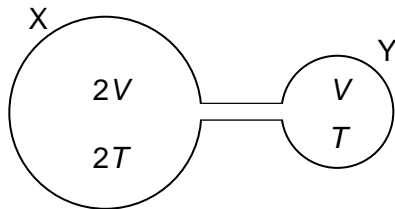


- 12** An ideal gas is contained in two spherical containers X and Y of volume $2V$ and V respectively, connected by a hollow tube of negligible volume. The containers X and Y are maintained at temperatures $2T$ and T respectively. The setup is shown in the diagram below.



What is the ratio $\frac{\text{number of moles of gas in container X}}{\text{number of moles of gas in container Y}}$?

A 0.25

B 1

C 2

D 4