

**30** Technetium-99 has a half life of 6.0 hrs. It is used as a biological tracer. A sample of technetium-99 with an activity of  $8.0 \times 10^{10}$  Bq is injected into the bloodstream of a patient. 20 hours later, when the technetium-99 is assumed to be uniformly distributed throughout the blood of the patient, a  $10 \text{ cm}^3$  blood sample is obtained. The activity from the sample is found to be  $1.6 \times 10^7$  Bq.

What is the total volume of blood in the patient?

- A**     $3100 \text{ cm}^3$     **B**     $4700 \text{ cm}^3$     **C**     $5000 \text{ cm}^3$     **D**     $6300 \text{ cm}^3$

**End of paper**