

- 20** An ideal gas undergoes an expansion in volume from  $1.3 \times 10^{-4} \text{ m}^3$  to  $3.6 \times 10^{-4} \text{ m}^3$  at a constant pressure of  $1.3 \times 10^5 \text{ Pa}$ . During this expansion, 24 J of heat is supplied to the gas.

What is the overall change in the internal energy of the gas?

- A** decrease of 54 J
  - B** decrease of 6 J
  - C** increase of 6 J
  - D** increase of 54 J
- 21** A long horizontal tube containing a fine powder is closed at one end. A loudspeaker connected