

- 6 (a) State what is meant by the term *threshold frequency* as applied to the photoelectric effect.

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- (b) In a typical set-up of the photoelectric experiment, a metal surface is illuminated with radiation of wavelength 450 nm, causing the emission of photoelectrons which are collected at an adjacent electrode.

- (i) Calculate the energy of a photon incident on the surface.

energy = J [2]

- (ii) The intensity of the incident radiation is $2.7 \times 10^3 \text{ W m}^{-2}$ and the area of the metal surface is 3.0 cm^2 .

Calculate the number of photons incident per second on the surface.

number per second = [2]

(iii) Fig. 6.1 shows a graph of how the photoelectric current I varies with the potential difference V between the electrodes.

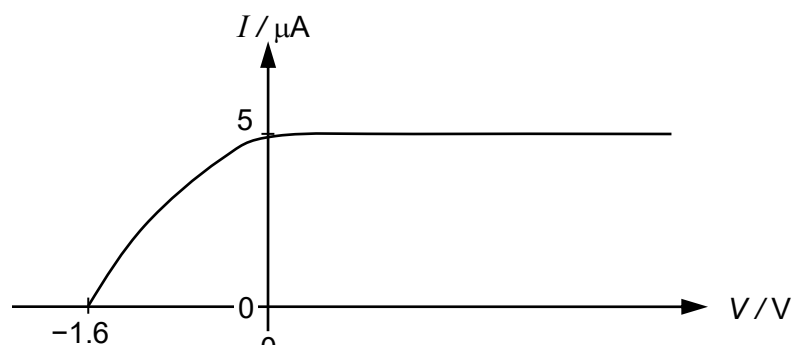


Fig. 6.1

Calculate the threshold wavelength of the metal.

wavelength = m [3]

- (c) The X-ray spectrum is first produced by an X-ray tube with tungsten (atomic number, $Z = 74$). Another X-ray spectrum is produced using barium (atomic number, $Z = 56$) and both spectrums are as shown in Fig. 6.2.

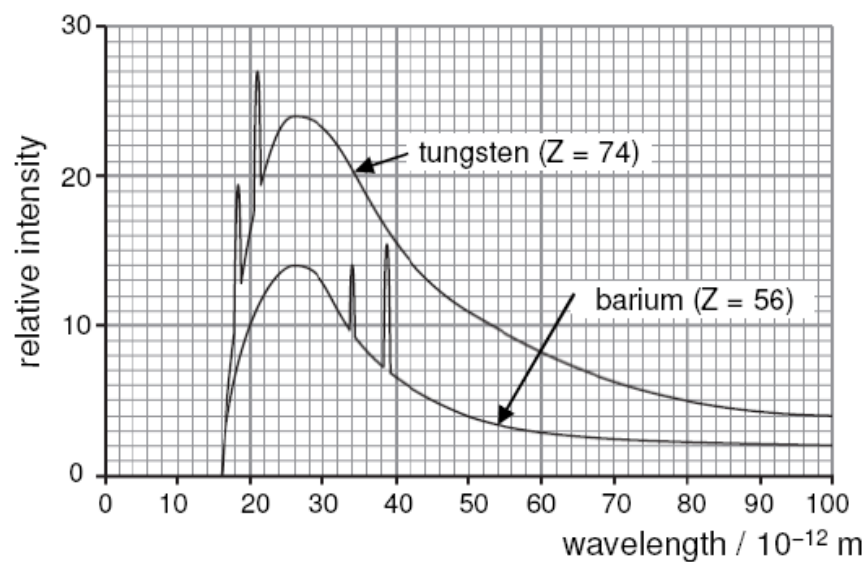


Fig. 6.2

- (i) The accelerating potential used to produce the X-ray spectra using tungsten and barium are the same.

State a feature in Fig. 6.2 that shows how this can be deduced.

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- (ii) Determine the accelerating potential.

accelerating potential = V [2]

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