

- 20 An ideal gas undergoes an expansion in volume from  $1.3 \times 10^{-4} \text{ m}^3$  to  $3.6 \times 10^{-4} \text{ m}^3$  at a constant pressure of  $1.3 \times 10^5 \text{ Pa}$ . During this expansion, 24 J of heat is supplied to the gas.

What is the overall change in the internal energy of the gas?

- A decrease of 54 J
- B decrease of 6 J
- C increase of 6 J
- D increase of 54 J

- 21 A long horizontal tube, containing a fine powder, is closed at one end. A loudspeaker connected