

# Electric Vehicles

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# Energy Storage -Introduction

- Energy storage mediates between variable sources and variable loads.
- Without storage, energy generation must equal energy demand.  
Energy storage works by moving energy through time.

## ***Classification of Electrical Energy Storage Technologies***

### ***Mechanical***

Pumped Hydro-PHS

Compressed Air-CAES

Flywheel-FES

### ***Electrochemical***

Secondary battery

Lead-acid/NaS/Li-ion

Flow battery

Redox flow/Hybrid flow

### ***Electrical***

Capacitor

Supercapacitor

Superconducting

Magnetic-SMES

### ***Thermochemical***

Solar fuels

Solar hydrogen

### ***Chemical***

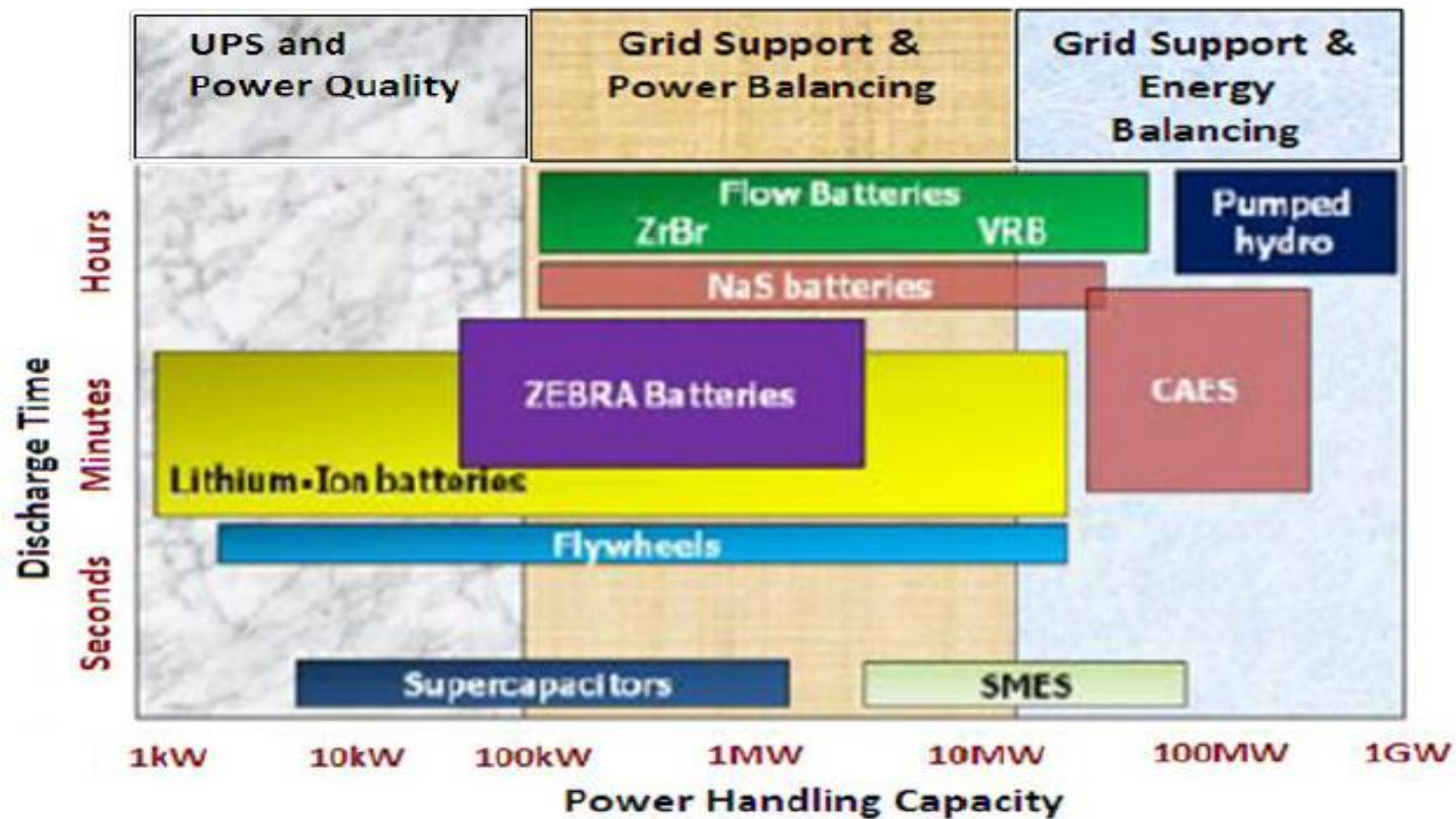
Hydrogen

Fuel cell/Electrolyser

### ***Thermal***

Sensible/latent

heat storage





# A comparative analysis among various Energy Storage technologies

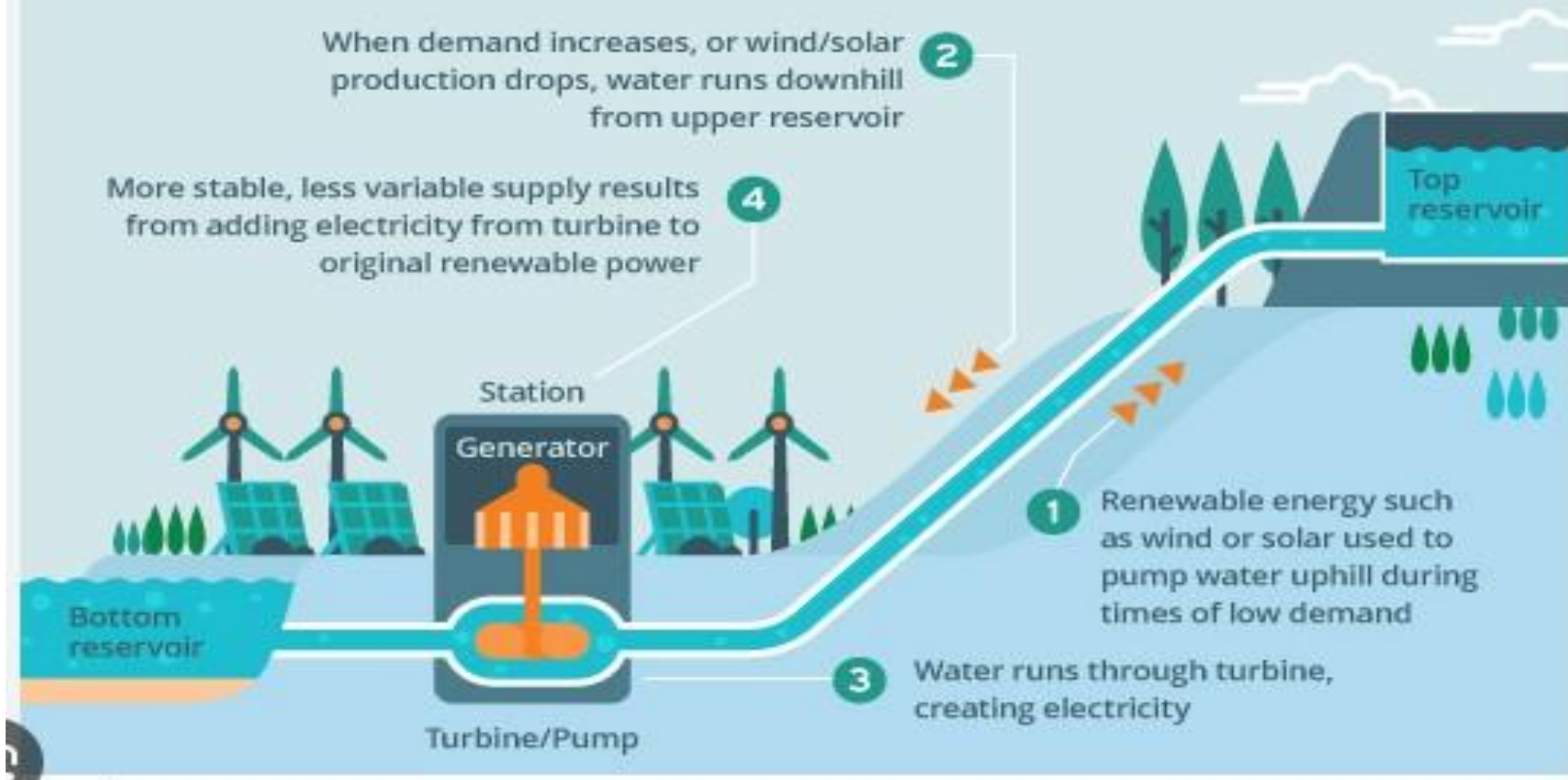


Technology	Typical Capacity	Response time	Discharge time	Efficiency	Life time	Development stage	Application
Pumped hydro energy storage (PHES)	5 MW – 2 GW	1 min (if standing still) 10 sec (if spinning)	4 - 100 h	55-85%	50+ years	Mature	Primary / secondary / tertiary control, energy arbitrage
Compressed air energy systems (CAES)	25 MW – 2.5 GW	10 - 15 min from cold start	2 - 24 h	40-70%	15-40 years	Mature	Tertiary control, energy arbitrage
Batteries	1 kW – 100 MW	Milli seconds to few Seconds	1 min - 10 h	65-85%	2 - 10 years (Even up to 20 years)	Premature / mature	Uninterruptible power supply, RES- fluctuation reduction, primary / secondary control
Flywheels	5 kW – 20 MW	Few seconds	4 sec - 15 min	90-95%	~20 years	Mature	Primary control, power quality
Hydrogen Fuel Cell Storage System (HFCSS)	1 kW – 10 GW	Depends on fuel cell	0.01 sec- days	20-40%	5-10 years	Prototype	RES- fluctuation reduction, tertiary reserve
Super magnetic energy storage (SMES)	10 kW – 1 MW	Milli Seconds to few sensonds	5 sec – 5 min	95%	~30 years	Premature	Uninterruptible power supply, power quality
Supercapacitors	< 150 kW	Milli seconds	1 sec – 1 min	85-95%	~10 years	Premature	Uninterruptible power supply, power quality

# Pumped Hydro

- 99% of world wide installed electrical storage capacity
- Two reservoirs at different elevations
- At off peak hours pump water from lower to upper reservoir
- When needed, water flows back from the upper to lower reservoir powering a turbine

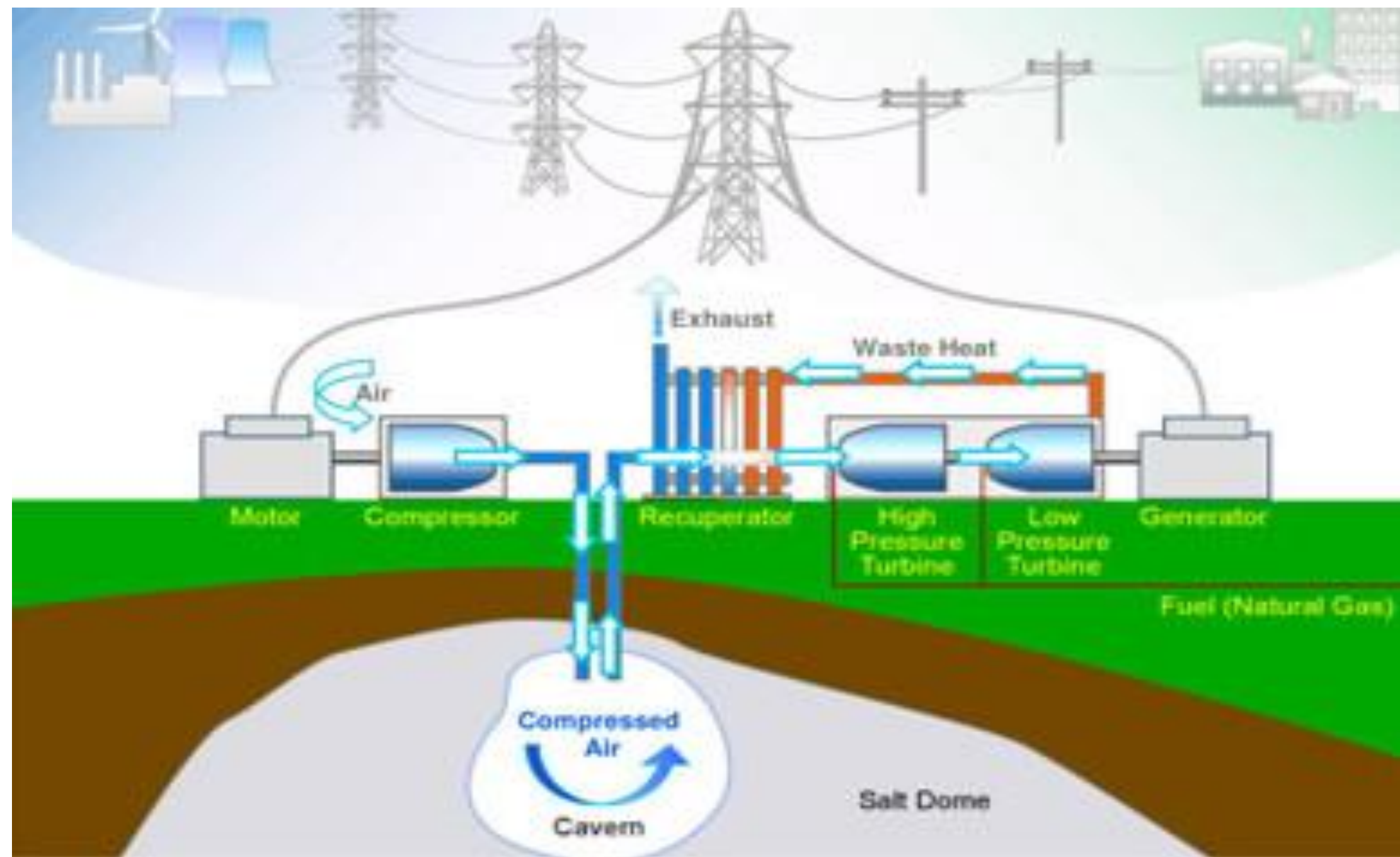
# PUMPED HYDRO STORAGE - HOW IT WORKS



# Compressed Air Energy Storage (CAES)

- Air is used as storage medium.
- CAES systems use off-peak electricity to compress air and store it in a reservoir either in an underground cavern or abandoned mines or above ground pipes or vessels.
- When electricity is needed, the compressed air is heated, expanded and directed through conventional turbine-generator to produce electricity.



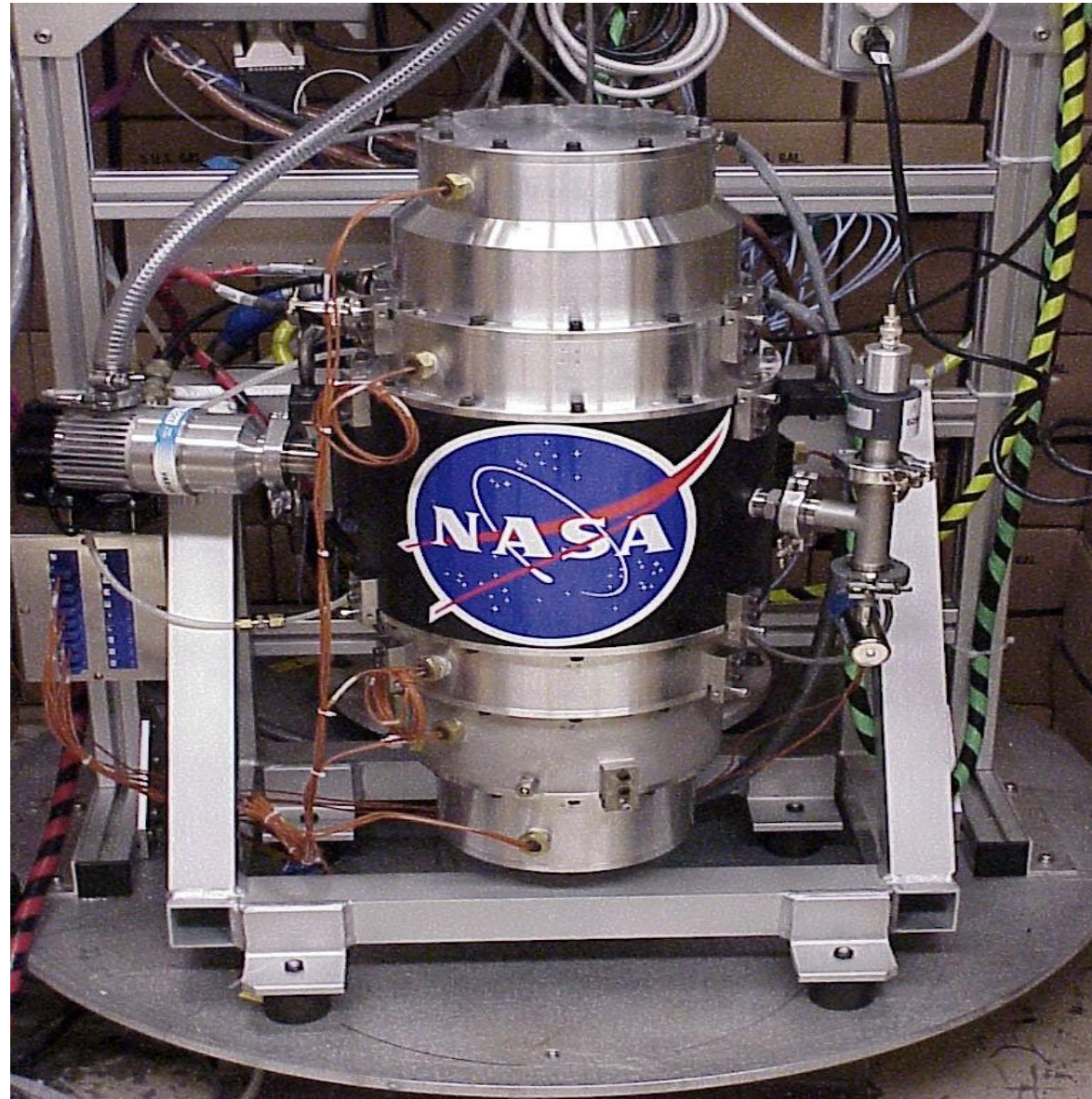


- Compression of air creates heat; the air is warmer after compression.
- Expansion removes heat. If no extra heat is added, the air will be much colder after expansion.
- If the heat generated during compression can be stored and used during expansion, then the efficiency of the storage improves considerably

Salt caverns are underground cavities created in salt deposits through solution mining, a process of dissolving the salt with water. They are used to store large quantities of energy resources like crude oil, natural gas, and hydrogen because of their natural sealing properties and low absorption of hydrocarbons, which create an impermeable barrier. These facilities are cost-effective and efficient alternatives to surface storage for [strategic petroleum reserves](#) and other energy applications.

# Flywheel Energy Storage

- Stores energy in the form of angular momentum of a spinning mass
- Work done to spin mass is stored in the form of kinetic energy
- A flywheel system transfers kinetic energy into ac power through the use of controls and power conversion systems
- High power density, low energy capacity

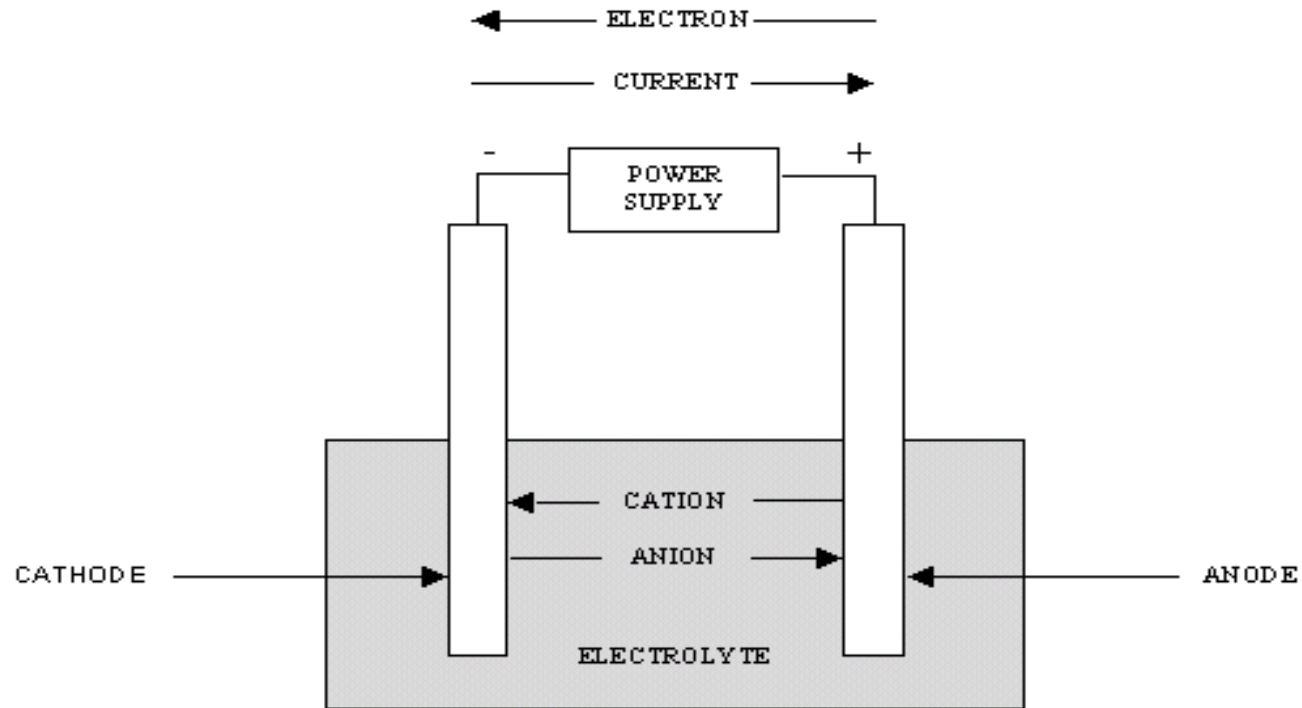




# Electrochemical Energy Storage

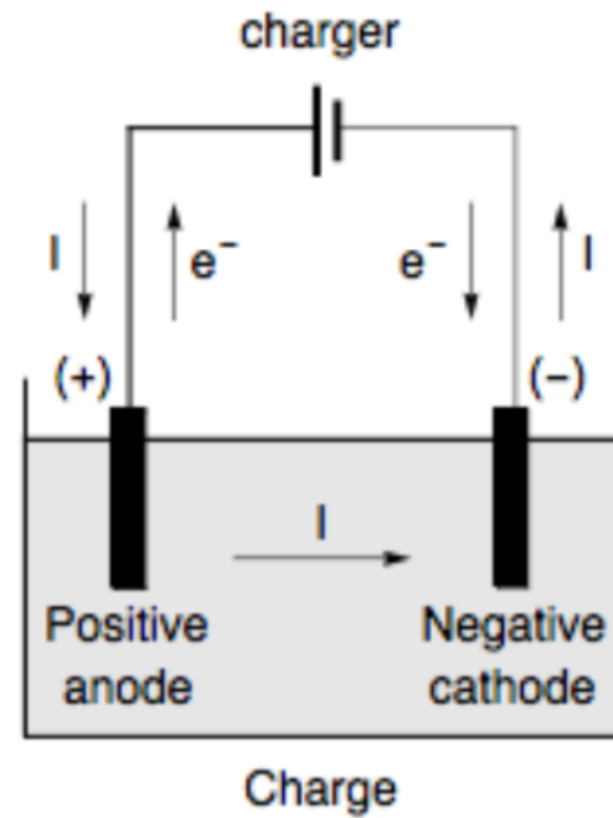
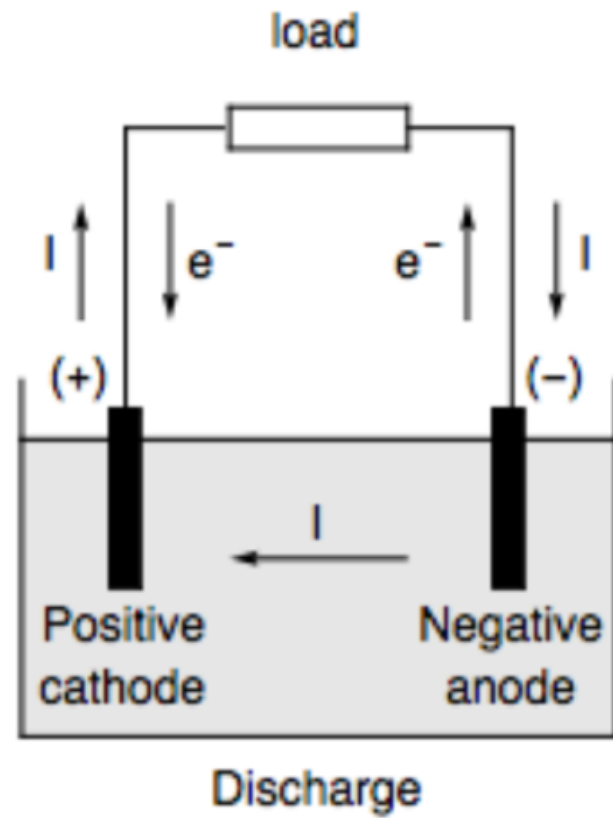
- Electrical energy is stored in chemical energy
- Both techniques share the same carrier, electron. Losses are limited.
- A battery consists of two electrodes, a positive and a negative placed in an electrolyte
- Secondary batteries are rechargeable, primary batteries are not
- The electrodes exchange ions with the electrolyte and electrons with external circuit.
- Common Batteries- Lead acid, Sodium Sulfur, Lithium Ion, Nickel Cadmium, Nickel metal hydrides

# Battery Cell



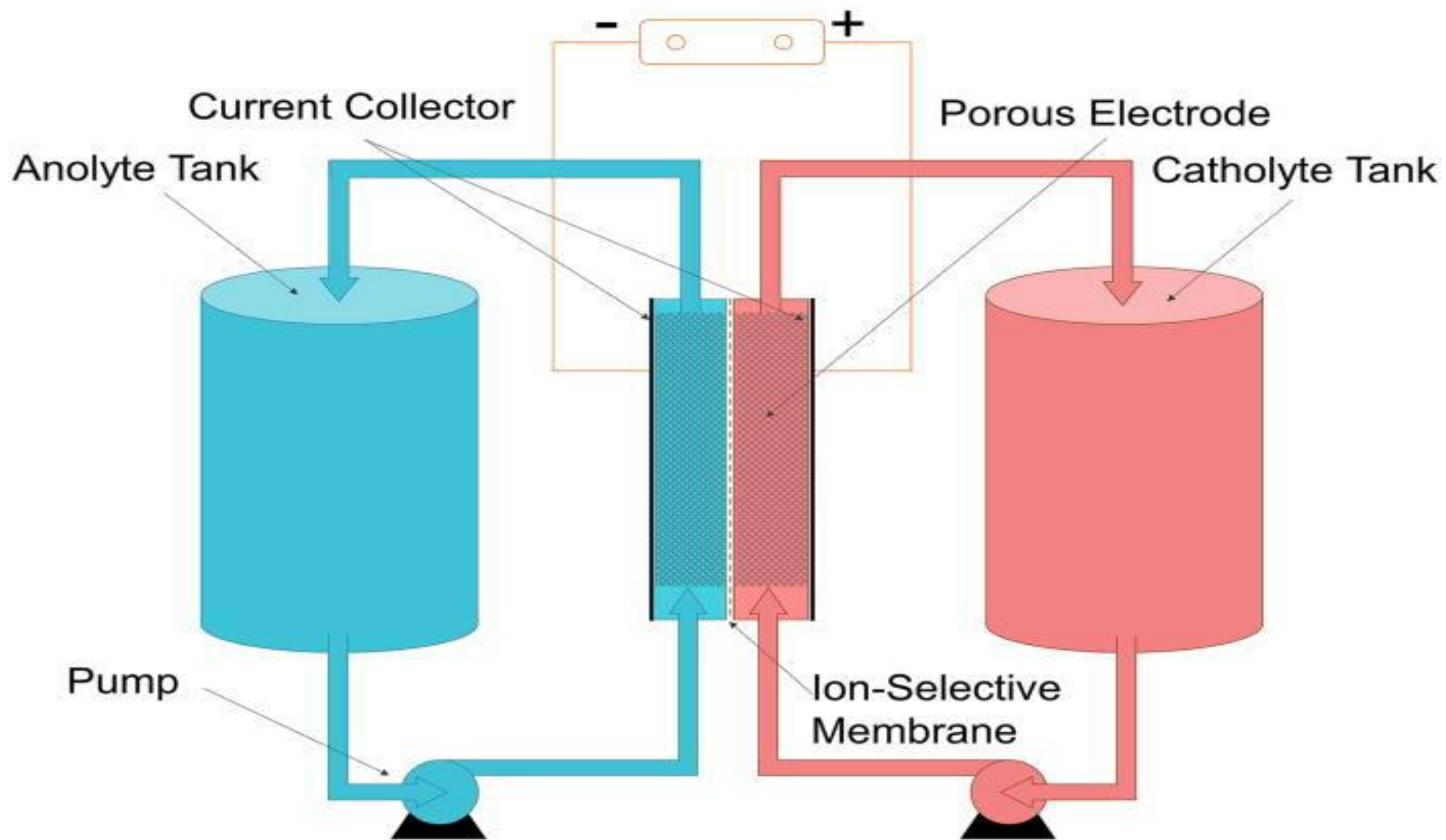
**Figure 3: Recharging a Cell**

# Battery Cell



# Flow Battery

- A **flow battery**, or **redox flow battery** (after [reduction–oxidation](#)), is a type of [electrochemical cell](#) where [chemical energy](#) is provided by two chemical components [dissolved](#) in liquids that are pumped through the system on separate sides of a membrane.
- [Ion transfer](#) inside the cell (accompanied by flow of electric current through an external circuit) occurs through the membrane while both liquids circulate in their own respective space.

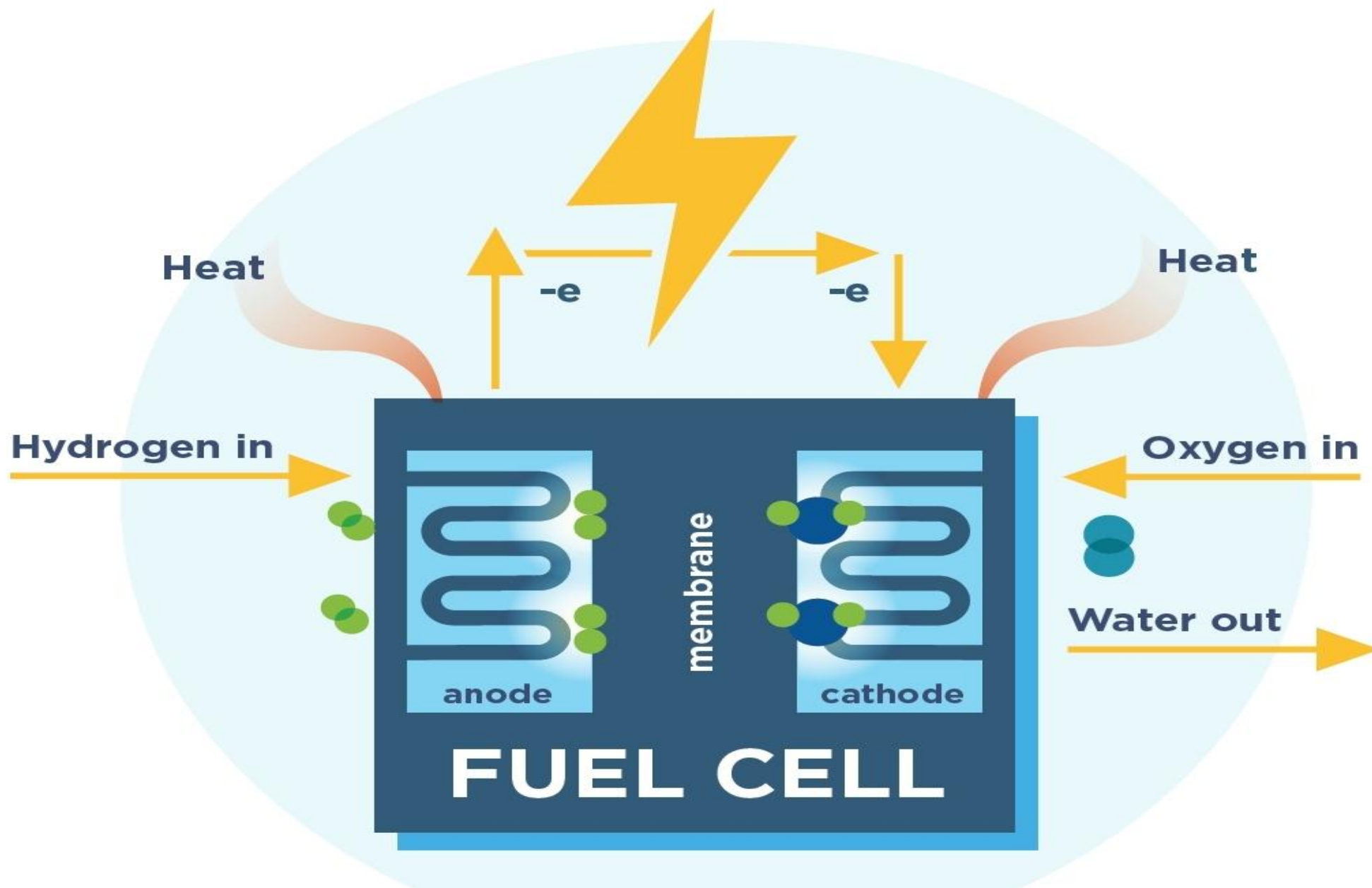




- The fundamental difference between conventional and flow batteries is that energy is stored in the electrode material in conventional batteries, while in flow batteries it is stored in the electrolyte.
- The power and energy ratings are independent in flow battery; the storage capacity being determined by the quantity of electrolyte used and the power rating determined by the active area of the cell stack.

# Chemical Energy Storage

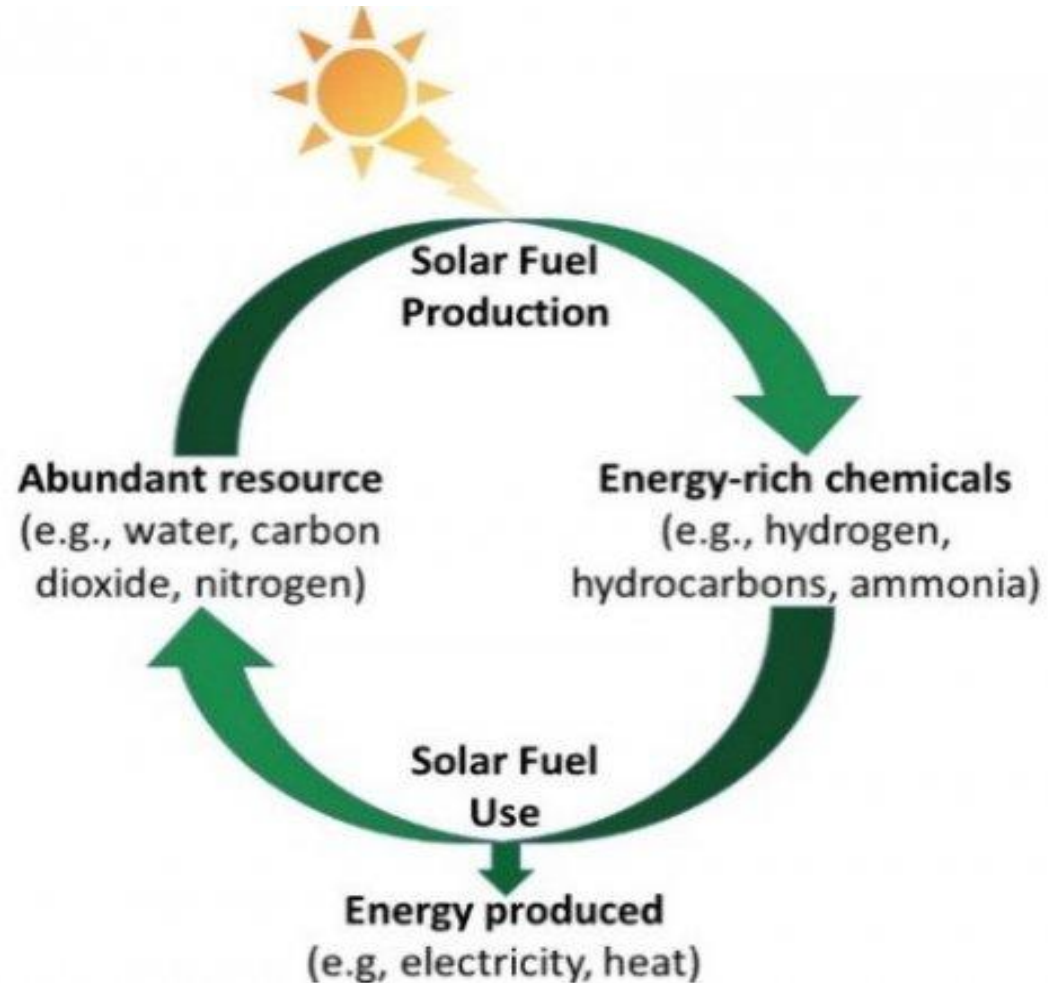
- Power to Gas (P2G) Technology
- Use excess electricity to produce hydrogen via water electrolysis
- Hydrogen is produced and can be stored upto TWH range and for greater periods of time
- Gaseous storage under high pressure in gas bottles or tanks
- Fuel cell is needed to generate water from hydrogen and oxygen
- Reverse of electrolysis –an electrochemical reaction takes place
- Heat produced can be used to generate electricity



# Fuel Cells

- Fuel cells work like batteries, but they do not run down or need recharging. They produce electricity and heat as long as fuel is supplied.
- A fuel cell consists of two electrodes—a negative electrode (or anode) and a positive electrode (or cathode)—sandwiched around an electrolyte.
- A fuel, such as hydrogen, is fed to the anode, and air is fed to the cathode.
- In a hydrogen fuel cell, a catalyst at the anode separates hydrogen molecules into protons and electrons, which take different paths to the cathode. The electrons go through an external circuit, creating a flow of electricity. The protons migrate through the electrolyte to the cathode, where they unite with oxygen and the electrons to produce water and heat.

# Solar Fuel – Thermo chemical



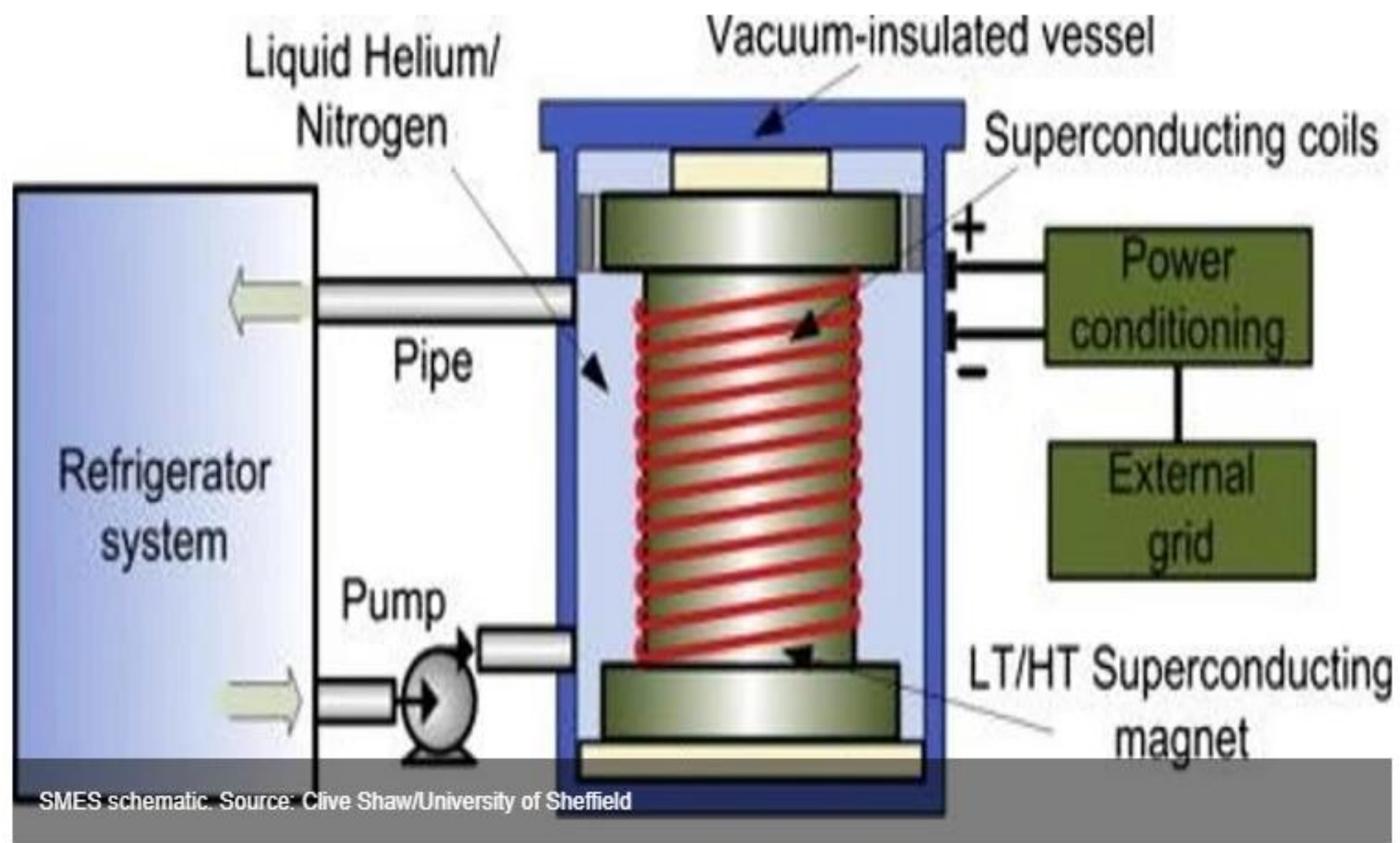


- **Solar fuels** are fuels made from common substances like water and carbon dioxide using the energy of sunlight.
- There is vast energy in sunlight striking the earth, but it is time-varying and dispersed, making it challenging to harness sunlight for practical use.
- Solar fuels could be an abundant supply of sustainable, storable, and portable energy.

# Superconducting Magnetic Energy Storage Systems (SMES)

- Superconducting magnetic energy storage (SMES) systems store energy in a magnetic field. This magnetic field is generated by a DC current traveling through a [superconducting coil](#).
- In a normal wire, as electric current passes through the wire, some energy is lost as heat due to electric resistance. However, in a SMES system, the wire is made from a superconducting material that has been cryogenically cooled below its critical temperature.
- As a result, electric current can pass through the wire with almost no resistance, allowing energy to be stored in a SMES system for a longer period of time.

- Common superconducting materials include mercury, vanadium, and niobium-titanium.
- SMES systems are an extremely efficient storage technology, but they have very low energy densities and are still far from being economically viable



# Super Capacitors

- A **supercapacitor (SC)**, also called an **ultracapacitor**, is a high-capacity [capacitor](#), with a [capacitance](#) value much higher than solid-state capacitors. It bridges the gap between [electrolytic capacitors](#) and [rechargeable batteries](#). It typically stores 10 to 100 times more [energy per unit volume or mass](#) than electrolytic capacitors, can accept and deliver charge much faster than batteries, and tolerates many more [charge and discharge cycles](#) than [rechargeable batteries](#).
- Supercapacitors are used in applications requiring many rapid charge/discharge cycles, rather than long-term compact energy storage



# Types

- Double-layer Capacitors
- Pseudo Capacitors
- Hybrid Capacitors

# Thermal Energy Storage System

- Excess electricity is used to heat up water.



District heating accumulation tower of Theiss, near [Krems an der Donau](#), [Lower Austria](#) with 50000 cubic meters volume

- The [sensible heat](#) of [molten salt](#) is also used for storing solar energy at a high temperature. It's termed molten-salt technology or molten salt energy storage (MSES).
- Molten salts can be employed as a thermal energy storage method to retain thermal energy. Presently, this is a commercially used technology to store the heat collected by [concentrated solar power](#) (e.g., from a [solar tower](#) or [solar trough](#)).
- The heat can later be converted into superheated steam to power conventional steam turbines and generate electricity at a later time.