

# Stochastic Simulation of Supply Chains

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# 1 Introduction to Supply Chains

A *supply chain* [2] is a network of all the parties involved in production, transportation, storage, distribution and delivery of products in order to fulfil a customer's request. Other than manufacturers and suppliers, a supply chain also includes warehouses, transporters, distributors, retailers and the customers themselves. A supply chain also includes the activities and functions that are aimed towards fulfilling a customer's demands. These functions include distribution, finance, quality assurance, information exchange and transportation.

We present a simple example of a supply chain that describes the functions of various stages when a customer buys a smartphone from a Walmart store.

Let us consider a customer purchasing a smartphone from Walmart. The supply chain that kicks-in to fulfil the customer's request includes - the customer, Walmart's store staff, Walmart's distribution centers, smartphone manufacturer, their suppliers, transportation partners and third-party organisations. The supply chain starts with the customer visiting Walmart to buy a smartphone. Walmart's store staff stocks its shelves from the store's inventory, that in turn is supplied from Walmart's distribution centers. Supplying goods requires involvement of transportation vehicles that might have been supplied from a third party. The distribution centers are stocked by supplies from smartphone manufacturers, that themselves use supplies from lower-tier suppliers. For example, the manufacturer might depend on a supplier for its semiconductor demands and another supplier for its packaging needs. This network of stages involves information flow amongst parties such as - i) point-of-sales information about the purchase that is registered at Walmart store, ii) data on available stocks in the store's inventory that is communicated with Walmart's distribution centers, iii) sales information and additional orders of smartphones placed by Walmart to smartphone manufacturers, just to name a few.

Several stages that are involved in a typical supply chain, as described in [2], include the customers, retailers, distributors, manufacturers and raw material suppliers.

A manufacturer usually depends on more than one suppliers, and similarly other stages of a supply chain depend on more than one parties for proper functioning of the supply chain. Figure 1 depicts a supply chain with eight processes, where process 3 depends on supplies from supply 1 and process 1. Similarly, process 5 depends on output materials from processes 3 and 4.

## 1.1 Risk and disruption in supply chains

Supply chains across the globe faced one of the biggest disruptions in the form of COVID-19 pandemic in 2020-2021. Supply chains at local and global levels were effected drastically due to the near-total reduction in public mobility and ocean freight. Reduction in workforce at manufacturing plants, ports and road transportation posed challenges to supply chains at almost all levels. Many

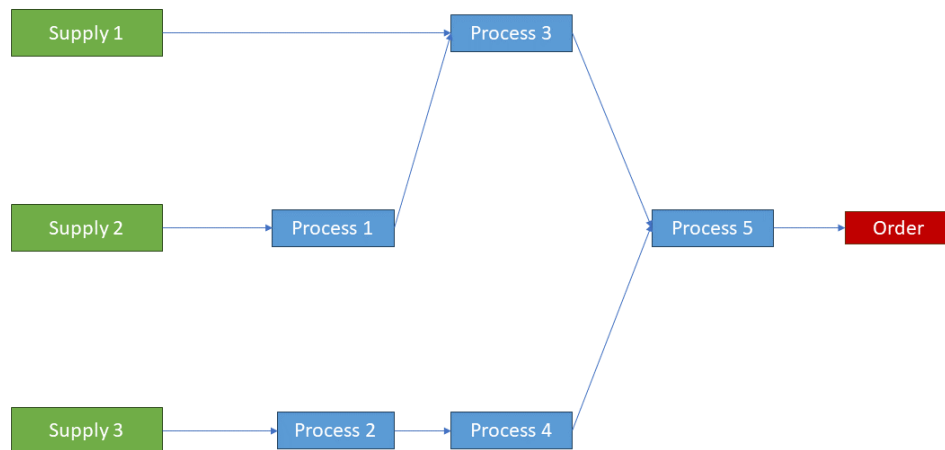


Figure 1: A supply chain with 8 processes.

countries still face economic challenges as an aftermath of disrupted and restructured supply chains. Given the far-reaching impact of the COVID-19 pandemic, which led to substantial reductions in public mobility and disrupted various aspects of supply chains, it becomes crucial to discuss the risks and underlying causes of such disruptions during the design and implementation of supply chains.

With rising global temperatures, extreme weather events are posing higher and higher risks to the smooth functioning of global supply chains. High level of inter-dependence between countries for raw materials and manufactured products implies that even a single disruption caused by extreme weather can hamper global supply chains, and hence lead to serious economic consequences.

As global supply chains become increasingly complex, they are susceptible to multiple points of failure, rendering them more vulnerable and less resilient. Some of the risk factors, as discussed in [2], are categorised in table 1:

## 1.2 Resilience of supply chains

With good network designs, supply chains can mitigate many risk factors mentioned in Table 1. According to D. Mavatoor in [3], “The concept of resilience shouldn’t assume that you won’t fail, but rather that you should be able to get back up fast.” Flexibility of networks, amongst others, plays a crucial role in ensuring resilience of supply chains. An example of the effectiveness of flexible networks is the levels of impact a supplier disruption had on two companies, Nokia and Ericsson

Category	Risk Drivers
Delays	High capacity utilization at supply source, inflexibility of supply source, poor quality or yield at supply source
Disruptions	Pandemics, war, natural disaster, supplier bankruptcy, labor disputes
Systems Risk	Information infrastructure breakdown, system integration or extent of systems being networked
Forecast Risk	Inaccurate forecasts due to small customer base, seasonality, product variety, information distortion
Procurement Risk	Exchange-rate risk, price of inputs, fraction purchased from a single source, industry-wide capacity utilization
Receivables Risk	Number of customers, financial strength of customers
Inventory Risk	Rate of product obsolescence, inventory holding cost, product value, demand and supply uncertainty
Capacity Risk	Cost of capacity, capacity flexibility

Table 1: Categorization of risk factors in supply chains

in 2000. Supply chains of Nokia and Ericsson were disrupted when a plant owned by Royal Philips Electronics caught fire in Albuquerque, New Mexico in 2000. Nokia bounced back quickly because of presence of multiple backups in their network of suppliers, and faced minimal losses. On the other hand, Ericsson lost \$400 million in revenues as it did not have a backup in their network, and hence could not bounce back fast.

Mitigation strategies come at a price and their own set of risks. Having multiple suppliers in the network increases cost of the supply chain due to increase in transportation, coordination, management and used resources. Having higher inventory incurs cost on storage, maintenance and raises risk of obsolescence. As discussed in [11], the trade-off in inventory control is between *producing*, which leads to increased inventory costs and management costs, and *idling*, which can lead to stock-outs and unsatisfied demands. Hence, customised mitigation strategies should be incorporated in designs of supply chains that achieve a trade-off between the risk mitigated and the incurred cost. Some of the tailoring strategies, as discussed in [2] are mentioned in table 2.

### 1.3 Types of processes in supply chains

The processes in supply chains can be categorised based on whether they operate in response to a customer's order, or in anticipation of customer orders [2]. *Push processes* are executed in anticipation of customer orders, whereas *pull processes* are triggered following a customer's order.

#### 1.3.1 Push Processes

Push processes are executed in anticipation of customer orders. They are based on forecasts of product demands in the future, and hence, are also referred to as *speculative processes* because they

Risk Mitigation Strategy	Tailored Strategies
Increase capacity	Focus on low-cost, decentralized capacity for predictable demand. Build centralized capacity for unpredictable demand. Increase decentralization as the cost of capacity drops.
Get redundant suppliers	More redundant supply for high-volume products, less redundancy for low-volume products. Centralize redundancy for low-volume products in a few flexible suppliers.
Increase responsiveness	Favor cost over responsiveness for commodity products. Favor responsiveness over cost for short-life cycle products.
Increase inventory	Decentralize inventory of predictable, lower value products. Centralize inventory of less predictable, higher value products.
Increase flexibility	Favor cost over flexibility for predictable, high-volume products. Favor flexibility for unpredictable, low-volume products. Centralize flexibility in a few locations if it is expensive.
Pool or aggregate demand	Increase aggregation as unpredictability grows.
Increase source capability	Prefer capability over cost for high-value, high-risk products. Favor cost over capability for low-value commodity products. Centralize high capability in a flexible source if possible.

Table 2: Tailored strategies for risk mitigation

operate in response to speculated demand. An example of a push process is a bakery producing a variety of bread and pastries early in the morning and pushing them onto store shelves for customers to purchase. Upstream processes, such as production of raw materials typically use push models based on forecasts of company demands.

### 1.3.2 Pull Processes

Pull processes are initiated in response to a customer order. These are also known as *reactive processes* as they react to an actual demand of a product. An example of a pull process is a company manufacturing products only when a customer order is received, ensuring that production is triggered by actual demand. Processes closer to end-customers, such as retailers and e-commercial websites use pull model as they operate only when a customer order is placed.

### 1.3.3 Hybrid Systems

In a hybrid system, some processes operate on a push model, whereas other processes operate on a pull model. Using hybrid systems enable companies to tailor their strategies to meet their limitations and maximise customer satisfaction at the same time. Hybrid systems offer the potential to optimise inventory levels, enhance customer satisfaction, reduce lead times, and improve overall supply chain performance.

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## 2 Chemical Reaction Modelling

*Continuous-time Markov chains* (CTMCs) are stochastic processes that follow the Markov property in which the system makes transitions in continuous time. Unlike discrete-time Markov chains where transitions can only take place at discrete time steps, CTMCs can transition through states at any point in time.

*Stochastic Reaction Networks* (SRNs) are a class of CTMCs that are used to model evolution of chemical systems where molecules of different chemical species can undergo a finite set of reactions [10]. SRNs are particularly useful when dealing with systems with small number of molecules as it helps in incorporating the inherent randomness of phase space of molecules in the evolution of the system.

In this chapter, we define CTMCs, SRNs, and discuss algorithms such as *Stochastic Simulation Algorithm* (SSA), *Next Reaction Method* (NRM), *Modified Next Reaction Method* (MNRM) and *tau-leap method*, that can be used to simulate SRN trajectories. SSA, NRM and MNRM are considered to be exact as the paths simulated by these methods follow correct statistical distributions. Tau-leap methods are used to simulate SRN paths using discrete time steps, in order to address some of the computational and performance related shortcomings of exact methods.

### 2.1 Continuous Time Markov Chains

Let  $\mathbf{X} = \{X(t), t \in [0, \infty)\}$  be a continuous-time stochastic process with a countable state space  $\mathbf{S}$ .  $\mathbf{X}$  is called a continuous-time Markov chain [12] if for any set of time indices  $0 \leq t_1 < t_2 < \dots < t_{s-1} < t_s < t$  and corresponding states  $c_1, c_2, \dots, c_{s-1}, c_s, c_t \in \mathbf{S}$  such that  $P(X(t_1) = c_1, X(t_2) = c_2, \dots, X(t_s) = c_s) > 0$ , the following is true:

$$P(X(t) = c_t | X(t_1) = c_1, \dots, X(t_s) = c_s) = P(X(t) = c_t | X(t_s) = c_s). \quad (1)$$

The property (1) is also called the Markov Property. We see from (1) that the probability of  $X$  being in a state  $c_t$  at time  $t$  is only conditioned on the state  $X(t_s)$  at time index  $t_s$ , and does not depend on the states of the process in time indices  $t_1, t_2, \dots, t_{s-1}$ .

### 2.2 Stochastic Reaction Networks

An SRN  $\mathbf{X}$  is a type of CTMC that describes evolution of a homogeneous chemical mixture of molecules in continuous time.

Given an underlying probability space  $(\Omega, \mathcal{F}, P)$ ,  $X : [0, T] \times \Omega \rightarrow \mathbb{Z}_{\geq 0}^N$  describes the time-evolution of a system of molecules belonging to  $N$  different species  $(S_1, S_2, \dots, S_N)$  in continuous time  $t \in [0, T]$ .

Elements in  $X(t) = (x_1(t), x_1(t), \dots, x_N(t))$  indicate the abundances of the  $N$  species of molecules at time  $t$ . The molecules can undergo a finite set of reactions  $\mathbf{R} = (\mathcal{R}_1, \mathcal{R}_2, \dots, \mathcal{R}_M)$ , where a reaction  $\mathcal{R}_j \in \mathbf{R}$  consists of a pair  $(\nu_j, a_j)$ . Each element in  $\nu_j = (\nu_{j,1}, \nu_{j,2}, \dots, \nu_{j,N})$ ,  $1 \leq j \leq M$  corresponds to the change in the number of molecules of that particular species when the  $j^{th}$  reaction fires in the mixture of molecules. The vectors  $\nu_j$  is also known as *stoichiometric vectors*. The *propensity*  $a_j : \mathbb{Z}_{\geq 0}^N \rightarrow \mathbb{R}_{\geq 0}$  of a reaction  $\mathcal{R}_j$  depends on the state  $X(t)$  at time  $t$  and describes the reaction rate of  $\mathcal{R}_j$  as follows:

$$P(X(t + \Delta t) = x + \nu_j | X(t) = x) = a_j(x)\Delta t + o(\Delta t). \quad (2)$$

A larger value of  $a_j(X(t))$  at time  $t$  indicates faster rate of reaction  $\mathcal{R}_j$  at time  $t$ . Probability of firing reaction  $\mathcal{R}_j$  in interval  $(t, t + \Delta t]$  increases as  $a_j(X(t))$  increases.

The probability that no reaction fires in the time interval  $(t, t + \Delta t]$  in the SRN  $\mathbf{X}$  is given by:

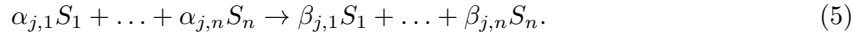
$$P(X(t + \Delta t) = x | X(t) = x) = 1 - \sum_{j=1}^M a_j(x)\Delta t + o(\Delta t). \quad (3)$$

Defining a new quantity  $a_0(x) := \sum_{j=1}^M a_j(x)$ , (3) can be rewritten as:

$$P(X(t + \Delta t) = x | X(t) = x) = 1 - a_0(x)\Delta t + o(\Delta t). \quad (4)$$

Given  $X(t) = x$  at time  $t$ , it can be derived from (4) that the time to fire next reaction follows an exponential distribution with parameter  $a_0(x)$ .

The stochastic mass-action kinetics principle can be represented by the diagram given below for some reaction  $\mathcal{R}_j \in \mathbf{R}$ :



As implied by (5), when a reaction  $\mathcal{R}_j$  is fired,  $\alpha_{j,1}, \dots, \alpha_{j,N}$  molecules of the corresponding species  $S_1, \dots, S_N$  are consumed and  $\beta_{j,1}, \dots, \beta_{j,N}$  molecules are produced. The stoichiometric vector  $\nu_j := (\beta_{j,1} - \alpha_{j,1}, \dots, \beta_{j,n} - \alpha_{j,n})$  can contain both positive and negative integers, and hence  $\nu_j \in \mathbb{Z}$ .

Given a positive reaction constant  $c_j$  and state  $X(t) = x = (x_1, \dots, x_N)$  at time  $t$ , the propensity function  $a_j(x)$  of reaction  $\mathcal{R}_j$  as derived from mass-action kinetics is given by:

$$a_j(x) = c_j \prod_{i=1}^N \frac{x_i!}{(x_i - \alpha_{j,i})!} \mathbf{1}_{\alpha_{j,i} \leq x_i} \quad (6)$$

where  $\alpha_{j,i}$  is the number of  $S_i$  molecules consumed when reaction  $R_j$  is fired and  $\mathbf{1}_{\alpha_{j,i} \leq x_i}$  is an

indicator function that can be described as below:

$$\mathbf{1}_{\mathbf{A}} = \begin{cases} 0, & \text{if } \mathbf{A} \text{ is false,} \\ 1, & \text{if } \mathbf{A} \text{ is true.} \end{cases}$$

In the remaining literature, (6) will be used to calculate propensities of reactions given the state  $X(t)$  at time  $t$ .

### Random Time Change Representation

Let  $x_0$  be the initial state of an SRN  $X(t), t \geq 0$  with reaction channels  $(\mathcal{R}_1, \mathcal{R}_2, \dots, \mathcal{R}_M)$ . The random time-change representation of  $\mathbf{X}$  as shown by T.G. Kurtz [9, 4, 8] is given as:

$$X(t) = x_0 + \sum_{j=1}^m \nu_j Y_j \left( \int_0^t a_j(X(s)) ds \right), \quad (7)$$

where  $Y_j : \mathbb{R}_{\geq 0} \times \Omega \rightarrow \mathbb{Z}_{\geq 0}$  are independent unit-rate Poisson processes. The representation (7) models the path of the process  $X(t), t \geq 0$  in continuous time as a combination of paths of  $M$  independent unit-rate Poisson processes. Each reaction is represented by a unit-rate Poisson process, and for  $1 \leq j \leq M$ , the value  $\int_0^t a_j(X(s)) ds$  is called the internal time of  $Y_j$ .

Later in the chapter, we will see how *Next Reaction Method* (NRM) and *Modified Next Reaction Method* (MNRM) make use of (7) to simulate exact paths of SRNs by sampling next-reaction times and next-reaction internal times respectively.

### Example - Gene transcription and translation

For illustration, we look at an example of an SRN that models gene transcription and translation. More details on this model can be found in [7]. The SRN, say  $X(t), t \geq 0$ , consists of three species  $(M, P, D)$  and five reactions as shown below:

- $\emptyset \rightarrow M$ , Transcription of a gene into mRNA.
- $M \rightarrow M + P$ , Translation of mRNA into proteins.
- $P + P \rightarrow D$ , Production of stable *Dimers* from proteins.
- $M \rightarrow \emptyset, P \rightarrow \emptyset$ , Degradation of mRNA and proteins respectively.



From the above reactions, we can construct  $\nu_j$  for each reaction as follows:

$$\nu_1 = \begin{bmatrix} 1 \\ 0 \\ 0 \end{bmatrix}, \quad \nu_2 = \begin{bmatrix} 0 \\ 1 \\ 0 \end{bmatrix}, \quad \nu_3 = \begin{bmatrix} 0 \\ -2 \\ 1 \end{bmatrix}, \quad \nu_4 = \begin{bmatrix} -1 \\ 0 \\ 0 \end{bmatrix}, \quad \nu_5 = \begin{bmatrix} 0 \\ -1 \\ 0 \end{bmatrix}.$$

The propensity functions of the five reactions as functions of the state  $X(t) = (m(t), p(t), d(t))^T$  at some time  $t$  are given by:

$$a_1(X(t)) = 25, \quad a_2(X(t)) = 10^3 m(t), \quad a_3(X(t)) = 10^{-3} p(t)(p(t) - 1), \\ a_4(X(t)) = 0.1 m(t) \quad \text{and} \quad a_5(X(t)) = p(t).$$

If the initial state  $X(0) = x_0$ , the process can be modeled with Kurtz' random time change representation (7) as:

$$X(t) = x_0 + Y_1 \left( \int_0^t 25 ds \right) \begin{bmatrix} 1 \\ 0 \\ 0 \end{bmatrix} + Y_2 \left( \int_0^t 10^3 m(s) ds \right) \begin{bmatrix} 0 \\ 1 \\ 0 \end{bmatrix} \\ + Y_3 \left( \int_0^t 10^{-3} p(s)(p(s) - 1) ds \right) \begin{bmatrix} 0 \\ -2 \\ 1 \end{bmatrix} + Y_4 \left( \int_0^t 0.1 m(s) ds \right) \begin{bmatrix} -1 \\ 0 \\ 0 \end{bmatrix} \\ + Y_5 \left( \int_0^t p(s) ds \right) \begin{bmatrix} 0 \\ -1 \\ 0 \end{bmatrix}$$

Where  $Y_j$ s are unit-rate Poisson processes with internal times given as  $\int_0^t a_j(X(s)) ds$ .

### 2.3 Simulation of SRNs

Analytical and numerical solutions to differential equations involving SRNs are usually infeasible. Some challenges met while finding analytical solutions for an SRN, let's say  $\mathbf{X}$ , include: i) non-linearity of propensity functions  $a_j$ s w.r.t. states of  $\mathbf{X}$ , ii) large number of state space of  $X(t), t \geq 0$ , to name a few. For this reason, simulating paths of SRNs using Monte Carlo method can help us circumvent these challenges and find approximate values of several quantities of interest including but not limited to time taken by process  $X(t), t \geq 0$  to reach a specified subset of its state space and expected value of an observable  $g$  at some time  $T$ , i.e.  $E[g(X(T))]$ .

**Exact Algorithms**

Exact algorithms are methods that can be used to simulate paths of an SRN  $X(t), t \geq 0$  that obey the probabilities described by (2) and (3). These paths have correct statistical distributions and hence can be used to estimate values such as expected value of an observable  $g$  at time  $T$ , i.e.  $E[g(X(t))]$ . We will discuss three exact methods: 1. *Stochastic Simulation Algorithm* (SSA), 2. *Next Reaction Method* (NRM) and 3. *Modified Next Reaction Method* (MNRM) [5, 1].

**2.3.1 Stochastic Simulation Algorithm (SSA)**

In [6], SSA was popularised by Gillespie to simulate paths of SRNs that could be used to describe evolution of chemical reactions. The algorithm involves sampling two random numbers at each step: i) first random number is used to sample the time to next reaction, let's say  $h$ , and ii) second random number is used to sample which reaction takes place after time  $h$ . The algorithm is based on an observation from (2) that, given  $X(t) = x$  at time  $t$ , the probability of  $\mathcal{R}_j$  being the only reaction that takes place in time interval  $(t, t + h)$  is given by  $a_j(x) \times \exp(-a_0(x)h)$  which can be split and rewritten as  $(a_j(x)/a_0(x)) \times a_0(x)\exp(-a_0(x)h)$ .

As observed from (4), given  $X(t) = x$  at time  $t$ , time to next reaction has an exponential distribution with parameter  $a_0(x)$ . Hence, the probability of firing reaction  $\mathcal{R}_j$  in time interval  $(t, t + h)$  can be split into two independent probabilities: probability of selecting reaction  $\mathcal{R}_j$  from set of all the reactions, and probability density that follows an exponential distribution with parameter  $a_0(x)$ .

SSA involves the following sequence of steps to produce an exact simulation of a path of an SRN  $X(t), t \geq 0$ :

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**Algorithm 1** Stochastic Simulation Method

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**Algorithm: Stochastic Simulation Method**  
**Initialize**  $\mathbf{x} = x_0, \mathbf{t} = 0$   
**Declare** array  $\mathbf{a}$   
**Declare constant**  $\mathbf{T}$ , array  $\nu$   
**while**  $\mathbf{t} < \mathbf{T}$  **do**  
  **for** each reaction-channel  $\mathbf{i}$  **do**  
    Calculate propensity  $\mathbf{a}[\mathbf{i}]$  using  $x$   
  **end for**  
  **Set**  $\mathbf{a}_0 = \sum_j \mathbf{a}[\mathbf{j}]$   
  **Generate** a random number  $\mathbf{r}_1$  from  $unif(0, 1)$   
  **Set**  $\mathbf{h} = \frac{1}{\mathbf{a}_0} \ln\left(\frac{1}{\mathbf{r}_1}\right)$   
  **Generate** a random number  $\mathbf{r}_2$  from  $unif(0, 1)$   
  **Find**  $\mathbf{m}$  such that  $\sum_{k=1}^{m-1} \mathbf{a}[\mathbf{k}] < \mathbf{a}_0 \mathbf{r}_2 \leq \sum_{k=1}^m \mathbf{a}[\mathbf{k}]$   
  **Set**  $\mathbf{x} = \mathbf{x} + \nu[\mathbf{m}], \mathbf{t} = \mathbf{t} + \mathbf{h}$   
**end while**

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It can be observed from the algorithm that for each reaction that is fired in the simulation, two

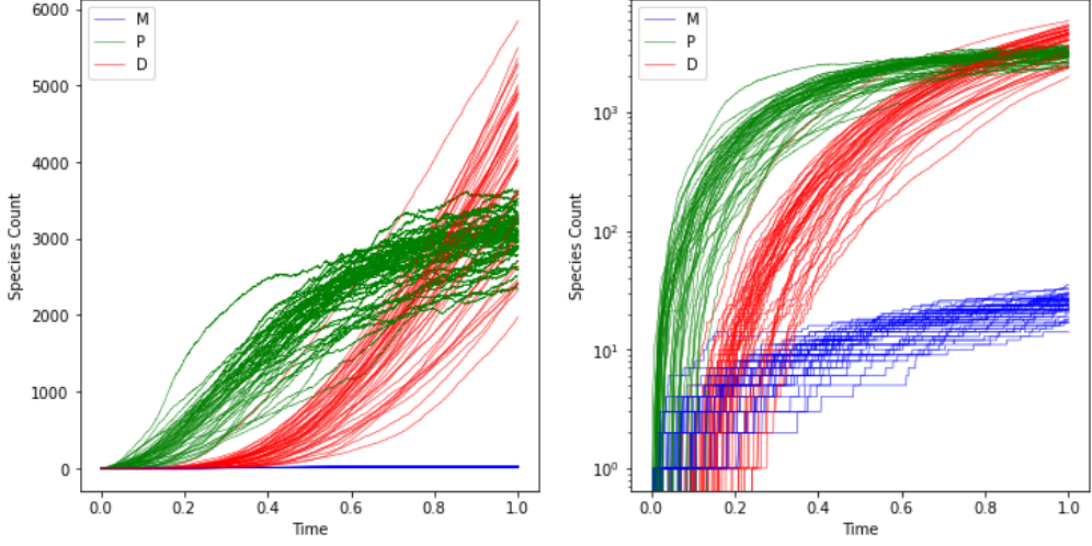


Figure 2: Fifty exact paths of gene transcription and translation SRN model, using SSA

random numbers are being generated, first to sample the time to next reaction and second to sample the reaction that should be fired based on their propensities. We will see in the following sections that NRM and MNRM require generation of only one random number each time a reaction is fired.

The plots in Figure 2 show 50 exact paths of the gene transcription and translation SRN discussed above using Stochastic Simulation Algorithm.

### 2.3.2 Next Reaction Method (NRM)

The representation (7) of SRN paths as shown by Kurtz is given as, assuming SRN  $\mathbf{X}$  is initially at state  $x_0$ :

$$X(t) = x_0 + \sum_{j=1}^m \nu_j Y_j \left( \int_0^t a_i(X(s)) ds \right).$$

The representation (7) separates the randomness of the system from the state of the system, as the randomness is only contained in the  $Y_j$ s which are unit-rate independent Poisson processes that count the number of times the corresponding reactions  $\mathcal{R}_j$  fired until time  $t$ . The intensities of the  $Y_j$ s are given by  $a_j(X(t))$ . Hence, each  $Y_j$  comes with its own time-frame, apart from the actual time frame of the chemical system. We can define  $T_j(t) = \int_0^t a_j(X(s)) ds$ , to represent the *internal times* of the corresponding  $Y_j$ . Using internal times  $T_j(t)$  in simulating SRN paths requires solving two crucial challenges: i) how to sample firing times of each  $Y_j$  in its own time frame and ii) how to translate these firing times to the actual time of the chemical system. We will see that NRM and MNRM try to solve these challenges in two different ways but both essentially use the same property of separation of randomness from the state of the system as suggested in (7).

For an SRN with  $M$  reactions, NRM first calculates the firing times of all the reactions, say  $\delta t_j, j \leq M$  at time  $t = 0$  by sampling from exponential distributions with internal times of  $Y_j$ s as parameters, based on the assumption the  $a_j$  remain unchanged until  $j^{th}$  reaction is fired. The reaction with the least firing time is fired first. After firing a reaction, the change in state of the system leads to change in propensities of the reaction-channels. This change in propensities requires the firing times to be updated, but the internal times of  $Y_j$  until their next firing remain the same. In the subsequent steps, the reaction with the nearest firing time is fired, and the firing times of the rest of the reactions are updated accordingly. For the reaction that got fired, new reaction time is sampled based on the updated propensity. The simulation is carried out until it is explicitly stopped or time limit of the simulation is reached.

Lets say, at some time  $t$ , we know  $X(t)$ , internal times  $T_j = T_j(t)$  and propensities  $a_j = a_j(X(t))$ . Assuming  $a_j$  remains constant over the next firing time of  $Y_j$ , we can also derive  $\delta t_j$  for each reaction. Now, the next reaction after time  $t$  that has to be fired should be the one with the minimum  $\delta t_j$ . Let us denote this value, i.e.  $\min_j \delta t_j$  as  $\Delta$ , and let  $\mu$  be the value of  $j$  with minimum  $\delta t_j$ . The system can then be updated to  $\bar{t} = t + \Delta$  and all the propensities can be updated to  $\bar{a}_j = a_j(X(\bar{t}))$ . For  $j = \mu$ , next firing time should be sampled using the updated propensity  $a_\mu(X(\bar{t}))$ . As the propensity functions have changed for all the reactions, the next firing time would not be same as old firing times for  $j \neq \mu$ . But it is to be noted that the internal times of the reactions until their next firing remain the same. Internal times that have passed until  $\bar{t}$  are given by  $T_j(\bar{t}) = T_j(t) + a_j \Delta$ . Hence the amount of internal time that should pass before firing  $Y_j$  is given by

$$(T_j(t) + a_j \delta t_j) - (T_j(t) + a_j \Delta) = a_j(\delta t_j - \Delta).$$

Also, we note that the difference in internal times between  $t$  and  $\bar{t}$  for  $Y_j$  are given by  $\bar{a}_j \delta \bar{t}_j$ . Hence, equating the differences between internal times for reactions, we get

$$\delta \bar{t}_j = \frac{a_j}{\bar{a}_j}(\delta t_j - \Delta).$$

NRM generates  $M$  random numbers during initialisation to sample the firing times of all the reactions. After the first step, it requires only one random variable per firing to sample the next firing time of the reaction that fired in the current step. NRM algorithm involves following sequence of steps:

Unlike SSA where two random variables were required in each step, NRM requires only one random number per firing of a reaction.

### 2.3.3 Modified Next Reaction Method (MNRM)

MNRM is similar to NRM in the way that it uses Kurtz' representation of evolution of  $X$  to simulate paths of  $\mathbf{X}$ , but unlike NRM, it does not require time conversions and works explicitly with internal

**Algorithm 2** Next Reaction Method

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**Algorithm: Next Reaction Method**  
**Initialize**  $\mathbf{x} \leftarrow x_0$ ,  $\mathbf{t} \leftarrow 0$   
**Declare** array  $\mathbf{a}$ , array  $\tau$   
**Declare constant**  $\mathbf{T}$ , array  $\nu$   
**for** each reaction-channel  $\mathbf{i}$  **do**  
    **Calculate** propensity  $\mathbf{a}[\mathbf{i}]$  using  $x$   
    **Generate** random number  $\mathbf{r}$  from  $unif(0, 1)$   
    **Set**  $\tau[\mathbf{i}] = (1/\mathbf{a}[\mathbf{i}]) \ln(1/\mathbf{r})$   
**end for**  
**while**  $\mathbf{t} < \mathbf{T}$  **do**  
    **Set**  $\mathbf{t}_{\text{next}} = \min_j \tau[\mathbf{j}]$ ,  $\mu = \arg \min_j \tau[\mathbf{j}]$   
    **Set**  $\mathbf{x} = \mathbf{x} + \nu[\mu]$ ,  $\mathbf{t} = \mathbf{t}_{\text{next}}$   
    **for** each reaction-channel  $\mathbf{i}$  **do**  
        **Calculate** propensity  $\bar{\mathbf{a}}[\mathbf{i}]$  using  $x$   
        **if**  $\mathbf{i} \neq \mu$  **then**  
            **Set**  $\tau[\mathbf{i}] = \mathbf{t} + (\mathbf{a}[\mathbf{i}]/\bar{\mathbf{a}}[\mathbf{i}])(\tau[\mathbf{i}] - \mathbf{t})$   
        **end if**  
    **end for**  
    **Generate** a random number  $\mathbf{r}$  from  $unif(0, 1)$   
    **Set**  $\tau[\mu] = \mathbf{t} + (1/\bar{\mathbf{a}}[\mu]) \ln(1/\mathbf{r})$   
    **for** each reaction-channel  $\mathbf{i}$  **do**  
        **Set**  $\mathbf{a}[\mathbf{i}] = \bar{\mathbf{a}}[\mathbf{i}]$   
    **end for**  
**end while**

---

times to calculate next firing times of reactions. At some time  $t$ , let  $P_j$  denote the first firing internal time of  $Y_j$  that is strictly larger than  $T_j$ :  $P_j = \min_s \{s > T_j : Y_j(s) > Y_j(T_j)\}$ , where  $T_j$ s are the internal times of  $Y_j$ s until  $t$ . Hence the next firing time of the reaction can be given as:

$$\delta t_j = (P_j - T_j)/a_j.$$

The internal time until the next firing of a reaction  $Y_j$  should remain constant irrespective of the evolution of the system until  $Y_j$  is fired, only the actual firing time changes based on the changes in propensities. MNRM uses this fact and lets us sample internal times, here  $P_j$ , directly instead of sampling actual next times of reactions. MNRM algorithm is as follows:

MNRM calculates  $M$  random numbers during initialisation to sample the internal times of first firings of all the  $Y_j$ s, and subsequently uses only one random number per iteration to re-sample the internal time of the fired reaction. MNRM and NRM differ in the way they use random numbers. MNRM utilizes random numbers to sample the internal times until the first firing, eliminating the need for time-conversions as in NRM.

**Algorithm 3** Modified Next Reaction Method

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**Algorithm: Modified Next Reaction Method****Initialize**  $\mathbf{x} = x_0$ ,  $\mathbf{t} = 0$ , array  $\mathbf{P} = 0$ , array  $\mathbf{T} = 0$ **Declare** array  $\mathbf{a}$ , array  $\delta t$ **Declare constant**  $\mathbf{T}$ , array  $\nu$ **for** each reaction-channel  $\mathbf{i}$  **do**    Calculate propensity  $\mathbf{a}[\mathbf{i}]$  using  $\mathbf{x}$     Generate random number  $\mathbf{r}$  from  $unif(0, 1)$     Set  $\mathbf{P}[\mathbf{i}] = (1/\mathbf{a}[\mathbf{i}]) \ln(1/\mathbf{r})$ **end for****while**  $\mathbf{t} < \mathbf{T}$  **do**    **for** each reaction-channel  $\mathbf{i}$  **do**        Set  $\delta t[\mathbf{i}] = (\mathbf{P}[\mathbf{i}] - \mathbf{T}[\mathbf{i}])/\mathbf{a}[\mathbf{i}]$     **end for**    Set  $\Delta = \min_i \delta t[\mathbf{i}]$ ,  $\mu = \arg \min_i \delta t[\mathbf{i}]$     Set  $\mathbf{t} = \mathbf{t} + \Delta$ ,  $\mathbf{x} = \mathbf{x} + \nu[\mu]$     **for** each reaction-channel  $\mathbf{i}$  **do**        Set  $\mathbf{T}[\mathbf{i}] = \mathbf{T}[\mathbf{i}] + \mathbf{a}[\mathbf{i}]\Delta$     **end for**    Generate a random number  $\mathbf{r}$  from  $uniform(0, 1)$     Set  $\mathbf{P}[\mu] = \mathbf{P}[\mu] + \ln(1/\mathbf{r})$     **for** each reaction-channel  $\mathbf{i}$  **do**        Calculate  $\mathbf{a}[\mathbf{i}]$  using  $\mathbf{x}$     **end for****end while**

---

**2.3.4 Approximate Algorithms - Tau-Leap Algorithm**

Exact algorithms, although being useful in simulating paths of SRNs with correct distributions, may not be feasible for many reasons. High propensities can lead to lower and lower reaction times, and hence increase computation load of the algorithms. Approximate algorithms mitigate these challenges by simulating paths of SRNs over discrete time steps. One such algorithm is the Tau-leap algorithm [10].

In this algorithm, a small time interval  $\tau$  is fixed and a path of an SRN  $\mathbf{X}$  is simulated by calculating  $X(t)$  at  $t = 0, \tau, 2\tau, \dots, N\tau$  where  $N\tau$  is the time limit of the simulation. After  $k$  time-steps, when  $t = k\tau$ , the algorithm progresses by sampling the number of each kind of reaction that would take place in the interval  $(k\tau, (k+1)\tau]$  and update the state of the system accordingly. Let the number of reactions that take place in the interval  $(k\tau, (k+1)\tau]$  for  $M$  reaction channels be  $(\mu_{k+1,1}, \mu_{k+1,2}, \dots, \mu_{k+1,M})$ , then the system is updated as follows:

$$X((k+1)\tau) = X(k\tau) + \sum_{j=1}^M \mu_{k+1,j} \nu_j.$$

If the initial state of the SRN is  $X(0) = x_0$ , and if the time interval  $(0, T]$  is partitioned into  $N$

time intervals,  $(0, s_1], (s_1, s_2], \dots, (s_{N-1}, s_N]$ , the integral  $\int_0^t a_j(X(s))ds$  in Kurtz's representation (7) can be approximated as a summation over discrete time intervals so that we get

$$\int_0^t a_j(X(s))ds \approx \sum_{q=0}^{N-1} a_j(X(s_q))(s_{q+1} - s_q).$$

Let us represent the path simulated by approximate methods by  $\bar{X}(t)$  to differentiate it from exact paths. The approximate path  $\bar{X}$  that can be modeled by approximating the integrals in random time change representation (7) is thus given as:

$$\bar{X}(t) = x_0 + \sum_{j=1}^m Y_j \left( \sum_{q=0}^{N-1} a_j(\bar{X}(s_q))(s_{q+1} - s_q) \right) \nu_j. \quad (8)$$

While simulating  $\bar{X}(t + \tau)$  given the state  $\bar{X}(t)$ , the second term in (8) becomes  $Y_j \left( \sum_{q=0}^{N-1} a_j(\bar{X}(s_q))(s_{q+1} - s_q) + a_j(\bar{X}(t))\tau \right)$ .  $Y_j$  becomes a Poisson process with internal time as sum of two terms. Hence,  $Y_j$  can be split into two terms:  $Y_j \left( \sum_{q=0}^{N-1} a_j(\bar{X}(s_q))(s_{q+1} - s_q) \right)$  and a Poisson random Variable  $\mathcal{P}_j(a_j(\bar{X}(t))\tau)$ . The first term is already known as it was used to arrive at the state  $\bar{X}(t)$ . Hence the expression for  $\bar{X}(t + \tau)$  becomes:

$$\bar{X}(t + \tau) = \bar{X}(t) + \sum_{j=1}^M \mathcal{P}_j(a_j(x)\tau) \nu_j,$$

where  $\mathcal{P}_j(\lambda_j), j = 1, 2, \dots, M$ , are independent Poisson processes with intensity  $\lambda_j$ .

Tau-Leap algorithm involves the following sequence of steps:

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**Algorithm 4** Tau-Leap Algorithm

---

```

Initialize  $\mathbf{x} = x_0, \mathbf{t} = 0$ 
Declare array  $\mathbf{a}$ , array  $\mathbf{p}$ 
Declare constant  $\tau_0, \mathbf{T}$ , array  $\nu$ 
while  $\mathbf{t} < \mathbf{T}$  do
  for each reaction-channel  $\mathbf{i}$  do
    Calculate propensity  $\mathbf{a}[\mathbf{i}]$  using  $\mathbf{x}$ 
    Generate random number  $\mathbf{p}[\mathbf{i}]$  from Poisson distribution  $\mathcal{P}(\mathbf{a}[\mathbf{i}]\tau)$ 
  end for
  Set  $\mathbf{t} = \mathbf{t} + \tau, \mathbf{x} = \mathbf{x} + \sum_{j=1}^M \mathbf{p}[\mathbf{j}]\nu[\mathbf{j}]$ 
end while

```

---

The plots in Figure 3 show 50 approximate paths of gene transcription and translation SRN simulated using tau-leap algorithm.

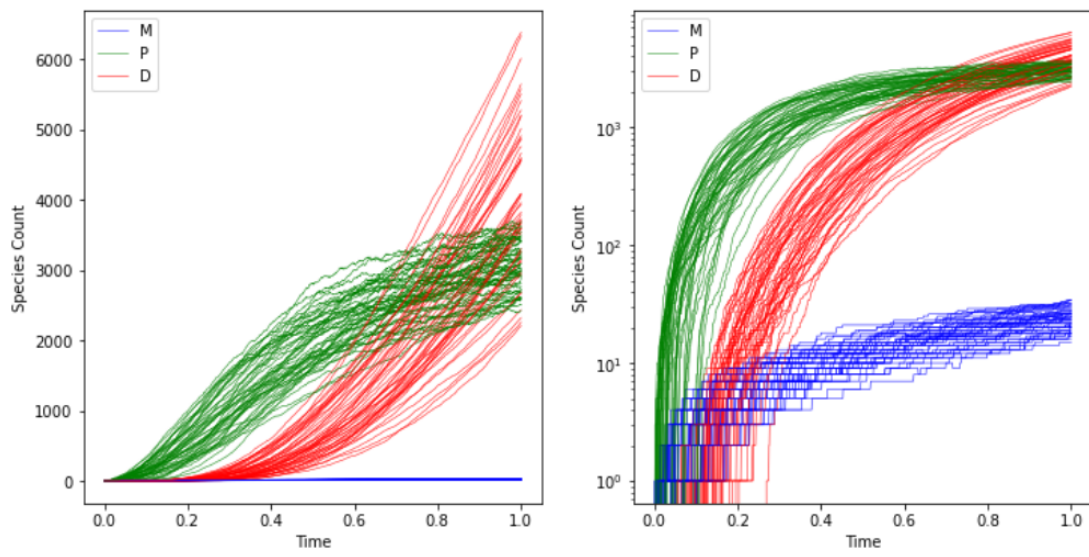


Figure 3: Fifty approximate paths of gene transcription and translation SRN model, using Tau-leap Algorithm



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## 3 Analogy between Supply Chains and Chemical Reaction Modelling

### 3.1 Discrete-event simulation (DES) of supply chains

The DES is a widely used algorithm to simulate logistics and supply chains. The *events* and *state* in DES can be modelled subject to the specific goal and usecase of the simulation. Each event that takes place while simulation of a path, the state of the system changes instantaneously. One kind of DES is the next-event time-advance algorithm where the time leaps to the next nearest event among the list of future events. It is worth noting that next-even time-advance is analogous to the stochastic simulation reaction (SSA) that was mentioned in section (2.3.1). The events in the DES algorithms that are used in simulating supply chains are analogous to chemical reactions in SSA, whereas the state of . the complexity of the algorithms depends on the number of events that take place within a time period for a given set of events.

### 3.2 Time-bucket method

In the time-bucket method, a time interval  $\tau$  is fixed and the algorithm progresses by simulating the number of each kind of events that take place in the time intervals  $(0, \tau], (\tau, 2\tau], \dots$  until the time limit of the simulation is reached, unlike the DES where each discrete event is simulated at their exact simulated times. The time-bucket method is, thus, an approximation of the DES, similar to how the tau-leap algorithm (2.3.4) used in simulating evolution of chemical systems, is an approximation of the exact method SSA.

In a simulation of supply chains using time-bucket method, we consider  $(P_1, P_2, \dots, P_d)$  as the supply chain's parts that are manufactured and transported between different processes. These parts are analogous to the species of molecules  $(S_1, S_2, \dots, S_d)$  in a chemical reaction as discussed in section 2.2. There are  $J$  processes, where each process  $j \leq J$  can be represented as follows:

$$\sum_{i \in A_j} c_{ij} P_i = \sum_{i \in B_j} p_{ij} P_i,$$

where  $A_j$  and  $B_j$  are the sets of parts consumed and produced respectively by process  $j$ . If we consider  $c_{ij}$  and  $p_{ij}$  as the  $i^{th}$  elements of vectors  $C_j \text{ in } \mathbb{Z}_+^d$  and  $P_j \text{ in } \mathbb{Z}_+^d$  respectively, then the vector  $P_j - C_j$  is equivalent to the stoichiometric vector  $\nu_j$  of an reaction(here process)  $j$ . Given state  $X(t) = (x_1(t), \dots, x_d(t))$  at time  $t$ , each process  $j \leq J$  is also associated with a production rate  $a_j(X(t))$  the describes the rate of consumption and production of parts in a single occurrence of process.

In the time-bucket method of simulation of supply chains, a time interval  $\Delta$  is fixed and the

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approximate state of the system  $\bar{X}(t) \in \mathbb{Z}_+^d$  is simulated at  $t = \Delta, 2\Delta, \dots, t \leq T$  where  $T$  is the time limit of the simulation. The state  $\bar{X}(t + \Delta)$ , given the state  $\bar{X}(t)$  is updated as follows:

$$\bar{X}(t + \Delta) = \bar{X}(t) + \sum_{j=1}^J \mathcal{P}_j(a_j(\bar{X}(t))\Delta)\nu_j,$$

where  $\mathcal{P}_j$ s are independent Poisson processes with rates given by  $a_j(\bar{X}(t))\Delta$ .

## 4 L-Leap Method

Each process in a supply chain is usually associated with a delay. A delay for a process  $j$  is the amount of time that passes between consumption and production of parts by that process. L-Leap method extends D-Leap method used in simulating chemical reactions with delays, by incorporating several features of a supply chain such as transportation, production time, inventory management and push and pull systems. The simulation of supply chains using time-bucket method can be split into two phases: instantaneous consumption and delayed production.

### 4.1 Consumption

### 4.2 Production

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