Cambridge International AS & A Level

CANDIDATE NAME					
CENTRE NUMBER			CANDIDATE NUMBER		

PHYSICS 9702/43

Paper 4 A Level Structured Questions

May/June 2023

2 hours

You must answer on the question paper.

No additional materials are needed.

INSTRUCTIONS

- Answer all questions.
- Use a black or dark blue pen. You may use an HB pencil for any diagrams or graphs.
- Write your name, centre number and candidate number in the boxes at the top of the page.
- Write your answer to each question in the space provided.
- Do **not** use an erasable pen or correction fluid.
- Do not write on any bar codes.
- You may use a calculator.
- You should show all your working and use appropriate units.

INFORMATION

- The total mark for this paper is 100.
- The number of marks for each question or part question is shown in brackets [].

This document has 24 pages. Any blank pages are indicated.

PMT

Data

$= 9.81 \mathrm{m s^{-2}}$
=

speed of light in free space
$$c = 3.00 \times 10^8 \,\mathrm{m \, s^{-1}}$$

elementary charge
$$e = 1.60 \times 10^{-19} \,\mathrm{C}$$

unified atomic mass unit
$$1 \text{ u} = 1.66 \times 10^{-27} \text{ kg}$$

rest mass of proton
$$m_{\rm p} = 1.67 \times 10^{-27} \, \rm kg$$

rest mass of electron
$$m_{\rm e} = 9.11 \times 10^{-31} \, \rm kg$$

Avogadro constant
$$N_A = 6.02 \times 10^{23} \,\mathrm{mol}^{-1}$$

molar gas constant
$$R = 8.31 \,\mathrm{J}\,\mathrm{K}^{-1}\,\mathrm{mol}^{-1}$$

Boltzmann constant
$$k = 1.38 \times 10^{-23} \,\mathrm{J \, K}^{-1}$$

gravitational constant
$$G = 6.67 \times 10^{-11} \,\mathrm{N \, m^2 \, kg^{-2}}$$

permittivity of free space
$$\varepsilon_0 = 8.85 \times 10^{-12} \, \mathrm{F \, m^{-1}}$$

$$(\frac{1}{4\pi\varepsilon_0} = 8.99 \times 10^9 \, \mathrm{m \, F^{-1}})$$

$$^{1}4\pi\varepsilon_{0}$$

Planck constant
$$h = 6.63 \times 10^{-34} \,\mathrm{J}\,\mathrm{s}$$

Stefan–Boltzmann constant
$$\sigma = 5.67 \times 10^{-8} \,\mathrm{W \, m^{-2} \, K^{-4}}$$

Formulae

uniformly accelerated motion
$$s = ut + \frac{1}{2}at^2$$
$$v^2 = u^2 + 2as$$

$$v^2 = u^2 + 2as$$

hydrostatic pressure
$$\Delta p = \rho g \Delta h$$

upthrust
$$F = \rho gV$$

Doppler effect for sound waves
$$f_o = \frac{f_s v}{v \pm v_s}$$

electric current
$$I = Anvq$$

resistors in series
$$R = R_1 + R_2 + \dots$$

resistors in parallel
$$\frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_2} + \dots$$

gravitational potential	ϕ	= _	$\frac{GM}{r}$
-------------------------	--------	-----	----------------

gravitational potential energy
$$E_{\rm P} = -\frac{GMm}{r}$$

pressure of an ideal gas
$$p = \frac{1}{3} \frac{Nm}{V} \langle c^2 \rangle$$

simple harmonic motion
$$a = -\omega^2 x$$

velocity of particle in s.h.m.
$$v = v_0 \cos \omega t$$

$$v = \pm \omega \sqrt{({x_0}^2 - x^2)}$$

electric potential
$$V = \frac{Q}{4\pi\varepsilon_0 r}$$

electrical potential energy
$$E_{\rm P} = \frac{Qq}{4\pi\varepsilon_0 r}$$

capacitors in series
$$\frac{1}{C} = \frac{1}{C_1} + \frac{1}{C_2} + \dots$$

capacitors in parallel
$$C = C_1 + C_2 + ...$$

discharge of a capacitor
$$x = x_0 e^{-\frac{t}{RC}}$$

Hall voltage
$$V_{\rm H} = \frac{BI}{ntq}$$

alternating current/voltage
$$x = x_0 \sin \omega t$$

radioactive decay
$$x = x_0 e^{-\lambda t}$$

decay constant
$$\lambda = \frac{0.693}{t_{\frac{1}{2}}}$$

intensity reflection coefficient
$$\frac{I_R}{I_0} = \frac{(Z_1 - Z_2)^2}{(Z_1 + Z_2)^2}$$

Stefan–Boltzmann law
$$L = 4\pi\sigma r^2 T^4$$

Doppler redshift
$$\frac{\Delta \lambda}{\lambda} \approx \frac{\Delta f}{f} \approx \frac{\textit{v}}{\textit{c}}$$

1	(a)	(i)	Define gravitational field.
		(ii)	Define electric field.
		(iii)	State one similarity and one difference between the gravitational potential due to a point
		()	mass and the electric potential due to a point charge. similarity:
			difference:
	(b)	The	[2] isolated uniform conducting sphere has mass <i>M</i> and charge <i>Q</i> . Is gravitational field strength at the surface of the sphere is <i>g</i> . Is electric field strength at the surface of the sphere is <i>E</i> .
		(i)	Show that $\frac{M}{Q} = \alpha \frac{g}{E}$
			where α is a constant.
		(ii)	Show that the numerical value of α is 1.35 × 10 ²⁰ kg ² C ⁻² .

[1]

(c)	Assume that the Earth is a uniform conducting sphere of mass 5.98×10^{24} kg.
	The surface of the Earth carries a charge of -4.80×10^5 C that is evenly distributed.

(i)	Use the information in (b) to determine the electric field strength at the surface of the
	Earth. Give a unit with your answer.

unit [2]	electric field strength =	
e Earth compares with the	State how the direction of the electric field at the surface of the direction of the gravitational field.	(ii)
[1]		
[Total: 11]		

A steel sphere of mass 0.29 kg is suspended in equilibrium from a vertical spring. The centre of the sphere is 8.5 cm from the top of the spring, as shown in Fig. 2.1.

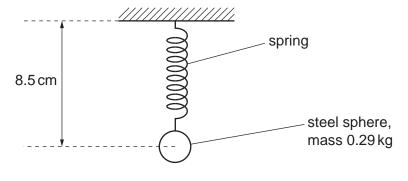


Fig. 2.1

The sphere is now set in motion so that it is moving in a horizontal circle at constant speed, as shown in Fig. 2.2.

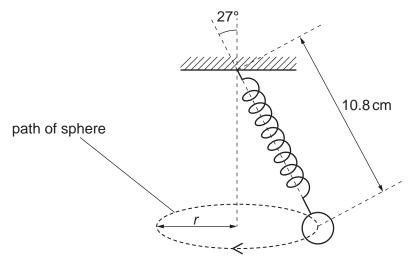


Fig. 2.2

The distance from the centre of the sphere to the top of the spring is now 10.8 cm.

(a)	Explain, with reference to the forces acting on the sphere, why the length of the spring in Fig. 2.2 is greater than in Fig. 2.1.

(b) The angle between the linear axis of the spring and the vertical is 27° .

(i)	Show that the radius r of the circle is 4.9 cm.	
(ii)	[1 Show that the tension in the spring is 3.2 N.]
(iii)	The spring obeys Hooke's law. $ \label{eq:calculate} $	[]
(c) (i)	spring constant =	1
(ii)	centripetal acceleration =	:]
	period = s [2 [Total: 12	

		s that are at the	same temperature are describe	ed as being ir
•••••				[1]
		vith temperature	e of the densities of mercury	and of water
		densi	tv	
	mercury	densi	·	
Ó	temperature/°C	100	o temperature/°C	100
		Fig. 3.1		
Tem	perature may be measured	using the variati	ion with temperature of the dens	sity of a liquid.
			·	
Sug	gest willy, for measuring term	perature over tr	ils temperature range.	
(i)	mercury is a suitable liquid			
				[1]
(ii)	water is not a suitable liquid			
Λ اـ -	adron contains a lignidat of			
	Fig. betv Terr Sug (i)	Temperature may be measured suggest why, for measuring tem (i) mercury is a suitable liquid water is not a suitable liquid A beaker contains a liquid of ma	Fig. 3.1 shows the variations with temperature between 0 °C and 100 °C. densi mercury Fig. 3.1 Temperature may be measured using the variations suggest why, for measuring temperature over the content of the conten	Fig. 3.1 shows the variations with temperature of the densities of mercury between 0°C and 100°C. density mercury water temperature/°C Fig. 3.1 Temperature may be measured using the variation with temperature of the dense Suggest why, for measuring temperature over this temperature range: (i) mercury is a suitable liquid

(c) A beaker contains a liquid of mass 120 g. The liquid is supplied with thermal energy at a rate of 810 W. The beaker has a mass of 42 g and a specific heat capacity of 0.84 J g⁻¹ K⁻¹. The beaker and the liquid are in thermal equilibrium with each other at all times and are insulated from the surroundings.

Fig. 3.2 shows the variation with time *t* of the temperature of the liquid.

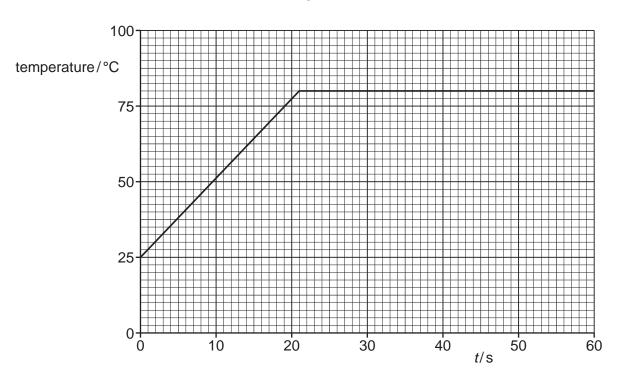


Fig. 3.2

(i) State the boiling temperature, in °C, of the liquid.

(ii) Determine the specific heat capacity, in $Jg^{-1}K^{-1}$, of the liquid.

(d) The experiment in (c) is repeated using water instead of the liquid in (c). The mass of liquid used, the power supplied, and the initial temperature are all unchanged. The specific heat capacity of water is approximately twice that of the liquid in (c). The boiling temperature of water is 100°C.

On Fig. 3.2, sketch the variation with time t of the temperature of the water between t = 0 and t = 60 s. Numerical calculations are not required. [2]

[Total: 11]

4 (a) State two of the basic assumptions of the kinetic theory of gases.

1	
2	
	[2]

(b) An ideal gas has amount of substance *n*.

The gas is initially in state X, with pressure 2p and volume V.

The gas is cooled at constant volume to state Y, with pressure p.

The gas is then heated at constant pressure to state Z, with volume 2V.

Finally, the gas returns at constant temperature to state X.

(i) Determine an expression for the temperature T of the gas in state X, in terms of n, p and V.

Identify any other symbols that you use.

[2]

(ii) On Fig. 4.1, sketch the variation with volume of pressure for the gas as the gas undergoes the three changes. The state X is labelled. Label states Y and Z.

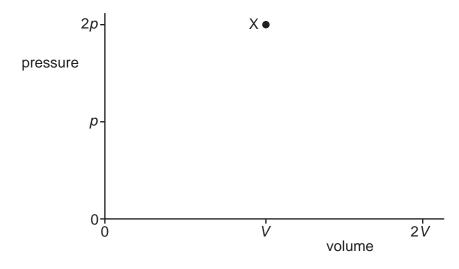


Fig. 4.1

[3]

(iii) During the change of state from Y to Z, the increase in internal energy of the gas is *U*. During the change of state from Z to X, the work done on the gas is *W*.

Complete Table 4.1 to indicate, for each of the three changes of state, the increase in internal energy of the gas, the thermal energy transferred to the gas and the work done on the gas, in terms of p, V, U and W.

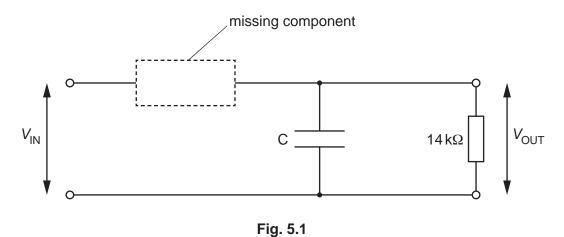
Table 4.1

change	increase in internal energy of gas	thermal energy transferred to gas	work done on gas
X to Y			
Y to Z	+U		
Z to X			+W

[5]

[Total: 12]

5 Part of an electric circuit is shown in Fig. 5.1.



The circuit is used to produce half-wave rectification of an alternating voltage of potential difference (p.d.) V_{IN} .

The output p.d. across the $14 \,\mathrm{k}\Omega$ resistor is V_{OUT}

- (a) (i) A component is missing from the circuit of Fig. 5.1.
 - Complete the circuit diagram in Fig. 5.1 by adding the circuit symbol for the missing component, correctly connected. [1]
 - (ii) A capacitor C is shown in the circuit of Fig. 5.1.

State the effect on $V_{\rm OUT}$ of including the capacitor in the circuit.

.....[1]

(b) Fig. 5.2 shows the variation with time t of V_{IN} .

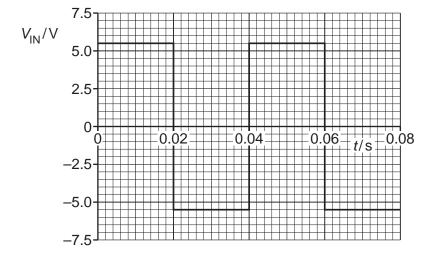


Fig. 5.2

Fig. 5.3 shows the variation with t of V_{OUT} .

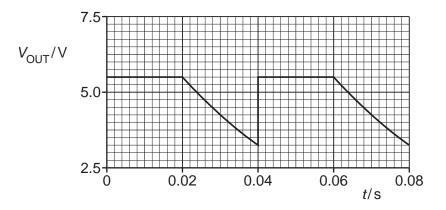


Fig. 5.3

(i) Determine the frequency of V_{IN} .

(ii) Show that the time constant τ for the discharge of the capacitor through the resistor is 0.038s.

[2]

(iii) Calculate the capacitance of C. Give a unit with your answer.

capacitance = unit [2]

(c) The circuit of Fig. 5.1 is modified so that it produces full-wave rectification of an input voltage. Suggest, with a reason, how $V_{\rm OUT}$ now varies with time when $V_{\rm IN}$ is as shown in Fig. 5.2.

[Total: 9]

6	(a)	State what is meant by a magnetic field.
		[2]
	(b)	A long, straight wire P carries a current into the page, as shown in Fig. 6.1.
		wire P current into page

8

Fig. 6.1

On Fig. 6.1, draw four field lines to represent the magnetic field around wire P due to the current in the wire.

(c) A second long, straight wire Q, carrying a current of 5.0A out of the page, is placed parallel to wire P, as shown in Fig. 6.2.

wire P	wire Q
\otimes	•
current	current 5.0 A
into page	out of page

Fig. 6.2

The flux density of the magnetic field at wire Q due to the current in wire P is 2.6 mT.

(i) Calculate the magnetic force per unit length exerted on wire Q by wire P.

force per unit length = Nm⁻¹ [2]

[Total: 10]

(ii)	State the direction of the force exerted on wire Q by wire P.	
		[1]
(iii)	The flux density of the magnetic field at wire P due to the current in wire Q is 1.5 mT.	
	Determine the magnitude of the current in wire P. Explain your reasoning.	
	current = A	[2]

/	(a)	State what is meant by the de Broglie wavelength.

(b) Fig. 7.1 shows a glass tube in which electrons are accelerated through a high p.d. to form a beam that is incident on a thin graphite crystal.

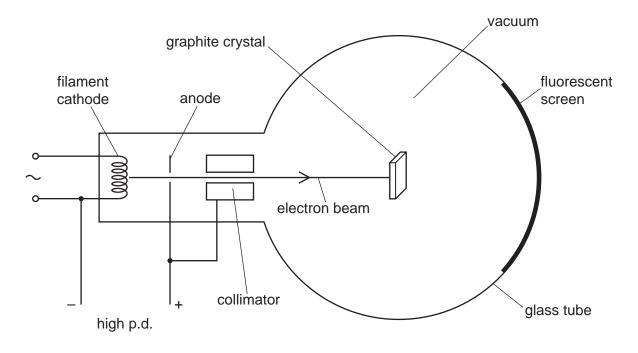


Fig. 7.1 (not to scale)

After passing through the graphite crystal, the electrons reach the fluorescent screen. The screen glows where the electrons strike it.

Fig. 7.2 shows the fluorescent screen viewed end-on, from the right-hand side of Fig. 7.1.

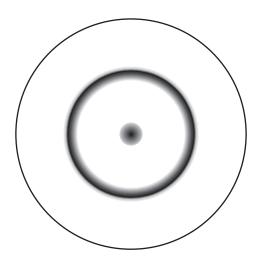


Fig. 7.2

(c)

(i)	State the name of the phenomenon demonstrated by the pattern shown in Fig. 7.2. [1]
(ii)	Explain what can be concluded from the pattern in Fig. 7.2 about the nature of electrons.
	electrons in (b) are now accelerated through a greater potential difference between the node and the anode.
(i)	On Fig. 7.3, sketch the pattern that is now seen on the fluorescent screen in Fig. 7.1.
	Fig. 7.3
(ii)	Explain, with reference to de Broglie wavelength, the change in the pattern on the fluorescent screen.
	[3]
	[Total: 9]

(a) Table 8.1 shows some data relating to the properties of air, gel and body tissue. The data are 8 given to three significant figures.

Table 8.1

material	density/kgm ⁻³	speed of sound/ms ⁻¹	specific acoustic impedance/kg m ⁻² s ⁻¹
air		340	440
gel	1200	1400	
tissue	1090		1.68 × 10 ⁶

	air			340	440	
	gel		1200	1400		
	tissue		1090		1.68 × 10 ⁶	
	(i)	Sho	ow that the specific acousti	c impedance of gel is 1.68	$\times 10^6 \mathrm{kg}\mathrm{m}^{-2}\mathrm{s}^{-1}.$	
	(ii)		nplete Table 8.1 by calcula ce below for any working t		three significant figures. Us	[1] se the
(b)) Use (i)		information in (a) to calcul air–tissue boundary	ate the intensity reflection	coefficient for:	[2]
	(ii)	a ge	intensity refle el–tissue boundary.	ction coefficient =		[2]

intensity reflection coefficient =[1]

Use your answers in (b) to explain why gel is applied to the skin during ultrasound scanning.	(c)
[2	
[Total: 8	

[Total: 9]

()	Define half-life.					
(b)	A sample contains	N_0 nuclei of car	bon-11 and r	no other nuc	lei at time $t = 0$.	
	On Fig. 9.1, sketch	n the variation wi	ith <i>t</i> of the nu	mber of nuc	lei of boron-11 i	n the sample.
	number of n					
		0.5 N ₀				
		0	20	40	60 8 t/min	0
			Fig. 9.1			
(c)	• •	reference to the mple in (b) decre			ctive decay, why	the activity of
	'ii) Ctoto with roo	acana whathar a	, radiation da	tootor place	d noor to the con	anla of carbon
	indicates a me	asons, whether a easured count ra ity of the sample	ite from the s			
	indicates a me	easured count ra ity of the sample	ite from the s	ample that is	s less than, the sa	ame as or gre
	indicates a me	easured count ra ity of the sample	ite from the s	ample that is	s less than, the sa	ame as or gre

[Total: 9]

0	(a)	State Hubbl	e's law. Identify a	any symbols that	you use.		
	(h)	Δ etar of lun	ningsity 3.8 v 10			m from the Earth.	[2]
	(D)		•				
		Calculate th	e radiant flux int	ensity at the Earl	h of the radiat	ion emitted by the	star.
				radiant flux inte	ensity =		W m ⁻² [2]
	(c)	have a wav		m. This spectral		light from this gal t from the galaxy o	
			why the wavele axy is known to h		n the Earth is	different from the	wavelength that
							[2]
		(ii) Determ	ine a value for th	ne Hubble consta	int H_0 .		
					$H_0 =$		s ⁻¹ [3]

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