

- 3 (a) Define *specific heat capacity*.

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[2]

- (b) A student carries out an experiment to determine the specific heat capacity of a liquid using the apparatus illustrated in Fig. 3.1.

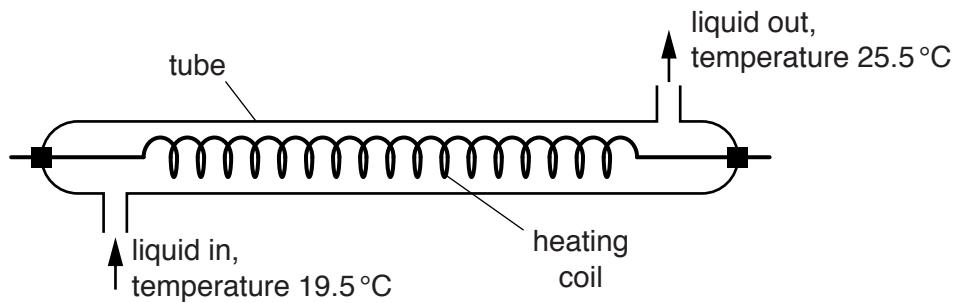


Fig. 3.1

Liquid enters the tube at a constant temperature of 19.5°C and leaves the tube at a temperature of 25.5°C . The mass of liquid flowing through the tube per unit time is m . Electrical power P is dissipated in the heating coil.

The student changes m and adjusts P until the final temperature of the liquid leaving the tube is 25.5°C .

The data shown in Fig. 3.2 are obtained.

$m/\text{g s}^{-1}$	P/W
1.11	33.3
1.58	44.9

Fig. 3.2

- (i) Suggest why the student obtains data for two values of m , rather than for one value.

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[1]

- (ii) Calculate the specific heat capacity of the liquid.

Show your working.

$$\text{specific heat capacity} = \dots \text{J kg}^{-1} \text{K}^{-1} [3]$$

- (c) When the heating coil in (b) dissipates 33.3W of power, the potential difference V across the coil is given by the expression

$$V = 27.0 \sin(395t).$$

The potential difference is measured in volts and the time t is measured in seconds.

Determine the resistance of the coil.

$$\text{resistance} = \dots \Omega [3]$$

[Total: 9]