

28 An electron moves with a constant velocity of $1.5 \times 10^6 \text{ m s}^{-1}$. If its momentum is measured to a precision of 0.2%, what is the uncertainty associated with its position?

A $1.3 \times 10^{-12} \text{ m}$

B $1.3 \times 10^{-10} \text{ m}$

C $2.4 \times 10^{-9} \text{ m}$

D $2.4 \times 10^{-7} \text{ m}$