## Vivekananda College of Engineering & Technology, Puttur [A Unit of Vivekananda Vidyavardhaka Sangha Puttur ®] Affiliated to VTU, Belagavi & Approved by AICTE New Delhi



CRM08

Rev 1.15(2022 scheme)

<FY>

<21/06/2024>

## CONTINUOUS INTERNAL EVALUATION - 2

Dept: FY(Chemistry)	Sem / Div: II/CS A and B	Sub: Chemistry for computer science and allied branches	S Co BCF		202	
Date:26/06/24	Time: 3:00-4:30	Max Marks: 50	Elec	Elective:N		
Note: Answer any 2 full questions, choosing one full question from each part.						
QN	Questions			Marks <sup>RBT</sup> CO's		
PART A						
l a Discuss the working principle of electrochemical gas 10 L2 CO1 sensors for the detection of SO <sub>x</sub> and No <sub>x</sub> .						
b Explain the steps involved in extraction of gold from e-waste.				8	L2 CO5	
c Explain the detection of pharmaceutical diclofenac using electrochemical sensor.				7	L2 CO1	
OR						
2 a Discuss the optical sense	•	ing and applications	of	10	L2 CO1	
b Define a battery. Explain the construction and working of Li- ion battery.				8	L2 CO1	
c Explain acid using el	the detection ectrochemical sens			7	L2 CO1	
PART B						
3 a Explain the classification of electronic memory devices with examples.				10	L2 CO2	
b What are nano materials? Explain any properties and applications of poly vinyl carbazole (PVK) suitable for				8	L1/ CO2 L2	

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optoelectronic devices. c Explain the types of organic memory devices by taking L2 CO2 p-type and n-type semiconducting materials. OR 4 a Explain the classification of liquid crystals. 10 L2 CO1 b What is OLED? Mention any four properties and 8 L1/COI applications of OLED. L2 c Explain the role of producers and consumers in e-waste L2 CO5 management.

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