

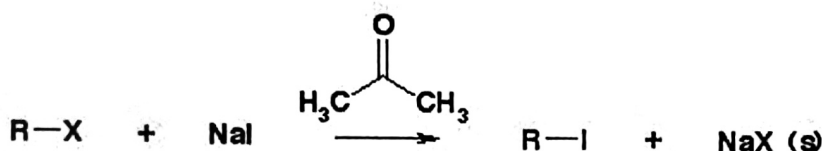
Name: Max Shir

Date: 8/25/19

Exp 7-  $\text{S}_\text{N}^2$  reaction  
Synthesis of alkyl iodide from alkyl halide.

Halides can react with sodium iodide in acetone and undergo  $\text{S}_\text{N}^2$  reaction which is evident by formation of precipitate of  $\text{NaX}$  (see below).

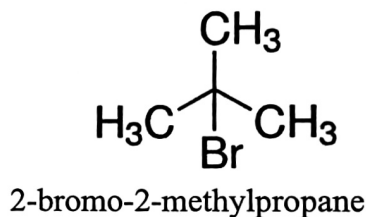
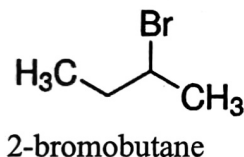
If the reaction proceed then you will see a precipitate. If the reaction doesn't go then you will not see a precipitate.



Answer questions 1-6:

(1) Which alkyl bromide will react fastest with sodium iodide in acetone:

1-bromobutane, 2-bromobutane or 2-bromo-2-methylpropane?



(2) Which alkyl bromide will react slowest?

2-bromo-2-methylpropane

(3) Explain how the structure of the alkyl halide affects the rate of an  $\text{S}_\text{N}^2$  reaction

The more substituted the alpha carbon is, the more steric hindrance there is for the backside attack, which makes  $\text{S}_\text{N}^2$  slower.

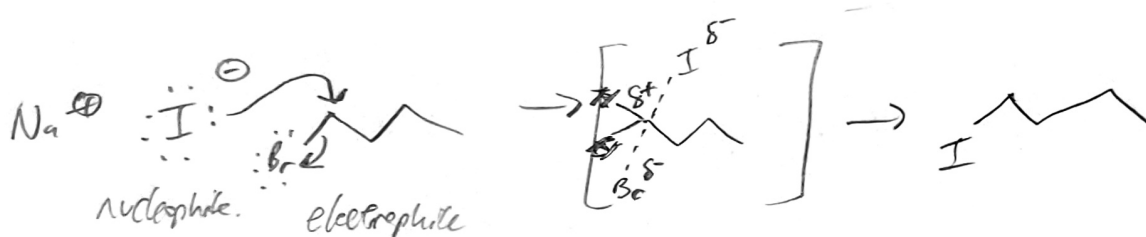
- (4) Which halide will faster with sodium iodide in acetone:  
1-bromobutane or 1-chlorobutane?

1-bromobutane will react faster

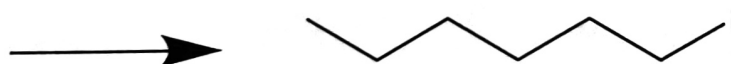
- (5) Explain how the nature of the leaving group affects the rate of an  $S_N2$  reaction. Would 1-iodobutane react faster or slower than the other halides?

The leaving group better leaves group will have a faster reaction rate in  $S_N2$ . Br is less electronegative and larger than Cl, making it a better leaving group. Thus, 1-iodobutane would be faster.

- (6) (a) Write balanced equations for substitution reaction and Propose transition states for the reaction of sodium iodide with 1-bromobutane  
(b) Label electrophile and nucleophile.



- (7) Perform a Scifinder search to find out number reactions reported for the the preparation of 1-iodohexane: Paste the screen shot of the results.



1-iodohexane

~~197925~~ reactions.  
~~93~~ 93

*[Handwritten signature]*

(Signed by TA or Instructor):

Reaction Structure substructure > reactions (93)

**REACTIONS** **Get References** **Tools** **Send to SciPlanner**

**Analyze** **Refine** Group by: No Grouping Sort by: Relevance **0 of 93** Reactions Selected Page: 1 of 7 [Display Options](#)

Analyze by:   
 Reagent

I <sub>2</sub>	31
KI	17
NaI	7
NaOH	5
O <sub>2</sub>	5
H <sub>3</sub> PO <sub>4</sub>	4
HCl	4
PPh <sub>3</sub>	4
LiAlH <sub>4</sub>	3
P <sub>2</sub> O <sub>5</sub>	3

[Show More](#)

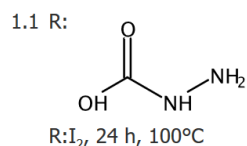
- ☐ 1. [View Reaction Detail](#) [Link](#) [Similar Reactions](#)

**Single Step** *Hover over any structure for more options.*



**Overview**

**Steps/Stages**



**Notes**

no solvent, optimization study, optimized on stoichiometry of reagents and time, pressure reactor used, Reactants: 1, Reagents: 2, Steps: 1, Stages: 1, Most stages in any one step: 1

**References**

[Preparation of alkyl halide](#)

[Quick View](#)

By Hur, Nam Hwi et al

From PCT Int. Appl., 2017155356, 14 Sep 2017

- ☐ 2. [View Reaction Detail](#) [Link](#) [Similar Reactions](#)

**Single Step** *Hover over any structure for more options.*



**Overview**

**Steps/Stages**

**Notes**