

Section A

Answer all the questions in this section.

- 1 (a) State what is meant by *latent heat of vaporisation*.

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 [2]

- (b) A student carries out an experiment to determine the specific latent heat of vaporisation of water.

Water is boiled in a beaker by means of an electric heater, as shown in Fig. 1.1.

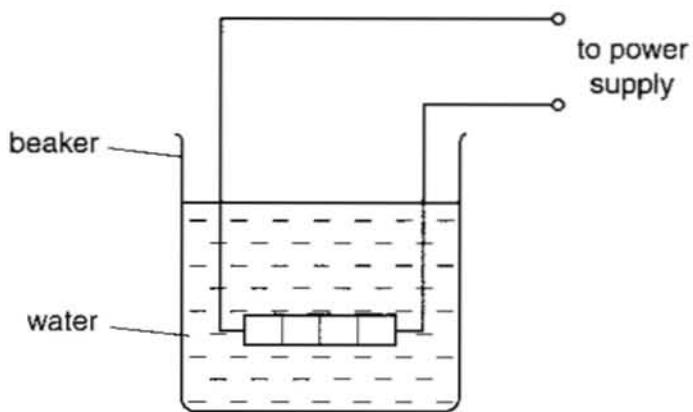


Fig. 1.1

The power P supplied to the heater is measured. When the water is boiling at a constant rate, the mass m of water evaporated in 5.0 minutes is determined.

Data for the power P and the mass m for two different values of P are shown in Fig. 1.2.

| P/W | m/g |
|-------|-------|
| 140 | 14.1 |
| 95 | 8.2 |

Fig. 1.2

- (i) Suggest why, in order to obtain a reliable result for the specific latent heat, the mass m is determined for two different values of P .

.....
.....

[1]

- (ii) Calculate the value for the specific latent heat L of vaporisation.

$$L = \dots \text{ J g}^{-1} [3]$$