

10 (a) By reference to the photoelectric effect, explain what is meant by *work function energy*.

.....  
.....  
..... [2]

(b) In an experiment, electromagnetic radiation of frequency  $f$  is incident on a metal surface.

The results in Fig. 10.1 show the variation with frequency  $f$  of the maximum kinetic energy  $E_{\text{MAX}}$  of electrons emitted from the surface.

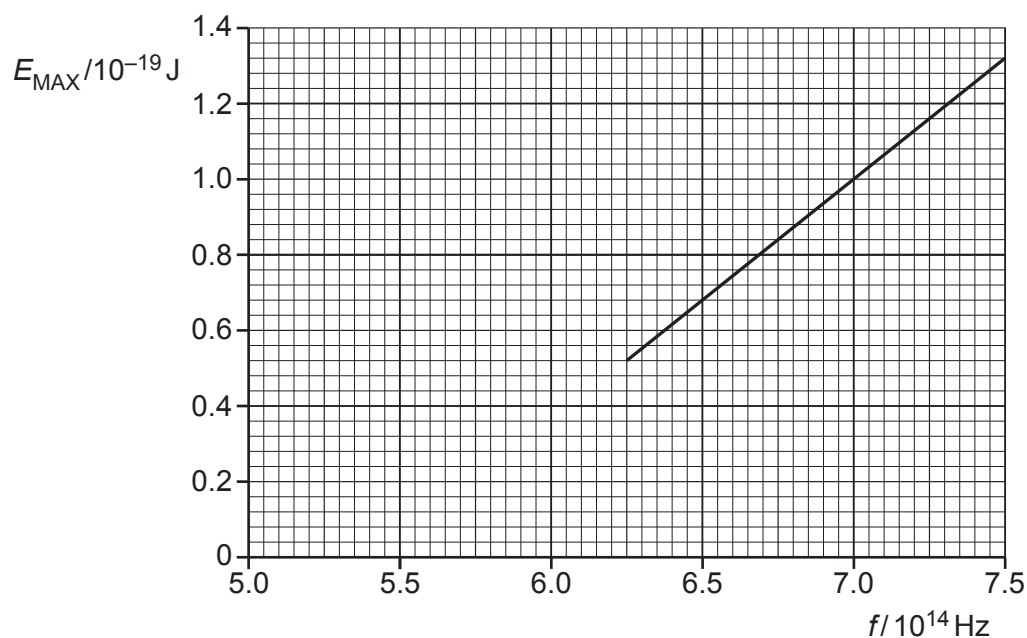


Fig. 10.1

(i) Determine the work function energy in J of the metal used in the experiment.

work function energy = ..... J [2]

- (ii) The work function energy in eV for some metals is given in Table 10.1.

**Table 10.1**

metal	work function / eV
tungsten	4.49
magnesium	3.68
potassium	2.26

Determine the metal used in the experiment. Show your working.

.....  
 ..... [1]

- (c) The intensity of the electromagnetic radiation for one particular frequency in (b) is increased.

State and explain the change, if any, in:

- (i) the maximum kinetic energy of the emitted electrons

.....  
 ..... [1]

- (ii) the rate of emission of photoelectrons.

.....  
 ..... [1]