

- 29** In the lungs, there are tiny sacs of air known individually as alveolus. The average diameter of an alveolus is 0.250 mm. Consider an oxygen molecule of mass  $5.30 \times 10^{-26}$  kg that is trapped in an alveolus.

What is the order of magnitude of the uncertainty in the velocity of this oxygen molecule?

- A**  $10^{-5} \text{ m s}^{-1}$
- B**  $10^{-8} \text{ m s}^{-1}$
- C**  $10^{-10} \text{ m s}^{-1}$
- D**  $10^{-12} \text{ m s}^{-1}$