

7 The photoelectric effect may be represented by the equation

photon energy = work function energy + maximum kinetic energy of electron.

(a) State what is meant by *work function energy*.

.....  
.....[1]

(b) The variation with frequency  $f$  of the maximum kinetic energy  $E_K$  of photoelectrons emitted from the surface of sodium metal is shown in Fig. 7.1.

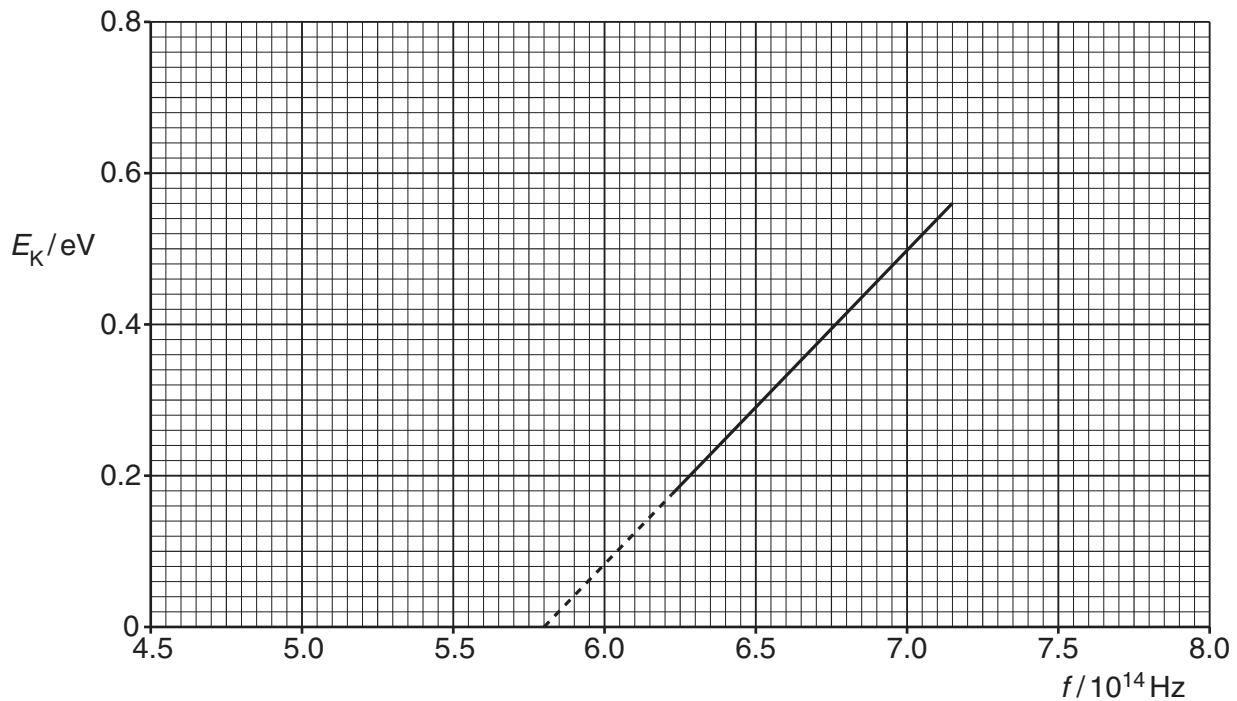


Fig. 7.1

Use the gradient of the graph of Fig. 7.1 to determine a value for the Planck constant  $h$ . Show your working.

$h = \dots\dots\dots \text{Js}$  [2]

- (c) The sodium metal in (b) has a work function energy of 2.4 eV. The sodium is replaced by calcium which has a work function energy of 2.9 eV.

On Fig. 7.1, draw a line to show the variation with frequency  $f$  of the maximum kinetic energy  $E_K$  of photoelectrons emitted from the surface of calcium. [3]