

9 Fluorine-18 (${}^{18}_9\text{F}$) decays by beta-plus (β^+) emission with a half-life of 110 minutes.

- (a) (i) State the name of the beta-plus particle.

..... [1]

- (ii) Show that the decay constant of fluorine-18 is $1.05 \times 10^{-4}\text{s}^{-1}$.

..... [1]

- (iii) Determine the activity of $2.1 \times 10^{-12}\text{kg}$ of fluorine-18.

activity = Bq [3]

- (b) A small sample of fluorine-18 injected into the body acts as a tracer for use in medical imaging.

- (i) Describe how the interaction of a β^+ particle with an electron in the body enables the formation of an image.

.....
.....
.....
.....
..... [3]

- (ii) Suggest why 110 minutes is a suitable half-life for a nuclide used as a tracer in medical diagnosis.

.....
.....
..... [2]