

# Acetate: A New Feedstock for Biomanufacturing

## Key Takeaway

Acetate is emerging as a versatile, low-cost, and potentially carbon-negative platform feedstock that can decouple biomanufacturing from food-derived sugars while enabling high-yield production of amino acids, polymers, fuels, and specialty chemicals. Integration with CO<sub>2</sub>-to-acetate electrolysis and syngas acetogenesis positions acetate at the heart of hybrid electro-bio manufacturing systems, but realizing its promise demands advances in strain tolerance, energy/redox management, process intensification, economical separation, and robust supply chains.

## 1. Introduction and scope

Decarbonizing industrial chemistry and bioprocessing requires pivoting away from food-competing sugars to alternative carbon sources with lower cost, greater stability, and scalable availability. **Acetate** stands out among next-generation feedstocks (C1/C2 streams) due to its broad natural occurrence in waste valorization chains (anaerobic digestion, lignocellulosic hydrolysates), its efficient microbial uptake via the Pta–AckA and Acs routes, and its central position as a direct precursor to acetyl-CoA for widespread biosynthetic pathways <sup>[1] [2] [3] [4]</sup>. Electrochemical and gas fermentation routes now enable direct conversion of CO<sub>2</sub>/CO/H<sub>2</sub> into acetate at relevant rates, creating the prospect of circular carbon manufacturing where acetate serves as an electro-sourced, logistically friendly “liquid CO<sub>2</sub>” for precision fermentation <sup>[5] [6] [7]</sup>.

This review critically evaluates acetate as a new biomanufacturing feedstock, covering: (i) sustainable production routes and supply characteristics; (ii) cellular physiology and metabolic engineering for acetate assimilation; (iii) emerging applications across chemicals, fuels, and materials; (iv) techno-economics and life-cycle aspects; and (v) outstanding challenges and a forward-looking perspective.

## 2. Sustainable acetate supply: production routes, scale, and quality

### 2.1 Thermochemical–biological routes (syngas → acetate)

Thermochemical gasification of lignocellulose yields syngas (CO, CO<sub>2</sub>, H<sub>2</sub>) that feeds **acetogenic** bacteria via the Wood–Ljungdahl pathway to produce acetate. Classic acetogens such as *Acetobacterium woodii* and *Moorella thermoacetica* achieve multi–tens g/L acetate titers under optimized conditions, with reported values up to ~51 g/L for *A. woodii* and ~31 g/L for *M. thermoacetica*, depending on gas composition, reactor type, and pH control <sup>[5]</sup>. Immobilization and bubble-column reactors have been used to stabilize continuous production and mitigate washout, achieving productivities on the order of 1 g/L/day and beyond <sup>[8] [9]</sup>. Gas composition (CO/CO<sub>2</sub>/H<sub>2</sub> ratio) crucially sets the electron economy and the selectivity toward acetate versus more reduced products <sup>[5]</sup>.

### 2.2 Electrochemical CO<sub>2</sub>/CO reduction to acetate

Electrocatalysis has advanced rapidly, enabling tandem CO<sub>2</sub> → CO and CO → acetate systems with **>50% carbon selectivity** and single-pass CO<sub>2</sub>-to-acetate conversion near 25% using nanostructured copper catalysts, with integration compatible with biological fermentation <sup>[6]</sup>. Studies report improved selectivity via constrained Cu domains and Ag–Cu surfaces, achieving up to ~90% acetate selectivity at elevated CO partial pressure, highlighting a viable path to high-purity acetate streams for fermentation with reduced downstream treatment <sup>[9] [6]</sup>. In hybrid “electro-agriculture,” electrosynthesized acetate can replace photosynthesis to support growth of microbes, algae, and even plant tissues, with calculated solar-to-biomass conversion efficiencies far exceeding conventional agriculture <sup>[6]</sup>. Technoeconomic analyses suggest acetate from electrolysis, when powered by low-cost renewables, can decrease fermentation costs versus glucose by ~16% and improve price stability <sup>[10] [7]</sup>.

## 2.3 Lignocellulosic and waste-derived acetate

Pretreatment and hydrolysis of biomass liberate acetyl groups from hemicellulose, generating acetate in hydrolysates (often in the 5–10 g/L range), while pyrolysis liquids can contain 5–17% acetate equivalents; food waste anaerobic digestion also yields acetate as a key intermediate [3]. Because these streams can be inhibitory and heterogeneous, detoxification and purification are often required, yet they enable co-location with waste management and biorefinery infrastructure [1] [3].

### Figure 1. Acetate production rates across methods

[11]

## 3. Physiology and metabolic engineering of acetate assimilation

### 3.1 Acetate uptake and central carbon rerouting

In model hosts, acetate is activated to acetyl-CoA via two routes: the reversible Pta–AckA pathway and ATP-dependent Acs. Growth on acetate demands anaplerotic carbon conservation through the glyoxylate shunt (AceA/GlcB) and careful partitioning of carbon between the TCA cycle (energy generation) and gluconeogenesis (precursors) [12] [13] [14]. Systems-biology analyses reveal that acetate is not merely a toxic overflow byproduct; at low glycolytic flux, acetate becomes a co-substrate that buffers total carbon uptake and can enhance growth, with dynamic switching between production and consumption around a ~10 mM threshold in *E. coli* [15] [14] [16].

### 3.2 Addressing overflow metabolism and toxicity

Industrial *E. coli* processes suffer from acetate overflow under aerobic glucose excess, reducing yields and inhibiting growth. Comparative engineering strategies to minimize overflow include: tuning flux through Pta/AckA/PoxB, increasing TCA and glyoxylate capacity (e.g., *iclR* deletion), and limiting sugar uptake; these interventions improve robustness to mixing and feed gradients and facilitate acetate reassimilation after perturbations [17] [12] [15]. Cross-scale physiology remains challenging; even small mismatches between feed rate and cellular demand can trigger runaway acetate accumulation in large tanks, arguing for strains pre-adapted to high-acetate regimes and for advanced feed and mixing control [12].

### 3.3 Energy and redox constraints relative to sugars

Relative to glucose, acetate provides less intrinsic reducing power, placing a premium on ATP and NAD(P)H management during acetate-based growth and production. Reviews emphasize energy/redox balancing, cofactor supply, and pathway coupling as central bottlenecks when broadening the acetate product spectrum, with design levers including optimized glyoxylate flux, alternative electron donors, and oxygenation strategies [18] [19]. Hybrid processes that decouple carbon supply (acetate) from energy/electron supply (H<sub>2</sub>, electricity) are promising but introduce complexity in reactor integration and safety [5] [20].

### 3.4 Host selection and adaptive evolution

Beyond *E. coli*, emerging hosts are being adapted for acetate valorization. *Vibrio natriegens*, with extremely fast growth, has been evolved and engineered for efficient acetate uptake and conversion to PHB, achieving ~0.27 g/L/h productivity and ~46% PHB of cell mass in fed-batch, underscoring the potential of next-generation chassis for acetate bioprocesses [21]. *Corynebacterium glutamicum* is being explored for upgrading acetate to proteins and recombinant products at industrially relevant titers, leveraging its robustness and secretion capacity [22].

## 4. What can be made from acetate? Scope, titers, and selectivities

Acetate-to-product landscapes are expanding rapidly across classes of chemicals and materials, aided by modular pathway design and host-specific strengths. The following selection highlights quantitative state-of-the-art achievements drawn from recent literature.

### 4.1 Amino acids and proteinogenic metabolites

Using *E. coli* W hosts, **homoserine** and **threonine** production from acetate have reached 44.1 g/L and 45.8 g/L, respectively, via modular engineering of acetate assimilation (Acs and Pta/AckA), TCA/glyoxylate tuning, and CoA biosynthesis enhancement; reported yields are up to ~65% of theoretical for threonine in fed-batch [23]. These levels demonstrate that amino acid fermentations can be competitive on acetate feedstock when energy/redox coupling is optimized.

### 4.2 Polymers and materials

Multiple organisms accumulate polyhydroxyalkanoates (PHAs) from acetate. A fast-growing *Vibrio* platform reached PHB content of ~45.7% of cell dry weight with >0.25 g/L/h productivity after adaptive evolution and pathway rewiring, highlighting a route to scalable, low-cost bioplastics from acetate [21]. Broader PHA production from acetate has been shown in *Pseudomonas*, *Synechocystis*, *Aeromonas*, and phototrophs, and acetate co-feeding can outperform glucose in some strains [2].

### 4.3 Platform solvents and diols

Acetone can be produced as a primary product from acetate using a hybrid acetone pathway in *E. coli*, with ~113 mM accumulation aided by gas stripping [24]. Recent *E. coli* work has produced mixed diols (2,3-butanediol and acetoin) at ~1.56 g/L on acetate via optimization of acetate activation and malate-to-pyruvate conversion, with gains reproduced in fed-batch [25].

### 4.4 Biofuels and higher alcohols

Although current acetate-to-isobutanol titers remain modest (~0.16 g/L), pathway designs coupling acetate assimilation, anaplerosis, and engineered pyruvate supply continue to improve, and further evolution/energy supplementation strategies may close the gap with sugar-based systems [3]. Ethanol and other fuels are accessible in native acetogens and mixed cultures but require careful electron balancing to avoid over-reduction away from acetate [5].

### 4.5 Food, feed, and cellular agriculture

Hybrid inorganic–biological systems demonstrate that acetate from CO<sub>2</sub> electrolysis can support the growth of yeast, algae, fungi, and even plant calli in the dark, with energy conversion efficiencies significantly exceeding agricultural photosynthesis. Diverse crop tissues can incorporate <sup>13</sup>C-acetate into central metabolism and biomass, suggesting a long-term path to electro-enabled vertical agriculture and novel food systems [6].

## Table 1. Selected bioproducts from acetate feedstock

[26]

## 5. Process engineering, separations, and integration

### 5.1 Bioreactor strategies for acetate supply and uptake

Acetate's dual role as product inhibitor and co-substrate necessitates tailored feeding profiles, pH control to manage undissociated acetic acid, and dynamic dissolved oxygen strategies. For acetogenic acetate generation, gas–liquid mass transfer and biofilm/immobilization techniques are pivotal for stable continuous operation [8] [9]. For electrosynthesis, designing electrolyte compositions and gas-diffusion electrodes that deliver acetate in low-salt effluents reduces cost and complexity of downstream microbial integration [6] [7].

## 5.2 Downstream processing and cost drivers

Acetic acid recovery from dilute aqueous media is energy intensive. Technoeconomic analyses indicate that solvent choice (e.g., tertiary amines versus ethyl acetate) and steam/electricity integration strongly influence minimum selling prices (roughly USD ~750–900 per ton in modeled cases), motivating co-location, heat integration, and hybrid extraction strategies [27]. For acetate-as-feedstock scenarios, impurity control (salts, organics, metals) and neutralization steps must be balanced against conductivity needs in electrolysis and bioreactor compatibility in fermentation [6] [7].

## 5.3 System-level integration: electro-bio “acetate hubs”

The most promising near-term architectures couple renewable electricity to CO<sub>2</sub>/CO electrolysis for on-site acetate generation, which then feeds dedicated fermentation units for amino acids, polymers, and specialty chemicals. Tandem electrolysis with high acetate selectivity, paired with acetate-tolerant, energy-optimized production strains, can deliver carbon-negative or carbon-minimal footprints, particularly when powered by low-carbon grids and integrated with waste heat and water management [28] [9] [7].

### Figure 2. Integrated acetate biorefinery concept

[29]

## 6. Controversies, constraints, and knowledge gaps

Despite success stories, several core debates and bottlenecks shape the field's trajectory.

1. Toxicity versus co-substrate role. Acetate has been long treated as a toxic byproduct to be minimized; new kinetic models and experiments show condition-dependent benefits to growth and robustness when glycolytic flux is constrained. Translating these insights to 100–500 m<sup>3</sup> reactors under variable gradients remains an unresolved scale-up problem [15] [14] [16].
2. Energy/redox economics. Relative to glucose, acetate's lower reducing equivalent content necessitates either higher oxygen demand or supplemental electron sources. Comparing aerobic and anaerobic C<sub>1</sub>/C<sub>2</sub> routes across products reveals trade-offs between ATP yields, CO<sub>2</sub> emissions, and product selectivities; rigorous, product-specific yield ceilings and process metrics are still emerging [30] [31] [18].
3. Electrosynthesis maturity and purity. While tandem CO<sub>2</sub>/CO systems can reach high acetate selectivity, stable long-term operation, electrolyte minimization, and impurity management at scale require further advances before routine direct feeding to microbes without polishing [9] [6] [7].
4. Downstream energy intensity. Separations for acetic acid/acetate remain a major cost and carbon sink; advances in membranes, extractants, reactive separations, and in situ product removal integrated with bioreactors are needed to lower MSPs and open new process windows [27].
5. Host chassis and product scope. Most acetate bioproduction has focused on a narrow set of hosts and products. Next-generation fast growers (e.g., *Vibrio* spp.), acetate-resilient bacteria (*Corynebacterium*, *Pseudomonas*), and eukaryotes (yeasts, microalgae) need standardized tools and pipelines for broadening the acetate product portfolio [21] [22] [3].

## 7. Technoeconomics and sustainability

Analyses suggest that substituting electrochemical acetate for glucose in fermentation can reduce production costs by ~16% and improve price stability, especially as renewable power prices decline. Life-cycle assessments of gas fermentation to solvents (e.g., acetone, isopropanol) indicate the potential for **carbon-negative** production by fixing waste CO/CO<sub>2</sub>, provided high selectivity and continuous operation are maintained [28] [19] [7]. However, for acetic acid as product, energy-intensive separations dominate costs and emissions unless integrated energy systems or co-location strategies are used [27]. Market pull exists for bio-based acetates and downstream esters (e.g., ethyl acetate), with growing demand in coatings, adhesives, and pharma, but price parity hinges on process intensification and scale [32] [33] [34].

## 8. Outlook: where acetate biomanufacturing is heading (5–10 years)

A credible path exists to position acetate as a mainstream feedstock in precision fermentation and biorefining:

- Hybrid electro-bio nodes. Onsite CO<sub>2</sub> → acetate via tandem electrolysis will be increasingly integrated with fermentation suites to produce amino acids (threonine, homoserine), PHAs, and specialty monomers, prioritizing high acetate selectivity at low electrolyte strength and microbes engineered for high acetate uptake and tolerance [6] [7].
- Chassis diversification and adaptive evolution. Fast-growing and robust hosts (*Vibrio natriegens*, *Corynebacterium glutamicum*, *Pseudomonas putida*) will be evolved for acetate formatotrophy/methylotrophy coupling and for efficient acetyl-CoA channeling to products, enabled by long-read genome editing, multiplex ALE, and systems-guided regulation of the glyoxylate/TCA balance [21] [22] [3] [15].
- Energy/redox co-supply. Processes will increasingly decouple carbon (acetate) from electrons/ATP via H<sub>2</sub> co-feeding, bioelectrochemical augmentation, or controlled oxygenation, tuned to product stoichiometry and redox demand, with dynamic control to avoid over-reduction or CO<sub>2</sub> losses [5] [35] [20].
- Process intensification and separations. Expect progress in membrane-assisted extraction, reactive distillation, and in situ product removal for acetic acid and acetate esters, alongside continuous fermentation and immobilized catalysts to boost space–time yields and energy efficiency [8] [27].
- Electro-agriculture pilots. Dark, acetate-fed cultivation of yeasts, algae, and engineered plants will expand for protein, oils, and specialty carbohydrates, catalyzed by high solar-to-biomass conversion, urban siting, and grid flexibility services [36] [10] [6].

If these advances converge, acetate-fed manufacturing could materially reduce land use, enable carbon-negative chemical production, and provide a robust anchor for renewable-electricity-coupled bioprocessing.

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1. <https://academic.oup.com/femsle/article-abstract/365/20/fny226/5101431>
2. <https://pmc.ncbi.nlm.nih.gov/articles/PMC7699770/>
3. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10891810/>
4. <https://www.frontiersin.org/articles/10.3389/frmbi.2024.1441290/full>
5. <https://elifesciences.org/articles/63661>
6. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10548627/>
7. <https://academic.oup.com/jimb/advance-article-pdf/doi/10.1093/jimb/kuad025/51331438/kuad025.pdf>
8. <https://pubmed.ncbi.nlm.nih.gov/28648471/>
9. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10891810/>
10. <https://pmc.ncbi.nlm.nih.gov/articles/PMC9319612/>
11. <https://news.biobuzz.io/2025/07/16/a-biomanufacturing-bottleneck-threatens-to-blunt-biotechs-boom/>
12. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10899681/>
13. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10143712/>
14. <https://pmc.ncbi.nlm.nih.gov/articles/PMC6999358/>

15. <https://elifesciences.org/articles/63661>
16. <http://downloads.hindawi.com/journals/bmri/2010/761042.pdf>
17. <https://www.frontiersin.org/articles/10.3389/fbioe.2024.1339054/full>
18. <http://arxiv.org/pdf/2205.03920.pdf>
19. <https://www.sciencedirect.com/science/article/pii/S2667370322000273>
20. <https://pmc.ncbi.nlm.nih.gov/articles/PMC9727924/>
21. <http://biorxiv.org/lookup/doi/10.1101/2025.06.23.661028>
22. <https://sfamjournals.onlinelibrary.wiley.com/doi/full/10.1111/1751-7915.14138>
23. <https://pubmed.ncbi.nlm.nih.gov/38796054/>
24. <https://microbialcellfactories.biomedcentral.com/articles/10.1186/s12934-019-1054-8>
25. <https://www.embopress.org/doi/10.15252/embj.2022113079>
26. <https://www.marketsanddata.com/industry-reports/bio-based-ethyl-acetate-market>
27. <https://pmc.ncbi.nlm.nih.gov/articles/PMC11269247/>
28. <https://www.nature.com/articles/s41587-021-01195-w>
29. <https://www.imarcgroup.com/insight/ethyl-acetate-cost-model>
30. <https://www.sciencedirect.com/science/article/abs/pii/S0960852422013979>
31. <https://www.frontiersin.org/articles/10.3389/fbioe.2024.1339054/pdf?isPublishedV2=False>
32. <https://www.sciencedirect.com/science/article/abs/pii/S0960852422013979>
33. <https://pmc.ncbi.nlm.nih.gov/articles/PMC11341323/>
34. <https://pmc.ncbi.nlm.nih.gov/articles/PMC11054211/>
35. <https://www.chemistryworld.com/news/less-is-more-for-copper-catalyst-when-it-comes-to-synthesising-acetate-from-co/4017391.article>
36. <https://source.washu.edu/2024/10/how-to-grow-food-without-light/>
37. <https://iopscience.iop.org/article/10.1149/MA2024-02593924mtgabs>
38. <https://www.frontiersin.org/article/10.3389/fbioe.2019.00035/full>
39. <https://www.nature.com/articles/s43016-022-00530-x>
40. <https://pmc.ncbi.nlm.nih.gov/articles/PMC12471985/>
41. <https://academic.oup.com/femsre/article/doi/10.1093/femsre/fuaf011/8104276>
42. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10390867/>
43. <https://www.sciencedirect.com/science/article/pii/S0958166925000540>
44. <https://www.frontiersin.org/journals/bioengineering-and-biotechnology/articles/10.3389/fbioe.2020.00833/full>
45. <https://pmc.ncbi.nlm.nih.gov/articles/PMC7396591/>
46. <https://enviromicro-journals.onlinelibrary.wiley.com/doi/10.1111/1751-7915.70063>
47. <https://pubmed.ncbi.nlm.nih.gov/38399713/>
48. <https://www.sciencedirect.com/science/article/pii/S2667370322000273>
49. <https://pmc.ncbi.nlm.nih.gov/articles/PMC9035589/>
50. <https://www.frontiersin.org/journals/microbiology/articles/10.3389/fmicb.2022.865168/full>
51. <https://arpa-e.energy.gov/programs-and-initiatives/search-all-projects/acetate-platform-carbon-negative-production-renewable-fuels-and-chemicals>
52. <https://www.sciencedirect.com/science/article/abs/pii/S0960852422015516>
53. <https://www.sciencedirect.com/science/article/pii/S1385894725034679>
54. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10776380/>
55. <https://www.biorxiv.org/content/10.1101/2025.06.17.659982v1.full-text>
56. <https://pmc.ncbi.nlm.nih.gov/articles/PMC11225617/>
57. <https://sfamjournals.onlinelibrary.wiley.com/doi/10.1111/1751-7915.70088>

58. <https://pmc.ncbi.nlm.nih.gov/articles/PMC5288458/>
59. <https://pmc.ncbi.nlm.nih.gov/articles/PMC11334397/>
60. <https://onlinelibrary.wiley.com/doi/pdfdirect/10.1049/enb2.12030>
61. <https://arxiv.org/pdf/2310.09991.pdf>
62. <https://www.mdpi.com/2306-5354/10/6/744>
63. <https://www.mdpi.com/1420-3049/29/8/1878/pdf?version=1713582391>
64. <https://www.mdpi.com/1422-0067/21/22/8777/pdf>
65. <https://www.frontiersin.org/articles/10.3389/fbioe.2024.1339054/pdf?isPublishedV2=False>
66. <https://pmc.ncbi.nlm.nih.gov/articles/PMC11103475/>
67. <https://sfamjournals.onlinelibrary.wiley.com/doi/pdfdirect/10.1111/1751-7915.12796>
68. <https://pmc.ncbi.nlm.nih.gov/articles/PMC11610983/>
69. <https://pmc.ncbi.nlm.nih.gov/articles/PMC8451798/>
70. <https://pmc.ncbi.nlm.nih.gov/articles/PMC7658081/>
71. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10899681/>
72. <https://pubmed.ncbi.nlm.nih.gov/39628702/>
73. <https://pmc.ncbi.nlm.nih.gov/articles/PMC7571159/>
74. <https://pmc.ncbi.nlm.nih.gov/articles/PMC9686113/>
75. <https://pubmed.ncbi.nlm.nih.gov/33036784/>
76. <https://pubmed.ncbi.nlm.nih.gov/37205870/>
77. <https://www.sciencedirect.com/science/article/pii/S2405844024068610>
78. <https://www.openpr.com/news/4210730/acetate-salt-market-to-reach-usd-1-6-billion-by-2035-driven>
79. <https://academic.oup.com/plphys/article-abstract/194/1/5/7237768>
80. <https://arpa-e.energy.gov/programs-and-initiatives/search-all-projects/acetate-platform-carbon-negative-production-renewable-fuels-and-chemicals>
81. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10948735/>
82. <https://www.sciencedirect.com/science/article/abs/pii/S0167779920302419>
83. <https://www.sciencedirect.com/science/article/pii/S2212982025001945>
84. <https://www.mdpi.com/2311-5637/10/6/285/pdf?version=1716889737>
85. <https://www.linkedin.com/pulse/north-america-acetate-salt-market-demand-drivers-key-challenges-tucnf>
86. <https://www.linkedin.com/pulse/global-bio-based-ethyl-acetate-market-segmentation-nydqf>
87. <https://pmc.ncbi.nlm.nih.gov/articles/PMC11610983/>
88. <https://pmc.ncbi.nlm.nih.gov/articles/PMC5502627/>
89. <https://pmc.ncbi.nlm.nih.gov/articles/PMC10549214/>
90. <https://pmc.ncbi.nlm.nih.gov/articles/PMC7658081/>
91. <https://iopscience.iop.org/article/10.1149/MA2024-02282136mtgabs>
92. <https://pmc.ncbi.nlm.nih.gov/articles/PMC7584085/>
93. <https://www.pnas.org/doi/10.1073/pnas.2201330119>
94. <https://journals.asm.org/doi/10.1128/AEM.02959-20>
95. <https://www.mdpi.com/2306-5354/10/12/1357>
96. <https://www.sciencedirect.com/science/article/pii/S0263876225001340>
97. <https://scholars.uky.edu/en/projects/acetate-as-a-platform-for-carbon-negative-production-of-renewable>
98. <https://arpa-e.energy.gov/programs-and-initiatives/search-all-projects/acetate-platform-carbon-negative-production-renewable-fuels-and-chemicals>
99. <https://www.sciencedirect.com/science/article/abs/pii/S0167779920302419>
100. <https://pubs.acs.org/doi/10.1021/acssuschemeng.4c08968>

101. <https://pmc.ncbi.nlm.nih.gov/articles/PMC11727576/>
102. <https://www.frontiersin.org/journals/bioengineering-and-biotechnology/articles/10.3389/fbioe.2024.1339054/full>
103. <https://www.frontiersin.org/articles/10.3389/fbioe.2022.874612/full>
104. <https://source.washu.edu/2024/06/alterd-carbon-points-toward-sustainable-manufacturing/>
105. <https://link.springer.com/10.1007/s11306-022-01912-9>
106. <http://bmcbiotechnol.biomedcentral.com/articles/10.1186/s12896-016-0284-7>
107. <https://www.semanticscholar.org/paper/271bedcac149981f4f293c4808e78791c86a9beb>
108. <https://www.semanticscholar.org/paper/b7320d8e31334d6467ad5bbcf332855d34747e3>
109. <https://pubs.acs.org/doi/10.1021/jf500355p>
110. <https://www.nature.com/articles/s41467-023-38072-w>
111. [http://link.springer.com/10.1007/978-3-642-02481-8\\_163](http://link.springer.com/10.1007/978-3-642-02481-8_163)
112. <http://microbialcellfactories.biomedcentral.com/articles/10.1186/s12934-016-0577-5>
113. <https://microbialcellfactories.biomedcentral.com/articles/10.1186/1475-2859-11-79>
114. <https://pmc.ncbi.nlm.nih.gov/articles/PMC12012870/>
115. <https://pmc.ncbi.nlm.nih.gov/articles/PMC8864926/>
116. <https://pmc.ncbi.nlm.nih.gov/articles/PMC8021400/>
117. <https://pmc.ncbi.nlm.nih.gov/articles/PMC6330746/>
118. <https://pubs.rsc.org/en/content/articlepdf/2020/gc/d0gc02395g>
119. <https://pmc.ncbi.nlm.nih.gov/articles/PMC5348906/>
120. <https://pmc.ncbi.nlm.nih.gov/articles/PMC7137189/>
121. <http://biorxiv.org/lookup/doi/10.1101/2025.06.28.662091>
122. <https://www.sciencedirect.com/science/article/pii/S0167779924001525>
123. <https://aiche.onlinelibrary.wiley.com/doi/full/10.1002/aic.18684>
124. <https://pubs.acs.org/doi/10.1021/acssynbio.4c00839>
125. <https://www.sciencedirect.com/science/article/abs/pii/S0921344922002397>
126. <https://www.frontiersin.org/journals/microbiology/articles/10.3389/fmicb.2016.00694/full>
127. <https://pubmed.ncbi.nlm.nih.gov/40103233/>
128. <https://netl.doe.gov/node/12678>
129. <https://www.frontiersin.org/journals/bioengineering-and-biotechnology/articles/10.3389/fbioe.2024.1339054/full>
130. <https://research-hub.nrel.gov/en/publications/syngas-mediated-microbial-electrosynthesis-for-co2-to-acetate-con>
131. <https://www.pnas.org/doi/10.1073/pnas.0337684100>
132. <https://arpa-e.energy.gov/programs-and-initiatives/search-all-projects/acetate-platform-carbon-negative-production-renewable-fuels-and-chemicals>
133. <https://www.sciencedirect.com/science/article/pii/S2667370322000273>
134. <https://journals.asm.org/doi/10.1128/msystems.00221-18>
135. <https://analyticalsciencejournals.onlinelibrary.wiley.com/doi/10.1002/pmic.201600303>
136. <https://onlinelibrary.wiley.com/doi/10.1002/sus2.117>
137. <https://www.mdpi.com/2304-8158/14/11/2004>
138. <https://www.mdpi.com/2079-4991/12/22/3959>
139. <https://onlinelibrary.wiley.com/doi/10.1111/gcb.16925>
140. <https://pubs.acs.org/doi/10.1021/acs.accounts.3c00098>
141. <https://biotechnologyforbiofuels.biomedcentral.com/articles/10.1186/s13068-024-02547-9>
142. <https://www.mdpi.com/2073-4441/16/12/1645>
143. <https://advanced.onlinelibrary.wiley.com/doi/10.1002/aenm.202302647>
144. <https://chemistry-europe.onlinelibrary.wiley.com/doi/10.1002/slct.202405152>



145. <https://journals.asm.org/doi/10.1128/mbio.01976-25>
146. <https://www.frontiersin.org/articles/10.3389/fbioe.2020.00833/pdf>
147. <https://pmc.ncbi.nlm.nih.gov/articles/PMC4491173/>
148. <https://pubs.rsc.org/en/content/articlepdf/2021/fd/d0fd00132e>
149. <https://pmc.ncbi.nlm.nih.gov/articles/PMC8788776/>
150. <https://pmc.ncbi.nlm.nih.gov/articles/PMC3514249/>
151. <https://www.mdpi.com/2306-5354/10/12/1357/pdf?version=1700992968>