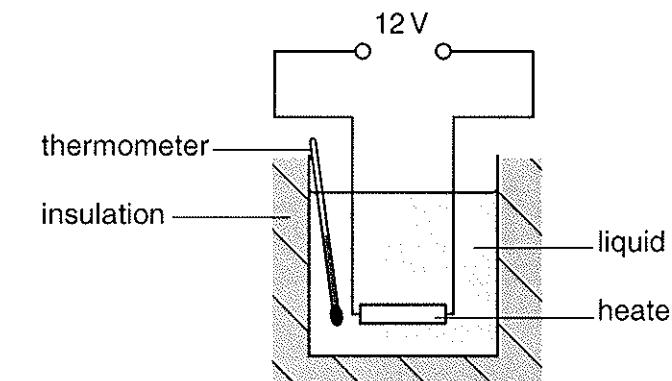




- 2 (a)** A student uses the apparatus shown in Fig. 2.1 to determine the specific heat capacity of a liquid.

For  
Examiner's  
Use



**Fig. 2.**

A power supply of potential difference 12V is connected to a heater of resistance  $2.4\Omega$ . The mass of liquid is 140g. The temperature of the liquid changes from  $25^\circ\text{C}$  to  $45^\circ\text{C}$  in 210s.

## Calculator

- (i) the power of the heat

power = ..... W [2]

- (ii) the student's value for the specific heat capacity of the liquid.

specific heat capacity = ..... J kg<sup>-1</sup>K<sup>-1</sup> [3]

- (b) The accepted value for the specific heat capacity of this liquid is  $4200 \text{ J kg}^{-1} \text{ K}^{-1}$ . The uncertainties in the measurements in (a) cannot account for the difference. Suggest a cause for this difference.

