

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

CANDIDATE NAME						
CENTRE NUMBER			CANDIDA NUMBER			

1912650567

PHYSICS 0625/53

Paper 5 Practical Test

October/November 2011

1 hour 15 minutes

Candidates answer on the Question Paper

Additional Materials: As listed in the Confidential Instructions

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name in the spaces at the top of the page.

Write in dark blue or black pen.

You may use a pencil for any diagrams, graphs or rough working.

Do not use staples, paper clips, highlighters, glue or correction fluid.

DO NOT WRITE IN ANY BARCODES.

Answer all questions.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.

For Examiner's Use			
1			
2			
3			
4			
Total			

This document consists of 10 printed pages and 2 blank pages.



1 In this experiment, you will determine the density of modelling clay by two methods.

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(a) Method 1.

Carry out the following instructions referring to Fig. 1.1. You are provided with a piece of modelling clay.

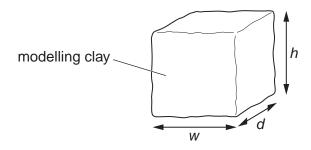


Fig. 1.1

(i)	Mould the piece of modelling clay into a shape that is approximately a cu	ube.
	Measure the height <i>h</i> , width <i>w</i> and depth <i>d</i> of the piece of modelling clay.	

h=	 cm
w=	 cm
ما	

(ii) Calculate the volume V of the piece of modelling clay using the equation V = hwd.

$$V = \dots cm^3$$

(iii) Measure the mass m of the piece of modelling clay using the balance provided.

$$m = \dots g$$

(iv) Calculate the density ρ of the modelling clay using the equation $\rho = \frac{m}{V}$.

$$\rho$$
 =[3]

(b) Method 2.

Carry out the following instructions referring to Fig. 1.2. You are provided with a beaker containing water at room temperature.

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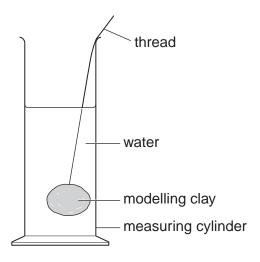


Fig. 1.2

Break the piece of modelling clay into two pieces with one piece approximately twice the size of the other piece.

(i) Using the **smaller** piece of modelling clay, measure its mass m_s .

 $m_{\rm S}$ =

(ii) Pour approximately $50 \,\mathrm{cm}^3$ of water into the measuring cylinder. Record the volume of water V_1 in the measuring cylinder.

 $V_1 = \dots V_1 = \dots V_n = \dots V_n$

(iii) Tie the thread around the smaller piece of modelling clay and lower it into the measuring cylinder until it is completely covered with water. Record the new volume V_2 .

V₂ =

(iv) Calculate the volume V_s of the modelling clay using the equation $V_s = V_2 - V_1$.

V_s =

(v) Calculate the density ρ of the modelling clay using the equation $\rho = \frac{m_{\rm s}}{V_{\rm s}}$.

ρ =[5]

(c)	Assuming that the experiment has been carried out with reasonable care, suggest two reasons why the two values obtained for the density of the modelling clay may not be the same.	For Examiner's Use
	1	
	2	
	[2]	
	[Total: 10]	

5

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			U		
2			is experiment, you will investigate the energy changes that occur when hot and color are mixed.		
	You are provided with a supply of hot water and a supply of cold (room temperature) wat				
	(a)	(i)	Pour 50 cm³ of cold water into the beaker labelled A . Measure and record the temperature $\theta_{\rm c}$ of this water.		
			$\theta_{\rm c}$ =		
		(ii)	Pour 50 cm³ of hot water into the beaker labelled ${\bf B}.$ Measure and record the temperature $\theta_{\rm h}$ of this water.		
			$\theta_{h} = \dots$		
		(iii)	As soon as you have taken the temperature, pour the 50 cm³ of water from beaker A into beaker B . Briefly stir the mixture then measure and record the temperature $\theta_{\rm m}$ of the mixture.		
			θ_{m} =		
			[3]		
	(b)	(i)	Calculate the gain in thermal energy E_c of the cold water using the equation $E_c = k(\theta_m - \theta_c)$		
			where $k = 210 \mathrm{J/^\circ C}$.		
			$E_{c} = \dots$		
		(ii)	Calculate the loss in thermal energy $E_{\rm h}$ of the hot water using the equation $E_{\rm h} = k(\theta_{\rm h} - \theta_{\rm m})$		
			where $k = 210 \mathrm{J/^\circ C}$.		
			_		
			$E_{h} = \dots $ [2]		
	(c)		udent suggests that all the thermal energy lost by the hot water is gained by the cold er. $E_{\rm c}$ and $E_{\rm h}$ should therefore be equal.		
		(i)	State whether your experimental results support this suggestion and justify your statement by reference to your results.		
			statement		
			justification		
			[2]		
		(ii)	Suggest a practical reason in this experiment why $E_{\rm c}$ might be different from $E_{\rm h}$.		

(d) Another student was asked to suggest quantities that should be kept constant if this experiment is repeated in order to check the readings. Table 2.1 shows the suggestions. Place a tick (✓) in the second column of the table next to each correctly suggested quantity.

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Table 2.1

suggestion	
avoid parallax errors when taking readings	
number of stirs	
room temperature	
starting temperature of cold water	
starting temperature of hot water	
use only two or three significant figures for the final answers	

[2]

[Total: 10]

3 In this experiment, you will investigate the resistance of a wire.

You will use the circuit shown in Fig. 3.1.

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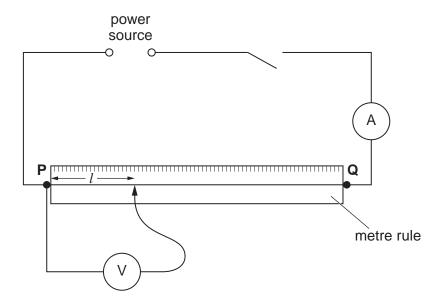


Fig. 3.1

- (a) (i) Switch on. Measure and record in Table 3.1 the current l in the circuit and the p.d. V across a length $l = 0.250 \,\mathrm{m}$ of the wire **PQ** as shown in Fig. 3.1. Switch off.
 - (ii) Calculate the resistance R of the length $l = 0.250 \,\mathrm{m}$ of wire using the equation

$$R = \frac{V}{I}$$
.

Record this value of R in the table.

- (iii) Repeat steps (i) and (ii) using *l* values of 0.500 m and 0.750 m.
- (iv) Complete the heading for each column of the table.

Table 3.1

1/	V/	1/	R/
0.250			
0.500			
0.750			

[5]

(b)	Use data from the table to suggest and justify a relationship between the length $\it l$ of the wire and its resistance $\it R$. Show your working.	For Examiner's Use
	relationship	
	justification	
	[3]	
(c)	Use your results to predict the resistance of a 1.500 m length of the same wire. Show your working.	
	prediction =[2]	
	[Total: 10]	

4 In this experiment, you will investigate the formation of images by a converging lens.

Carry out the following instructions referring to Fig 4.1.



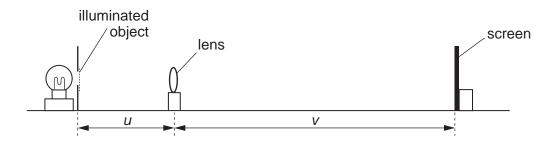


Fig. 4.1

- (a) (i) Place the screen about 1.0 m from the illuminated object.
 - (ii) Place the lens between the object and the screen so that the centre of the lens is at a distance $u = 0.200 \,\mathrm{m}$ from the object. Adjust the position of the screen until a clearly focused image is formed on the screen.
 - (iii) Measure (in metres) the distance *v* between the centre of the lens and the screen.
 - (iv) Record the values of u and v in Table 4.1.
 - (v) Repeat the steps in (i) (iv) using values for u of 0.250 m, 0.300 m, 0.350 m and 0.400 m.

Table 4.1

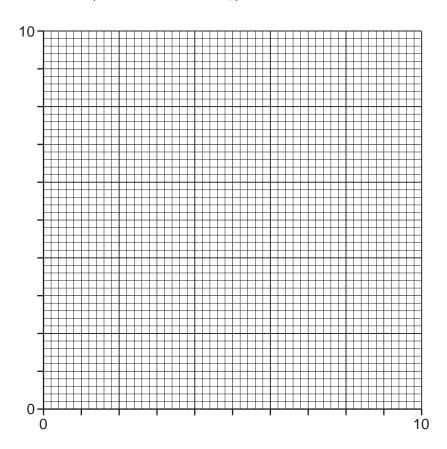
u/m	v/m	$\frac{1}{u}/\frac{1}{m}$	$\frac{1}{v}/\frac{1}{m}$

(vi) Calculate the values of $\frac{1}{u}$ and $\frac{1}{v}$ and enter them in the table.

[3]

(b) Plot the graph of $\frac{1}{V} / \frac{1}{M}$ (y-axis) against $\frac{1}{U} / \frac{1}{M}$ (x-axis).





[4]

(c) (i) Use the graph to find the intercept on the *y*-axis.

intercept on the *y*-axis =

(ii) Use the graph to find the intercept on the *x*-axis.

intercept on the *x*-axis =[2]

(d) State and briefly explain one precaution you took in order to obtain reliable measurements.

precaution

. .

.....[1]

[Total: 10]

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