18 Which statement about molecules in a gas is correct?

A In Brownian motion experiments, the molecules can be seen moving randomly in all directions.

B The pressure exerted by a gas is caused by molecules bouncing against each other and changing kinetic energy.

C The pressure exerted by a gas is caused by molecules rebounding from the walls of a container and changing momentum.

D When the average speed of the molecules in a closed container increases, the density must also increase.