Microrobotics: challenges & applications

- Applications: environmental monitoring, biomedical applications (patient recovery time reduction, less medical complication, minimally non-invasive) $| \bullet$ Magnetics effects: ①Magnetic force between current carrying wires $\sim L^4$ biocompatible manner \oplus various motion)/Acoustic/Thermal(\oplus Large S/Vleads to rapid heat thermal response #soft microrobot actuation possible)/Light(#light-responsive materials e.g. shape changing polymers are often soft and biomimetic which is biocompatible #stimulator for nano-actuation due to molecular switches actuated by photonic excitation)/Chemical(\(\phi\)Onboard chemical fuel cells or reactions allow for self propulsion

 bubble formation gives high velocities

 Cytotoxic aspect makes biomedical applications impractical) Major concern reason: microrobot has to overcome adhesive forces for actuation, needs constant power supply ▶Onboard power is bad because power scales with volume and even when using principles that scale well(e.g. electrostatics) we can't provide enough power@Localization(visual/electromagnetic/magnetic/ultrasound) **®Biocompatibility**(Coatings-PVD/CVD/Electropolymerization:Stainless steel hard to coat, Gold. Titanium low weight and cheap, Polymer many choices, Carbon) **Functionalization**(Non-fouling to prevent absorption: Lubrication to improve movement; Antibacterial to prevent bacterial sticking) ▶ Etch microrobot to make it (nano)porous(SA→ED to coat) ▶ Coat microrobot with porous polymer or liposomes to absorb drugs
- Short term biocompatibility: Cytotoxicity tests(fluorescent dye to monitor cells death cultured on the material) + Tissue culture tests(materials implanted in vitro in a tissue) • Long term biocompatibility: Animal test ing + Clinical trials on human •In vivo: within plant/animal living body; •In vitro:glass tubes/petridish; • In silico:silicon-based computing devices
- •Bio-inspired robot: Engineered device that borrows concepts of bi ology but has some freedom in its implementation \oplus nature is experienced exact copy impossible Why: ©Same challenges from environmental constraints 2 robot/biology mutually benefit (biological research tools/new robot principles) ▶Locomotion: Crawl(inchworm), Walk(strider) Jump(locust), Fly(fruit flies), Swim(Jellyfish: contraction&recoil, Flagella:rotation) •Biomimetic robot: Engineered device that produces exactly the target biological systems

 nature offers good starting point

 risk of taking nature's examples out of context •Biological robot: Emulation of a biological system used to better understand the biological system itself \oplus allow testing of a hypothesis in impossible conditions on actual organism \ominus errors during biological-artificial system transfer causes false scientific claims
- Micromanipulation: Use vacuum or Electromagnetic Forces DS urface Tension(object trapping inside small fluidic film) @Electrostatic Forces @Var der Waals Forces(increased surface roughness)

2 Scaling effects for small devices

• Scaling laws: heat loss $\sim L^2$, heat generation $\sim L^3$; Capillary tubes: weight $\sim L^3$, surface tension $\sim L^1$;

Reynolds	Fourier		Biot	;	Mach
Inertia F. Viscous F.	Diffusion Storage Rate		onvection R onduction R		Velocity Speed of sounds
Weber	Bond		Foude]	
Inertia Surface Tension	Gravity Surface Tens	ion	Inertia Gravity		

- Mechanical effects: \mathbb{Q} Gravity $\sim L^3$ & Inertia negligible (Quick velocity changes; High Resonant frequencies devices vibrate faster) @Friction dominates 3 Van der Waals forces $\sim L^2$ (Adhesion dominates $\frac{F_{vdw}}{G} \sim L^{-1}$) 4 Classic Newtonian physics(> nm); Quantum effects(< nm)
- Fluidic effects: ①Laminar flow dominates: Re< 2000: Turbulent flow Re> 3000 @Hagen-Poiseuille law for volumetric flow through a capillary flow rate $\sim L^3$ indicates even small amount of aterial oclusion influences
- Thermal effects: ①Energy required to heat a volume $\sim L^3$ ② Heat transfer $\sim L^2$ 3 Thermal equilibrium time $\sim L^2$ instantaneously \triangleright Thermal Actuator used as Relay: upward/downward movement by heating bottom/upper beam \blacktriangleright Thermal Microgripper: $W_r = \int_0^Q U_r dq = \int F dx$
- Electronic effects: Electromagnetic forces dominate © Conductor resistance $\sim L^{-1}$ @ Parallel plate capacitors $C = \epsilon_0 \frac{A}{d} \sim L^1$; Assume constant charge density $\frac{Q}{A} \sim L^0 \Rightarrow \text{Voltage } V = \frac{dQ}{cdA} \sim L^1$ Electrostatic Force •Magnetization M(induced field): $B = \mu_0(H + M)$

 $F = \frac{Q^2}{2c_{\star}4} \sim L^2$ @Electromagnetic induction: voltage of conductor loop in- Φ Hard magnetic materials: once magnetized M is constant, independent side magnetic field $U = BA\omega sin(\omega t) \sim L^2$

- ullet Challenges: ullet Power: Magnetic (\oplus precise wireless control in a highly Porce of a magnet on a current carrying wire $\sim L^3$ Torque between two magnets $\sim L^3$ @Force between two magnets $\sim L^2$ @Force/weight required to lift an object against gravity $\sim L^{-1}$
 - Chemistry effects: higher efficiency with larger S/V ratio
 - Power Density: $@Weight/Electromagnetic \sim L^0.5$; Electrostatic/Fluidic $\sim L^{-1}$; Surface tension $\sim L^{-2.5}$

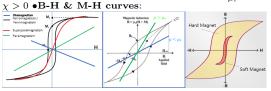
3 Electromagnetics (dominate microrobotics actuation(best candidate) & sensing

- •Fundamental Forces: Strong, Weak, Gravity, Electromagnetic
- •Subset-Electrostatics: Major function and failure mechanism of MEMS devices are dependent on electrostatic forces
- •Coulomb interactions: reduced in dielectric materials due to dielectric constant ϵ_r (due to polarization of particles of the dielectric medium - either induced or permanent dipoles around a free charge will be oriented to terminate some of the field lines coming from free charge)
- •Electric field & potential energy: $E = \frac{1}{d\pi\epsilon_0} \frac{Q}{r^2} \vec{r}, \vec{U} = \frac{Q}{4\pi\epsilon_0} \frac{1}{r}$
- •Piezoelectricity: interaction between mechanical stress and electrical charge distribution high voltage requirement short stroke length
- •Electromagnetic forces at atomic scale: Ionic bonds, Metallic bonds Covalent bonds, F_{VdW} , Hydrophobicity, Hydrogen bonds, Solvation forces •Van der Waals force: due to electron charge distributions of two atom
- s/molecules/surfaces; weak(\sim surface area & 1/distance)
- •Surface Tension: caused by cohesive forces within a liquid (attraction between molecules by various intermolecular forces)▶ describes interface force between liquid and vapor▶ hydrophobic/hydrophilic results in large/s mall contact angle and form a round/flat-spread-out dropplet ►Capillary effects with gravity(large Bond number): strong adhesive forces pull water up(concave); strong cohesive forces push mercury down(convex) ▶ Capillary effects without gravity(small Bond number): hydrophilic micro-channels fill $\vec{M} = \chi \vec{H_i}$, $\vec{H_i} = \vec{H} + \vec{H_d} = \vec{H} - N\vec{M} \Rightarrow \vec{M} = \chi [\mathbf{I} + \chi N]^{-1} \mathbf{H} = \chi_{\alpha} \vec{H}$ quickly; hydrophobic micro-channels hard to fill but empty easily

4 Magnetism

- Atomic Magnetism: protons and neutrons in nucleus. e⁻ in pairs with different spins cancel magnetic fields. Unpaired e^- can align with an external field and exhibit a magnetic field.
- Categoties: Diamagnetic(no unpaired e⁻, tiny attenuation of external field) $@Paramagnetic(small number of unpaired <math>e^-$, small intensification) \Im Ferromagnetic(large number of unpaired e^- , large intensification, some can retain magnetic field)
- Gauss's laws and Ampere's law: magnetic flux is divergence free and electrical fields are caused by electric charges; $\nabla \times \mathbf{H} = \mathbf{J} + \frac{\partial \mathbf{D}}{\partial t} = J$ (can be neglected when low f)
- Difference H and B: $B = \mu H$, A magnetic field H(A/m) gives rise to a magnetic induction or flux density B(T) in a medium with permeability $\mu = \mu_0 \mu_r (N/A^2)$. H does not depend on medium whereas B does. B is a measure of how a material reacts to a magnetic filed H. \blacktriangleright Diamagnetic ($\mu_r < 1$) Paramagnetic $(\mu_r = 1...10)$, Ferromagnetic $(\mu_r \gg 10)$
- Superparamagnetism: appears at different size depending on material behave like a paramagnetic but susceptibility is much larger(found in ferro magnetic below Curie T) • Paramagnetism: have magnetic susceptibility only under external field (found in paramagnetic and ferromagnetic at tem perature below Curie T) • Ferromagnetism: Dipoles tend to align at short range without any applied field to reduce exchange energy (Weiss domains) become paramagnetic when heated beyond Curie temperature(heating randomizes magnetic orientation and removes any remanence) > Soft magnetic materials: domain walls will again reorient at random orientations > Hard | • Low Reynolds Numbers: @Navier-Stokes law becomes time-independent magnetic materials: permanent magnet
- Hysteresis loop: @Microrobot control: high saturation; Precise: high co ercivity makes magnetic moment stable; Robust: linear(low remanence and coercivity), slow rising slope@Electromagnets: low remanence(min leftover magnetization); low coercivity(easily magnetized); low permeability(better

dent of H \otimes Soft magnetic materials: $M = \chi H(\text{linear region}), M =$ m_s (saturation region), susceptibility tensor $\chi = \frac{\mu}{\mu_s} - I$, DM: $\chi < 0$; PM,FM:



•Magnetic force: only depend on the gradient of H; F_{max} when magnetization is aligned with external field

•Magnetic torque: depend on the magnitude of H; T_{max} when ①Hard magnetic materials: H and M are perpendicular @Soft magnetic: depends.



• Force and Torque on Soft Magnets: Def - Magnetization will minimize total energy of the material (Exchange, Anisotropy, Zeeman energy). ► Anisotropy energies: align the magnetization vector in a given direction (Shape anisotropy: originates from interaction between dipoles in the material and tries to align the magnetization along the geometry of the body) ▶Shape anisotropy: ①M depends linearly on the field actually present inside the material $\mathbf{M} = \chi \mathbf{H}_i$ 2 Internal field is the sum of applied and demagnetizing field $\mathbf{H_i} = \mathbf{H} + \mathbf{H_d}$ ($H_d = -NM$) 3Demagnetizing tensor $N = diag(n_x, n_y, n_z), tr(N) = 1, smallest n_i represents easy axis(longest)$ because small demagnetizing field corresponds to larger internal field and hence larger magnetization • Calculation of M in soft magnetic bodies: where $\mathcal{X}_a = \operatorname{diag}\left(\frac{1}{n_x}, \frac{1}{n_y}, \frac{1}{n_z}\right)$ ©Soft magnetic body not free to move: \vec{M} will lie in between easy axis and applied field @free to move: \vec{M} will cause

torque to align easy axis and $\vec{H} \bullet \textbf{Example}$: Torque on Magnetic swimmers \mathbb{O} Permanent magnet: constant M in swimming direction (no driving τ except when perpendicular) 2 Soft magnet (linear): stable with shape change • Generating magnetic field: Biot-Savart Law: Along the axis of a circular coil 2Helmholtz coils(same current direction, H(0.5a) =0.7455Ni/a)(uniform filed) 3Maxwell coils(opposite current direction)(constant gradient) \oplus Electromagnets: long solenoids($L \gg D$) center: $|\mathbf{H}| = \frac{Ni}{L}$ end: $|\mathbf{H}| = \frac{Ni}{2L}$

•Magnetics and scaling: Orelative geometry and the magnitude of the field map around the magnet remain unchanged @Quantities depending on the field gradient increase: $T, F \sim L^3$ but $T/V, F/V \sim L^0$ \ominus Electrostatic, drag forces scale better \oplus Moving electromagnets closer compensates scaling loss because $H \sim d^{-3}$

•Benefits of magnetic interactions:

Permanent forces: permanent magnets provide constant magnetic field $\oplus Bistability$ suspensions: keep a system in a given configuration without energy consumption ⊕Remote state switching: external fields can magnetize soft magnetics parts of a MEMS to change its behavior \(\phi\)Long-range actuation: magnetic fields and gradients can be effective even over long distances relative to MEMS device and actuators with large motion possible

Contactless actuation: helps locomotion inside body

Safe: for human except at high frequencies

5 Liquid(Lamilar flow independent if object shape)

and reciprocal motion (ABCBABCBA) doesn't work at this regime 2A micro-swimmer must generate non-reciprocal(ABCDABCDA) motion in order to produce a net displacement(having a turning tail like flagella or cilia power stroke) •Propulsion matrix: ①Vertical balancing: Tune ω until ABF does not move out of focus($u = 0 \Rightarrow b = F_{ext}/\omega$, $F_{ext} =$ $F_{buoy} - F_{qrav}$) @Horizontal swimming: put ABF in horizontal direction and propulse forward with a rotating magnetic field, record speed u at different perature •Non-Newtonian Fluids: Stress depends non-linearly on strain or strain rate ①NN are viscoelastic if time-dependent and shear rate and shear strain are related to shear stress 2NN are inelastic if time-independent and Electron Microscopy (SEM) shear strain is non-linear of shear stress Common properties: relaxation: creep; effective stiffness is a function of strain rate; hysteresis; frictional resistance; acoustic attenuation •Non-Newtonian Biofluids: Blood(arteris: NF; capillaries: NNF); Vitreous body of eye(combination of several NNF) •Random Walks and Brownian Motion: diffusivity $D = kT/(6\pi R\mu)$; Def: thermally induced particle motion of particles within a gas or liquid due to particle-particle interactions

6 Observation tools

• Optical Microscope

Operating	Visible light (electromagnetic radiation in the visible spectrum)	
Principle	and a system of lenses is used to magnify images of small objects.	
Resolution	around 200nm (fluorescence 2-5 nm possible)	
Applications	everyday lab use, observe without altering samples, determine	
	protein concentration in living cells using fluorescence imaging	
Limitations	low resolution limited by diffraction, transparent samples diffi-	
	cult to image	

(1000x possible; depth of field(DOF: range of distance along the optical axis where specimen can move without losing image sharpness) and working distance decrease with higher magnification and NA @Nature of light: Diffraction(light rays bend around edges and generate new wavefronts, smaller aperture bigger diffraction); Dispersion(light seperation into its constituent wavelengths) **Resolution**: $d = 0.61\lambda/NA$, where numerical aperture $NA = nsin\alpha$ (light gathering capabilities of objective lens) \blacktriangleright Enhancement: smaller d higher resolution; Fluorescence microscopy: high energy beam UV excites molecules to emit visible light, different λ be separated with filters **4**Contrast:

Bright field (sample appears dark)	Dark field (sample appears bright)		
full aperture illuminated;	obstruction blocks central light cone;		
brightness differences poorly shown;	indirect illumination enhances contrast;		
light absorption areas appear dark.	light scattering areas appear bright.		

Saberration: ▶Spherical aberration: light waves passing through the edge of an uncorrected convex lens are not brought into focus with those pass through the center ▶Chromatic aberration: failure of lens to focus all colors to the same point due to different n for different λ , minimized by achromatic lens ▶Astigmatism: optical system not axisymmetric due to manufacturing errors like optical surface shape and component misalignment @Confocal Laser Scanning Microscope(CLS): Laser scans single plane point by point, pinhole blocks light reflected or emitted from others than focal plane, stage moves up and down, optical slices assembled to create 3D model

•Scanning Probe Microscopes: probe size limits resolution; Ad compared with OM, EM: resolution not limited by light or e^- diffraction

(I) Atomic Force Microscopy (AFM): probe with a sharp tipped cantilever

Operating Principle	A laser diode is used to detect the deflection of a cantilever probe moving in very close proximity to the surface	
Resolution	sub nanometres	
Application	Observe and manipulate nanometer-sized objects (surface), measure biological samples in wet	
Limitations	Slow scanning rate & area; sharp, vertical edges & overhangs can't be imaged; tip can damage sample	

Contact mode: use low stiffness cantilevers to boost deflection signal high speed \oplus rough samples \ominus sample damage **Tapping mode**: higher stiffness. oscillated near resonance f \oplus higher lateral resolution \oplus lessen sample damage #liquid environment #slower Non-contact mode: oscillated slightly above resonance $f \oplus no$ surface force \bigcirc lower resolution \bigcirc slowest

2 Magnetic Force Microscopy (MFM): magnetic features are hidden

- @Magnetic Resonance Imaging(MRI) @Magnetic Particle Imaging(MPI)
- OM \(\rightarrow\) Wave nature: energy $eV = 0.5mv^2$, $\lambda_e = h/(mv) = h/\sqrt{2meV} = 1$ ation \bigcirc contamination; Laser \oplus no radiation low contamination \bigcirc expensive) order

frequencies $(F=0) \Rightarrow u=-b\omega/a$, a can be obtained through extracting $12.3/\sqrt{V}$, electrons are charged particles accelerated in an electrostatic \triangleright Sputtering: Potential difference ionizes argon \rightarrow ions are accelerated \rightarrow hit the slope) \(\text{3}\) Vertical Free Fall: put in vertical direction and measure free-fall field whose trajectory can be deflected by electrostatic and magnetic target atoms which are released \(\text{-travel}\) to substrate \(\text{-form layers of atoms} \) velocity($u = (F_{ext} - b\omega)/a, \tau = 0 \Rightarrow c = -bu/\omega$) •Viscosity: resistance to fields **\rightharpoonup** •Resolution: $d = 0.6\lambda_e$, considering aberrations up to 0.2 nm oshadowing effect (non-conformal coating) **©Chemical Vapor Deposition** material flow(shear stress/shear rate) •Newtonian Fluids: viscosity inde- | Electron beam interaction with material: elastic electron-matter | (CVD): Mechanism: Diffusion of reactants(gas) -absorption of reactants pendent of shear rate and velocity, altered by material components and tem- interactions (incident electrons with energy pass through a sample are scat- on surface \rightarrow chemical reaction \rightarrow surface diffusion \rightarrow desorption of volatile surtered without energy transfer), inelastic electron-matter interactions(energy face reaction product ⊝slow ③Electrodeposition: ▶Mechanism: Chemical transfer from electron to specimen, causing various effects) (Scanning process in solution Process: anode & cathode are connected to an external

1	10 ()
Operating	electron beam scanned over conductive sample using electromag-
Principle	netic coils, scattered e^- are detected, each pixel represents signal strength of specific position on sample
Resolution	few nanometres
Application	image microstructures, surface topography, material composition
Limitations	samples(no wet, living cells) must be solid, high vacuum & con-
	ducting coating is necessary→alters sample, e ⁻ interaction with larger area limits resolution, charging effects
l	larger area minus resolution, charging enects

Detector: SE(sample surface), BSE(material contrast), In-lens(edge)

Transmissions Floatron Microscopy (TFM)

	Transmissions Electron wicroscopy (1EM)		
٦	Operating	e beam interacts with and eventually passes through an ultra-	
	Principle	thin sample and is magnified by using electromagnetic lenses.	
]		coils, scattered e^- are detected, each pixel represents signal	
ı		strength of specific position on sample	
ı	Resolution	few nanometres(better than SEM)	
٦	Application	Observing thin samples, crystallography	
J	Limitations	Long sample preparation time; resolution limited by e^- diffrac-	
- 1		tion: need high cacuum	

•Types of solid materials

Ionic bonds	one neutral atom gives up e^- (cation), one takes e^- up(anion), obtain full valence shell; bonded due to coulombic forces
Covalent	sharing of e^- pairs between non-metallic atoms; e^- delocalized
Metallic bonds	metallic atoms come together, e from outer shell share space and can move freely within atom orbitals, electrostatic attrac-
bolids	tion forces between delocalized e^- & metal ions

•Nanoscale materials and structures: ①Nanoparticle(< 100nm metal lic, metalloid, ceramic, organic, polymeric) $2Nanowires(\sim 10nm \text{ same ex.})$ 3Nanorods 4Nanotubes(organic or inorganic tubular structures with an inner diameter in nanometer range) ⑤Nanoporous materials ▶Carbon Al lotropic: Diamond, Graphite, Lonsdaleite, Fullerene C_{60,540,70}, Amorphous carbon, Carbon nanotube
Graphene: chemically most reactive form of car bon; high electrical and thermal conductivity; harder than diamond and 300 times harder than steel; electrons travel through graphene very fast at Fermi velocity 10⁶ (applications: graphics cards, touchscreens)

•Physical properties compared with macroscale materials ①Gravitational forces negligible and electromagnetic forces dominate 2Greater surface-to-volume ratios 3Random molecular motion more important @Quantum mechanics is used to describe motion and energy

•High surface to volume ratio in nano structure materials: large area for different coatings to functionalize different structures (nickel, hydrophilic coating) •Magnetic properties of nano structured materials: Super paramagnetism describes the effect of random flips of the magnetization direction in small ferromagnetic particles under the influence of temperature (magnetic saturation and susceptibility drastically decline) •Optical properties of.: the size of nanoparticle defines its color due to frequency of incident light and resonance frequency of the surface electrons

•Top-down: Motivated by manufacturing histroy

well-established method(thin film deposition, photolithography, etching) olimited resolution •Bottom-up: motivated by nature's way of growing things (nanomanipulation, self-assembly, chemical methods) olong range order difficult to achieve •Additive processes ①Physical Vapor Deposition (PVD): ▶Thermal evaporation: source material is heated until it evaporates. The evaporated material travels to the substrate where it condenses and is de-

supply of direct current →immersed in a solution called electrolyte →metal ions or conductive polymers reduce at cathode to solid atoms & form deposit ### | Harge-scale to small-scale ##fast cheap simple highly-tunable ### | high-aspect ratio structures possible

•Lithography:

Mechanism: UV light and a mask with desired pattern are used to expose the photoresist and transmit the pattern onto a wafer ▶Process: silicon coated with photoresis→prebake to evaporate solvent→align and expose \rightarrow develop(remove exposed resist) \rightarrow deposition(etch) \rightarrow Liftoff(resist removal) ▶ Photoresist: ①positive (light degrades polymers resulting in photoresist being more soluble in developers) 2negative(light polymerizes rubbers in photoresist to strengthen its resistance to dissolving in developers) ► Exposure: ©Contact printing ⊕ high resolution ⊝ mask & wafer easily damaged ②Proximity printing⊕long mask life ⊝low resolution(diffraction effect) ③Projection printing⊕high resolution ⊝complicated, expensive

•Substractive processes Wet Etching: using liquid etchants based on chemical reaction; etch rate depends on etchant concentration, temperature, crystal orientation, agitation @Isotropic: etch at the same rate in all directions \oplus fast \ominus undercuts mask \ominus limited diffusion @Anisotrophic: etch at different rates depending on orientation of exposed crystal plane ⊕not undercut mask ⊝slow ⊝limited reaction rate **2Dry Etching**: gas etchants physically knock off material @Sputter etching: physical by high energy bombardment 2Plasma: chemical reaction between gas, molecules. sample surface 3Reactive ion: combination of both Oslower 3Laser Micromachining: 500nm resolution, used to machine metals, plastics.composites.wafers.diamond

•3D Laser Lithography using Nanoscribe: ▶Process: CAD→laser writes structure onto photoresist resist developed to give the final structure (twophoton photopolymerization)

•Case study: Ocular Microrobot Fabrication: 1. Deposit sacrificial copper layer on silicon wafer using PVD 2. Apply negative photoresist mold 3. Electroplate (electrodeposition) nickel into mold 4. Strip photoresist 5. Etch sacrificial copper to release nickel parts 6. Planar parts are assembled to get 3D shape 7. Magnetization of robot

9 Nanofabrication

•Electron-beam Lithography: use electron beam to expose an electronsensitive resist(patterns directly written into resist by SEM; main way to fabricate nanostructures combined with lift-off electron scattering in resist and substrate limit resolution hack scattered and secondary electrons expose resist, resulting in beam spreading and reducing resolution •Extreme Ultraviolet Lithography:⊕extend minimum line without throughput loss ⊝EUV strongly absorbed in all materials, must performed in vacuum oneed special masks & mirrors expensive •X-ray Lithography: #large aspect ratios possible omask substrate and pattern must be thin ocomplex mask fabrication due to stress on thin mask •AFM-based (non-contact mode) exposure and Lithography: detailed imaging of substrate precise alignment between substrate and AFM tip •Dip-Pen Lithography(DPN): AFM tip is dipped into solution containing small concentration of molecules of interest which flow from tip onto surface enabled by water meniscus formed between tip and surface •Localized Electrochemical Deposition(LECD)(3D nanostructures): current for electrodeposition confined onto sharp tip, material only deposited near tip, by moving the tip, the electrolyte material can be grown along a given trajectory ⊕surface roughness modified ⊙time consuming ⊙only electrolytes materials •Focused-Ion-Beam etching/milling: similar to SEM but use heavy ions beams for etching instead of electrons •FIB-CVD: fabricate 3D nanostructures odeposition area is posited. (Resistance heating evaporation #no radiation @contamination; limited •Self-assembly: structural reversible self-organization of physical • Electron Microscopy: influenced by material properties compared with | Electron beam evaporation \(\pmathrm{1}\) blow contamination \(\pmathrm{1}\) radiation; RF \(\pmathrm{1}\) no radiation; without external guidance into stable \(\partial \) well-defined pattern of higher