

- 8 (a) State what is meant by the work function energy of a metal.

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- (b) Ultraviolet radiation of frequency  $1.36 \times 10^{15}$  Hz is incident, in a vacuum, on a metal surface. The power of the radiation incident on the surface is 8.36 mW. Photoelectrons are emitted with a maximum kinetic energy of  $3.09 \times 10^{-19}$  J.

- (i) Determine the number of photons incident on the surface per unit time.

$$\text{number per unit time} = \dots \text{ s}^{-1} [2]$$

- (ii) Calculate the work function energy  $\Phi$  of the metal.

$$\Phi = \dots \text{ J} [2]$$

- (c) The frequency of the radiation incident on the surface in (b) is increased while the power remains constant.

State and explain the effect of this change on:

- (i) the maximum kinetic energy of the photoelectrons

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- (ii) the rate of emission of photoelectrons.

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