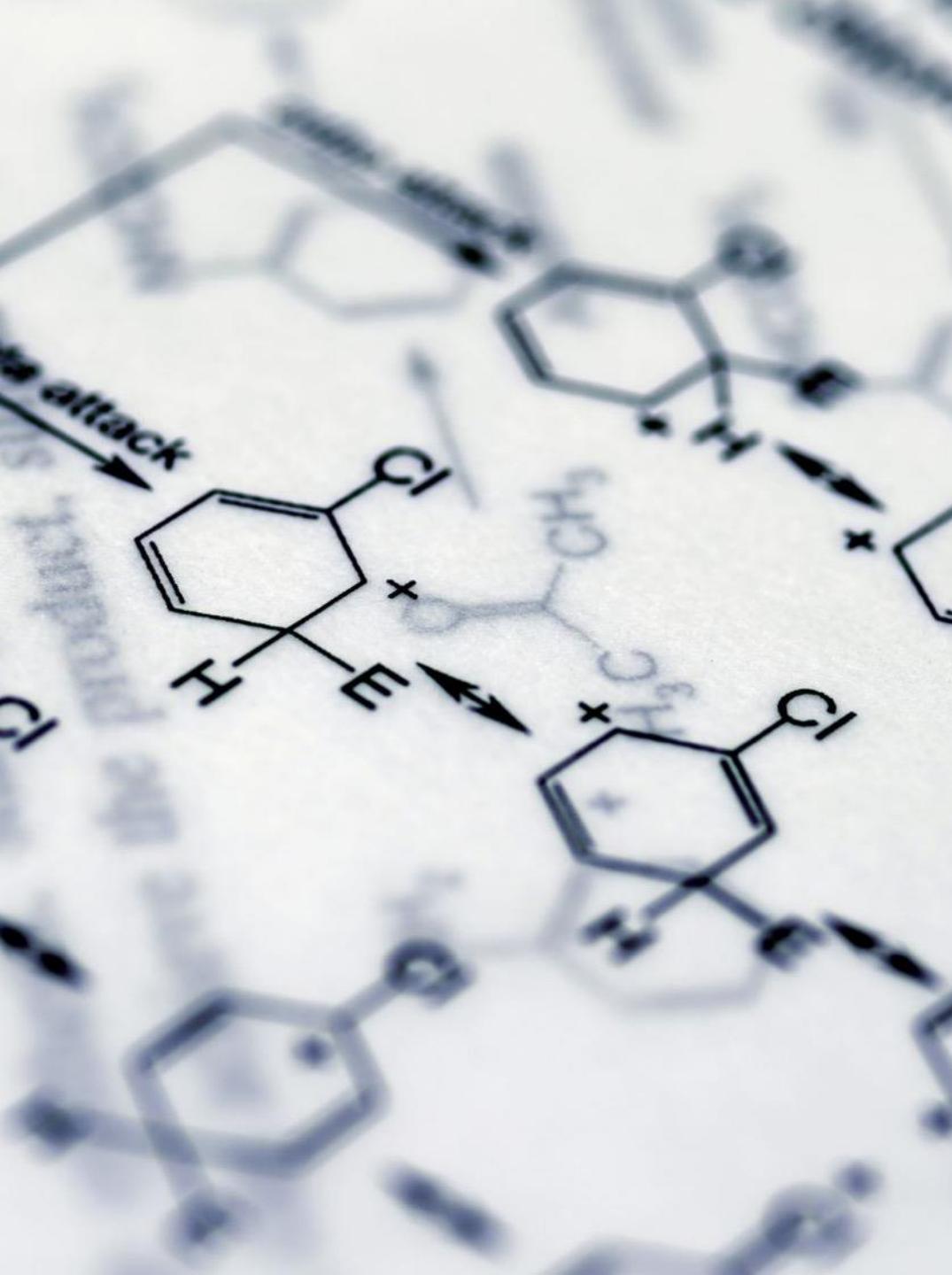


# Chapter 14 Fundamentals of Package Encapsulation, molding and sealing

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Georgia Institute of Technology, Atlanta, GA, USA



## Outline

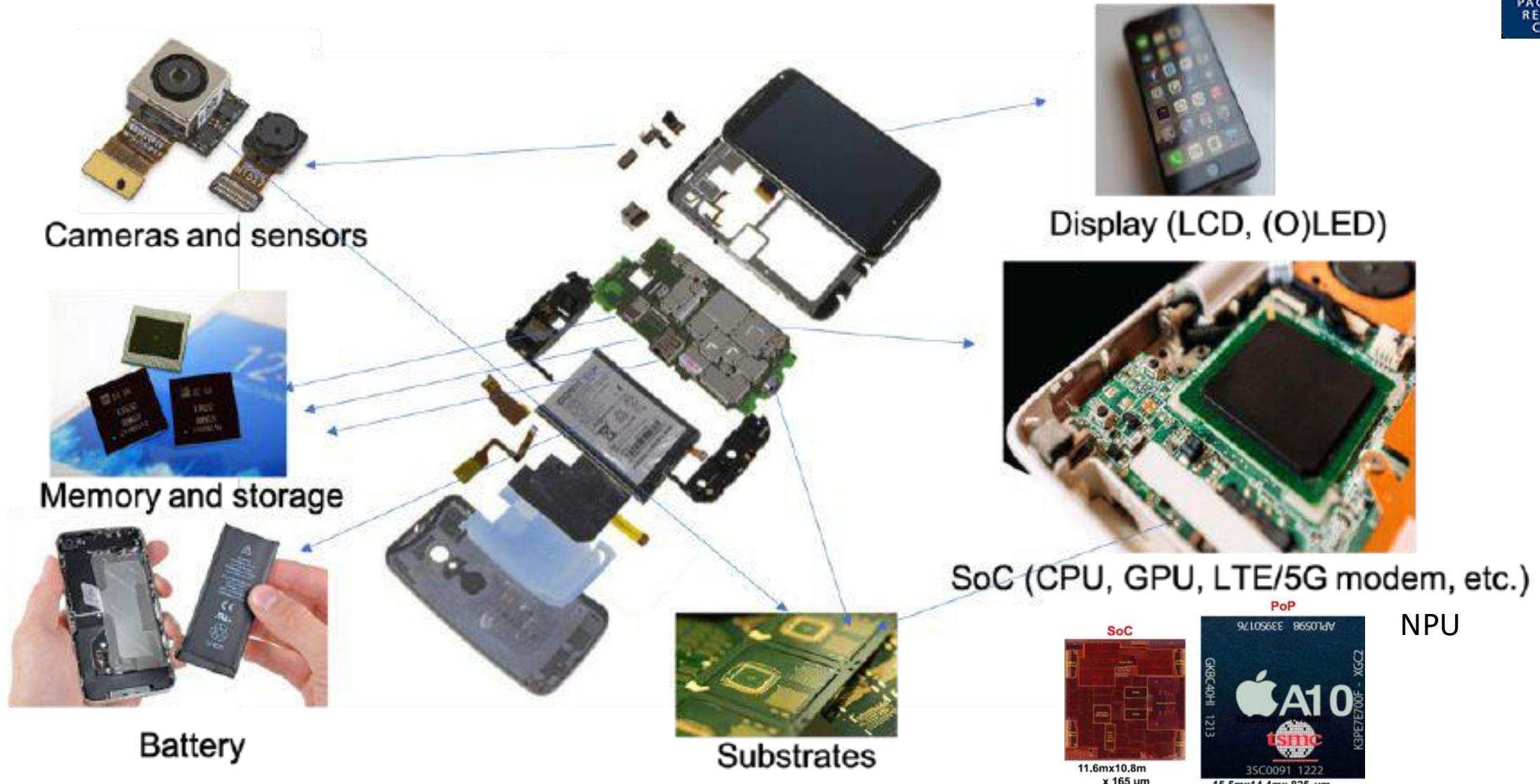
- 14.0 Materials in Electronic Packaging
- 14.1 What Is Sealing and Encapsulation and Why?
- 14.2 Anatomy of an Encapsulated and a Sealed Package
- 14.3 Properties of Encapsulants
- 14.4 Encapsulation Materials
- 14.5 Encapsulation Processes
- 14.6 Hermetic Sealing
- 14.7 Summary and Future Trends

# Electronics in various areas and demands

- Consumer electronics
    - Cell phones
    - Display
    - Computers
  - Aerospace
  - Automotive
  - Data center
- 
- Mobility; small, thin, multifunctional, flexible, etc.
  - High speed computing
  - High bandwidth communication (RF)
  - High resolution visualization
  - Safe operation and complex control/functions
  - Reliability

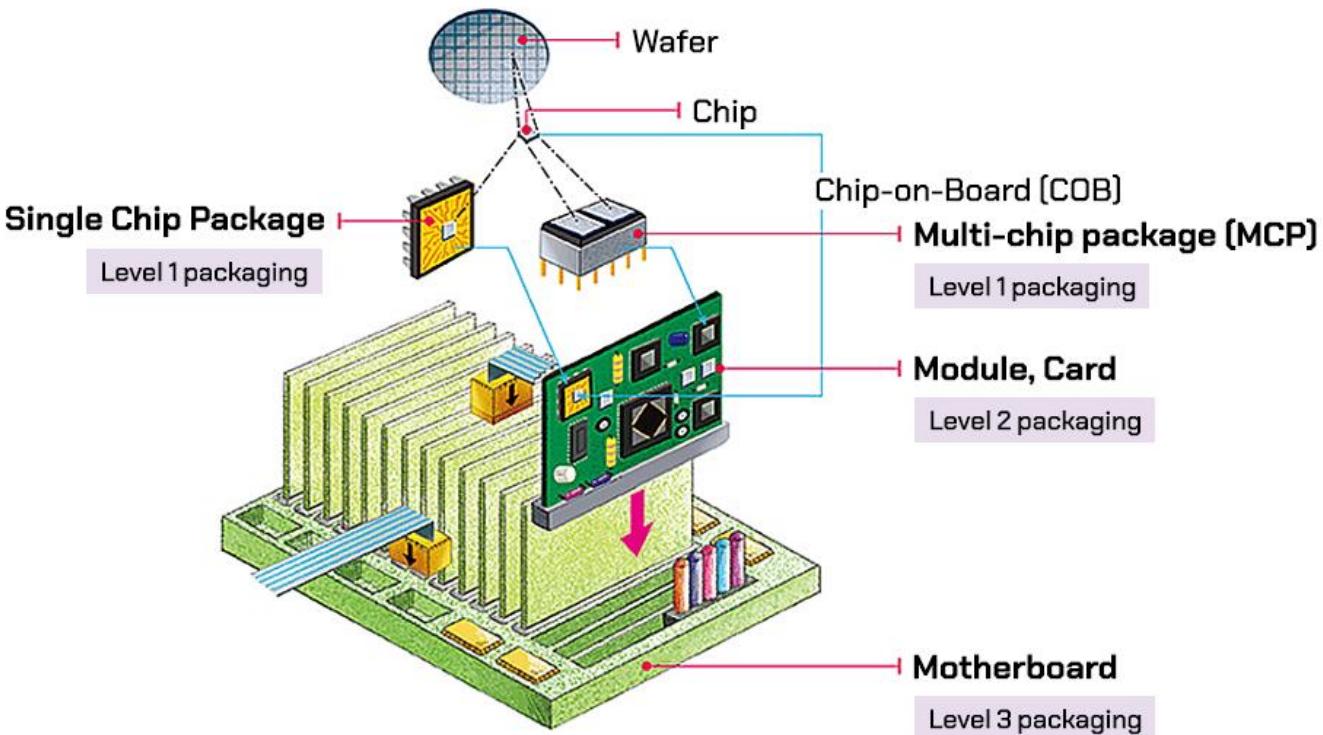


# Electronics anatomy and packaging



# Electronic Packaging

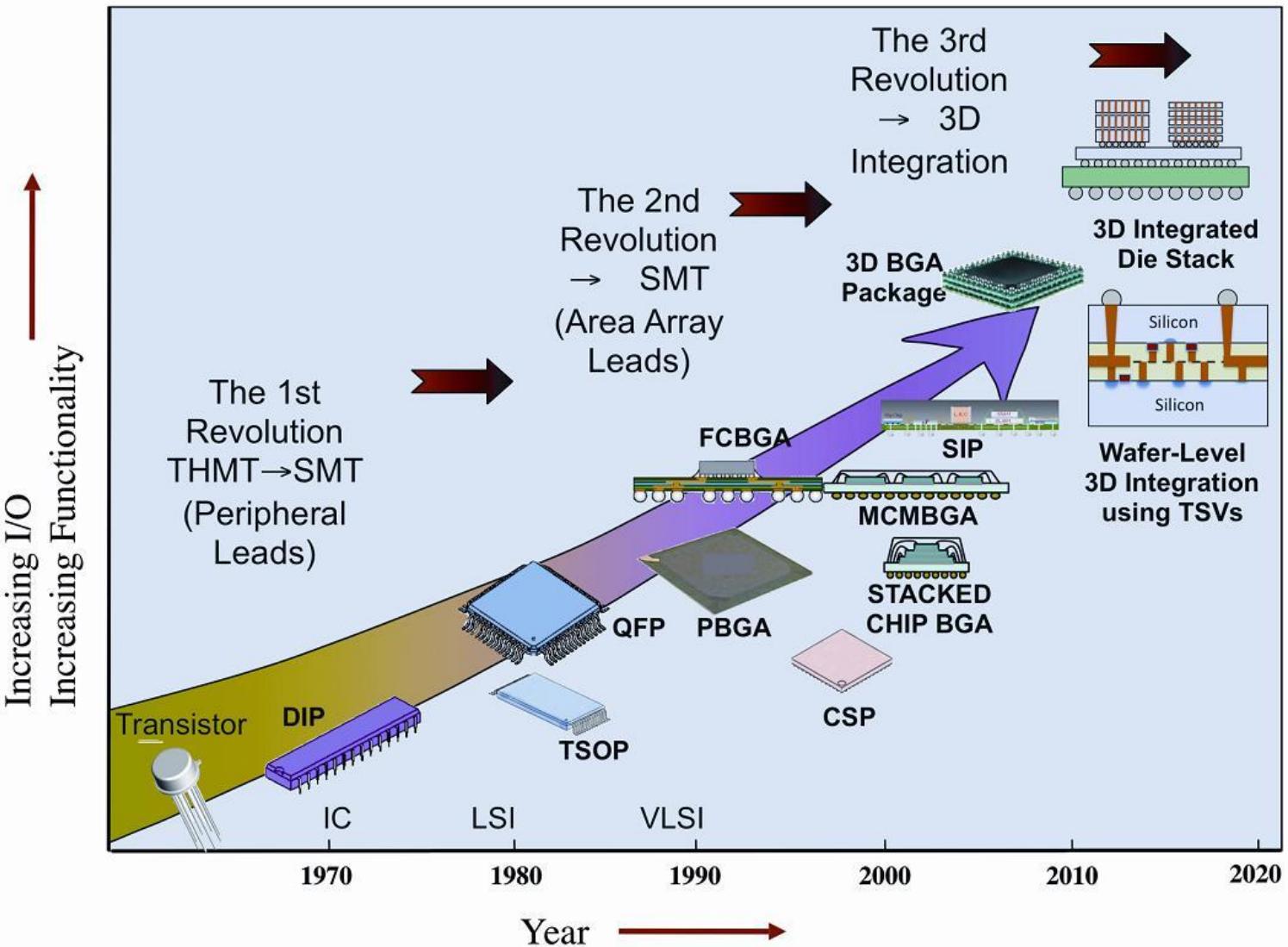
- Electronic packaging is a system integration. Defined as bridge that interconnects integrated circuits and other components into system-level board to form electronic products.
- Functions
  - *Power Distribution*
  - *Signal Distribution*
  - *Heat Dissipation*
  - ***Protection (physical/chemical)***
    - *Reliability*



Source: Principle of Electronic Packaging, p. 5

# Microelectronics Packaging Evolution

**Interconnection Technology**



# Microelectronics Packaging

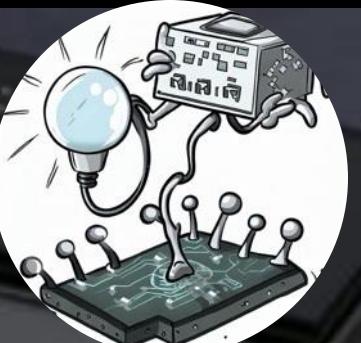
Power  
Distribution

Heat  
Dissipation

Packaging

Signal  
Distribution

Protection



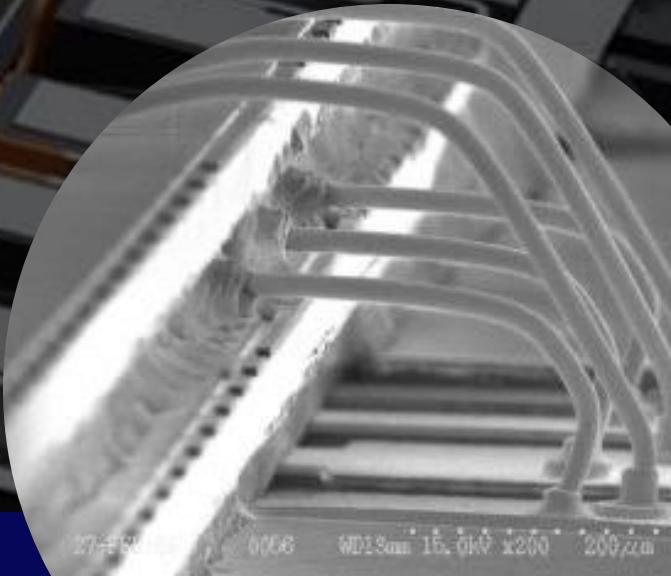
Powering



Cooling

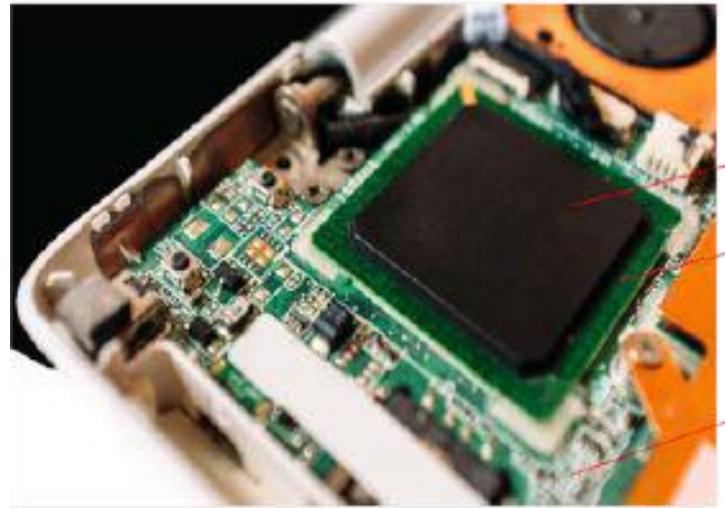


Protecting IC

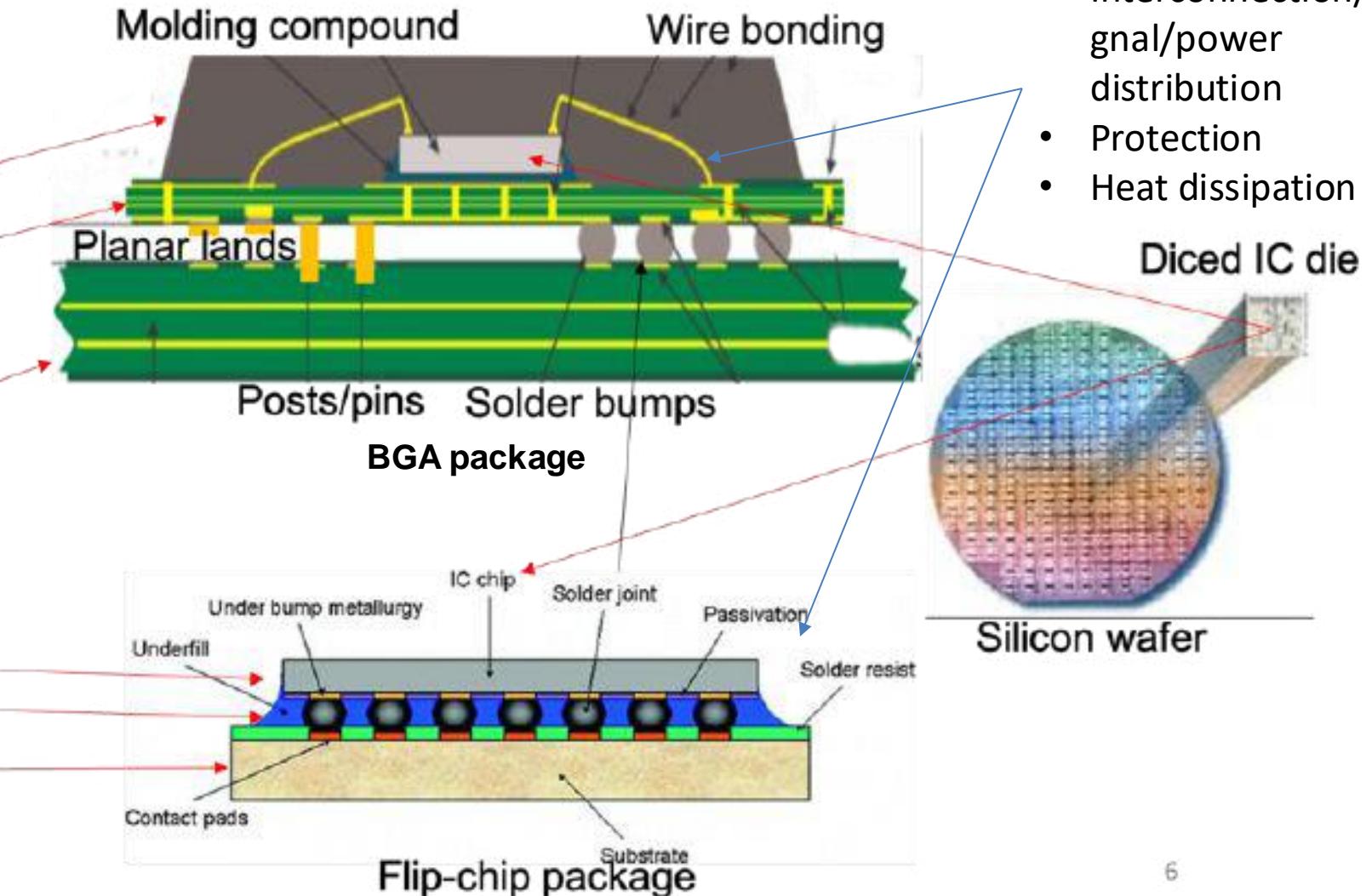


Interconnecting

# IC packaging and components

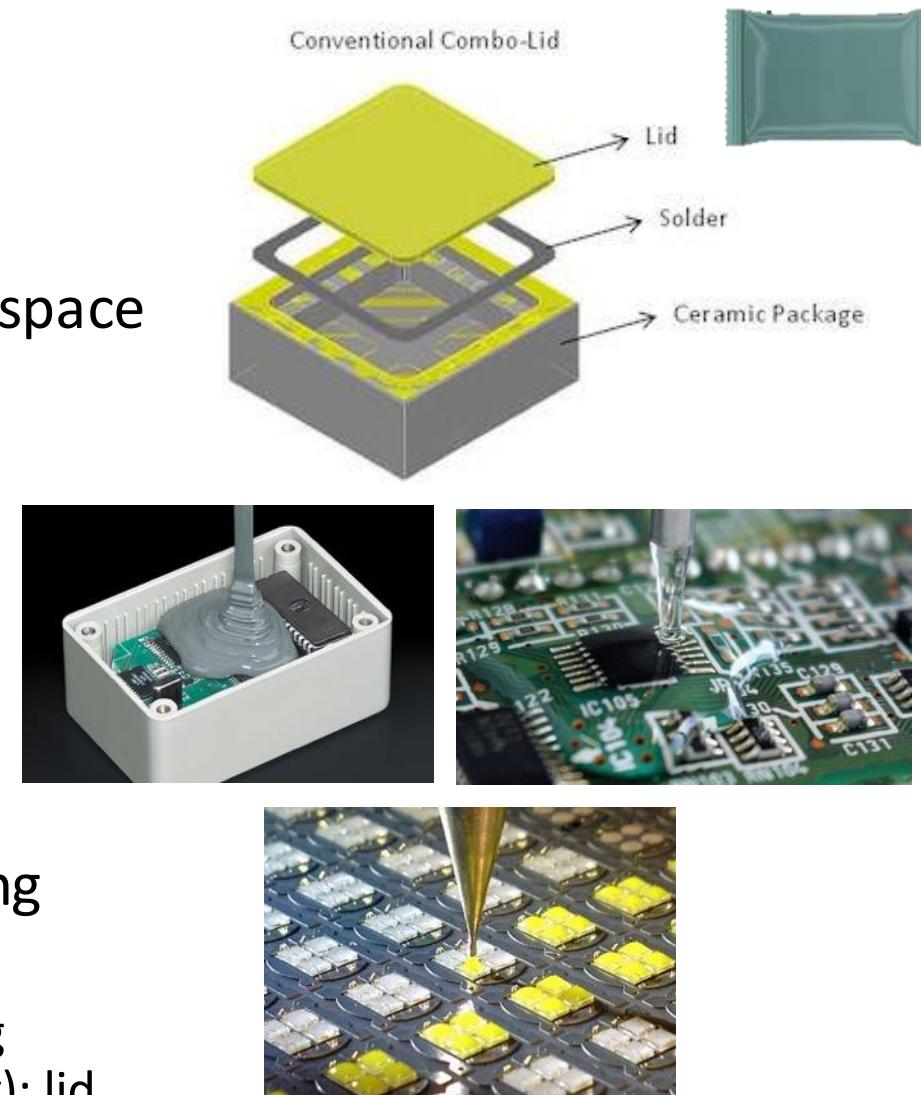


PBGA in a cell phone



# 14. 1 What is Sealing and Encapsulation and Why

- Sealing/hermetic packaging
  - Gas tight or impervious to gas flow
  - Metal, ceramics and glass
  - High reliability; medical, optical, automotive, aerospace applications
- Encapsulation/near hermetic packaging
  - Commercial products/applications
  - Low temperature and low cost
  - Polymers
- Great progress made in polymer-based sealing/encapsulation for some of the most demanding applications (chemical, optical)
  - Not only protection but also providing functionality by using functional materials such as electrical (ESD and conductivity); lid bonding, fuel tank sealants, etc.



SOURCE; Google 9

# 14.2 Anatomy of an Encapsulated and a Sealed Package

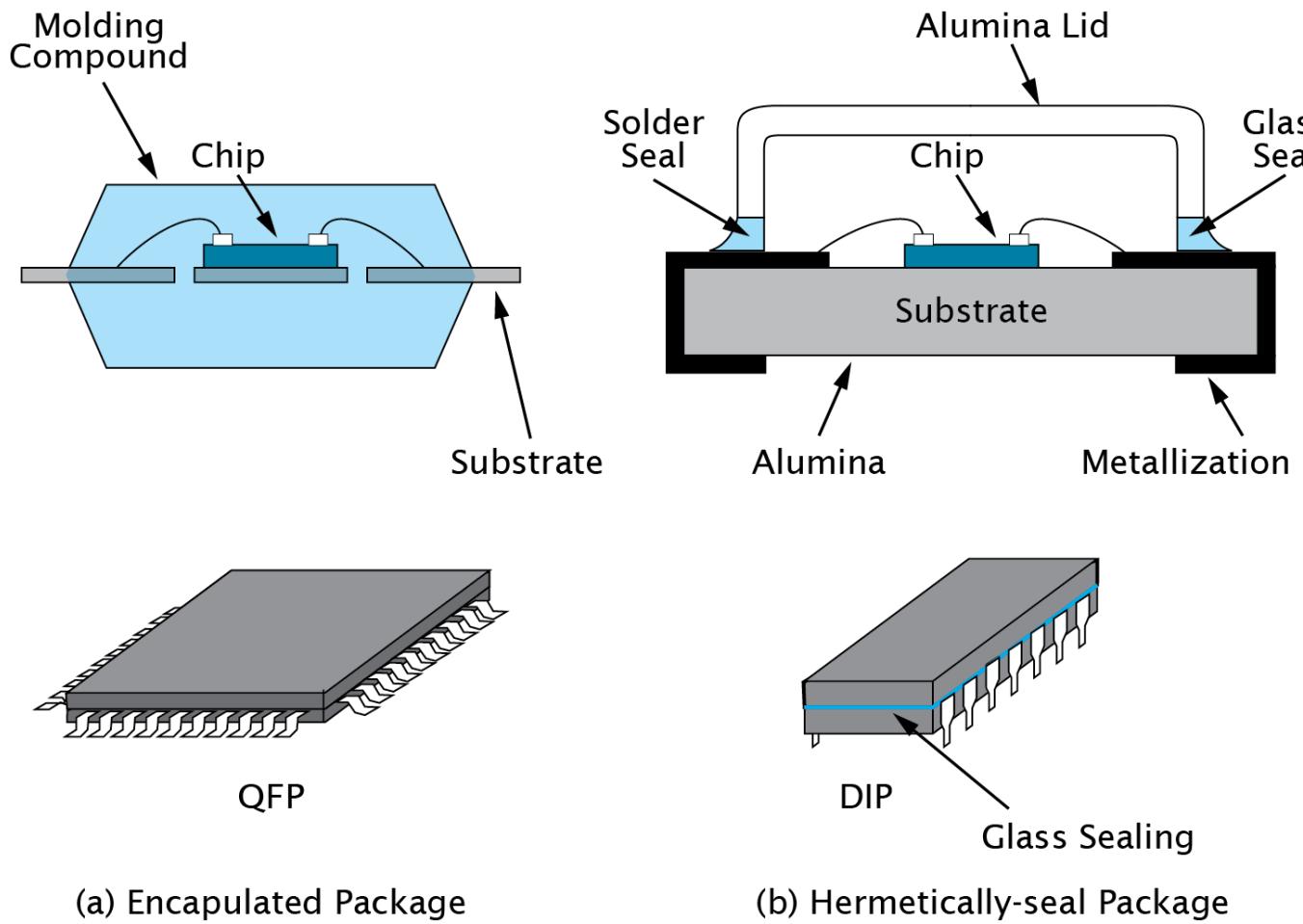
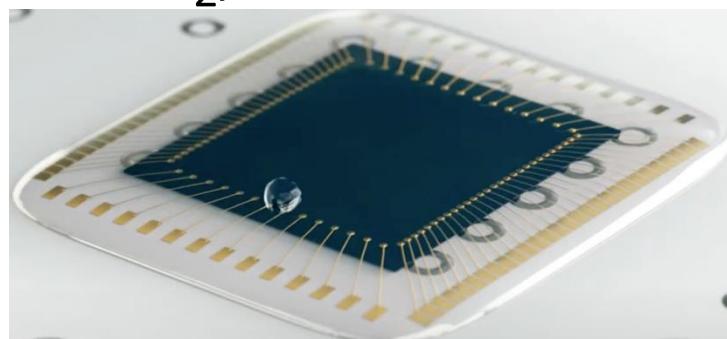


Fig 14.01 - Example of an encapsulated

# Fundamentals of Encapsulation and Sealing



- Chemical Protection
  - Moisture; Primary contributor to failure/reliability, delamination, popcorn cracking, require bake-out prior to board assembly, moisture to cured/uncured resin resulting in lowered Tg, modulus, and adhesion, hygrostress by swelling
  - From Salts; mobile ions, corrosion in fine-pitch,
  - From Biological Organisms; insects, or so
  - Atmospheric contaminants; Corrosive acids-NOx and SO<sub>2</sub>, acid rain
- Mechanical protection
- Functionality



# Protecting ICs Chemically

Environmental pollutants, corrosive ingredients such as salts, biological secretions, mobile ions ( $\text{Na}^+$ ,  $\text{Cl}^-$ , etc.)

Moisture (absorption/desorption)

- Delamination, popcorn cracking
- Reduce  $T_g$  and  $E$  of plastic encapsulants (both the uncured and cured/hydrolysis)
- Deteriorate adhesion of solder balls to epoxy
- Swelling-hygrostress, diffusion controlled, exponential

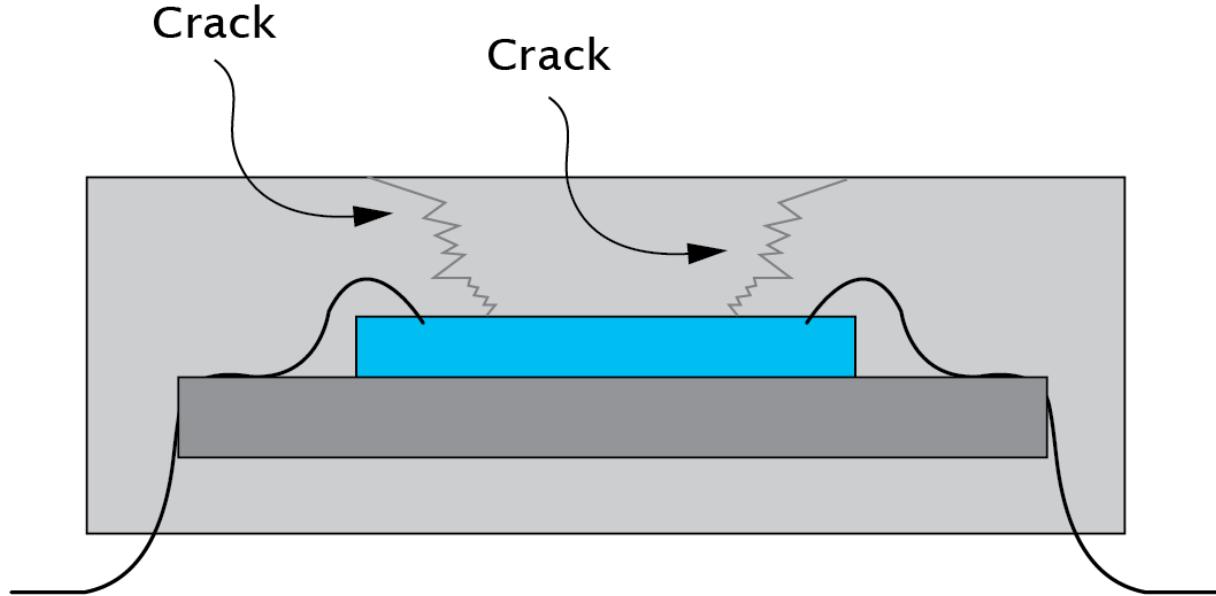
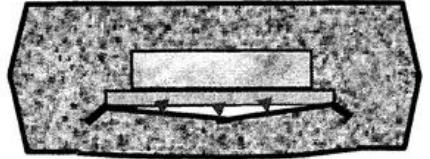


Fig 14.02 - Popcorn effect

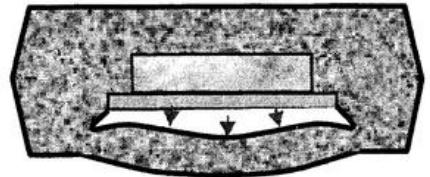
$$\frac{W_t}{W_e} = \frac{4}{L} \sqrt{\left( \frac{Dt}{\pi} \right)} \quad \text{Eq 14.1}$$

$$W_e = KH_\alpha \quad \text{Eq 14.2}$$

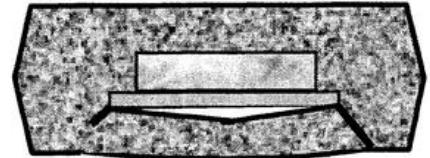
# Popcorn cracking in memory die-EMC



- Delamination
- Crack growth into EMC

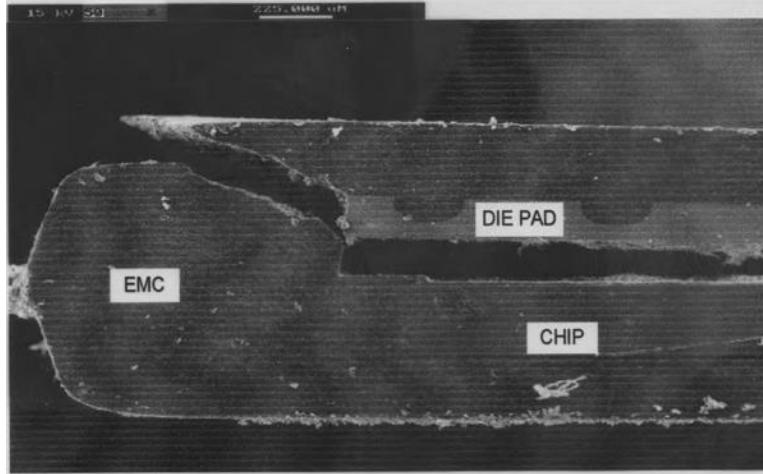


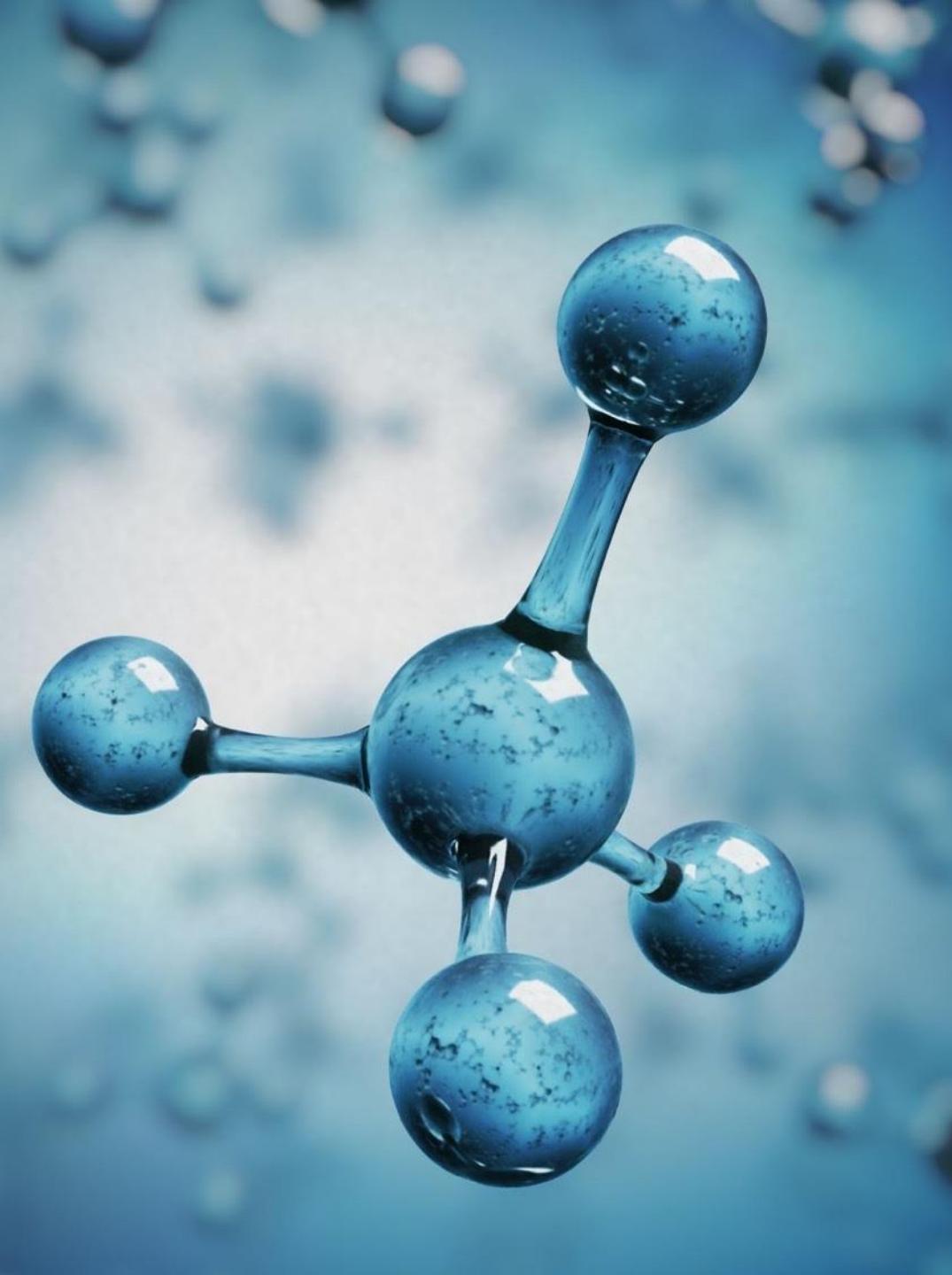
- Plastic deformation inhibiting crack growth
- Vapor pressure increases at elevated temperature



- Critical vapor pressure reaches
- Plastic collapse occurs
- Pressure release to deflate dome

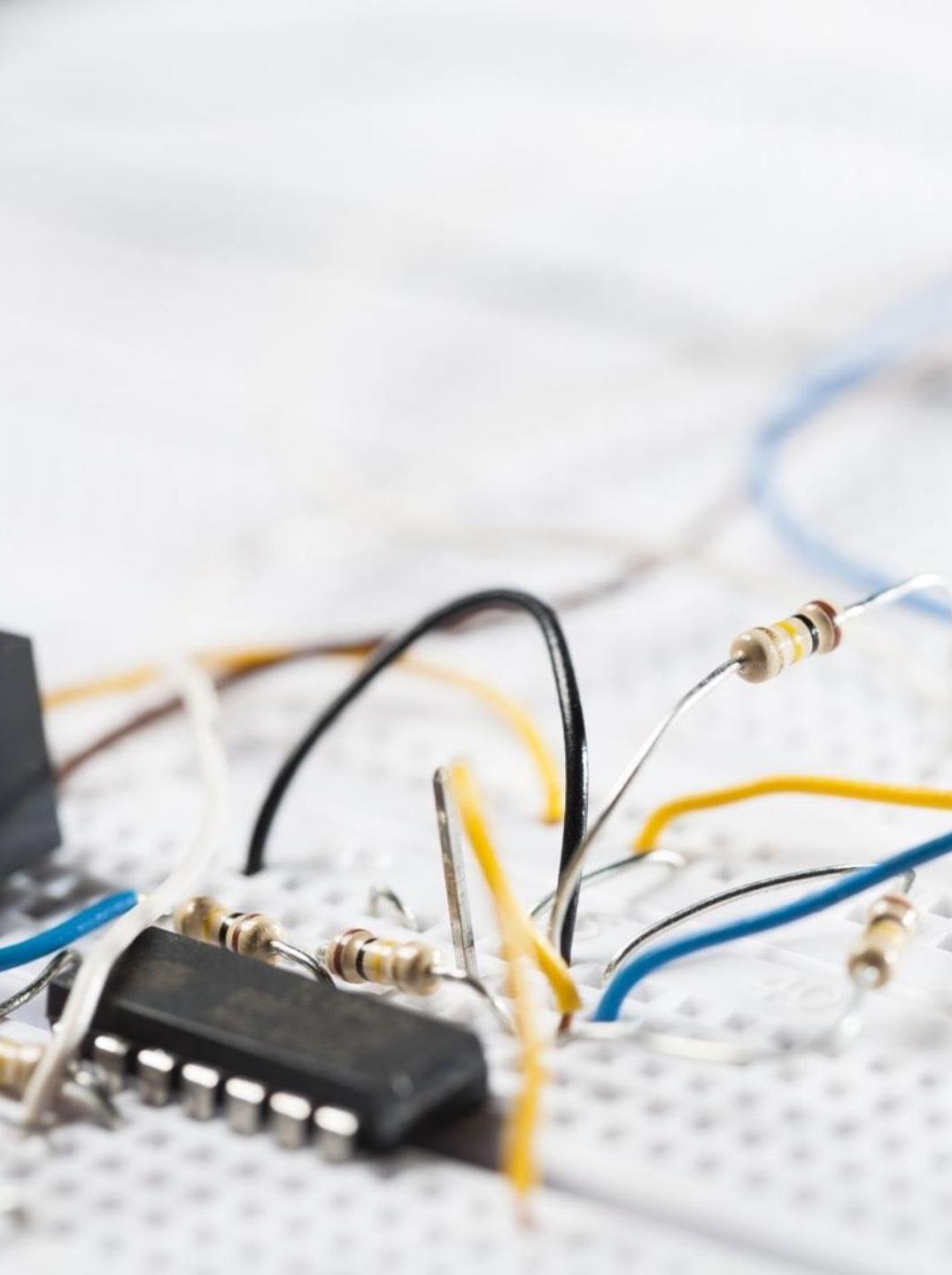
Under standard reliability test (HAST, thermal cycling, thermal shock, etc. with temperature, pressure, bias, or so





## Protecting ICs Chemically (cont'd) Salts, biological organisms and atmospheric contaminants

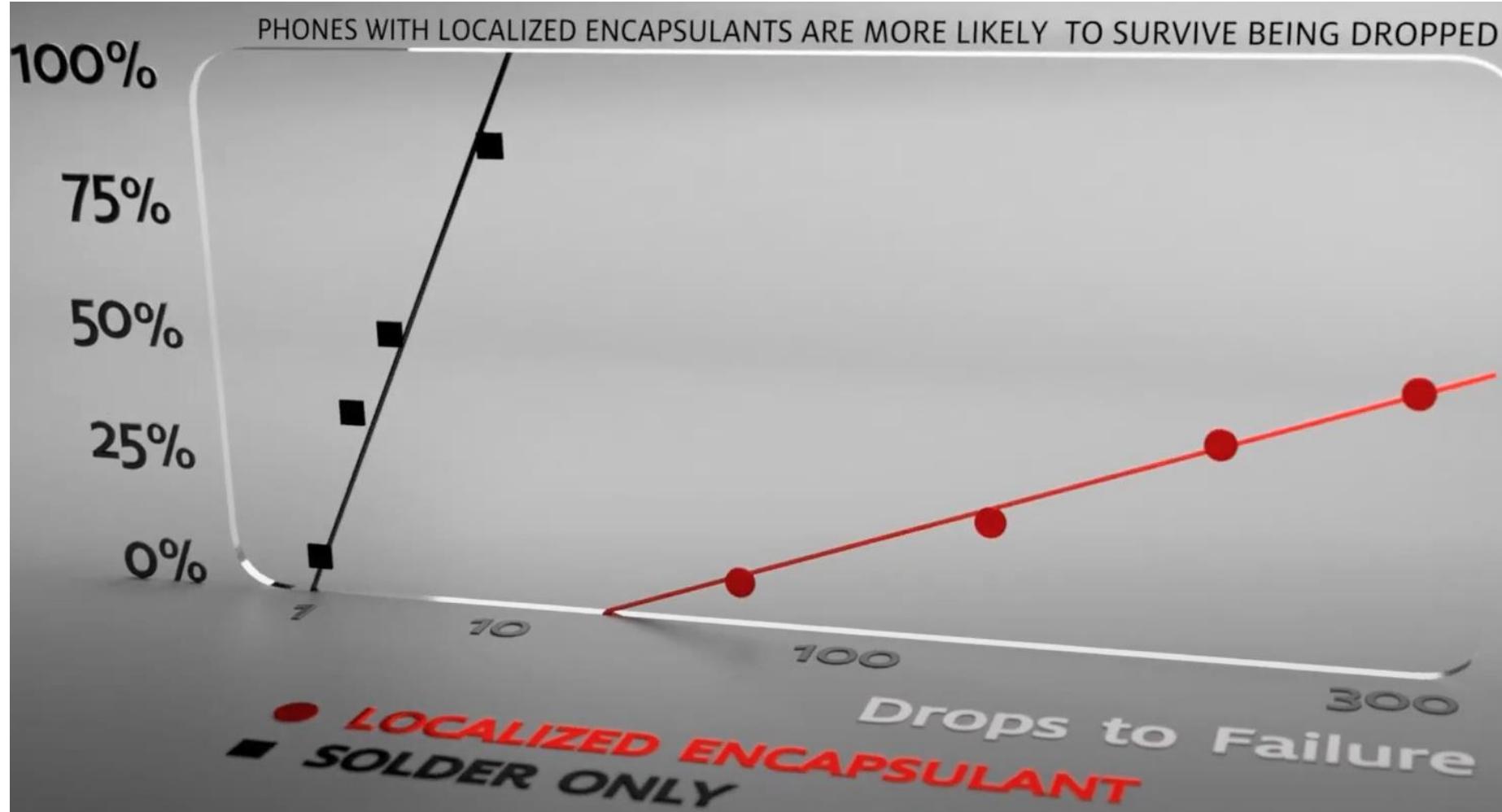
- Ionic contaminants; K<sup>+</sup>, Cl<sup>-</sup>, Na, cause electrolytic corrosion, electrochemical migration, very influential onto submicron sized pitch of electrical transmission lines
- Insects, fire-ants cause damage to some telephone switches
- Corrosive gases; NO<sub>x</sub> and SO<sub>2</sub> from fossil fuel burning to form corrosive acids, brings about acid rains



## Protecting ICs Mechanically/thermally

- From mechanical impact
- Protecting interconnections (wires and bumps) via encapsulating ICs
- Minimizing strain and distributing stress in the solder joints by underfills

# For what Encapsulants are employed



# Hermetic vs Non-hermetic

- Criteria of hermeticity; to prevent Helium's diffusion below  $10 \text{ cm}^3/\text{s}$
- Organic vs inorganic
- Early days Hermetic packages; sealed at IC level by SiN or sealed in metal cases (military and aerospace industries)
- Hermeticity by non-hermetic way; plastic packages (non-hermetic) evolved for acceptable reliability and now account for 90-95% of device packages

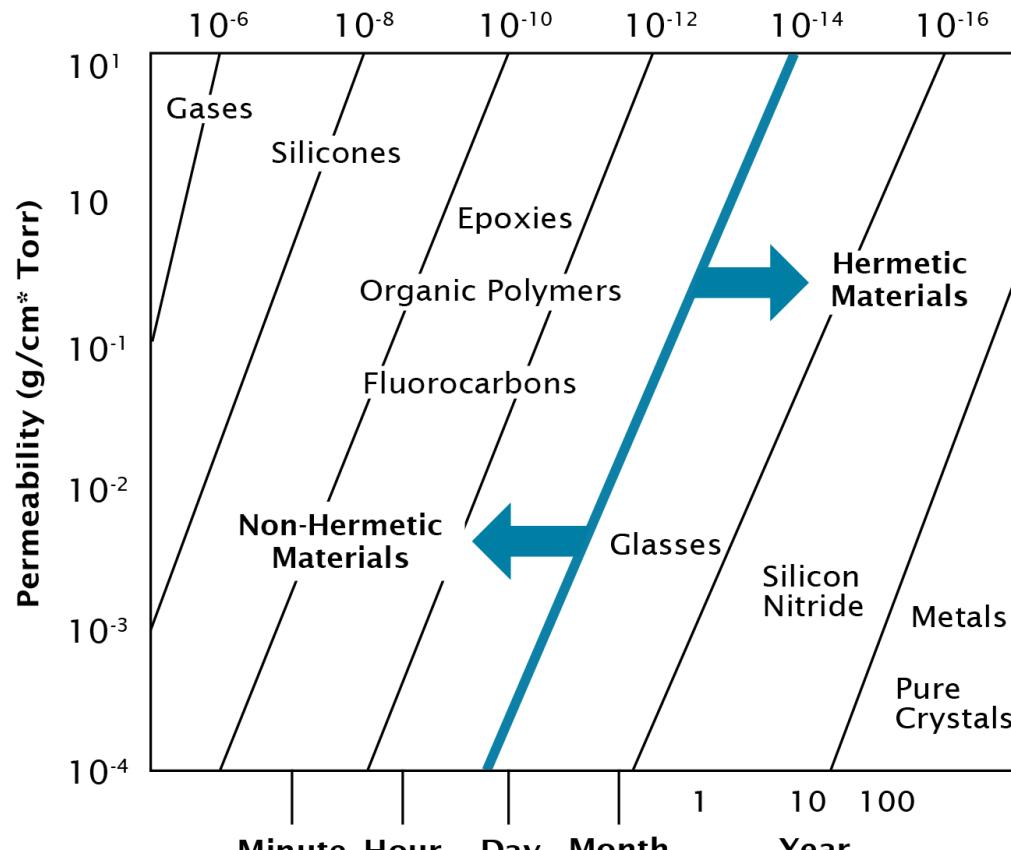


Fig 14.03 - Permeability of water

# Moisture effects on Plastic Packages

Acting as a debonding agent between organic materials and substrates

- To form weak hydrated oxide surface
- To break chemical bonds
- To induce depolymerization or degradation; plasticizing, E and T<sub>g</sub> reduction

Greater issues in organic materials due to the high water permeability

# Trend to Polymer Encapsulants

Early stage of using polymers in electronic package struggled from moisture, interface, stress, etc.

Great progress in polymer chemistry and understanding in filler technologies enabled to achieve enhanced reliability of plastic packaged electronics

Most of modern packages are employing polymers;

- **High purity chemicals, sophisticatedly developed backbone molecules, additives and advanced filler technologies**
  - *Purity → Ionic impurity*
  - *Hydrophobicity → backbone, reduced polarity and free volume*
  - *Filler interfaces and loading techniques*

# Accelerated testing-evaluation

To assess non-hermetic packaging

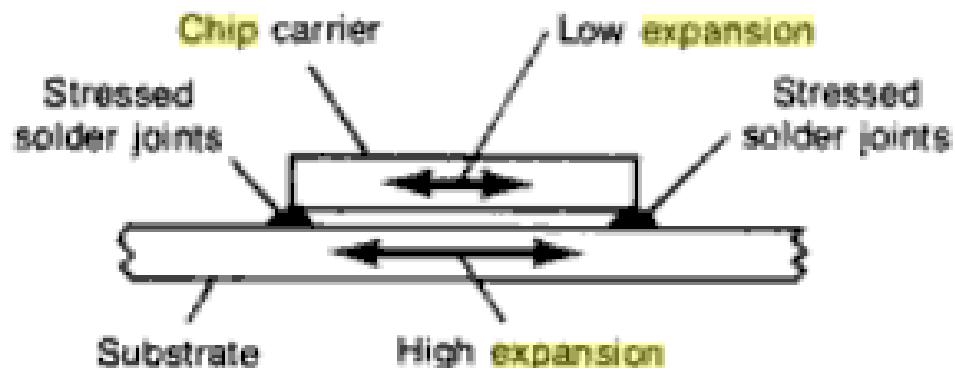
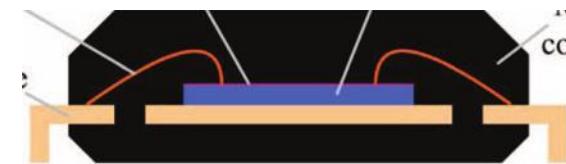
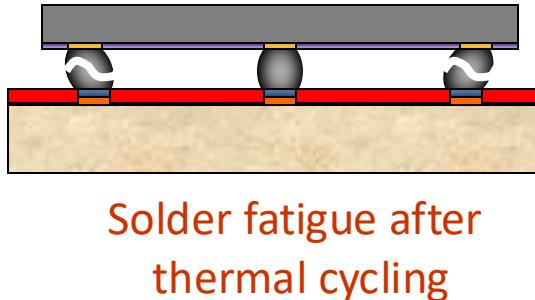
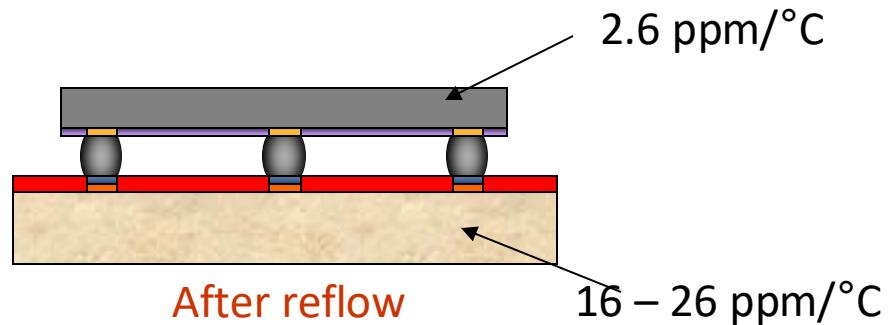
Temperature cycling to induce thermal stress

Correlation between the accelerated tests and real life time is not clearly established yet

## Examples

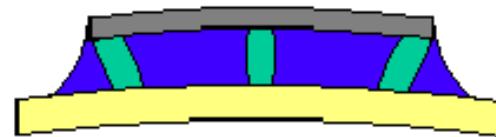
- Thermal shock: test method:  $-65^{\circ}\text{C}$  to  $150^{\circ}\text{C}$ ; cycle time: 10 s; dwell time: 5 minutes at each extreme 1000 cycles.
- Salt-spraying time: 24, 48, 96, 240 hours; salt concentration: 0.5–3 percent (NaCl) pH 6.0–7.5,  $95^{\circ}\text{F}$ ; deposition rate: 10,000–50,000 mg/m for 24 hours at  $35^{\circ}\text{C}$ .
- Autoclave:  $121^{\circ}\text{C}$ , 100 percent relative humidity (RH), 30 psi (2 atm), with or without bias (pressure pot). Thermal shock

# Thermal Stress in Packages



$$\Delta\gamma = \frac{D_{NP} \Delta(\alpha\Delta T)}{2h}$$

where  $D_{NP}$  = largest diagonal distance between solder joints,  
and  $h$  = height



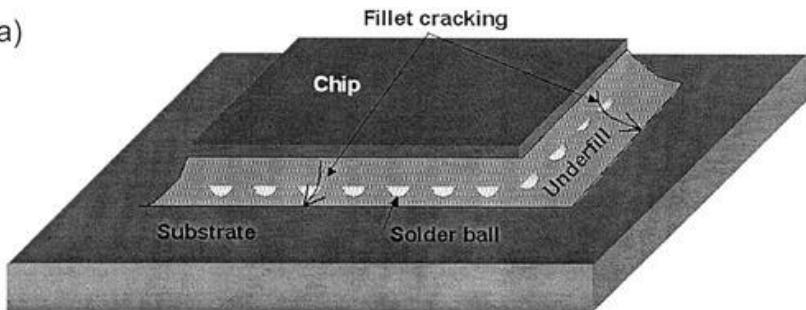
$$S = E \times \epsilon = E \times \alpha \times dT$$

Stress/Modulus/Strain/CTE/Temp

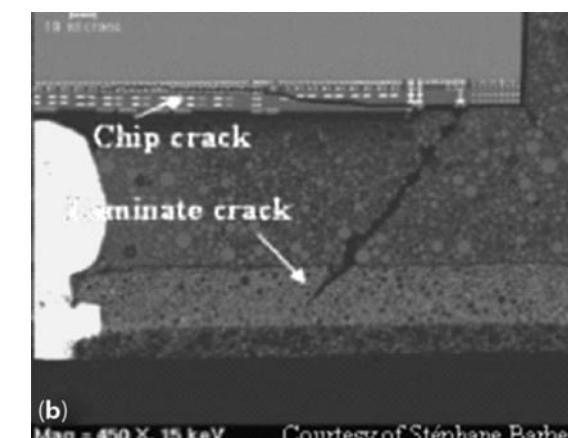
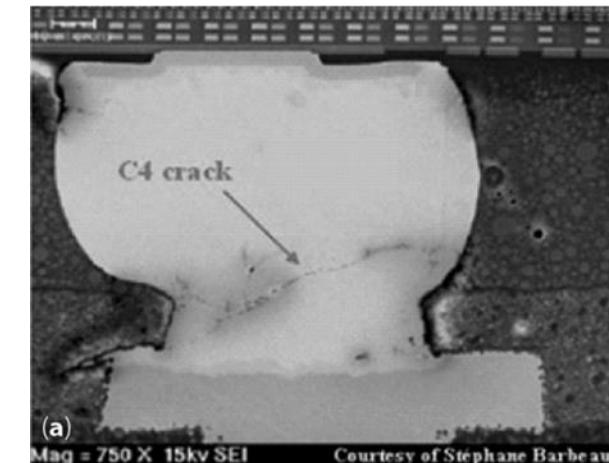
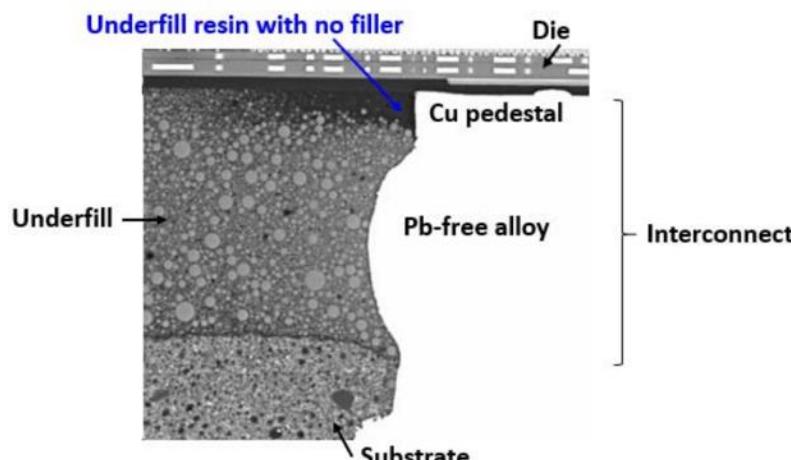
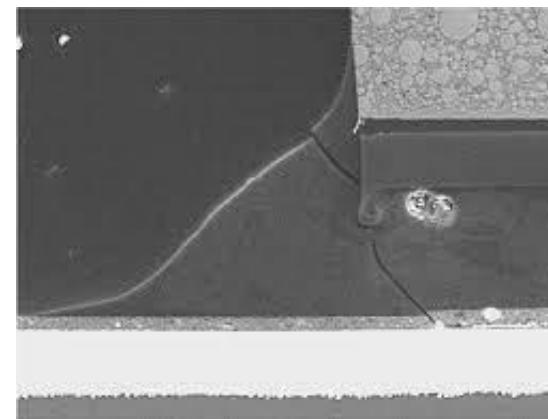
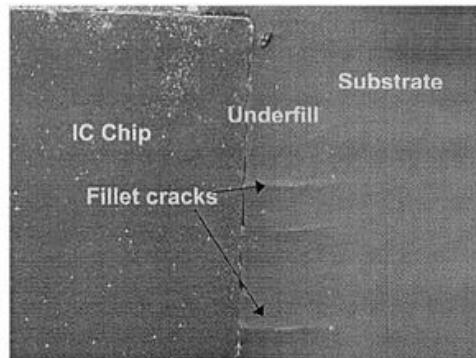


# Stressed induced cracking/delamination

(a)

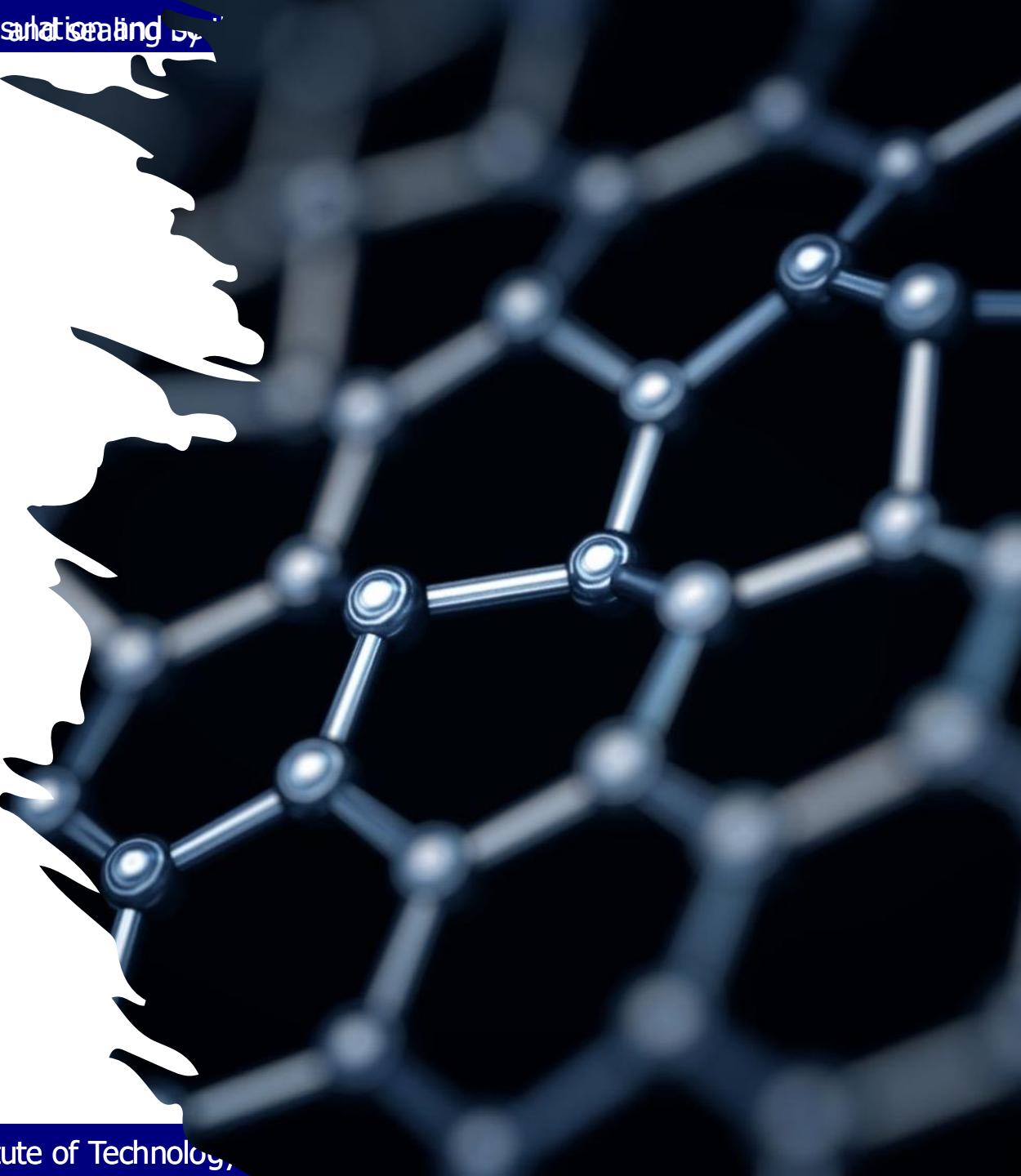


(b)



# 14.3 Properties of Encapsulants

- Thermo-mechanical properties
  - CTE
  - T<sub>g</sub>
  - Young's modulus, storage and loss modulus
  - Elongation
- Chemical properties
- Physical properties
  - Flow property-Rheology
  - Adhesion



# Material requirement for underfill

Properties	Desirable Values	Comments
Flow	>0.5 mm/s	Fast flow with no air bubbles entrapment
Adhesion	>50 MPa shear force	Key to device protection
CTE	18–30 ppm/°C	Matches CTE of solder (26 ppm/°C)
Elongation	>1 percent	Resists CTE mismatch stress
Modulus	5–8 GPa	Provides mechanical coupling
T <sub>g</sub>	>130°C	Maintains dimensional stability
Stress after cure	<10 MPa	Minimizes internal stress caused by shrinkage of polymer
Water pickup	<1 percent	Reduces moisture-induced failures
Ionic impurities (Na <sup>+</sup> , K <sup>+</sup> , Cl <sup>-</sup> , Br <sup>-</sup> )	<10 ppm	Prevents corrosion and metal electromigration
Thermal stability, 1 percent weight loss	>260°C	Prevents underfill decomposition during solder reflow
Curing time at 160°C	<0.5 hr	Maintains good product output
Volatility during cure	<1 percent weight loss	Maintains correct stoichiometry
Pot life at RT, 20 percent increase in viscosity	>8 hr	Provides long usable underfill life

**Table 1.** Comparison of properties of typical materials used for MUF and CUF processes.

Property	Units	MUF	CUF
Filler Wt Content	%	85	65
Nom. Filler Size	um	6.1	0.6
Max Filler Size	um	25	3
Viscosity	Pa-S	9.5	50
T <sub>g</sub>	C	153	117
CTE Alpha 1	ppm/c	12	31
CTE Alpha 2	ppm/c	45.25	90
Flex Modulus (< T <sub>g</sub> )	Gpa	18.875	11
Flex Modulus (> T <sub>g</sub> )	Gpa	0.4	0.15

<https://shorturl.at/biqr7>

- Table 14.1 Flip-Chip Underfill Material Requirements

# Stress-strain curves for underfills

- Elastic failure
- Viscoelastic failure
- Toughness
  - Energy to absorb impact
  - Stress intensity factor/Stress release energy

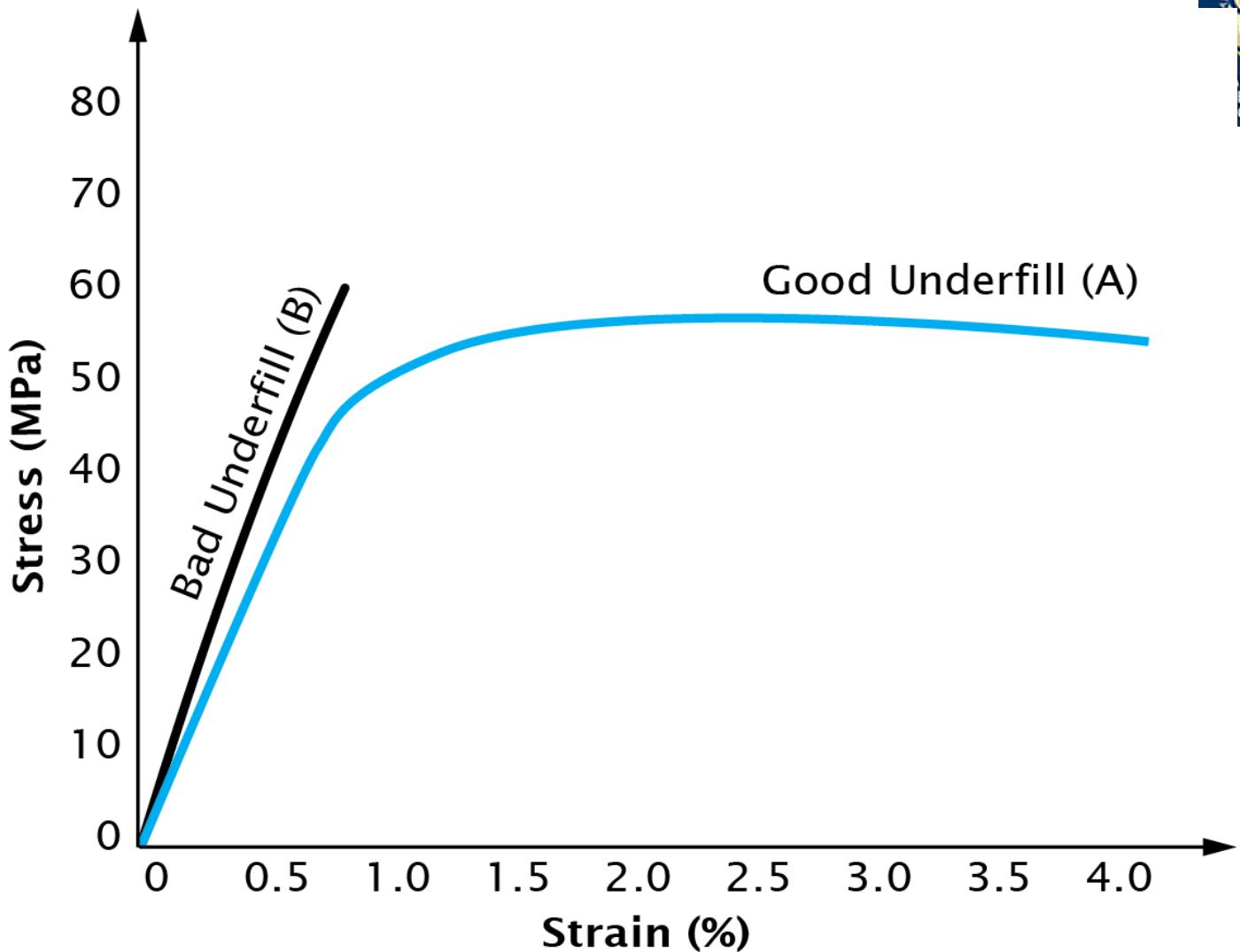


Fig 14.04 - Stress Strain Curves

**Table 14.2** Physical Properties of Some Materials of Interest

Material	CTE (ppm/°C)	Modulus (GPa)	Density (g/cc)
Silicon	2.6	107	2.33
Silicon dioxide	0.5	119	2.60
Alumina	6.6	345	3.90
Solder (63Sn/37Pb)	25	50	8.40
Aluminum	23	79	2.90
Molding compound	15	14.2	2.30
FR-4	16	20	1.85

# Residual stress

- Major sources of residual stress
  - Initial residual stress during encapsulation (molding/sealing) due to polymerization of polymers
  - Polymerization (curing for thermoset polymers such as epoxy) causes Volume shrinkage → 1~2% (linear dimensions) and 3-6% by volume → ~20MPa internal stress
  - Once cured, thermal stress brings about reliability issue in packages due to Mismatch in coefficient of thermal expansion (CTE) between encapsulants and mating/adherent substances such as Si dies and substrates

$$\sigma = k \int_{25}^{T_g} E(a_e - a_s) dT$$

E is the elastic modulus of the epoxy.

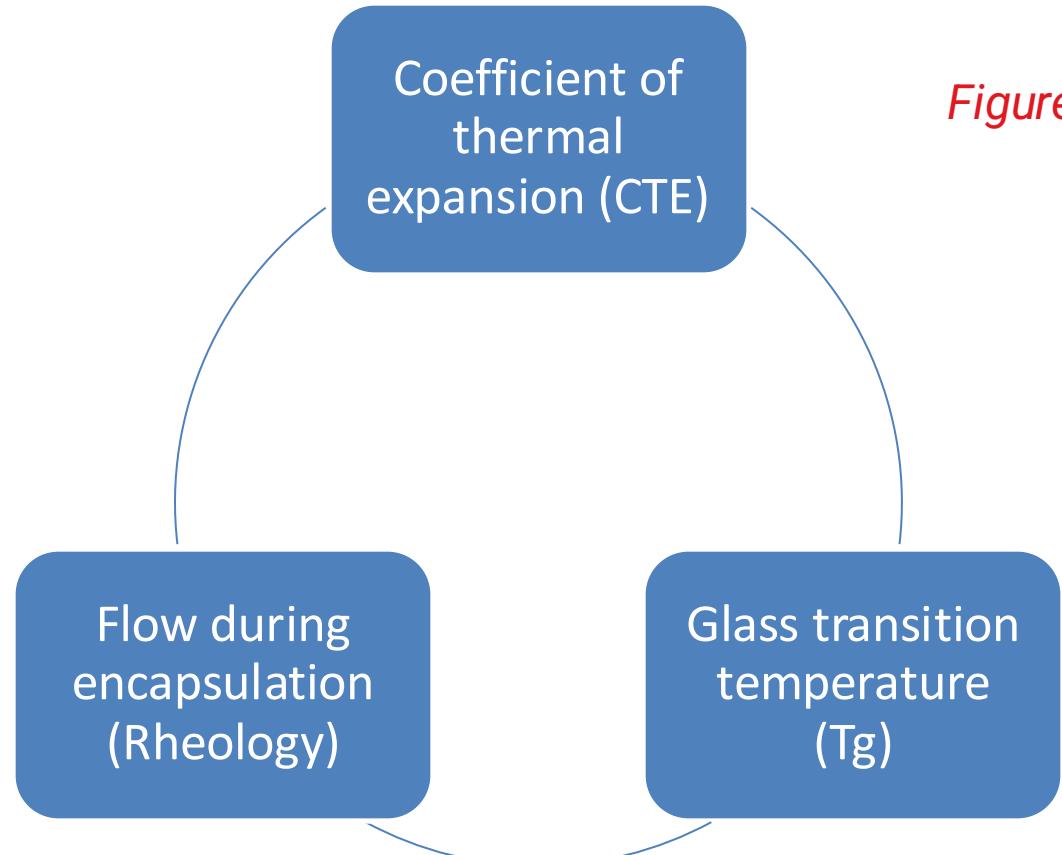
k is a constant.

$(\alpha_e - \alpha_s)$  is the difference in CTEs between the epoxy and the substrate.

dT is the change in temperature between T of the epoxy and room temperature.

EQ. 14.3

# Thermomechanical properties



*Figure 14.5 Glass transition temperature of polymers.*

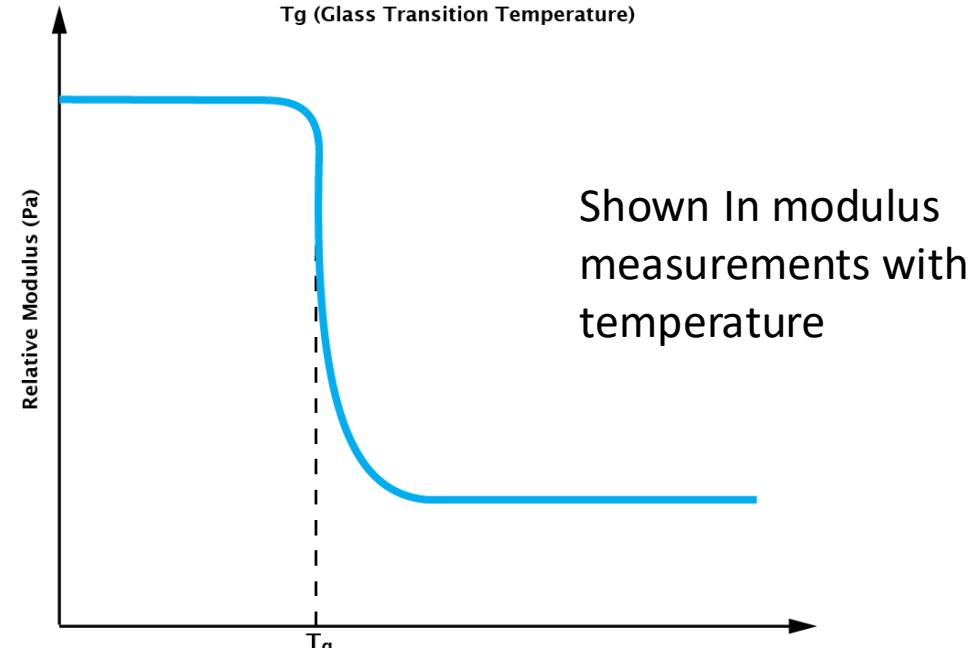
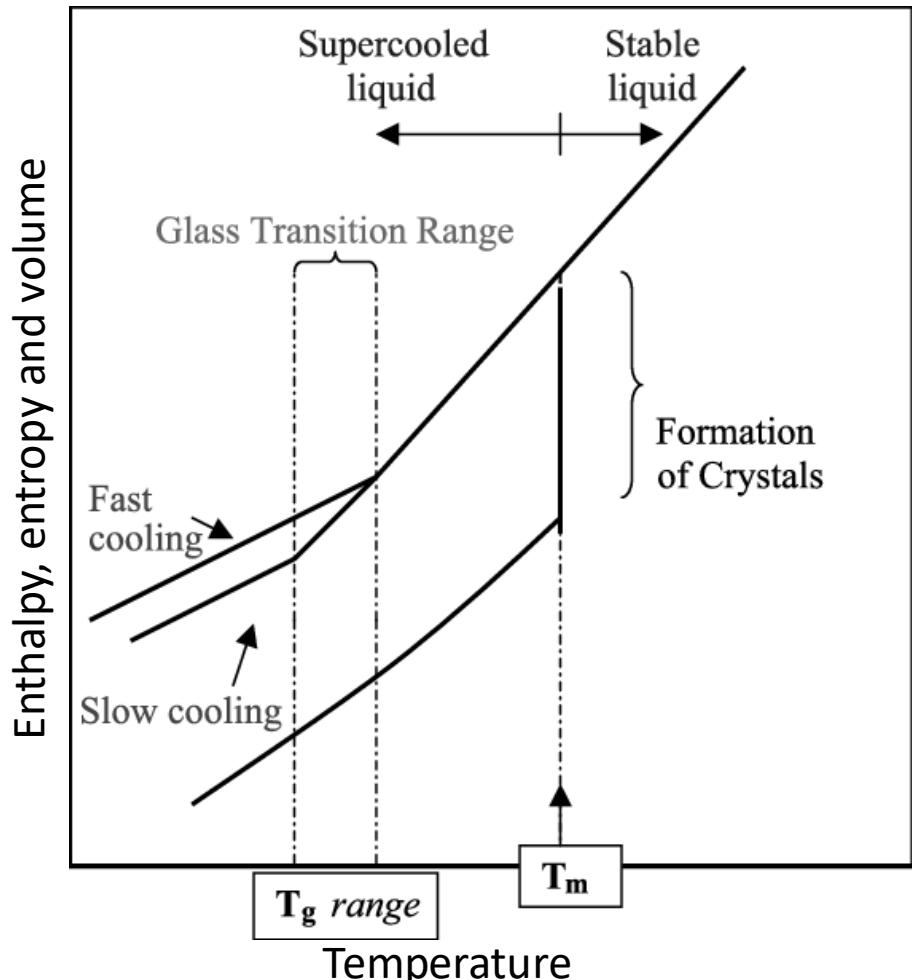


Fig 14.05 - Glass transition temperature

Low viscosity, high surface tension,  
small wetting angle

# Phase transition



- Crystalline materials
  - Metals/ceramics/crystalline polymers)
  - $T_m/T_f$ , latent heat, 1<sup>st</sup> order
- Glassy materials
  - Organic, amorphous alloy
  - Continuous transition/2<sup>nd</sup> order
- Mixed
  - Crystalline and amorphous
  - Some polymers

# Characterization of Thermomechanical Properties

## Uncured resin

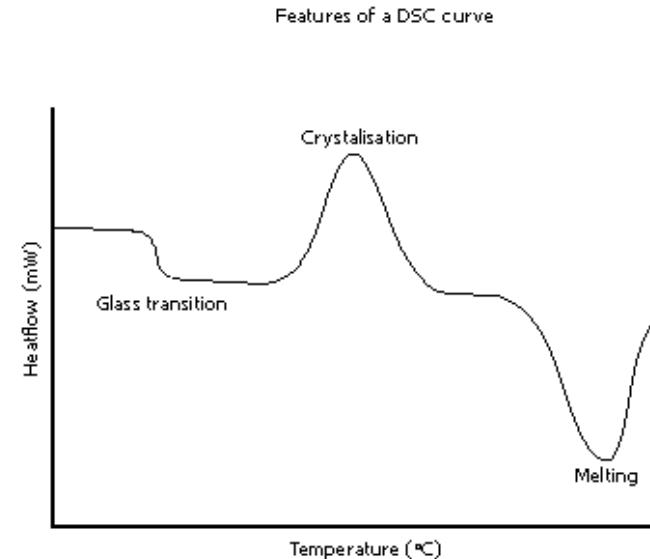
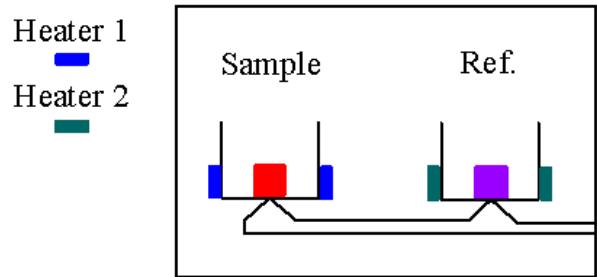
- Cure time and temperature
  - DSC (differential scanning calorimeter)
  - Flowability (rheometer/viscometry)

## Cured polymer

- Coefficient of thermal expansion (CTE)
  - TMA (thermomechanical analyzer)
- Glass transition temperature (Tg)
  - DSC, TMA, DMA (dynamic mechanical analyzer)
  - 2<sup>nd</sup> order transition (without involving latent heat)
  - Heat capacity change
- Elastic modulus
  - DMA (storage/loss moduli)
  - Tensile/bending test
- Strain at break (tensile/bending test)

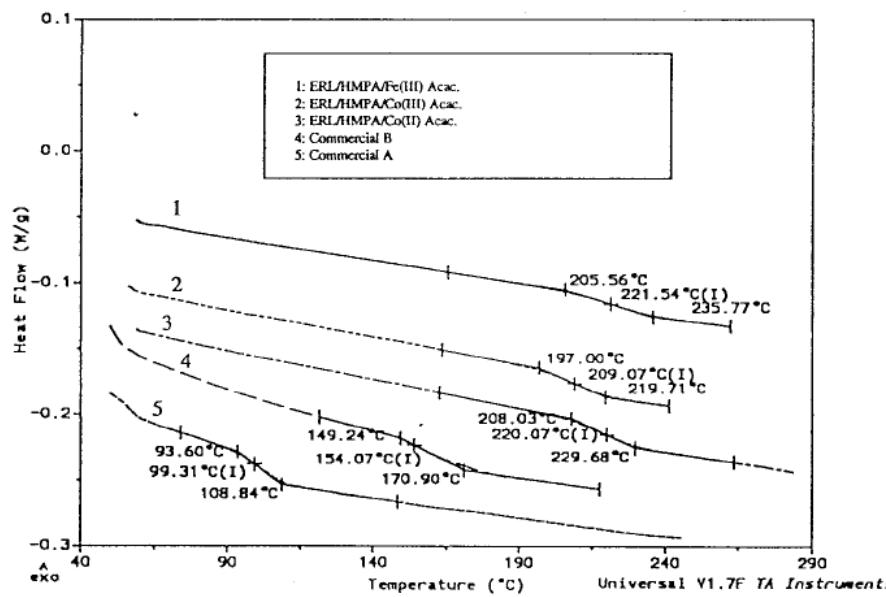
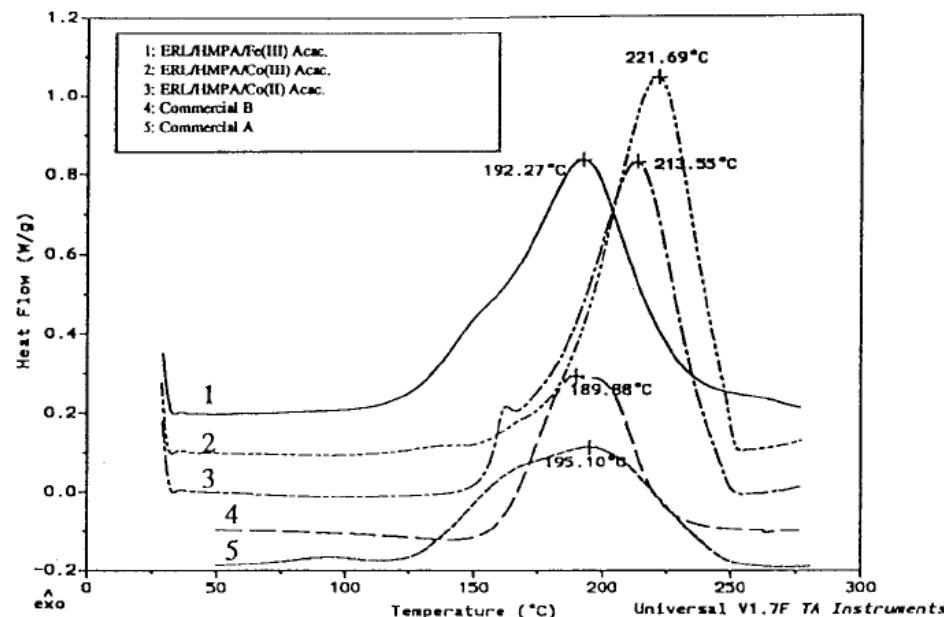
# DSC (Differential Scanning Calorimeter)

- Difference in the amount of heat required to increase the temperature of a sample and reference is measured as a function of temperature
- To determine exothermic and endothermic behavior



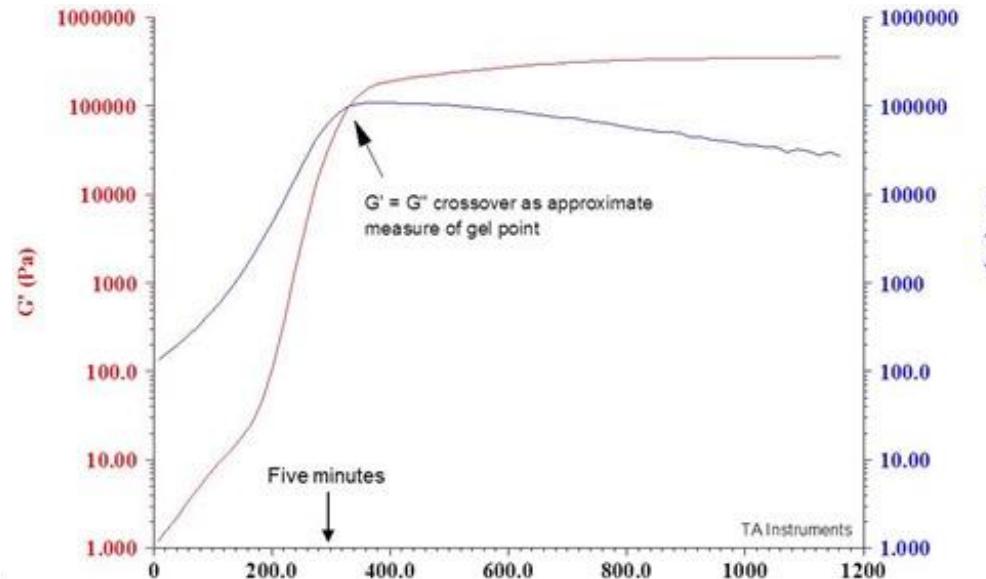
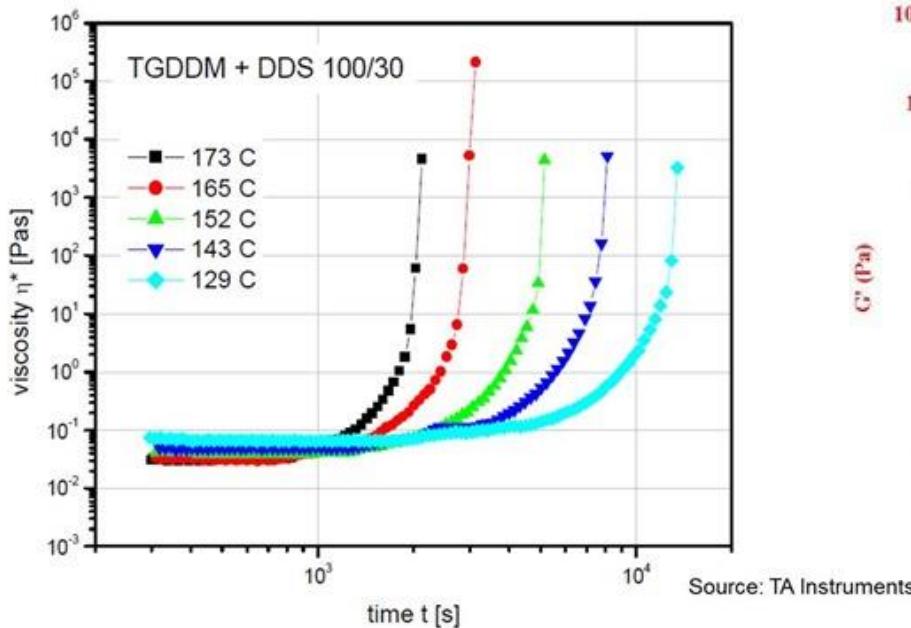
# DSC profiles

- To characterize curing, melting, crystallization, specific heat
- Isothermal/dynamic scan-->cure kinetics
- Glass transition temperature



# Rheometer/viscometer/spiral flow

- Time @minimum viscosity
- Gel time



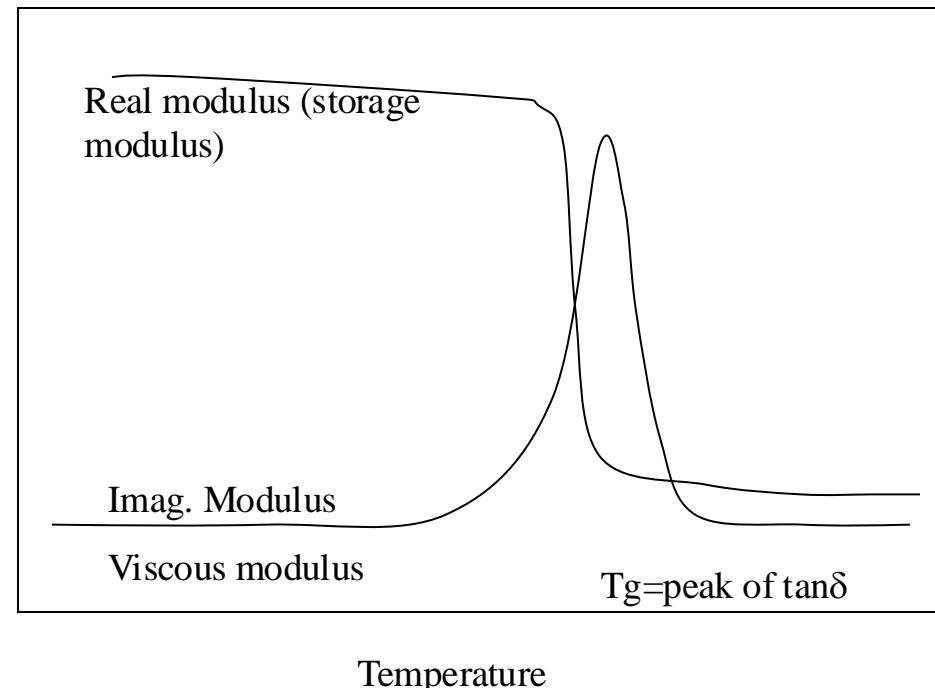
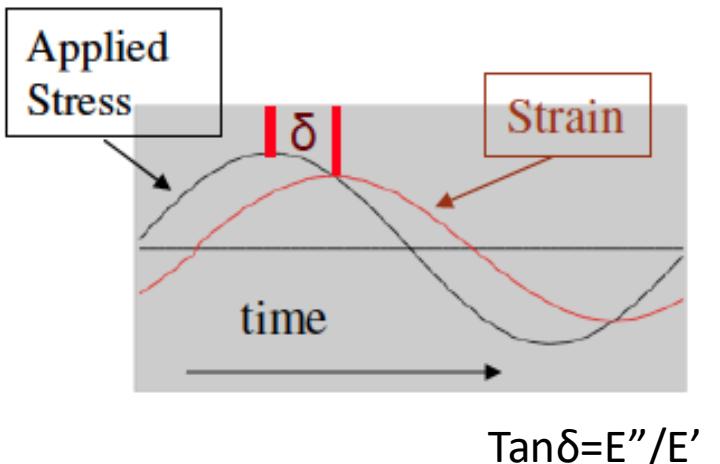
Source; H. H. Winter et. al., Journal of Rheology, v. 31, p. 683-697, 1987

33

# DMA

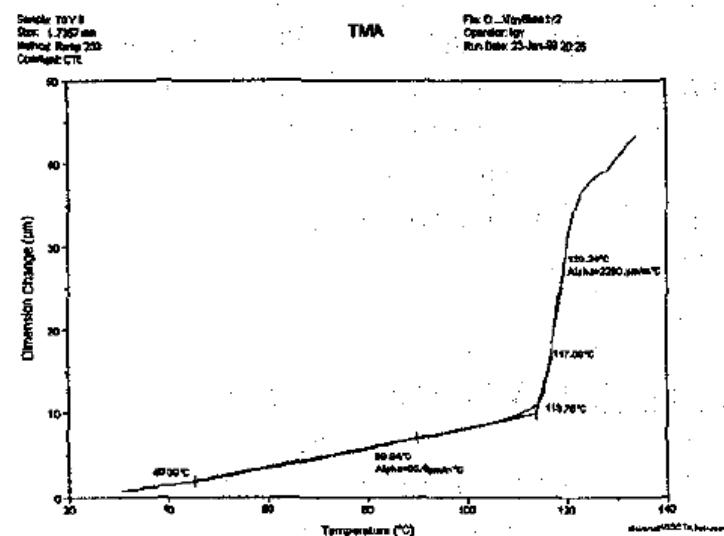
## (Dynamic Mechanical Analyzer)

- For viscoelastic behavior of polymers
- Sinusoidal stress applied and strain in the material measured, to determine the complex modulus (storage/loss modulus)
- $E = E' + iE''$



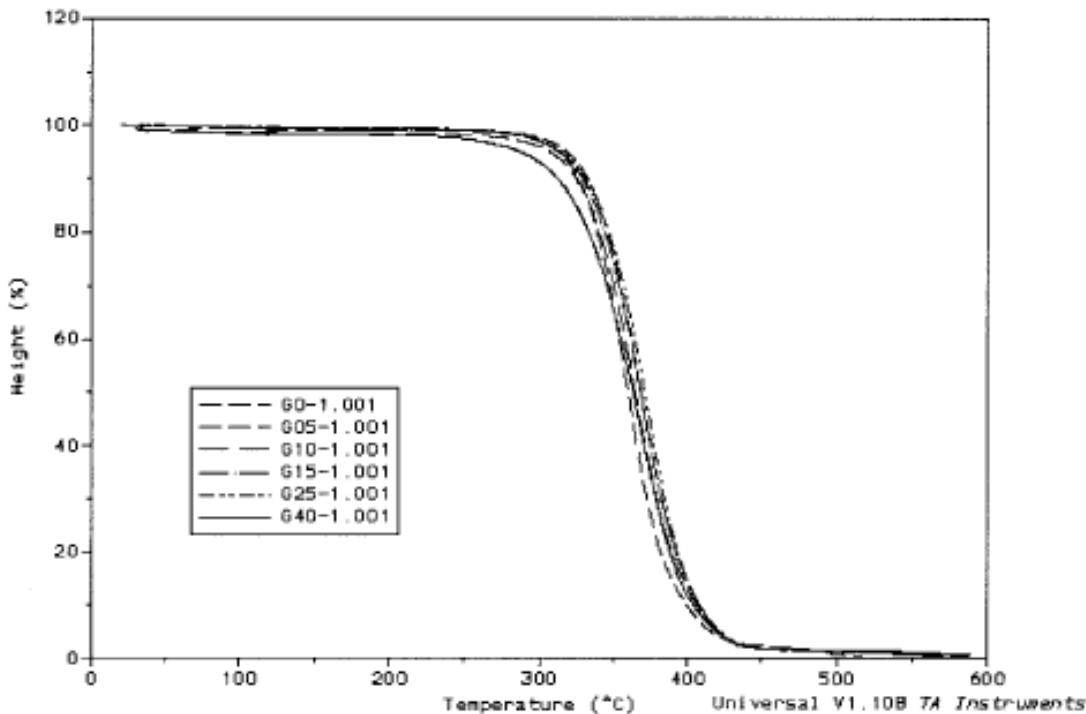
# TMA/Dilatometry (Thermomechanical Analyzer)

- Measure dimension change with temperature
- Coefficient of Thermal expansion (CTE)



# TGA (Thermogravimetry)

- Weight change with temperature
- Residue



# Physical Properties

## Adhesion

- Failures; damage → microcracking → debonding growth → interfacial delamination
- Bonding mechanisms
  - Primary/Covalent
  - Secondary/Van der Waals and Hydrogen bonding
- Adhesion promotion
  - Chemical/physical methods; interfacial bonding/crack suppressors (toughening)

## Interfaces

- Hard/soft passivations (SiO<sub>2</sub>, SiN, SiON or PI ,BCB)
- Solder bumps, wires
- Solder mask
- Conductor leads,
- Substrates (BT, FR-4, ceramics)

## 14.4. Encapsulation Materials

- Epoxy molding compounds (EMC)-leadframe packages, over-molding, MUF
- Underfill material; flip-chip types
- Gloptop; SMT top, LED top (optical function)
- Anisotropically conductive film/paste (ACF/P)
- Non conductive film/paste (NCP)

# Example of underfill composition

**Table 14.3 Example of a Flip-Chip Underfill Composition**

Ingredient	Weight %	Functionality
Bisphenol A diepoxy	5.8	Resin
Cycloaliphatic epoxy ERL4221	12.5	Diluent, cross-linker
HMPA anhydride	13.8	Cross-linker, hardener
2-ethyl-4-methyl imidazole	0.3	Curing accelerator
Pigment (C-black)	0.1	Color coding
Spherical silica filler	67.5	CTE reducer

# Material properties required for encapsulants

Low water uptake  
Low stress  
High strength  
High adhesion  
High toughness  
High flowability

- Base resin
- Filler engineering (loading level, surface treatment, functional property)
- Additives

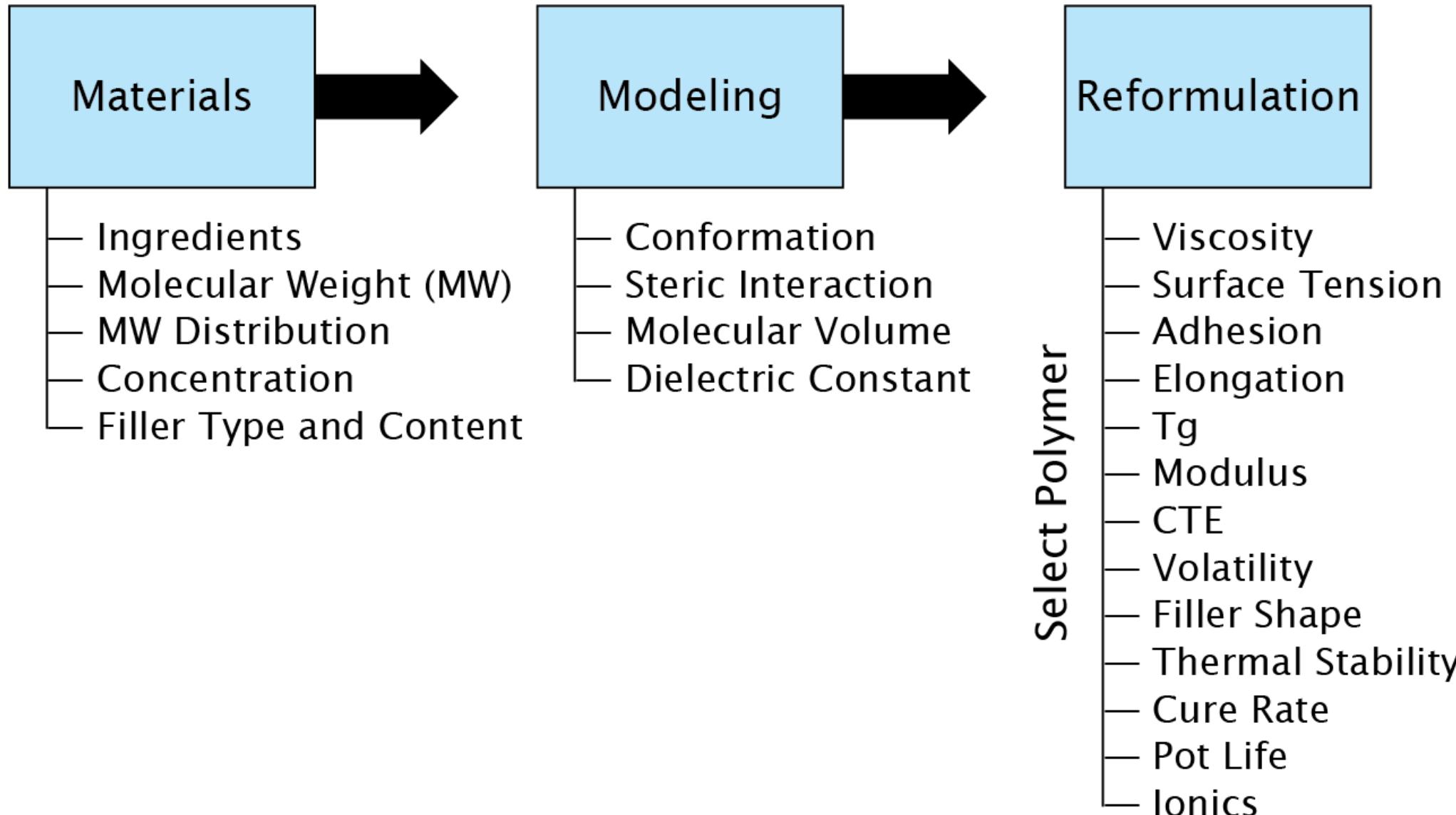
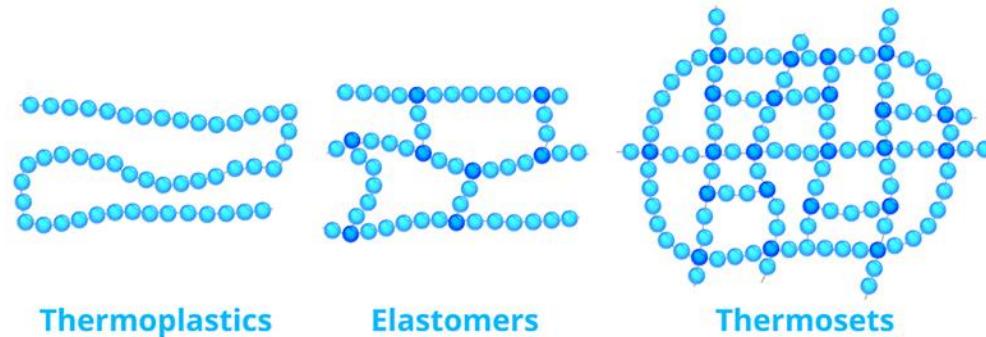
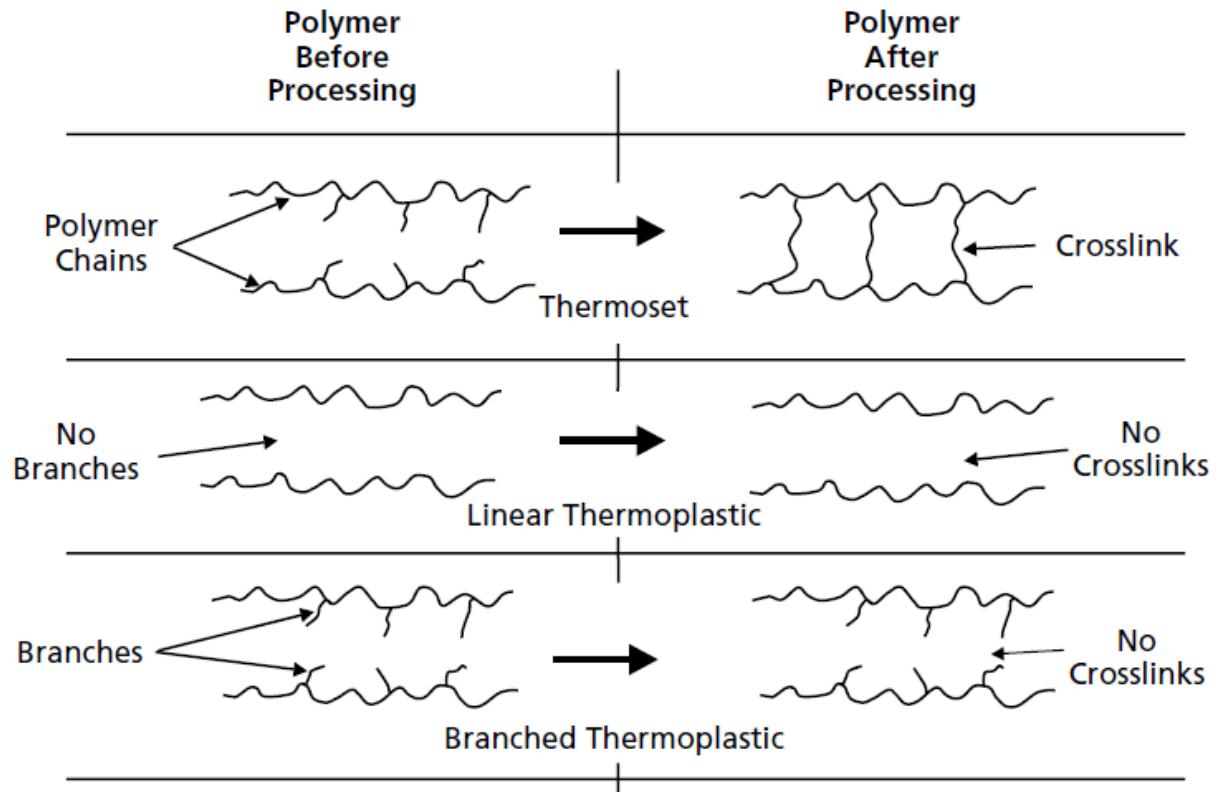


Fig 14.06 - Encapsulation formulation optimization

# Epoxy and related materials

- Epoxy
- Cyanate ester
- Urethane
- Silicone

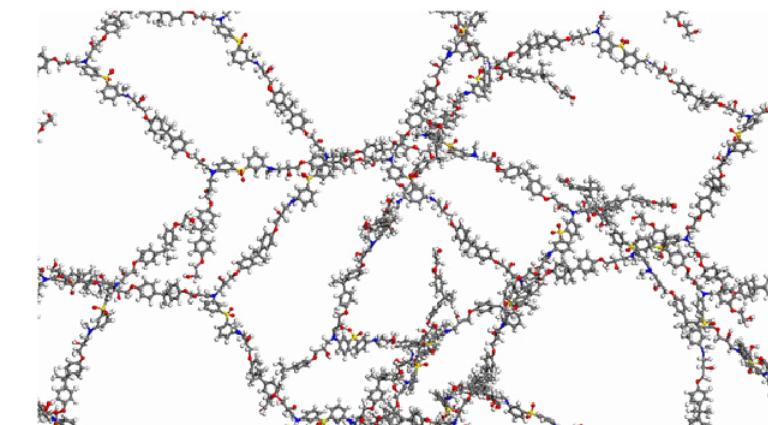
# Polymer types



Source; Bufo, Rodolfo Morales Ibarra's thesis

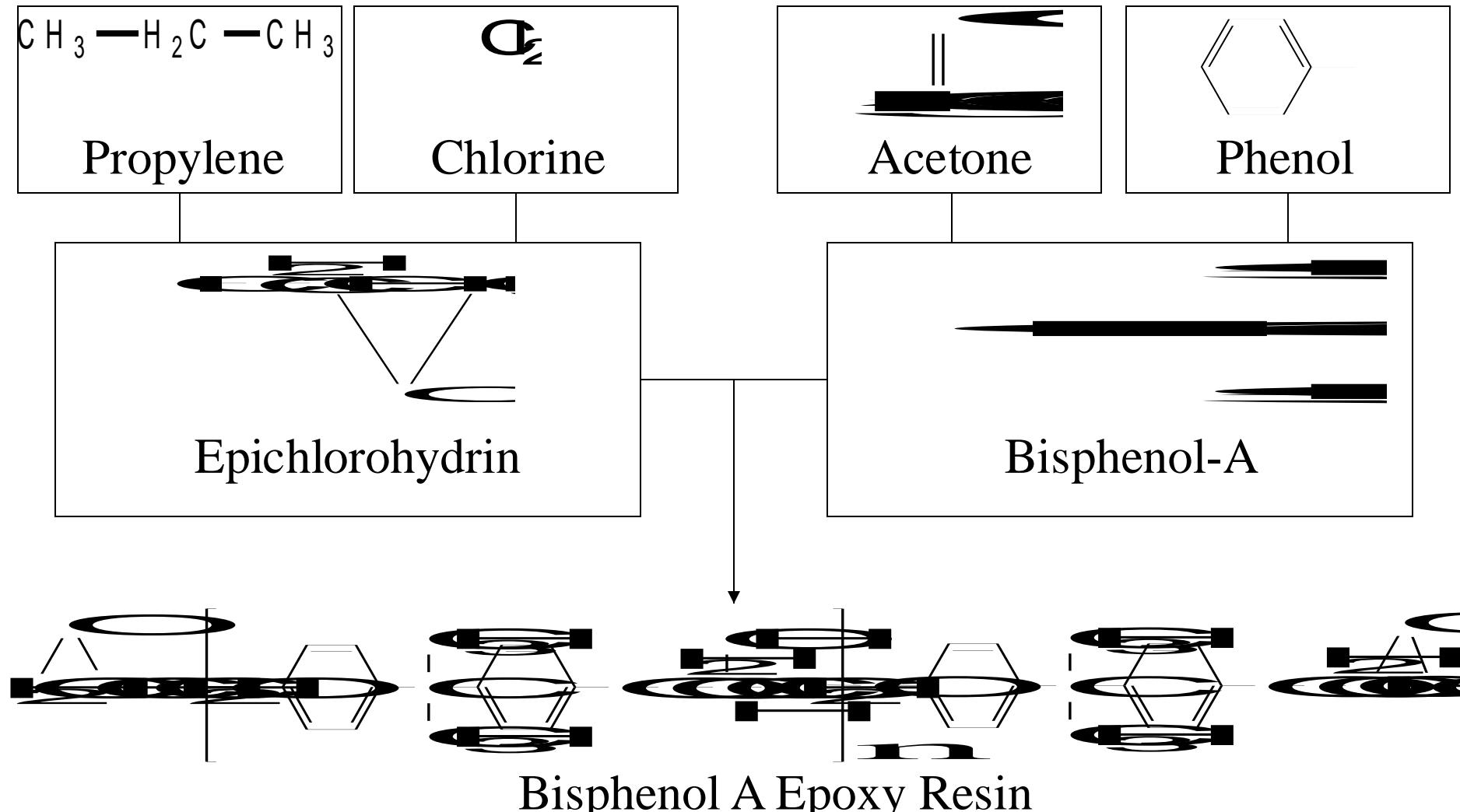
# Epoxy

- Containing epoxide ring group
- Various curing agents to open the ring and crosslink → curing
  - Nitrogen containing; Aliphatic/cycloaliphatic/aromatic amines
  - Oxygen containing; Anhydride, carboxylic, phenol
  - Sulfur containing; p-sulfide/mercaptans
  - Catalysts
- # of functional groups
- Oligomers/Branched molecules
- B-stage → 3D network



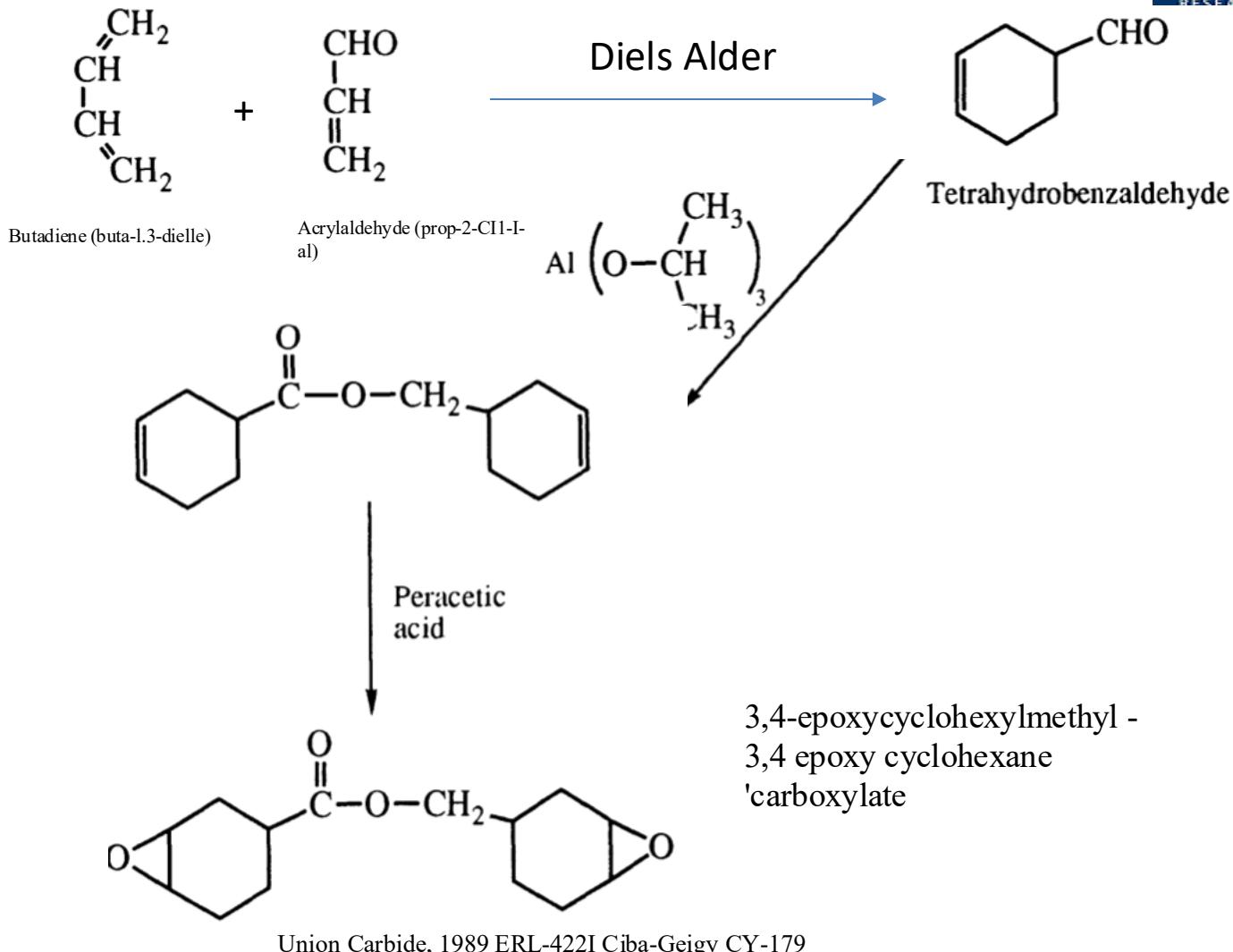
DOI: [10.1088/0965-0393/22/2/025013](https://doi.org/10.1088/0965-0393/22/2/025013)

# Manufacturing Process for BIS A-Based Epoxy Resins

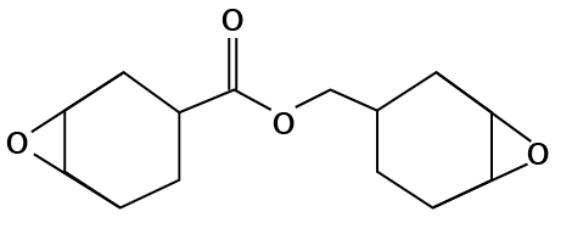
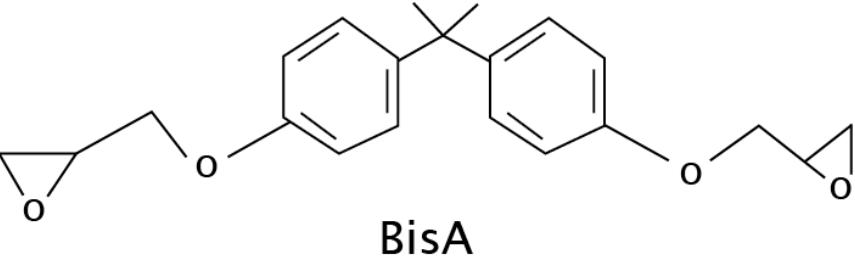


# Cycloaliphatic epoxy

- Free of hydrolysable chloride and inorganic salts (ash)
- No contain aromatic compounds and hence are more stable to UV exposure than BPA



# Anhydride- Cycloaliphatic epoxy



Low viscosity  
Low ion impurity

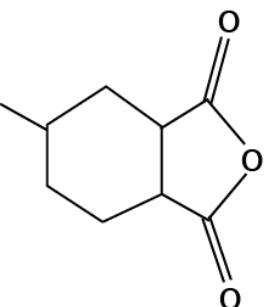
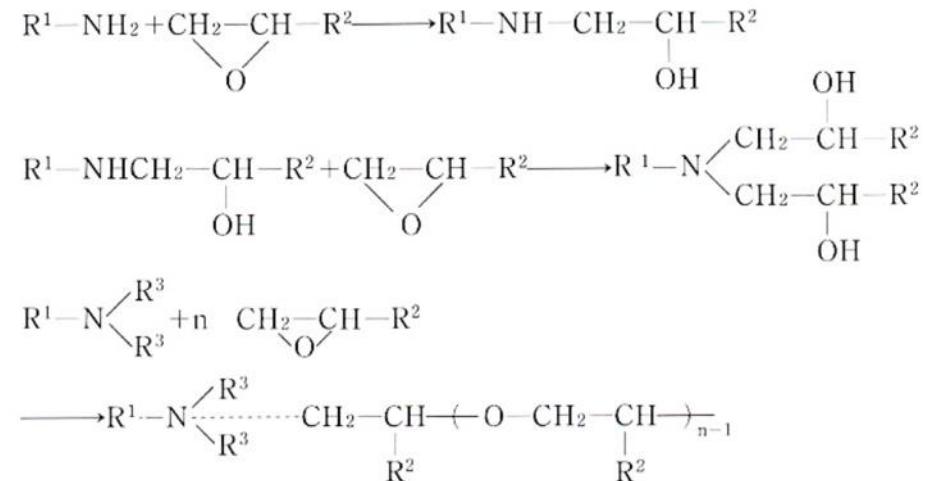


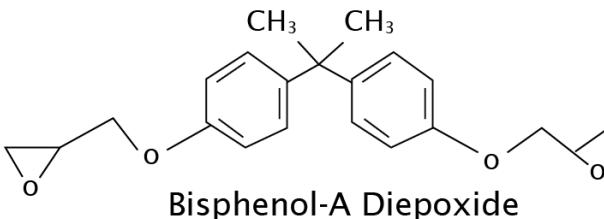
Fig 14.07 - Structures of common ingredients

- Low temperature cure

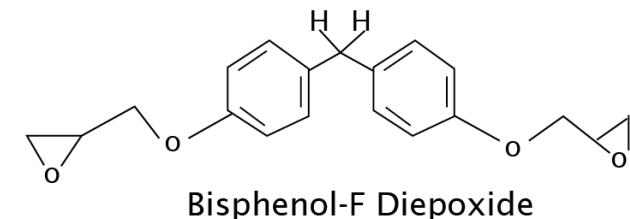
# Amine epoxy



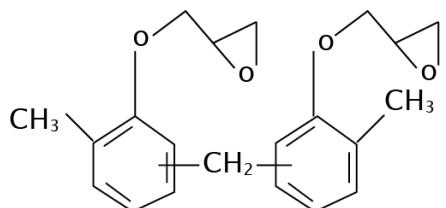
# Phenolic epoxy



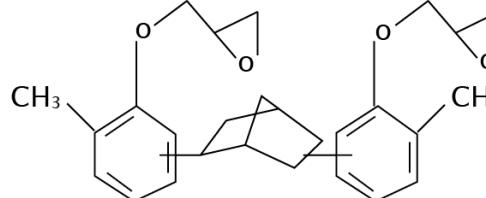
## Bisphenol-A Diepoxide



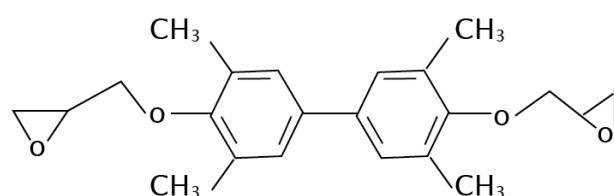
## Bisphenol-F Diepoxide



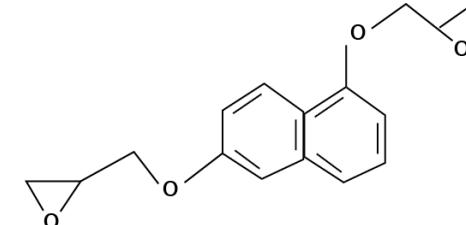
## Cresol Novalac Diepoxide



## Dicyclopentadiene Cresol Novalac Diepoxide



## Tetramethylbiphenol Diepoxyde



## 1,6-Naphthalene Diepoxide

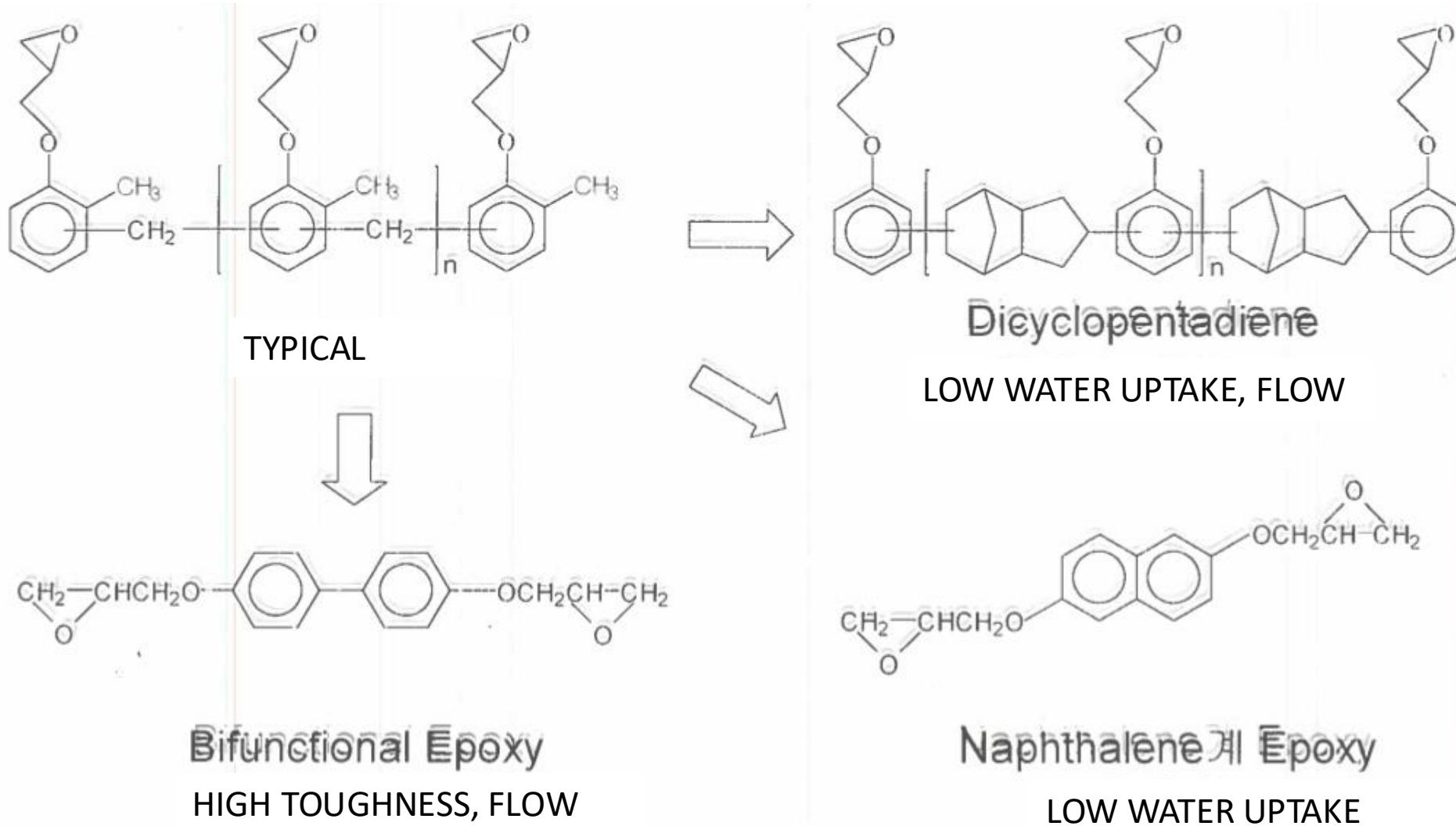
Fig 14.08 - Structures of common aromatic epoxy

# EMC



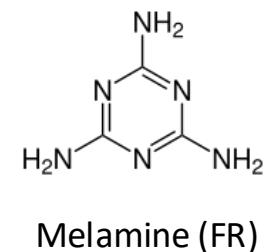
- Typical recipes
- Epoxy
- Curing agents
- Additives
- Filler
- High filler loading

# Epoxies for high performance EMC



# EMC Composition

No	Component	Remarks	Compounding ratio
1	Epoxy	- Binder	4 – 15
2	Hardener	- Curing agent	4 -10
3	Br-Epoxy	- Organic flame retardant	2 >
4	Sb <sub>2</sub> O <sub>3</sub>	- Inorganic flame retardant	3 >
5	Filler	- Improving strength, thermal conductivity - Lowering thermal expansion coefficient, moisture absorption	70 – 90
6	Coupling agent	- Improving adhesion between filler and organic materials	1 >
7	Accelerator	- Accelerating reaction of epoxy with hardener	1 >
8	Flexibilizer	- Lowering thermal stress	5 >
9	Wax	- Releasing agent for warability - Improving degree of mixing	2 >
10	Colorant	- Coloring	1 >



Functional groups in silanes

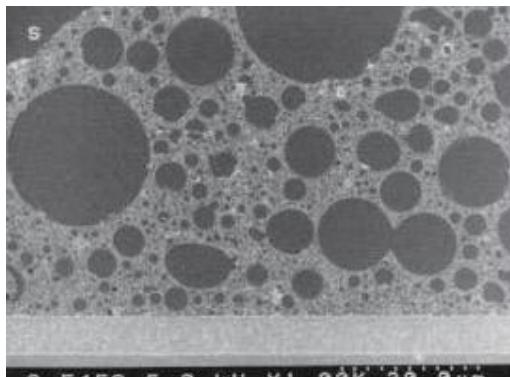
X-TBN (butadiene acrylonitrile)

Reactive silicone

Sea-island

# Filler

- Critical element to govern the properties of compounds due to significant loading levels
- Fused silica (pure, low CTE, low k)
- Different sizes/shapes; multimodal sized with maximum packing factor
- Surface modification with various reactive groups

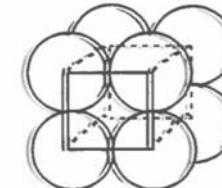


## Mooney's Equation

$$\ln\left(\frac{\eta_C}{\eta_0}\right) = \frac{K_E \phi_f}{1 - \phi_f/\phi_m}$$

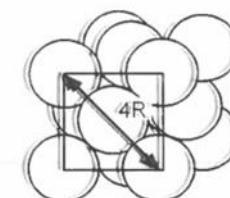
$$\phi_m = \frac{\text{True volume of the filler}}{\text{Apparent volume occupied by the filler}}$$

### 1) Simple cubic

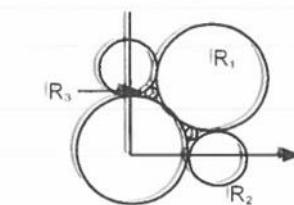
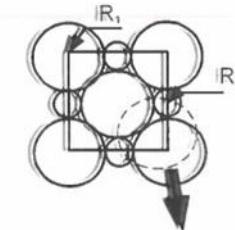


$$\phi_m = \frac{1}{(2R)^3} \left( \frac{4}{3} \pi R^3 \times \frac{1}{4} \times 4 \right) = 0.52$$

### 2) Face centered cubic



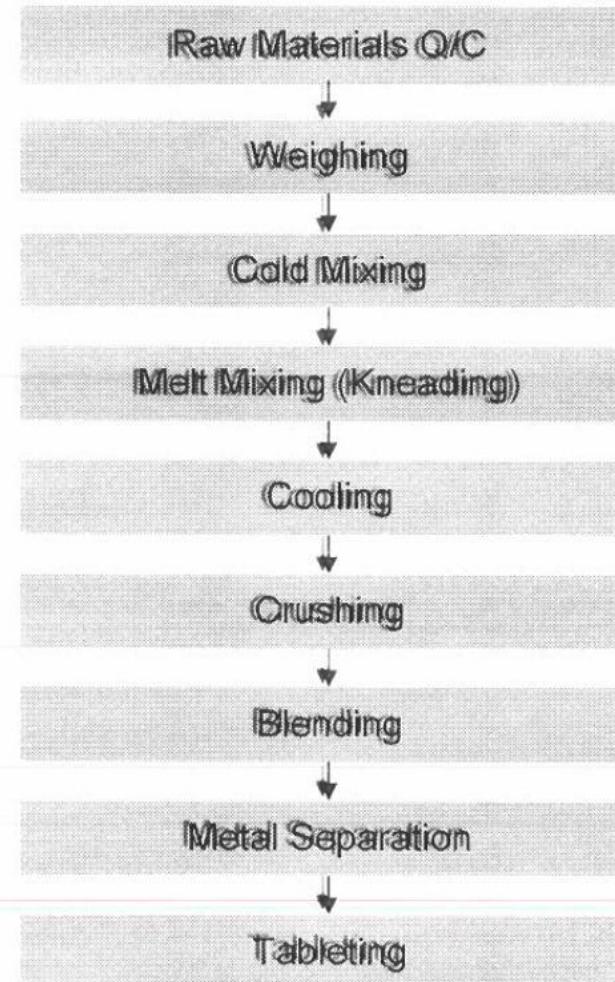
$$\phi_m = \frac{1}{2\sqrt{2}R^3} \left( \frac{4}{3} \pi R^3 \times 4 \right) = 0.74$$



$$R_1 : R_2 : R_3 = 1 : 0.414 : 0.114$$

$$\phi_m = 0.793$$

# Making EMC



# No-Flow Underfill Technology



## Generic Composition of (No)-Flow Underfill Material

- **Base resin; flowability**
- **Curing agent (hardener); flowability**
- **Latent catalyst; control curing profile for bump materials**
- **(Fluxing agent); enabling in-situ soldering**
- **Modifiers; toughness**
- **Silica filler; CTE control**

# Cyanate ester

- High Tg
- Lower water uptake
- Low dielectric permittivity
- Epoxy-cyanate ester (cost reduction and less polar)
- High temperature applications

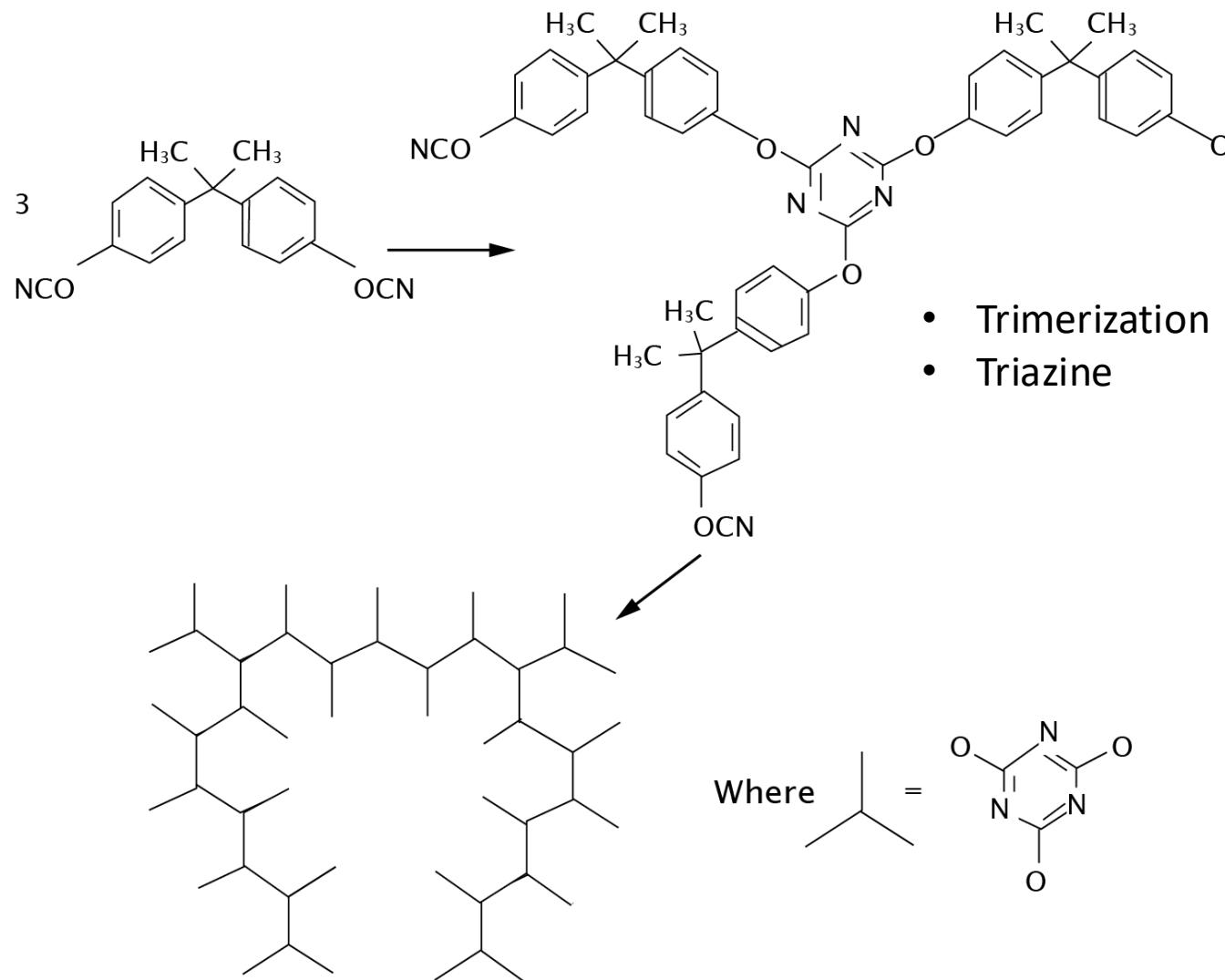


Fig 14.09 - Structures of cyanate ester polymerization products

# Urethane

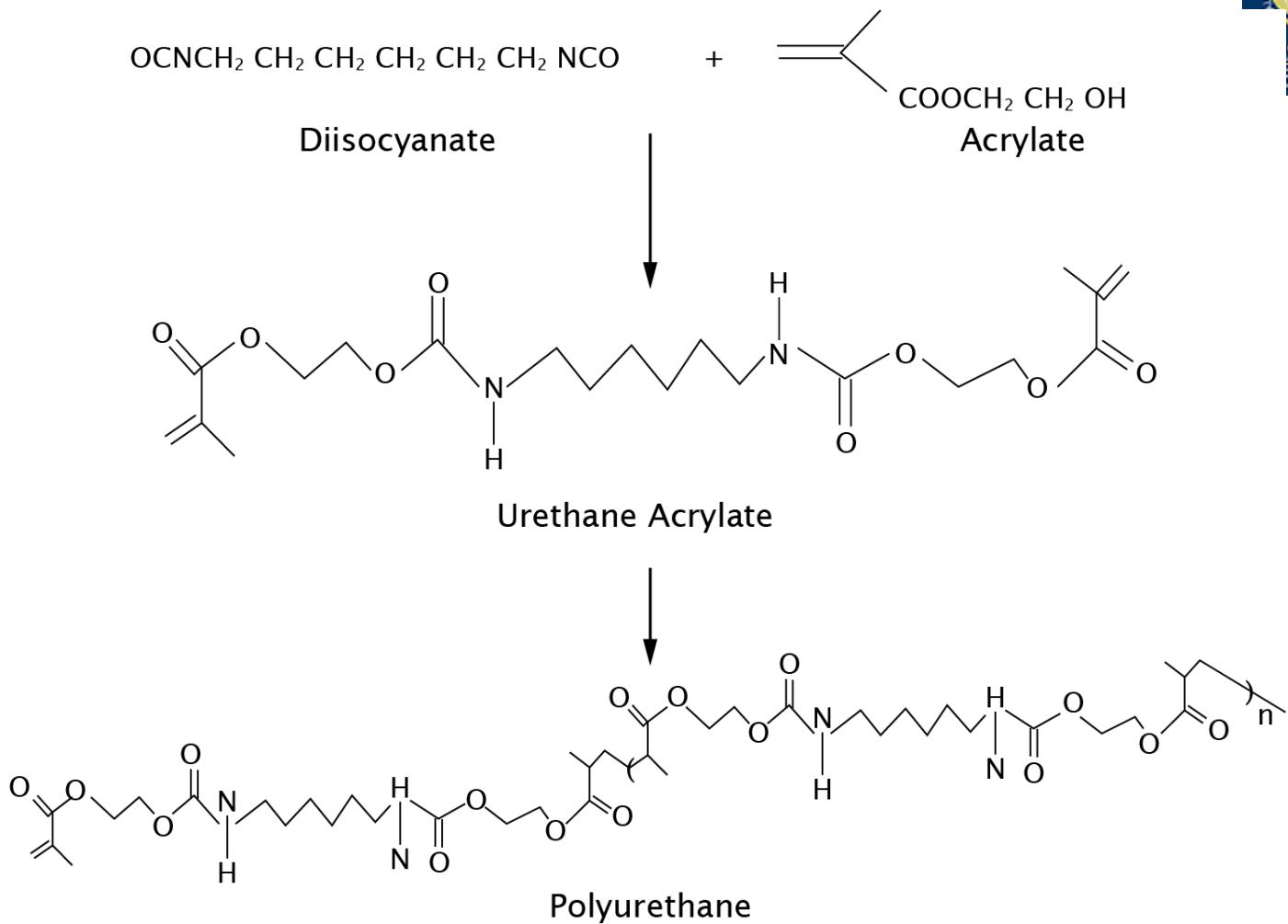


Fig 14.10 - Structures of urethane

# Silicone

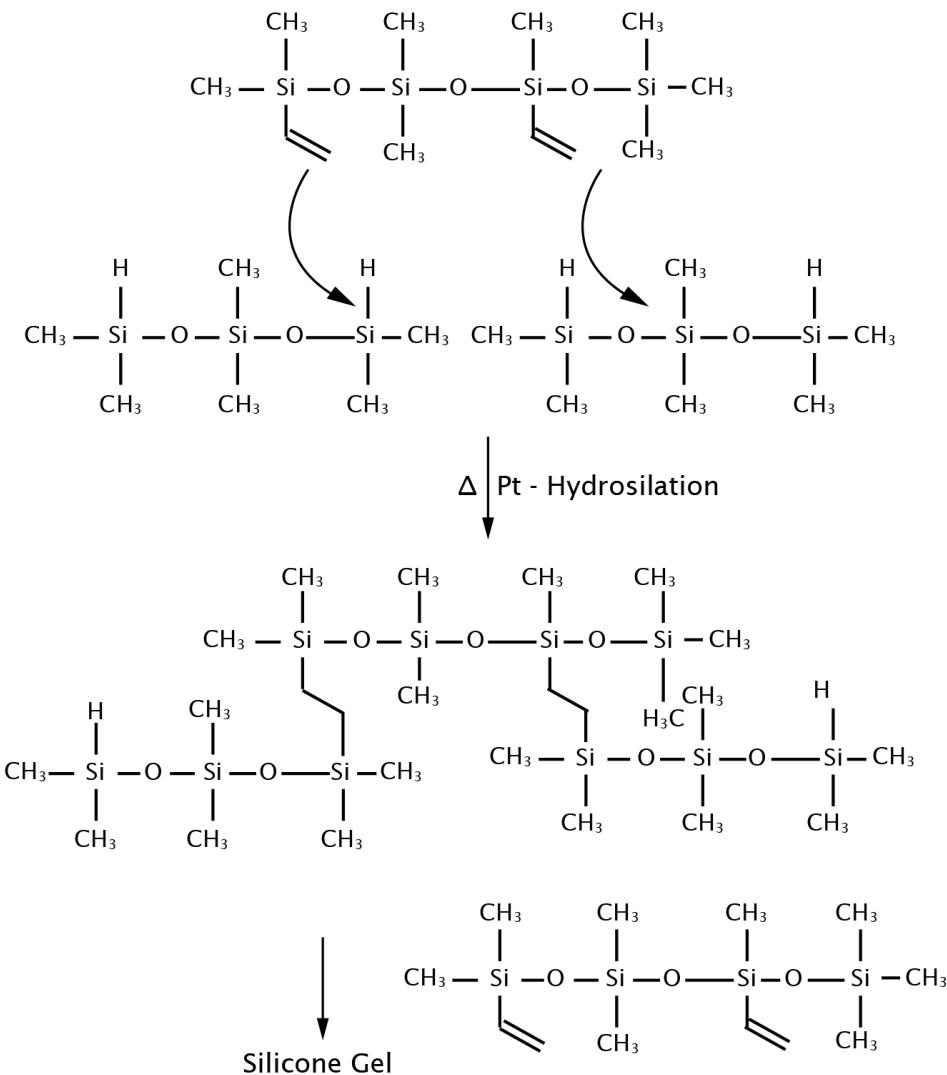


Fig 14.11 - Silicon cross-linking reaction

# Silicone

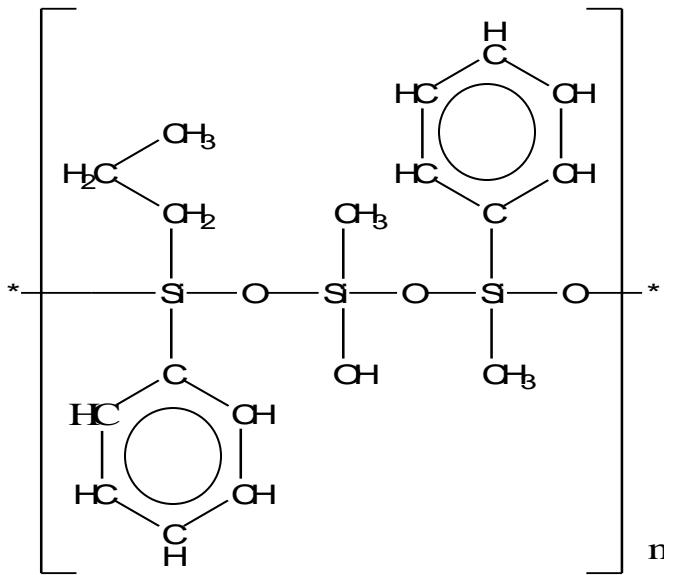
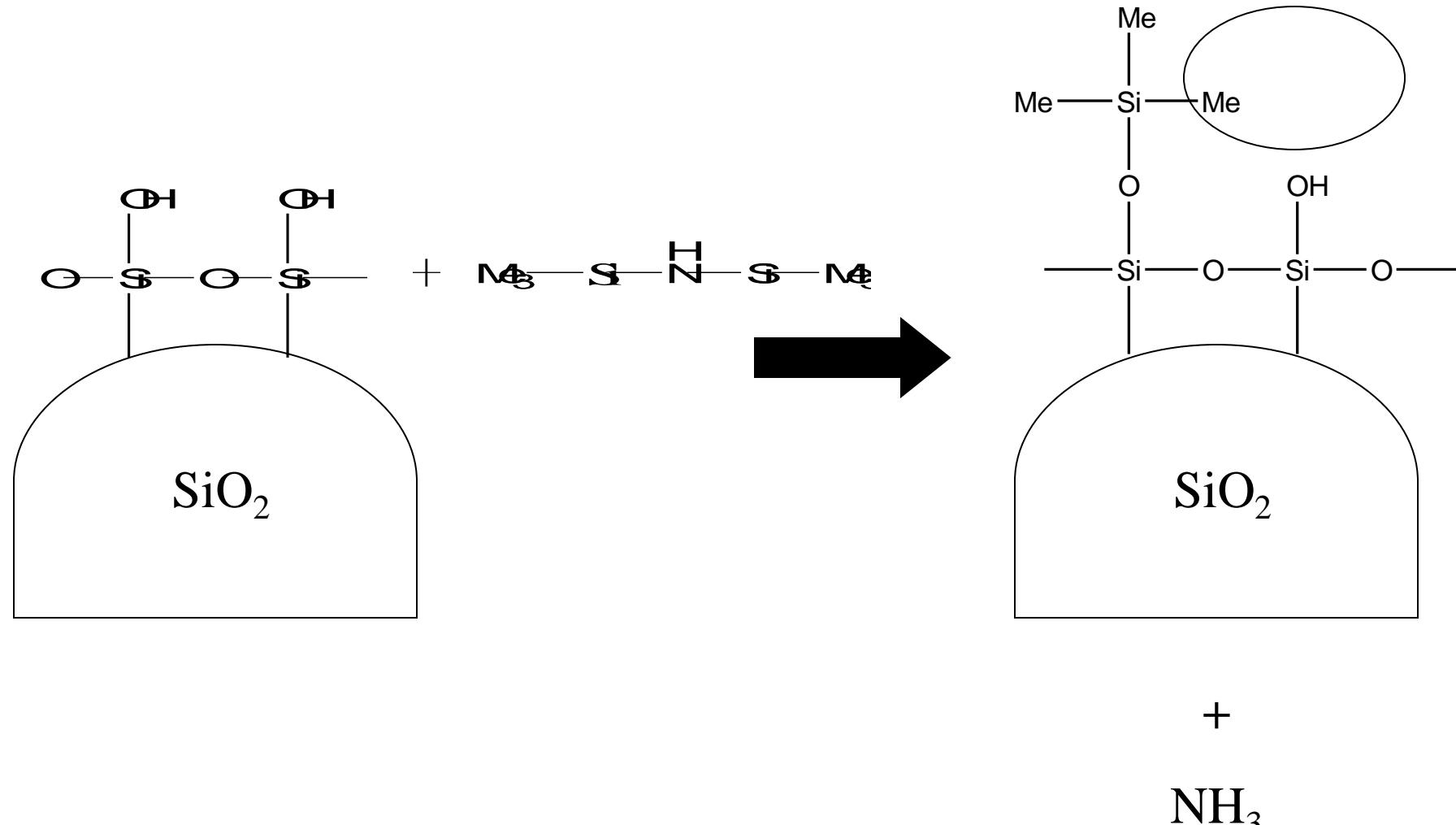


Figure: Silicone structure has a Si-O-Si backbone that provides thermal stability of the material. Hydrocarbons that attach to silicon atoms provide water-repelling properties.

## General Properties of Silicone

- Superior thermal properties (-100° C → 200-300° C)
- Non-wettable by liquid water
- Resistance to UV and other radiations
- Excellent electrical properties
- Low surface tension
- With additive and thicken agent → Good lubricant
- Long chain PDMS → Biological non-active

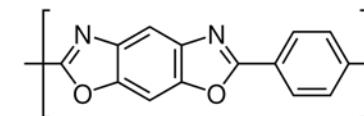
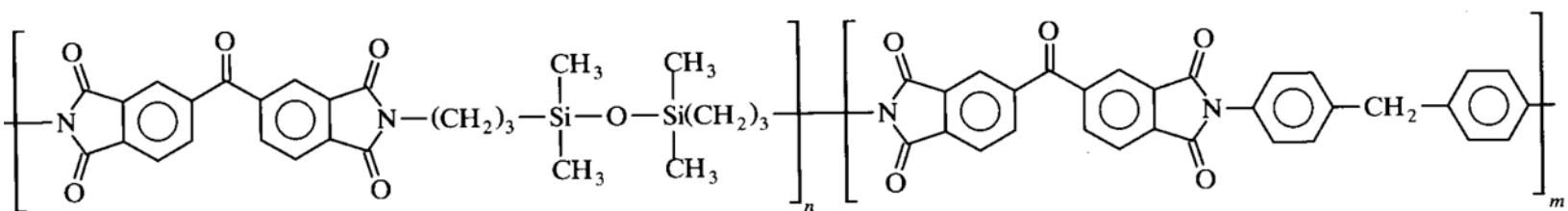
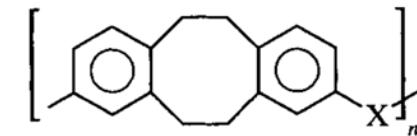
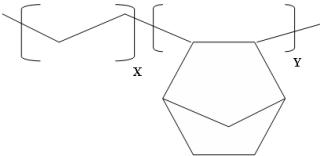
# Deactivation of CAB-O-SIL



# Other polymers

- Parylenes (Poly-Para-Xylylenes)
- Benzocyclobutene (BCB)
- Silicone-polyimide
- Cyclo-olefin copolymers
- PBO (polybenzoxazole)

Type of Parylenes	Deposition Rate ( $\mu\text{m/hr}$ )	Dissipation Factor	Properties	Dielectric Constant	Water Absorption (%)
N Type:		1	0.0002	0.02	
C Type:		3 - 5	0.02	0.06	
D Type:		10 - 15	0.01	0.07	
F Type:		?	<0.0002	<0.02	



# 14.5 Encapsulation processes

- Cakes Molding
  - Transfer molding
  - Compression molding
- Liquid
  - Cavity filling
  - Glob top
  - Underfilling



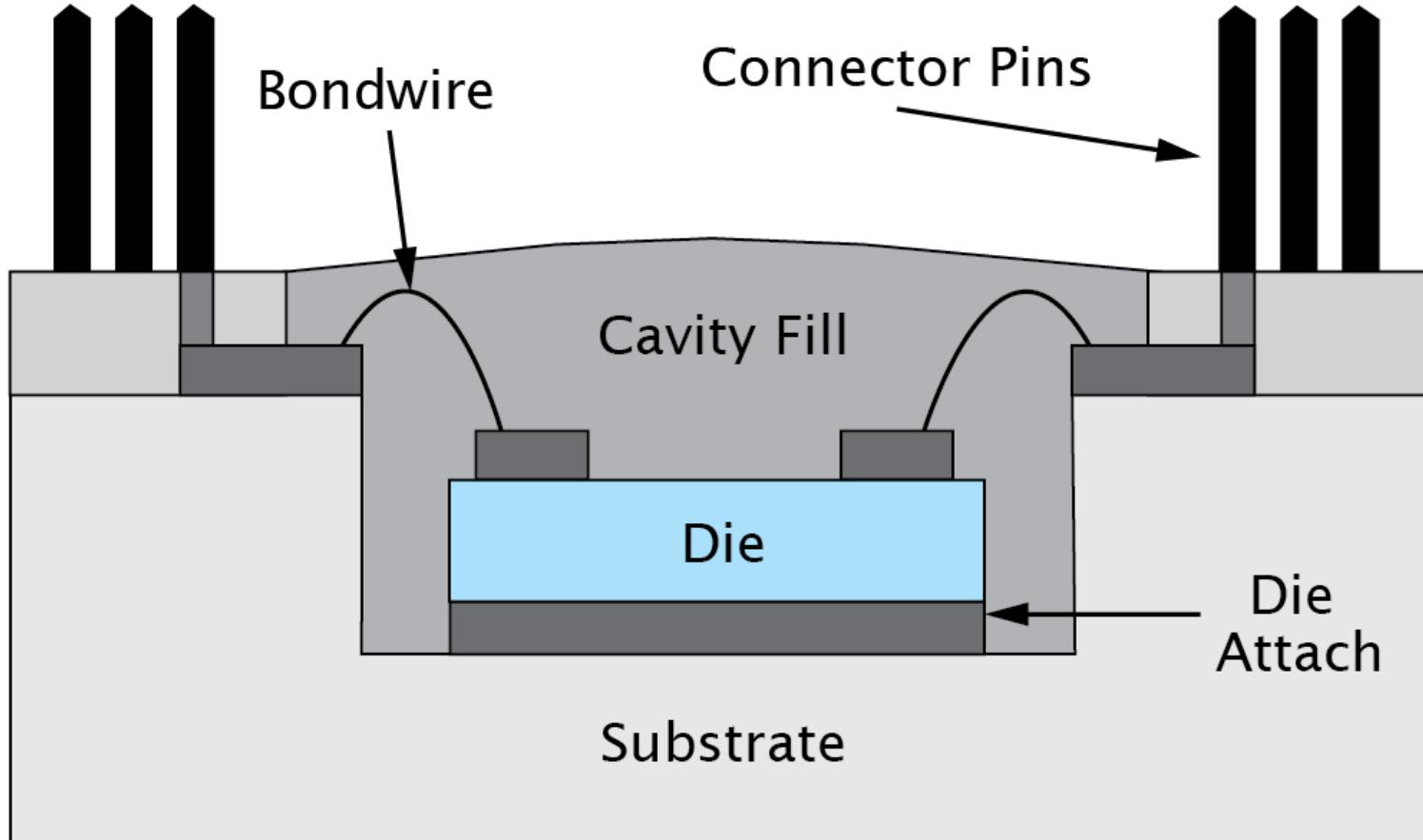
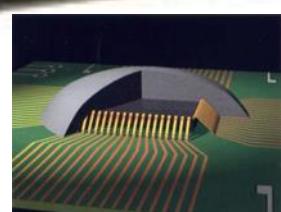


Fig 14.12 - Cavity filling of PGA

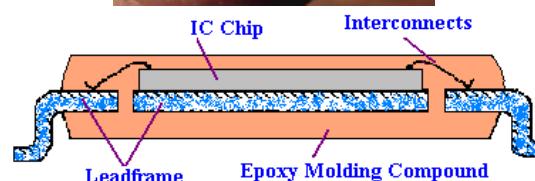
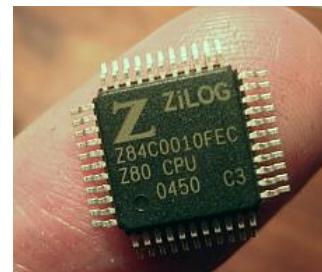


# Package Encapsulants

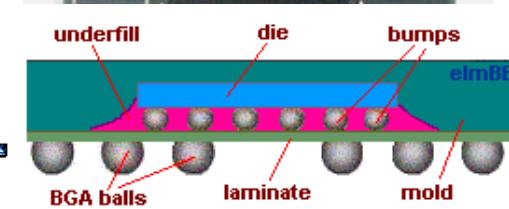
- *Physical/chemical protection of functional chips*
  - *Glob tops*
  - *Epoxy molding compounds,*
  - *Underfills for flip-chip type packages*



Glob top



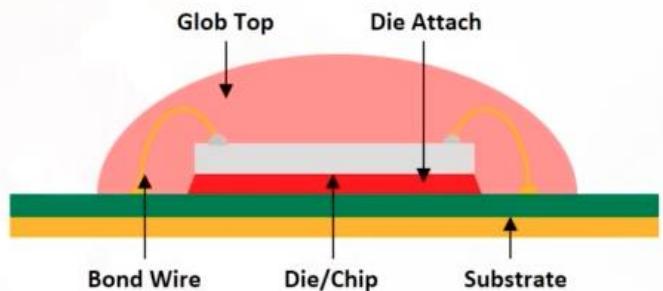
EMC



Underfills

# Glob top vs Dam/Fill

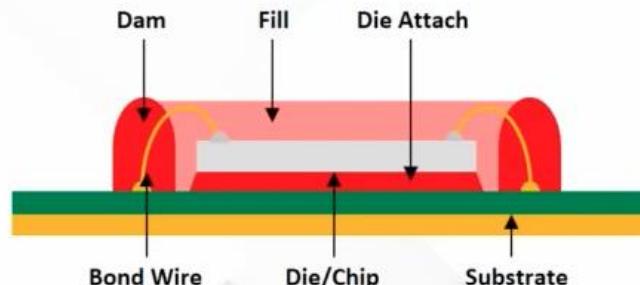
**Glob Top**



- For **small** die sizes below 2x2mm
- Typically higher thickness
- Low viscosity and fine-tuned thixotropy
- **One step** process
- Difficult to control the shape



**Dam & Fill**



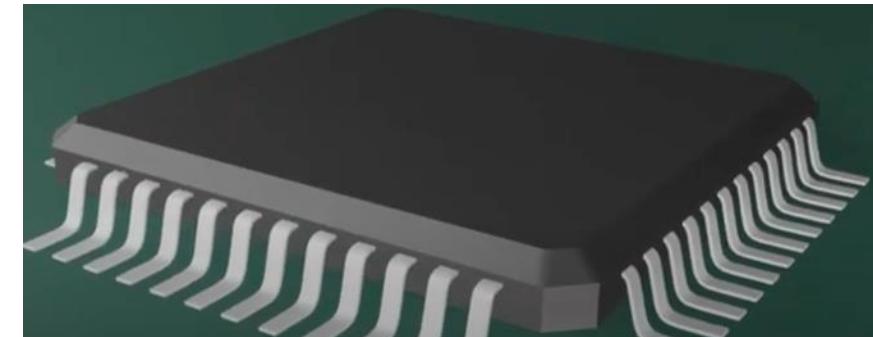
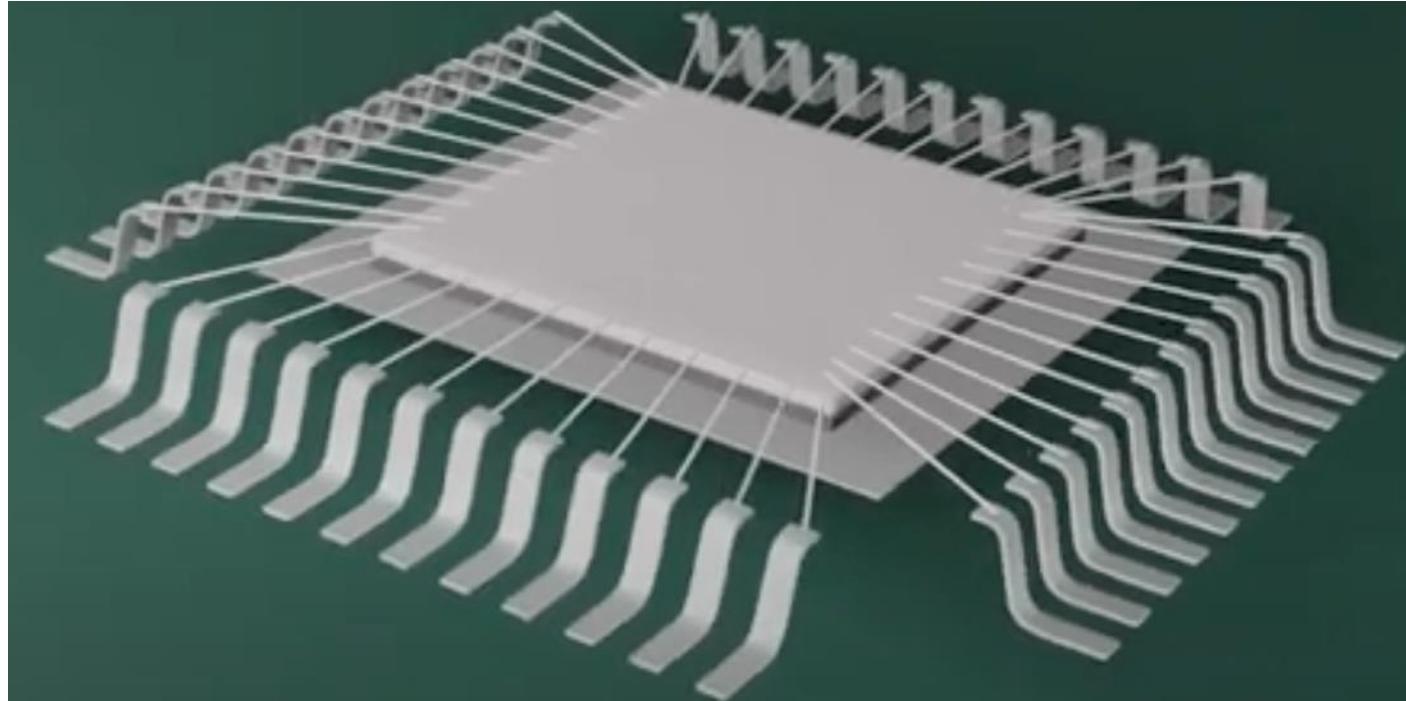
- For **large** die sizes up to 10x10mm
- Square shaped to evenly distribute stress after cure
- Requires High viscosity DAM + Self levelling FILL
- **Two step** process
- Tight space and shape control

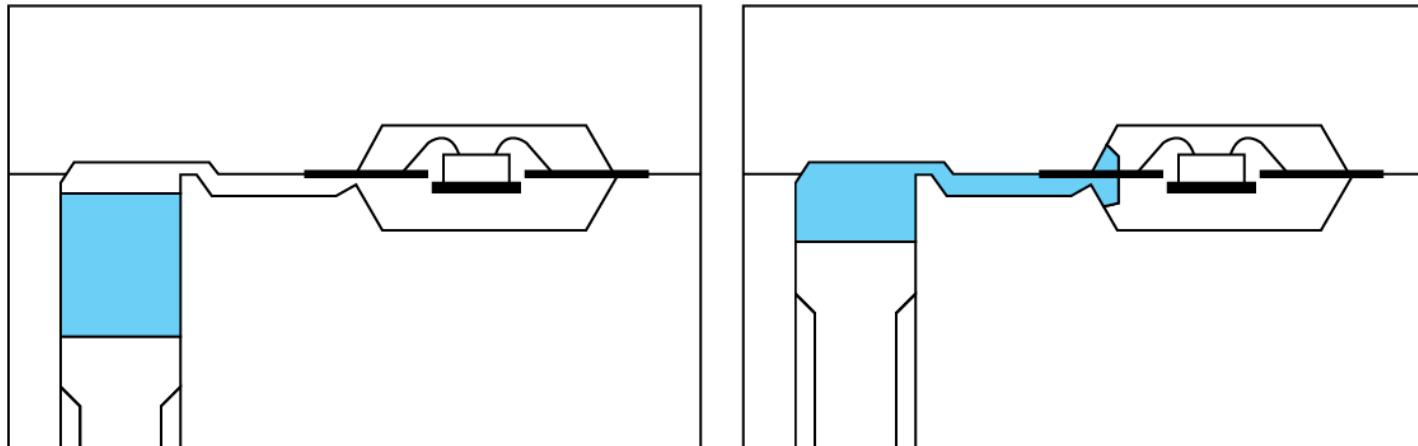


From Capling

65

# Molding leadframe based package





(a)

(c)

(b)

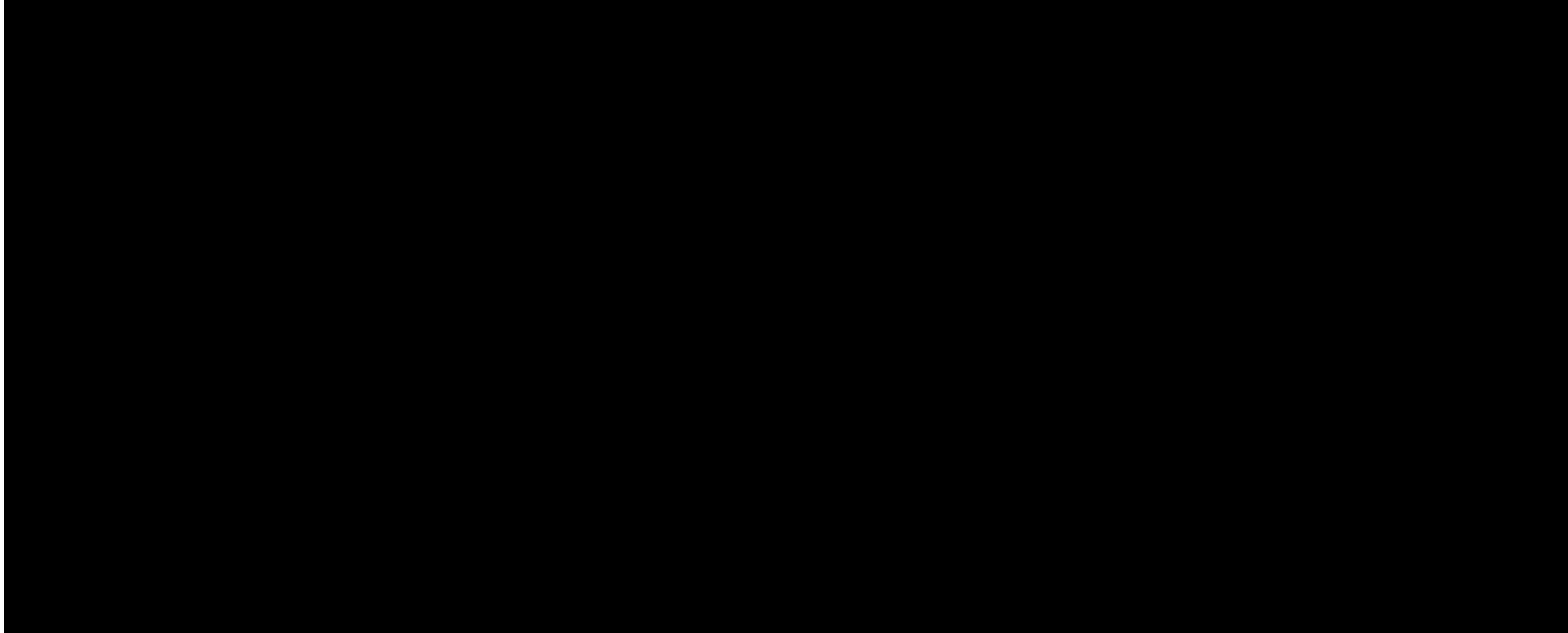
(d)



- Spiral flow; cm @molding temperature
- Minimum viscosity by capillary rheometer

Fig 14.13 - A typical transfer model

# Transfer molding



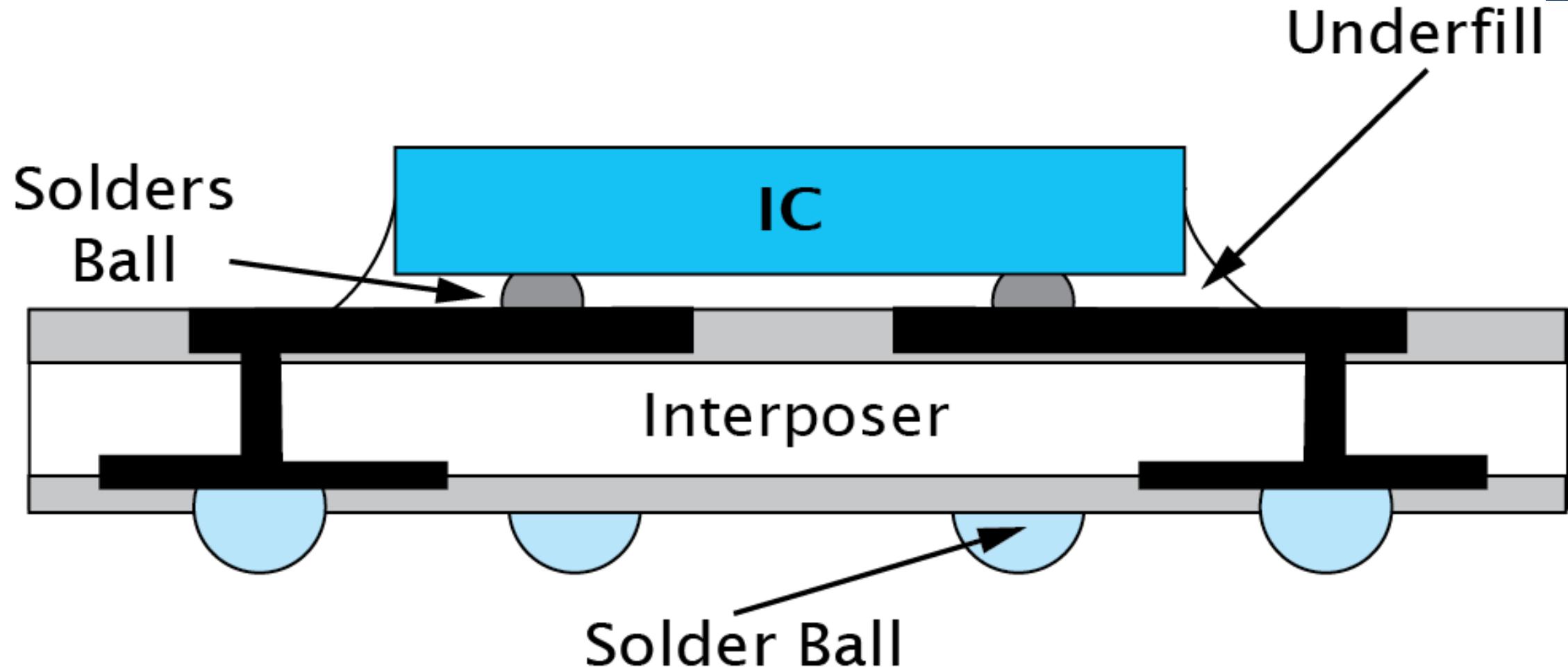


Fig 14.15 - A typical flip-chip BGA

- Underfilling techniques
  - Capillary filling
  - Pre-applied/No-flow
  - Wafer level coating/lamination
  - Molded UF

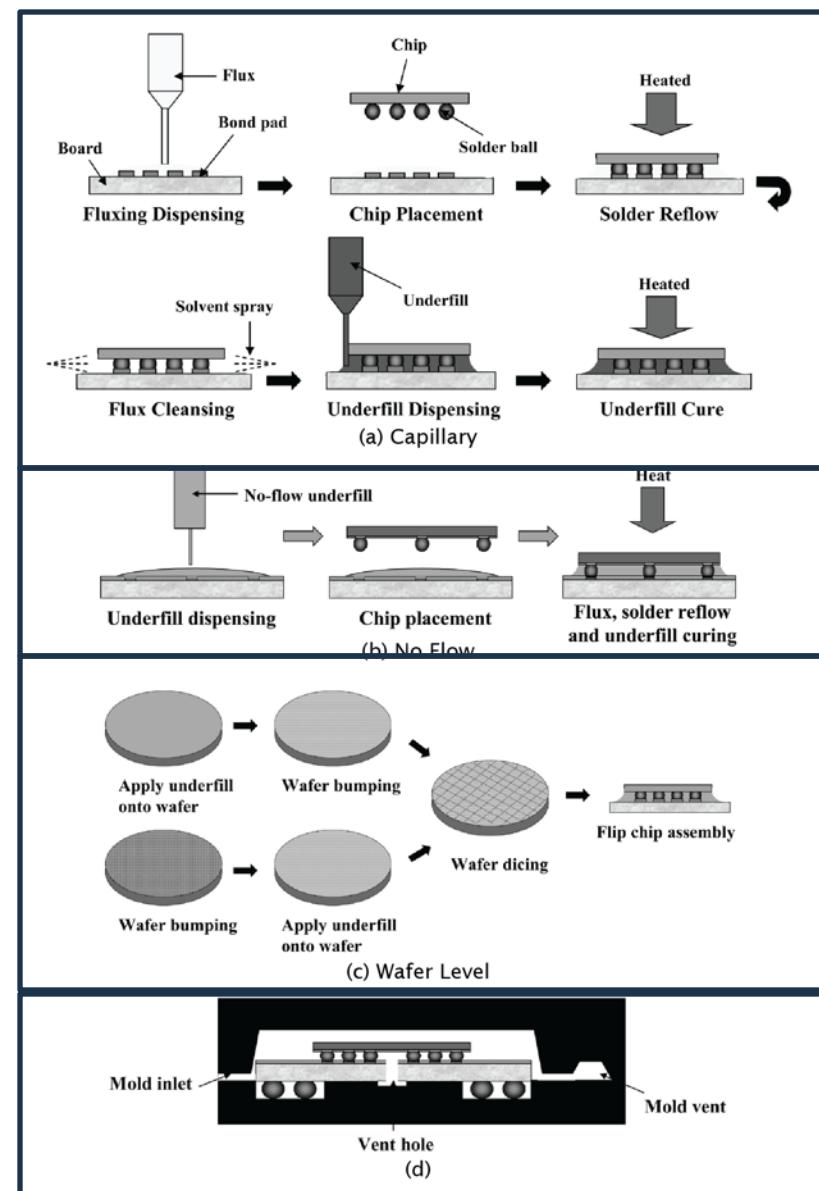
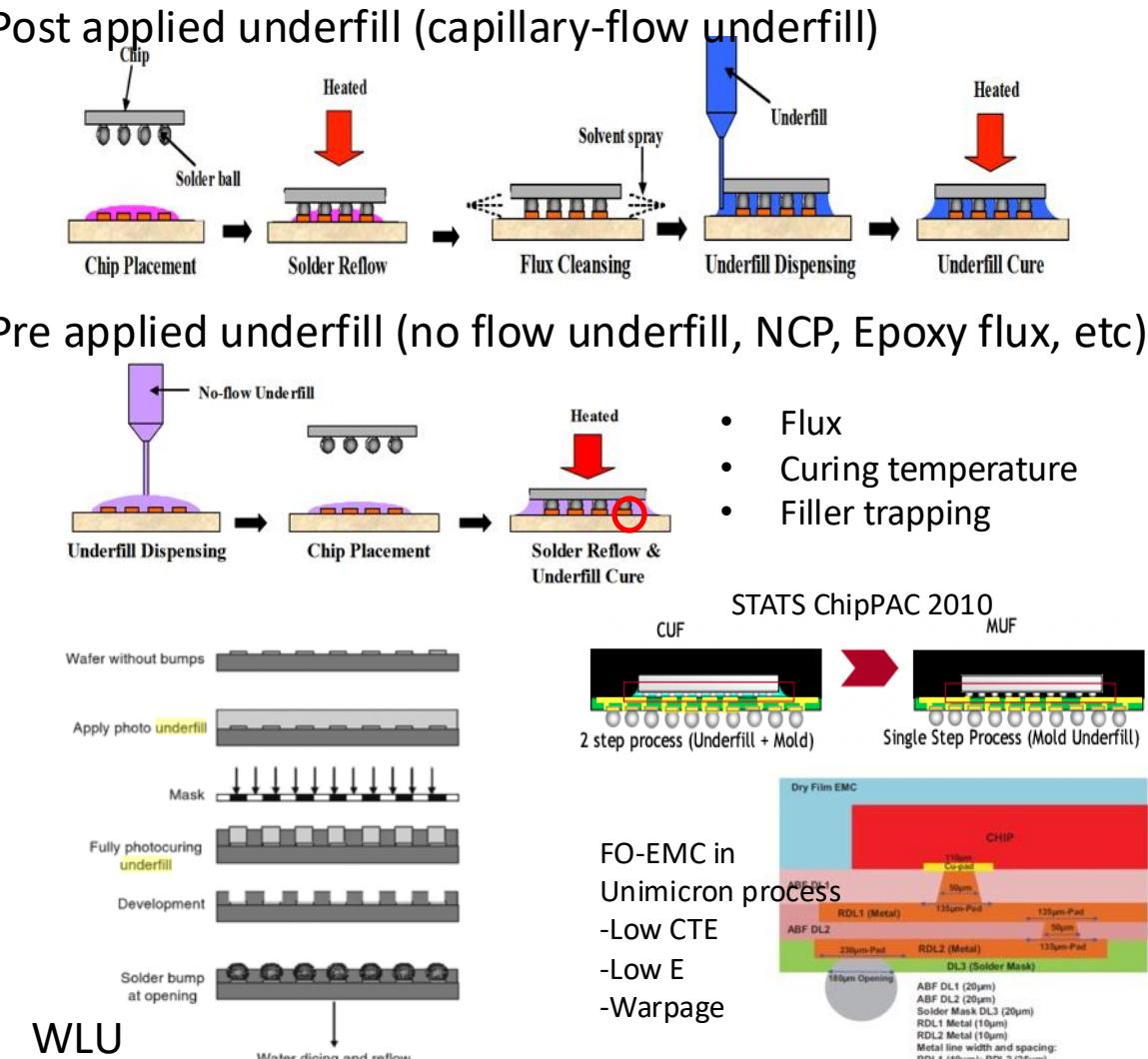


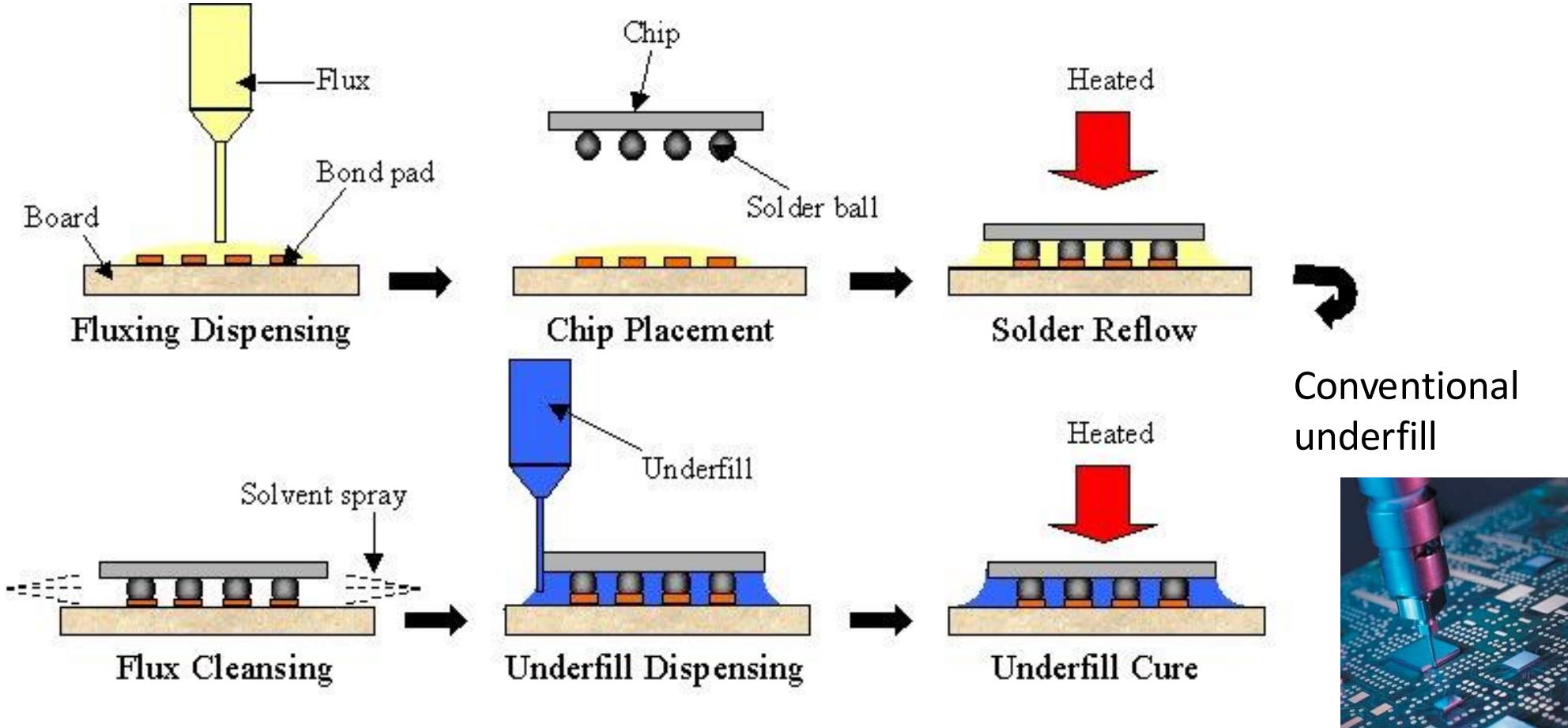
Fig 14.17 - Four underfilling processes

# Stress Redistribution of interconnecting joints (chip/wafer/panel level)

- Underfills
  - Capillary flow underfill
  - No-flow underfill
    - Process compatible chemistry
    - Flux, cure temperature
    - TCB compatible
  - Reworkable underfill
    - Chemically or thermally decomposed
    - Expensive boards
  - Wafer level underfill
    - Burn-in and test WF level
  - Molded underfill
  - EMC for FOWLP;



# Capillary Flow Underfilling



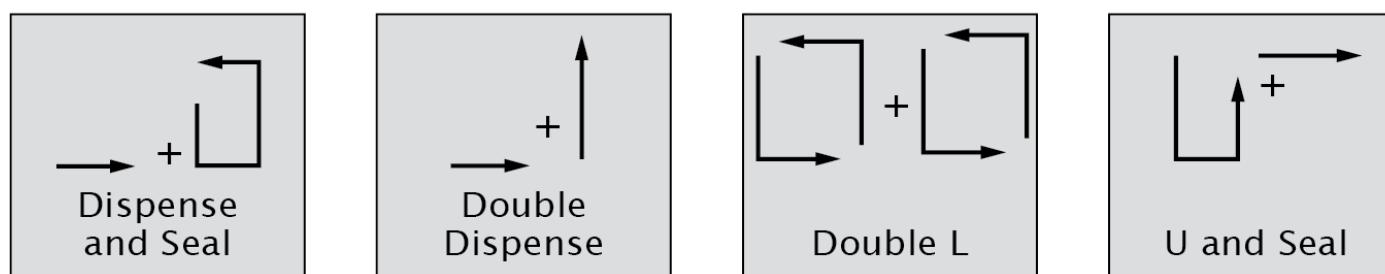
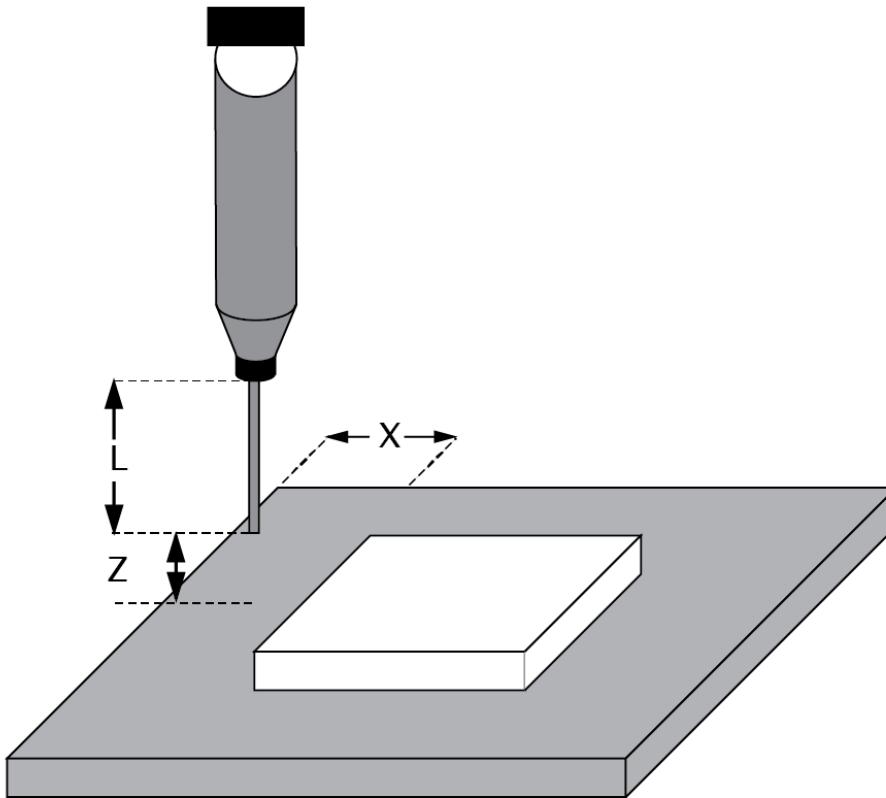
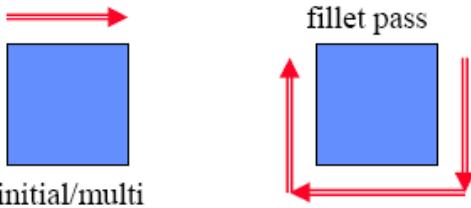


Fig 14.16 - Underfill dispensing

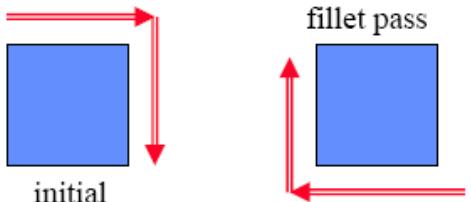
# Underfill Dispense Patterns

- ❑ Underfill Flow Depends
  - ❖ Chip Size/Shape( $W_{\text{chip}}$ ,  $L_{\text{chip}}$ )
  - ❖ Bump/Interconnect Pattern
  - ❖ Gap Size ( $h$ )
  - ❖ Underfill Viscosity
  - ❖ Underfill Surface Tension
  - ❖ Underfill Wetting Angle
- ❑ General Guidelines
  - ❖ Chip < 0.25 in: Single Side
  - ❖ Chip Between 0.25 & 0.5 in: L-Shaped (single pass)
  - ❖ Chip > 0.5 in: Single Multi-Pass
  - ❖ Selection governed by cycle time/throughput requirements

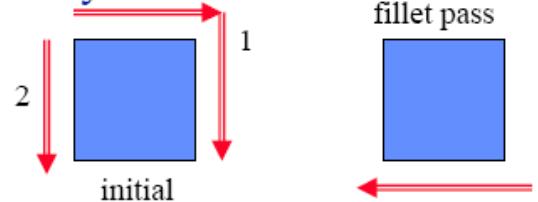
## ❑ Single Side and Multiple



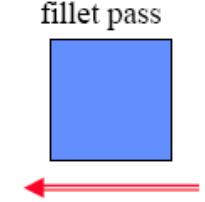
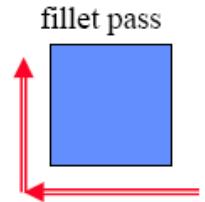
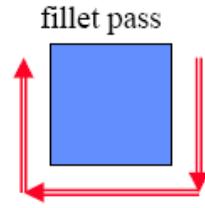
## ❑ L-Shaped



## ❑ Hybrid

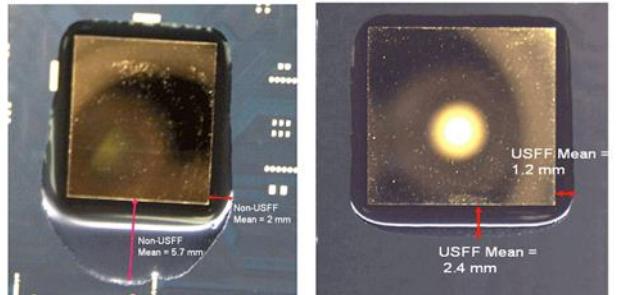


## ❑ Single Side and Multiple

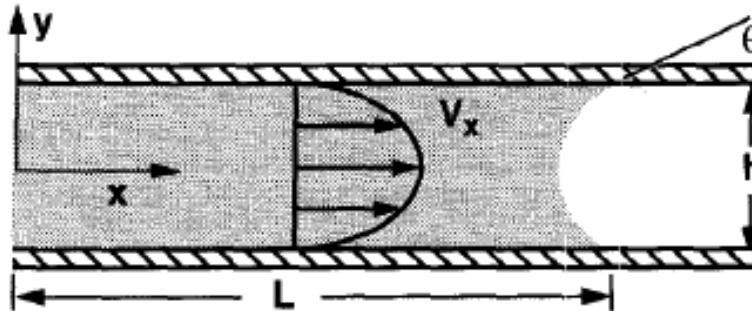
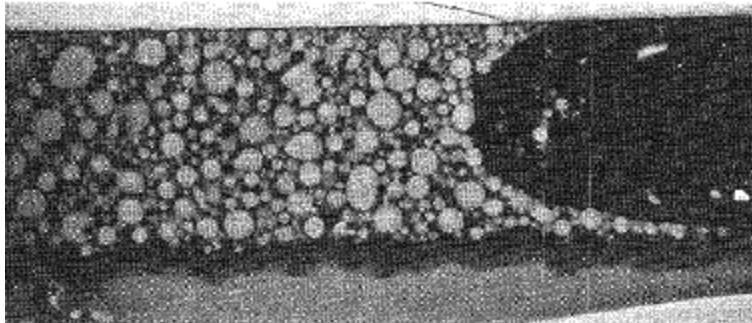


# Capillary underfilling

$$t = \frac{3hL^2}{hg \cos \theta} \quad \text{Eq 14.4}$$



$t$  is flow time,  $\eta$  is viscosity,  $L$  is distance of flow,  $h$  is die-substrate gap,  $\gamma$  is surface tension and  $\theta$  is wetting angle.



- No solder bumps
- No surface roughness
- No flux
- No other obstructions

IEEE TRANSACTIONS ON COMPONENTS AND PACKAGING TECHNOLOGIES , VOL. 28, NO. 2, JUNE 2005

# Underfill Dispense Volume Calculation

## Dispense Volume

$$V_{underfill} = V_{Standoff\ Gap} - V_{Bumps} + V_{Fillet}$$

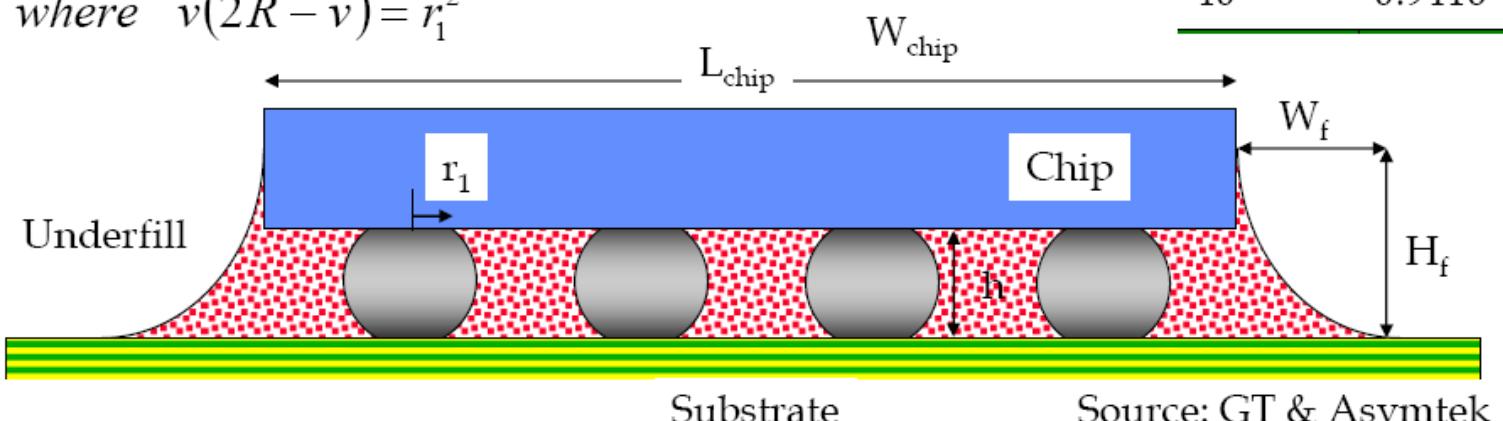
$$V_{Standoff\ Gap} = L_{chip} W_{chip} h$$

$$V_{Bumps} = N_{bumps} \left[ \frac{4}{3} \pi R^3 - \frac{\pi}{3} v^2 (3R - v) \right]$$

$$V_{Fillet} = [2(L_{chip} + W_{chip})(H_f W_f) + (W_f^2 H_f)] / 3 SF$$

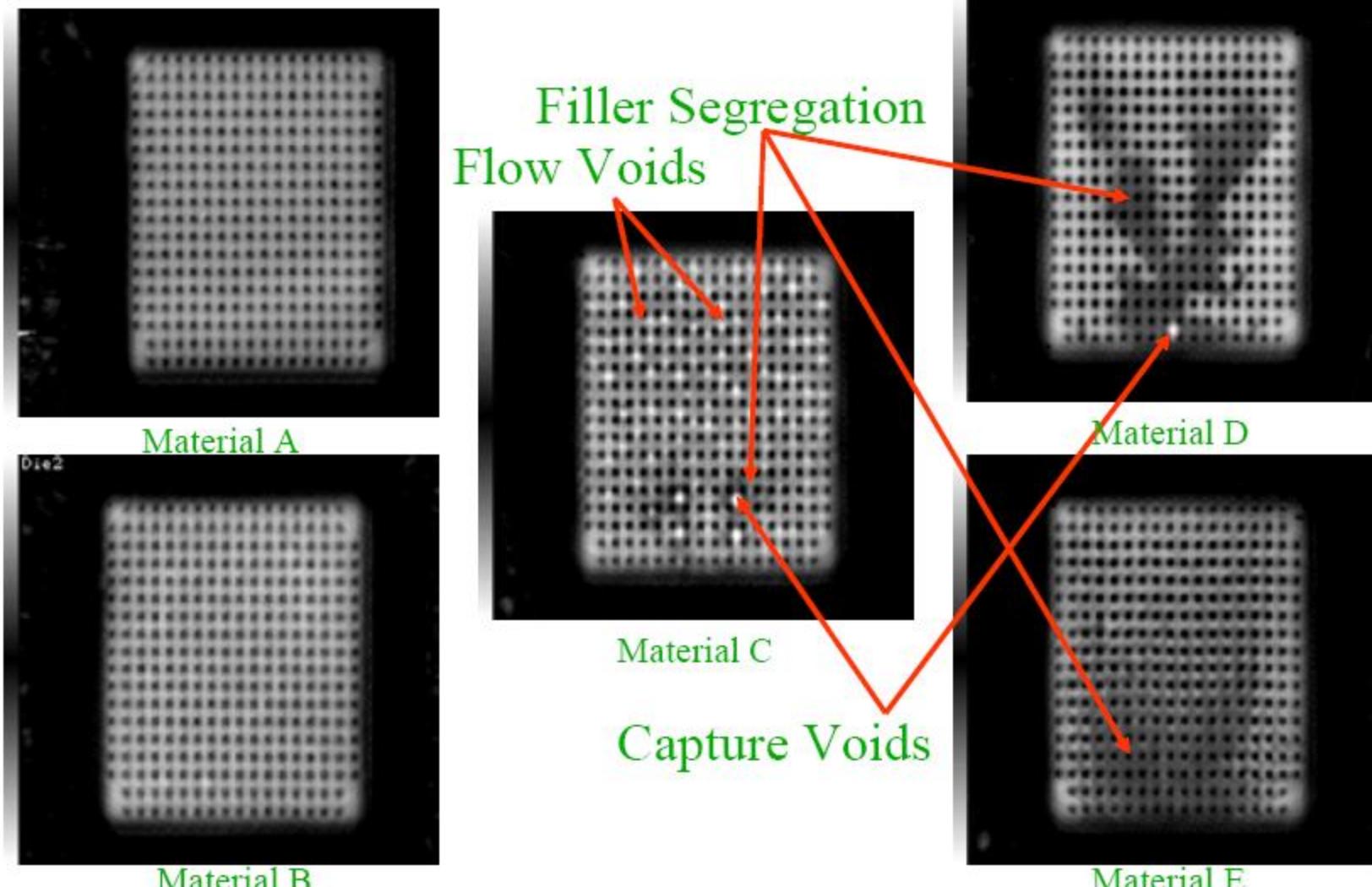
where  $v(2R - v) = r_1^2$

Contact Angle (°)	Shape Factor
1	0.1109
5	0.3202
10	0.4739
15	0.5849
20	0.6742
25	0.7505
30	0.8187
35	0.8817
40	0.9416



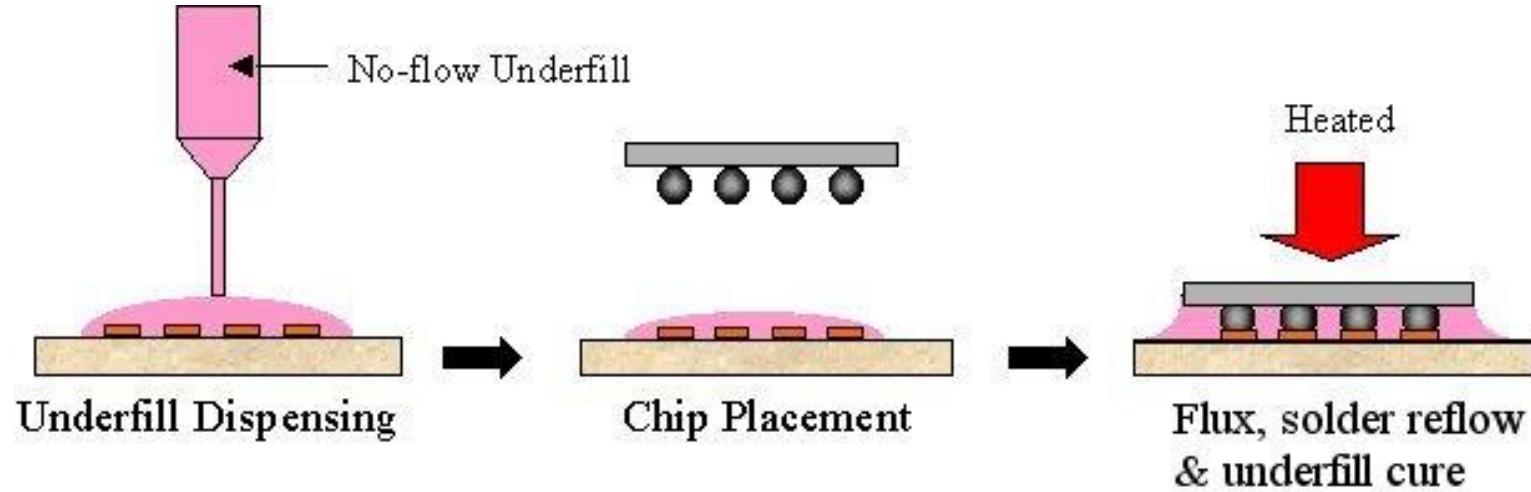
Source: GT & Asymtek

# Flow Induced Underfill Defects



C-SAM (scanning acoustic microscopy)

# No-Flow Underfill/Epoxy flux



Key material property; self-fluxing

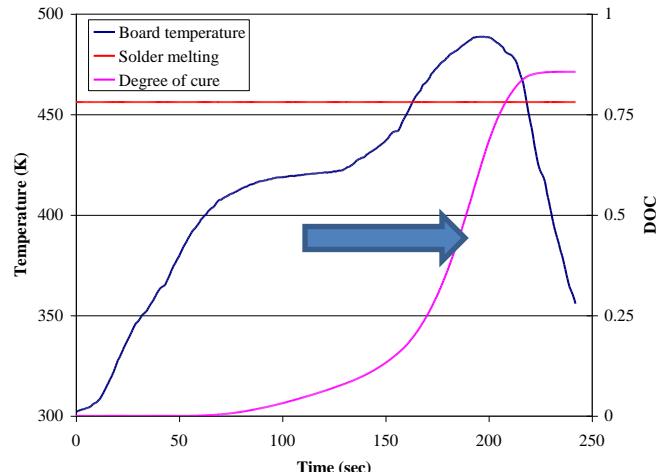
# No-Flow Underfill Technology

## Advantages:

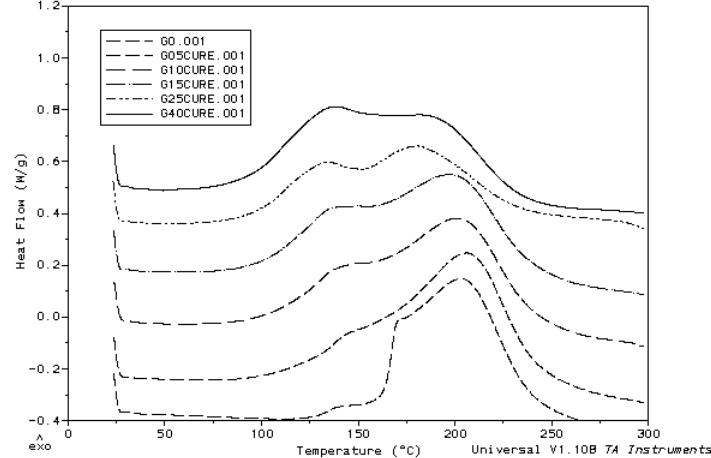
- Elimination of flux dispensing and cleaning steps
- Elimination of underfill capillary flow
- Combination of solder bump reflow and underfill cure
- Higher assembling efficiency and lower cost

# Key Characteristics of No-Flow Underfill

- Latent curing ability
  - Maintain low viscosity before solder joint formation
  - Completion of reaction after reflow
- Fluxing capability
  - Eliminate the solder oxide and facilitate solder wetting
  - Influence of flux on the material properties of underfill



Reflow profile and modeled curing degree with time

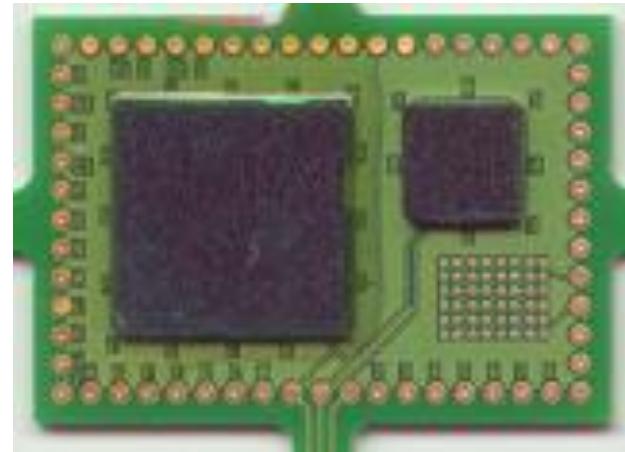
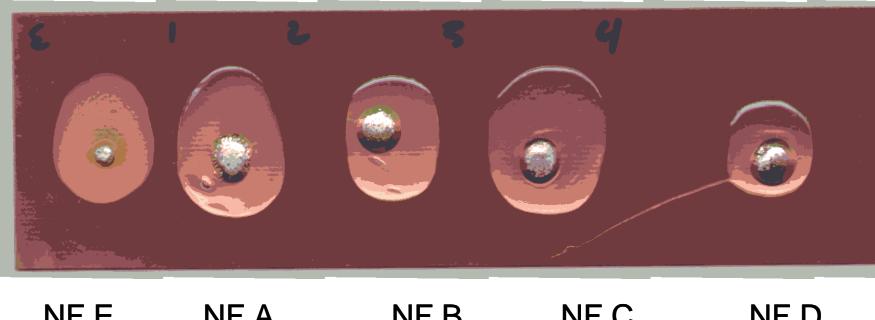
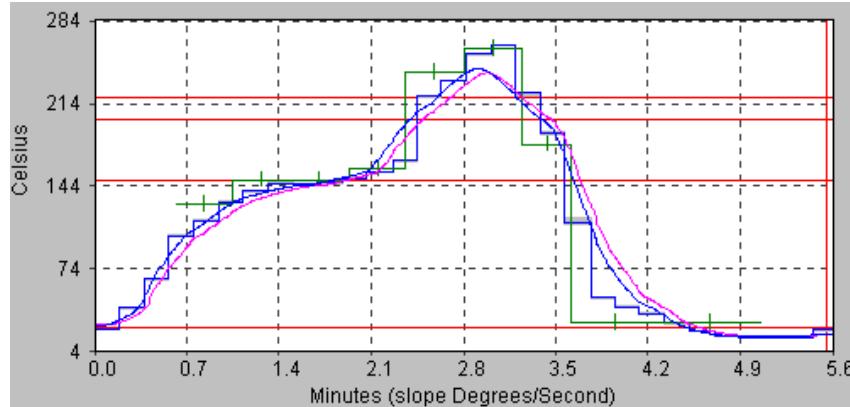


Influence of flux concentration on curing behavior

# No-Flow Underfill for Lead-Free

## Challenges

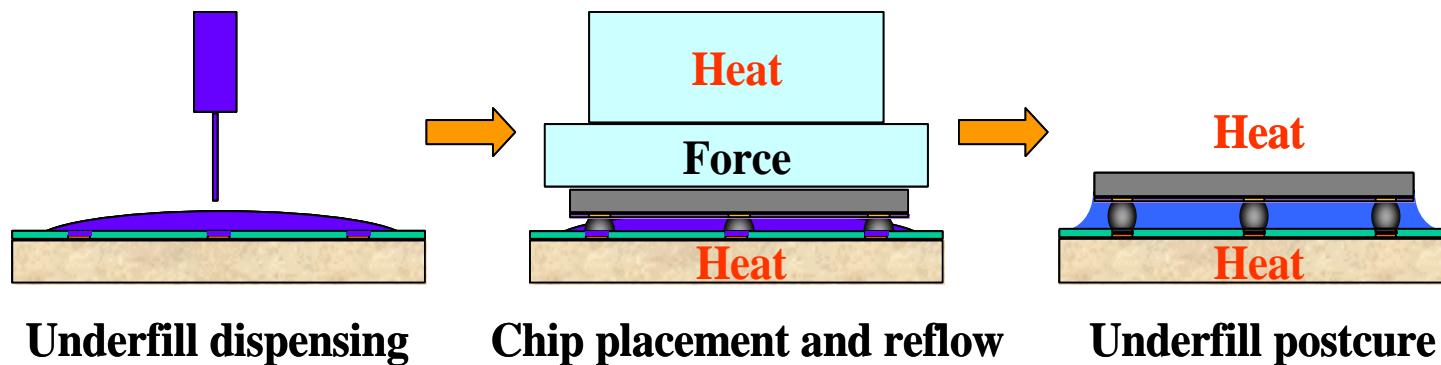
- Higher reflow temperature requires higher curing latency
- Lead-free solder needs better fluxing capability



Daisy chained, periphery array,  
pitch size: 200  $\mu\text{m}$   
**100 % yield**

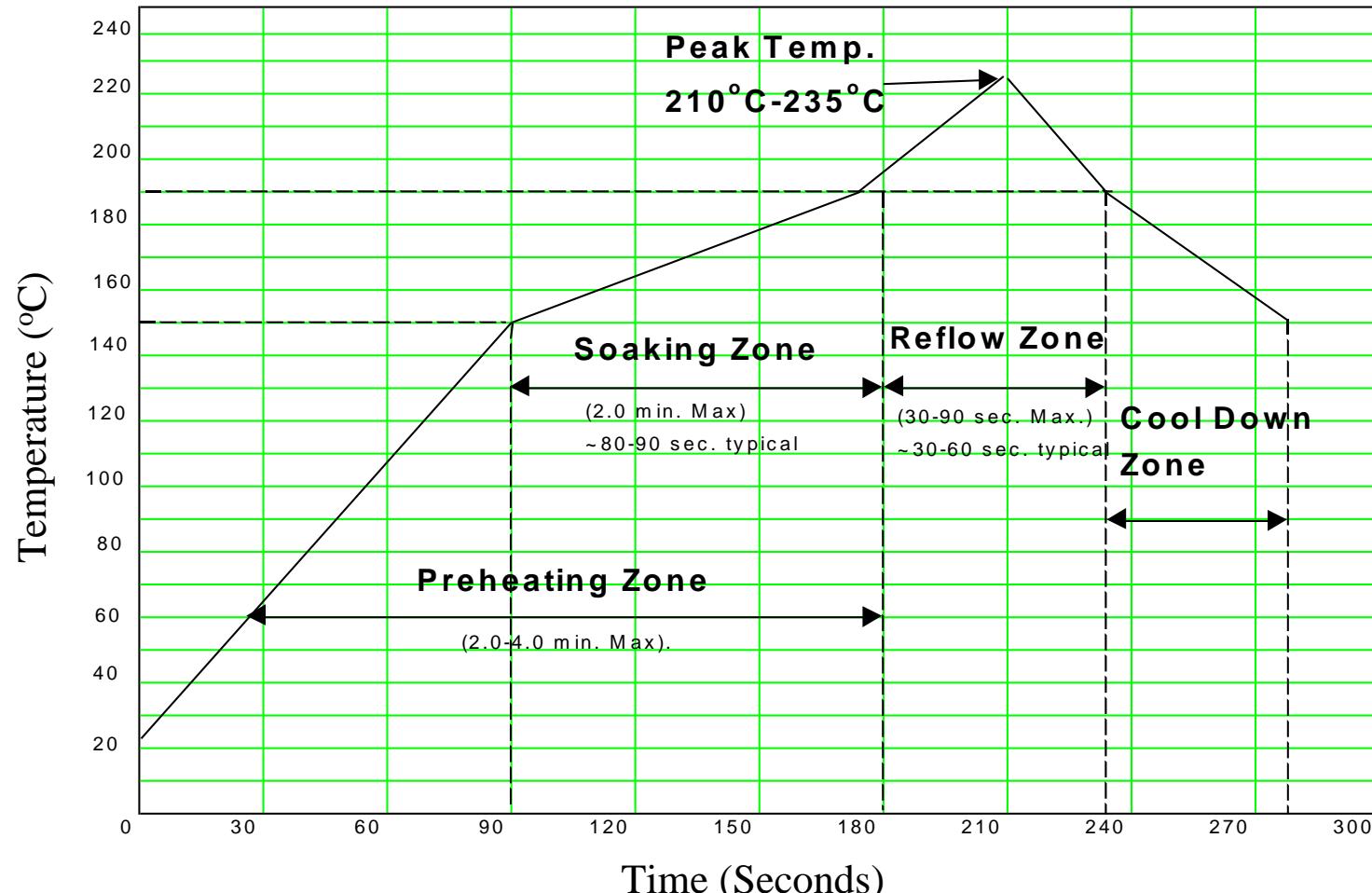
# Reliability of No-Flow Packages

- Failure mode difference for no-flow underfill
  - Localized CTE mismatch between the chip, the underfill, the board, and the solder material is serious
  - Low fracture toughness and high CTE mismatch result in underfill cracking
- Reliability enhancement for no-flow underfill
  - Improve the fracture toughness of no-flow underfill (Moon, Wong, 51<sup>st</sup> ECTC)
  - Low Tg, low modulus underfill (Liu, et al., 51<sup>st</sup> ECTC)
  - Incorporate nano-silica filler into no-flow underfill (Zhang, Wong, 52<sup>nd</sup> ECTC)

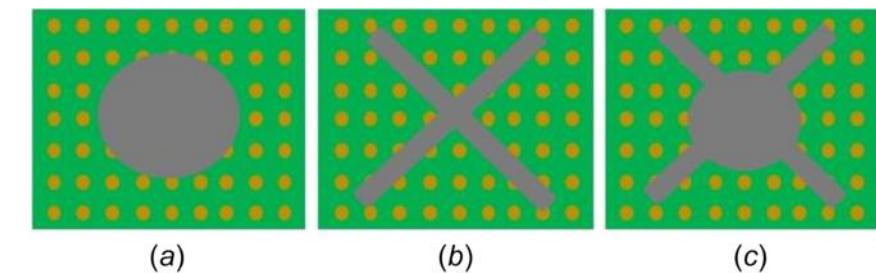
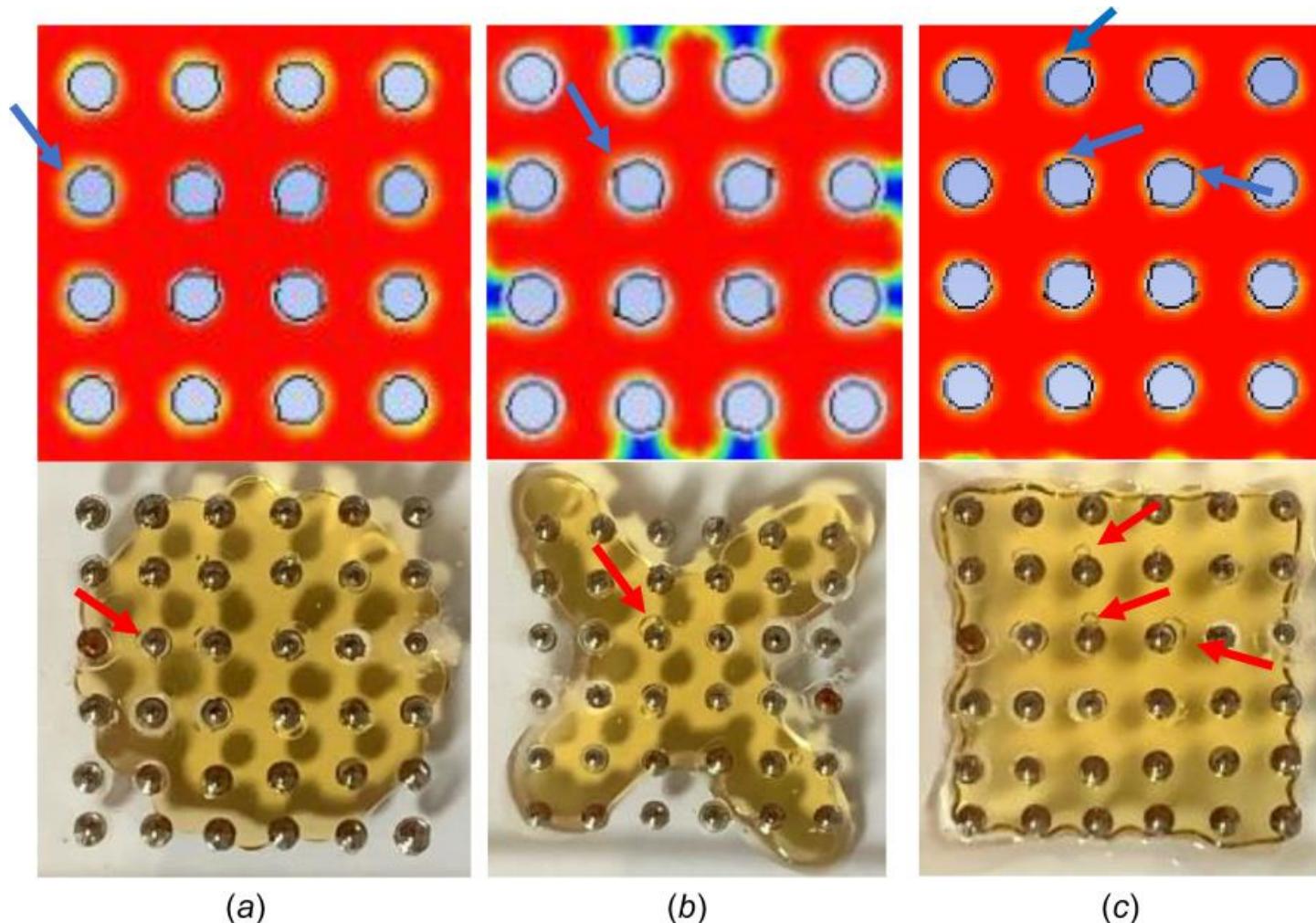


# No-Flow Underfill Technology (Cont'd)

## Typical Reflow Profile of Eutectic Sn/Pb Solder



# Dispensing patterns for no-flow underfills



Dot

Cross

Combined

# No-Flow Underfill Technology (Cont'd)

## Challenges

- **Dispensing**
  - Proper viscosity and surface tension
  - No process voiding
- **Pre-heating zone**
  - Significant decrease of viscosity
  - Easy compressing down of flip-chip
- **Soaking zone**
  - Sufficient fluxing capability
  - Maintenance of low viscosity
- **Reflow zone**
  - Maintenance of low viscosity before solder melting
  - Quick hardening after solder melting
- **Cooling zone**
  - Low shrinkage and low residual stress
  - Excellent thermal, physical, and mechanical properties

# No-Flow Underfill Technology (Cont'd)



## Generic Composition of No-Flow Underfill Material

- **Base resin**
- **Curing agent (hardener)**
- **Latent catalyst**
- **Fluxing agent**
- **Modifiers**
- **Silica filler**

# No-Flow Underfill Technology (Cont'd)

## Possible Base Epoxy Resin System

- **Epoxy Resins**
  - Cycloaliphatic epoxy resins
  - Bisphenol A type epoxy resins
  - Bisphenol F type epoxy resins
  - Naphthalene type epoxy resins
  - Novolac type epoxy resins
  - A mixture of the above epoxy resins
- **Cyanate ester resins**

# No-Flow Underfill Technology (Cont'd)

## Possible Curing Agents

- Active Hydrogen Compounds
  - Polyamines
  - Carboxylic acids and anhydrides
  - Polymercaptans
  - Polyphenols (such as novolac resins)
- Ionic Initiators
  - Tertiary amines
  - Imidazole and its derivatives
  - Onium salts
  - Phosphorus-containing compounds
  - Metal chelates
- Cross-Linkers
  - Melamine-formaldehyde resins
  - Phenol-formaldehyde resins
  - Urea-formaldehyde resins

# No-Flow Underfill Technology (Cont'd)



## Possible Latent Catalysts

- Imidazoles and their derivatives
- Quaternary phosphonium compounds
- Metal acetylacetones
- Tertiary amines
- Thermally and/or photo-sensitive onium salts

## No-Flow Underfill Technology (Cont'd)

### Possible Fluxing Agents

- **Organic carboxylic acids**
- **Polyols**
- **Resinous (polymeric) fluxes**

# No-Flow Underfill Technology (Cont'd)

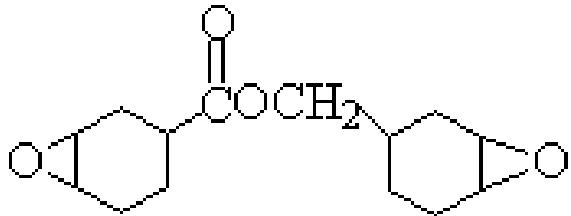
## Possible Modifiers

- **Toughening agents**
  - Reactive type
    - CTBN modified epoxy resins
    - Reactive CTBN resins
    - Rubber modified epoxy resins
    - Alicyclic acids
    - Reactive polyesters
    - Polyamide resins
    - Urethane elastomers
    - Silicone-epoxy copolymer
  - Non-reactive type
    - Silicone powder
    - Rubber powder
    - Thermoplastic powder
  - **Viscosity modifier (such as reactive diluent)**

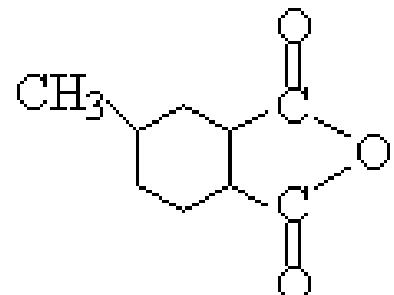
# No-Flow Underfill Technology (Cont'd)

## GT No-Flow Formulation

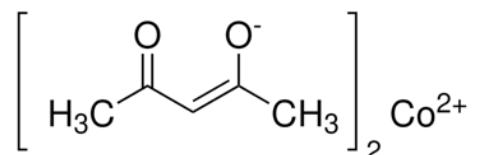
ERL 4221



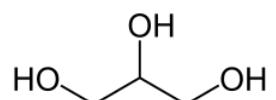
HHMPA



Co(II) Acetylacetone

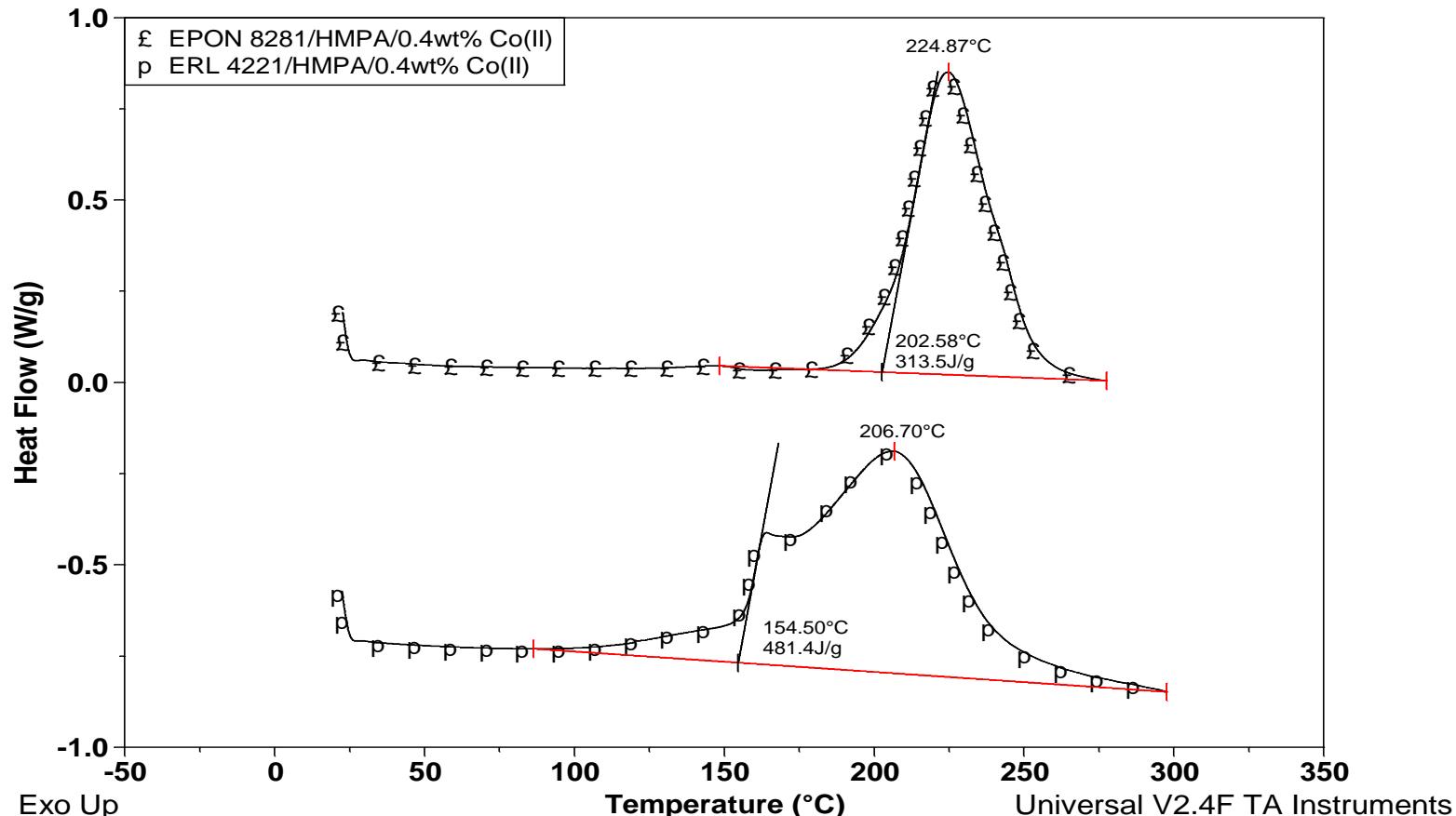


Glycerol



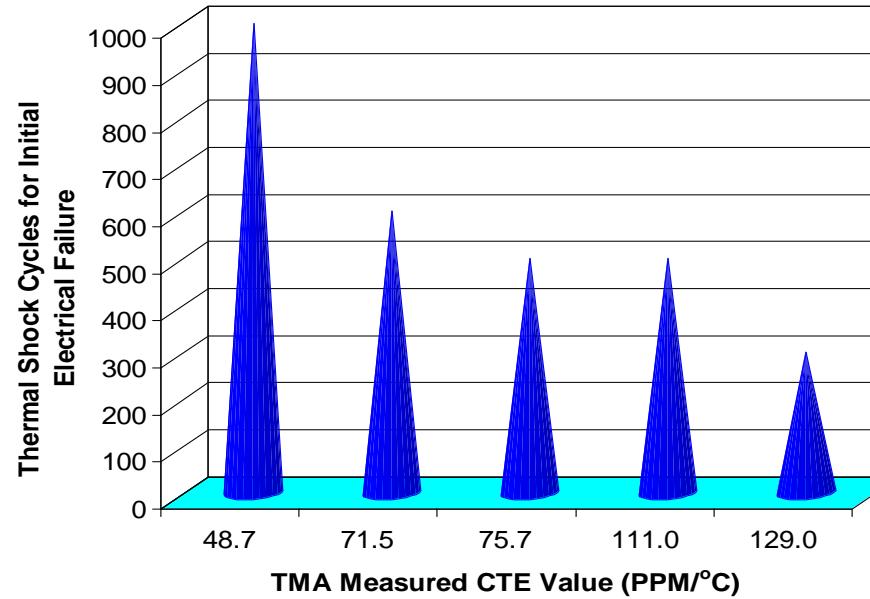
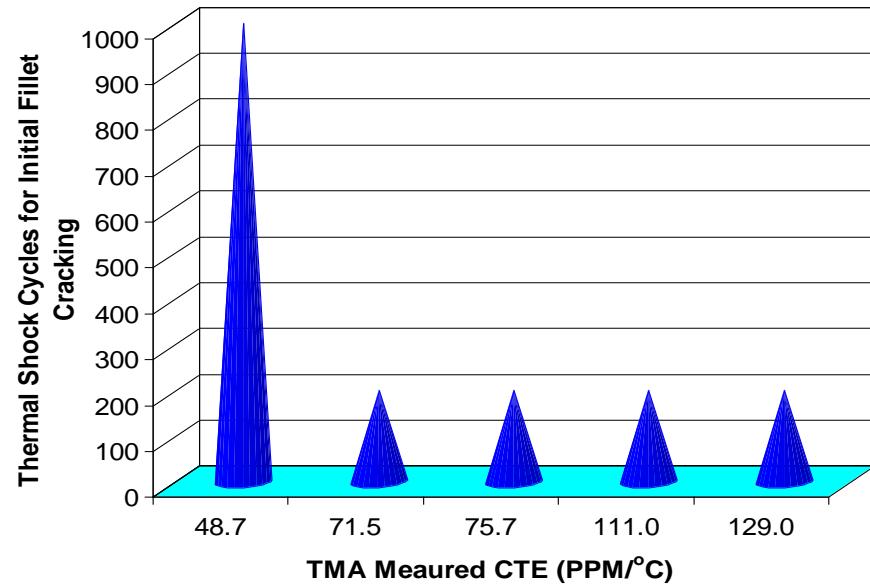
# No-Flow Underfill Technology (Cont'd)

## Curing of Co(II) Acac. in Epoxy/Anhydride systems



# No-Flow Underfill Technology (Cont'd)

## CTE Effects on Fillet Cracking and Package Reliability



# Reworkable Underfill Technology

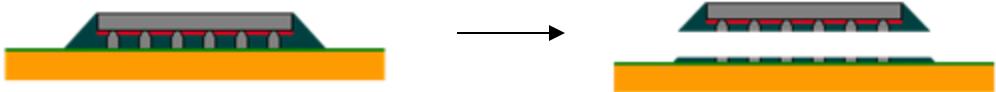
## Flip-Chip Rework

### Importance

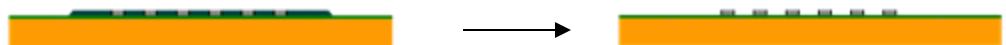
- KGD
- Cost
- Reusing boards

### Process

Localized chip removal



Underfill removal and site preparation



Solder replenish  
New chip replacement  
Underfill dispensing

# Reworkable Underfill Technology (Cont'd)

## Reworkable Underfill Process

= Capillary - Flow Underfill + Rework

### Rework Process

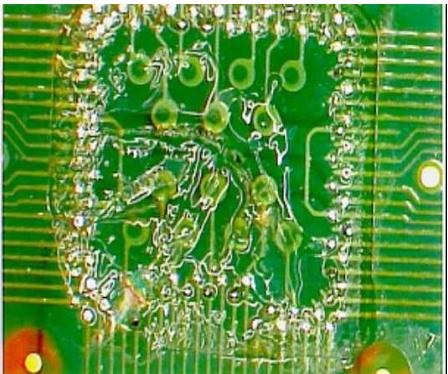
- Localized chip removal
- Underfill removal and site preparation
- Solder replenish
- New chip replacement
- Underfill dispensing

# Reworkable Underfill Technology (Cont'd)

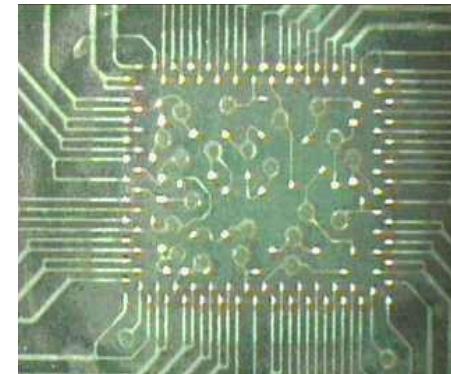
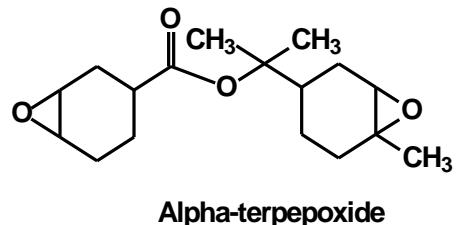
## Epoxy Based Reworkable Underfills (Cornell)

### Thermally Degradable Epoxyes

- ☺ Degrade at 200-300°C
- ☺ Allow die to be removed with tweezers on hot-plate
- ☺ Brushing to remove underfill residue



After die removal



After underfill residue removal

# Reworkable Underfill Technology (Cont'd)

## Non-Epoxy Based Reworkable Underfills

Thermoplastic (National Starch & Chemical)

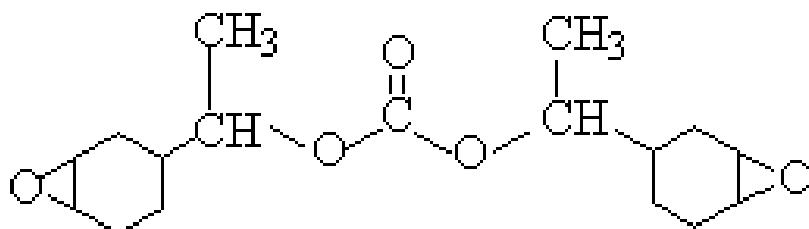
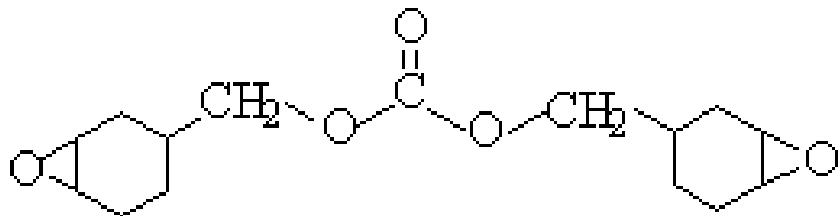
- ☺ Easy to rework (die removal & residue removal)
- ☹ Low reliability

# Reworkable Underfill Technology (Cont'd)

## Georgia Tech's Approach I

### Thermally degradable epoxies

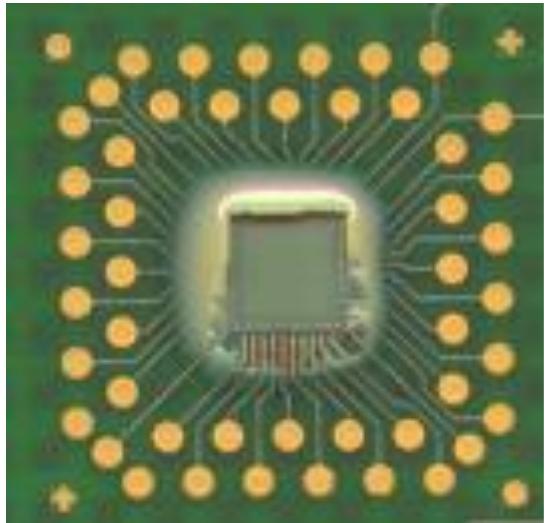
- ☺ Degrade ~ 220°C
- ☺ Have good overall properties



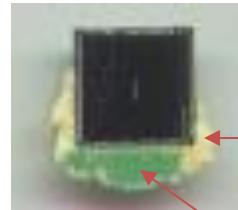
# Reworkable Underfill Technology (Cont'd)

## Rework Test

Board and Die after Rework



Board



Die

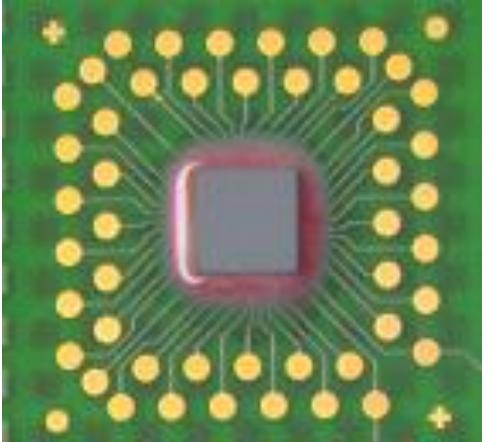
Underfill  
Solder mask

Non-Reworkable Underfill

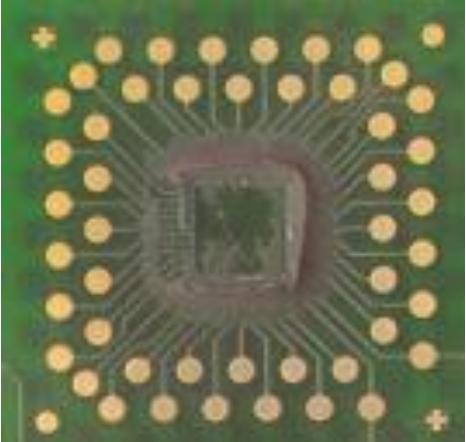
# Reworkable Underfill Technology (Cont'd)

## Rework Test

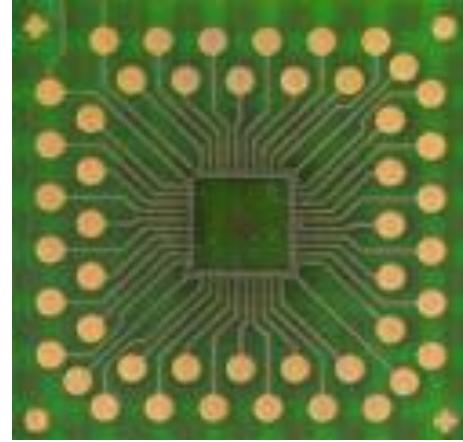
Board Site



Before rework



After die removal



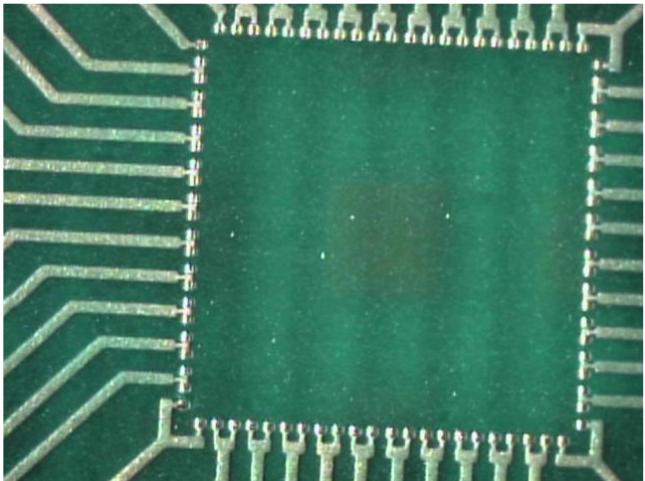
After underfill  
residue  
removal

Reworkable Underfill

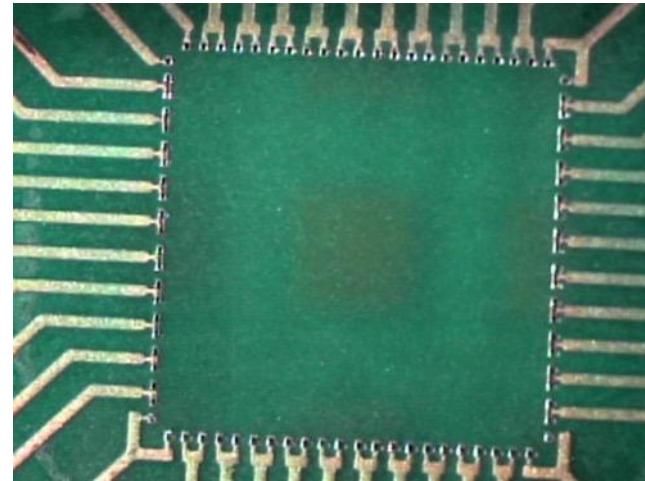
# Reworkable Underfill Technology (Cont'd)

## Rework Test

Bond Pads



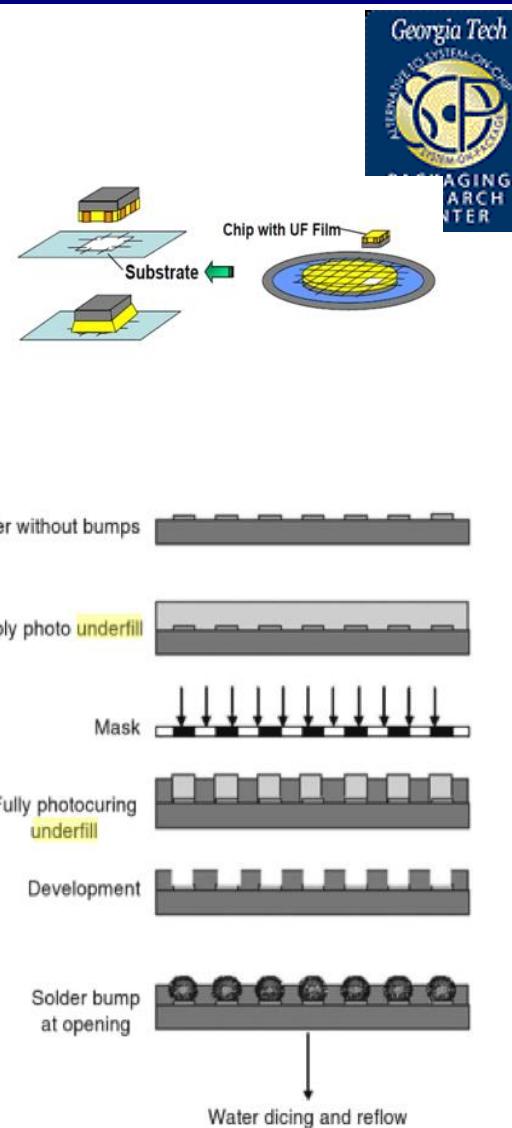
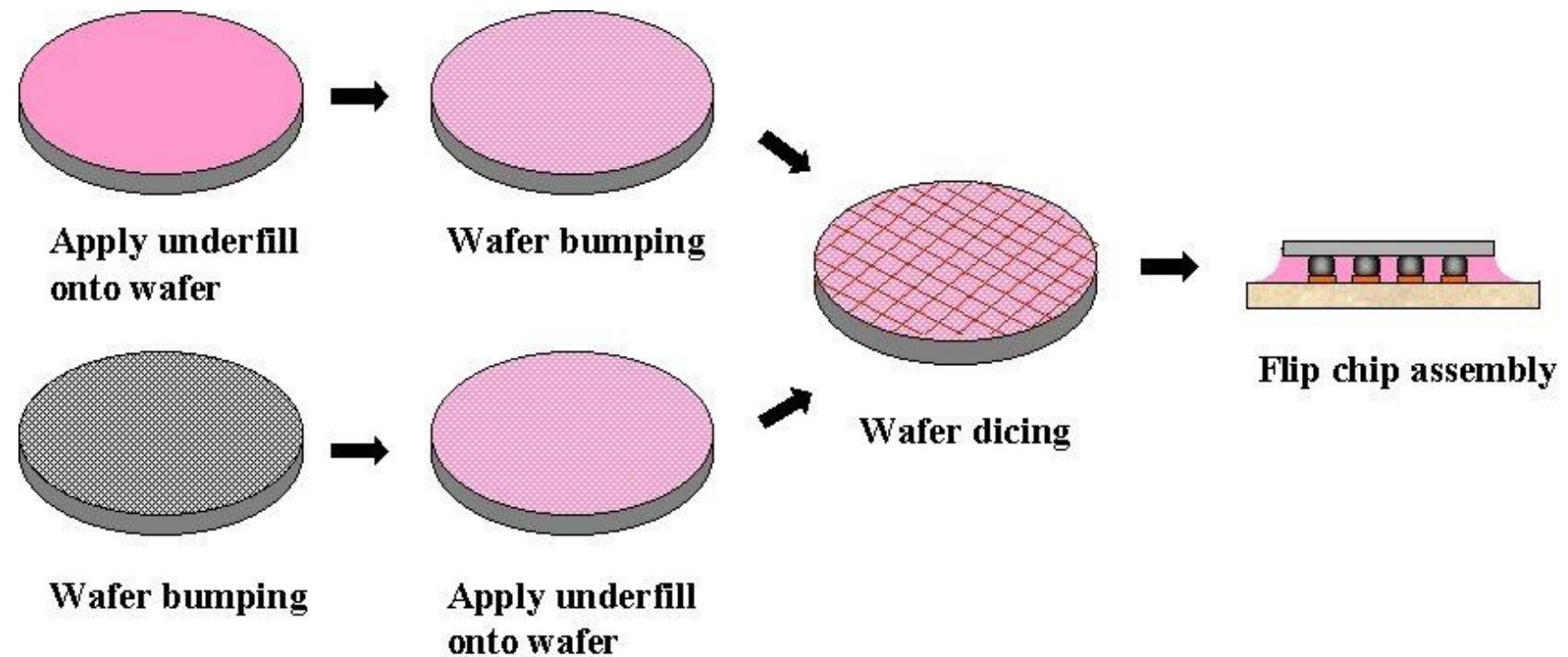
Before rework



After rework

## Reworkable Underfill

# Wafer Level Underfill



Key material property; B-staged No Flow Underfill  
024 (Patented by GT in 2002)

# Wafer Level Underfill Technology



**Underfill is applied onto the wafer before it is diced.**

## Advantages:

- Simplification process
- Burn-in and test at wafer level
- Reduction of cost, higher yields
- True SMT transparency of flip-chip

# Molded Underfill Technology

## Conventional underfilling process



## Molding process



# Molded Underfill Technology (Cont'd)



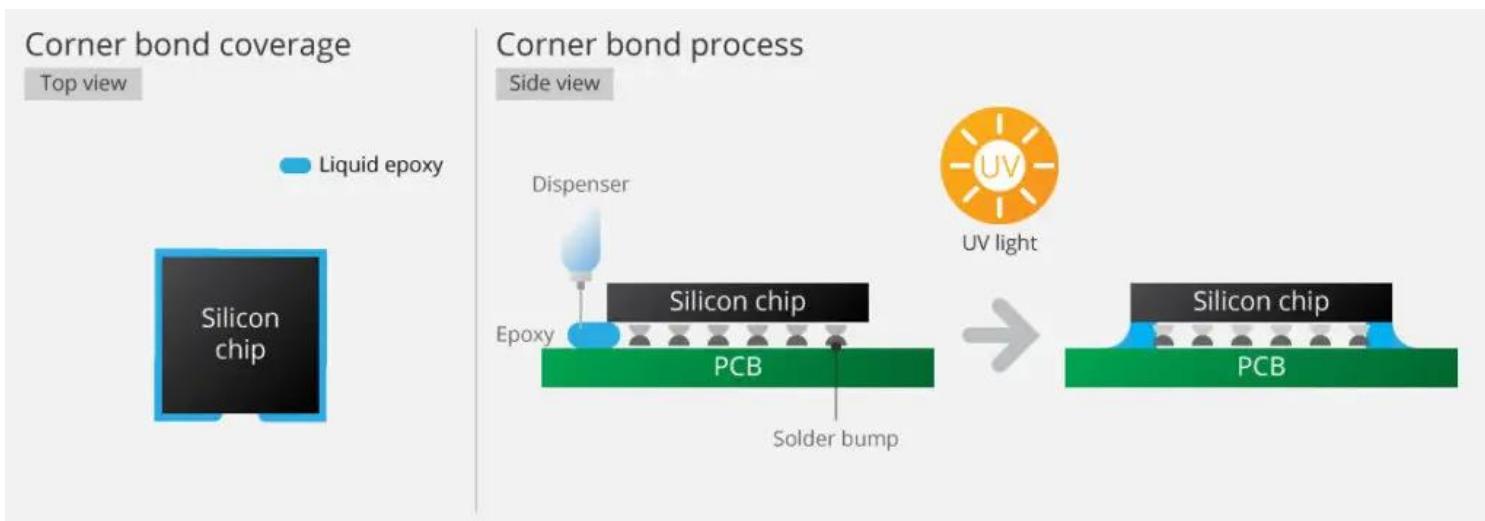
## Advantages:

- Replacing underfilling with transfer molding – high throughput
- Based on Epoxy Molding Compound – long pot-life, high reliability
- Extendable to wafer form – low cost

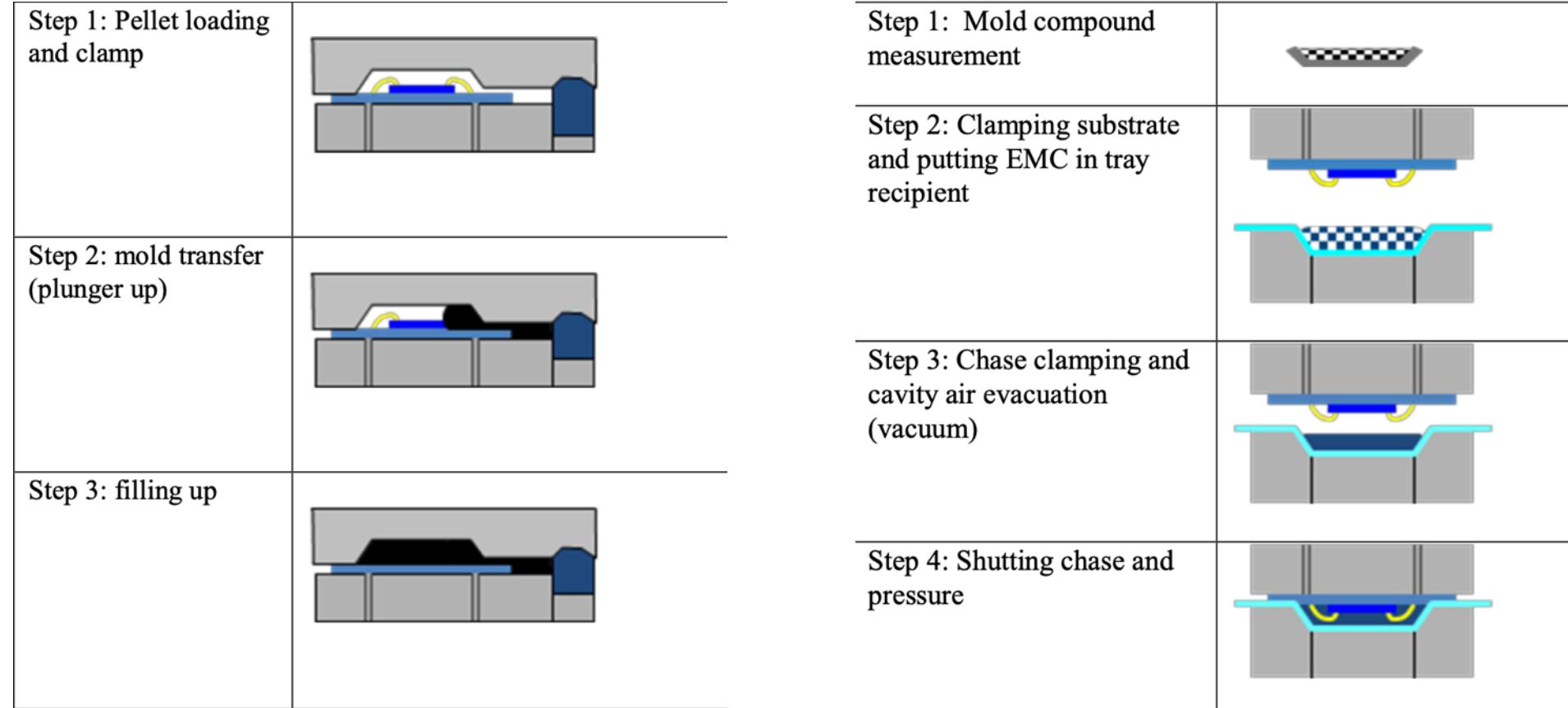
## Applications:

Flip-chip in packages (BGA, CSP)

Evolving into FOWLP



# Transfer/Compression molding



Not much flowing as in transfer molding

Board level Encapsulant

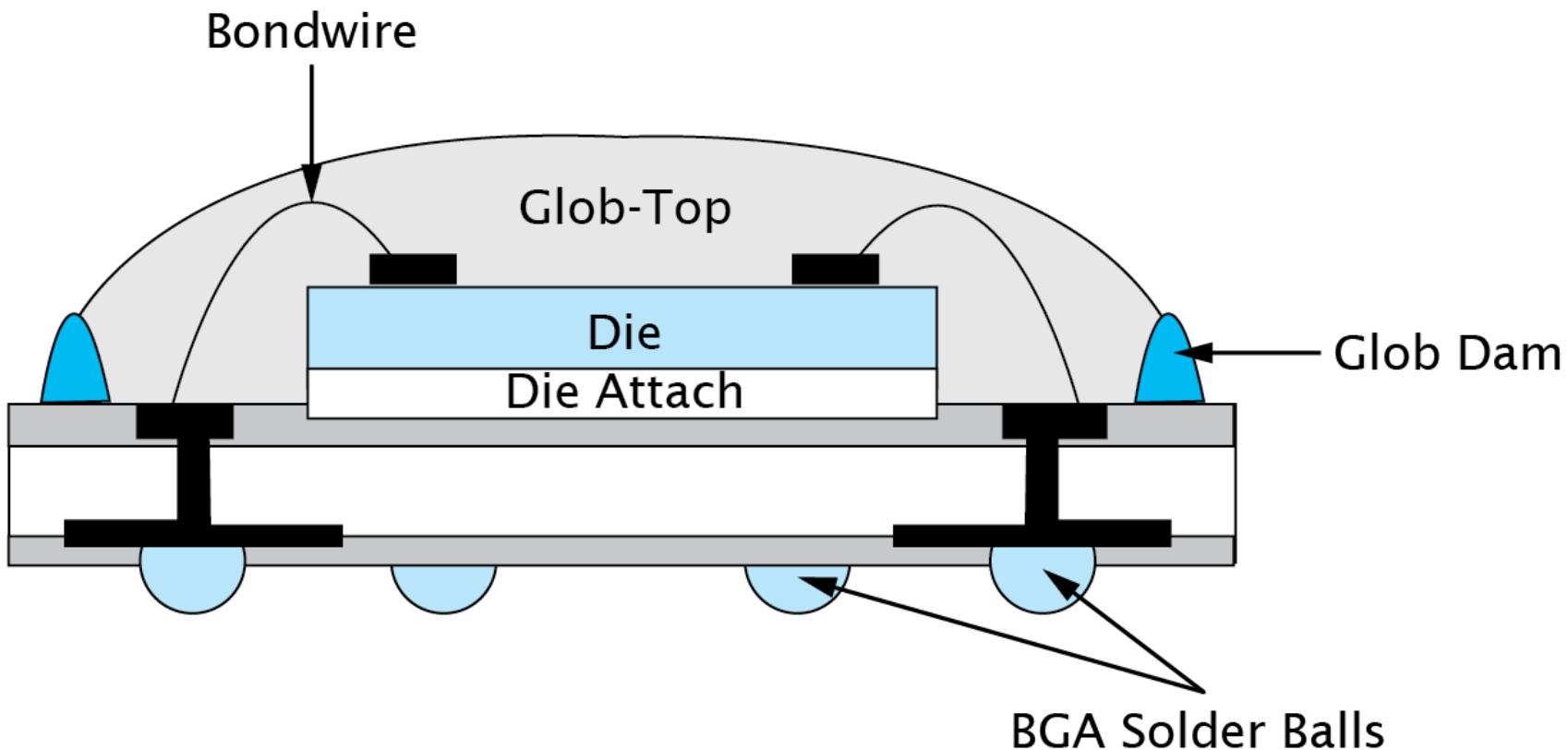
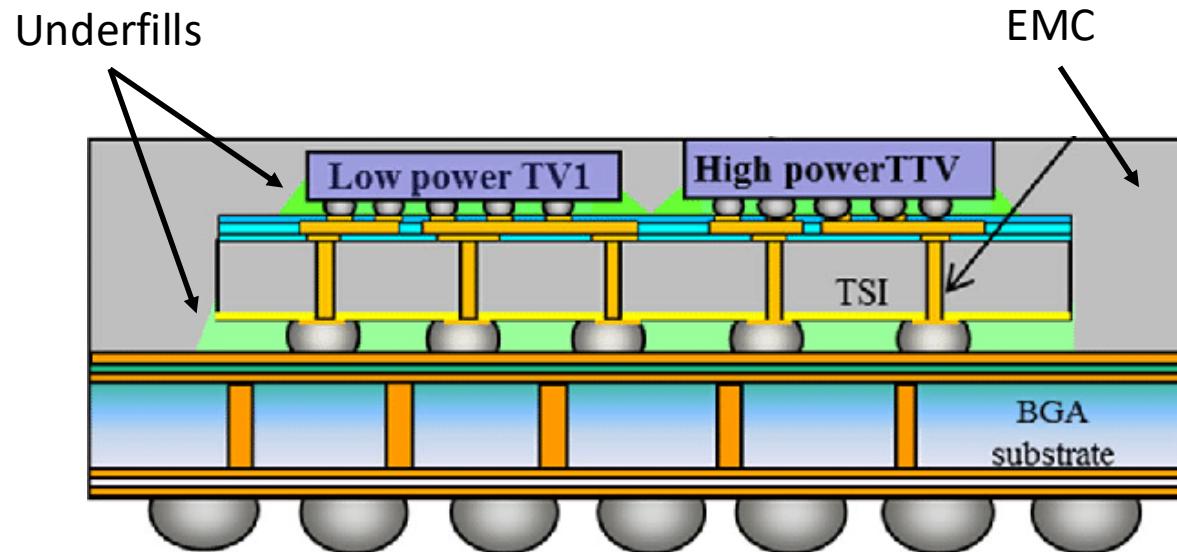
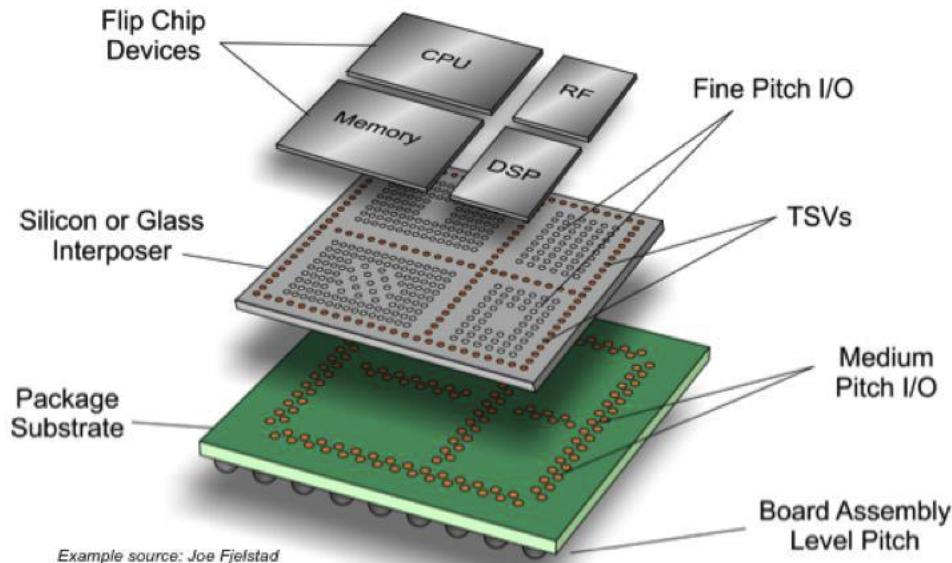


Fig 14.14 - Glob topping a BGA device



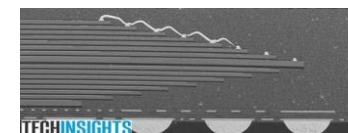
# Encapsulant Materials used in the state of the art 2.5D/3D Advanced Packaging Solutions



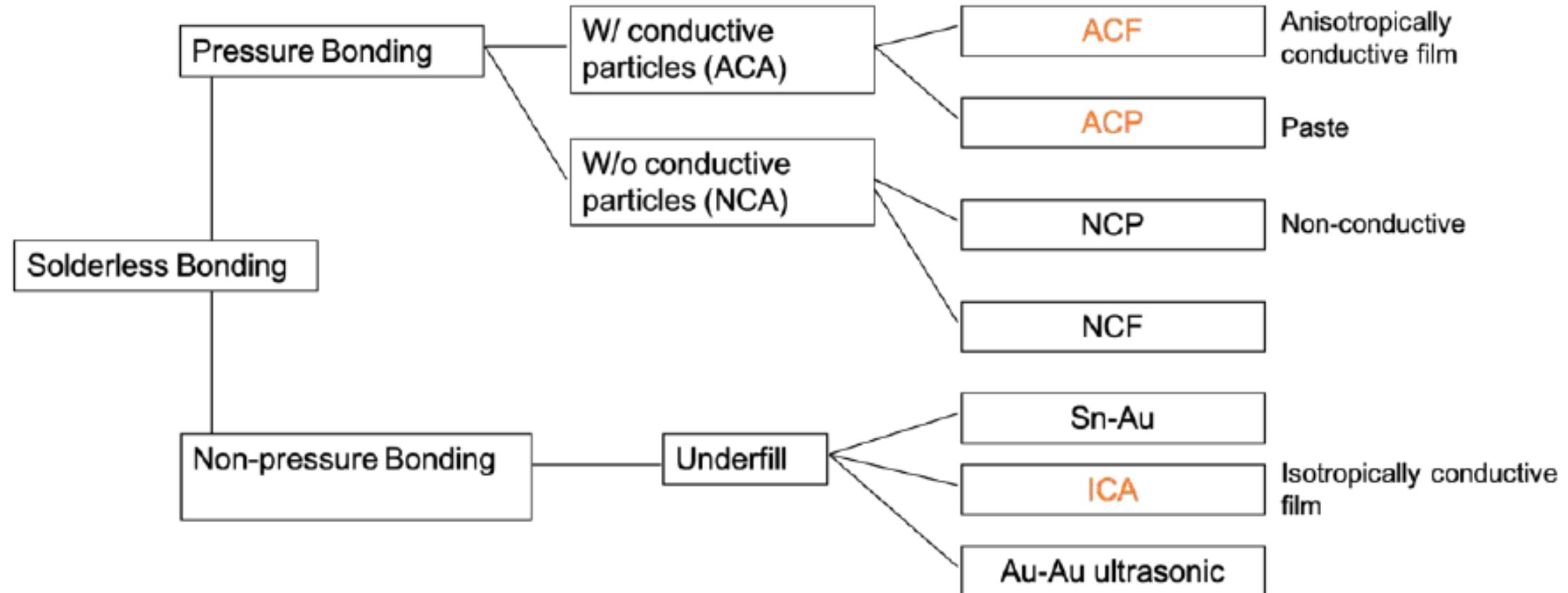
General elements of a 2.5D package (Illustration courtesy of Joseph Fjelstad)

# Encapsulation/sealing materials with specific functionality

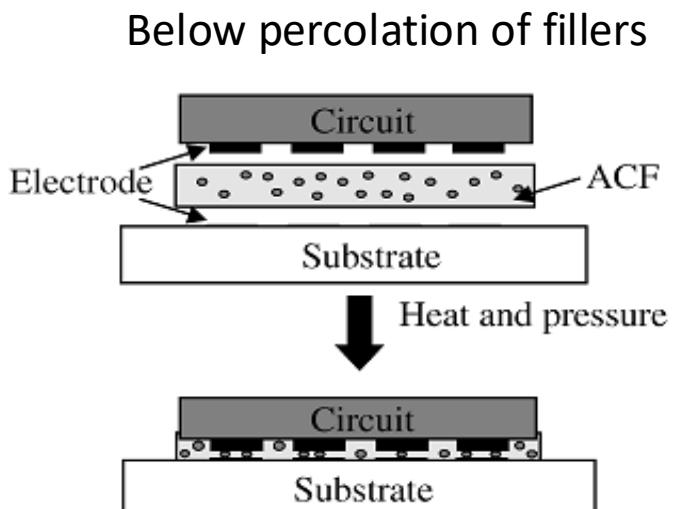
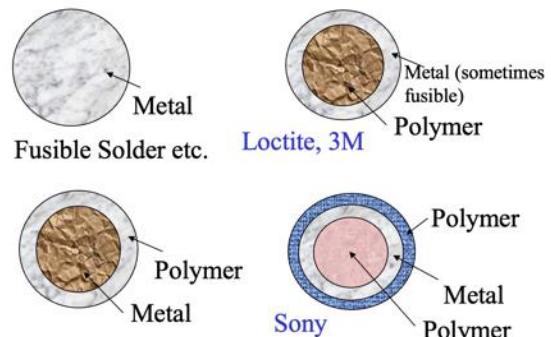
- Underfill capability with electrical interconnections
- EMC/gloptop for EMI solution
- Encapsulation with electrostatic discharge (ESD) acting like a varistor
- EMC for High thermal conductivity
- Lid bonding materials, fuel tank sealants for ESD
- Die attach film (DAF); 3D die stacking



# Encapsulation with electrical interconnection



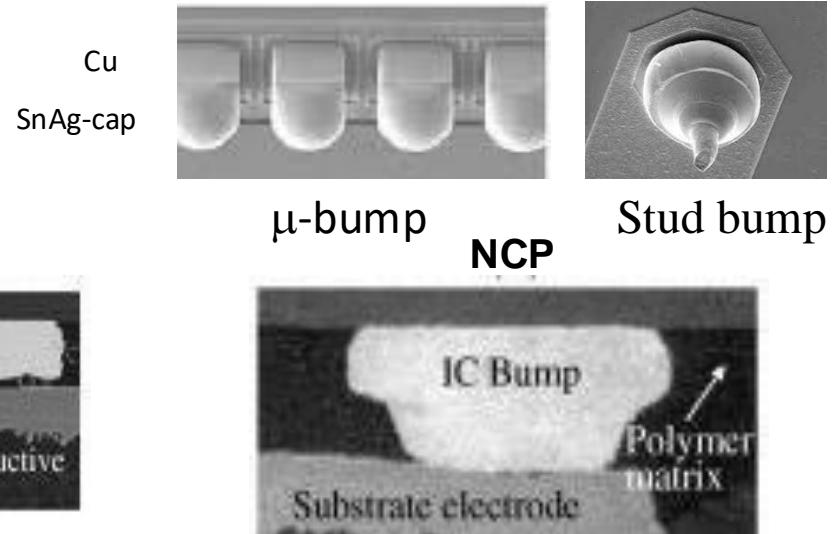
# ACF/P and NCF/P → Electrical and Mechanical bonding → UFB function



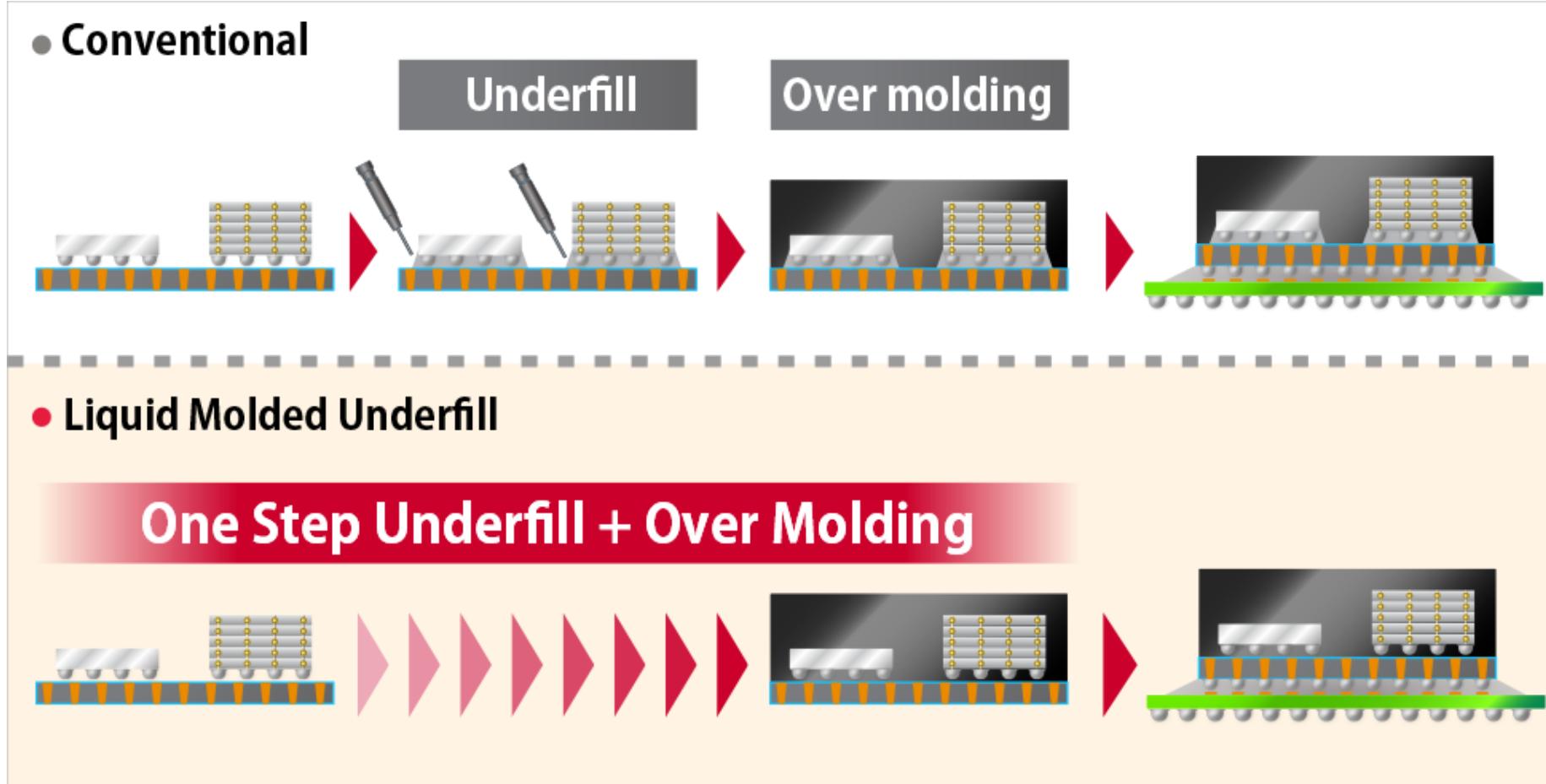
- High Tg, aromatic polymers (polydivinylbenzene, melamin, etc.)

Figure 10. Thermo-compression bonding using an ACF.

- For fine pitch COG/COF (glass/flex)
- Thermocompression bonding (TCB); pressure and temperature/ultrasound

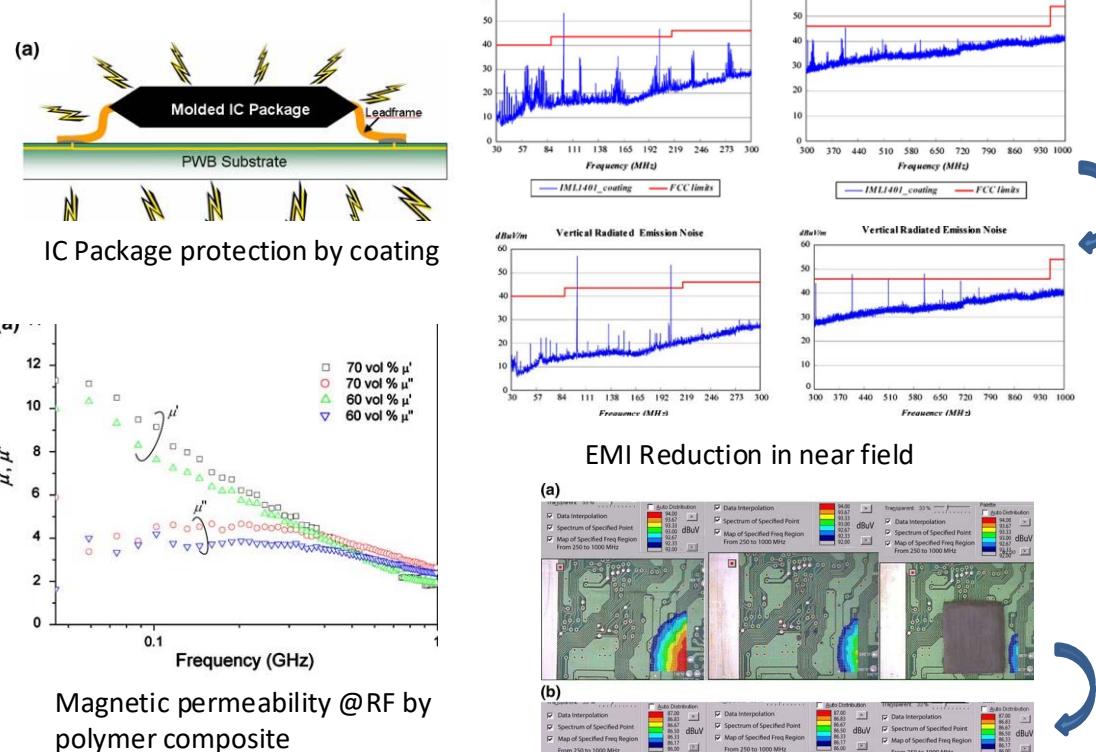


# Liquid UF for 2.5D/SiP

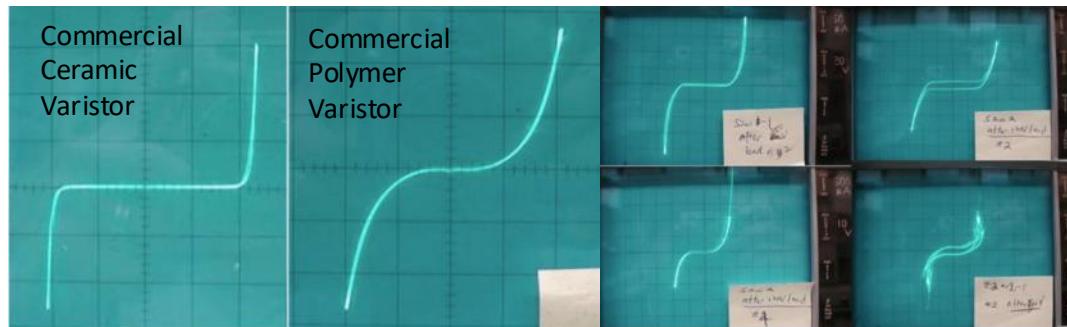
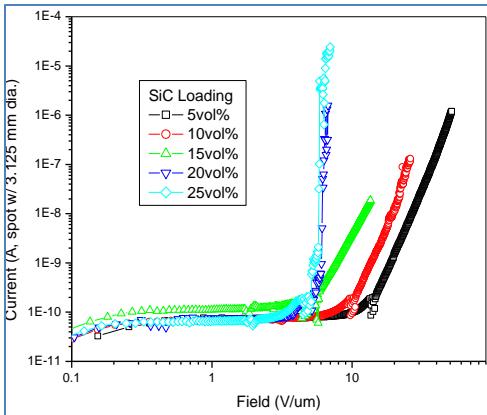


# Gloptop for EMI suppressor

- Objectives: Enhancing electromagnetic compatibility (EMC) of IC packaging
- Approach: Incorporation of soft magnetic fillers (ferrite) into polymers
- Summary: The absorber composite efficiently suppresses MW noise (H-field) in far- and near-field modes.



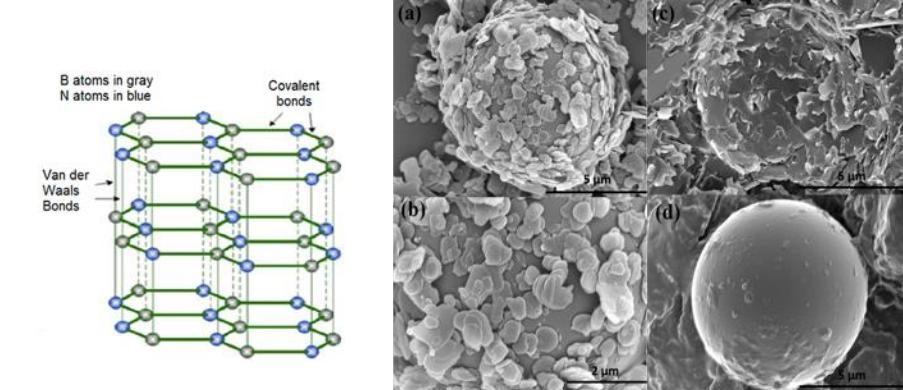
# Gloptop as varistor for IC Protection from ESD (Wafer coating), SiC, BC and metal in polymer



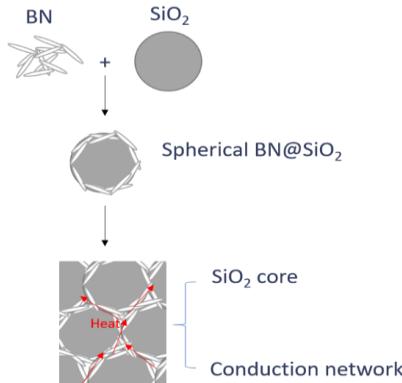
# Fillers for High thermal conductivity

- Filler interface chemistry to design molecular couplers to alleviate phonon scattering at interfaces
- High K, low Dk materials

	K(W/mK)	$\rho(\text{g}/\text{cm}^3)$	CTE (ppm/K)	Dk, 1 MHz, 20C	Cost(\$/kg)	Comment
Fused $\text{SiO}_2$	1.3	2.2	0.5	4.3	1	Main filler in EMC
$\text{Al}_2\text{O}_3$	30	3.9	8	9.6	5	Conventional for high thermal
AlN	140~180	3.26	4.5	9	40	Irregular shape
h-BN	400 (in plane)	1.9	38(out of plane) -2.72(in plane)	4.4	40-50	2-D shape, hard to mix
SiC	120	3.1	4	10	36	Low Eg
diamond	2000	3.5	0.8	5.7	>1200	Military/aerospace applications
$\text{Si}_3\text{N}_4$	10-43	2.4~3.3	1.4-3.7	9.5-10.5		
Epoxy	0.2	1.2	60	4~5	5	

hBN coated  $\text{SiO}_2$  ( $\text{BN}@\text{SiO}_2$ )

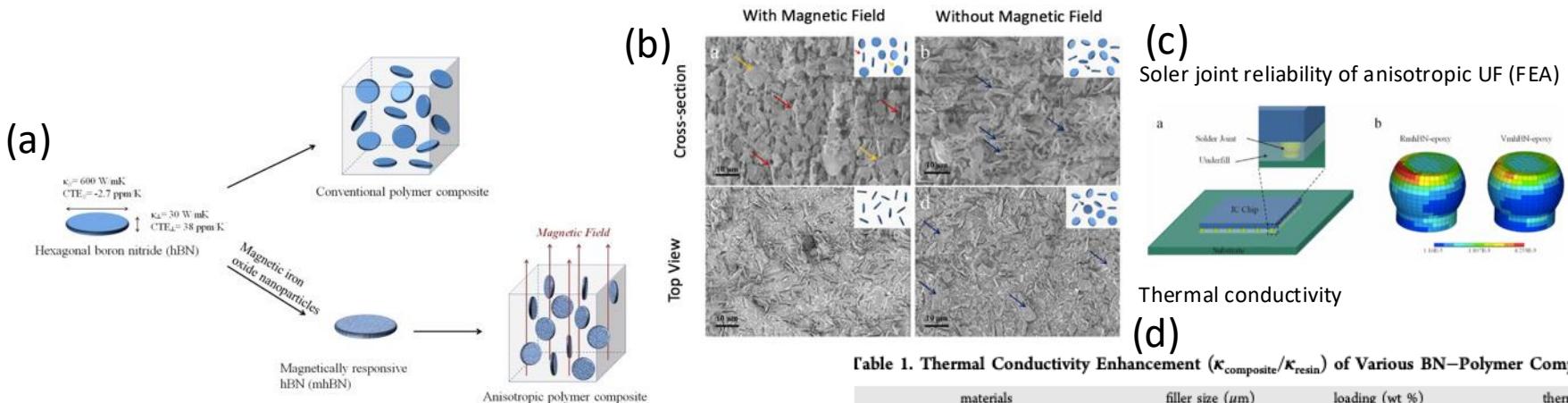
- Chemical bonding of hBN and  $\text{SiO}_2$  via respective functionalization of each
- Enhanced flowability via reduced surface area
- $\text{SiO}_2$  core as co-filler for cost and CTE control
- Phonon conduction path via in-plane hBN



J. Li, K. Moon, CP Wong, et al, ECTC, 2020

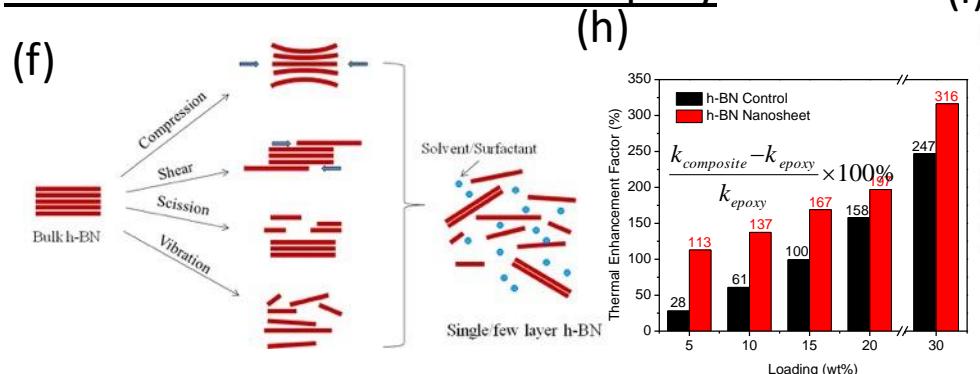
# High thermal K Epoxy-cont'd

## Magnetic Alignment of h-BN in Polymer Matrix/Underfill

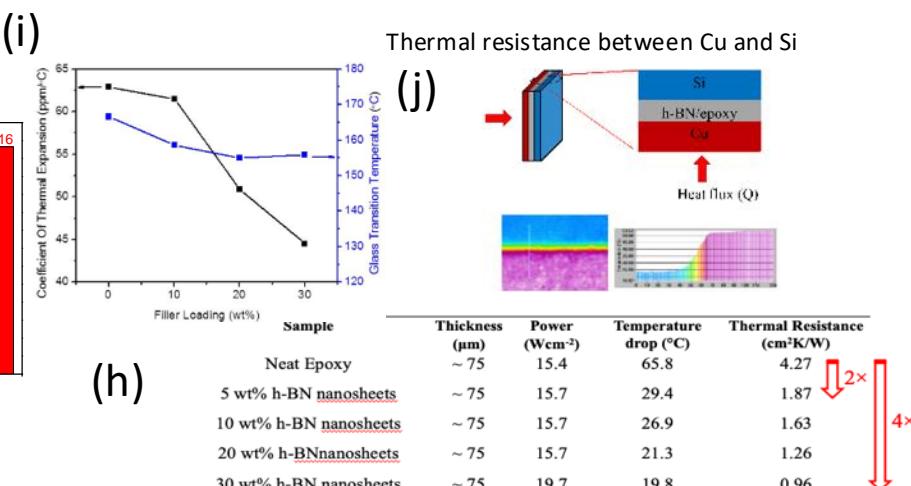


Z Lin, Y Liu, S Raghavan, K Moon, SK Sitaraman, C Wong ACS applied materials & interfaces 5 (15), 7633-7640, 2013.

## h-BN exfoliated nanosheets in epoxy



Z. Lin, K Moon, CP Wong, Composites Science and Technology 90, 123-128, 2014.

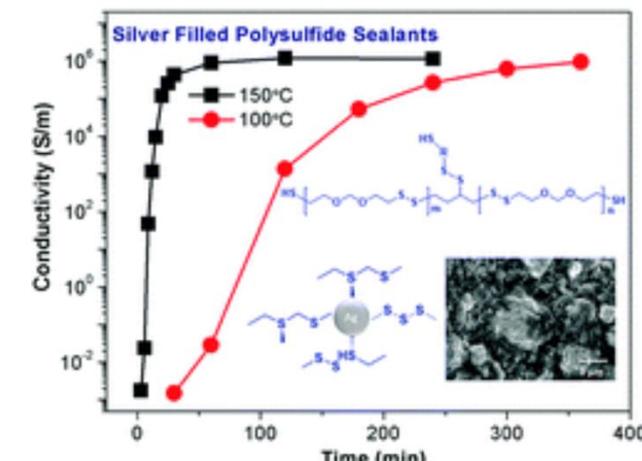


# Polysulfide sealants

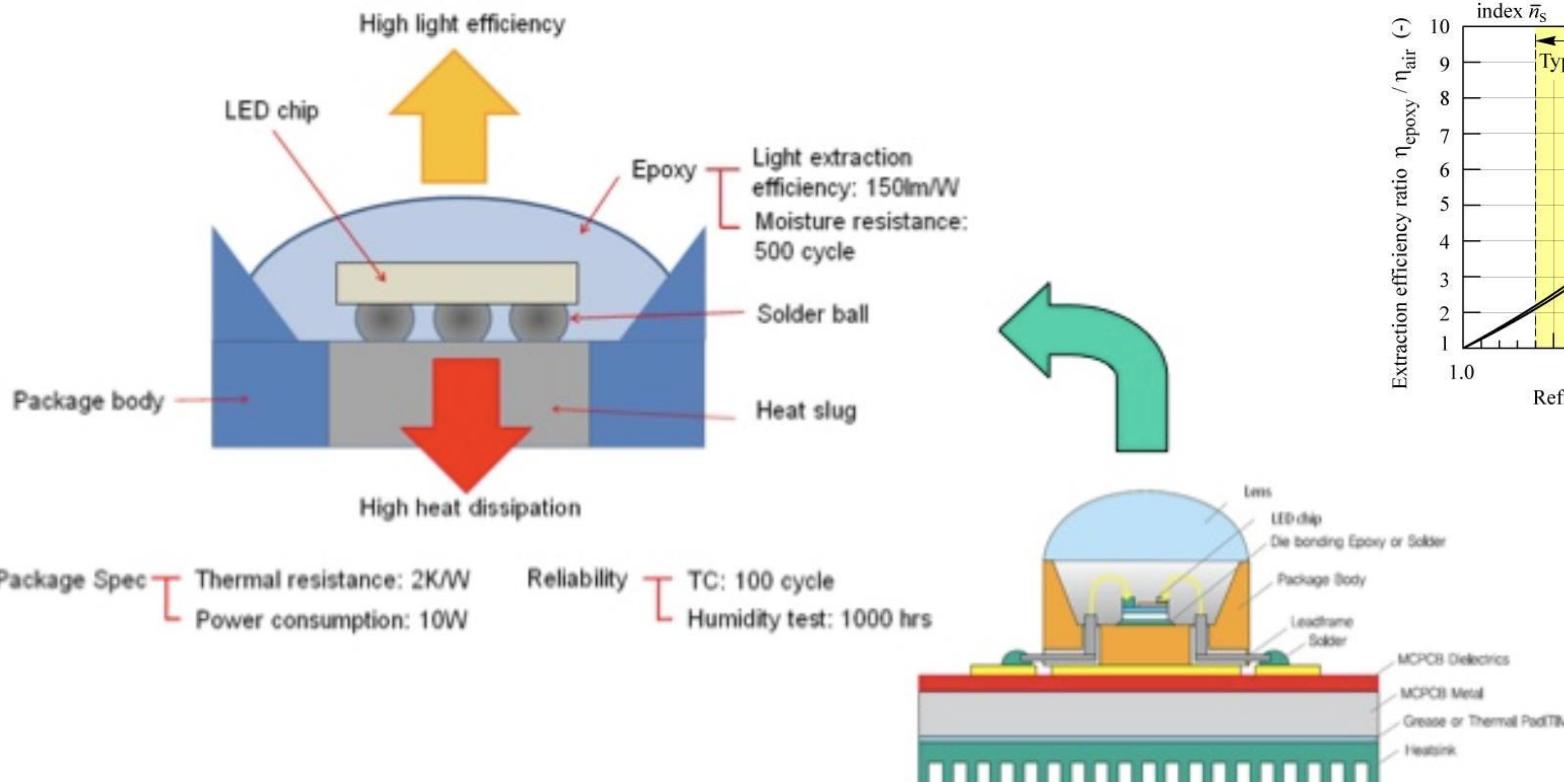
- PS containing ~37% sulfur content demonstrates exceptional resistance to gasoline, fuels, and aviation gas, superior to epoxy, polyurethane, and polyacrylates
- To dissipate static charges
- Unique filler–resin interactions caused low electrical conductivity

<https://doi.org/10.1039/C8SM02004C>

Processing and characterization of silver-filled conductive polysulfide sealants for aerospace applications



# HB-LED encapsulant for light extraction



<https://shorturl.at/jpDO9>

High Refractive Index and Transparent Nanocomposites as Encapsulant for High Brightness LED Packaging

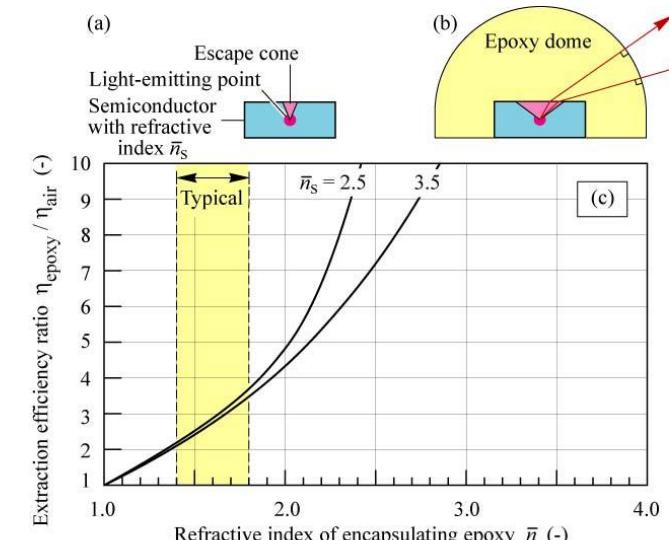
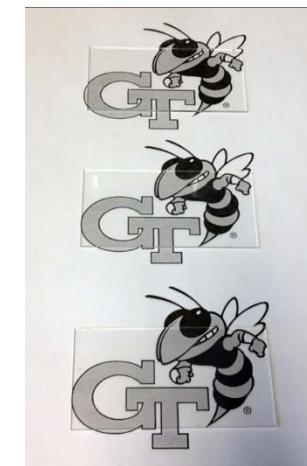
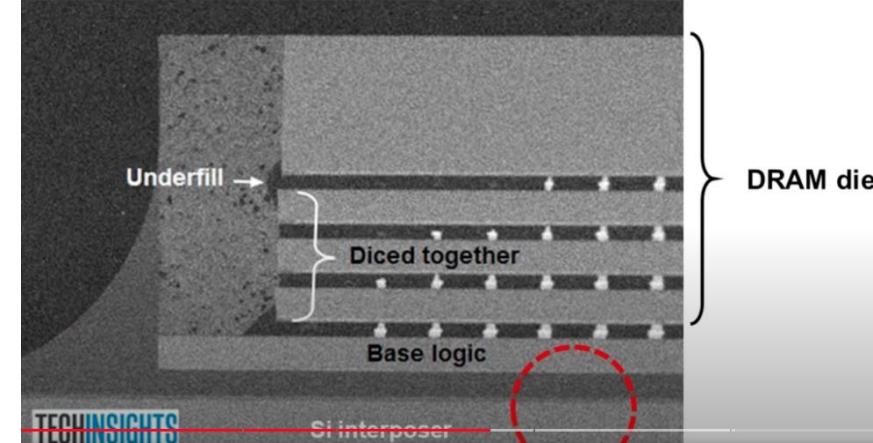
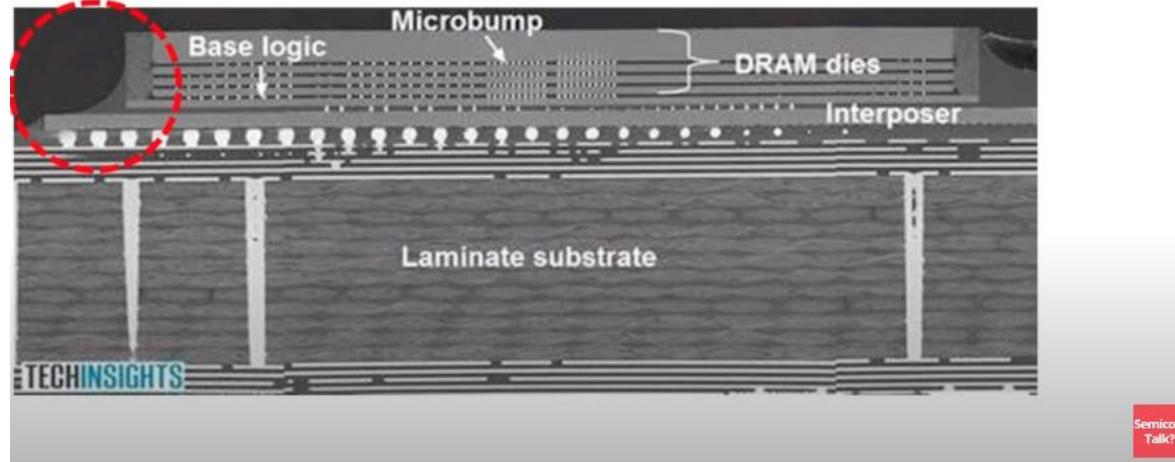


Fig. 5.6. (a) LED without and (b) with dome-shaped epoxy encapsulant. A larger escape angle is obtained for the LED with an epoxy dome. (c) Calculated ratio of light extraction efficiency emitted through the top surface of a planar LED with and without an epoxy dome. The refractive indices of typical epoxies range between 1.4 and 1.8 (adopted from Nuese *et al.*, 1969).

E. F. Schubert  
*Light-Emitting Diodes* (Cambridge Univ. Press)  
www.LightEmittingDiodes.org



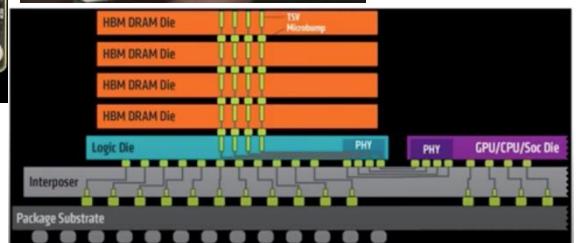
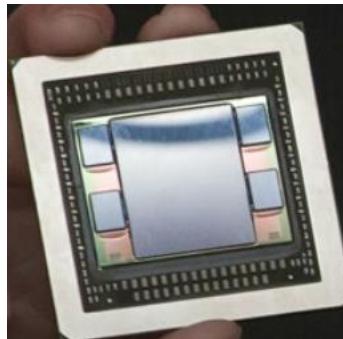
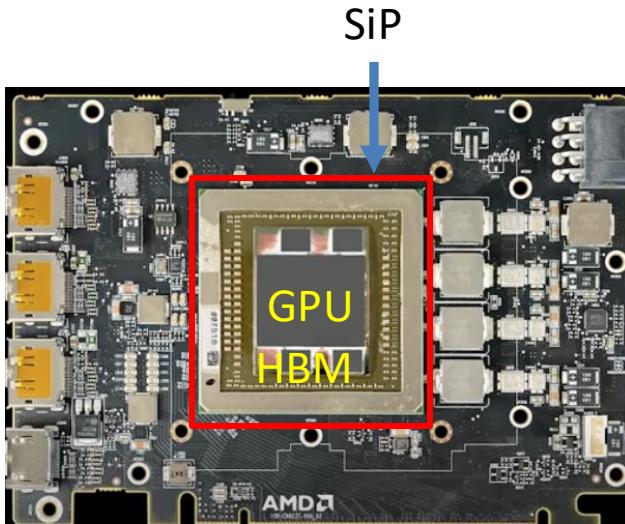
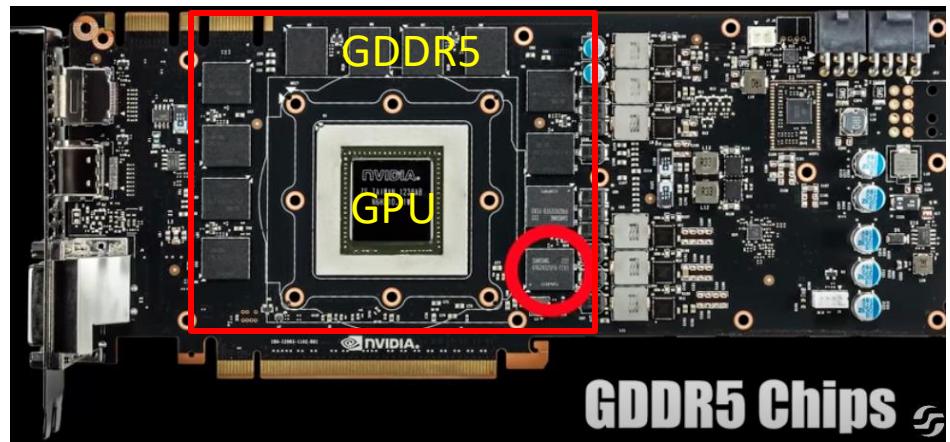
# Encapsulation in HBM-SiP



- In die stacking;
  - Materials; DAF, NCP and UF
  - Assembly; thermocompression bonding (TCB)

# GDDR vs HBM-SiP

- Graphic DDR (video memory); separate packages between GPU and DRAMs
- High Band Width; unified package into SiP

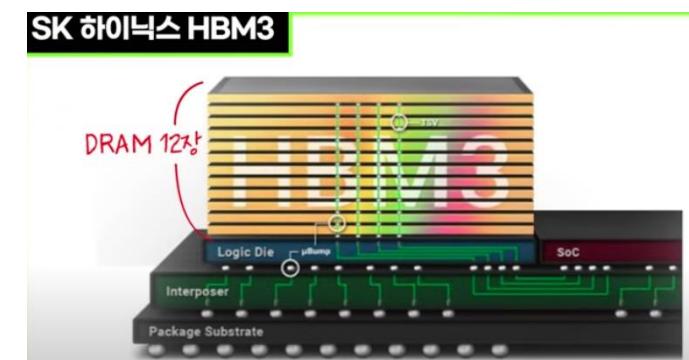
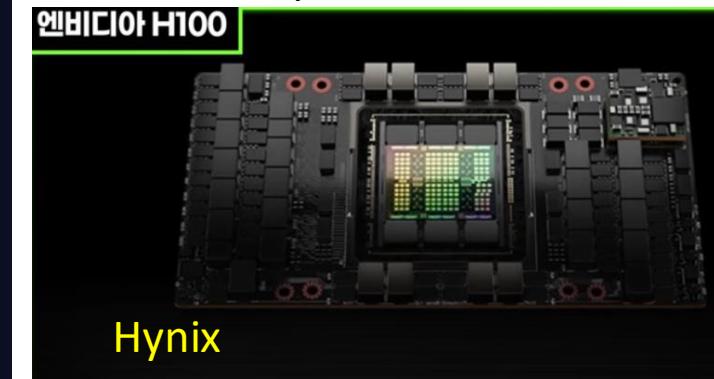


# GDDR6 vs HBM

	<b>GDDR6</b>	<b>HBM3</b>
대역폭	64GB/s	819.2GB/s
핀 수	32K	1024K
용량	2GB	24GB
속도(핀 당)	16GB/s	6.4GB/s



Samsung HBM3 found in servers ;  
Aurora, ElCapitan



Speed; Intentionally reduced for mitigating thermal and power issues

# Decapsulation/Rework

- Decapsulation
  - Failure analyses (to remove encapsulants-top mold, overmold, fillets)
  - Chemical; acids (concentrated/fuming nitric, sulfuric acids, etc), to selectively etching (Al, Cu, Ag, etc.)
  - Physical; laser ablation, CF4-RIE, Plasma (microwave induced), etc.
- Rework
  - Reworkable underfills
    - Process development/optimization
    - Expensive substrates



Kim, Moon, et al., IEEE Trans. CPMT, 9(8), Aug. 2019.

124

# Summary

- Early days for microelectronics packaging dominated by Hermetic sealing at device level by inorganic thin-films such as silicon oxide and nitride and at package level by metal packages
- Early polymers were incapable of dealing with moisture, thermomechanical properties (CTE, modulus), adhesion, impurity level etc.
- Evolution of Resin chemistry, filler loading technologies and interface engineering enabled near hermetic plastic packaging to meet the requirements for device and system level applications.
- Consequently, polymer encapsulation or molded packaging has established itself as the primary high-volume and low-cost approach and is expected to continue to grow across all but highly reliable space, deep sea, and high-temperature automotive applications.
- Demands on high performance and versatility from polymer encapsulants/sealants and one of key components to enable advanced packaging
  - High temperature sustaining (>250C)
  - High thermal dissipation
  - Tunable electric/dielectric/magnetic properties