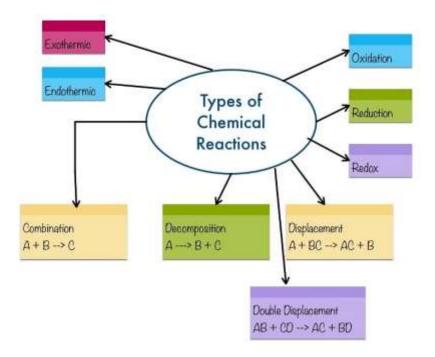
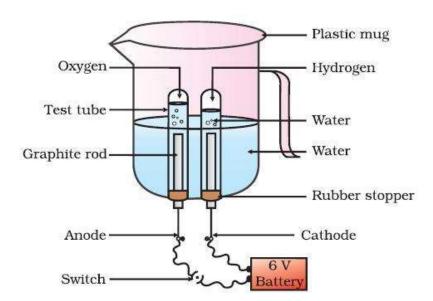
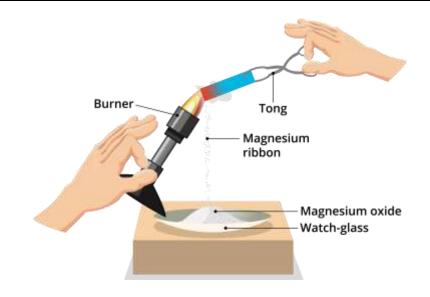
Introduction

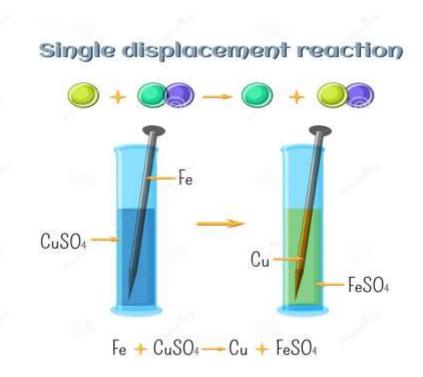




Electrolysis is an example of **decomposition** reaction. Electrolysis is the process of using electricity to split water into hydrogen and oxygen. This reaction takes place in a unit called an electrolyzer. During electrolysis, positively charged ions move to the negative electrode and negatively charged ions move to the positive electrode.



Burning Of magnesium ribbon- Write the balanced chemical equation.



This is an example of dispalcement Rection.

When an iron nail is dipped in CuSO4 solution iron can displace copper from its solution. The iron nail gets coated with a brown layer of copper and the colour of the blue copper sulphate solution changes to pale green