

- 8 (a) Define the term radioactive *decay constant*.

.....  
 .....  
 .....[2]

- (b) State the relation between the activity  $A$  of a sample of a radioactive isotope containing  $N$  atoms and the decay constant  $\lambda$  of the isotope.

.....[1]

- (c) Radon is a radioactive gas with half-life 56 s. For health reasons, the maximum permissible level of radon in air in a building is set at 1 radon atom for every  $1.5 \times 10^{21}$  molecules of air. 1 mol of air in the building is contained in  $0.024 \text{ m}^3$ .

Calculate, for this building,

- (i) the number of molecules of air in  $1.0 \text{ m}^3$ ,

number = .....

- (ii) the maximum permissible number of radon atoms in  $1.0 \text{ m}^3$  of air,

number = .....

- (iii) the maximum permissible activity of radon per cubic metre of air.

activity = ..... Bq  
[5]

