

30 The iron isotope Fe-56 has an atomic number of 26 and a nucleon number of 56.

The atom of the isotope has a mass of $55.93493\text{ }u$.

What is the mass defect of Fe-56?

(mass of proton = $1.00728\text{ }u$, mass of neutron = $1.00867\text{ }u$, mass of electron = $0.00055\text{ }u$)

A $8.45 \times 10^{-28}\text{ kg}$

B $8.54 \times 10^{-28}\text{ kg}$

C $8.72 \times 10^{-28}\text{ kg}$

D $8.78 \times 10^{-28}\text{ kg}$

