

- 12 A school laboratory has dimensions $10\text{ m} \times 3.0\text{ m} \times 4.0\text{ m}$. The laboratory contains 5000 moles of air, with a molar mass of 29 g, at an atmospheric pressure of $1.0 \times 10^5\text{ Pa}$.

What is the root-mean-square (r.m.s.) speed of the gas molecules in the air?

A 16 ms^{-1}

B 290 ms^{-1}

C 500 ms^{-1}

D $2.5 \times 10^5\text{ ms}^{-1}$

