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Unit I: Review of concepts leading to Quantum Mechanics



Week #3 Class #10

- Concept of observables of the state of a system
- Operators in quantum mechanics
- Expectation values of observables

Unit I: Review of concepts leading to Quantum Mechanics



- > Suggested Reading
 - 1. Concepts of Modern Physics, Arthur Beiser, Chapter 5
 - 2. Learning Material prepared by the Department of Physics
- > Reference Videos
 - 1. Video lectures: MIT 8.04 Quantum Physics I

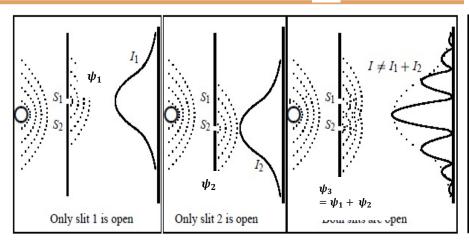
Observables

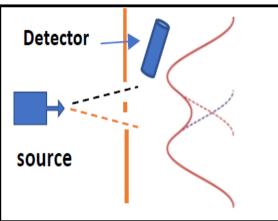
- All experimentally measurable parameters of a physical system are observables
 - **Position**
 - > momentum
 - > Energy of a state
 - > life time of electrons
 - > Spin of a system
- Multiple measurements yield average values of the parameters



Double slit experiment revisited

- Quantum mechanically observations on physical systems do not yield information about the system
- Observations lead to a collapse of the wave function double slit experiment with an "observer"
- Observables in quantum mechanics have to be estimated probabilistically







Operators

- Wave functions contain information about the quantum system
- Mathematical operators can be used to extract information about the physical state in terms of the observables



Operators

- A normalized wave function contain information about the quantum system $\psi = Ae^{\frac{i}{\hbar}(px-Et)}$ eigen function
- A mathematical operator $\widehat{\textbf{G}}$ operating on the wavefunction can result in the eigen value G of the observable
- The eigen value equation $\hat{G}\psi = G\psi$



Operators

Momentum operator -



$$\frac{\partial \psi}{\partial x} = (\frac{ip}{\hbar})\psi$$

$$\left\{-i\hbar\frac{\partial}{\partial x}\right\}\psi=p\psi$$

• The momentum operator $\hat{p} = \left\{-i\hbar \frac{\partial}{\partial x}\right\}$

Operating on the eigen function yields the momentum eigen value



Operators

Kinetic energy operator -

• The second derivative of ψ with respect to position yields

$$\frac{\partial^2 \psi}{\partial x^2} = \left(\frac{ip}{\hbar}\right)^2 \psi$$
$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} = \frac{p^2}{2m} \psi = KE\psi$$

The kinetic energy operator
$$\widehat{KE} = \left\{ -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \right\}$$

operating on the eigen function yields the eigen value of the kinetic energy of quantum system



Operators

Total energy operator -



$$\frac{\partial \psi}{\partial t} = \left(-\frac{iE}{\hbar}\right)\psi$$
$$\left\{i\hbar\frac{\partial}{\partial t}\right\}\psi = E\psi$$

The total energy operator
$$\widehat{\pmb{E}} = \left\{ \pmb{i}\hbar \frac{\partial}{\partial t} \right\}$$

Operating on the eigen function yields the eigen value of the total energy of quantum system

This is also called as the Hamiltonian operator $\widehat{m{H}}$



Operators

Position operator -

- The position operator has to be discussed in the momentum space
- The position operator \widehat{x} operating on ψ

$$\widehat{x}\psi = x\psi$$

yields the eigen value of position of the quantum system



Operators

Potential energy operator -

- Potential energy operator is not explicitly described
- The eigen value of the potential energy can be inferred as the difference of the total energy and the kinetic energy
- The eigen value equation for the potential energy is

$$\widehat{V}\psi = V\psi$$



Expectation values of observables

Quantum mechanics predicts only the most probable values of the observables of a physical system

 the expectation values ≡ the average of repeated measurements on the system.

The eigen value equation for momentum

$$\widehat{p}\psi = p\psi$$

The operation $\psi^*\widehat{p}\psi=\psi^*p\psi=p\psi^*\psi$

 $\psi^*\psi$ is the probability density

 $p\psi^*\psi$ should be the probability of the eigen value



Expectation values of observables

The spread in the wave packet can yield a range of $p\psi^*\psi$

Integrated over the range of *x* for the extend of the wave packet

$$\int \psi^* \widehat{p} \psi \, dx = \int \psi^* p \psi \, dx = \langle p \rangle \int \psi^* \psi \, dx$$

 $\langle p \rangle$ is the most probable value of the momentum.

Thus the expectation value of the momentum is written as

$$\langle \boldsymbol{p} \rangle = \frac{\int \boldsymbol{\psi}^* \widehat{\boldsymbol{p}} \boldsymbol{\psi} \, dx}{\int \boldsymbol{\psi}^* \boldsymbol{\psi} \, dx}$$



Expectation values of observables

In general an operator \hat{G} of the observable g

Gives the expectation value of the observable

$$\langle g \rangle = \frac{\int \boldsymbol{\psi}^* \widehat{\boldsymbol{G}} \boldsymbol{\psi} \, dx}{\int \boldsymbol{\psi}^* \boldsymbol{\psi} \, dx}$$

And in three dimensional space

$$\langle g \rangle = \frac{\int \Psi^* \widehat{G} \Psi \, dV}{\int \Psi^* \Psi \, dV}$$



Class #10 Quiz....



Understanding concepts of observables, operators and expectation values

- 1. Observables in quantum mechanics are obtained as results of experiments
- 2. The total energy operator is $\left\{i\hbar\frac{\partial}{\partial t}\right\}$
- 3. The position operator as obtained from the momentum space would be $\left\{-i\hbar\frac{\partial}{\partial p}\right\}$
- 4. The total energy operator is Hamiltonian operator
- 5. Expectation value of position in 1d is $\langle x \rangle = \frac{\int \psi^* \hat{x} \psi dx}{\int \psi^* \psi dx}$



THANK YOU

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