

- 9 (a) (i) State what is meant by the *decay constant* of a radioactive isotope.

For
Examiner's
Use

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[2]

- (ii) Show that the decay constant λ and the half-life $t_{\frac{1}{2}}$ of an isotope are related by the expression

$$\lambda t_{\frac{1}{2}} = 0.693.$$

[3]

- (b) In order to determine the half-life of a sample of a radioactive isotope, a student measures the count rate near to the sample, as illustrated in Fig. 9.1.

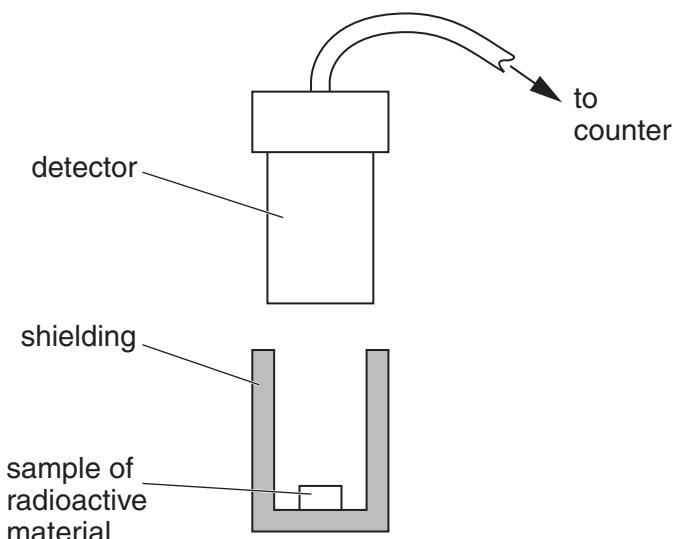


Fig. 9.1

Initially, the measured count rate is 538 per minute. After a time of 8.0 hours, the measured count rate is 228 per minute.

Use these data to estimate the half-life of the isotope.

$$\text{half-life} = \dots \text{ hours} [3]$$

- (c) The accepted value of the half-life of the isotope in (b) is 5.8 hours.
The difference between this value for the half-life and that calculated in (b) cannot be explained by reference to faulty equipment.

Suggest two possible reasons for this difference.

1.

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2.

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[2]