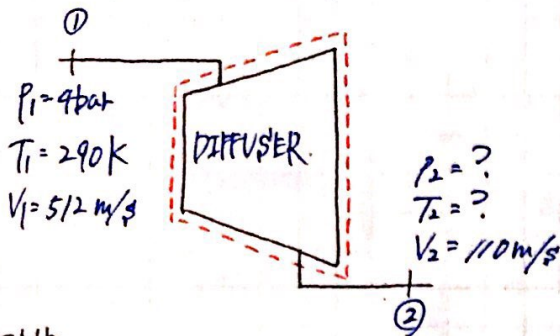


<1>

GIVEN & CFFD AIR



FIND

- (a) P_2 , T_2 when $k = 1.4$
 (b) using data from table

ASSUMP

open sys, SSST, 1-DUF, $\dot{Q} = 0$, $\Delta P_E = 0$, $\dot{W} = 0$, $\Delta S = 0$ (isotropic), ideal gas

EQN

$$\frac{dm}{dt}|_{\text{sys}} = \sum_i \dot{m}_i - \sum_e \dot{m}_e, \quad \frac{dE}{dt}|_{\text{sys}} = \dot{Q} - \dot{W} + \sum_i \dot{m}_i (h + pe + ke) - \sum_e \dot{m}_e (h + pe + ke)$$

$$\frac{ds}{dt}|_{\text{sys}} = \sum_i \frac{\dot{Q}_i}{T} + \sum_i \dot{m}_i s_i - \sum_e \dot{m}_e s_e + \dot{S}_{\text{gen}}, \quad \frac{T_2}{T_1} = \left(\frac{P_2}{P_1}\right)^{\frac{k-1}{k}}, \quad P_2 = P_{\text{air}} T, \quad P_{\text{air}} = 0.287 \text{ kJ/kg}\cdot\text{K}$$

SOLN

$$\dot{m}_1 = \dot{m}_2 = \dot{m} \quad \dots \textcircled{1}$$

$$0 = \dot{m} \left(h_1 + \frac{V_1^2}{2} - h_2 - \frac{V_2^2}{2} \right) \quad \dots \textcircled{2}$$

(a)

from table $h_1 = 290.1 \text{ kJ/kg}$

using ②

$$h_2 = h_1 + \frac{V_1^2}{2} \times 10^{-3} - \frac{V_2^2}{2} \times 10^{-3}$$

$$= 290.1 \text{ kJ/kg} + (512 \text{ m/s})^2 \times \frac{10^{-3}}{2} - (110 \text{ m/s})^2 \times \frac{10^{-3}}{2} = 415.022 \text{ kJ/kg}$$

from table

$$T_2 = 415 \text{ K}$$

and

$$P_2 = P_1 \left(\frac{T_2}{T_1} \right)^{\frac{k}{k-1}} = (4 \text{ bar}) \left(\frac{415 \text{ K}}{290 \text{ K}} \right)^{\frac{1.4}{0.4}} \approx 14.02$$

$$P_2 = 14.0 \text{ bar}$$

(b) if $T_2 = 415 \text{ K}$ from tables

$$P_{r1} = 1.2311$$

$$\text{interpolation } P_{r2} = (415 \text{ K} - 410 \text{ K}) \frac{(4.522 - 1.2311)}{(420 \text{ K} - 410 \text{ K})} + 1.2311 = 4.3375$$

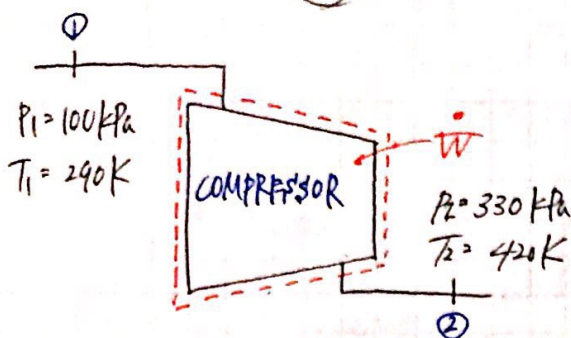
then

$$\frac{P_2}{P_1} = \frac{P_{r2}}{P_{r1}} \Leftrightarrow P_2 = P_1 \left(\frac{P_{r2}}{P_{r1}} \right) = (4 \text{ bar}) \left(\frac{4.3375}{1.2311} \right) \approx 14.0 \text{ bar}$$

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GIVEN $\delta < \text{FFD}$ (AIR)

FIND

(a) specific work, w , kJ/kg (b) $\frac{\dot{Q}_{\text{gen}}}{\dot{m}}$, kJ/kg-K (c) isentropic efficiency η 

ASSUMP

open sys, SSSF, 1-DUF, $\dot{Q} = 0$, $\Delta P_E = \Delta P_F = 0$, ideal gas

EQN

$$\frac{dm}{dt}|_{\text{sys}} = \sum \dot{m}_i - \sum \dot{m}_e, \quad \frac{dE}{dt}|_{\text{sys}} = \dot{Q} - \dot{W} + \sum \dot{m}_i(h + pe + ke) - \sum \dot{m}_e(h + pe + ke)$$

$$\frac{ds}{dt}|_{\text{sys}} = \sum \frac{\dot{Q}_j}{T_j} + \sum \dot{m}_i s_i - \sum \dot{m}_e s_e + \dot{J}_{\text{gen}}, \quad p_2 = p_{\text{air}} T \quad (p_{\text{air}} = 0.287 \text{ kJ/kg-K})$$

SOLN

$$\dot{m}_1 = \dot{m}_2 = \dot{m} \quad \dots (1)$$

from table

$$\dot{w} = h_1 - h_2 \quad \dots (2)$$

$$h_1 = 290.1 \text{ kJ/kg} \quad s_1^0 = 1.669 \text{ kJ/kg-K}$$

$$h_2 = 421.4 \text{ kJ/kg} \quad s_2^0 = 2.042 \text{ kJ/kg-K}$$

$$\frac{\dot{J}_{\text{gen}}}{\dot{m}} = s_2 - s_1$$

$$= s_2^0 - s_1^0 - R \ln \frac{p_2}{p_1} \quad \dots (3)$$

(a)

$$\therefore (2) \quad \dot{w} = 290.1 \text{ kJ/kg} - 421.4 \text{ kJ/kg} = -131.3 \text{ kJ/kg} \approx \boxed{-131 \text{ kJ/kg}}$$

(b) $\therefore (3)$

$$\frac{\dot{J}_{\text{gen}}}{\dot{m}} = 2.042 \text{ kJ/kg-K} - 1.669 \text{ kJ/kg-K} - (0.287 \text{ kJ/kg-K}) \ln \left(\frac{330 \text{ kPa}}{100 \text{ kPa}} \right) \approx \boxed{0.030344 \text{ kJ/kg-K}}$$

(c) if $\Delta S = 0$

$$\text{using (3)} \quad s_{2s}^0 = s_1^0 + R \ln \frac{p_2}{p_1} = 1.669 \text{ kJ/kg-K} + (0.287 \text{ kJ/kg-K}) \ln \left(\frac{330 \text{ kPa}}{100 \text{ kPa}} \right) \approx 2.0117$$

the corresponding (Temp.) \rightarrow enthalpy h_{2s} will be

$$\text{interpolating} \quad h_{2s} = (2.0117 \text{ kJ/kg-K} - 1.993 \text{ kJ/kg-K}) \frac{(411.2 - 401.1) \text{ kJ/kg}}{(2.018 - 1.993) \text{ kJ/kg-K}} + 401.1 \text{ kJ/kg} \approx 408.65 \text{ kJ/kg}$$

$$\therefore \dot{w}_{\text{isentropic}} = h_1 - h_{2s} = (290.1 - 408.65) \text{ kJ/kg} = -118.55 \text{ kJ/kg}$$

$$\text{then} \quad \eta = \frac{\dot{w}_{\text{isentropic}}}{\dot{w}} = \frac{-118.55 \text{ kJ/kg}}{-131.3 \text{ kJ/kg}} \approx 0.9029$$

$$\boxed{\eta = 90.3 \%}$$