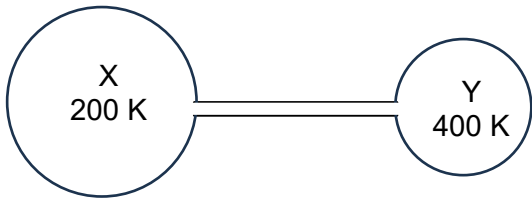


- 10** In the diagram, the volume of bulb X is twice that of bulb Y. The system is filled with an ideal gas and a steady state is established with the bulbs X and Y at temperature 200 K and 400 K respectively.



If the mass of gas in X is  $m$ , what is the mass of gas in Y?

- A**  $\frac{m}{4}$
- B**  $\frac{m}{2}$
- C**  $m$
- D**  $4m$

- 11** One mole of ideal gas is contained within a cylinder by a frictionless piston and is initially at