



**Aalto University**  
School of Chemical  
Engineering

# Sensitivity analysis of a control-oriented model of supercritical extraction

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What are advantages of supercritical fluid extraction?

- $CO_2$  non-flammable and non-toxic
- properties of  $CO_2$  can be 'tuned' with operating conditions
- the critical point of  $CO_2$  is easily accessible
- high selectivity
- easy separation of the solvent (if there is no co-solvent)

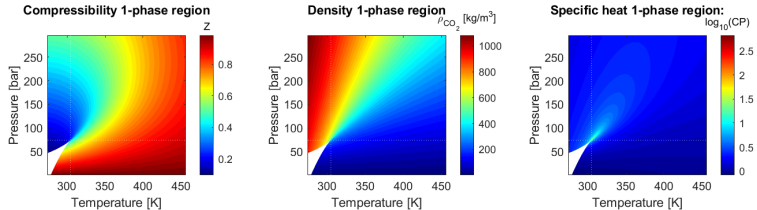
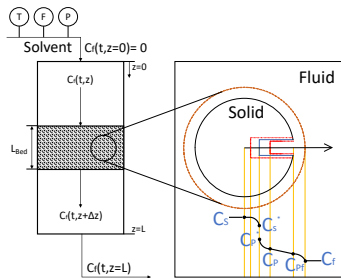


Figure 1: Properties of  $CO_2$



**Figure 2:** Schematic representation of SFE process

## Assumptions:

- Quasi-one-dimensional model
- Negligible external diffusion
- Low-Mach number assumption
- No pressure drop
- Pseudo-single component
- Negligible thermal properties of solid phase
- Uniform distribution of the particles
- Varying void fraction along the extractor

# Model of an extractor

(1) Fluid phase mass balance [1]

$$\frac{\partial c_f(t, z)}{\partial t} + \underbrace{\frac{1}{\phi} \frac{\partial (c_f(t, z)u)}{\partial z}}_{\text{Convection}} = \underbrace{\frac{1 - \phi}{\phi} r_e(t, z)}_{\text{Kinetic}} + \underbrace{\frac{1}{\phi} \frac{\partial}{\partial z} \left( D_e^M \frac{\partial c_f(t, z)}{\partial z} \right)}_{\text{Diffusion}}$$

(2) Solid phase mass balance [1]

$$\frac{\partial c_s(t, z)}{\partial t} = - \underbrace{r_e(t, z)}_{\text{Kinetic}}$$

(3) Heat balance [2]

$$\frac{\partial (\rho_f(T(t, z), P(t))h(t, z)A_f)}{\partial t} - \frac{\partial (P(t)A_f)}{\partial t} = - \underbrace{\frac{\partial (\rho_f(T(t, z), P(t))h(t, z)A_f v)}{\partial z}}_{\text{Convection}} + \underbrace{\frac{\partial}{\partial z} \left( k \frac{\partial T(t, z)}{\partial z} \right)}_{\text{Diffusion}}$$

Extraction kinetic [1]

$$r_e(t, z) = - \frac{D_i}{\mu l^2} \left( c_s(t, z) k_m - c_f(t, z) \right)$$

Output function [3]

$$y(t) = g(x(t)) = \int_0^{t_f} Q(t) c_f(t, z)|_{z=L} dt = \int_0^{t_f} u(t) A c_f(t, z)|_{z=L} dt = \iff \frac{\partial y(t)}{\partial t} = u(t) A c_f(t, z)|_{z=L}$$

Equation of state - Peng-Robinson

$$P(T(t, z), \rho(t, z)) = \frac{RT(t, z)}{v(t, z) - b} - \frac{a}{v(t, z)(v(t, z) + b) + b(v(t, z) - b)}; \quad v(t, z) = \frac{1}{\rho(t, z)}$$

[1] E. Reverchon, Mathematical modeling of supercritical extraction of sage oil, AIChE J 42 (6), 1996

[2] John D. Anderson. Computational fluid dynamics the basic with applications. McGraw-Hill, 1995

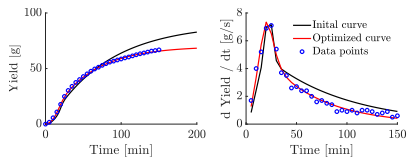
[3] H. Sovova, R. Komers, J. Kucnera, and J. Jezou. Supercritical carbon dioxide extraction of caraway essential oil. Chemical Engineering Science, 1994

# Model discretization

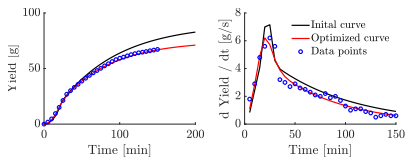
$$\frac{dx(t)}{dt} = \begin{bmatrix} \frac{dc_{f_1}(t)}{dt} \\ \vdots \\ \frac{dc_{f_{N_z}}(t)}{dt} \\ \frac{dc_{s_1}(t)}{dt} \\ \vdots \\ \frac{dc_{s_{N_z}}(t)}{dt} \\ \frac{dh_1(t)}{dt} \\ \vdots \\ \frac{dh_{N_z}(t)}{dt} \\ \frac{dP(t)}{dt} \\ \frac{dy(t)}{dt} \end{bmatrix} = \begin{bmatrix} F_1(c_f(t), c_s(t), T(t), y(t); \Theta) \\ \vdots \\ F_{N_z}(c_f(t), c_s(t), T(t), y(t); \Theta) \\ F_{N_z+1}(c_f(t), c_s(t), T(t), y(t); \Theta) \\ \vdots \\ F_{2N_z}(c_f(t), c_s(t), T(t), y(t); \Theta) \\ F_{2N_z+1}(c_f(t), c_s(t), T(t), y(t); \Theta) \\ \vdots \\ F_{3N_z}(c_f(t), c_s(t), T(t), y(t); \Theta) \\ F_{3N_z+1}(c_f(t), c_s(t), T(t), y(t); \Theta) \\ F_{3N_z+2}(c_f(t), c_s(t), T(t), y(t); \Theta) \end{bmatrix} = F(t; \Theta)$$

where  $x \in \mathbb{R}^{N_x=3N_z+2}$  and  $\Theta \in \mathbb{R}^{N_\Theta=N_\theta+N_u}$

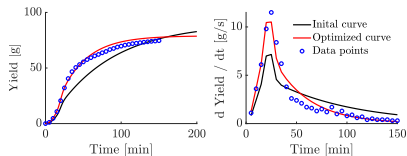
# Results of the simulation



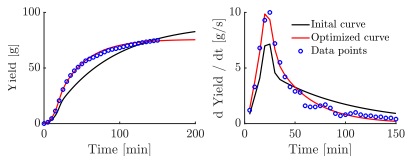
(a) Experiment at 40°C and 200 bar



(b) Experiment at 50°C and 200 bar



(c) Experiment at 40°C and 300 bar



(d) Experiment at 50°C and 300 bar

# Sensitivity Analysis

Local derivative-based methods involve taking the partial derivative of the output with respect to an input parameter. In sensitivity analysis the original system of equations is solved simultaneously with  $\frac{\partial F(x(t); \Theta)}{\partial \Theta}$ , where  $\Theta$  is a vector of all the parameters and the control variables.

$$\begin{aligned}
 Z &= \frac{\partial x(t, \Theta)}{\partial \Theta} \\
 \dot{Z} &= \frac{\partial Z(t, \Theta)}{\partial t} = \frac{\partial}{\partial t} \left( \frac{\partial x(t, \Theta)}{\partial \Theta} \right) = \frac{\partial}{\partial \Theta} \left( \frac{\partial x(t, \Theta)}{\partial t} \right) = \frac{dF(x(t); \Theta)}{d\Theta} \\
 \frac{dF(x(t); \Theta)}{d\Theta} &= \frac{\partial F(x(t); \Theta)}{\partial x(t)} \frac{\partial x(t)}{\partial \Theta} + \frac{\partial F(x(t); \Theta)}{\partial \Theta} \\
 &= J_x(x(t); \Theta) S(x(t); \Theta) + J_\Theta(x(t); \Theta)
 \end{aligned}$$

$$\boxed{\frac{dF(x(t); \Theta)}{d\Theta}} = \boxed{\frac{\partial F(x(t); \Theta)}{\partial x(t)}} \boxed{\frac{\partial x(t)}{\partial \Theta}} + \boxed{\frac{\partial F(x(t); \Theta)}{\partial \Theta}}$$

$(N_x \times N_\Theta)$        $(N_x \times N_x)$        $(N_x \times N_\Theta)$        $(N_x \times N_\Theta)$

# Results of the sensitivity analysis - Internal diffusion

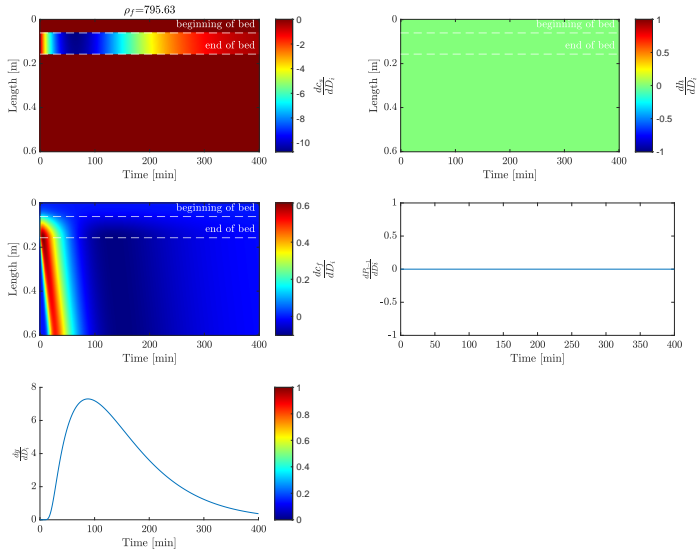


Figure 4: 50°C, 200 bar



# Results of the sensitivity analysis - Flow-rate

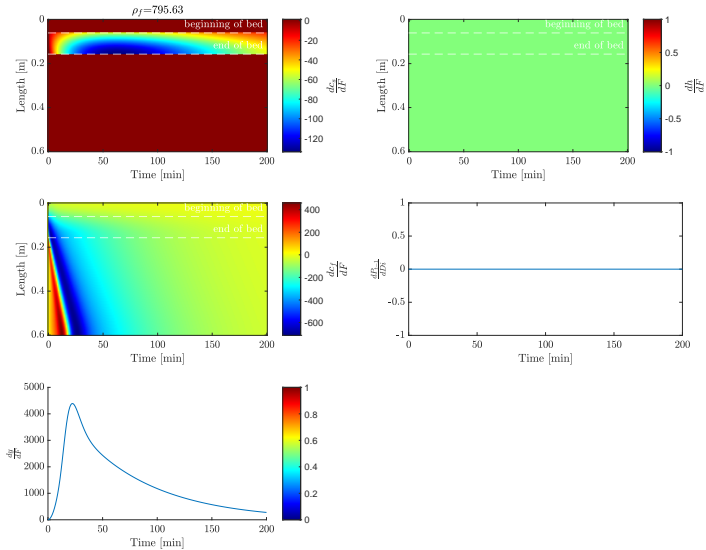


Figure 5: 50°C, 200 bar

## Results of the sensitivity analysis - Flow-rate

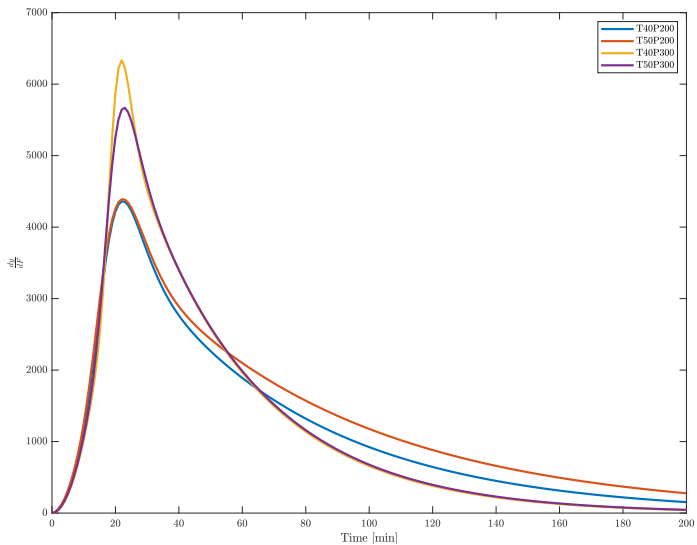


Figure 6: 40 – 50°C, 200 – 300 bar

The forward sensitivity analysis allows to

- better understanding of the analysed model
- evaluated an influence of controls and parameters
- identify sensitive and none-sensitive parameters
- spot modelling errors