

- 11 An ideal gas is confined in a cylinder fitted with a frictionless piston. When 90 J of heat is supplied and the gas is allowed to expand $2.5 \times 10^{-4} \text{ m}^3$ in volume at a constant pressure of $2.0 \times 10^5 \text{ Pa}$, the temperature rises by 4.5 K.

What is the increase in internal energy during this process?

A 40 J

B 50 J

C 90 J

D 140 J

