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In a heating experiment, energy is supplied at a constant rate to a liquid. The temperature of the liquid rises at 4.0 K per minute just before it begins to boil.

After 40 minutes of boiling, all the liquid has boiled away.

For this liquid, what is the ratio  $\frac{\text{specific latent heat of vaporisation}}{\text{specific heat capacity}}$ ?

**A**

$$\frac{1}{160} \text{ K}$$

**B**

$$\frac{1}{40} \text{ K}$$

**C**

$$40 \text{ K}$$

**D**

$$160 \text{ K}$$