

28 An electron moves with a constant velocity of $1.5 \times 10^6 \text{ m s}^{-1}$. If its momentum is measured to a precision of 0.2%, what is the uncertainty associated with its position?

A $1.2 \times 10^{-9} \text{ m}$

B $2.4 \times 10^{-9} \text{ m}$

C $2.4 \times 10^{-7} \text{ m}$

D $1.2 \times 10^{-6} \text{ m}$