微纳光电子材料与器件工艺原理

Film Deposition Part IV: PVD

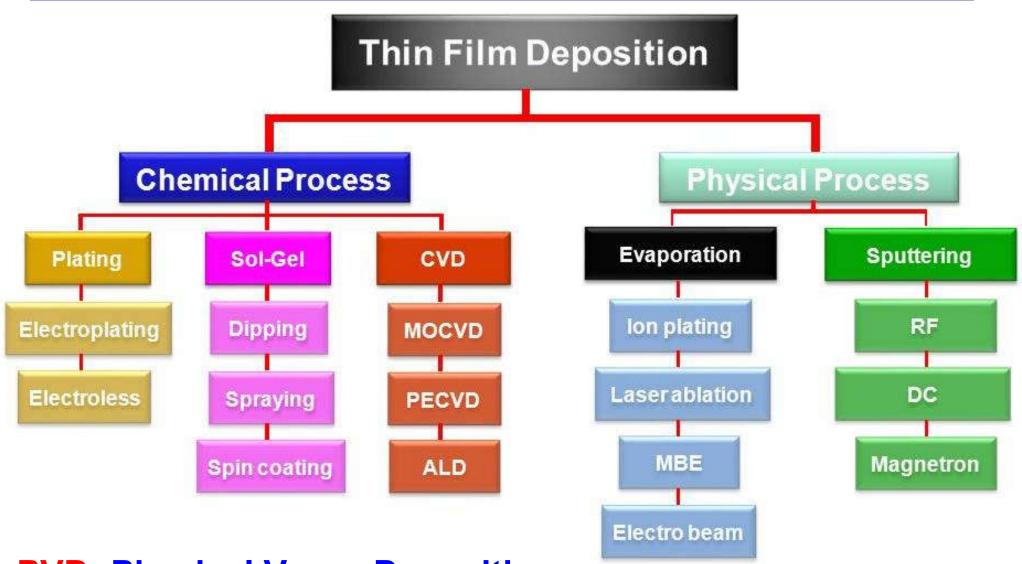
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Film Deposition

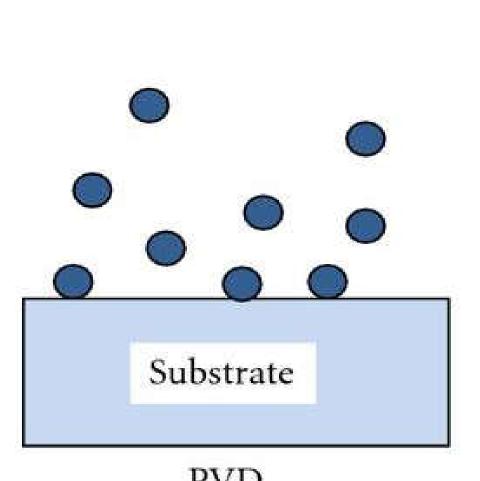


PVD: Physical Vapor Deposition

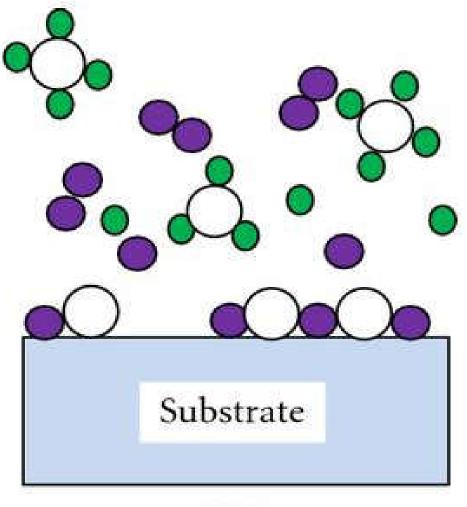
CVD: Chemical Vapor Deposition

PVD vs. CVD

Physical process

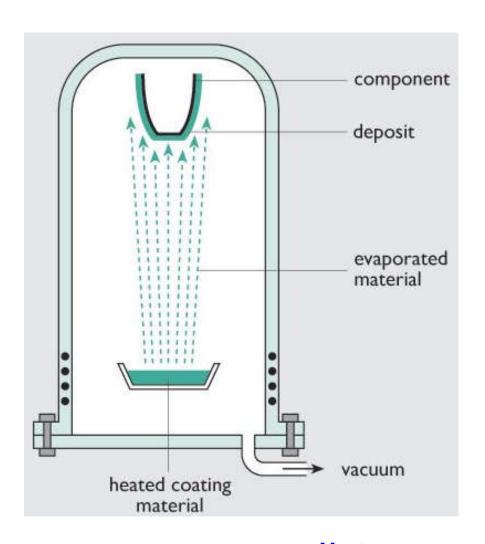


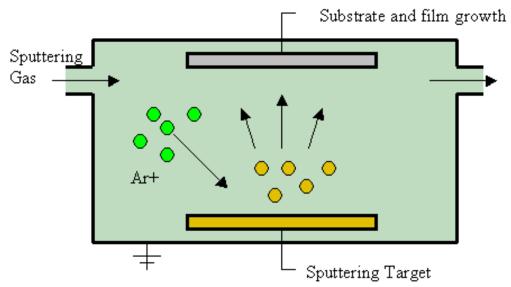
Chemical reactions



CVD

PVD





Evaporation (蒸发)

Sputter (溅射)

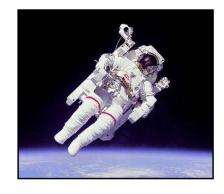
Vacuum Basics

Units

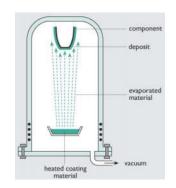
- \Box 1 Pa = 1 N/m²
- **1** 1 atm = 760 torr = 760 mm Hg = $1.013*10^5$ Pa
- **1** bar = 10^5 Pa = 750 torr
- □ 1 torr = 133.3 Pa



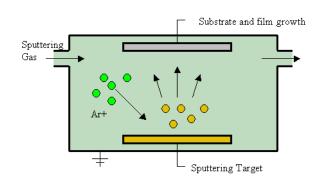
Pressure cooker ~ 1.5 atm



outer space < 10⁻¹⁰ Pa

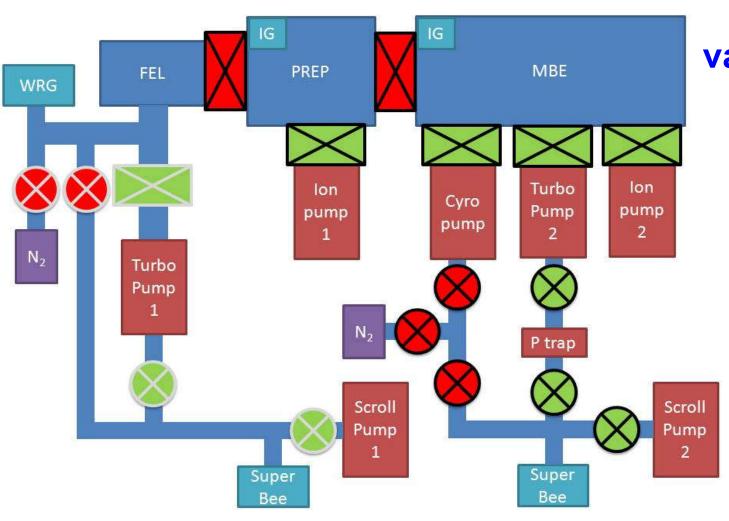


Evaporation < 10⁻⁷ Pa



Sputter ~ 10⁻¹ Pa

Vacuum Systems



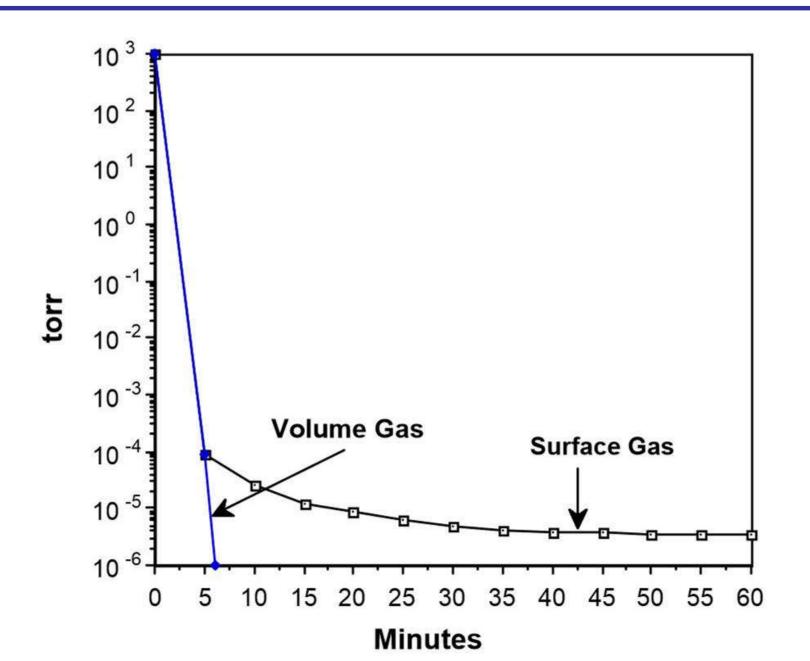
vacuum ~ 10⁻¹⁰ Pa



MBE: Molecular Beam Epitaxy

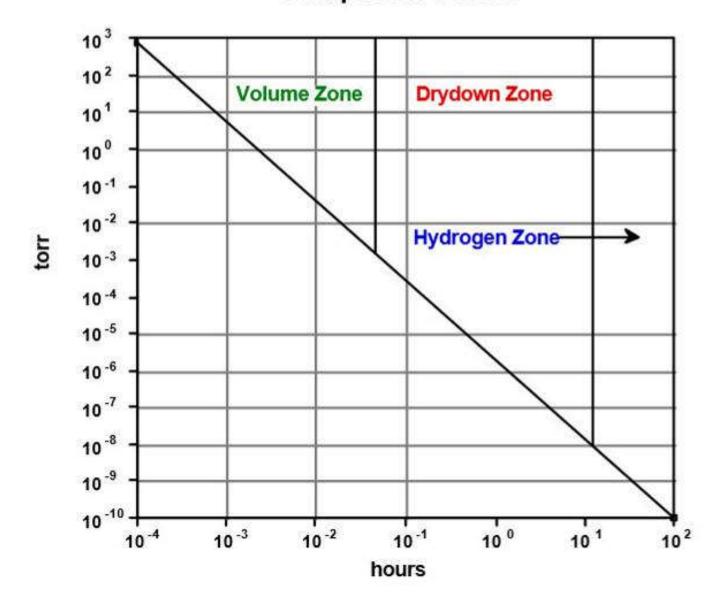
分子束外延

Vacuum Pumpdown



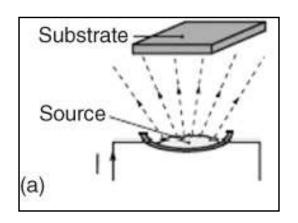
Vacuum Pumpdown

Pumpdown Zones

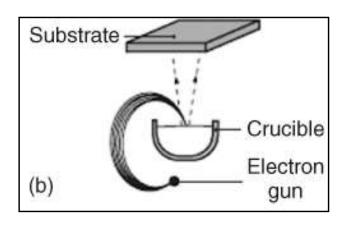


- 1. air
- 2. water
- 3. hydrogen

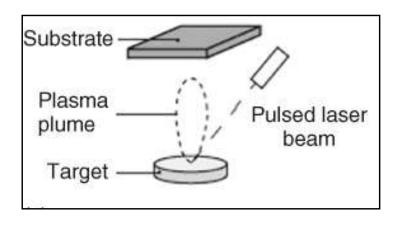
Evaporation



Thermal



Electron Beam (Ebeam)



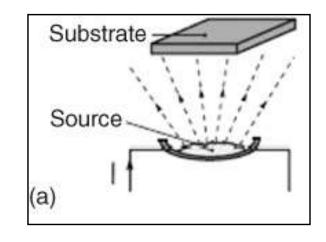
Pulsed Laser Deposition (PLD)

Evaporation

- Reduce the impurities (N₂, O₂, H₂O, ...)
- Prevent oxidation



lacktriangle molecular mean free path λ

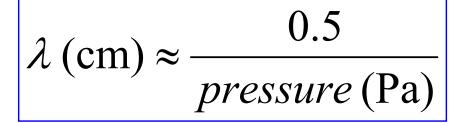


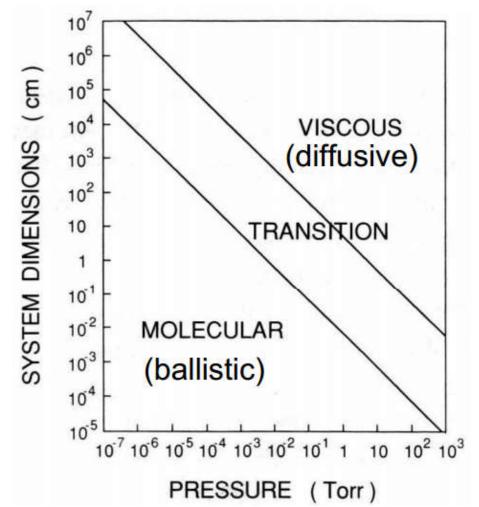
$$\lambda = \frac{kT}{\sqrt{2\pi r^2 p}}$$

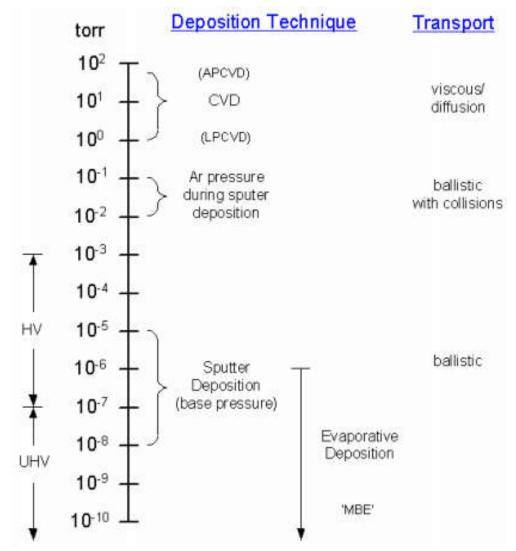
$$\lambda \text{ (cm)} \approx \frac{0.5}{pressure \text{ (Pa)}}$$

Q: Required pressure?

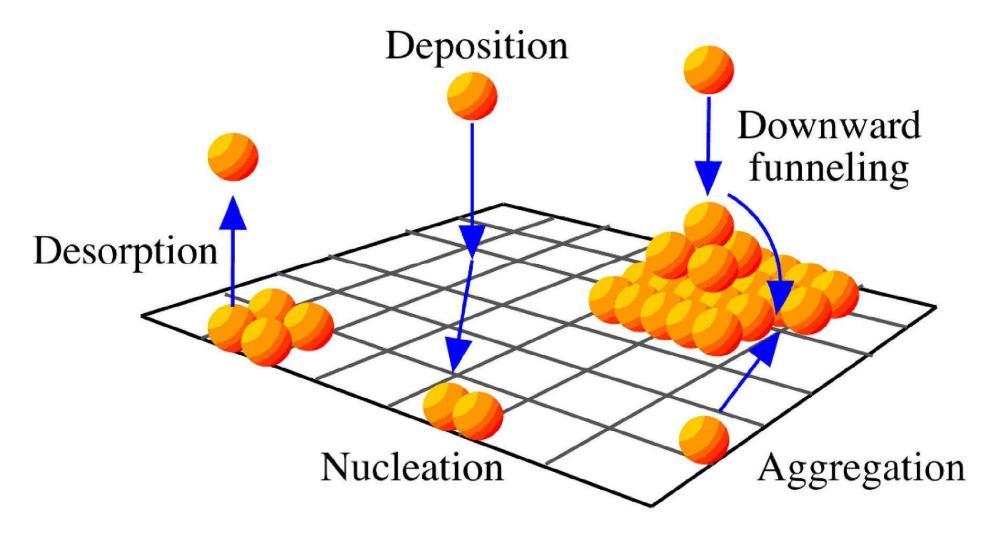
Mass Transport







Mass Transport



absorption - movement - desorption

Evaporation Rate

Langmuir-Knudsen Theory

$$R_{evap} = 5.83 \times 10^{-2} A_s \sqrt{\frac{m}{T}} P_e$$

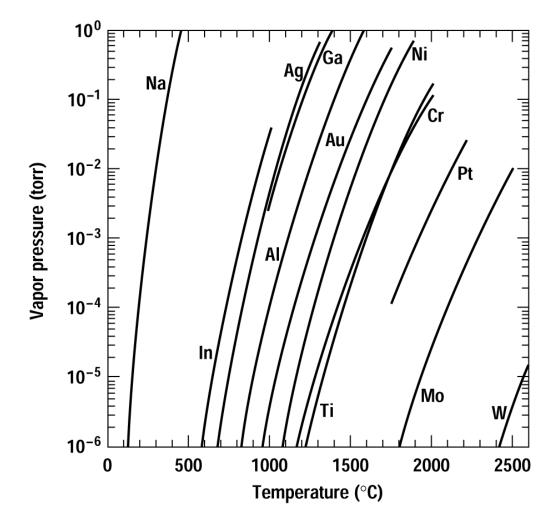
 R_{evap} : Evaporation rate (g/s)

 A_s : area of sources (cm²)

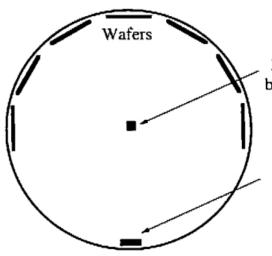
m: molecular weight (g/mol)

T: temperature (K)

P_e: vapor pressure of sources (Torr) (*not* chamber pressure)

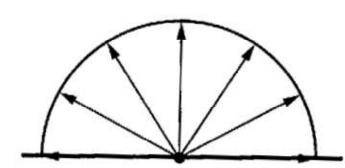


Evaporation Sources

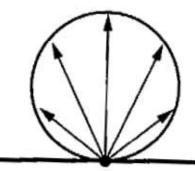


Location of source which behaves like an ideal point source

Location of source which behaves like an ideal small area surface source

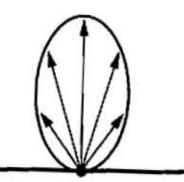


(a) Uniform (isotropic) entission from a point source



 (b) Ideal cosine emission from a small planar surface source.

 $(n = 1 \text{ in } \cos^n \theta)$ distribution)



(c) Non-ideal, more anisotropic emission from a small planar surface source.

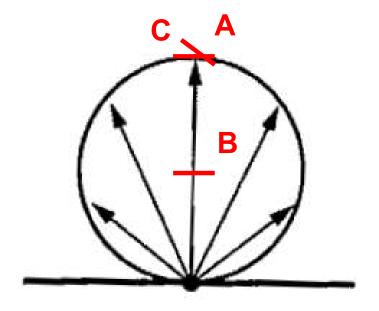
 $(n > 1 \text{ in } \cos^n \theta$ distribution)

Deposition rate

Question:

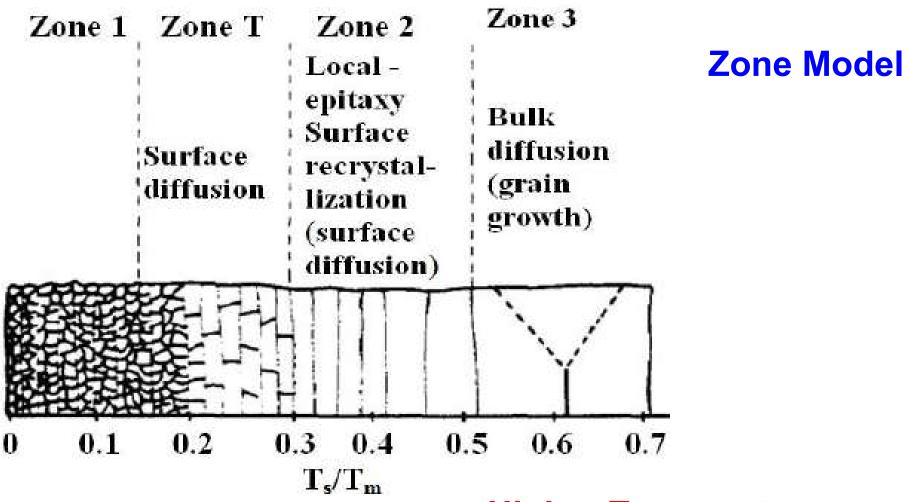
1. R_A : R_B = ?

2. R_A : R_C = ?



Ideal cosine emission from a small planar surface source.

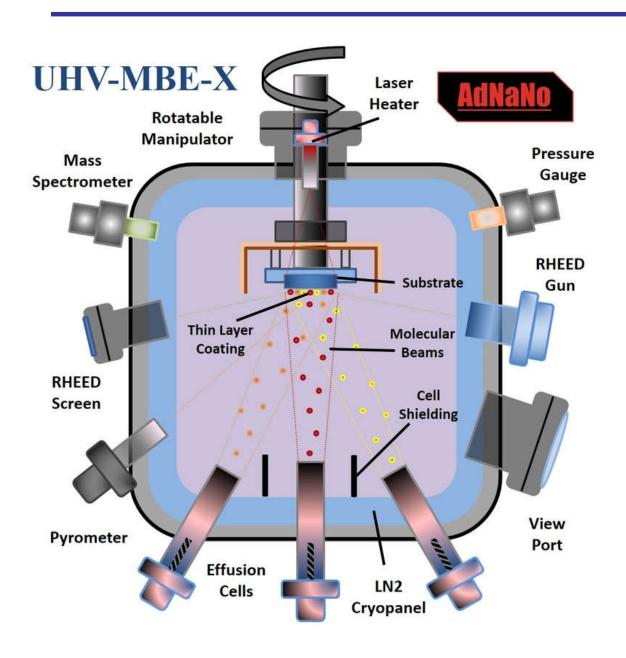
Effects of Substrate Temperature



Higher Temperature

- -> Larger Atom Mobility
- -> Larger Grain Size

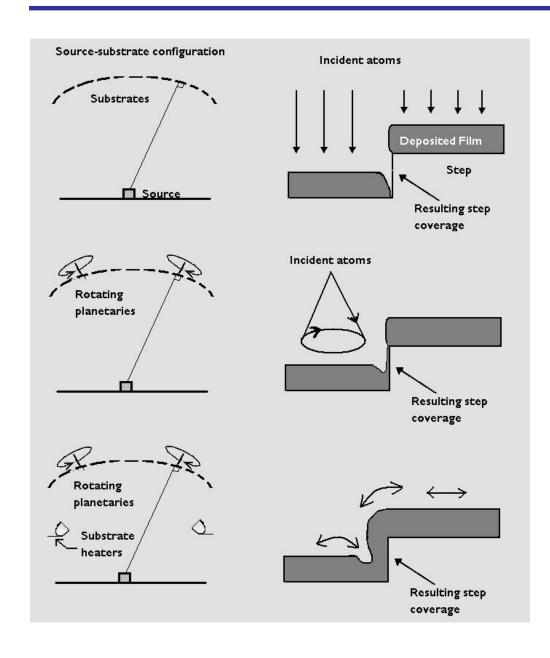
MBE: Molecular Beam Epitaxy



Ultrahigh Vacuum
High Substrate Temperature
Lattice Matched Substrate

High Quality, Single Crystal Films

Step Coverage



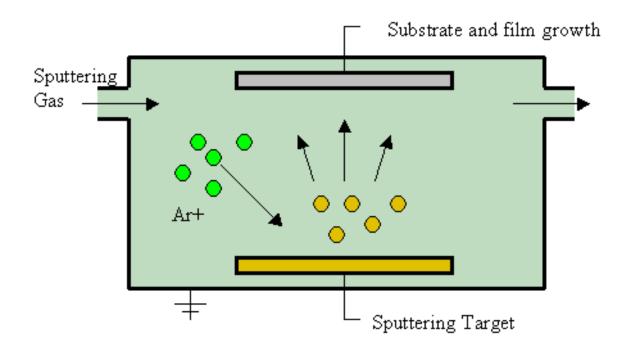
Substrate rotation and heating improve step coverage

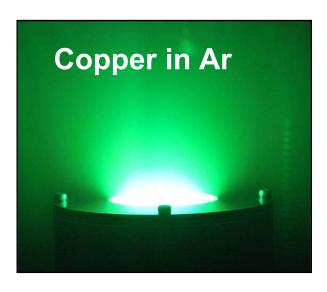
Challenges of Evaporation

- Materials with high melting points / low vapor pressure
 - □ W, Mo, SiO₂, ...
- Compounds and alloys (non-stoichiometry)
 - FeCoB alloy
 - \Box TiO₂ -> TiO_x
- Radiation damage generated by Ebeam
 - electron beam and X-ray radiation
- Poor step coverage
 - via filling

Sputter (溅射)

- Plasma (e.g. Ar) assisted transport
 - high energy
 - high deposition rate

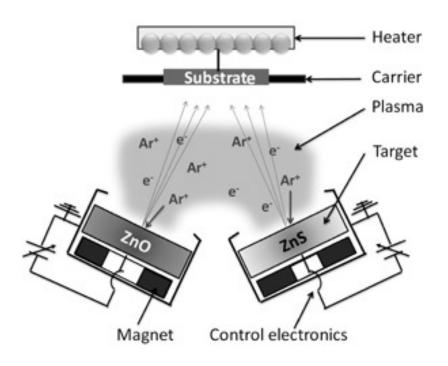




Plasma

Co-Sputter

- Deposit more than one material
 - composition control





Sputter: Pros. & Cons.

Advantages

- Higher pressure than evaporation
- Higher deposition rate
- Better uniformity and step coverage
- Better stoichiometry control
- Work for most materials

Disadvantages

- Plasma induced damages (etching)
- More impurities and defects
- Not good for single crystal epitaxy
- Mostly polycrystalline and amorphous films

Sputter

Process Parameters

- □ Type: DC, RF/AC, Magnetron, ...
- Substrate temperature
- \square Gas type (Ar, O₂, N₂, ...)
- Chamber pressure
- Sputter power
- **-**

Control Parameters

- Deposition rate
- Crystallinity
- Film quality (defects, ...)
- **---**

Sputter

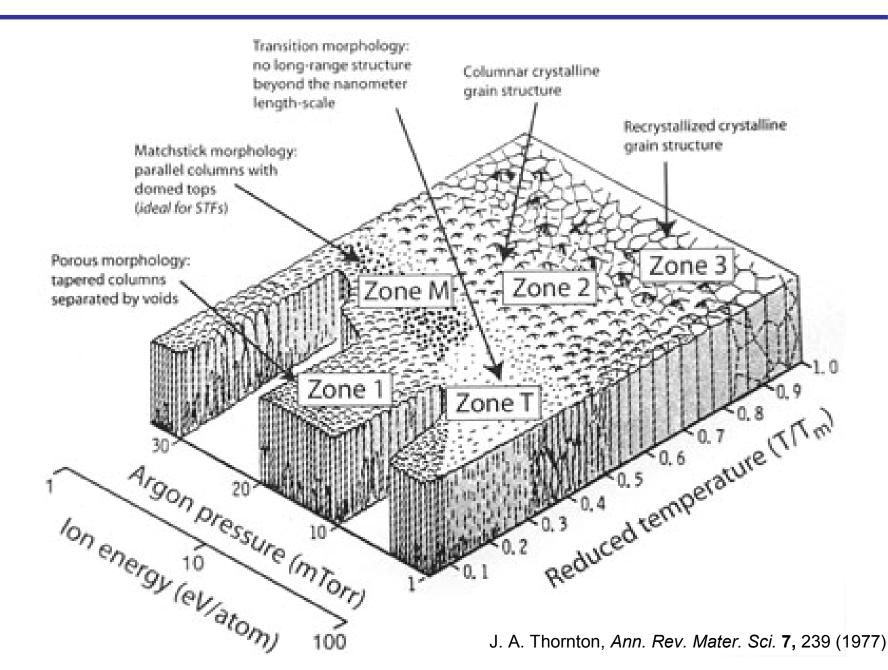
Sputtering Process Trend for typical metals and films

	Base Pressure	Sputtering Pressure	Power	Substrate RF Bias
Deposition Rate	\Leftrightarrow	Below~3mT Above~8mT	1	-
Stress (+ tensile, - compressive)	\Leftrightarrow	1	1	1
Step Coverage/ Sidewall coverage	\Leftrightarrow	***************************************	2 nd order effect depending on geometry	Can cause re-dep onto sidewalls thru collisions
Resistivity	1		2 nd order effect with substrate or target heating on some films	2 nd order effect with some films by changing density or stress

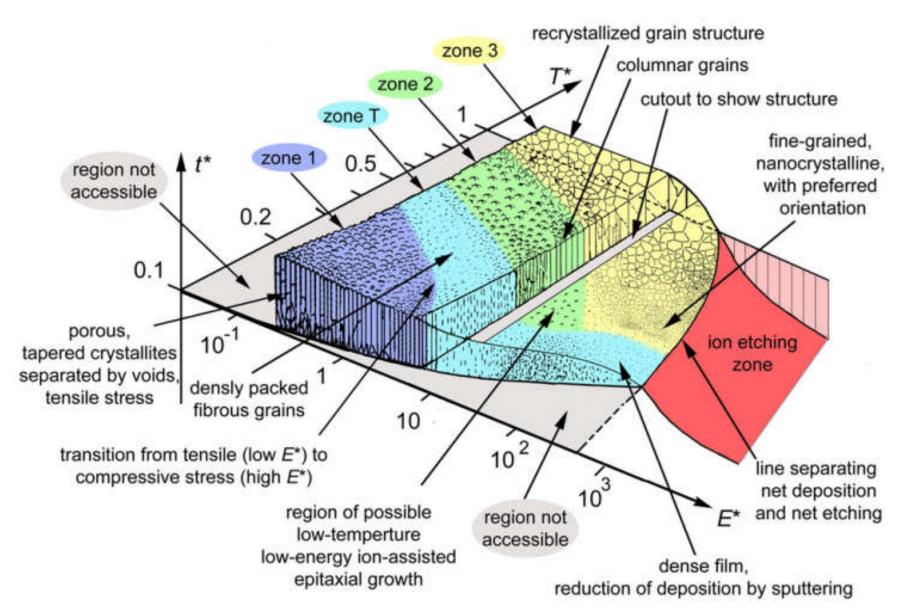
Legend



Thornton's Zone Model



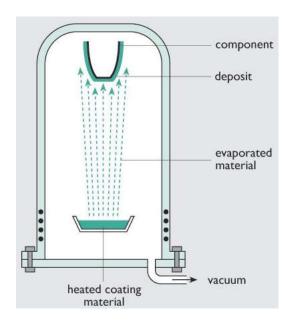
Refined Zone Model



Evaporation vs. Sputter

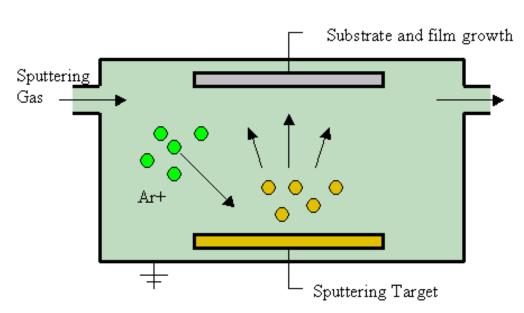
Evaporation:

- higher temperature
- radiation (Ebeam)
- lower pressure
- poor step coverage

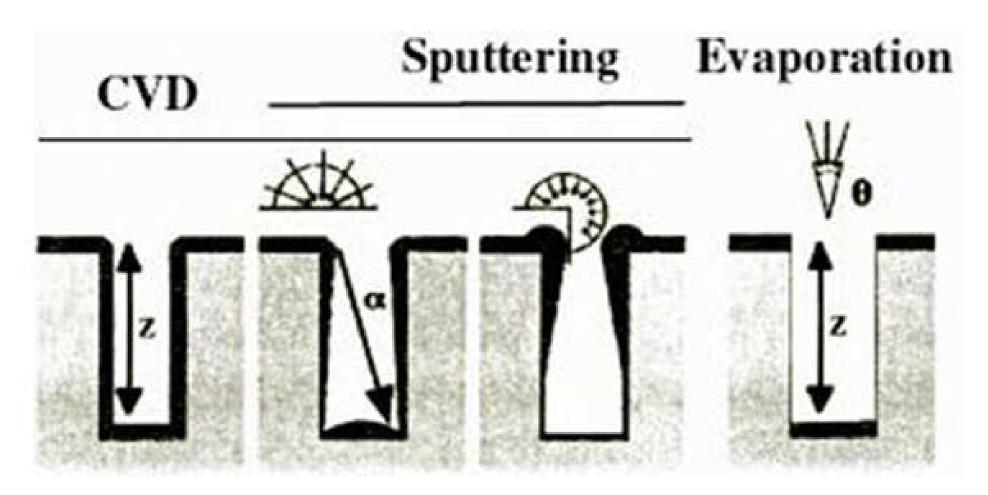


Sputter:

- lower temperature
- plasma damage
- higher pressure
- better step coverage



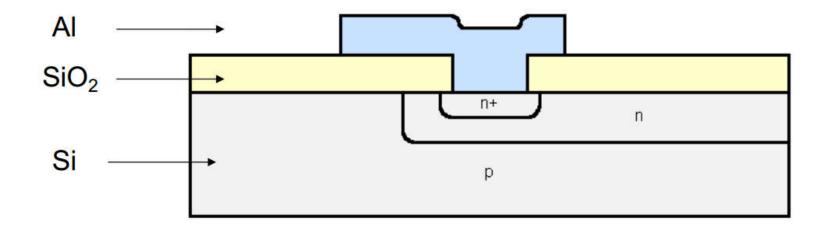
Step Coverage



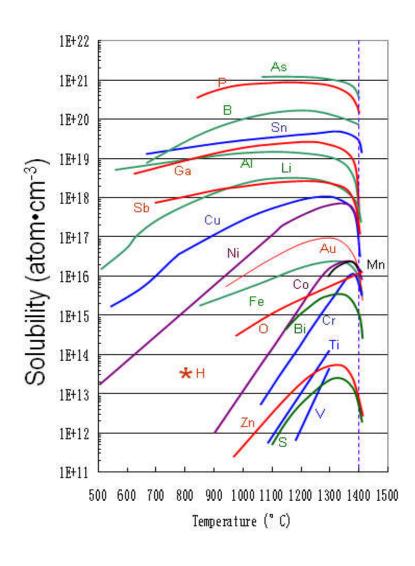
surface reaction

ballistic transport

- Al has good adhesion on Si and SiO₂
 - Al has high solubility in Si
 - \square Al + SiO₂ = Al₂O₃ + Si



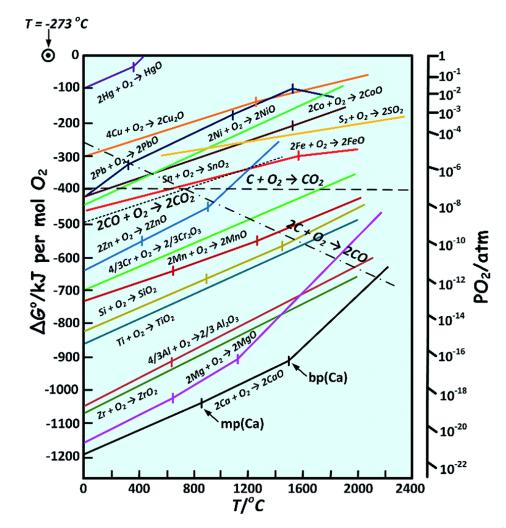
Q: How about Cu and Au?



solubility of metals in Si

Ellingham diagram

Formation of metal oxides



- Metals like Ag and Au tend to have poor adhesion on Si and SiO₂
- Substrate clean
- Deposit a thin (~5 nm) Ti or Cr layer for adhesion
- Plasma treatment
- Monolayer bonding





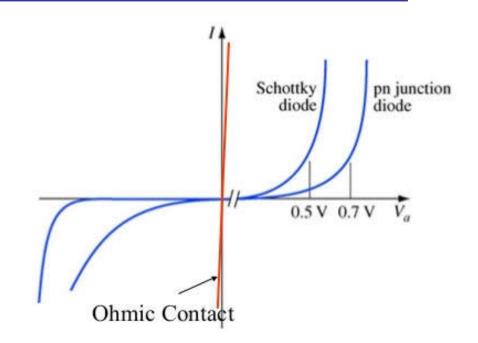
Typical Ohmic Contacts for III-V

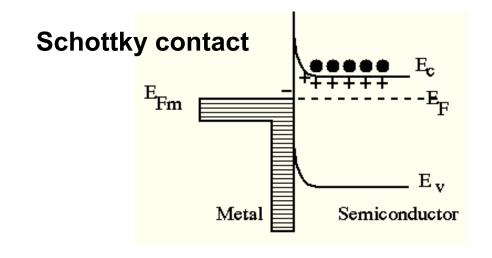
GaAs

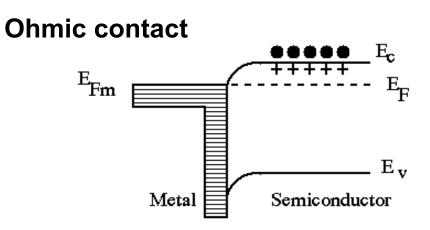
- □ p-GaAs Be/Au, Ti/Pt/Au, ...
- □ n-GaAs Ge/Ni/Au, Pd/Ge, ...

GaN

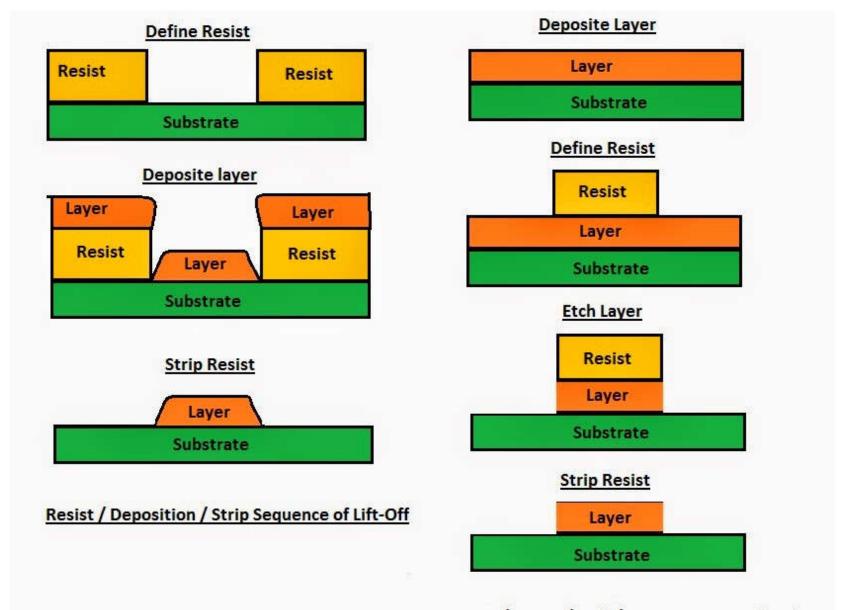
- □ p-GaN Ni/Au
- □ n-GaN Ti/Al/Au





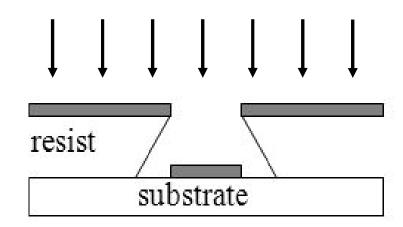


Thin Film Patterning

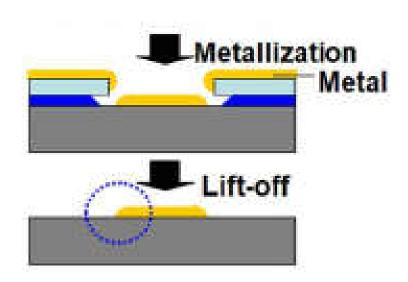


Requirements for Liftoff

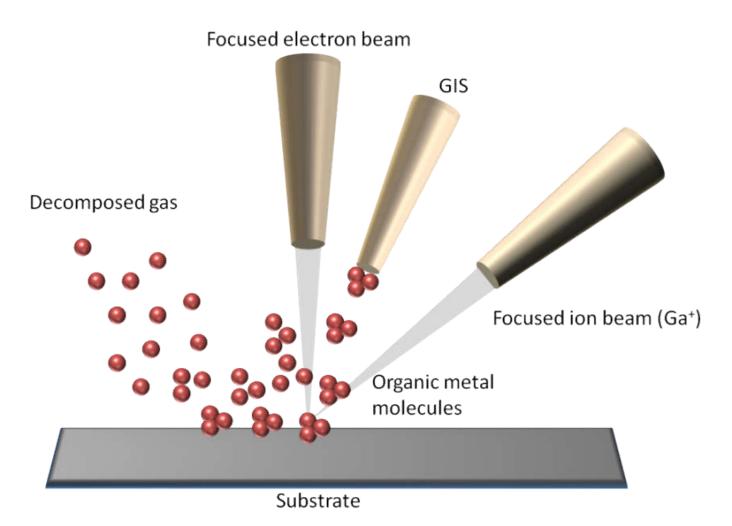
- Avoid Step Coverage
- PVD instead of CVD
 - avoid high temperature

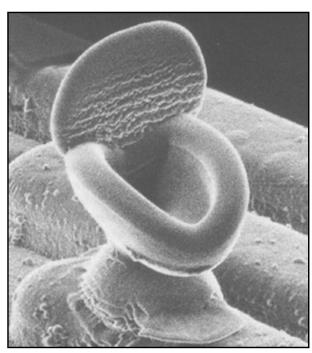


- Photoresist (PR) process
 - negative PR preferred (Why?)
 - increase PR thickness
 - multilayer PR



Focused Ion Beam (FIB) Deposition





world's smallest toilet

Etch: Ga

Deposition: Pt

Thank you for your attention