

FEDERAL PUBLIC SERVICE COMMISSION COMPETITIVE EXAMINATION-2020 FOR RECRUITMENT TO POSTS IN BS-17 UNDER THE FEDERAL GOVERNMENT

Roll Number

PHYSICS, PAPER-I

TIME ALI PART-I(M	LOWED: THREE HOURS ICQS): MAXIMUM 30 MINUTES	PART-I (MCQS) PART-II	MAXIMUM M MAXIMUM M	
NOTE: (i) (ii) (iii) (iv) (v) (vi) (vii)	Write Q. No. in the Answer Book in acco No Page/Space be left blank between the crossed. Extra attempt of any question or any part	PART-II. ALL questions ust be attempted at one plordance with Q. No. in the ne answers. All the blank	lace instead of at diff e Q.Paper. k pages of Answer I	erent places.
	PAI	RT – II		
Q. 2. (a) (b)	What is the curl of a vector field? Explain What is vector triple product? Show that $ABCACBABC\Box$	at	nce.	(10) (6)
	(4) (20)			
(c)	If $\phi \Box 2x \ y \ z^{3 \ 2 \ 4}$ then find the div grad \emptyset .			
Q. 3. (a)	State and explain Kepler's law of areas			(8)
(b)	A spaceship of mass $m = 4.50 \times 103$ kg $r \square 8.00 \ 10^{\square}$ 6 m and period $T_0 = 118.6$			(6) in
	the forward direction to decrease the sp the period T of the resulting elliptical o	eed to 96.0% of the orig	ginal speed. What is	(6) (20)
(c)	Which has greater magnitude, the anguassociated with its rotation on its axis of the center of its orbit) associated with its	or the angular momentu	m of the Earth (rela	
Q. 4. (a)	Explain the equivalence of mass and er	nergy.		(6)
(b)	Explain two tests of time dilation i.e m	-	-	
(c)	The mean lifetime of stationary muons lifetime of high-speed muons in a burst measured to be 16.000 µs. To five sign these cosmic-rays' muons relative to E	of cosmic rays observe	d from Earth is	(6) (20)
Q. 5. (a)	What is viscosity? Explain in detail. W	hat is the effect of tempe	erature on viscosity	? (8)
(b)	Castor oil, which has a density of 0.96 a pipe of circular cross section by a pur	$\times 10^3$ kg/m ³ at room ten	nperature, is forced	(5) through

has a diameter of 2.6 cm and a length of 65 cm. The castor oil emerging from the free end of

- the pipe at atmospheric pressure is collected. After 90 s, a total of 1.23 kg has been collected. What is the coefficient of viscosity of the castor oil at this temperature?
- (c) A liquid flow through a horizontal pipe whose inner radius is 2.52 cm. The pipe (7) (20) bends upward through a height of 11.5 m where it widens and joins another horizontal pipe of inner radius 6.14 cm. What must the volume flux be if the pressure in the two horizontal pipes is the same?
- **Q. 6. (a)** What is damped harmonic oscillator? Write its equation of motion and find its solution. (10)
 - (b) The amplitude of a lightly damped oscillator decreases by 3.0% during each cycle. (4) What percentage of the mechanical energy of the oscillator is lost in each cycle?
 - (c) An insulating vessel containing 1.8 kg of water is placed on a hot plate, both the (6) (20) water and hot plate being initially at 20°C. The temperature of the hot plate is raised very slowly to 100°C, at which point the water begins to boil. What is the entropy change of the water during this process?

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- Q. 7. (a) What are travelling waves? Find the rate at which energy is transported by a wave (5) travelling along a string.
 - (b) A string has linear density $\mu = 525$ g/m and is under tension T = 45 N. We send a (5) sinusoidal wave with frequency f = 120 Hz and amplitude $y_m = 8.5$ mm along the string. At what average rate does the wave transport energy?
 - (c) Two sinusoidal waves with the identical wavelengths and amplitudes travel in (10) (20) opposite directions along a string with a speed of 10 cm/s. If the time interval between instants when the string is flat is 0.50 s, what is the wavelength of the waves?
- Q. 8. (a) Explain the volume and pressure corrections in ideal gas law as suggested by van (10) der Waals.
 - For oxygen the van der Waals coefficients have been measured to be (5) $a = 0.138 \text{ J} \cdot \text{m}^3/\text{mol}^2$ and $b = 3.18 \times 10^{-5} \, \text{m}^3/\text{mol}$. Assume that 1.00 mol of oxygen at $T = 50 \, \text{K}$ is confined to a box of volume 0.0224 m³. What pressure does the gas exert according to (a) the ideal gas law and (b) the van der Waals equation?
 - (c) State and explain the zeroth law of thermodynamics. (5) (20)
