

- 8 The element strontium has at least 16 isotopes. One of these isotopes is strontium-89. This isotope has a half-life of 52 days.

- (a) State what is meant by *isotopes*.

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.....  
.....

[2]

- (b) Calculate the probability per second of decay of a nucleus of strontium-89.

$$\text{probability} = \dots \text{s}^{-1} [3]$$

- (c) A laboratory prepares a strontium-89 source.

The activity of this source is measured 21 days after preparation of the source and is found to be  $7.4 \times 10^6 \text{ Bq}$ .

Determine, for the strontium-89 source at the time that it was prepared,

- (i) the activity,

$$\text{activity} = \dots \text{Bq} [2]$$

- (ii) the mass of strontium-89.

$$\text{mass} = \dots \text{g} [2]$$