

Task A51 240979

$$m = 60 \text{ kg}$$

$$\rho = 1000 \text{ kg/m}^3 \quad / \quad c = 4200 \text{ J/kg-K}$$

$$1) \quad \frac{dQ}{dt} = \underbrace{\epsilon}_{\substack{\text{ideal given} \\ \downarrow}} \underbrace{\sigma}_{\substack{\text{Stefan's const} \\ \rightarrow}} A T^4 \quad (2.728)^4$$

$$\cancel{A} \quad \text{Vol} = \frac{m}{\rho} = \frac{60}{1000} = 0.06 \text{ m}^3 = \frac{4}{3} \pi r^3$$

$$\Rightarrow r = 0.242859 \text{ m}$$

$$A = 4\pi r^2 = 0.741170783 \text{ m}^2$$

$$\Rightarrow \frac{dQ}{dt} = (1) (5.67 \times 10^{-8}) (A) (2.728)^4 \\ = 2.8274 \times 10^{-6} \text{ Watt}$$

$$2) \text{ avg. microwave } \lambda \text{ be } 0.01 \text{ m.}$$

$$\begin{aligned} \text{no. of photon per second} &= \frac{\left(\frac{dQ}{dt}\right) \lambda}{hc} \\ &= \frac{(2.8274 \times 10^{-6}) (0.01)}{(6.626 \times 10^{-34}) (3 \times 10^8)} \\ &= 0.11708 \times 10^{18} = [1.1708 \times 10^{17}] \end{aligned}$$



3) Energy given in time (t)

$$\left(\frac{dq}{dt}\right) t = m C \Delta T$$

$$t = \frac{(60)(4200)(1)}{2.3274 \times 10^6}$$

$$= 108.27 \times 10^9$$

$$= 1.0827 \times 10^{11} \text{ sec}$$

4+5)