

Historical Introduction to the Elementary Particles

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The Classical Era (1897-1932)



- The foundation of atomic theory dates back to ancient Greece (Democritus)
- The true birth of modern elementary particle physics began in 1897 with the discovery of the electron by J. J. Thomson
- The foundations were laid through successive discoveries shaping atomic theory

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Discovery of Electron (1897)



- Cathode ray experiments showed particles with a negative charge.
- Measurement of the charge-to-mass ratio indicated a very small mass.

The term electron, first introduced by George Johnstone Stoney (1891), was

later adopted.

Atomic theory

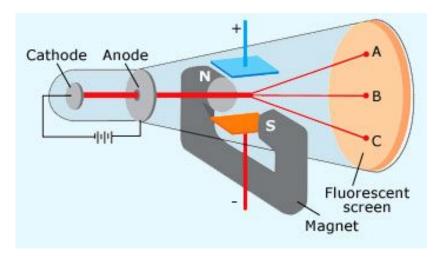


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Rutherford's Gold Foil Experiment (1909)



 Alpha particles were expected to pass through a thin gold foil with minimal deflection.

Some particles were deflected at large angles, indicating a dense central

nucleus.

This led to the nuclear model of the atom.

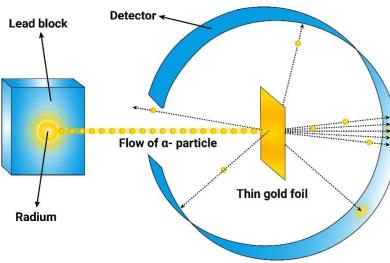


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Discovery of Proton & Neutron



- 1917: Rutherford identified the proton as a fundamental particle.
- Concluded Hydrogen nuclei present in other atoms
- 1932: James Chadwick discovered the **neutron**, resolving inconsistencies in atomic mass calculations. eg(He & Li)
- With electrons, protons, and neutrons, the classical atomic model was complete.

Introduction to the Photon (1900-1924) IISER MOHALI

- Unlike other particles, the photon was not discovered by a single individual.
- Contributions from Planck and Einstein helped establish the concept of quantized light energy.

Planck's Contribution



- Ultraviolet Catastrophe: Classical physics predicted that a blackbody should emit infinite energy at high frequencies, which was clearly incorrect.
- Max Planck (1900) solved this by introducing energy quantization (E = hv), assuming radiation is emitted in discrete packets.
- His assumption led to the foundation of quantum mechanics.

Einstein's Radical Interpretation



- Extended the Idea of Max Planck
- Photoelectric Effect: Light behaves as discrete packets (photons) rather than continuous waves.
- Not Intensity but Frequency
- Einstein's Equation
- Significance: This provided the foundation for quantum mechanics and confirmed the particle-like nature of light.

Millikan & Compton



- Millikan (1916) confirmed Einstein's equation but resisted the photon theory.
- Compton Scattering (1923) demonstrated photon momentum with the equation:

$$\lambda' = \lambda + rac{h}{mc}(1-\cos heta)$$

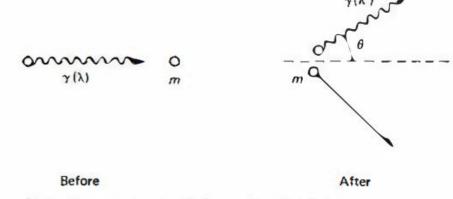


Fig. 1.2 Compton scattering. A photon of wavelength λ scatters off a particle, initially at rest, of mass m. The scattered photon carries wavelength λ' given by Equation 1.4.

Mesons and Strong Force (1934-1947)



- The big question: What holds the nucleus together?
- The presence of multiple positively charged protons should cause repulsion.
- A powerful but short-ranged force was postulated: the strong nuclear force.

Yukawa's Theory



- Yukawa (1934) theorized that a massive particle mediates the strong force, similar to how the photon mediates electromagnetism.
- Forces are not just fields: In quantum mechanics, forces arise from the exchange of particles rather than continuous fields.
- Range of the Strong Force: Given by the equation:

Range~ħ/mc

- He predicted a meson with a mass ~300 times the electron's mass.
- Anderson & Neddermeyer (1937) discovered a particle with the expected mass in cosmic rays

The Crisis-Something Went Wrong



- Initial cosmic ray discoveries showed meson-like particles, but they did not interact as expected.
- These particles had incorrect lifetimes and did not strongly interact with nuclei.

First Resolution-Pion & Muon



Further studies revealed **two different particles**:

- **Pion** (π -meson): The true Yukawa meson, mediating the strong force.
- Muon (μ-meson): A separate particle behaving like a heavy electron, unrelated to the strong force.

End of an Early Era



- The discovery of pions and muons refined our understanding of particle interactions.
- This set the stage for further discoveries leading to the Standard Model of particle physics.