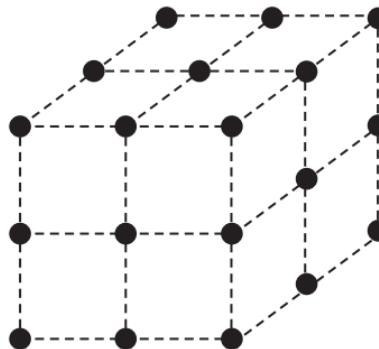


21 The diagram shows the arrangement of atoms in a particular crystal.



Each atom is at the corner of a cube.

The mass of each atom is  $3.5 \times 10^{-25}$  kg. The density of the crystal is  $9.2 \times 10^3$  kg m<sup>-3</sup>.

What is the shortest distance between the centres of two adjacent atoms?

- A**  $3.8 \times 10^{-29}$  m
- B**  $6.2 \times 10^{-15}$  m
- C**  $3.4 \times 10^{-10}$  m
- D**  $3.0 \times 10^{-9}$  m