

- 28** When electromagnetic radiation falls on a particular metal surface, photoelectrons may be emitted. The table below gives the maximum kinetic energy of the photoelectrons emitted at a given frequency.

frequency of incident radiation	maximum kinetic energy of photoelectrons
$1.0 \times 10^{15} \text{ Hz}$	$2.6 \times 10^{-19} \text{ J}$
$2.5 \times 10^{15} \text{ Hz}$	E

What is the value of E ?

- A** $1.3 \times 10^{-18} \text{ J}$ **B** $9.9 \times 10^{-19} \text{ J}$ **C** $7.3 \times 10^{-19} \text{ J}$ **D** $6.5 \times 10^{-19} \text{ J}$

