

a. Steady $\Rightarrow \dot{E}_{in} = \dot{E}_{out}$

$$\dot{Q}_{in} + \dot{W}_{in} + \dot{m} \left(h + \frac{V^2}{2} + gz \right)_{in} = \dot{Q}_{out} + \dot{W}_{out} + \dot{m} \left(h + \frac{V^2}{2} + gz \right)_{out}$$

$$\dot{m} \left(h + \frac{V^2}{2} \right)_{in} = \underbrace{\dot{Q}_{out} - \dot{Q}_{in}} + \dot{W}_{out} + \dot{m} \left(h + \frac{V^2}{2} \right)_{out}$$

$$\dot{m} \left(h + \frac{V^2}{2} \right)_{in} = \Delta Q + \dot{W}_{out} + \dot{m} \left(h + \frac{V^2}{2} \right)_{out}$$

~~For~~ $h_{out} = h_g @ 100 \text{ kPa} = 234.46 \text{ kJ/kg}$, $V_{out} = 20 \text{ m/s}$

$h_{in} = 340.55 \text{ kJ/kg}$, $V_{in} = 100 \text{ m/s}$

$$1500 \left(340.55 + \frac{100^2}{2 \cdot 1000} \right) = \Delta Q + 50000 + 1500 \left(234.46 + \frac{20^2}{2 \cdot 1000} \right)$$

$$518325 = \Delta Q + 50000 + 351900$$

$$116335 = \Delta Q \Rightarrow \dot{Q}_{out} > \dot{Q}_{in} \Rightarrow \text{this is a } \dot{Q}_{out}$$

$$116335 \text{ W} = 116.335 \text{ kW}$$

b. $\Delta S = \dot{m}(s_2 - s_1) = 1500(1.0836 - 0.9591) = 197.535 \text{ kW/K}$

$$s_2 = 0.9591$$

$$s_1 = 1.0836$$

$$\dot{Q}_{out} = -116335$$

$$\Delta S = \frac{\dot{Q}}{T} + S_{gen}$$

$$\Delta S_{out} = -197.535$$

$$-197.535 = \frac{-116335}{25+273.15} + S_{gen}$$

$$S_{gen} = 192.65 \text{ kW/K}$$

c. ~~Entropy generation is from the second law, as this is not a reversible system, Energy has a flow in this system, so~~

1. Entropy generation is due to this system being irreversible, as some parts of this turbine are irreversible, such as the energy conversion to \dot{W}_{out} .