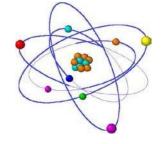


# Chemistry 10600 Hunter College of the City University of New York Department of Chemistry & Biochemistry



TA: Amanda Whittaker (pronouns: she/her/hers)

Office: 1317N

Email: awhittaker@gradcenter.cuny.edu

Office hours: Fridays 12-1 pm

Lab Coordinators: Dr. Gabriela Smeureanu (Room 1320HN) <u>asmeurea@hunter.cunv.edu</u>

Dr. Nadya Kobko (Room 1320HN) <u>nkobko@hunter.cuny.edu</u>

## Welcome to Chemistry 10600

Chemistry is a challenging and often abstract science, but as you progress through this course we hope you will discover that chemistry is also exciting and that many of the key concepts in chemistry are both important and relevant to life on earth. Throughout this semester we will provide you with the basic skills and knowledge to think and feel like a chemist. You will learn that chemistry is exciting!

The lab course runs synchronous in BB collaborate. Make sure you have access to your class in Blackboard and the email listed in your account there is the right email you use and check regularly. That's the only way we will be able to communicate with you.

#### Blackboard:

You must use your Hunter/Cuny credentials to log in on Blackboard.

Instructions on how to access the course website on blackboard can be found at: <a href="http://bb.hunter.cuny.edu">http://bb.hunter.cuny.edu</a>. You should check the site regularly. It can also be used to communicate with your classmates.

I will use Blackboard to post announcements, additional resources and helpful hints.

**Email:** You can find information about Hunter email through a link at: <a href="http://www.hunter.cuny.edu/icit/help-docs/e-mail-faq">http://www.hunter.cuny.edu/icit/help-docs/e-mail-faq</a>

Your username and password should have been mailed to you. Please make sure that if you are not using this email account, you set your mail to forward to an account that you do use. Instructions on how to forward mail can be found through the link above. Important emails will be sent through Blackboard. The information you may miss is likely to impact your grade if I don't have email contact with you.

#### A. DESCRIPTION

This course will present essential facts, laws, and theories of general chemistry.

## B. OBJECTIVES

Upon successful completion of this course, the student will be able to develop problem-solving skills, use definition, concepts, and ideas to predict how a chemical process will occur. This course will also test the understanding of the basic concepts of chemistry.

### C. GRADING POLICY

Your overall letter grade will be based on the total number of points you earn in the course. The total number of points that can be earned in this course is **1450**. This total will be converted to a percentage (out of 100%) and scaled according to the Hunter College Grading system.

12 Laboratories (Lab Reports)	100 pts
Scientific Presentations × 2	100 pts
Excel Exercise	40 pts
Safety Certificate	10 pts
Attendance/class participation/summaries	108 pts

#### D. LABORATORY POLICY:

Lab are running online synchronous in the days and at the times they we scheduled. Attendance for each laboratory experiment is **MANDATORY**. You will be allowed **one excused absence** for the semester\*. Any subsequent absences will result in a grade of zero for that experiment. If you are more than 10 minutes late for your online lab in BB collaborate you will not be allowed to complete the experiment. This will count as your one excused absence for the semester. **NO MAKE-UP labs**.

Any students who disrupt the class will be asked to leave.

Attendance points are 2 pts./class (24 classes), so students who attend every class get a small points boost at the end.

#### E. SUMMARIES:

A <u>typed</u> summary of each lab is to be submitted in Blackboard- this is to ensure you have read the instructions for the lab that will be performed that day. Lab summaries should consist of 3-4 sentences and are worth 2 points each as part of weekly Attendance/Class participation.

Videos of the lab will be provided, and the TA will discuss the labs with you. Then sets of data will be given to you to be able to write your lab reports and answer the questions.

#### F. LAB REPORTS:

Lab reports are to resemble the format found at the back of your lab manual and be turned in promptly in BB by the due date. Each student is responsible for submitting/uploading in blackboard their own lab report. Lab reports will be graded, and feedback will be provided. Your lowest lab report will be dropped- I would suggest completing all the lab reports in case you find that one is more difficult than another. LATE LAB REPORTS WILL NOT BE ACCEPTED. Some lab reports may take a full day to complete, so I recommend doing a little each day so that you're not pulling all-nighters to complete them. DO NOT WAIT UNTIL THE LAST MINUTE!

No title page or TOC - double-sided ideal. No more than 2 pages ideally (for the written portion) - obviously the graphs / data will add more pages to the report. If you can fit it all on 1 page, great, but your work must answer the grading criteria and rubric.

This is not a creative writing class. We don't want to see content like...

- "Joe walked briskly to the sink and scrubbed the beaker with all his might before using the acetone to dry it."
- 1) Definitely don't describe the washing process in your report.
- 2) Science writing is very straight to the point. If you are unsure about how to do science writing, you can look it up on the web to see how it's done. Procedure must be in bullets.

If an error was done in the experiment, explain what happened. Don't be vague or try to guess what happened - own it and you will get full points.

To get ahead on writing lab reports: Write your summaries like your introductions so you can get a head start on the next sections of the lab! Make sure that it syncs with the criteria for what an introduction should look like per the lab manual. Also, if you've read this far, please e-mail me with "Kylo Ren is the best Star Wars character" for a surprise.

Data tables and graphs must be labeled and with proper names. Students submitting identical reports will result in a zero for that report and will be reported to the Chemistry department. Calculations <u>must be typed</u> unless otherwise stated (I can't always read everyone's handwriting!).

Plagiarism will not be tolerated.

# G. MATERIALS TO BRING EACH CLASS:

Lab manual, calculator, personal laptop, notebook for recording observations.

#### In the first day of lab you will be required to have:

- Laboratory Manual: Chemistry 106 "General Chemistry Laboratory" by G. Smeureanu and S. Geggier (available for purchase ONLY at Shakespeare bookstore or Hunter College online bookstore)
- Carbon Copy Laboratory Notebook 100 pages or any notebook -- (available for purchase ONLY at Shakespeare bookstore or Hunter College online bookstore)
- Calculator
- Computer

- 1. <u>Academic Integrity Statement</u>: "Hunter College regards acts of academic dishonesty (e.g., plagiarism, cheating on examinations, obtaining unfair advantage, and falsification of records and official documents) as serious offenses against the values of intellectual honesty. The College is committed to enforcing the CUNY Policy on Academic Integrity and will pursue cases of academic dishonesty according to the Hunter College Academic Integrity Procedures."
- 2. <u>ADA Statement</u>: "In compliance with the ADA and with Section 504 of the Rehabilitation Act, Hunter College is committed to ensuring educational access and accommodations for all its registered students. Hunter College's students with disabilities and medical conditions are encouraged to register with the Office of AccessABILITY for assistance and accommodation. For information and appointment contact the Office of AccessABILITY located in Room E1214 or call (212) 772-4857 /or VRS (646) 755-3129."
- 3. <u>Hunter College Policy on Sexual Misconduct</u> "In compliance with the CUNY Policy on Sexual Misconduct, Hunter College reaffirms the prohibition of any sexual misconduct, which includes sexual violence, sexual harassment, and gender-based harassment retaliation against students, employees, or visitors, as well as certain intimate relationships. Students who have experienced any form of sexual violence on or off campus (including CUNY-sponsored trips and events) are entitled to the rights outlined in the Bill of Rights for Hunter College.
- a. Sexual Violence: Students are strongly encouraged to immediately report the incident by calling 911, contacting NYPD Special Victims Division Hotline (646-610-7272) or their local police precinct, or contacting the College's Public Safety Office (212-772-4444).
- b. All Other Forms of Sexual Misconduct: Students are also encouraged to contact the College's Title IX Campus Coordinator, Dean John Rose (jtrose@hunter.cuny.edu or 212-650-3262) or Colleen Barry (colleen.barry@hunter.cuny.edu or 212-772-4534) and seek complimentary services through the Counseling and Wellness Services Office, Hunter East 1123.

CUNY Policy on Sexual Misconduct Link: <a href="http://www.cuny.edu/about/administration/offices/la/Policy-on-Sexual-Misconduct-12-1-14-with-links.pdf">http://www.cuny.edu/about/administration/offices/la/Policy-on-Sexual-Misconduct-12-1-14-with-links.pdf</a>"

# Hunter College of the City University of New York Chemistry and Biochemistry Department

# ~ Tentative Syllabus for Chemistry 10600~

Thursday 8/27	Intro and Safety Safety video and certification Excel exercise	
Tuesday 9/01	What is around us? "Chemistry is LiFe" (Lab 1)	
Thursday 9/03	Is volume conserved? "The sneaky ethanol molecules" (Lab 2)	
Tuesday 9/08	Data Analysis "Up in the air" (Lab 3 Part A)	
Thursday 9/10	Data Analysis "Up in the air" (Lab 3 Part B)	
Tuesday 9/15	What is in my sample? "Let's find the cations (Lab 4 Part A)	
Thursday 9/17	What is in my sample? "Let's find the cations (Lab 4 Part B)	
Tuesday 9/22	Molecular models (Lab 5)	
Thursday 9/24	VSEPR and molecular shape "How does it look?" (Lab 6)	
Tuesday 9/29	No Class	
Thursday 10/01	How much is a mole? "Avogadro's Number Dilemma" (Lab 7 Part A)	
Tuesday 10/06	How much is a mole? "Avogadro's Number Dilemma" (Lab 7 Part B)	
Thursday 10/08	Molecules and reactions "Foiled again" (Lab 8 Part A)	
Tuesday 10/13	Molecules and reactions "Foiled again" (Lab 8 Part B)	
Thursday 10/15	Scientific Presentation 1	
Tuesday 10/20	Reactants and products – An Online Simulatio "I am in equilibrium" (Lab 9)	
Thursday 10/22	Le Châtelier's Principle	
•	"Where do I shift?" (Lab 10)	
Tuesday 10/27	"Where do I shift?" (Lab 10)  Acid, bases and Ka "Where is my Lewis pair?" (Lab 11)	
Tuesday <b>10/27</b> Thursday <b>10/29</b>	Acid, bases and Ka	
	Acid, bases and Ka "Where is my Lewis pair?" (Lab 11)	

Tuesday 11/10	Titration curves "Drop the Base" (Lab 15)
Thursday 11/12	Galvanic cell "LEO the lion goes GER" (Lab 16)
Tuesday <b>11/17</b>	Calorimetry and Thermochemistry "Stealing Joules" (Lab 17)
Thursday 11/19	Coffee-Cup Calorimetry "Burning Food. Where are my carbs?"(Lab 18)
Tuesday 11/24	Kinetics Iodine-Clock reaction (Lab 19)
Thursday 11/26	Thanksgiving Holiday
Tuesday 12/01	Kinetics Data, Activation energy (Lab 20)
Thursday 12/03	Scientific Presentation 2