

- 8 Uranium-234 is radioactive and emits α -particles at what appears to be a constant rate.

A sample of Uranium-234 of mass $2.65 \mu\text{g}$ is found to have an activity of 604 Bq.

(a) Calculate, for this sample of Uranium-234,

(i) the number of nuclei,

$$\text{number} = \dots \quad [2]$$

(ii) the decay constant,

$$\text{decay constant} = \dots \text{ s}^{-1} \quad [2]$$

(iii) the half-life in years.

$$\text{half-life} = \dots \text{ years} \quad [2]$$

- (b) Suggest why the activity of the Uranium-234 appears to be constant.

.....
..... [1]

- (c) Suggest why a measurement of the mass and the activity of a radioactive isotope is not an accurate means of determining its half-life if the half-life is approximately one hour.

.....
..... [1]

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