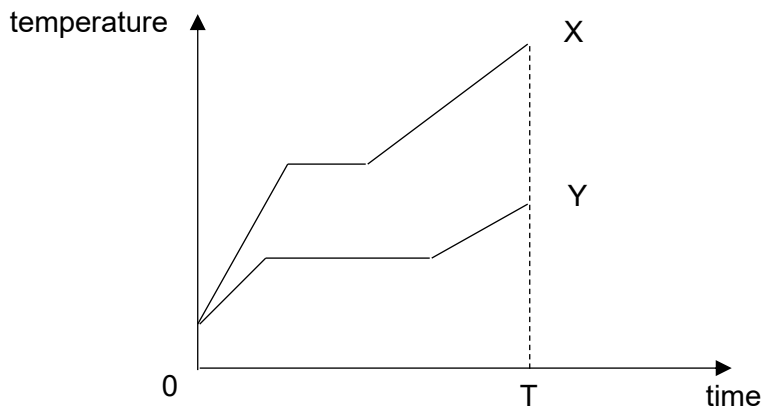


- 15** Equal masses of two solids X and Y are heated from the same temperature. Energy is supplied by heating at the same constant rate to each solid.

The graph shows how the temperature of each solid varies with time after heating starts.



What can be deduced from the graph?

- A** X has the higher boiling point.
- B** X has the larger specific latent heat of fusion.
- C** When X and Y are solids, Y has the larger specific heat capacity.
- D** At time T, X gained more internal energy than Y.