

- 8 Uranium-234 is radioactive and emits  $\alpha$ -particles at what appears to be a constant rate.

A sample of Uranium-234 of mass  $2.65 \mu\text{g}$  is found to have an activity of  $604 \text{ Bq}$ .

- (a) Calculate, for this sample of Uranium-234,

(i) the number of nuclei,

number = ..... [2]

(ii) the decay constant,

decay constant = .....  $\text{s}^{-1}$  [2]

(iii) the half-life in years.

half-life = ..... years [2]

- (b) Suggest why the activity of the Uranium-234 appears to be constant.

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..... [1]

- (c) Suggest why a measurement of the mass and the activity of a radioactive isotope is not an accurate means of determining its half-life if the half-life is approximately one hour.

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..... [1]

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