

- 11 (a) State **three** pieces of evidence provided by the photoelectric effect for a particulate nature of electromagnetic radiation.

1.

.....
2.

.....
3.

[3]

- (b) The work function energies of some metals are shown in Fig. 11.1.

	work function energy/eV
sodium	2.4
calcium	2.9
zinc	3.6
silver	4.3

Fig. 11.1

Each metal is irradiated with electromagnetic radiation of wavelength 380 nm.

- (i) Calculate the energy, in eV, of a photon of electromagnetic radiation of wavelength 380 nm.

$$\text{energy} = \dots \text{eV} \quad [3]$$

- (ii) Determine which metals will give rise to the emission of photoelectrons. Explain your answer.

.....
..... [2]

- (c) Photons of wavelength 380 nm are incident normally on a metal surface at a rate of $7.6 \times 10^{14} \text{ s}^{-1}$.

All the photons are absorbed in the surface and no photoelectrons are emitted.

Calculate the force exerted on the metal surface by the incident photons.

force = N [3]

[Total: 11]