



- 2 (a) A student uses the apparatus shown in Fig. 2.1 to determine the specific heat capacity of a liquid.

For
Examiner's
Use

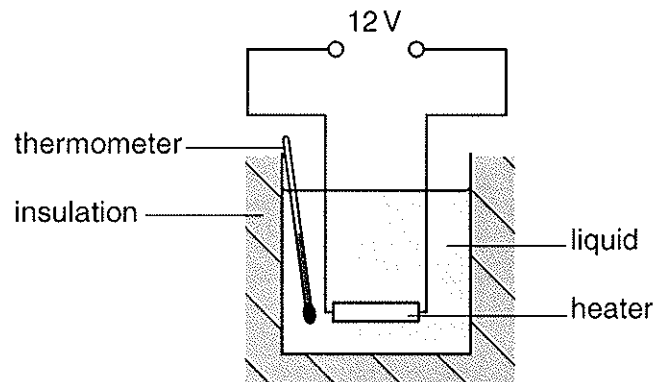


Fig. 2.1

A power supply of potential difference 12V is connected to a heater of resistance 2.4Ω . The mass of liquid is 140g. The temperature of the liquid changes from 25°C to 45°C in 210s.

Calculate

- (i) the power of the heater,

power = W [2]

- (ii) the student's value for the specific heat capacity of the liquid.

specific heat capacity = $\text{J kg}^{-1} \text{K}^{-1}$ [3]

- (b) The accepted value for the specific heat capacity of this liquid is $4200 \text{ J kg}^{-1} \text{K}^{-1}$. The uncertainties in the measurements in (a) cannot account for the difference. Suggest a cause for this difference.

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 [1]

