

- 6 (a) Use the theory of the particulate nature of electromagnetic radiation to explain why there is a threshold frequency for the photoelectric effect.

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- (b) An experimental setup to investigate the photoelectric effect is shown in Fig. 6.1. Electromagnetic radiation is incident on the emitter E of the photocell and the photoelectrons move towards the collector C.

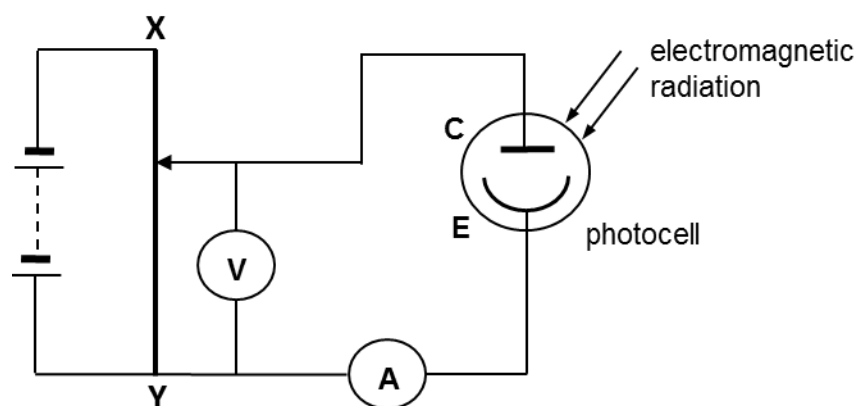


Fig. 6.1

The wavelength of the electromagnetic radiation incident on the photocell was varied. For two values of wavelength λ , the stopping potential V_s required for the ammeter reading to become zero was measured. The results are shown in Fig. 6.2.

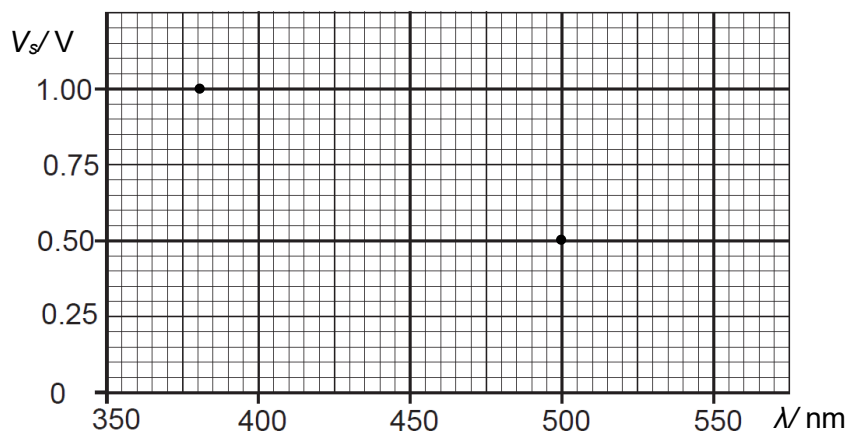


Fig. 6.2

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- (i) Calculate the maximum kinetic energy of a photoelectron emitted from the metal surface by radiation of wavelength 500 nm.

maximum kinetic energy = J [2]

- (ii) Calculate the energy of a photon of wavelength 500 nm.

energy = J [2]

- (iii) Hence, determine

1. the work function energy of the metal surface, and

work function energy = J [1]

2. the maximum photon wavelength that can liberate a photoelectron.

maximum wavelength = nm [2]

- (iv) Suggest why it is not possible to deduce the maximum photon wavelength that can liberate a photoelectron from this metal plate directly from the data in Fig. 6.2.

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