

- 6 (a) To produce X-rays, a metal target inside an X-ray tube is bombarded with high speed electrons that have been accelerated over a large potential difference.

Fig. 6.1 shows the variation with wavelength of the emitted radiation from the tube of the X-ray intensity. It consists of a continuous spectrum with two sharp peaks, labelled  $K_\alpha$  and  $K_\beta$ . The X-ray spectrum cuts off at a minimum wavelength of  $\lambda_0$ .

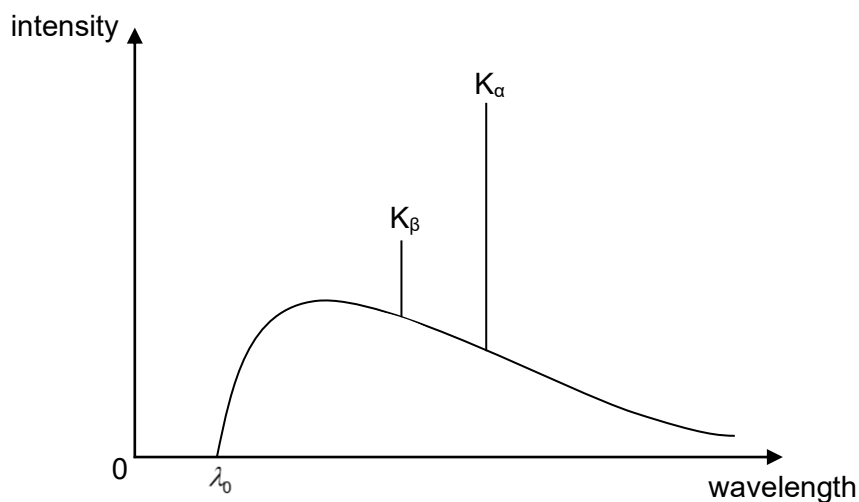


Fig. 6.1

- (i) Explain the formation of the continuous spectrum.

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..... [3]

- (ii) Determine  $\lambda_0$  if the electrons are accelerated over a potential difference of 50 kV.

$$\lambda_0 = \dots\dots\dots \text{ m} \quad [2]$$

(iii) Besides the  $K_\alpha$  and  $K_\beta$  peaks, two other peaks  $L_\alpha$  and  $L_\beta$  can also be observed on the X-ray spectrum in Fig 6.1.

(b) The  $L_\alpha$  and  $L_\beta$  peaks are due to photons produced from electronic transitions within the metal atoms from energy levels  $n = 3$  to  $n = 2$ , and  $n = 4$  to  $n = 2$  respectively. In a photoelectric experiment, a metal target in a vacuum tube is bombarded with electromagnetic radiation.

Fig. 6.2 shows the variation with the frequency  $f$  of the incident radiation of the stopping potential  $V_s$ . Sketch the  $L_\alpha$  and  $L_\beta$  peaks on Fig. 6.1. Label the two peaks clearly. [2]

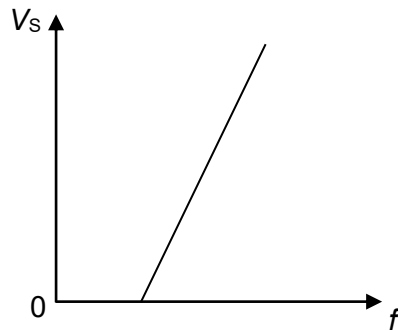


Fig. 6.2

(i) Determine, with clear explanations, the gradient of the graph.

$$\text{gradient} = \dots\dots\dots [2]$$

(ii) When potassium was used as the metal target, it was found that no photocurrent was generated when the frequency of the incident electromagnetic radiation was below  $5.55 \times 10^{14} \text{ Hz}$ .

Determine the work function energy of potassium.

$$\text{work function energy} = \dots\dots\dots \text{ eV} \quad [2]$$

- (iii) On Fig. 6.2, sketch a graph to represent the variation with  $f$  of  $V_s$  if tungsten was used as the metal target instead. The work function energy of tungsten is 4.50 eV.

[1]

**End Section A**  
**Section B**

Answer **one** question from this Section in the spaces provided.