

## Supporting Information

# Solvent triggered shape morphism of 4D printed hydrogels

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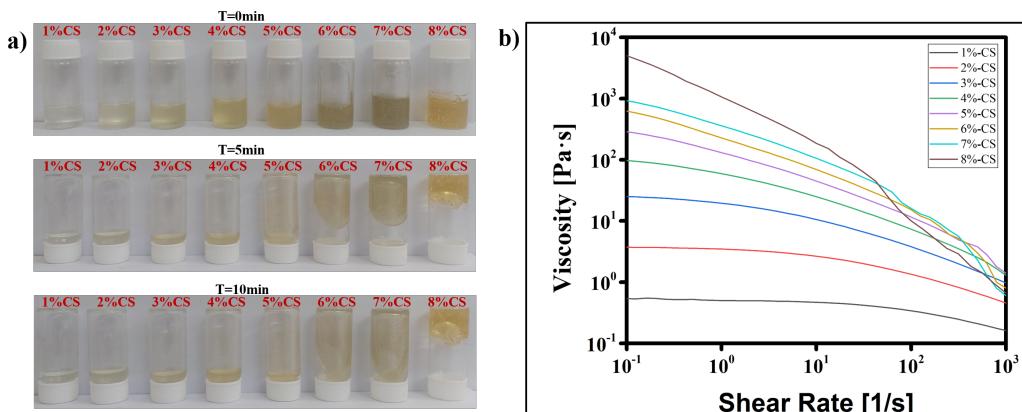
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<sup>1</sup> **1. Rheological characterization**

<sup>2</sup> In rheological testing, flow behaviour, shear thinning behaviour and an-  
<sup>3</sup> gular frequency sweep of the hydrogels were investigated by varying the per-  
<sup>4</sup> centage of chitosan from 1-8%. Fig. S1a) revealed the flow behaviours of  
<sup>5</sup> different hydrogels, in which sol-like 1-5% chitosan hydrogels exhibited a  
<sup>6</sup> good gravity-induced flow. The 6% and 7% hydrogels remained stable after  
<sup>7</sup> inverting the hydrogel-containing tubes for 5 min after which it flowed due  
<sup>8</sup> to gravity. The 8% chitosan hydrogel exhibited a good gel-like behaviour and  
<sup>9</sup> didn't flow even at 10 min, indicating that it can form stable structures post-  
<sup>10</sup> printing. In the shear thinning test, as shown in fig. S1b), it was found that  
<sup>11</sup> the viscosity of all the hydrogel samples was decreased with the increased  
<sup>12</sup> shear rate, demonstrating a shear thinning behaviour. Besides, 1-5% exhib-  
<sup>13</sup> ited the lowest viscosity among the entire shear rate, suggesting that they  
<sup>14</sup> had a poor printability. Interestingly, the viscosity of 6-8% were higher,  
<sup>15</sup> showing that chitosan acts a viscosity modifier to enhance printability. The  
<sup>16</sup> viscosity of 8% was highest at low shear rate among all and did not flow  
<sup>17</sup> after 10 min due to gravity, therefore the ink with 8% chitosan was the most  
<sup>18</sup> suitable composition to form stable 3D structures.

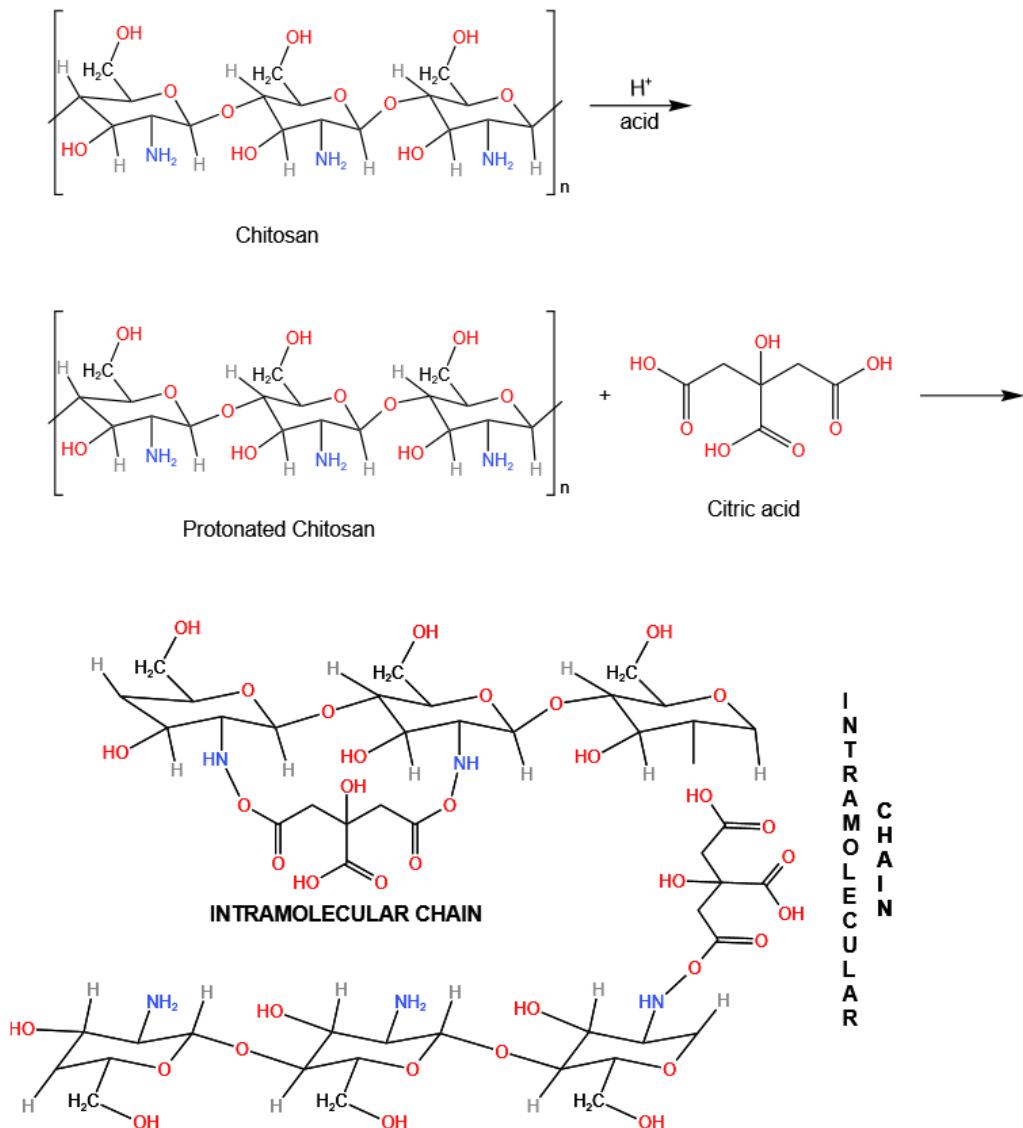


**Figure S1:** Printability of 1-8% chitosan hydrogel: a) flow behaviour of each hydrogel, b) viscosity as a function of shear rate at room temperature

<sup>19</sup> **2. Crosslinking procedure and mechanism**

<sup>20</sup> The samples were 3D printed using chitosan ink (the ink contains citric  
<sup>21</sup> acid, but it's not yet cross-linked). All the samples were treated at 50°C in

22 a vacuum oven for 24 hours and the samples obtained were termed an un-  
 23 crosslinked printed samples. To obtain cross-linked samples, the dried uncross  
 24 linked samples containing citric acid were treated at 150°C for 10 min in an  
 25 oven for the crosslinking reaction to take place as shown in fig. S2 [1].



**Figure S2:** Crosslinking mechanism of chitosan with citric acid

26 To study the effect of crosslinking and mechanism of crosslinking, swelling  
27 measurements and FTIR analysis were conducted.

28 • **Swelling measurements**

29 Samples of diameter 20 mm were 3D printed for the swelling mea-  
30 surements. The swelling degree of the uncrosslinked, and cross-linked  
31 printed samples were measured by gravimetric method. Samples of each  
32 system were immersed into 100ml of distilled water. Weight of each sam-  
33 ple was measured before ( $W_1$ ) and after immersion ( $W_2$ ) for 24 hours  
34 at room temperature. Swelling degree was calculated by the following  
35 equation:

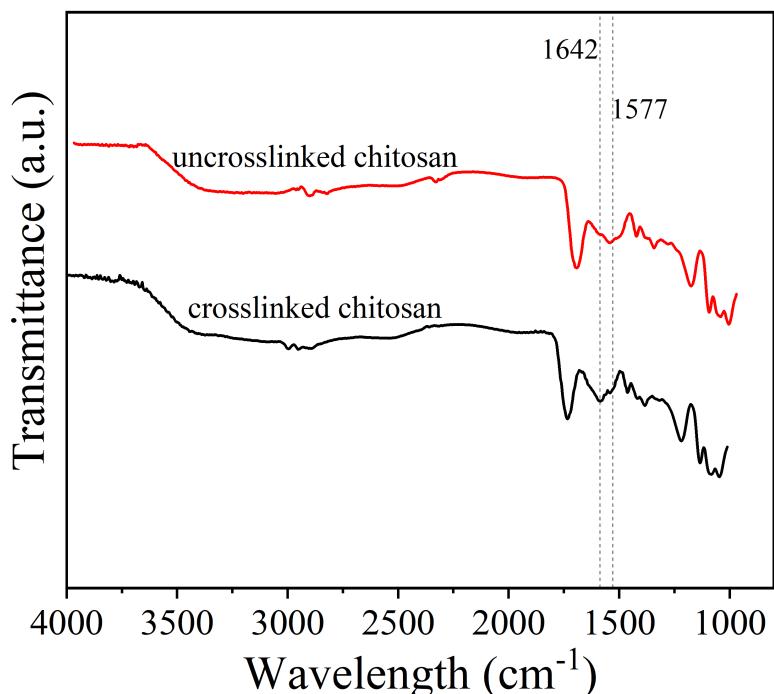
$$S\% = (W_2 - W_1)/W_1 * 100 \quad (1)$$

36 Measurements were taken in triplicate for each system. The results  
37 showed that the uncrosslinked samples and cross-linked samples had  
38 swelling degree of 1550% and 250% respectively. The decreasing in the  
39 swelling values in the citric acid cross-linked structures could be ex-  
40 plained by a higher crosslinking degree, which reduced the free volume  
41 and chain mobility, thus, hindering the water entry in the networks[2,  
42 3].

43 • **Fourier Transform Infrared Spectrophotometer**

44 The FTIR spectra of chitosan samples were obtained by using Bruker's  
45 fourier transform infrared spectrophotometer. 64 scans were collected in  
46 the range 800-4000 cm<sup>-1</sup> with a resolution of 8 cm<sup>-1</sup>. The IR spectra  
47 were analysed using lab solutions series software.

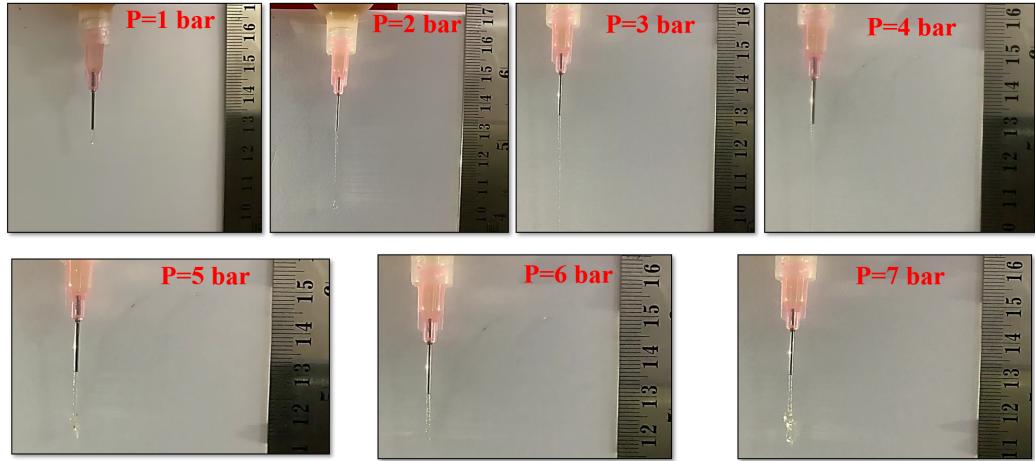
48 Fig.S3 shows the ATR-FTIR spectrum of the chitosan(CS) and chitosan-  
49 citric acid(CS-CA) cross-linked films. The main characteristic bands as-  
50 sociated with chitosan molecular were detected from the spectra. The  
51 amide I band is associated with stretching vibrations of C-O and ap-  
52 pears at 1642 cm<sup>-1</sup>. While, the amide II associated with C-N stretching  
53 and N-H vibration, has a band at 1577 cm<sup>-1</sup>. The ATR-FTIR result im-  
54 plies that citric acid has been cross-linked with the chitosan molecules  
55 at 150°C for 10 min successfully.



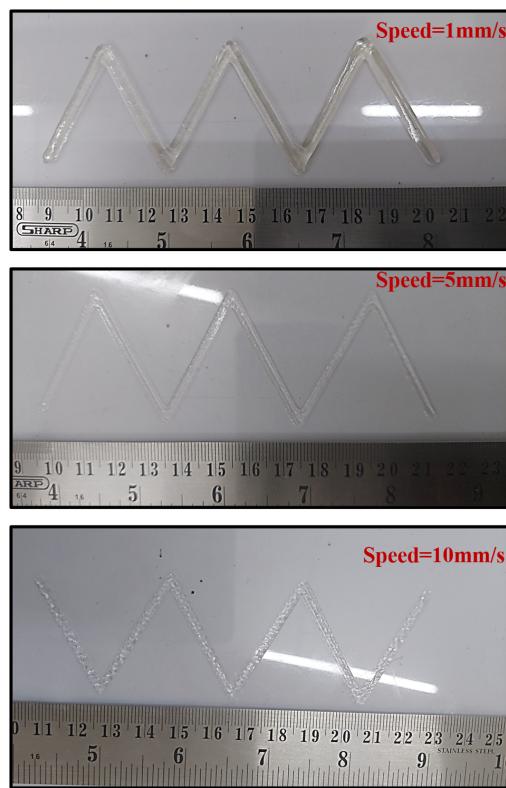
**Figure S3:** FTIR spectra of 3D printed chitosan films before and after the crosslinking with citric acid

<sup>57</sup> **3. Printability and shape fidelity**

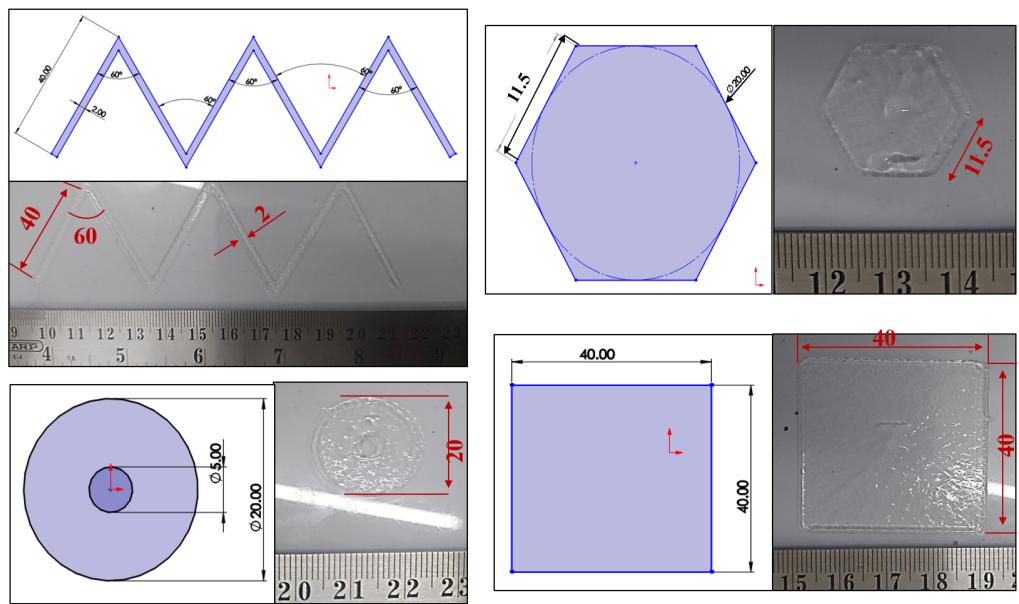
- <sup>58</sup> • Study of extrusion pressure (see figure S4)
- <sup>59</sup> • Study of printing speed (see figure S5)
- <sup>60</sup> • Analysis of shape fidelity (see figure S6)



**Figure S4:** Study of extrusion pressure at different pressure(1-7 bar)



**Figure S5:** Study of printability at different printing speed(1-10 mm/s)



**All dimensions are in mm**

**Figure S6:** Comparison of designed part with printed structures using optimal printing parameters

61 **References**

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