

**28** An electron moves with a constant velocity of  $1.5 \times 10^6 \text{ m s}^{-1}$ . If its momentum is measured to a precision of 0.2%, what is the uncertainty associated with its position?

**A**  $1.3 \times 10^{-12} \text{ m}$

**B**  $1.3 \times 10^{-10} \text{ m}$

**C**  $2.4 \times 10^{-9} \text{ m}$

**D**  $2.4 \times 10^{-7} \text{ m}$