# CH-412-LA Experiment 7

#### Separation of Metal Ions

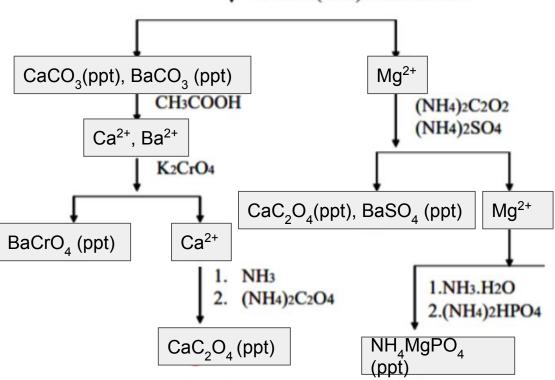
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I pledge my honor that I have abided by the Stevens Honor System.

## **Reaction Flowchart**

#### Ba Mg Ca ion solution

- 1. Add NH<sub>4</sub>Cl/NH<sub>3</sub> buffer, pH=9.0
- 2. Add (NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub> solution



### **Post Lab Questions**

1. What is the flame test for each Ba2+, Ca2+, Mg2+ ions?

The flame test is used to determine the presence of a metal ion by the color a flame turns when exposed to a metal. Ba2+ will have a green flame, Ca2+ will have an orange flame, and Mg2+ will have a white flame.

2. Draw a flow chart to show separation of Ag, Cu, Zn, and Sr metalions. (flow chart on the next slide)

