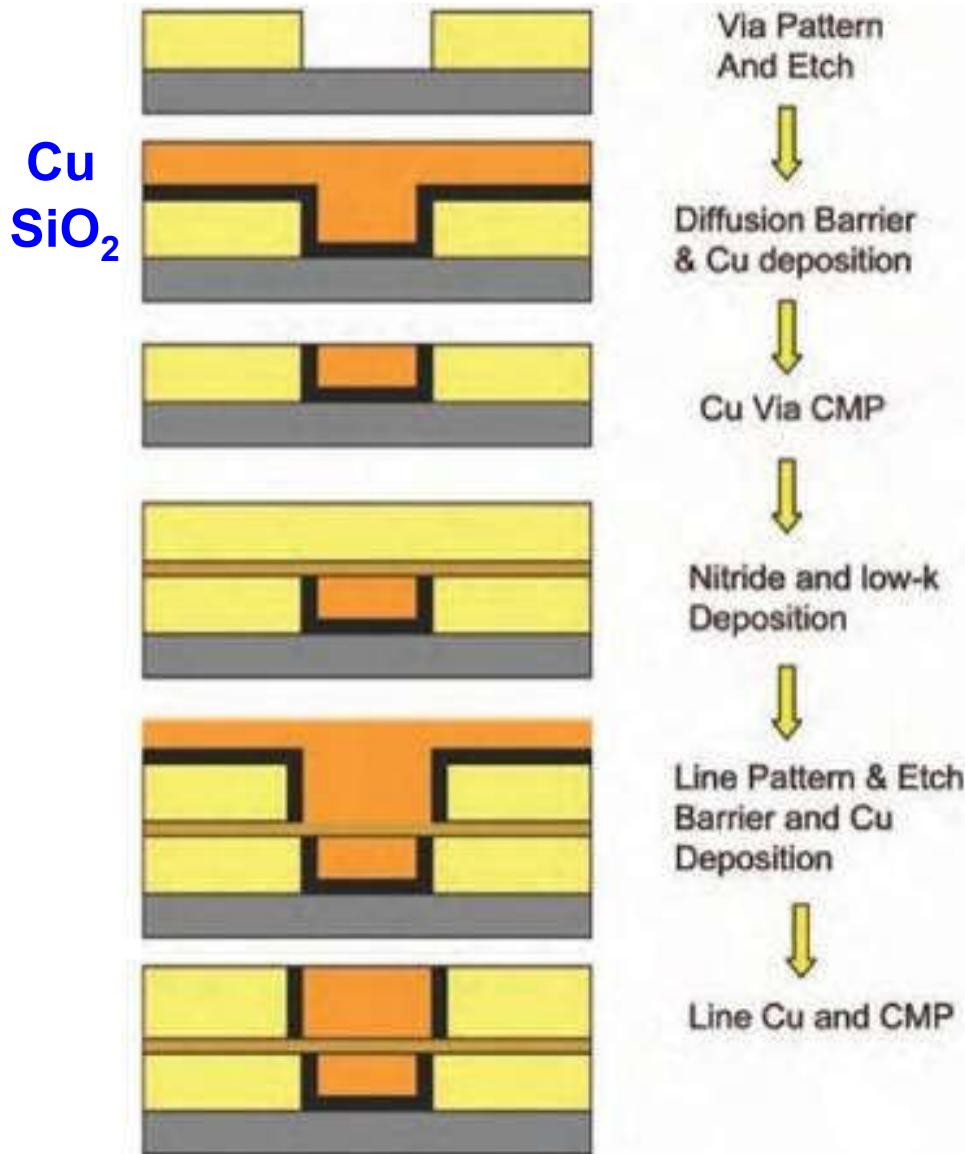


$$(1) l_m \cong 0.6 \frac{\lambda}{NA} \quad \text{want small } l_m$$

$$(2) DOF = \pm \frac{\lambda}{2(NA)^2} \quad \text{want large } DOF$$

smaller λ , larger NA
----> smaller DOF

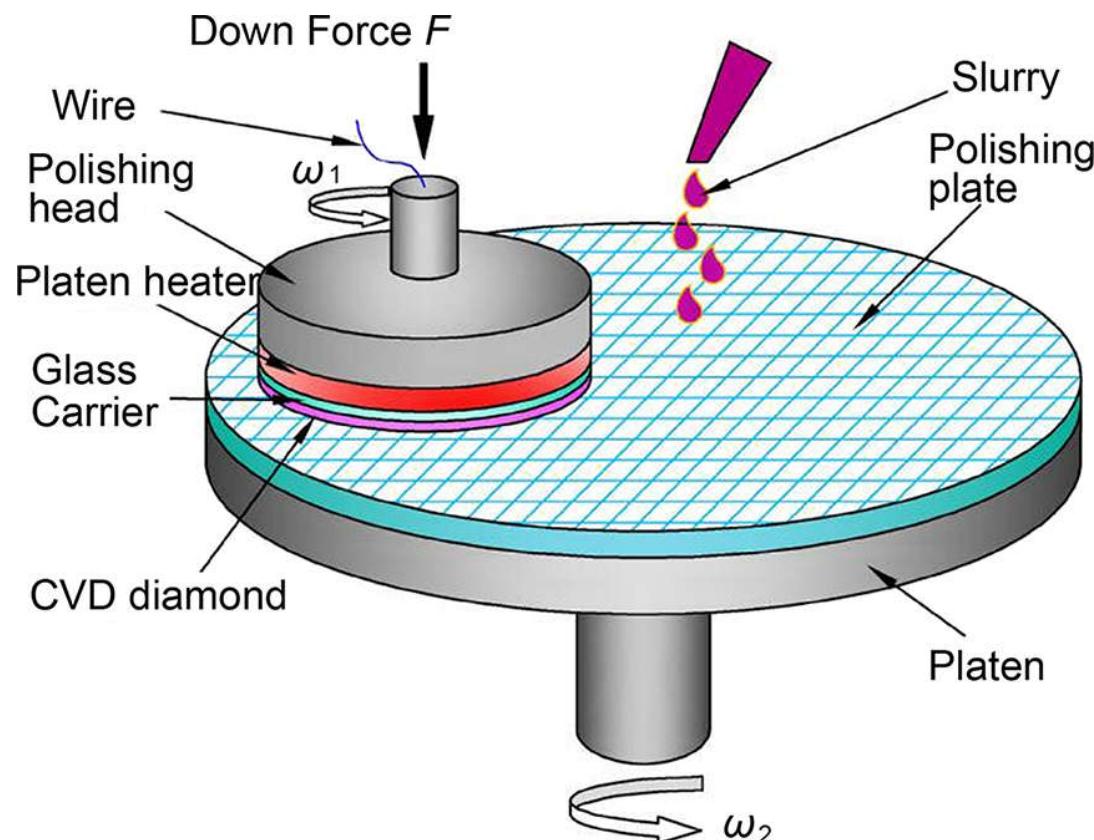
Damascene Process



ancient art work

CMP: Chemical Mechanical Polishing

- Chemical selectivity + Mechanical Planarization
- applied for Cu, W, SiO₂, ...

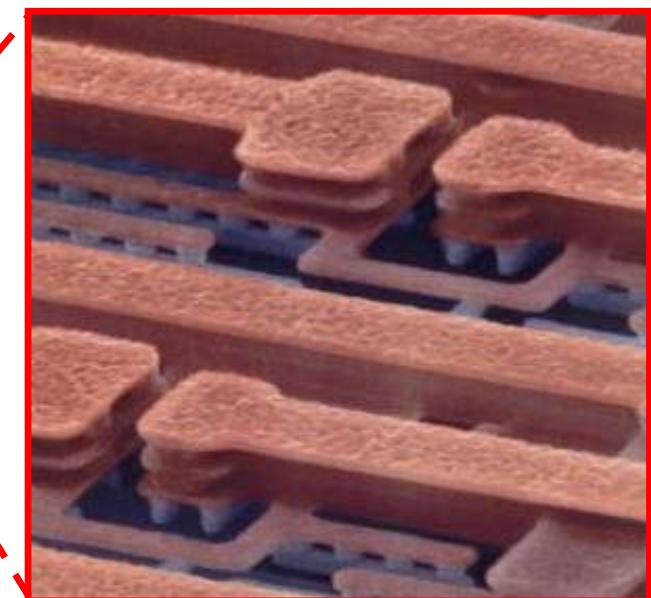
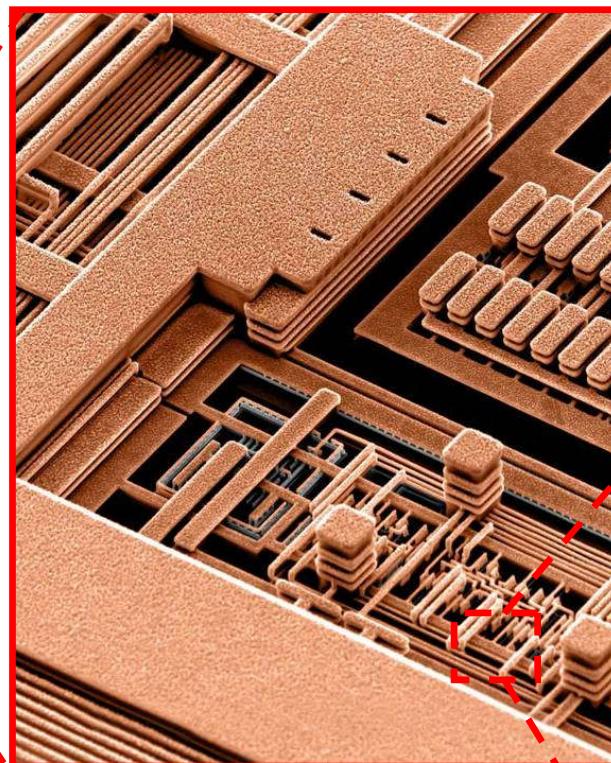


Video

3D IC



Electroplating + CMP
dirtiest process for the most advanced IC



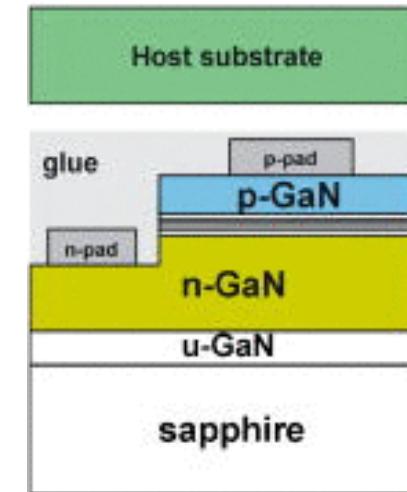
Other Methods for Etching

- **Laser Lift-Off**
- **FIB: Focused Ion Beam**
- **Laser Milling**
- ...

GaN Laser Liftoff

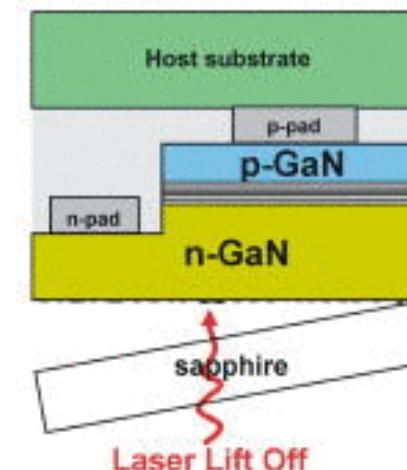
- GaN devices grown on sapphire

- low cost
- low thermal conductivity
- electrically insulating
- sapphire is very difficult to etch

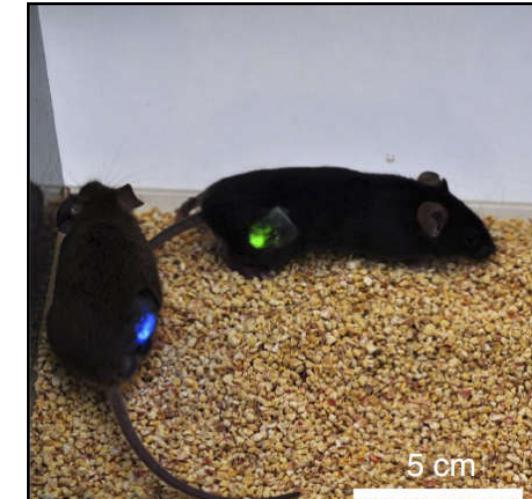
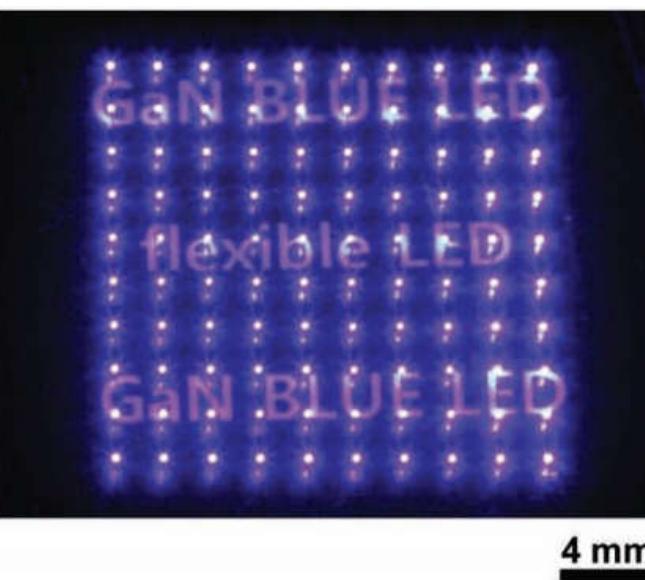
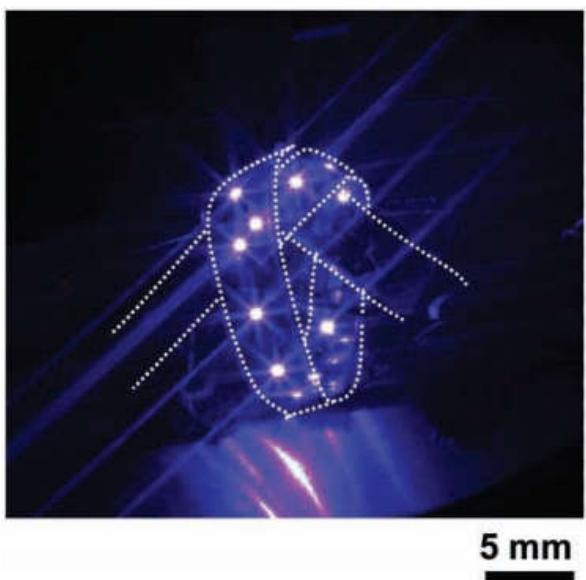
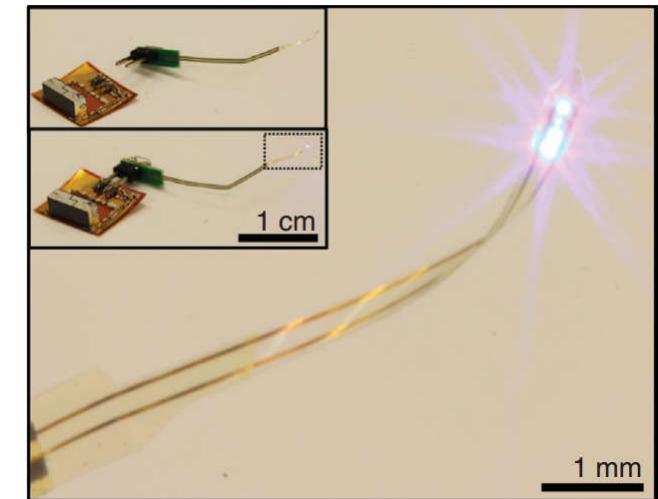
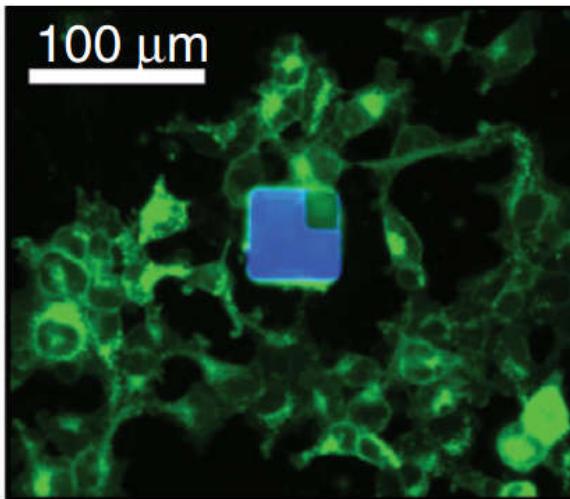
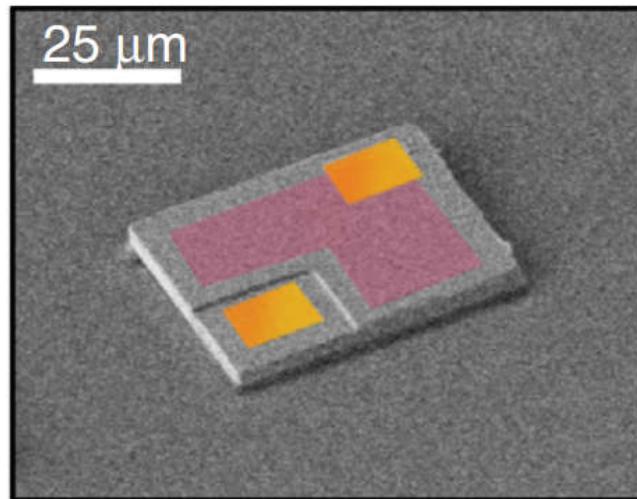


- Release by laser liftoff

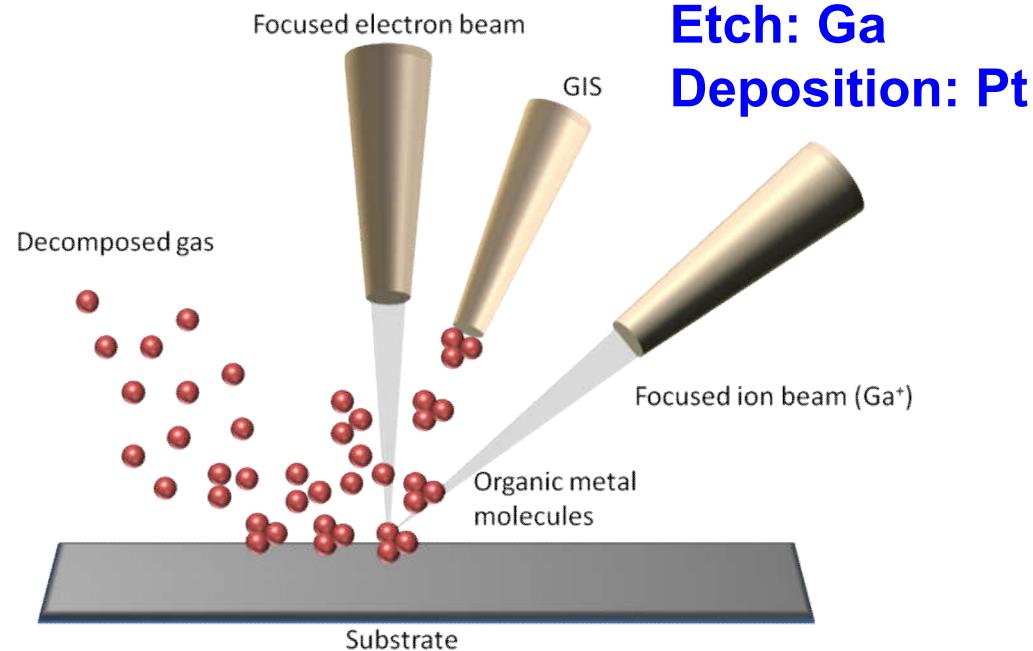
- KrF excimer UV laser (248 nm)
- $\text{GaN} = \text{Ga} + \text{N}_2$ (gas)
- bonding onto new substrates



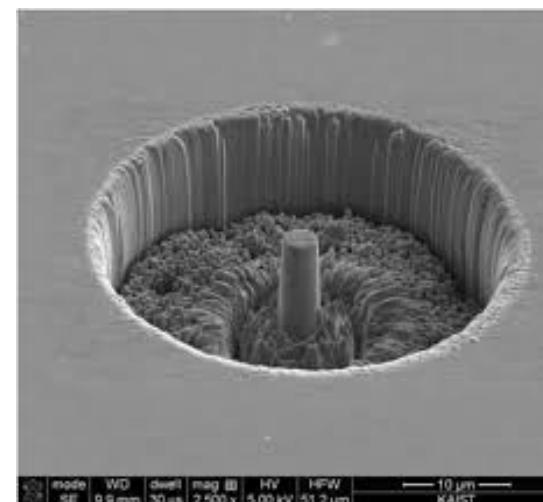
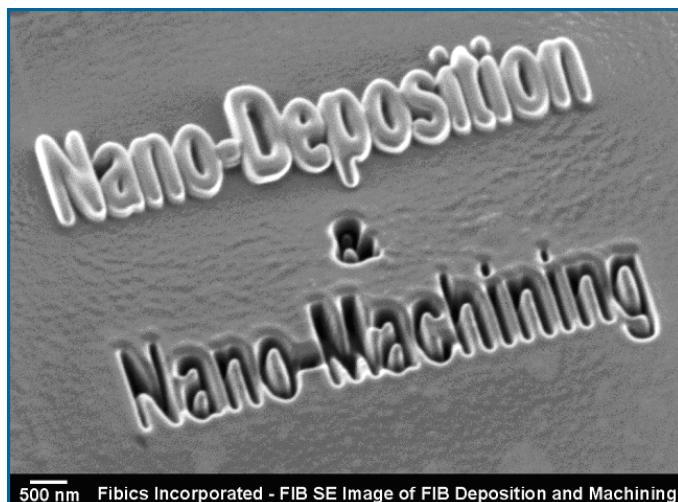
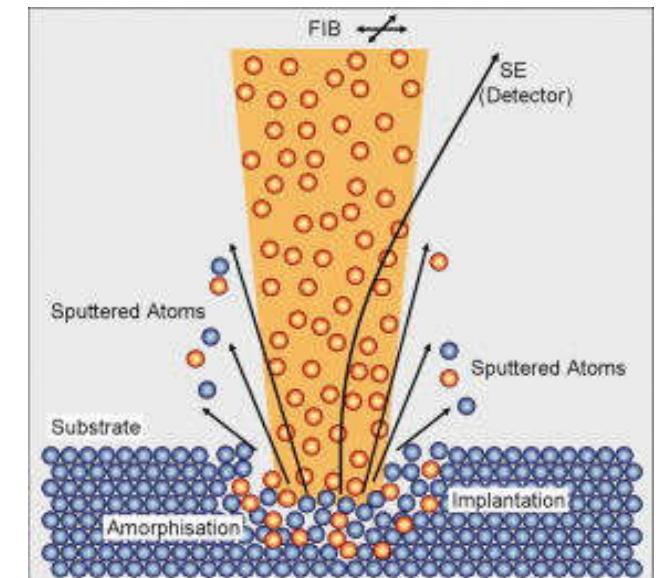
Flexible GaN blue LEDs



FIB: Focused Ion Beam



**Etch: Ga
Deposition: Pt**



- **nanostructures**
- **SEM/TEM sample preparation**
- **doping**

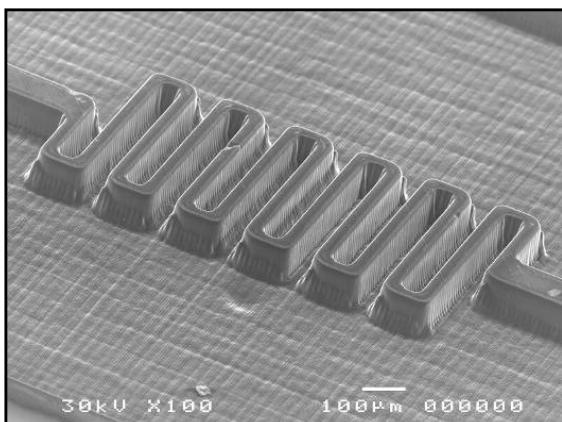
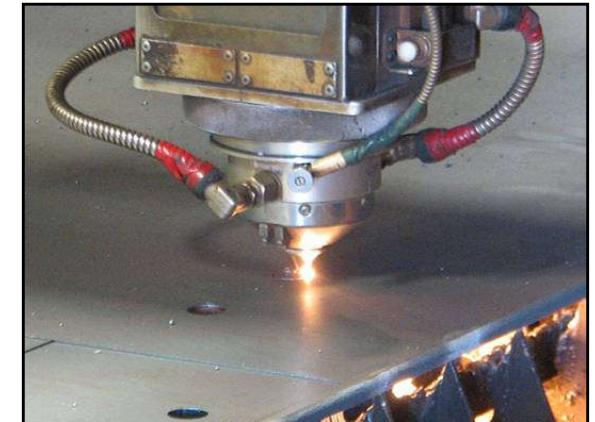
Laser Milling

■ Types

- CO₂ laser 10 μm
- Near-IR laser 1064 nm
- UV laser 365 nm
- Excimer laser 248 nm, 193 nm
- ps/fs laser ...

■ Applications

- die cut, PCB cut, ...



**shorter wavelength
shorter pulse**



better resolution