

- 11** The number of hydrogen molecules per cm^3 in a container is 6.80×10^{15} . The root-mean-square speed of the molecules is $1.90 \times 10^3 \text{ m s}^{-1}$. The mass of one mole of hydrogen molecule is 2.02 g.

What is the pressure exerted by the gas in the container?

- A** 0.014 Pa
- B** 26 Pa
- C** 27 Pa
- D** 82 Pa