

- 6 The energy spectrum of the beta particles emitted from an initially stationary parent nuclei is shown in Fig. 6.1.

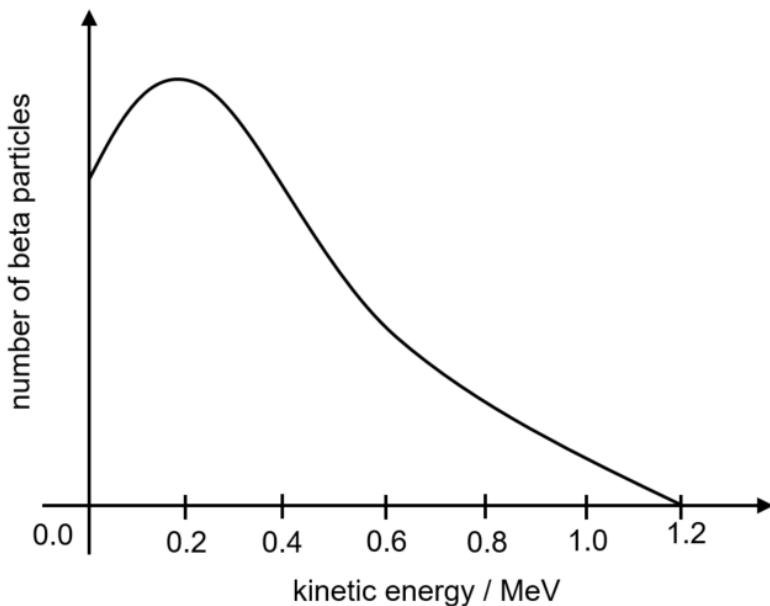


Fig. 6.1

- (a) Explain why a continuous spectrum of kinetic energy for the beta particles is obtained.

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[3]

- (b) The radioisotope Bismuth $^{210}_{83}\text{Bi}$ undergoes beta decay into a stable isotope of Polonium (Po).

- (i) Write down a nuclear equation that describes the above nuclear reaction.

[1]

- (ii) Using data from Fig. 6.1 and Fig. 6.2, determine the mass of the Polonium nucleus in terms of u .

particle	mass / u
$^{210}_{83}\text{Bi}$ nucleus	209.939
proton	1.00729
neutron	1.00867
electron	0.000548795
neutrino	negligible

Fig. 6.2

mass of polonium nucleus = u [4]

(iii) Research has shown that stable isotopes of heavy elements have an optimal neutron-to-

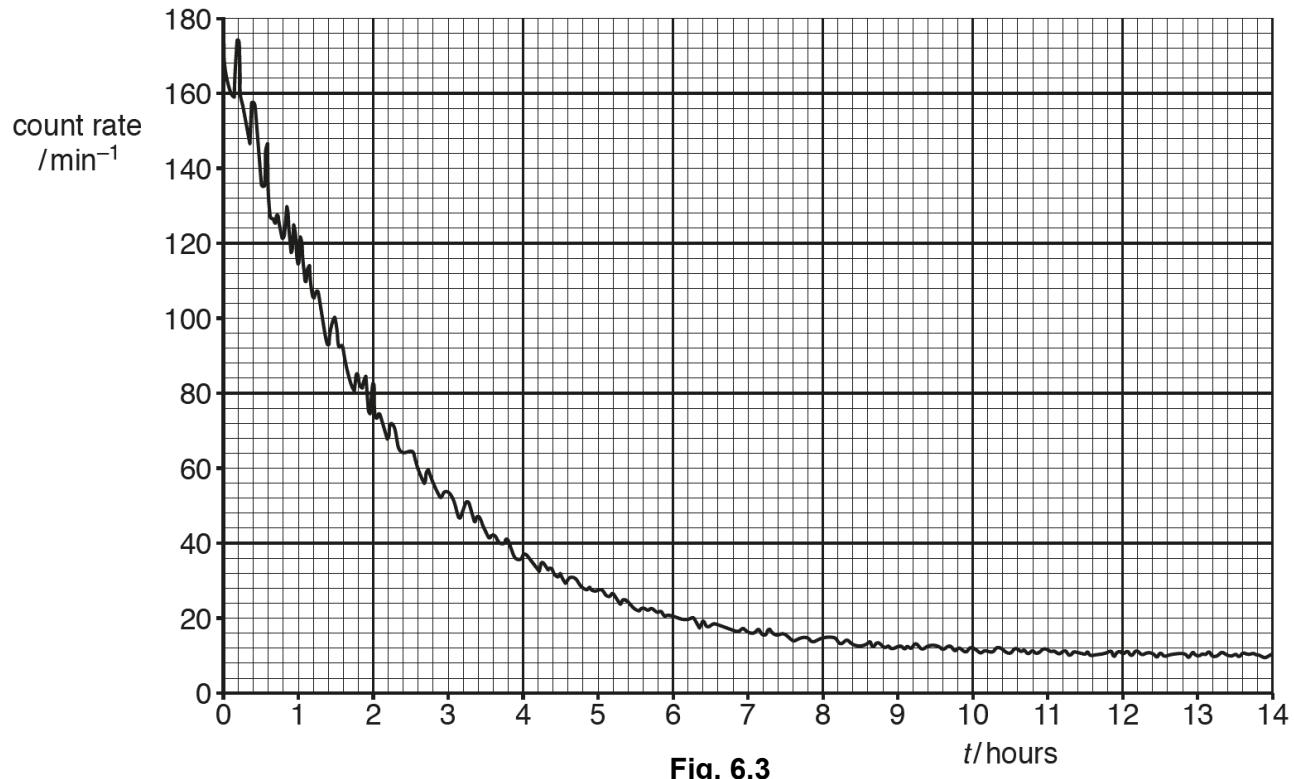
proton ratio and lies on a “belt of stability”. Unstable isotopes will undergo transmutation through radioactive decay such that the resulting nucleus achieves this optimal ratio.

Using your answer in (b)(i) or otherwise, suggest if Bismuth-210 has an excess of neutrons or protons, as compared to the optimal ratio.

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[2]

(c) A Geiger-Muller tube used in an experiment is positioned at a fixed distance from a radioactive sample to measure the variation of count rate with time, as shown in Fig. 6.3.



(i) Explain why the data points on the graph do not lie on a smooth curve.

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[1]

- (ii) Suggest a reason why the count rate recorded is not equal to the activity of the radioactive sample.

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[1]

[Total: 12]

Section B

Answer any **one** question in this Section in the spaces provided.