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Unit III: Application of Quantum Mechanics to Electrical transport in Solids



- > Suggested Reading
 - 1. Fundamentals of Physics, Resnik and Halliday, Chapter 41
 - 2. Solid state Physics, S.O Pillai, Chapter 6
 - 3. Concepts of Modern Physics, Arthur Beiser, Chapter 9
 - 4. Learning material prepared by the department-Unit III

- > Reference Videos
 - 1. Physics Of Materials-IIT-Madras/lecture-26.html

Unit III: Application of Quantum Mechanics to Electrical transport in Solids



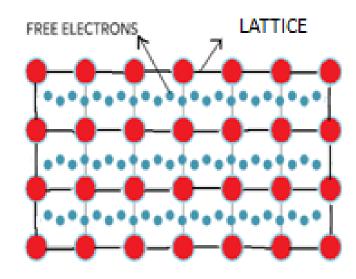
Class #28

Merits of Quantum free electron theory:

- > Temperature dependence of resistivity
- > Heat capacity due to free electrons
- > Conductivity dependence on electron concentration

Temperature dependence of Resistivity





- ➤ According to the CFET ionic centers are stationary and the resistivity originates from the scattering of the free electrons.
- The QFET takes into account of the thermal vibrations of the ionic array which accounts for the scattering of electrons.

Temperature dependence of Resistivity

- \triangleright The amplitude of the vibrations r of the lattice ions increase with increasing temperature.
- \triangleright When ions vibrate the lattice presents an effective cross sectional area $A = 2r^2$ for scattering.
- \triangleright Effective cross sectional area A is proportional to temperature A α T
- > This will increase the probability of electron scattering at higher temperatures .
- \succ Results in reduction of mean free path λ of the electrons $\lambda \propto 1/T$



Temperature dependence of Resistivity



Expression for is conductivity
$$\sigma = \frac{ne^2\tau}{m} = \frac{ne^2\lambda}{mv_f}$$

This implies that the conductivity is proportional to the mean free path

The electron mean free path λ is inversely proportional to the scattering cross section.

Hence the conductivity $\sigma \alpha 1/T$ or resistivity $\rho \alpha T$

Heat capacity due to free electrons





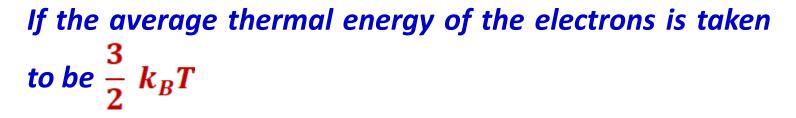
U = No of conduction electrons * Average energy

At temperature T electrons can gain an energy K_BT , but only electrons close to the Fermi level participate in the conduction.



Heat capacity due to free electrons

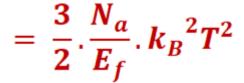




Then the total energy of electrons in one mole of the material is given by $U=n_{eff}$. $\frac{3}{2}k_BT$ $=\frac{3}{2}k_BT \cdot \frac{N_a}{E_f} \cdot k_BT$



Heat capacity due to free electrons



Electronic specific heat is given by $C_{el} = \frac{dU}{dT}$

$$=3.\frac{N_a}{E_f}.k_B^2T=3R.\frac{k_BT}{E_f}$$

A more accurate evaluation of the electronic specific heat of mono valent metals results in the relation

$$C_{el} = \frac{\pi^2}{2} N \left(\frac{k_B^2 T}{E_f} \right)$$

This analysis gives the correct correlation with the experimental results.



Conductivity variations with electron concentrations



According to QFET the conductivity expression is given by

$$\sigma = \frac{ne^2\tau}{m}$$
 or $= \frac{ne^2\left(\frac{\lambda}{v_F}\right)}{m}$

According to the above equation σ not only depends on the number of electrons per unit volume but also depends on the λ/v_F ratio.

Class 28. Quiz ...

The concepts which are correct are....

- 1. The quantum free electron theory takes accounts of the thermal vibrations of the ionic array, which accounts for the scattering of electrons.
- 2. The probability of electron scattering decreases with increasing temperature.
- 3. According to QFET resistivity of the metal is found to be inversely proportional to temperature.
- 4. For a given mole of electron gas, the fraction of electron that gain energy KT is $n_{eff} = \frac{N_a}{E_f} \ kT$





THANK YOU

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