

3 (a) Define *specific latent heat*.

For
Examiner's
Use

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.....

[2]

(b) The heater in an electric kettle has a power of 2.40 kW.

When the water in the kettle is boiling at a steady rate, the mass of water evaporated in 2.0 minutes is 106 g.

The specific latent heat of vaporisation of water is 2260 J g^{-1} .

Calculate the rate of loss of thermal energy to the surroundings of the kettle during the boiling process.

rate of loss = W [3]