

- 11 (a) Electromagnetic radiation is incident on a metal surface.

It is observed that there is a minimum frequency of electromagnetic radiation below which emission of electrons does not occur.

This observation provides evidence for a particulate nature of electromagnetic radiation.

State **two** other observations associated with photoelectric emission that provide evidence for a particulate nature of electromagnetic radiation.

1. ....

.....

2. ....

.....

[2]

- (b) The maximum kinetic energy  $E_{\text{MAX}}$  of electrons emitted from a metal surface is determined for different wavelengths  $\lambda$  of the electromagnetic radiation incident on the surface.

The variation with  $\frac{1}{\lambda}$  of  $E_{\text{MAX}}$  is shown in Fig. 11.1.

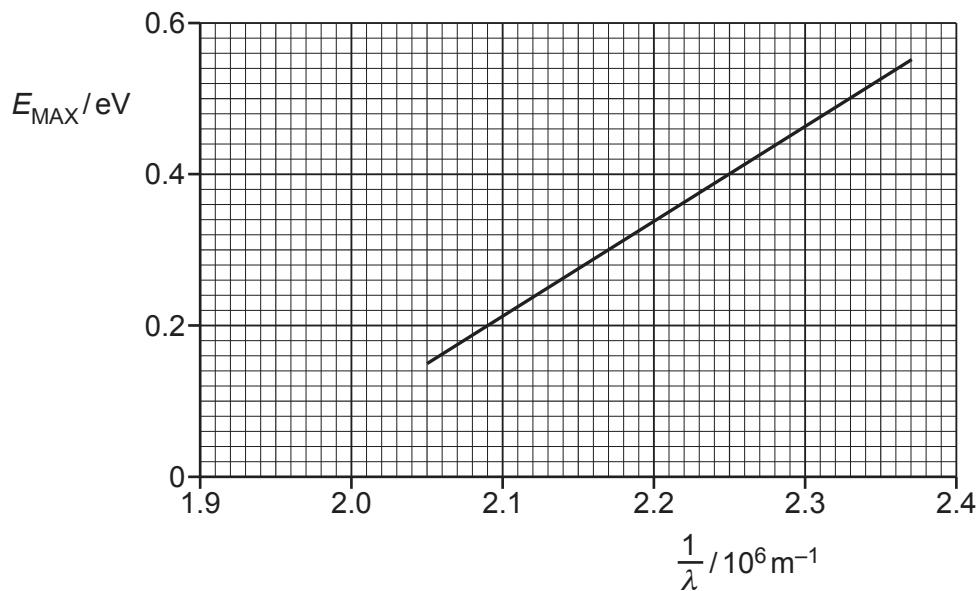


Fig. 11.1

- (i) Use Fig. 11.1 to determine the threshold frequency  $f_0$ .

$$f_0 = \dots \text{ Hz} [2]$$

- (ii) Use the gradient of the line on Fig. 11.1 to determine a value for the Planck constant  $h$ . Explain your working.

$$h = \dots \text{ Js} [4]$$

- (c) The electromagnetic radiation is now incident on a metal with a larger work function energy than the metal in (b).

On Fig. 11.1, sketch the variation with  $\frac{1}{\lambda}$  of  $E_{\text{MAX}}$ .

[2]

[Total: 10]