

- 9 (a) An electron is travelling in a vacuum towards an electrode with kinetic energy of $8.55 \times 10^{-19} \text{ J}$.

Calculate the stopping potential V_s required to stop the electron.

$$V_s = \dots\dots\dots \text{ V [2]}$$

- (b) (i) The electron in (a) is emitted from a material whose work function is 2.80 eV. Calculate the wavelength of the radiation responsible for causing the emission of the electron.

$$\text{wavelength} = \dots\dots\dots \text{ m [2]}$$

- (ii) Suggest the type of radiation which has the wavelength in (b)(i).

$$\text{type of radiation} = \dots\dots\dots [1]$$

- (c) (i) Calculate the de Broglie wavelength of an electron travelling with speed $1.85 \times 10^7 \text{ m s}^{-1}$.

$$\text{wavelength} = \dots\dots\dots \text{ m [2]}$$

- (ii) Graphite, with its layered structure as shown in Fig. 9.1, acts as a natural diffraction grating when used in electron diffraction experiments. The distance between each layer of graphite is 0.335 nm.

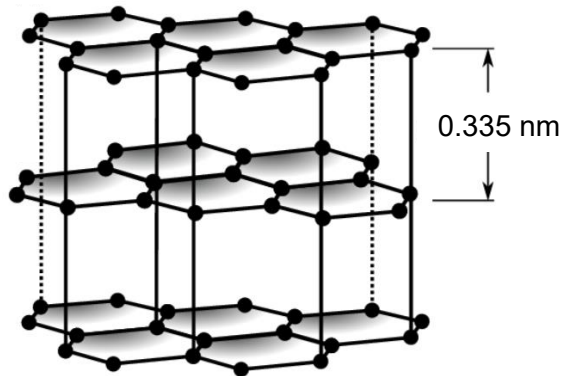


Fig. 9.1

Explain whether electrons having the speed of $1.85 \times 10^7 \text{ m s}^{-1}$ can be used to demonstrate electron diffraction.

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[2]

- (d) Tungsten, a transition metal, is commonly used as a target metal to produce X-rays. The energy levels of the K- to M-shells for tungsten are shown in Fig. 9.2 below.

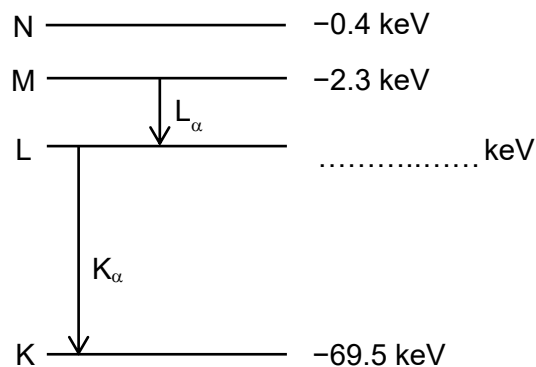


Fig. 9.2 (not to scale)

The wavelength of the photon produced by the K_{α} transition is 21.2 pm.

- (i) Complete Fig. 9.2 by filling in the energy level of the L-shell for tungsten. Show your working clearly.

- (ii) The intensity of various photon wavelengths from electron bombardment of a tungsten target metal is shown in Fig. 9.3. The peak representing K_α transition is labelled.

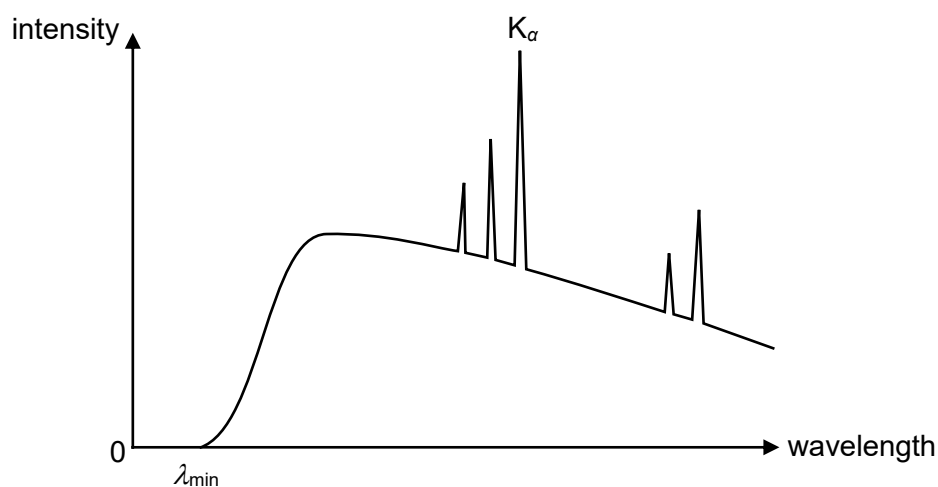


Fig. 9.3

1. On Fig. 9.3, label the peak for L_α transitions. [1]
2. Explain the existence of a minimum wavelength λ_{\min} .

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.....[2]

- (iii) With reference to Fig. 9.2, state the minimum energy of the bombarding electrons to produce the characteristic X-rays lines shown in Fig. 9.3.

minimum energy = keV [1]

- (iv) Explain your answer in (d)(iii).

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.....[1]

(e) Fig. 9.4 below shows a typical setup for producing such X-ray beams.

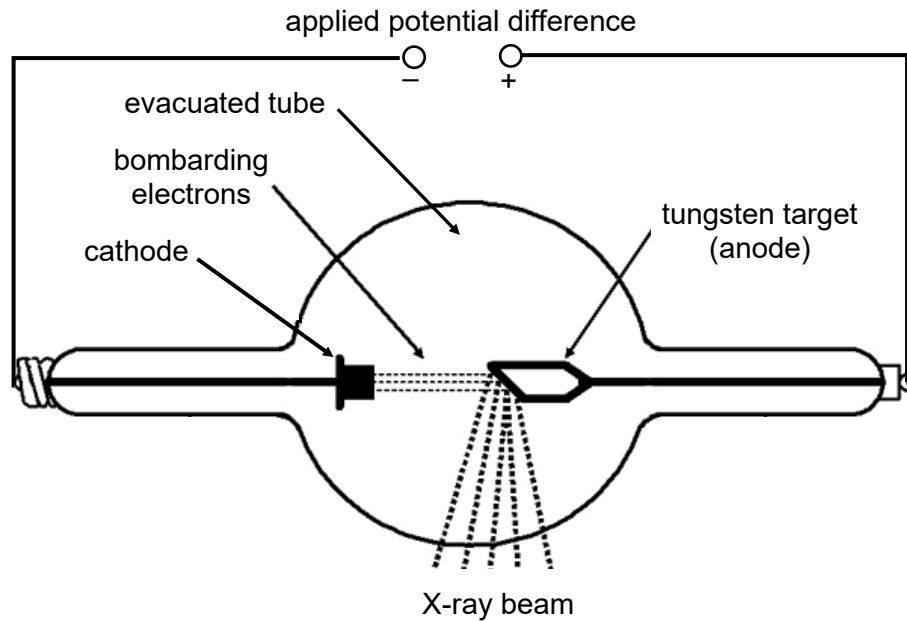


Fig. 9.4

- (i) For safety reasons, the wavelength of radiation used for medical X-rays should not be shorter than 50 pm.

Suggest why the wavelength of X-rays radiation should not be shorter than 50 pm.

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[1]

- (ii) Determine the minimum applied potential difference for medical X-rays.

minimum potential difference = V [2]