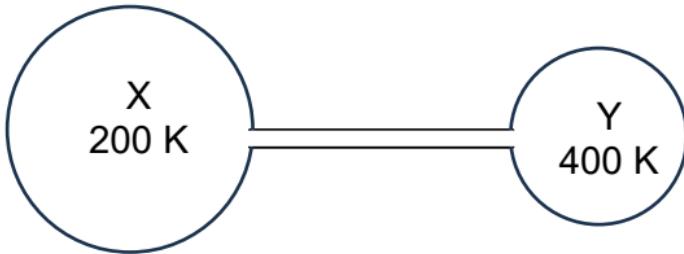


- 10 In the diagram, the volume of bulb X is twice that of bulb Y. The system is filled with an ideal gas and a steady state is established with the bulbs X and Y at temperature 200 K and 400 K respectively.



If the mass of gas in X is m , what is the mass of gas in Y?

A $\frac{m}{4}$

B $\frac{m}{2}$

C m

D $4m$

- 11 One mole of ideal gas is contained within a cylinder by a frictionless piston and is initially at