

- 30** The momentum of an electron is measured to be  $6.42 \times 10^{-23} \text{ kg m s}^{-1}$  with a percentage uncertainty of 1.00 %.

What is the minimum uncertainty in its position?

- A**  $8.07 \times 10^{-24} \text{ m}$
- B**  $1.03 \times 10^{-9} \text{ m}$
- C**  $2.06 \times 10^{-9} \text{ m}$
- D**  $6.50 \times 10^{-9} \text{ m}$

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