Prof. Ernst Otto Fischer and Prof. Geoffrey Wilkinson

Max Shi and Jeremy Meyerberg

Prof. Fischer and Prof. Wilkinson



Professor Ernst Otto Fischer
Technical University, Munich, Federal Republic
of Germany
November 10, 1918 - July 23, 2007

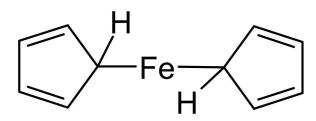


Professor Geoffrey Wilkinson Imperial College, London, United Kingdom July 14, 1921 - September 26, 1996

1973 Nobel Prize for Chemistry

- Awarded jointly "for their pioneering work, performed independently, on the chemistry of organometallic, so called sandwich compounds."
- Prepared dicyclopentadienyl iron (ferrocene) in 1951, at the same time
- Compound was extraordinarily stable chemically and thermally
- Insufficiently explained by ionic-covalent resonance





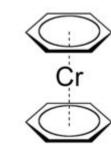
Previously proposed structure of ferrocene

1973 Nobel Prize for Chemistry (cont.)

- Fischer used x-ray crystal structure determination to establish true geometry of compound.
- Wilkinson used chemical, physical, and spectroscopic (NMR) studies for the same conclusion.
- Contributed to the creation of the field of "metallocene chemistry"
- Fischer and Wilkinson were also consistent contributors to the new field
 - Fischer went on to create bis-benzene chromium
 - Wilkinson also created Wilkinson's catalyst, used for hydrogenation of alkenes

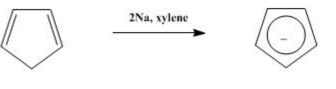
Ph₃P—Rh—CI Ph₃P

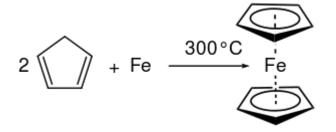
Wilkinson's Catalyst

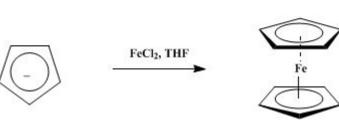


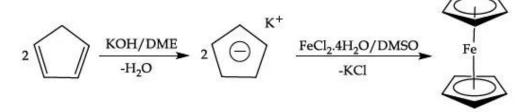


Mechanism of the Reaction



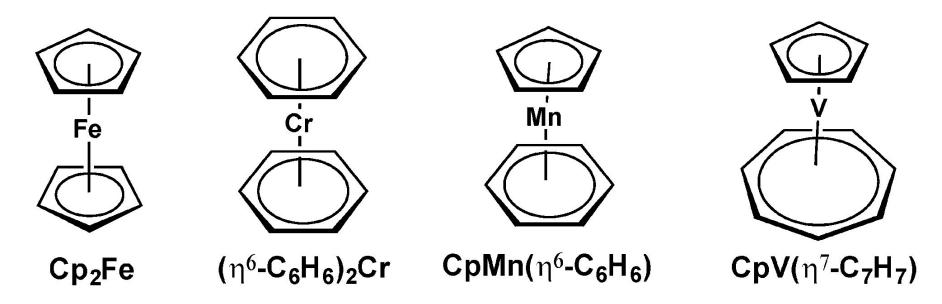




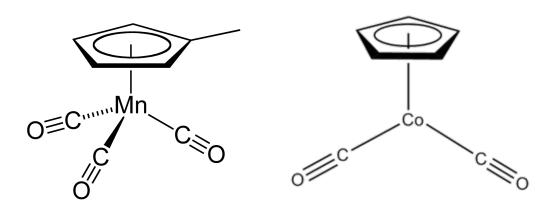


 Reaction is between the d-orbitals of Fe²⁺ with the p-orbitals of two cyclopentadienyl ligands

Examples of Sandwich Compounds

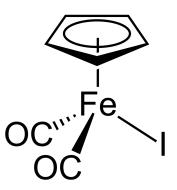


Half-Sandwich Compounds



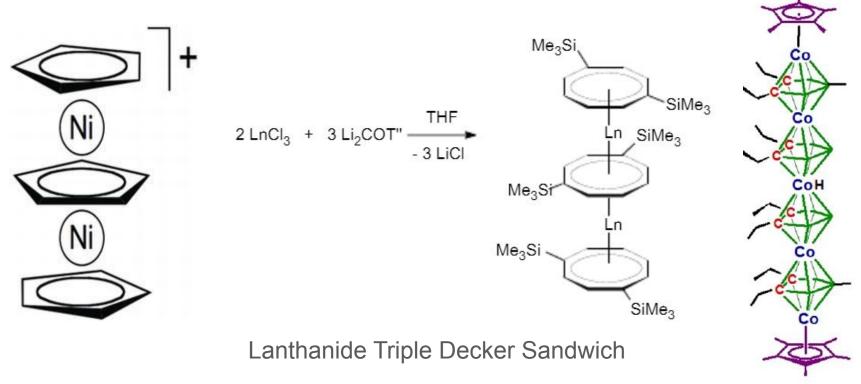


Catalyst for Pyridine Synthesis



Intermediate in preparation of organoiron compounds.

Multidecker Sandwich Compounds



Cobaltacarborane hexadecker

Application of Sandwich Compounds

- Fuel additive
 - o Anti-knock compounds that reduce engine knocking and increase a fuel's octane rating
- Half-sandwich compounds catalysts for transfer hydrogenation
 - Addition of hydrogen without using gaseous H₂
- Various catalysts for reactions
- Some bent metallocenes are precatalysts for polymerization of propylene



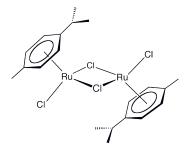
1, M = Zr, Hf



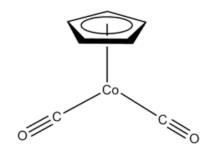
2, M = Zr, Hf



3, M = Zr, Hf



Derivatives of this sandwich used for transfer hydrogenation



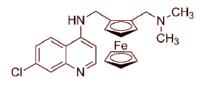
Catalyzes formation of pyridines from alkynes and nitriles

Compounds used for polymerization

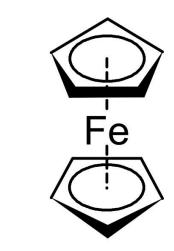
Application

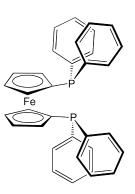
Ferrocene

- Catalyst for the production of carbon nanotubes
- Ligand scaffold
- Safer fuel additives
 - Replaced tetraethyllead as anti-knock agent
 - Provide rocket propellant with heat stability
 - Reduces smoke and sulfur trioxide generation in burning coal
- Pharmaceuticals
 - Ferroquine is an antimalarial in clinical trials
 - Anticancer properties



Ferroquine





1,1'-Bis(diphenylphosphino)ferrocene, used for palladium coupling reactions

Works Cited

"Ferrocene." 2021. In Wikipedia. https://en.wikipedia.org/w/index.php?title=Ferrocene&oldid=1015769972.

"Laureate - Ernst Otto Fischer." n.d. Lindau Nobel Mediatheque. Accessed May 4, 2021.

https://www.mediatheque.lindau-nobel.org/laureates/fischer.

"Pauson - 1973 - Recognition for Organometallic Chemistry.Pdf." n.d. Accessed May 4, 2021.

https://www.nature.com/articles/246003a0.pdf.

Pauson, P. L. 1973. "Recognition for Organometallic Chemistry." Nature 246 (5427): 3–3. https://doi.org/10.1038/246003a0.

Seyferth, Dietmar, and Alan Davison. 1973. "The 1973 Nobel Prize for Chemistry." Science 182 (4113): 699–701.

https://doi.org/10.1126/science.182.4113.699.

"Sir Geoffrey Wilkinson | British Chemist." n.d. Encyclopedia Britannica. Accessed May 4, 2021.

https://www.britannica.com/biography/Geoffrey-Wilkinson.

"The Nobel Prize in Chemistry 1973." n.d. NobelPrize.Org. Accessed May 4, 2021.

https://www.nobelprize.org/prizes/chemistry/1973/summary/.