TABLE 2-153 Heat Capacities of Inorganic and Organic Liquids [J/(kmol·K)]

	2-153 Heat Capacities of I	norganic c	Tria Organic	ridoias [1	/ (KIIIOI·K)]	ı		I					
Cmpd. no.	Name	Formula	CAS no.	Mol. wt.	C1	C2	C3	C4	C5	$T_{\min}$ , K	$C_p$ at $T_{\min}$ $\times$ 1E-05	$T_{\rm max}$ , K	$C_p$ at $T_{ma}$ × 1E-05
1	Acetaldehyde	C <sub>2</sub> H <sub>4</sub> O	75-07-0	44.053	115,100	-433	1.425			150.15	0.8221	294.00	1.1097
2	Acetamide	$C_2H_5NO$	60-35-5	59.067	102,300	128.7				354.15	1.4788	571.00	1.7579
3	Acetic acid	$C_2H_4O_2$	64-19-7	60.052	139,640	-320.8	0.8985			289.81	1.2213	391.05	1.5159
4	Acetic anhydride	$C_4H_6O_3$	108-24-7	102.089	36,600	511	0.5055			250.00	1.6435	350.00	2.1545
5	Acetone	$C_3H_6O$	67-64-1	58.079	135,600	-177	0.2837	0.000689		178.45	1.1696	329.44	1.3271
6	Acetonitrile	$C_2H_3N$	75-05-8	41.052	97,582	-122.2	0.34085	0.00000		229.32	0.8748	354.75	0.9713
7	Acetylene	$C_2H_2$	74-86-2	26.037	-122,020	3,082.7	-15.895	0.027732		192.40	0.8021	250.00	0.8853
8	Acrolein	$C_3H_4O$	107-02-8	56.063	103,090	-247.8	1.0343	0.021102		253.00	1.0660	379.50	1.5801
9	Acrylic acid	$C_3H_4O_2$	79-10-7	72.063	55,300	300	1.0010			286.15	1.4114	375.00	1.6780
10	Acrylonitrile	$C_3H_4O_2$ $C_3H_3N$	107-13-1	53.063	109,900	-109.75	0.35441			189.63	1.0183	400.00	1.2271
11	Air	Mixture	132259-10-0	28.960	-214,460	9,185.1	-106.12	0.41616		75.00	0.5307	115.00	0.7132
12	Ammonia [use Eq. (2)]	H <sub>3</sub> N	7664-41-7	17.031	61.289	80,925	799.4	-2,651		203.15	0.7575	401.15	4.1847
13	Anisole	C <sub>7</sub> H <sub>8</sub> O	100-66-3	108.138	150,940	93.455	0.23602	2,001		298.15	1.9978	484.20	2.5153
14	Argon	Ar	7440-37-1	39.948	134,390	-1,989.4	11.043			83.78	0.4523	135.00	0.6708
15	Benzamide	C <sub>7</sub> H <sub>7</sub> NO	55-21-0	121.137	161,440	260.66	11.040			403.00	2.6649	563.15	3.0823
16	Benzene	$C_6H_6$	71-43-2	78.112	129,440	-169.5	0.64781			278.68	1.3251	353.24	1.5040
16	Benzene	$C_6H_6$	71-43-2	78.112	162,940	-344.94	0.85562			278.68	1.3326	500.00	2.0437
17	Benzenethiol	$C_6H_6S$	108-98-5	110.177	119,780	180.34	0.05502			258.27	1.6636	442.29	1.9954
18	Benzoic acid	$C_6H_6O_2$	65-85-0	122.121	-5,480	647.12				395.45	2.5042	450.00	2.8572
19	Benzonitrile	$C_7H_6O_2$ $C_7H_5N$	100-47-0	103.121	93,383	242.61				260.40	1.5656	464.15	2.0599
20	Benzophenone	$C_{7}H_{5}N$ $C_{13}H_{10}O$	119-61-9	182.218	156,130	454.49				321.35	3.0218	640.00	4.4700
21	Benzyl alcohol	$C_{13}H_{10}O$ $C_7H_8O$	100-51-6	108.138	-334,997	3,644.21	-7.77514	0.00591102		257.85	1.8905	478.60	2.7617
22	Benzyl ethyl ether	$C_{9}H_{12}O$	539-30-0	136.191	87,500	480	-1.11514	0.00591102		275.65	2.1981	458.15	3.0741
23	Benzyl mercaptan	$C_9H_{12}O$ $C_7H_8S$	100-53-8	124.203	100,320	346.89				243.95	1.8494	472.03	2.6406
24	Biphenyl	$C_{12}H_{10}$	92-52-4	154.208	121,770	429.3				342.20	2.6868	533.37	3.5075
25	Bromine	C <sub>12</sub> Π <sub>10</sub>	7726-95-6	159.808	179,400	-667.11	1.0701			265.90	0.7768	331.90	0.7587
26	Bromobenzene	$\mathrm{Br}_2$ $\mathrm{C}_6\mathrm{H}_5\mathrm{Br}$	108-86-1	157.008	121,600	-007.11 -9.45	0.358			293.15	1.4960	495.08	2.0467
27	Bromoethane	$C_6H_5Br$ $C_2H_5Br$	74-96-4	108.965	94,364	-9.45 -109.12	0.338			160.00	0.8818	320.00	1.0453
28	Bromomethane	CH <sub>3</sub> Br	74-83-9	94.939	129,730	-596.54	2.16	-0.0024234		184.45	0.3313	276.71	0.7870
29	1,2-Butadiene	$C_4H_6$	590-19-2	54.090	135,150	-311.14	0.97007	-0.0024234		136.95	1.1034	290.00	1.2279
30	1,3-Butadiene	$C_4\Pi_6$ $C_4H_6$	106-99-0	54.090	128,860	-311.14 -323.1	1.015	0.0001323		165.00	1.0333	350.00	1.4148
31			106-99-0	58.122	191,030	-323.1 -1,675	12.5	-0.03874	4.6121E-05	134.86	1.1272	400.00	2.2237
32	Butane	$C_4H_{10}$	584-03-2	90.121	55.136	314,200	280.19	1,413.9	4.0121E-05	220.00	1.1272	670.00	5.2045
33	1,2-Butanediol [use Eq. (2)] 1,3-Butanediol [use Eq. (2)]	$C_4H_{10}O_2$	107-88-0	90.121	42.152		517.35	1,415.9		196.15	0.6251	670.00	5.2437
34		$C_4H_{10}O_2$	71-36-3	74.122		324,580 -730.4		1,449.5				391.90	
35	1-Butanol 2-Butanol	$C_4H_{10}O \\ C_4H_{10}O$	78-92-2	74.122	191,200 426,790	-730.4 -3,694.6	2.2998	-0.0135		183.85 158.45	1.3465	372.90	2.5817
აა 36		$C_4H_{10}O$	106-98-9	56.106			13.828		4.5027E-05		1.3485 1.1015	380.00	2.7190
	1-Butene	$C_4H_8$			182,050	-1,611	11.963	-0.037454	4.5027E-05	87.80			1.8103
37 38	cis-2-Butene	$C_4H_8$	590-18-1	56.106	126,680	-65.47	-0.64	0.002912		134.26	1.1340	350.00	1.5022
	trans-2-Butene	$C_4H_8$	624-64-6	56.106	112,760	-104.7	0.5214			167.62	1.0986	274.03	1.2322
39	Butyl acetate	$C_6H_{12}O_2$	123-86-4	116.158	111,850	384.52	0.52005			298.15	2.2649	399.26	2.6537
40	Butylbenzene	$C_{10}H_{14}$	104-51-8	134.218	182,470	-13.912	0.72897	0.0000015		185.30	2.0492	400.00	2.9354
41	Butyl mercaptan	$C_4H_{10}S$	109-79-5	90.187	232,190	-804.35	2.7063	-0.0023017		157.46	1.6365	390.00	1.9359
42	sec-Butyl mercaptan	$C_4H_{10}S$	513-53-1	90.187	197,890	-491.54	1.7219	-0.0012499		133.02	1.6003	370.00	1.8844
43	1-Butyne	$C_4H_6$	107-00-6	54.090	136,340	-300.4	1.0216			147.43	1.1426	298.15	1.3759
44	Butyraldehyde	C <sub>4</sub> H <sub>8</sub> O	123-72-8	72.106	65,682	1,329.1	-7.1579	0.012755		176.75	1.4741	300.00	1.6459
45	Butyric acid	$C_4H_8O_2$	107-92-6	88.105	237,700	-746.4	1.829			267.95	1.6902	436.42	2.6031
46	Butyronitrile	C <sub>4</sub> H <sub>7</sub> N	109-74-0	69.105	104,000	174				161.25	1.3206	390.75	1.7199
47	Carbon dioxide	$CO_2$	124-38-9	44.010	-8,304,300	104,370	-433.33	0.60052		220.00	0.7827	290.00	1.6603
48	Carbon disulfide	$CS_2$	75-15-0	76.141	85,600	-122	0.5605	-0.001452	2.008E-06	161.11	0.7577	552.00	1.3125
49	Carbon monoxide [use Eq. (2)]	CO	630-08-0	28.010	65.429	28,723	-847.39	1,959.6		68.15	0.5912	132.00	6.4799
50	Carbon tetrachloride	CCl <sub>4</sub>	56-23-5	153.823	-752,700	8,966.1	-30.394	0.034455		250.33	1.2763	388.71	1.6374
51	Carbon tetrafluoride	$CF_4$	75-73-0	88.004	104,600	-500.6	2.2851			89.56	0.7810	145.10	0.8007
52	Chlorine	$Cl_2$	7782-50-5	70.906	63,936	46.35	-0.1623			172.12	0.6711	239.12	0.6574
53	Chlorobenzene	C <sub>6</sub> H <sub>5</sub> Cl	108-90-7	112.557	-1,307,500	15,338	-53.974	0.063483		227.95	1.3617	360.00	1.8101
54	Chloroethane	C <sub>2</sub> H <sub>5</sub> Cl	75-00-3	64.514	127,900	-345.15	0.915			134.80	0.9800	340.00	1.1632
55	Chloroform	CHCl <sub>3</sub>	67-66-3	119.378	124,850	-166.34	0.43209			233.15	1.0956	366.48	1.2192
56	Chloromethane	CH₃Cl	74-87-3	50.488	96,910	-207.9	0.37456	0.000488		175.43	0.7460	373.15	0.9684
57	1-Chloropropane	C <sub>3</sub> H <sub>7</sub> Cl	540-54-5	78.541	132,280	-153.27	0.50836			150.35	1.2073	319.67	1.3523
58	2-Chloropropane	C <sub>3</sub> H <sub>7</sub> Cl	75-29-6	78.541	69,362	215.01				200.00	1.1236	308.85	1.3577

TABLE 2-153 Heat Capacities of Inorganic and Organic Liquids [J/(kmol·K)] (Continued)

IADLE	2-133 near capacines or in	ioi gaine a	ila Organi	c Elquius	[3/ (KIIIOI-IC)]	(Commoeu)							
Cmpd. no.	Name	Formula	CAS no.	Mol. wt.	C1	C2	C3	C4	C5	$T_{\min}$ , K	$C_p$ at $T_{\min}$ $\times$ 1E-05	$T_{ m max}$ , K	$C_p$ at $T_{\rm max}$ $\times$ 1E-05
	C1	CHO	108-39-4	108.138	-246,700	3,256.8	-7.4202	0.0060467		285.39	2.1005	400.00	2.5578
59 60	m-Cresol	C <sub>7</sub> H <sub>8</sub> O									2.1895		
	o-Cresol	C <sub>7</sub> H <sub>8</sub> O	95-48-7	108.138	-185,150	3,148	-8.0367	0.007254		304.20	2.3297	400.00	2.5243
61	p-Cresol	C <sub>7</sub> H <sub>8</sub> O	106-44-5	108.138	259,980	-1,112.3	4.9427	-0.0054367		307.93	2.2740	400.00	2.5794
62	Cumene	$C_9H_{12}$	98-82-8	120.192	61,723	494.81				177.14	1.4937	425.56	2.7229
63	Cyanogen	$C_2N_2$	460-19-5	52.035	65,516	144.7	0.063229			200.08	0.9700	300.08	1.1463
64	Cyclobutane	$C_4H_8$	287-23-0	56.106	101,920	-215.81	0.8103			190.00	0.9017	298.15	1.0961
65	Cyclohexane	$C_6H_{12}$	110-82-7	84.159	-220,600	3,118.3	-9.4216	0.010687		279.69	1.4836	400.00	2.0323
66	Cyclohexanol	$C_6H_{12}O$	108-93-0	100.159	-40,000	853				296.60	2.1300	434.00	3.3020
67	Cyclohexanone	$C_6H_{10}O$	108-94-1	98.143	6,110.4	600.94				290.00	1.8038	489.75	3.0042
68	Cyclohexene	$C_6H_{10}$	110-83-8	82.144	105,850	-60	0.68			169.67	1.1525	356.12	1.7072
69	Cyclopentane	$C_5H_{10}$	287-92-3	70.133	122,530	-403.8	1.7344	-0.0010975		179.28	0.9956	322.40	1.3584
70	Cyclopentene	$C_5H_8$	142-29-0	68.117	125,380	-349.7	1.143			138.13	0.9888	317.38	1.2953
71	Cyclopropane	$C_3H_6$	75-19-4	42.080	89,952	-196.63	0.65237			150.00	0.7514	298.15	0.8932
72	Cyclohexyl mercaptan	$C_6H_{12}S$	1569-69-3	116.224	177,560	-179.12	0.76723			189.64	1.7118	431.95	2.4334
73	Decanal	C <sub>10</sub> H <sub>20</sub> O	112-31-2	156.265	150,460	586.63	******			267.15	3.0718	488.15	4.3682
74	Decane	$C_{10}H_{22}$	124-18-5	142.282	278,620	-197.91	1.0737			243.51	2.9409	460.00	4.1478
75	Decanoic acid	$C_{10}H_{20}O_2$	334-48-5	172.265	219,840	140.41	0.9968			304.75	3.5521	543.15	5.9017
76	1-Decanol	$C_{10}H_{20}O_2$ $C_{10}H_{22}O$	112-30-1	158.281	4,988,500	-52,898	216.35	-0.37538	0.00023674	280.05	3.5373	503.00	5.0169
77	1-Decene	$C_{10}H_{20}$	872-05-9	140.266	417,440	-1,616.5	5.3948	-0.004348	0.00025014	206.89	2.7541	443.75	3.8250
78	Decyl mercaptan	$C_{10}H_{20}$ $C_{10}H_{22}$ S	143-10-2	174.347	314,570	-1,010.9 -160.93	0.95561	-0.004040		247.56	3.3330	512.35	4.8297
79	1-Decyne	$C_{10}H_{18}$	764-93-2	138.250	276,900	-371.23	1.5774			229.15	2.7466	447.15	4.2629
80	Deuterium	$D_{2}$	7782-39-0	4.032	270,900	-371.23	1.5774			229.13	2.7400	447.13	4.2029
81			557-91-5		140 400	221.0	0.5946			210.15	1.2695	201.15	1 4749
	1,1-Dibromoethane	C <sub>2</sub> H <sub>4</sub> Br <sub>2</sub>		187.861	149,400	-231.8				210.15		381.15	1.4743
82	1,2-Dibromoethane	C <sub>2</sub> H <sub>4</sub> Br <sub>2</sub>	106-93-4	187.861	200,560	-491.44	0.9187			282.85	1.3506	410.00	1.5350
83	Dibromomethane	CH <sub>2</sub> Br <sub>2</sub>	74-95-3	173.835	202,580	-726.3	1.3377			240.00	1.0532	370.10	1.1701
84	Dibutyl ether	C <sub>8</sub> H <sub>18</sub> O	142-96-1	130.228	270,720	-259.83	0.95427			175.30	2.5450	450.00	3.4704
85	m-Dichlorobenzene	C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	541-73-1	147.002	114,880	187.25				248.39	1.6139	400.00	1.8978
86	o-Dichlorobenzene	$C_6H_4Cl_2$	95-50-1	147.002	93,093	183.97	0.2314			273.15	1.6061	528.75	2.5506
87	p-Dichlorobenzene	$C_6H_4Cl_2$	106-46-7	147.002	133,950	-24.84	0.48191			326.14	1.7711	513.56	2.4829
88	1,1-Dichloroethane	$C_2H_4Cl_2$	75-34-3	98.959	126,340	-94.63	0.32			176.19	1.1960	330.45	1.3001
89	1,2-Dichloroethane	$C_2H_4Cl_2$	107-06-2	98.959	179,170	-444.74	0.93009			237.49	1.2601	356.59	1.3885
90	Dichloromethane	$CH_2Cl_2$	75-09-2	84.933	98,968	-62.941	0.23265			180.00	0.9518	320.00	1.0265
91	1,1-Dichloropropane	$C_3H_6Cl_2$	78-99-9	112.986	144,560	-53.605	0.30617			180.00	1.4483	361.25	1.6515
92	1,2-Dichloropropane	$C_3H_6Cl_2$	78-87-5	112.986	111,560	149.44				275.00	1.5266	369.52	1.6678
93	Diethanol amine	$C_4H_{11}NO_2$	111-42-2	105.136	184,200	286				301.15	2.7033	541.54	3.3908
94	Diethyl amine	$C_4H_{11}N$	109-89-7	73.137	101,330	243.18				223.35	1.5564	328.60	1.8124
95	Diethyl ether	$C_4H_{10}O$	60-29-7	74.122	44,400	1,301	-5.5	0.008763		156.92	1.4698	460.00	3.3202
96	Diethyl sulfide	$C_4H_{10}S$	352-93-2	90.187	238,520	-1,038.4	4.0587	-0.0044691		181.95	1.5703	322.08	1.7579
97	1,1-Difluoroethane [use Eq. (2)]	$C_2H_4F_2$	75-37-6	66.050	67.155	105,580	310.21	-490.54		154.56	0.9915	359.98	1.6874
98	1,2-Difluoroethane	$C_2H_4F_2$	624-72-6	66.050	82,577	109.85				215.00	1.0619	283.65	1.1374
99	Difluoromethane	$CH_2F_2$	75-10-5	52.023	263,980	-1,791.1	4.3666			200.00	0.8042	250.00	0.8912
100	Di-isopropyl amine	$C_6H_{15}N$	108-18-9	101.190	98,434	429.04				275.00	2.1642	357.05	2.5162
101	Di-isopropyl ether	$C_6H_{14}O$	108-20-3	102.175	163,000	-4.5	0.62			187.65	1.8399	341.45	2.3375
102	Di-isopropyl ketone	$C_7H_{14}O$	565-80-0	114.185	179,270	28.37	0.5375			204.81	2.0763	410.00	2.8126
103	1.1-Dimethoxyethane	$C_4H_{10}O_2$	534-15-6	90.121	187,790	-313.41	1.1023			159.95	1.6586	337.45	2.0755
104	1.2-Dimethoxypropane	$C_5H_{12}O_2$	7778-85-0	104.148	199,930	-191.5	0.87664			226.10	2.0145	366.15	2.4734
105	Dimethyl acetylene	$C_4H_6$	503-17-3	54.090	88,153	124.16				240.91	1.1806	300.13	1.2542
106	Dimethyl amine	$C_2H_7N$	124-40-3	45.084	-214,870	3,787.2	-13.781	0.016924		180.96	1.1947	298.15	1.3779
107	2,3-Dimethylbutane	$C_6H_{14}$	79-29-8	86.175	129,450	18.5	0.608	0.010021		145.19	1.4495	331.13	2.0224
108	1,1-Dimethylcyclohexane	$C_8H_{16}$	590-66-9	112.213	134,500	8.765	0.81151			239.66	1.8321	392.70	2.6309
109	cis-1 9-Dimethylevelohevane	$C_8H_{16}$	2207-01-4	112.213	150,130	-62.38	0.8851			223.16	1.8029	402.94	2.6870
110	trans-1 2-Dimethylevelohovono	$C_8H_{16}$	6876-23-9	112.213	155,560	-145.26	1.0932			184.99	1.6610	396.58	2.6989
110	trans-1,2-Dimethylcyclohexane Dimethyl disulfide	$C_8\Pi_{16}$ $C_2H_6S_2$	624-92-0	94.199	171,580	-145.26 -256.67	0.5727			188.44	1.4355	360.00	1.5340
111	Dimethyl alsulfide Dimethyl ether	$C_2H_6S_2$ $C_2H_6O$	115-10-6	46.068	110,100	-256.67 -157.47	0.51853			131.65	0.9836	250.00	1.0314
	NN Dimethyl Committee	$C_2H_6O$ $C_3H_7NO$											
113	N,N-Dimethyl formamide		68-12-2	73.094	147,900	-106	0.384			273.82	1.4767	466.44	1.8200
114	2,3-Dimethylpentane	$C_7H_{16}$	565-59-3	100.202	146,420	59.2	0.604			90.00	1.5664	380.00	2.5613
115	Dimethyl phthalate	$C_{10}H_{10}O_4$	131-11-3	194.184	206,560	325.75				274.16	2.9587	360.00	3.2383
116	Dimethylsilane	C <sub>2</sub> H <sub>8</sub> Si	1111-74-6	60.170	131,810	ı	1	'	ı	298.15	1.3181	298.15	1.3181

	117	Dimethyl sulfide	$C_2H_6S$	75-18-3	62.134	146,950	-380.06	1.2035	-0.00084787		174.88	1.1276	310.48	1.1959
	118	Dimethyl sulfoxide	C <sub>2</sub> H <sub>6</sub> OS	67-68-5	78.133	240,300	-595	1.013			291.67	1.5293	422.15	1.6965
	119	Dimethyl terephthalate	$C_{10}H_{10}O_4$	120-61-6	194.184	190,020	431.04				423.40	3.7252	466.35	3.9104
	120	1,4-Dioxane		123-91-1	88.105	956,860		9.6124			284.95	1.5306	374.47	2.2277
			$C_4H_8O_2$				-5,559.9	9.0124				1.5500		
	121	Diphenyl ether	$C_{12}H_{10}O$	101-84-8	170.207	134,160	447.67				300.03	2.6847	570.00	3.8933
	122	Dipropyl amine	$C_6H_{15}N$	142-84-7	101.190	49,120	562.24				277.90	2.0537	407.90	2.7846
	123	Dodecane	$C_{12}H_{26}$	112-40-3	170.335	508,210	-1,368.7	3.1015			263.57	3.6292	330.00	3.9429
	124	Eicosane	$C_{20}H_{42}$	112-95-8	282.547	352,720	807.32	0.2122			309.58	6.2299	616.93	9.3154
	125	Ethane [use Eq. (2)]	$C_2H_6$	74-84-0	30.069	44.009	89,718	918.77	-1,886		92.00	0.6855	290.00	1.2444
	126	Ethanol		64-17-5	46.068	102.640	-139.63	-0.030341	0.0020386		159.05	0.8787	390.00	1.6450
			$C_2H_6O$						0.0020360			0.0101		
	127	Ethyl acetate	$C_4H_8O_2$	141-78-6	88.105	226,230	-624.8	1.472			189.60	1.6068	350.21	1.8796
	128	Ethyl amine	$C_2H_7N$	75-04-7	45.084	121,700	38.993				192.15	1.2919	289.73	1.3300
	129	Ethylbenzene	$C_8H_{10}$	100-41-4	106.165	154,040	-142.29	0.80539			178.20	1.5426	409.35	2.3075
	130	Ethyl benzoate	$C_9H_{10}O_2$	93-89-0	150.175	124,500	370.6				238.45	2.1287	486.55	3.0482
	131	2-Ethyl butanoic acid	$C_6H_{12}O_2$	88-09-5	116.158	56,359	603.02				258.15	2.1203	466.95	3.3794
	132	Ethyl butyrate	$C_6H_{12}O_2$	105-54-4	116.158	82,434	422.45	0.20992			285.50	2.2015	428.25	3.0185
	133	Ethylcyclohexane	$C_8H_{16}$	1678-91-7	112.213	132,360	72.74	0.64738			161.84	1.6109	404.95	2.6798
	134	Ethylcyclopentane	$C_7H_{14}$	1640-89-7	98.186	178,520	-518.35	2.3255	-0.0016818		134.71	1.4678	301.82	1.8767
	135	Ethylene	$C_2H_4$	74-85-1	28.053	247,390	-4,428	40.936	-0.1697	0.00026816	104.00	0.7012	252.70	0.9758
	136	Ethylenediamine	$C_2H_8N_2$	107-15-3	60.098	184,440	-150.2	0.37044			284.29	1.7168	390.41	1.8226
	137	Ethylene glycol	$C_2H_6O_2$	107-21-1	62.068	35,540	436.78	-0.18486			260.15	1.3666	493.15	2.0598
	138	Ethyleneimine	$C_2H_5N$	151-56-4	43.068	46,848	205.35				250.00	0.9819	329.00	1.1441
	139	Ethylene oxide	$C_2H_4O$	75-21-8	44.053	144,710	-758.87	2.8261	-0.003064		160.65	0.8303	283.85	0.8693
								2.0201	-0.003004					
	140	Ethyl formate	$C_3H_6O_2$	109-94-4	74.079	80,000	223.6				254.20	1.3684	374.20	1.6367
	141	2-Ethyl hexanoic acid	$C_8H_{16}O_2$	149-57-5	144.211	207,670	-17.907	1.0493			235.00	2.6141	510.10	4.7157
	142	Ethylhexyl ether	$C_8H_{18}O$	5756-43-4	130.228	146,040	458.22				298.15	2.8266	417.15	3.3719
	143	Ethylisopropyl ether	$C_5H_{12}O$	625-54-7	88.148	106,250	292.15				298.15	1.9335	326.15	2.0153
	144	Ethylisopropyl ketone	$C_6H_{12}O$	565-69-5	100.159	229,250	-404.54	1.1382			204.15	1.9410	386.55	2.4295
	145	Ethyl mercaptan	C <sub>2</sub> H <sub>6</sub> S	75-08-1	62.134	134,670	-234.39	0.59656			125.26	1.1467	315.25	1.2007
	146	Ethyl propionate	$C_5H_{10}O_2$	105-37-3	102.132	76,330	400.1	0.00000			298.15	1.9562	410.00	2.4037
		Ethyl propionate						2.0047	0.0040057			1.6686		
	147	Ethylpropyl ether	C <sub>5</sub> H <sub>12</sub> O	628-32-0	88.148	103,680	726.3	-2.6047	0.0040957		145.65		320.00	2.0358
	148	Ethyltrichlorosilane	C <sub>2</sub> H <sub>5</sub> Cl <sub>3</sub> Si	115-21-9	163.506	105,150	85.318	0.46693			167.55	1.3255	371.05	2.0109
	149	Fluorine	$\mathbf{F}_2$	7782-41-4	37.997	-94,585	7,529.9	-139.6	1.1301	-0.0033241	58.00	0.5541	98.00	0.5966
	149	Fluorine	$\mathbf{F}_2$	7782-41-4	37.997	1724,400	-59,924	537.85			53.48	0.5798	56.00	0.5535
	150	Fluorobenzene	$C_6H_5F$	462-06-6	96.102	-991,200	11,734	-40.669	0.047333		239.99	1.3675	319.99	1.5018
	151	Fluoroethane	C <sub>2</sub> H <sub>5</sub> F	353-36-6	48.060	85,663	-118.56	0.55459			140.00	0.7994	240.00	0.8915
	152	Fluoromethane	CH <sub>3</sub> F	593-53-3	34.033	74,746	-132.32	0.53772			140.00	0.6676	220.00	0.7166
	153		CH <sub>2</sub> O	50-00-0			28.3	0.55112			204.00	0.6767	234.00	0.6852
		Formaldehyde			30.026	61,900								
	154	Formamide	CH <sub>3</sub> NO	75-12-7	45.041	63,400	150.6				292.00	1.0738	493.00	1.3765
	155	Formic acid	$CH_2O_2$	64-18-6	46.026	78,060	71.54				281.45	0.9820	380.00	1.0525
	156	Furan	$C_4H_4O$	110-00-9	68.074	114,370	-215.69	0.72691			187.55	0.9949	304.50	1.1609
	157	Helium-4	He	7440-59-7	4.003	387,220	-465,570	211,800	-42,494	3212.9	2.20	0.1087	4.60	0.2965
	157	Helium-4	He	7440-59-7	4.003	410,430	-464,890	135,100	,		1.80	0.1135	2.10	0.2995
	158	Heptadecane	C <sub>17</sub> H <sub>36</sub>	629-78-7	240.468	376,970	347.82	0.57895			295.13	5.3005	575.30	7.6869
	159	Heptadecane   Heptanal	$C_{17}H_{36}$ $C_{7}H_{14}O$	111-71-7	114.185	222,360	-105.17	0.65074			229.80	2.3256	381.25	2.7685
				111-/1-/					0.547.0			4.5250		
	160	Heptane [use Eq. (2)]	$C_7H_{16}$	142-82-5	100.202	61.26	314,410	1,824.6	-2,547.9		182.57	1.9989	520.00	4.0657
	161	Heptanoic acid	$C_7H_{14}O_2$	111-14-8	130.185	194,570	-23.206	0.88395			265.83	2.5087	496.15	4.0065
	162	1-Heptanol	$C_7H_{16}O$	111-70-6	116.201	2,416,800	-26,105	110.03	-0.19172	0.00011968	239.15	2.3590	448.60	3.8766
	163	2-Heptanol	$C_7H_{16}O$	543-49-7	116.201	283,127	-1,037.63	3.44064			230.00	2.2649	432.90	4.7873
	164	3-Heptanone	$C_7H_{14}O$	106-35-4	114.185	270,730	-399.89	1.0601			234.15	2.3522	480.00	3.2303
	165	2-Heptanone	$C_7H_{14}O$	110-43-0	114.185	265,040	-375.68	1.0024			238.15	2.3242	490.00	3.2163
	166	1-Heptene	$C_7H_{14}$	592-76-7	98.186	267,950	-1,315.9	6.5242	-0.011994	9.3808E-06	154.12	1.8150	366.79	2.4096
	167			1639-09-4		236,870		0.78982	0.011334	J.5000E-00	229.92	2.4229	460.00	
		Heptyl mercaptan	C <sub>7</sub> H <sub>16</sub> S		132.267		-158.01							3.3131
	168	1-Ĥeptyne	$C_7H_{12}$	628-71-7	96.170	46,798	761.13	-0.62882			200.00	1.7387	372.93	2.4319
	169	Hexadecane	$C_{16}H_{34}$	544-76-3	226.441	370,350	231.47	0.68632			291.31	4.9602	560.01	7.1521
	170	Hexanal	$C_6H_{12}O$	66-25-1	100.159	117,700	329.52				217.15	1.8926	401.45	2.4999
	171	Hexane	$C_6H_{14}$	110-54-3	86.175	172,120	-183.78	0.88734			177.83	1.6750	460.00	2.7534
	172	Hexanoic acid	$C_6H_{12}O_2$	142-62-1	116.158	161,980	44.116	0.709			269.25	2.2526	478.85	3.4568
	173	1-Hexanol	$C_6H_{14}O$	111-27-3	102.175	1,638,600	-17,261	71.721	-0.12026	0.000071087	228.55	1.9821	460.00	3.5197
	174	2-Hexanol	$C_6H_{14}O$	626-93-7	102.175	267,628	-1,033.06	3.35185	"	0.0000.1001	223.00	2.0394	585.30	8.1124
	175	2-Hexanone	$C_6H_{12}O$	591-78-6	100.159	208,250	-107.47	0.2062	0.00070293		217.35	2.0185	460.00	2.7087
2-167									0.00070293			2.0133		2.7632
7	176	3-Hexanone	$C_6H_{12}O$	589-38-8	100.159	235,960	-345.94	0.94278			217.50		460.00	
57	177	1-Hexene	$C_6H_{12}$	592-41-6	84.159	164,640	-200.37	0.8784			133.39	1.5354	336.63	1.9673
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TABLE 2-153 Heat Capacities of Inorganic and Organic Liquids [J/(kmol·K)] (Continued)

		J			, (	,							
Cmpd.											$C_p$ at $T_{\min}$		$C_p$ at $T_{ m max}$
no.	Name	Formula	CAS no.	Mol. wt.	C1	C2	C3	C4	C5	$T_{\min}$ , K	×1E-05	$T_{\rm max}$ , K	×1E-05
178	3-Hexvne	$C_6H_{10}$	928-49-4	82.144	82.795	283.4				300.00	1.6781	354.35	1.8322
179	Hexyl mercaptan	$C_6H_{10}$ $C_6H_{14}S$	111-31-9	118.240	303,320	-1,009	3.3885	-0.002762		192.62	2.1495	430.00	2.7639
180			693-02-7	82.144	93,000	326	ა.აიია	-0.002762		200.00	1.5820	344.48	2.7639
	1-Hexyne	$C_6H_{10}$	764-35-2	82.144	94,860	254.15				300.00			
181 182	2-Hexyne	$C_6H_{10}$	302-01-2			50.929	0.042270			274.69	1.7110	357.67	1.8576
	Hydrazine	$H_4N_2$		32.045	79,815		0.043379	450.05			0.9708	653.15	1.3158
183	Hydrogen [use Eq. (2)]	H <sub>2</sub>	1333-74-0	2.016	66.653	6,765.9	-123.63	478.27		13.95	0.1262	32.00	1.3122
184	Hydrogen bromide	HBr	10035-10-6	80.912	57,720	9.9				185.15	0.5955	206.45	0.5976
185	Hydrogen chloride	HCl	7647-01-0	36.461	47,300	90	0.0000			165.00	0.6215	185.00	0.6395
186	Hydrogen cyanide	CHN	74-90-8	27.025	95,398	-197.52	0.3883			259.83	0.7029	298.85	0.7105
187	Hydrogen fluoride	HF	7664-39-3	20.006	62,520	-223.02	0.6297	1 000		189.79	0.4288	292.67	0.5119
188	Hydrogen sulfide [use Eq. (2)]	H <sub>2</sub> S	7783-06-4	34.081	64.666	49,354	22.493	-1,623		187.68	0.6733	370.00	4.9183
189	Isobutyric acid	$C_4H_8O_2$	79-31-2	88.105	127,540	-65.35	0.82867	0.0000010		270.00	1.7031	427.65	2.5114
190	Isopropyl amine	C <sub>3</sub> H <sub>9</sub> N	75-31-0	59.110	-32,469	1,977.1	-7.0145	0.0086913		177.95	1.4621	320.00	1.6671
191	Malonie acid	$C_3H_4O_4$	141-82-2	104.061	157,850	-41.619	0.42817			407.95	2.1213	603.75	2.8880
192	Methacrylic acid	$C_4H_6O_2$	79-41-4	86.089	146,290	-58.59	0.3582			288.15	1.5915	434.15	1.8837
193	Methane [use Eq. (2)]	$CH_4$	74-82-8	16.042	65.708	38,883	-257.95	614.07		90.69	0.5361	190.00	14.9780
194	Methanol	CH <sub>4</sub> O	67-56-1	32.042	105,800	-362.23	0.9379			175.47	0.7112	400.00	1.1097
195	N-Methyl acetamide	C <sub>3</sub> H <sub>7</sub> NO	79-16-3	73.094	62,600	243.4				359.00	1.4998	538.50	1.9367
196	Methyl acetate	$C_3H_6O_2$	79-20-9	74.079	61,260	270.9				253.40	1.2991	373.40	1.6241
197	Methyl acetylene	$C_3H_4$	74-99-7	40.064	79,791	89.49				200.00	0.9769	249.94	1.0216
198	Methyl acrylate	$C_4H_6O_2$	96-33-3	86.089	275,500	-1,147	2.568			196.32	1.4930	353.35	1.9084
199	Methyl amine	CH <sub>5</sub> N	74-89-5	31.057	92,520	37.45				179.69	0.9925	266.82	1.0251
200	Methyl benzoate	$C_8H_8O_2$	93-58-3	136.148	125,630	279.75				260.75	1.9857	472.65	2.5785
201	3-Methyl-1,2-butadiene	$C_5H_8$	598-25-4	68.117	135,370	-133.34	0.63868			159.53	1.3035	314.56	1.5662
202	2-Methylbutane	$C_5H_{12}$	78-78-4	72.149	108,300	146	-0.292	0.00151		113.25	1.2328	310.00	1.7048
203	2-Methylbutanoic acid	$C_5H_{10}O_2$	116-53-0	102.132	74,200	417.4				321.50	2.0839	481.50	2.7518
204	3-Methyl-1-butanol	$C_5H_{12}O$	123-51-3	88.148	247,870	-1,145	3.4223			155.95	1.5254	404.15	3.4411
205	2-Methyl-1-butene	$C_5H_{10}$	563-46-2	70.133	149,510	-247.63	0.91849			135.58	1.3282	304.31	1.5921
206	2-Methyl-2-butene	$C_5H_{10}$	513-35-9	70.133	151,600	-266.72	0.90847			139.39	1.3207	311.71	1.5673
207	2-Methyl-1-butene-3-yne	$C_5H_6$	78-80-8	66.101	81,919	181.01				298.15	1.3589	305.40	1.3720
208	Methylbutyl ether	$C_5H_{12}O$	628-28-4	88.148	177,850	-171.57	0.74379			157.48	1.6928	343.31	2.0661
209	Methylbutyl sulfide	$C_5H_{12}S$	628-29-5	104.214	198,390	-220.35	0.76096			175.30	1.8315	510.00	2.8394
210	3-Methyl-1-butyne	$C_5H_8$	598-23-2	68.117	105,200	191.1				200.00	1.4342	299.49	1.6243
211	Methyl butyraté	$C_5H_{10}O_2$	623-42-7	102.132	102,930	129.1	0.62516			277.25	1.8678	415.87	2.6474
212	Methylchlorosilane	CH <sub>5</sub> ClSi	993-00-0	80.589	47,726	338.4				250.00	1.3233	325.00	1.5771
213	Methylcyclohexane	$C_7H_{14}$	108-87-2	98.186	131,340	-63.1	0.8125			146.58	1.3955	320.00	1.9435
214	1-Methylcyclohexanol	$C_7H_{14}O$	590-67-0	114.185	50,578	508.59				300.00	2.0315	441.15	2.7494
215	cis-2-Methylcyclohexanol	$C_7H_{14}O$	7443-70-1	114.185	118,600	447.07				300.00	2.5272	438.15	3.1448
216	trans-2-Methylcyclohexanol	C <sub>7</sub> H <sub>14</sub> O	7443-52-9	114.185	118,170	447.99				300.00	2.5257	440.15	3.1535
217	Methylcyclopentane	$C_6H_{12}$	96-37-7	84.159	155,920	-490	2.1383	-0.0015585		130.73	1.2492	366.48	1.8682
218	1-Methylcyclopentene	C <sub>6</sub> H <sub>10</sub>	693-89-0	82.144	53,271	327.92				200.00	1.1885	348.64	1.6760
219	3-Methylcyclopentene	C <sub>6</sub> H <sub>10</sub>	1120-62-3	82.144	46,457	346.93				200.00	1.1584	338.05	1.6374
220	Methyldichlorosilane	CH <sub>4</sub> Cl <sub>2</sub> Si	75-54-7	115.034	27,030	413				250.00	1.3028	350.00	1.7158
221	Methylethyl ether	C <sub>3</sub> H <sub>8</sub> O	540-67-0	60.095	85,383	199.08	-0.061547			160.00	1.1566	280.50	1.3638
222	Methylethyl ketone	C <sub>4</sub> H <sub>8</sub> O	78-93-3	72.106	132,300	200.87	-0.9597	0.0019533		186.48	1.4905	373.15	1.7511
223	Methylethyl sulfide	C <sub>3</sub> H <sub>8</sub> S	624-89-5	76.161	161,240	-288.61	0.78179	0.0010030		167.23	1.3484	339.80	1.5344
224	Methyl formate	$C_2H_4O_2$	107-31-3	60.052	130,200	-396	1.21			174.15	0.9793	304.90	1.2195
225	Methylisobutyl ether	$C_5H_{12}O$	625-44-5	88.148	92,919	324.43	1.21			298.15	1.8965	350.00	2.0647
226	Methylisobutyl ketone	$C_6H_{12}O$	108-10-1	100.159	183,650	-79.862	0.60769			189.15	1.9029	389.15	2.4460
227	Methyl Isocyanate	$C_2H_3NO$	624-83-9	57.051	149,770	-529.82	1.3499			256.15	1.0263	366.00	1.3668
228	Methylisopropyl ether	$C_{2}H_{3}NO$ $C_{4}H_{10}O$	598-53-8	74.122	143,440	-154.07	0.7255			127.93	1.3560	310.00	1.6540
229	Methylisopropyl ketone	$C_{5}H_{10}O$	563-80-4	86.132	191,170	-331.04	0.7255			180.15	1.6348	440.00	2.3610
230	Methylisopropyl sulfide	$C_{5}H_{10}O$ $C_{4}H_{10}S$	1551-21-9	90.132	211,170	-661.97	2.4216	-0.0021383		171.64	1.5808	357.91	1.8641
231	Methyl mercaptan	$C_4\Pi_{10}S$ $CH_4S$	74-93-1	48.107	115,300	-263.23	0.60412	-0.0021303		150.18	0.8939	298.15	0.9052
232	Methyl mercaptan Methyl methacrylate	$C_5H_8O_2$	80-62-6	100.116	255,100	-265.25 -938.4	2.413			224.95	1.6611	373.45	2.4118
233	2-Methyloctanoic acid	$C_{5}H_{8}O_{2}$ $C_{9}H_{18}O_{2}$	3004-93-1	158.238	226,650	15.421	1.0578			240.00	2.9128	518.15	5.1864
234	2-Methylpentane	$C_9H_{18}O_2$ $C_6H_{14}$	107-83-5	86.175	142,220	-47.83	0.739			119.55	1.4706	333.41	2.0842
234	2-Methylpentane Methylpentylether	$C_6H_{14}$ $C_6H_{14}O$	628-80-8	102.175	251,890	-468.32	1.2209			176.00	2.0728	372.00	2.0842
236			75-28-5	58.122	172,370	-405.32 -1,783.9	14.759	-0.047909	0.00005805	113.54	0.9961	380.00	2.4003
230	2-Methylpropane	$C_4H_{10}$	10-20-0	30.122	172,570	-1,700.9	14.709	-0.047909	0.000000000	115.54	0.9901	300.00	2.0723

237	2-Methyl-2-propanol	C <sub>4</sub> H <sub>10</sub> O	75-65-0	74.122	-925,460	7,894.9	-17.661	0.013617		298.96	2.2016	460.00	2.9455
238	2-Methyl propene	$C_4H_8$	115-11-7	56.106	87,680	217.1	-0.9153	0.002266		132.81	1.0568	343.15	1.4596
239	Methyl propionate	$C_4H_8O_2$	554-12-1	88.105	71.140	335.5				300.00	1.7179	390.00	2.0198
240	Methylpropyl ether	C <sub>4</sub> H <sub>10</sub> O	557-17-5	74.122	144,110	-102.09	0.58113			133.97	1.4086	312.20	1.6888
241	Methylpropyl sulfide	C <sub>4</sub> H <sub>10</sub> S	3877-15-4	90.187	179,850	-264.1	0.79202			160.17	1.5787	368.69	1.9014
242	Methylsilane	CH <sub>6</sub> Si	992-94-9	46.144	113,470	201.1	0.10202			298.15	1.1347	298.15	1.1347
243	alpha-Methyl styrene	$C_9H_{10}$	98-83-9	118.176	76,822	421.6				249.95	1.8220	438.65	2.6176
	Methyl tert-butyl ether		1634-04-4				0.0022	0.0009795			1.5410	328.18	1.9954
244		C <sub>5</sub> H <sub>12</sub> O		88.148	134,300	94.356	-0.0032	0.0009795		164.55			
245	Methyl vinyl ether	C <sub>3</sub> H <sub>6</sub> O	107-25-5	58.079	73,600	184.7				151.15	1.0152	278.65	1.2507
246	Naphthalene	$C_{10}H_8$	91-20-3	128.171	29,800	527.5				353.43	2.1623	491.14	2.8888
247	Neon	Ne	7440-01-9	20.180	1,034,100	-138,770	7,154	-162.55	1.3841	24.56	0.3666	40.00	0.6980
248	Nitroethane	$C_2H_5NO_2$	79-24-3	75.067	187,740	-497.6	1.0691			183.63	1.3242	387.22	1.5536
249	Nitrogen	$N_2$	7727-37-9	28.013	281,970	-12,281	248	-2.2182	0.0074902	63.15	0.5593	112.00	0.7960
250	Nitrogen trifluoride	F <sub>3</sub> N	7783-54-2	71.002	101,400	-682.11	3.8912			117.00	0.7486	175.50	1.0154
251	Nitromethane	CH <sub>3</sub> NO <sub>2</sub>	75-52-5	61.040	116,270	-135.3	0.345			244.60	1.0382	473.15	1.2949
252	Nitrous oxide	N <sub>0</sub> O	10024-97-2	44.013	67,556	54.373				182.30	0.7747	200.00	0.7843
253	Nitric oxide	NO	10102-43-9	30.006	-2,979,600	76,602	-652.59	1.8879		109.50	0.6229	150.00	1.9909
254	Nonadecane	C <sub>19</sub> H <sub>40</sub>	629-92-5	268.521	342,570	762.08	0.20481	2.00.0		305.04	5.9409	603.05	8.7663
255	Nonanal	C <sub>9</sub> H <sub>18</sub> O	124-19-6	142.239	136,820	531.29	0.20101			255.15	2.7238	468.15	3.8554
256	Nonane	$C_9H_{20}$	111-84-2	128.255	383,080	-1,139.8	2.7101			219.66	2.6348	325.00	2.9890
257	Nonanoie acid	$C_9H_{18}O_2$	112-05-0	158.238	224,336	49.726	0.9813			285.55	3.1855	528.75	5.2498
258	1-Nonanol	$C_9H_{18}O_2$ $C_9H_{20}O$	143-08-8	144.255	10,483,000	-115,220	476.87	-0.85381	0.00056246	310.00	3.5059	460.00	4.6494
								-0.55551	0.00030240				
259	2-Nonanol	C <sub>9</sub> H <sub>20</sub> O	628-99-9	144.255	329,641	-1,046.78	3.61823			238.15	2.8555	649.50	11.7608
260	1-Nonene	$C_9H_{18}$	124-11-8	126.239	254,490	-298.06	1.1707			191.91	2.4041	420.02	3.3583
261	Nonyl mercaptan	$C_9H_{20}S$	1455-21-6	160.320	265,350	-46.22	0.79154			253.05	3.0434	492.95	4.3491
262	1-Nonyne	$C_9H_{16}$	3452-09-3	124.223	253,580	-366.3	1.4881			223.15	2.4594	423.85	3.6566
263	Octadecane	$C_{18}H_{38}$	593-45-3	254.494	399,430	374.64	0.58156			301.31	5.6511	589.86	8.2276
264	Octanal	$C_8H_{16}O$	124-13-0	128.212	130,650	463.61				246.00	2.4470	447.15	3.3795
265	Octane	$C_8H_{18}$	111-65-9	114.229	224,830	-186.63	0.95891			216.38	2.2934	460.00	3.4189
266	Octanoic acid	$C_8H_{16}O_2$	124-07-2	144.211	205,260	44.392	0.8956			289.65	2.9326	512.85	4.6358
267	1-Octanol	$C_8H_{18}O$	111-87-5	130.228	571,370	-4,849	19.725	-0.021532		250.00	2.5550	467.10	4.1566
268	2-Octanol	$C_8H_{18}O$	123-96-6	130.228	319,198	-1,042.21	3.52943			241.55	2.7338	452.90	5.7113
269	2-Octanone	$C_8H_{16}O$	111-13-7	128.212	300,400	-426.2	1.1172			252.86	2.6406	500.00	3.6660
270	3-Octanone	$C_8H_{16}O$	106-68-3	128.212	289,980	-417.27	1.2218			255.55	2.6314	440.65	3.4335
271	1-Octene	$C_8H_{16}$	111-66-0	112.213	509,420	-4,279.1	21.477	-0.044462	3.5028E-05	171.45	2.1327	394.41	2.8235
272	Octyl mercaptan	$C_8H_{18}S$	111-88-6	146.294	240,040	-33.198	0.67889			240.00	2.7118	472.19	3.7573
273	1-Octyne	$C_8H_{14}$	629-05-0	110.197	42,642	886.67	-0.69315			200.00	1.9225	399.35	2.8619
274	Oxalic acid	$C_2H_2O_4$	144-62-7	90.035	175,510	-381.36	0.64623			462.65	1.3740	603.00	1.8052
275	Oxygen	$O_2$	7782-44-7	31.999	175,430	-6,152.3	113.92	-0.92382	0.0027963	54.36	0.5365	142.00	0.9066
276	Ozone	$O_3$	10028-15-6	47.998	60,046	281.16				90.00	0.8535	150.00	1.0222
277	Pentadecane	$C_{15}H_{32}$	629-62-9	212.415	346,910	219.54	0.65632			283.07	4.6165	543.84	6.6042
278	Pentanal	$C_5H_{10}O$	110-62-3	86.132	112,050	257.78				200.00	1.6361	376.15	2.0901
279	Pentane	$C_5H_{12}$	109-66-0	72.149	159,080	-270.5	0.99537			143.42	1.4076	390.00	2.0498
280	Pentanoic acid	$C_5H_{10}O_2$	109-52-4	102.132	145,050	28.344	0.6372			239.15	1.8827	458.95	2.9228
281	1-Pentanol	$C_5H_{12}O$	71-41-0	88.148	201,200	-651.3	2.275			200.14	1.6198	389.15	2.9227
282	2-Pentanol	C <sub>5</sub> H <sub>12</sub> O	6032-29-7	88.148	251,596	-1,028.49	3.26306			200.00	1.7642	561.00	7.0158
283	2-Pentanone	C <sub>5</sub> H <sub>10</sub> O	107-87-9	86.132	194,590	-263.86	0.76808			196.29	1.7239	375.46	2.0380
284	3-Pentanone	C <sub>5</sub> H <sub>10</sub> O	96-22-0	86.132	193,020	-176.43	0.5669			234.18	1.8279	375.14	2.0661
285	1-Pentene	C <sub>5</sub> H <sub>10</sub>	109-67-1	70.133	156,100	-456.94	2.255	-0.003163	0.00000238	108.02	1.2939	350.00	1.7251
286	2-Pentyl mercaptan	$C_5H_{12}S$	2084-19-7	104.214	188,200	-140.84	0.63581	0.000100	0.00000230	160.75	1.8199	385.15	2.2827
287	Pentyl mercaptan	C <sub>5</sub> H <sub>12</sub> S	110-66-7	104.214	213,760	-324.4	0.9472			197.45	1.8664	399.79	2.3546
288	1-Pentyne	$C_5H_{12}$	627-19-0	68.117	86,200	256.6	0.0112			200.00	1.3752	313.33	1.6660
289	2-Pentyne	$C_5H_8$	627-21-4	68.117	68,671	246.66				200.00	1.1800	329.27	1.4989
290	Phenanthrene	$C_{14}H_{10}$	85-01-8	178.229	103,370	527.03				372.39	2.9963	500.00	3.6688
291	Phenol	$C_{6}H_{6}O$	108-95-2	94.111	101,720	317.61				314.06	2.0147	425.00	2.3670
292			103-33-2				0.29552			243.15	1.3080	489.75	2.3745
	Phenyl isocyanate	C <sub>7</sub> H <sub>5</sub> NO	85-44-9	119.121	60,834	215.89 252.4	0.29552			404.15	2.4741	489.75 557.65	
293 294	Phthalic anhydride	C <sub>8</sub> H <sub>4</sub> O <sub>3</sub>	85-44-9 463-49-0	148.116 40.064	145,400	252.4 98.275				200.00		238.65	2.8615
	Propadiene	C <sub>3</sub> H <sub>4</sub>			66,230		622.21	979.40			0.8589		0.8968
295	Propane [use Eq. (2)]	C <sub>3</sub> H <sub>8</sub>	74-98-6	44.096	62.983	113,630	633.21	-873.46		85.47	0.8488	360.00	2.6079
296	1-Propanol	C <sub>3</sub> H <sub>8</sub> O	71-23-8	60.095	158,760	-635	1.969	0.0144		146.95	1.0797	400.00	2.1980
297	2-Propanol	C <sub>3</sub> H <sub>8</sub> O	67-63-0	60.095	471,710	-4,172.1	14.745	-0.0144		185.26	1.1329	355.30	2.0487
298	Propenylcyclohexene	$C_9H_{14}$	13511-13-2	122.207	201,400	-450.6	1.7053			199.00	1.7926	431.65	3.2463
299	Propionaldehyde	C <sub>3</sub> H <sub>6</sub> O	123-38-6	58.079	99,306	115.73	1			200.00	1.2245	328.75	1.3735

TABLE 2-153 Heat Capacities of Inorganic and Organic Liquids [J/(kmol·K)] (Concluded)

Cmpd.	Name	Formula	CAS no.	Mol. wt.	C1	C2	C3	C4	C5	$T_{\min}$ , K	$C_p$ at $T_{\min}$ $\times$ 1E-05	$T_{ m max}$ , K	$C_p$ at $T_{ m max}$ $ imes 1  ext{E-05}$
300	Propionic acid	$C_3H_6O_2$	79-09-4	74.079	213,660	-702.7	1.6605			252.45	1.4209	414.32	2.0756
301	Propionitrile	$C_3H_5O_2$ $C_3H_5N$	107-12-0	55.079	118,190	-120.98	0.42075			180.26	1.1005	370.50	1.3112
302	Propyl acetate	$C_5H_{10}O_2$	109-60-4	102.132	83,400	384.1	0.12010			274.70	1.8891	404.70	2.3885
303	Propyl amine	$C_3H_9N$	107-10-8	59.110	139,530	78				188.36	1.5422	340.00	1.6605
304	Propylbenzene	$C_9H_{12}$	103-65-1	120.192	174,380	-101.8	0.79			173.55	1.8051	432.39	2.7806
305	Propylene	$C_3H_6$	115-07-1	42.080	114,140	-343.72	1.0905			87.89	0.9235	225.45	0.9208
306	Propyl formate	$C_4H_8O_2$	110-74-7	88.105	75,700	326.1	1.0000			298.15	1.7293	398.15	2.0554
307	2-Propyl mercaptan	$C_3H_2S$	75-33-2	76.161	138,390	-117.11	0.47059			142.61	1.3126	350.00	1.5505
308	Propyl mercaptan	$C_3H_8S$	107-03-9	76.161	167,330	-319.1	0.8127			159.95	1.3708	340.87	1.5299
309	1,2-Propylene glycol	$C_3H_8O_2$	57-55-6	76.094	58,080	445.2				213.15	1.5297	460.75	2.6321
310	Ouinone	$C_6H_4O_2$	106-51-4	108.095	45,810	368.33				388.85	1.8904	683.00	2.9738
311	Silicon tetrafluoride	F <sub>4</sub> Si	7783-61-1	104.079	829,380	-7,331.5	19.203			186.35	1.3000	253.15	2.0403
312	Styrene	$C_8H_8$	100-42-5	104.149	113,340	290.2	-0.6051	0.0013567		242.54	1.6749	418.31	2.2816
313	Succinic acid	$C_4H_6O_4$	110-15-6	118.088	244,770	-236.96	0.63148			460.65	2.6961	604.50	3.3228
314	Sulfur dioxide	$O_2S$	7446-09-5	64.064	85,743	5.7443				197.67	0.8688	350.00	0.8775
315	Sulfur hexafluoride	F <sub>6</sub> S	2551-62-4	146.055	119,500					230.15	1.1950	230.15	1.1950
316	Sulfur trioxide	$O_3S$	7446-11-9	80.063	258,090					303.15	2.5809	303.15	2.5809
317	Terephthalic acid	$C_8H_6O_4$	100-21-0	166.131	,								
318	o-Terphenyl	$C_{18}H_{14}$	84-15-1	230.304	182,900	635.09				329.35	3.9207	609.15	5.6977
319	Tetradecane	$C_{14}H_{30}$	629-59-4	198.388	353,140	29.13	0.86116			279.01	4.2831	526.73	6.0741
320	Tetrahydrofuran	$C_4H_8O$	109-99-9	72.106	171,730	-800.47	2.8934	-0.0025015		164.65	1.0721	339.12	1.3546
321	1,2,3,4-Tetrahydronaphthalene	$C_{10}H_{12}$	119-64-2	132.202	81,760	455.38				237.38	1.8986	480.77	3.0069
322	Tetrahydrothiophene	$C_4H_8S$	110-01-0	88.171	123,300	-130.1	0.6229			176.98	1.1979	394.27	1.6883
323	2,2,3,3-Tetramethylbutane	$C_8H_{18}$	594-82-1	114.229	43,326	630.73				375.41	2.8011	426.00	3.1202
324	Thiophene	$C_4H_4S$	110-02-1	84.140	84,864	91.725	0.13243			234.94	1.1372	357.31	1.3455
325	Toluene	$C_7H_8$	108-88-3	92.138	140,140	-152.3	0.695			178.18	1.3507	500.00	2.3774
326	1,1,2-Trichloroethane	$C_2H_3Cl_3$	79-00-5	133.404	103,350	159.3				236.50	1.4102	300.00	1.5114
327	Tridecane	$C_{13}H_{28}$	629-50-5	184.361	350,180	-104.7	1.0022			267.76	3.9400	508.62	5.5619
328	Triethyl amine	$C_6H_{15}N$	121-44-8	101.190	111,480	368.13				200.00	1.8511	361.92	2.4471
329	Trimethyl amine	$C_3H_9N$	75-50-3	59.110	136,050	-288	0.9913			156.08	1.1525	276.02	1.3208
330	1,2,3-Trimethylbenzene	$C_9H_{12}$	526-73-8	120.192	119,450	324.54				247.79	1.9987	449.27	2.6526
331	1,2,4-Trimethylbenzene	$C_9H_{12}$	95-63-6	120.192	178,800	-128.47	0.83741			229.33	1.9338	350.00	2.3642
332	2,2,4-Trimethylpentane	$C_8H_{18}$	540-84-1	114.229	95,275	696.7	-1.3765	0.0021734		165.78	1.8285	520.00	3.9095
333	2,3,3-Trimethylpentane	$C_8H_{18}$	560-21-4	114.229	388,620	-1,439.5	3.2187			280.00	2.3791	320.00	2.5757
334	1,3,5-Trinitrobenzene	$C_6H_3N_3O_6$	99-35-4	213.105	40,364	664.46				398.40	3.0508	475.47	3.5629
335	2,4,6-Trinitrotoluene	$C_7H_5N_3O_6$	118-96-7	227.131	133,530	514.64				354.00	3.1571	475.00	3.7798
336	Undecane	$C_{11}H_{24}$	1120-21-4	156.308	293,980	-114.98	0.96936			247.57	3.2493	433.42	4.2624
337	1-Undecanol	$C_{11}H_{24}O$	112-42-5	172.308	129,450	-3,039.5	27.927	-0.061847	4.3042E-05	289.05	3.9103	520.30	5.5127
338	Vinyl acetate	$C_4H_6O_2$	108-05-4	86.089	136,300	-106.17	0.75175			259.56	1.5939	389.35	2.0892
339	Vinyl acetylene	$C_4H_4$	689-97-4	52.075	68,720	135				200.00	0.9572	278.25	1.0628
340	Vinyl chloride	$C_2H_3Cl$	75-01-4	62.498	-10,320	322.8				200.00	0.5424	400.00	1.1880
341	Vinyl trichlorosilane	C <sub>2</sub> H <sub>3</sub> Cl <sub>3</sub> Si	75-94-5	161.490	49,516	420.35				178.35	1.2449	363.85	2.0246
342	Water	$H_2O$	7732-18-5	18.015	276,370	-2,090.1	8.125	-0.014116	9.3701E-06	273.16	0.7615	533.15	0.8939
343	m-Xylene	$C_8H_{10}$	108-38-3	106.165	133,860	7.8754	0.52265			217.00	1.6018	540.15	2.9060
344	o-Xylene	$C_8H_{10}$	95-47-6	106.165	36,500	1,017.5	-2.63	0.00302		247.98	1.7314	417.58	2.2269
345	p-Xylene	$C_8H_{10}$	106-42-3	106.165	-35,500	1,287.2	-2.599	0.002426		286.41	1.7697	600.00	3.2520

For the 11 substances, ammonia, 1,2-butanediol, 1,3-butanediol, carbon monoxide, 1,1-difluoroethane, ethane, heptane, hydrogen, hydrogen sulfide, methane, and propane, the liquid heat capacity  $C_{\rm pL}$  is calculated with Eq. (2) below. For all other compounds, Eq. (1) is used. For benzene, fluorine, and helium, two sets of constants are given for Eq. (1) that cover different temperature ranges, as shown in the table.

(1) 
$$C_{\rm pL} = C1 + C2T + C3T^2 + C4T^3 + C4T^4$$

(2) 
$$C_{\text{pL}} = \frac{\text{C1}^2}{t} + \text{C2} - 2\text{C1C3}t - \text{C1C4}t^2 - \frac{\text{C3}^2t^3}{3} - \frac{\text{C3C4}t^4}{2} - \frac{\text{C4}^2t^5}{5}$$

where  $t = 1 - T_r$ ,  $T_r = T/T_c$ , is the critical temperature from Table 2-141,  $C_{\rm pL}$  is in  $J/({\rm kmol\cdot K})$  and T is in K. All substances are listed by chemical family in Table 2-6 and by formula in Table 2-7. For temperatures less than the normal boiling point, the pressure is 1 atm. Above the normal boiling point, the pressure.

Values in this table were taken from the Design Institute for Physical Properties (DIPPR) of the American Institute of Chemical Engineers (AIChE), copyright 2007 AIChE and reproduced with permission of AICHE and of the DIPPR Evaluated Process Design Data Project Steering Committee. Their source should be cited as R. L. Rowley, W. V. Wilding, J. L. Oscarson, Y. Yang, N. A. Zundel, T. E. Daubert, R. P. Danner, DIPPR® Data Compilation of Pure Chemical Properties, Design Institute for Physical Properties, AIChE, New York (2007).

The number of digits provided for values at  $T_{\min}$  and  $T_{\max}$  was chosen for uniformity of appearance and formatting; these do not represent the uncertainties of the physical quantities, but are the result of calculations from the standard thermophysical property formulations within a fixed format.