All Chemists Know the FCC Structure - Do They, though?

Exploring the Concept of Packings of Sphere Packings

- What are packings of sphere packings (PSPs)?
- How many PSPs do we know?
- Do they occur in real structures?

What are PSPs?

- Sphere packings lie at the heart of structural inorganic chemistry. They provide a rationale for the classification of crystal structures. Both homogeneous [2] and heterogeneous [3] sphere packings are currently being compiled.
- Recently, we realized that a higher level of categorization is possible by considering *packings of sphere packings* (PSPs) [4].
- PSPs emerge whenever two (or more) sphere packings interpenetrate and thereby establish contact, provided the contiguity (network of contacts) in either of them is not disrupted.
- These two conditions set PSPs apart from most of the known interpenetrating structures like MgCu₂ or Cu₂O (and metal-organic frameworks, MOFs), in which the two substructures fail to establish contact.
- Higher-order PSPs (of more than two sphere packings) are also found, the fcc structure being a particularly notable example (Fig. 1).
- PSPs may not only serve for the classification of crystal structures, they sometimes show remarkable new characteristics in their own right as well, e.g. quasirandomness (Fig. 2) [4-6].

How many PSPs do we know?

• At the *Hemdsärmelkolloquium* at Augsburg (Germany) in March 2017, we initiated a search for PSPs by a call for contributions to an online compilation [1].

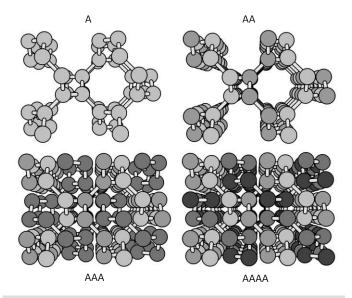


Fig. 1. The archetypal fcc structure (AAAA) - the first quaternary PSP!

- Four identical exemplars of the Laves net (A) interpenetrate and fit together perfectly, establishing contact without the need to readjust bond lengths or angles.
- A and AAA (the latter in itself a ternary PSP) are enantiomorphic.
- Either four A's of the same hand or two pairs of A's of opposite hand may be combined to form the same achiral structure AAAA (= fcc).
- This PSP epitomizes an intriguingly close relationship between the *most simple* (Laves net) and the *most efficient* (fcc) 3D-structures which exist! [Central projection parallel to a cubic unit cell edge shown.]

References:

- [1] http://www.staff.uni-marburg.de/~petrikma/welcome.html.
- [2] H. Sowa, Acta Cryst. 2018, A74, 143.
- [3] M. O'Keeffe, M. A. Peskov, S. J. Ramsden, O. M. Yaghi, Acc. Chem. Res.



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- 19. Vortragstagung der Anorganischen Chemie der Fachgruppen Wöhler-Vereinigung und Festkörperchemie und Materialforschung, 24.-27. September 2018, Universität Regensburg, Book of Abstracts, Gesellschaft Deutscher Chemiker GDCh, Frankfurt am Main, **2018**, p. P59.
 - The number of PSPs known has been growing steadily since 2017 so that now (as of September 2018) we are in a position to discuss 35 of them, for the most part previously unrecognized.

Do PSPs occur in real structures?

- The structure of a novel, quasirandom sphere packing isopointal with β -manganese consists of two interpenetrating sphere packings, or nets, forming a PSP [5].
- The two sphere packings forming this novel PSP are: on the one hand the three-connected Laves net (distorted), on the other a related net obtained by placing vertices in the centers of the bonds of the Laves net (again distorted but in a different manner).
- The Laves net may be viewed as a threedimensional variant of the *graphene* net. Then, by analogy, the second net in the foregoing structure is a kind of threedimensional *Kagome* net (vertices in place of bonds of the graphene net).
- ullet Thus, the novel PSP and consequently eta-manganese also is found to derive from two most simple and fundamental motifs: graphene and the Kagome net extrapolated into three dimensions!
- The novel PSP occurs in an almost ideal realization in the structure of RbAg_al_s where it forms the backbone network of iodine atoms (Fig. 2).

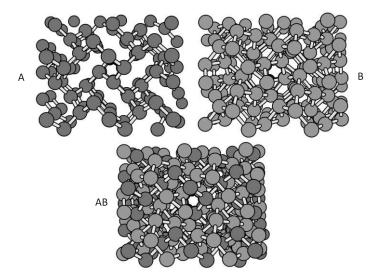


Fig. 2. The anion substructure (AB) of one of the best room-temperature ionic conductors known (RbAg₄I₅) - a remarkable PSP!

- An entanglement of a distorted Laves net (3D-folded distorted *graphene* net, A) with its similarly distorted *Kagome*-analog (i.e. bonds replaced by nodes, B).
- A, B and AB are all enantiomorphic.
- AB is isopointal with β -Mn, representing both the maximum in a minimal distance map [6] and a novel quasirandom sphere packing [4, 5]. [Central projection parallel to a cubic unit cell edge shown, centered on one of the empty iodine channels occupied by disordered Ag in RbAg₄I₅.]

2008, 41, 1782.

- [4] M. Petrik, W. Hornfeck, Z. Anorg. Allg. Chem. 2016, 642, 1023.
- [5] M. Petrik, W. Hornfeck, B. Harbrecht, *ibid.* **2014**, *640*, 2328.
- [6] W. Hornfeck, P. Kuhn, Acta Cryst. 2014, A70, 441.