## Scientific Abstract

## Water at $\alpha$ -Alumina Surfaces: Energetics, Dynamics and Kinetics

submitted by
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Since the last decades, the field of surface science is of growing interest to understand and improve catalytic processes. Not only experimental progress was made, but also computational methods were developed and applied with good success in understanding processes at the microscopic scale. This work contributes to this progress with three different projects at the  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> surface. In all projects the interaction of water with the surface is studied using methods of periodic density functional theory including dispersion corrections, mainly with the density functional PBE. Close collaboration with the experimental "Interfacial Molecular Spectroscopy" group of Dr. Kramer Campen from the Fritz Haber institute in Berlin was maintained.

The first project is focused on the most stable surface cut under ultra high vacuum (UHV) conditions, the (0001) surface. This surface has been studied earlier in our group extensively considering water adsorption, vibrational frequencies and dissociation and diffusion reactions. In contrast to previous work, we do not apply a plane wave basis but atom centered orbital basis functions. In that way, we calculate adsorption energies with PBE, the hybrid functional B3LYP (both with D3 dispersion corrections) and also with the wave function based local Møller-Plesset perturbation theory of second order (LMP2). In addition, we compute vibrational frequencies, which show better agreement with experimental spectra than previous plane wave based results.

From the literature it is known, that generalized gradient approximation (GGA) functionals like PBE, which is standard for such surface calculations nowadays, give too low barrier heights for reactions and hence overestimate reaction rate constants. To overcome this issue, we reoptimize for an exemplary hydrogen diffusion reaction (called Df-H-4-2) the transition state with a hybrid functional (B3LYP+D3) and evaluate the barrier height and reaction rate with B3LYP and also with LMP2. These methods give indeed higher barriers and slower rate constants, although with a higher computational cost. This leads to a better understanding of the system.

In a second project, molecular beam scattering at the (0001) surface was examined with the help of plane wave DFT. In a molecular beam experiment, a water beam is shot in UHV at the surface, resulting in a non-equilibrium situation. Recent experiments showed, that with this method water dissociation is enhanced in comparison to other probing techniques like pinhole dosing. For this *ab initio* molecular dynamics simulations were performed to reproduce and explain these experimental findings. We found molecular adsorption but also dissociative adsorption, the latter directly after impact and also after initial bouncing or molecular adsorption processes up

to 1 ps after the impact, which we call indirect. We used different surface and beam models, with and without temperature effects which allowed us not only to reproduce but also to understand this increased dissociation probability mechanistically.

The last project that makes up the largest part of this work studied the  $(11\bar{2}0)$  surface with periodic DFT in a plane wave basis and the functional PBE. It is the third most stable surface cut under UHV conditions and has not been studied to a large extent yet. The structure of the clean surface slab  $(2 \times 2 \text{ supercell})$  was optimized and analyzed to predict most probable adsorption sites. The surface is relatively complex with two distinct alumina atom dimers and both twofold and threefold coordinated oxygen atoms, offering a variety of adsorption sites.

The adsorption of water was studied in the low coverage limit, concerning the stability of the molecularly and dissociatively adsorbed water. In comparison to the molecular minimum, the dissociated structures are substantially more stable.

From these stable minima, reactions were investigated with nudged elastic band (NEB) to determine reaction rate constants. This was done for dissociation and diffusion reactions of H and OH residues to learn about the mobility of surface species. We found, that dissociation is very fast and that rates for diffusion reactions cover a wide range from  $10^{-13}$ - $10^6$ s<sup>-1</sup>, depending on the specific reaction and the type of surface oxygen atom involved.

In addition, vibrational frequencies for all minima were evaluated. Here, we used OD species to be able to compare to experimental results, that employ heavy water (D<sub>2</sub>O). For this SFG spectra (sum frequency generation, an interface specific vibrational spectroscopy method) were measured by Yanhua Yue from Kramer Campen's group and compared to theory with good agreement such that single modes could be assigned to their respective OD stretch vibration. Two of the most stable species contribute to the peaks, since they are likely to be populated with respect to Boltzmann distribution. Furthermore we studied higher water coverages and the effects on the vibrational frequencies.

In this work we were able to recalculate adsorption energies, vibrational frequencies with a better agreement to experiment and to improve the reaction rates with the help of hybrid functionals and higher level methods such as LMP2 using atom centered bases. Additionally, in this work the microscopic processes during molecular beam scattering at the (0001) surface could be observed and explained mechanistically, helping to understand dynamical processes better. This work could unravel open questions in the scientific community as the behavior of water at the  $\alpha$ -alumina(11 $\bar{2}0$ ) surface with low and higher coverages, the vibrations and reactivity. Although great progress was made, still more question arose during the work, which will be answered hopefully in future work.