

- 13** A  $350 \text{ cm}^3$  cup of hot tea takes 30 minutes to cool down from an initial temperature of  $90^\circ\text{C}$  to  $30^\circ\text{C}$ .

The specific heat capacity of tea is  $4200 \text{ J kg}^{-1} \text{ K}^{-1}$  and its density is  $1000 \text{ kg m}^{-3}$ .

What is the average rate of heat loss of the tea?

- A**  $49 \text{ W}$       **B**  $2900 \text{ W}$       **C**  $4.9 \times 10^4 \text{ W}$       **D**  $2.9 \times 10^6 \text{ W}$