

This Material is downloaded from <u>Education God</u>. You can get latest news, updates, reading material for upcoming exams without paying a single Rupee. Visit <u>Education God</u> to download More Such Material.

# Courses

- Engineering
- Law
- MBA
- MCA
- Medical
- Pharmacy
- Polytechnic

# **Study Material**

- Books
- Mock Tests
- Model Papers
- Previous Papers
- Sample Papers
- Tips and Tricks

Visit EducationGod.Com for further information.

# BITSAT

# **Solved Paper 2011**

#### Instructions

1. There are 150 questions in all. The number of questions in each part is as follows

Subjects	No. of Questions
Part I (Physics)	1-40
Part II (Chemistry)	41–80
Part III	1.05.22
(a) English Proficiency	81–95
(b) Logical Reasoning	96-105
Part IV (Mathematics)	106-150

- 2. All questions are multiple choice questions with four options, only one being correct.
- 3. Each correct answer fetches 3 marks while incorrect answer fetches -1 mark.

# Part I

### **Physics**

- 1. Suppose the gravitational force varies inversely as the nth power of distance. Then the time period of a planet in circular orbit of radius R around the sun will be proportional
  - (a)  $R^{(n+1)/2}$
- (b)  $R^{(n-1)/2}$
- (c) Rn
- (d)  $R^{(n-2)/2}$
- 2. Two wires are made of the same material and have the same volume. However wire 1 has cross-sectional area A and wire 2 has cross-sectional area 3A. If length of wire 1 increased by  $\Delta x$  on applying force F, how much force is needed to stretch wire 2 by the same amount?
  - (a) 4F
- (b) 6F
- (c) 9F
- (d) F
- 3. The satellite of mass m revolving in a circular orbit of radius r around the earth has kinetic energy E. Then its angular momentum will be

- (a)  $\sqrt{\frac{E}{mr^2}}$  (b)  $\frac{E}{2mr^2}$
- (c)  $\sqrt{2Emr^2}$
- 4. A galvanometer of resistance 100Ω gives full scale deflection with 0.01 A current. How much resistance should be connected in parallel to convert it into an ammeter of range 10 A?
  - (a) 0.100 Ω
- (b) 1.00 Ω
- (c) 10.00 Ω
- (d) 100.00Ω
- 5. A car is moving on a circular road of diameter 50 m with a speed of 5 m/s. It is suddenly accelerated at a rate of 1 m/s2. If the mass of the car is 500 kg, then the net force acting on the car is
  - (a) 5 N
- (b) 1000 N
- (c) 500√2 N

- 6. Hard X-rays for the study of fractures in bones should have a minimum wavelength of 10-11 m. The accelerating voltage for electrons in X-ray machine should be
  - (a)  $< 124 \, kV$
  - (b)  $> 124 \, kV$
  - (c) between 60 kV and 70 kV
  - $(d) = 100 \, kV$
- 7. Natural length of a spring is 60 cm and its spring constant is 4000 N/m. A mass of 20 kg is hung from it. The extension produced in the spring is (Take  $g = 9.8 \text{ m/s}^2$ )
  - (a) 4.9 cm
- (b) 0.49 cm
- (c) 9.4 cm
- (d) 0.94 cm
- 8. A point source of light is placed 4 m below the surface of water of refractive index  $\frac{5}{3}$ .

The minimum diameter of a disc, which should be placed over the source, on the surface of water to cut-off all right coming out of water is

- (a) infinite
- (b) 6 m
- (c) 4 m
- (d) 3 m
- 9. What is the maximum acceleration of the particle doing the SHM?

$$y = 2 \sin \left[ \frac{\pi t}{2} + \phi \right]$$
, where 2 is in cm

- (a)  $\frac{\pi}{2}$  cm/s<sup>2</sup> (b)  $\frac{\pi^2}{2}$  cm/s<sup>2</sup>

- (a) 10 days
- (b) 20 days
- (c) 40 days
- (d) None of these
- 13. The velocity of efflux of a liquid through an orific in the bottom of the tank does not depend upon
  - (a) size of orific
  - (b) height of liquid
  - (c) acceleration due to gravity
  - (d) density of liquid
- A neutron with velocity v strikes a stationary deuterium atom, its KE changes by a factor
  - (a)  $\frac{15}{16}$
- (c)  $\frac{2}{1}$
- (d) None of these
- 15. The Poisson's ratio of a material is 0.5. If a force is applied to a wire of this material, there is a decrease in the cross-sectional area by 4%. The percentage increase in the length is
  - (a) 1%
- (b) 2%
- (c) 2.5%
- (d) 4%
- 16. Lenz's law of electromagnetic induction corresponds to the
  - (a) law of conservation of charge
  - (b) law of conservation of energy
  - (c) law of conservation of momentum
  - (d) law of conservation of angular

- 19. The force constant of a spring gun is 50 N/m. If a ball of 20 g be shoot by the gun so, that its spring is compressed by 10 cm, the velocity of the ball is
  - (a) 5 m/s
- (b) 15 m/s
- (c) 25 m/s
- (d) 20 m/s
- 20. 1 g of water (volume 1 cm3) becomes 1671 cm3 of steam when boiled at a pressure of 1 atm. The latent heat of vapourisation is 540 cal/g, then the external work done is
  - $(1 \text{ atm} = 1.013 \times 10^5 \text{ N/m}^2)$
  - (a) 499.7 J
- (b) 40.3 J
- (c) 169.2 J
- (d) 128.57 J
- 21. A cube has a side of length  $1.2 \times 10^{-2}$  m. Calculate its volume.
  - (a)  $1.7 \times 10^{-6} \text{ m}^3$
- (b)  $1.73 \times 10^{-6} \text{ m}^3$
- (c)  $1.70 \times 10^{-6} \text{ m}^3$
- (d)  $1.732 \times 10^{-6} \text{ m}^3$
- 22. A ball is dropped from height h and another from 2h. The ratio of time taken by the two balls to reach the ground is
  - (a) 1:√2
- (b) √2:1
- (c) 2:1
- (d) 1:2
- 23. The linear momentum p of a body moving in one dimension varies with time t according to the equation  $p = a + bt^2$ , where a and b are positive constant. The net force acting on the body is
  - (a) a constant
  - (b) proportional to t2
  - (c) inversely proportional to t
  - (d) proportional to t
- 24. Which of the following is not an example of perfectly inelastic collision?
  - (a) A bullet fired into a block, if bullet gets embedded into block
  - (b) Capture of an electron by an atom
  - (c) A man jumping onto a moving boat
  - (d) A ball bearing striking another ball bearing
- 25. If a new planet is discovered rotating around sun with the orbital radius double that of the earth, then what will be its time period? (in earth's days)
  - (a) 1032
- (b) 1023
- (c) 1024
- (d) 1043

- 26. If density of earth increases 4 times and its radius becomes half of what it is, our weight
  - (a) be 4 times its present value
  - (b) be doubled
  - (c) remain same
  - (d) be halved
- 27. The magnitude of electric field intensity E, such that an electron placed in it would experience an electrical force equal to its weight, is given by
  - (a) mge
- (c)  $\frac{e}{mg}$
- (b)  $\frac{mg}{e}$ (d)  $\frac{e^2}{m^2}g$
- 28. The work done in placing a charge of 8 × 10<sup>-18</sup> C on a capacitor of capacity 100 μF

  - (a)  $32 \times 10^{-32} \text{ J}$  (b)  $16 \times 10^{-32} \text{ J}$
  - (c)  $3.1 \times 10^{-26}$  J (d)  $4 \times 10^{-10}$  J
- 29. A steady current flow in a metallic conductor non-uniform cross-section. quantity/quantities remaining constant along the whole length of the conductor is/are
  - (a) current, electric field and drift speed
  - (b) drift speed only
  - (c) current and drift speed
  - (d) current only
- 30. A galvanometer of 50Ω resistance has 25 divisions. A current of 4 × 10<sup>-4</sup>. A gives a deflection of one division. To convert this galvanometer into a voltmeter having a range of 25 V, it should be connected with a resistance of
  - (a)  $2500 \Omega$  as a shunt
  - (b)  $2950\Omega$  as in shunt
  - (c)  $2550\Omega$  in series
  - (d)  $2450\Omega$  in series
- 31. The cyclotron frequency of an electron gyrating in a magnetic field of 1 T is approximately
  - (a) 28 MHz
  - (b) 280 MHz
  - (c) 2.8 GHz
  - (d) 28 GHz

- 32. If M is magnetic moment and B is the magnetic field, then the torque is given by
  - (a) M · B
- (c) M × B
- (d) | M | B |
- 33. A coil of inductance L is carrying a steady current I what is the nature of its stored energy?
  - (a) Magnetic
  - (b) Electrical
  - (c) Both magnetic and electrical
  - (d) Heat
- 34. Energy conversion in a photoelectric cell takes place from
  - (a) chemical to electrical
  - (b) magnetic to electrical
  - (c) optical to electrical
  - (d) mechanical to electrical
- 35. If the ionisation potential of helium atom is 24.6 V, the energy required to ionise it will be
  - (a) 24.6 eV
- (b) 24.6 V
- (c) 13.6 V
- (d) 13.6 V
- 36. Fast neutrons can easily be solved down by
  - (a) the use of lead shielding
  - (b) passing them through water
  - (c) elastic collision with heavy nuclei
  - (d) applying a strong electric field

- 37. A film projector magnifies a 100 cm2 film strip on a screen. If the linear magnification is 4, the area of the magnified film on the screen is
  - (a) 1600 cm<sup>2</sup>
- (b) 400 cm<sup>2</sup>
- (c) 800 cm<sup>2</sup>
- (d) 6400 cm<sup>2</sup>
- **38.** If  $v_m$  is the speed of sound in moist air and  $v_d$ is the speed of sound in dry air under identical conditions of pressure and temperature, then
  - (a)  $v_m > v_d$
- (b)  $v_m < v_d$
- (c)  $v_m = v_d$
- (d)  $v_m \cdot v_d = 1$
- 39. A hot and a cold body are kept in vacuum separated from each other. Which of the following cause decrease in temperature of the hot body?
  - (a) Radiation
  - (b) Convection
  - (c) Conduction
  - (d) Temperature remains unchanged
- 40. An ideal refrigerator has a freezer at a temperature of -13°C. The coefficient of performance of the engine is 5. The temperature of the air (to which heat is rejected) will be
  - (a) 325°C
- (b) 325 K
- (c) 39°C
- (d) 320°C

# Part II

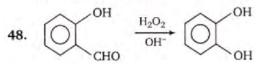
## Chemistry

- 41. The mutual heat of neutralisation of 40 g NaOH and 60 g CH3COOH will be
  - (a) 57.1 kJ
  - (b) less than 57.1 kJ
  - (c) more than 57.1 kJ
  - (d) 13.7 kJ
- 42. Which has the smallest size?
  - (a) Al3+
- (b) Mg2+
- (c) P5+
- (d) Na+
- 43. The treatment of benzene with iso-butene in the presence of sulphuric acid gives

  - (a) iso-butylbenzene (b) tert-butylbenzene
  - (c) n-butylbenzene (d) no reaction

- 44. Toluene on reaction with N-bromosuccinimide gives
  - (a) p-bromomethylbenzene
  - (b) o-bromomethylbenzene
  - (c) phenyl bromomethane
  - (d) m-bromomethylbenzene
- 45. Pinacolone is
  - (a) 2, 3-dimethyl-2, 3-butanediol
  - (b) 3, 3-dimethyl-2-butanone
  - (c) 1-phenyl-2-propanone
  - (d) 1, 1-diphenyl-1, 2-ethandiol
- 46. A synthetic rubber which is resistant to the action of oils, gasoline and other solvents is
  - (a) buna-S
- (b) polyisoprene
- (c) neoprene
- (d) polystyrene

- 47. Ozone depletion over Antarctica is due to the
  - (a) formation of chlorine nitrate (ClONO2)
  - (b) formation of HCl
  - (c) formation of HOCl and Cl<sub>2</sub> which are converted back into reactive Cl atoms
  - (d) None of the above



This reaction is called

- (a) Reimer-Tiemann reaction
- (b) Liebermann's nitroso reaction
- (c) Dakin reaction
- (d) Leader-Manase reaction
- 49. Which anion is the weakest base?
  - (a) C2H5O
- (b) NO<sub>3</sub>
- (c) F
- (d) CH<sub>3</sub>COO
- 50. K<sub>b</sub> for water is 0.52 K/m. Then 0.1 m solution of NaCl will boil approximately at
  - (a) 100.52°C
- (b) 100.052°C
- (c) 101.04°C
- (d) 100.104°C
- 51. One mole of P2O5 undergoes hydrolysis as

$$P_2O_5 + H_2O \longrightarrow H_3PO_4$$

The normality of the phosphoric acid formed is (The volume of solution is 1 L.)

- (a) 2
- (b) 12
- (c) 24
- (d) 4
- 52. 1 L of a gas is at a pressure of 10<sup>-6</sup> mm of Hg at 25°C. How many molecules are present in the vessel?
  - (a)  $3.2 \times 10^6$
- (b)  $3.2 \times 10^{13}$
- (c)  $3.2 \times 10^{10}$
- (d)  $3 \times 10^4$
- 53. Which of the following has the largest de-Broglie wavelength, given that all have equal velocity?
  - (a) CO2 molecule
- (b) NH3 molecule
- (c) Electron
- (d) Proton
- 54. 1 g of U-235 is converted into UF<sub>6</sub>. The radioactivity of UF<sub>6</sub> thus obtained is
  - (a) zero
  - (b) less than that of 1 g of U-235
  - (c) more than that of 1 g of U-235
  - (d) same as that of 1 g of U-235

- 55. In which of the following molecules S atom does not assume sp<sup>3</sup> hybridisation?
  - (a) SO<sub>4</sub><sup>2</sup>-
- (b) SF4
- (c) SF<sub>2</sub>
- (d) S<sub>8</sub>
- 56. For the reaction,

$$N_2(g) + 3H_2(g) \Longrightarrow 2NH_3(g)$$

the units of K are

- (a) L mol-1
- (b) L2 mol-2
- (c) mol L-1
- (d) No units
- A sulphuric acid solution has pH = 3. Its normality is
  - (a) 1/1000
- (b) 1/200
- (c) 1/2000
- (d) 1/100
- The oxidation number of N and Cl in NOClO<sub>4</sub> respectively are
  - (a) + 2 and + 7
- (b) + 3 and + 7
- (c) -3 and +5
- (d) + 2 and -7
- 59. Pyrolusite is a/an
  - (a) oxide ore
- (b) sulphide ore
- (c) carbide ore
- (d) Not an ore
- When potassium ferrocyanide crystals are heated with conc. H<sub>2</sub>SO<sub>4</sub>, the gas evolved is
  - (a) SO<sub>2</sub>
- (b) NH<sub>3</sub>
- (c) CO<sub>2</sub>
- (d) CO
- 61. The product/s of the reaction,

$$Na_2CO_3 + CO_2 + H_2O \longrightarrow is/are$$

- (a) 2NaOH + CO<sub>2</sub>
- (b)  $Na_2CO_3 + H_2CO_3$
- (c) 2NaHCO<sub>3</sub>
- (d) None of these
- 62. Which among the following is likely to show geometrical isomerism?
  - (a) CH<sub>3</sub>CH= NOH
  - (b)  $CH_3CH = CH_2$
  - (c)  $CH_2 = CH CH = CCl_2$
  - (d)  $CH_3C(Cl) = C(CH_3)_2$
- 63. A fuel has the same knocking property as a mixture of 70% iso-octane (2, 2, 4-trimethylpentane) and 30% n-heptane by volume. The octane number of the fuel is
  - (a) 100
- (b) 70
- (c) 50
- (d) 40
- 64. Sodium carbonate reacts with SO<sub>2</sub> in aqueous medium to give
  - (a) NaHSO<sub>3</sub>
- (b) Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>
- (c) NaHSO<sub>4</sub>
- (d)  $Na_2SO_4$

- **65.** For a given reaction  $t_{1/2} = 1/ka$ . The order of this reaction is
  - · (a) 0
- (b) 1
- (c) 2
- (d) 3
- 66. Which of the following compounds will react with two moles of CH3MgBr?
  - (a) C<sub>2</sub>H<sub>5</sub>COOH
- (b) CH3COOH
- (c)  $CH_3C \equiv CCH_3$
- (d)  $HC \equiv C CH_2OH$
- 67. The number of polypeptide chains present in a molecule of haemoglobin is
  - (a) four
- (b) one
- (c) two
- (d) three
- 68. The pentose sugar in DNA and RNA has the
  - (a) open chain structure
  - (b) pyranose structure
  - (c) furanose structure
  - (d) All of the above
- 69. Which of the following is an artificial edible colour?
  - (a) Saffron
- (b) Carotene
- (c) Tetrazine
- (d) Melamine
- 70. The number of unpaired electrons in nickel carbonyl is
  - (a) zero
- (b) one
- (c) four
- (d) five
- 71. The time taken for 90% of a first order reaction to complete is approximately
  - (a) 1.1 times that of half-life
  - (b) 2.2 times that of half-life
  - (c) 3.3 times that of half-life
  - (d) 4.4 times that of half-life
- 72. The pH of a 0.01 M HCN solution for which  $pK_a$  is 4 is
  - (a) 0.47
- (b) 1.2
- (c) 3.0
- (d) 4.0
- Which of the following does not contain any coordinate bond?
  - (a) H<sub>3</sub>O<sup>+</sup>
- (b) BF<sub>4</sub>
- (c) HF<sub>2</sub>
- (d) NH<sub>4</sub>
- 74. If E is the energy of the combining atomic orbitals, E1 and E2 are the energies of the bonding and anti-bonding molecular orbitals formed, then

- (a)  $E E_1 > E_2 E$
- (b)  $E E_1 < E_2 E$
- (c)  $E E_1 = E_2 E$
- (d) Any one of the above is possible
- 75. The equilibrium constant (K) for the reaction  $Cu(s) + 2Ag^{+}(aq) \rightarrow Cu^{2+}(aq) + 2Ag(s)$

will be [Given,  $E_{cell}^{\circ} = 0.46 \text{ V}$ ]

- (a)  $K_c = \text{Antilog } 15.6$  (b)  $K_c = \text{Antilog } 2.5$  (c)  $K_c = \text{Antilog } 1.5$  (d)  $K_c = \text{Antilog } 12.2$

- 76.  $E^{\circ}$  for Fe/Fe<sup>2+</sup> is +0.44 V and  $E^{\circ}$  for Cu/Cu<sup>2+</sup> is - 0.32 V. Then, in the cell
  - (a) Cu oxidises Fe2+ ion
  - (b) Cu2+ oxidises iron
  - (c) Cu reduces Fe2+ ion
  - (d) Cu2+ ion reduces Fe
- 77. Which of the following carbon atoms is most electronegative?

$$CH_3$$
— $CH_2$ — $C\equiv CH$ 

- (a) I
- (b) II
- (c) III
- (d) All are equally electronegative
- 78. The reaction/method that does not give an alkane is
  - (a) catalytic hydrogenation of alkenes
  - (b) hydrolysis of alkylmagnesium bromide
  - (c) Kolbe's electrolytic method
  - (d) dehydrohalogenation of an alkyl halide
- 79. Which of the following will yield a mixture of 2-chlorobutene and 3-chlorobutene on treatment with HCl?
  - (a)  $CH_2 = C = CH CH_3$
  - (b)  $H_2C = C CH = CH_2$ CH<sub>3</sub>
  - (c) CH<sub>2</sub>=CH-CH=CH<sub>2</sub>
  - (d) HC≡C—CH=CH<sub>2</sub>
- 80. The well known urinary urotropine is formed when formaldehyde reacts with
  - (a) NH2OH
- (b) NH<sub>3</sub>
- (c) NH<sub>2</sub>·NH<sub>2</sub>
- (d) C<sub>6</sub>H<sub>5</sub>NH · NH<sub>2</sub>

96. Which one number is wrong in the given series?

5, 10, 17, 24, 37

- (a) 10
- (b) 17
- (c) 24
- (d) 37
- 97. Find the next two letters in the given series.

EFHKO?

- (a) T, Z
- (b) Z, T
- (c) S, Z
- (d) T, Y
- 98. If MONKEY is coded as NNOJFX, what will be the code for TARGET?
  - (a) ZUSFFS
- (b) SFFSZU
- (c) UZSFSF
- (d) UZSFFS
- 99. Among six friends L, M, N, P, Q and S, each having a different height, N is shorter than Q and P but taller than M. S is shorter than only L. Which of the following represents the tallest among six friends?
  - (a) P
  - (b) Q
  - (c) L
  - (d) Cannot be determined
- 100. Manick is fourteenth from the right end in a row of 40 boys. What is his position from the left end?
  - (a) 24th
- (b) 25<sup>th</sup>
- (c) 26<sup>th</sup>
- (d) 27<sup>th</sup>
- 101. The missing number in the given figure is



- 16

- (a) 13
- (b) 15
- (c) 17
- (d) 19
- 102. Select the combination of numbers so that the letters arranged will form a meaningful word.

HNRCAB 1 23 456

- (a) 2, 5, 3, 4, 1, 6
- (b) 3, 5, 6, 4, 1, 2
- (c) 4, 1, 5, 6, 2, 3
- (d) 6, 3, 5, 2, 4, 1
- 103. Which of the given Venn diagrams out of (a), (b), (c) or (d) correctly represents the relationship among the following classes? Rose, Flower, Lotus









0

(d)

(a)

- (b)
- (c)
- 104. A piece of paper is folded and a cut is made as shown below. From the given responses indicate how it will appear when opened?

#### Question figures



(a)







#### Answer figures









105. Which answer figure will complete the question figure?

### Question figure



#### Answer figures



(a)







# Part IV

### Mathematics

- 106. The equation of the normal to the circle  $x^2 + y^2 = a^2$  at point (x', y') will be
  - (a) x'y xy' = 0
- (b) xx' yy' = 0
- (c) x'y + xy' = 0 (d) xx' + yy' = 0
- 107. Equation of the bisector of the acute angle between lines 3x + 4y + 5 = 012x - 5y - 7 = 0 is
  - (a) 21x + 77y + 100 = 0
  - (b) 99x 27y + 30 = 0
  - (c) 99x + 27y + 30 = 0
  - (d) 21x 77y 100 = 0
- 108. If  $z = \cos \theta + i \sin \theta$ , then the value of  $z^n + \frac{1}{z^n}$

will be

- (a)  $\sin 2n\theta$
- (b) 2 sin nθ
- (c)  $2\cos n\theta$
- (d)  $\cos 2n\theta$
- 109. If  $\alpha$  and  $\beta$  are the roots of the equation  $x^2 - 2x + 4 = 0$ , then the value of  $\alpha^n + \beta^n$  will
  - (a)  $i2^{n+1} \sin(n\pi/3)$  (b)  $2^{n+1} \cos(n\pi/3)$
  - (c)  $i2^{n-1}\sin(n\pi/3)$  (d)  $2^{n-1}\cos(n\pi/3)$
- 110. If  $A = \begin{bmatrix} 3 & 1 \\ -1 & 2 \end{bmatrix}$  and  $I = \begin{bmatrix} 1 & 0 \\ 0 & 1 \end{bmatrix}$ , then the

correct statement is

- (a)  $A^2 + 5A 71 = 0$
- (b)  $-A^2 + 5A + 7I = 0$
- (c)  $A^2 5A + 7I = 0$
- (d)  $A^2 + 5A + 7I = 0$
- 111. The value the determinant a-b-cb-c-a2bwill be 2c 2c c-a-b
  - (a)  $(a-b-c)(a^2+b^2+c^2)$
  - (b)  $(a+b+c)^3$
  - (c) (a + b + c)(ab + bc + ca)
  - (d) None of the above
- $(1+x)^n = C_0 + C_1x + C_2x^2 + ... + C_nx^n$ then  $C_0 - C_1 + C_2 - C_3 + ... + (-1)^n \cdot C_n$  is equal to

- (a)  $3^{n}$
- (b) 2<sup>n</sup>
- (c) 1
- (d) 0
- 113. If AM and HM between two numbers are 27 and 12 respectively, then their GM is
  - (a) 9
- (b) 18
- (c) 24
- (d) 36
- 114. For any two events A and B, if  $P(A \cup B) = 5/6,$  $P(A \cap B) = 1/3$ 
  - P(B) = 1/2, then P(A) is
  - (a) 1/2
- (b) 2/3
- (c) 1/3
- (d) None of these
- 115. A bag contains 3 white and 5 black balls. One ball is drawn at random. Then, the probability that it is white, is
  - (a)

- 116.  $\mathbf{a} \cdot (\mathbf{b} \times \mathbf{c}) = 0$ , then the correct statement is
  - (a) out of a, b, c any two vectors are parallel
  - (b) a, b, c are coplanar
  - (c) any two are equal a, b, c
  - (d) at least one above statement is correct
- 117. If 2i + j k and  $i 4j + \lambda k$  are perpendicular to each other, then \(\lambda\) is equal to
  - (a) -3
- (c) -1
- 118. If  $\frac{d}{dx}(\phi(x)) = f(x)$ , then  $\int_{1}^{2} f(x) dx$  is equal to
  - (a) f(1) f(2) (b)  $\phi(1) \phi(2)$
  - (c) f(2) f(1)
- 119.  $\int_{0}^{2} |1-x| dx$  is equal to
- (c)  $\frac{3}{2}$
- 120.  $\int \frac{\sin 2x}{\sin^4 x + \cos^4 x} dx$  is equal to
  - (a)  $2 \tan^{-1}(\tan^2 x) + C$
  - (b)  $\tan^{-1}(x \tan^2 x) + C$
  - (c)  $\tan^{-1}(\tan^2 x) + C$
  - (d) None of the above

- 121. The function  $\sin x + \cos x$  is maximum when x is equal to

- 122.  $\frac{d}{dx}(x^x)$  is equal to
  - (a)  $x^x \log(e/x)$
- (b)  $x^x \log ex$
- (c) log ex
- (d)  $x^x \log x$
- **123.**  $\lim_{x \to 0} \frac{\sin x}{x}$  is equal to
  - (a) 2
- (b) -1
- (c) 1
- (d) 0
- **124.** The set  $A = \{x : x \in R, x^2 = 16 \text{ and } 2x = 6\}$ equals
  - (a) o
- (b) {14, 3, 4}
- (c) {3}
- (d) {4}
- 125. In how many ways can 5 prizes be distributed among four students when every student can take one or more prizes?
  - (a) 1024
- (b) 625
- (c) 120
- (d) 600
- **126.** The value of  $(\sqrt{5} + 1)^5 (\sqrt{5} 1)^5$  is
  - (a) 252
- (b) 352
- (c) 452
- (d) 552
- 127. The value of

$$7 \log \left(\frac{16}{15}\right) + 5 \log \left(\frac{25}{24}\right) + 3 \log \left(\frac{81}{80}\right)$$
 is equal to

- (a) log 2
- (c) 5
- **128.** The value of  $\frac{2}{1!} + \frac{2+4}{2!} + \frac{2+4+6}{3!} + \dots \infty$  is
  - (a) e
- (c) 3e
- (d) None of these
- 129. The of the series sum  $\log_4 2 - \log_8 2 + \log_{16} 2 - \dots$  is
  - (a) e2
- (b) log, 2
- (c)  $\log_{a} 3 2$
- (d) 1 log, 2
- 130. If the domain of the function  $f(x) = x^2 - 6x + 7$  is  $(-\infty, \infty)$ , then the range of function is
  - (a)  $(-\infty, \infty)$
- (b)  $[-2, \infty)$
- (c) (-2, 3)
- (d)  $(-\infty, -2)$

- 131,  $\lim_{x\to 0} \frac{\cos(\sin x) 1}{x^2}$  is equal to
- (c)  $\frac{1}{2}$
- (b) -1(d)  $-\frac{1}{2}$
- **132.** In order that the function  $f(x) = (x+1)^{1/x}$  is continuous at x = 0, f(0) must be defined as
  - (a) f(0) = 0
- (b) f(0) = e
- (c)  $f(0) = \frac{1}{2}$
- (d) f(0) = 1
- **133.** The function f(x) = |x| at x = 0 is
  - (a) continuous but non-differentiable
  - (b) discontinuous and differentiable
  - (c) discontinuous and non-differentiable
  - (d) continuous and differentiable
- 134. The point (0, 5) is closer to the curve  $x^2 = 2y$ 

  - (a)  $(2\sqrt{2}, 0)$
- (b) (0, 0)
- (c) (2, 2)
- (d) None of these
- **135.** The function  $f(x) = x^{1/x}$  is
  - (a) increasing in (1, ∞)
  - (b) decreasing in (1, ∞)
  - (c) increasing in (1, e), decreasing in (e, ∞)
  - (d) decreasing in (1, e), increasing in (e, ∞)
- 136. The area bounded by the x-axis and the curve  $y = \sin x$  and x = 0,  $x = \pi$  is

  - (a) 1 sq unit (b) 2 sq units
  - (c) 0
- (d) 4 sq units
- 137. The order and degree of the differential equation  $\sqrt{\frac{dy}{dx}} - 4\frac{dy}{dx} - 7x = 0$  are
  - (a) 1 and  $\frac{1}{2}$
- (b) 2 and 1
- (c) 1 and 1
- (d) 1 and 2
- 138. The line x + y = 4 divides the line joining the points (-1, 1) and (5, 7) in the ratio
  - (a) 2:1
- (b) 1:2
- (c) 1:2 externally
- (d) None of these
- 139. The angle between the pair of lines given by equation  $x^2 + 2xy - y^2 = 0$ , is
- (c)  $\frac{\pi}{2}$
- (d) 0

- 140. The length of tangent from point (5, 1) to the circle  $x^2 + y^2 + 6x - 4y - 3 = 0$  is
  - (a) 81
- (b) 29
- (c) 7
- (d) 21
- 141. The length of the latusrectum of the parabola  $169\{(x-1)^2 + (y-3)^2\} = (5x-12y+17)^2$ 
  - (a)  $\frac{14}{13}$
- (b)  $\frac{12}{12}$
- (c)  $\frac{28}{13}$
- (d) None of these
- 142. The angle of intersection between the curves  $x^2 = 8y$  and  $y^2 = 8x$  at (0, 0) is

- 143. If the centre, one of the foci and semi-major axis of an ellipse be (0, 0), (0, 3) and 5, then its equation is
  - (a)  $\frac{x^2}{16} + \frac{y^2}{25} = 1$  (b)  $\frac{x^2}{25} + \frac{y^2}{16} = 1$
  - (c)  $\frac{x^2}{0} + \frac{y^2}{25} = 1$  (d) None of these
- 144. The radius of the director circle of the hyperbola  $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$  is
  - (a) a-b
- (b) √a − b
- (c)  $\sqrt{a^2 b^2}$
- (d)  $\sqrt{a^2 + b^2}$
- 145. If projection of any line on coordinate axes 3, 4 and 5, then its length is
  - (a) 12
- (b) 50
- (c) 5√2
- (d) 3√2

- **146.** If  $\tan \theta = \frac{1}{2}$  and  $\tan \phi = \frac{1}{3}$ , then the value of

  - (a)  $\frac{\pi}{6}$
- (b) π

- **147.** If  $\sin \theta = \frac{1}{2}$ ,  $\tan \theta = \frac{1}{\sqrt{3}}$ ,  $\forall n \in I$ , then most general values of  $\theta$  is
  - (a)  $2n\pi + \frac{\pi}{6}$ ,  $\forall n \in I$  (b)  $2n\pi + \frac{\pi}{4}$ ,  $\forall n \in I$
  - (c)  $2n\pi + \frac{\pi}{3}$ ,  $\forall n \in I$  (d)  $2n\pi + \frac{\pi}{3}$ ,  $\forall n \in I$
- 148. The principal value of  $\sin^{-1}\left(-\frac{\sqrt{3}}{2}\right)$  is
  - (a)  $-\frac{2\pi}{3}$
- (c)  $\frac{4\pi}{3}$  (d)  $\frac{5\pi}{3}$
- 149. A ladder rests against a wall so that its top touches the roof of the house. If the ladder makes an angle of 60° with the horizontal and height of the house be 6\sqrt{3} m, then the length of the ladder is
  - (a)  $12\sqrt{3}$  m
- (b) 12 m
- (c)  $\frac{12}{\sqrt{2}}$  m
- (d) None of these
- **150.** If angles A, B and C are in AP, then  $\frac{a+c}{b}$  is equal to

  - (a)  $2\sin\left(\frac{A-C}{2}\right)$  (b)  $2\cos\left(\frac{A-C}{2}\right)$

  - (c)  $\cos\left(\frac{A-C}{2}\right)$  (d)  $\sin\left(\frac{A-C}{2}\right)$