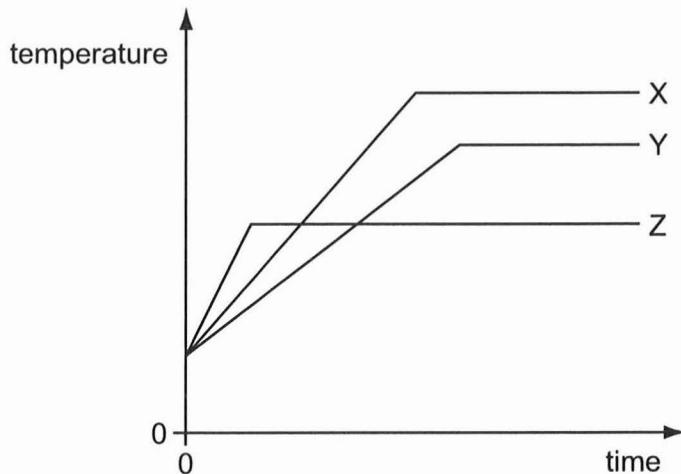


- 18 Equal masses of three liquids X, Y and Z are heated from room temperature. Energy is supplied by heating at the same rate to each liquid.

The graph shows how the temperature of each liquid varies with time after heating starts.



What can be deduced from the graph?

- A X has the highest melting point.
- B X gains the most internal energy.
- C Y has the largest specific heat capacity.
- D Z has the smallest specific latent heat of vaporisation.