

- 6 (a) State what is meant by the term *threshold frequency* as applied to the photoelectric effect.

.....

[1]

- (b) In a typical set-up of the photoelectric experiment, a metal surface is illuminated with radiation of wavelength 450 nm, causing the emission of photoelectrons which are collected at an adjacent electrode.

- (i) Calculate the energy of a photon incident on the surface.

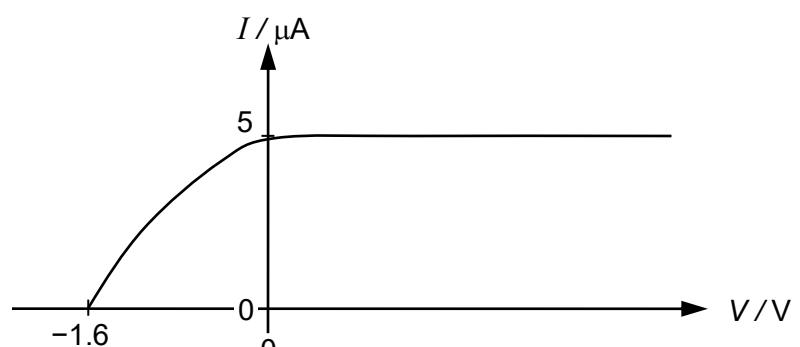
energy = ..... J [2]

- (ii) The intensity of the incident radiation is  $2.7 \times 10^3 \text{ W m}^{-2}$  and the area of the metal surface is  $3.0 \text{ cm}^2$ .

Calculate the number of photons incident per second on the surface.

number per second = ..... [2]

- (iii) Fig. 6.1 shows a graph of how the photoelectric current  $I$  varies with the potential difference  $V$  between the electrodes.



**Fig. 6.1**

Calculate the threshold wavelength of the metal.

wavelength = ..... m [3]

- (c) The X-ray spectrum is first produced by an X-ray tube with tungsten (atomic number,  $Z = 74$ ). Another X-ray spectrum is produced using barium (atomic number,  $Z = 56$ ) and both spectrums are as shown in Fig. 6.2.

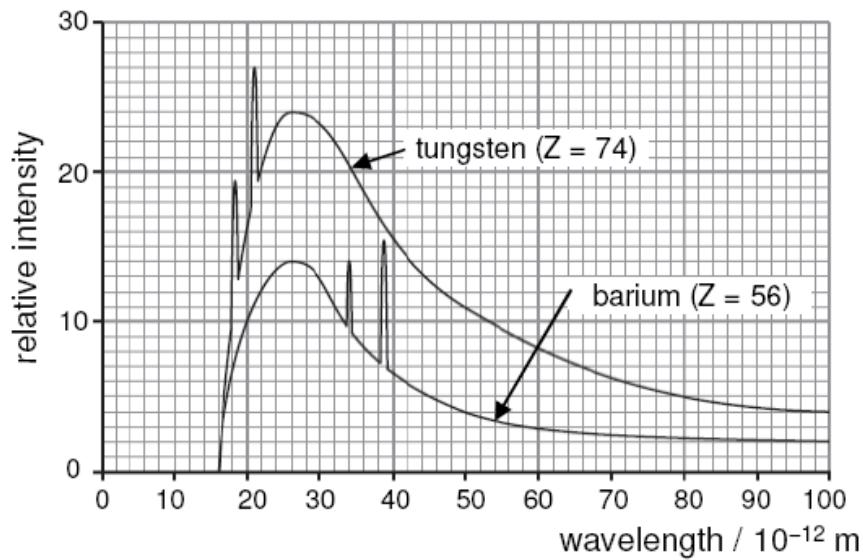


Fig. 6.2

- (i) The accelerating potential used to produce the X-ray spectra using tungsten and barium are the same.

State a feature in Fig. 6.2 that shows how this can be deduced.

..... [1]

- (ii) Determine the accelerating potential.

accelerating potential = ..... V [2]

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