

**Prob-1.** In the 'Open' form, the protein domains 'A' and 'B' can independently adopt 27 and 81 microscopic states, respectively, The ligand 'L' can always adopt 3 states.

When shifting from the 'Open' to the 'Bound' state, the interaction energies change as:

$$\Delta E_{A-B}: -35.0 \text{ kcal/mol}$$

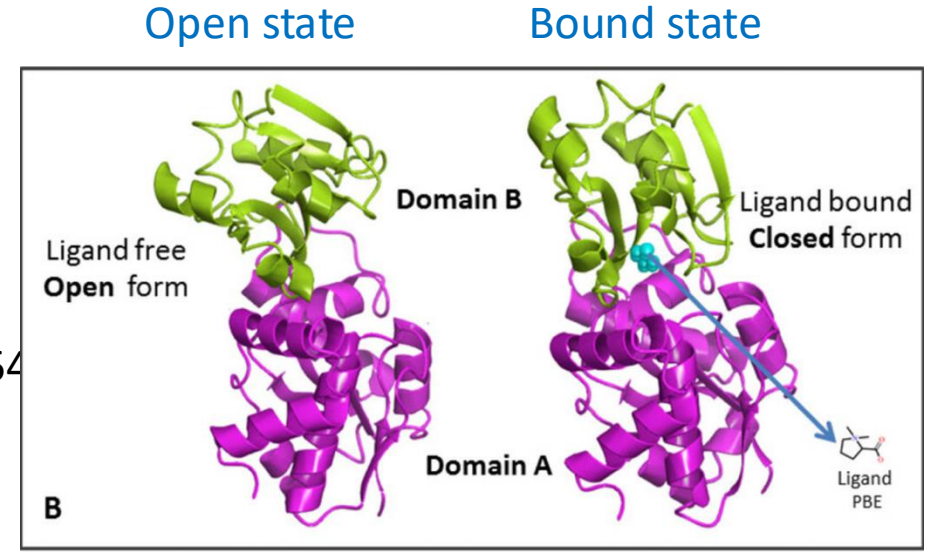
$$\Delta E_{A-L}: -7.0 \text{ kcal/mol}$$

$$\Delta E_{B-L}: -4.0 \text{ kcal/mol}$$

$$\Delta E_{A-B} = E_{(A+B)} - (E_A) - (E_B)$$

Also, the number of states in 'A' and 'B' in the closed form reduce to 18 and 54 respectively.

Determine if the 'Open' to 'Bound' transition is spontaneous.



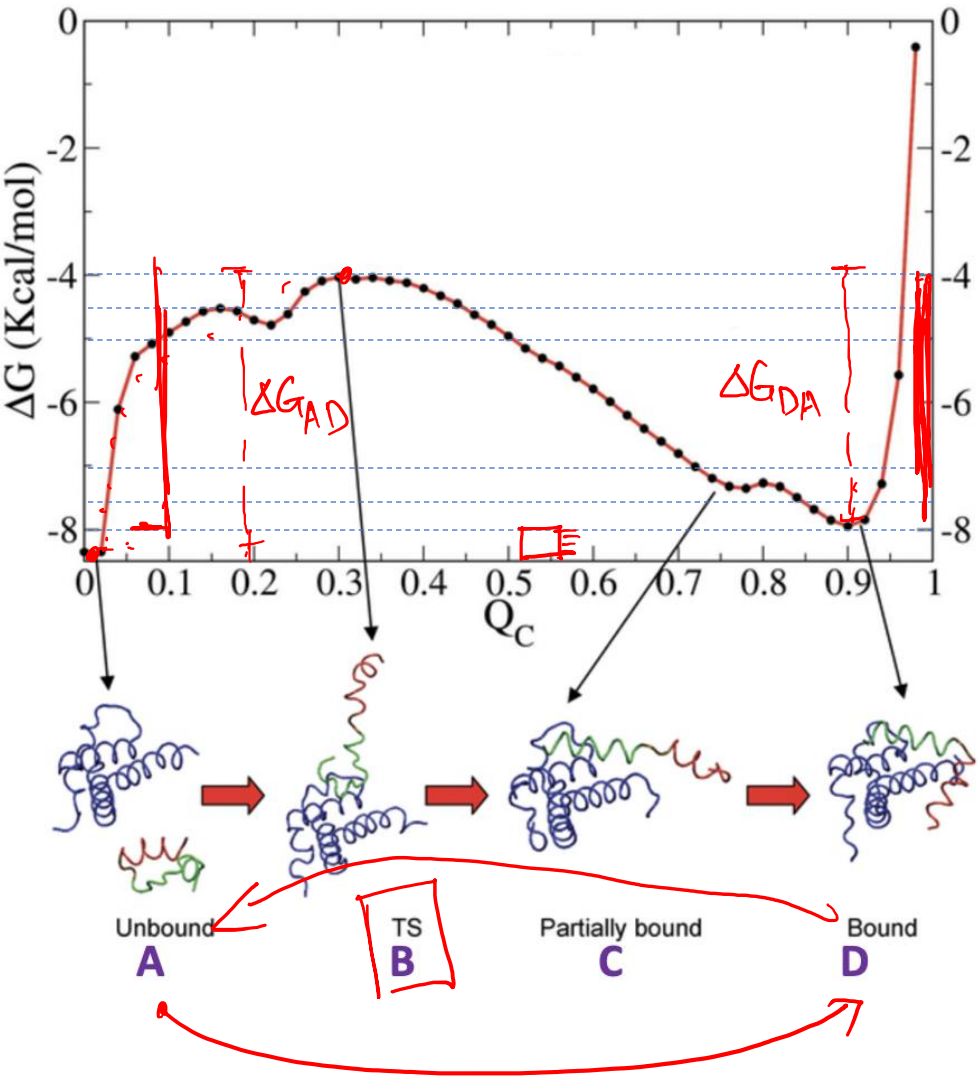
$$\Delta G = \Delta E - T\Delta S$$

$$= (\Delta E_{A-B} + \Delta E_{A-L} + \Delta E_{B-L}) - T(\Delta S_A + \Delta S_B + \Delta S_L)$$

$$= -46.0 - 300 R [\ln \Omega_{\text{final}} - \ln \Omega_{\text{initial}}]$$

IS THE PROCESS ENERGETICALLY OR ENTROPICALLY DRIVEN?

- Prob-2.** a) Find the ratio of the probabilities of finding states A, B, C and D at room temperature.  
b) Find the ratio of the transition rate from (D → A) and from (A → D), ignoring any intermediary metastable state.



a) 
$$\frac{P_A}{P_B} = \frac{e^{-\beta G_A}}{e^{-\beta G_B}} = e^{-\beta(G_A - G_B)} = e^{-\frac{1}{RT}(-8.5 - (-4.0))} \approx 2300 \text{ (check)}$$

Similarly, 
$$\frac{P_A}{P_C}, \frac{P_A}{P_D} \left\{ \begin{array}{l} P_A = P_B \\ P_C = P_D \end{array} \right. = e^{-\beta(G_A - G_C)} = e^{-\frac{1}{RT}(-8.5 - (-8.0))} \approx 2300 \text{ (check)}$$

b) 
$$\left. \begin{array}{l} k_{A \rightarrow D} \propto e^{-\beta \Delta G_{AD}} \\ k_{D \rightarrow A} \propto e^{-\beta \Delta G_{DA}} \end{array} \right\} \frac{k_{A \rightarrow D}}{k_{D \rightarrow A}} = e^{-\beta(\Delta G_{AD} - \Delta G_{DA})} = e^{-\beta(-8.5 - (-8.0))} = e^{-\beta(-0.5)} = e^{\beta(0.5)} \approx 2300 \text{ (check)}$$

$$\frac{k_{A \rightarrow D}}{k_{D \rightarrow A}} = e^{-\frac{1}{RT} \left[ \underbrace{0.25} \right]} \approx e^{-\frac{0.25}{0.58}} \approx 0.6$$

$$\Delta G_{AD} - \Delta G_{DA}$$

$$= 0.25$$

check

**Prob 3.** In an experiment (E1), the diffusion coefficient of human Hemoglobin was measured as  $8.0 \times 10^{-7} \text{ cm}^2/\text{s}$  at 35 deg C.

Another experiment (E2) under a different condition measured the value as  $6.8 \times 10^{-7} \text{ cm}^2/\text{s}$ .

If E2 reported a 10% decrease in the average radius of Hemoglobin, what is the closest approximation to the temperature corresponding to E2?

(Assume Hemoglobin's movement was purely diffusive)

Assumptions:  $6\pi\eta aD = k_B T$

- Spherical assumption
- $\eta$  remains constant

$$\frac{a_1 D_1}{a_2 D_2} = \frac{T_1}{T_2}$$

**Prob 4.** Find the closest approximation to 1 unit of room temperature thermal energy ( $k_B T_r$ ) in electron Volts (eV).  
[1 eV is the work done in moving a charge of 1 electron across a potential difference of 1 volt, in SI units]

Ans: Thermal energy is approximately 0.025 eV at 300K