

Class Assignment
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Questions (20 pts):

Q1. Determine the point groups for a-p

- a. **Ethane, $\text{CH}_3\text{-CH}_3$, (staggered conformation)**

D3d: It has a principal C_3 axis, a perpendicular C_2 axis, and multiple vertical mirror planes within the C_3 axis

- b. **Ethane (eclipsed conformation)**

D3h: It has a principal C_3 axis, a perpendicular C_2 axis, and a horizontal plane perpendicular to the C_3 axis

- c. **Chloroethane (staggered conformation)**

Cs: Only one mirror plane, no rotational axis

- d. **1,2-Dichloroethane (staggered *anti* conformation)**

The point group is C_{2h} . Has C_2 rotational axes passing through the mid point of the C-C bond and perpendicular to the Cl-C-C-Cl plane. It has one diamond h plane perpendicular to the C_2 rotational axis.

- e. **Ethylene**

Belongs to D_{2h} point group. Has 3 C_2 rotation axes and 3 σ planes of symmetry

- f. **Acetylene**

Acetylene is a linear molecule. It has C infinite rotational axis along the molecule axis. The principle axis is C infinite and it has an infinite number of C_2 rotational

axes perpendicular to the principal axis. It also has σ_h perpendicular to the principal axis. **The point group is $D_{\infty h}$.**

g. Naphthalene

Principal axis is C_2

2 perpendicular axes - D group

Horizontal mirror plane - D_{nh}

C_2 so $\rightarrow D_{2h}$

h. 1,8-dichloronaphthalene

Principal axis is C_2

No perpendicular axis - C group

No horizontal mirror plane, but has mirror plane that contains principal axis $\rightarrow C_{2v}$

i. 1,5-dichloronaphthalene

Principal axis is C_2

No perpendicular C_2 axes so C group

Has a horizontal mirror plane $\rightarrow C_{2h}$

j. 1,2-Dichloronaphthalene

No rotational axis, only one mirror plane of molecule $\rightarrow C_s$

k. Cyclohexane (in chair conformation)

C_6 is the principal axis with 3 C_2 axes perpendicular to the C_6 axis. Also has mirror plane containing principle axis $\rightarrow D_{3d}$

l. 1,1' – Dichloroferrocene

C_2 is the principal axis, and a horizontal plane is present that is perpendicular to the C_2 axis. Also, no C_2 axis perpendicular to the principal axis. Thus, its point group $\rightarrow C_{2h}$.

m. Cis-[PtCl₂(NH₃)₂]

C_2 rotation axis and two vertical planes

C_{2v} molecule

n. Trans-[PtCl₂(NH₃)₂]

C_2 rotation axis, three vertical planes, and inversion center

D_{2h} molecule

o. Diborane

Three perpendicular C_2 axes and three perpendicular mirror planes

D_{2h} molecule

p. Dibenzenechromium (eclipsed conformation)

C_6 , C_3 , C_2 perpendicular axes to the rings, six perpendicular C_2 axes, horizontal plane

D_{6h} molecule