A mathematical modeling toolbox for ion channels and transporters across cell membranes

Shadi Zaheria, Fatemeh Hassanipoura,*

^aDepartment of Mechanical Engineering, The University of Texas at Dallas, Richardson, TX, 75080, USA

- The following supplementary material is from " A mathematical modeling toolbox for ion channels
- 2 and transporters across cell membranes" manuscript. It contains an overview of all equations
- 3 related to Ion channels, Pumps, Cotransporters, and Symporters, organized in a table form. The
- 4 detailed transporters along with the descriptions of their equatuons can be found from here.

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^{*}Corresponding author

32 3.3. Sodium chloride cotransporter

Sodium chloride cotransporter	Ref
	[34, 54]
$J_{Na,NCC}^{M,N(net)} = [E]_{t} \left(\frac{(g_{ENaCl}^{M} Na'^{M} Cl^{M})(g_{E}^{N}) - (g_{ENaCl}^{N} Na'^{N} Cl^{N})(g_{E}^{M})}{R_{M} R_{NN} + R_{N} R_{MM}} \right) $ (134)	a)
$J_{Cl,NCC}^{M,N(net)} = [E]_t \left(\frac{(g_{ENaCl}^M Na'^M Cl^M)(g_E^N) - (g_{ENaCl}^N Na'^N Cl^N)(g_E^M)}{R_M R_{NN} + R_N R_{MM}} \right) $ (134)	b)
Where	
$[E]_t = [E]_M + [ECl]_M + [ENa]_M + [ENaCl]_M + [ENaCl]_N + [ENa]_N + [ECl]_N + [E]_M + [ECl]_M + [ECl]_$	l_N
$Na^{M} = \frac{[Na]_{M}}{K_{Na}^{M}} Na'^{M} = \frac{[Na]_{M}}{K_{ClNa}^{M}} Cl^{M} = \frac{[Cl]_{M}}{K_{Cl}^{M}} Na^{N} = \frac{[Na]_{N}}{K_{Na}^{N}} Na'^{N} = \frac{[Na]_{N}}{K_{ClNa}^{N}} Cl^{N} = \frac{[Cl]_{N}}{K_{Cl}^{N}}$ $R_{M} = (1 + Na^{M} + Cl^{M} + Na'^{M}Cl^{M}) R_{N} = (1 + Na^{N} + Cl^{N} + Na'^{N}Cl^{N})$ $R_{M} = a^{M} + a^{M} - Na'^{M}Cl^{M} R_{MM} = a^{N} + a^{N} - Na'^{N}Cl^{N}$	
$R_{MM} = (1 + Na^{M} + Cl^{M} + Na^{M}Cl^{M}) R_{N} = (1 + Na^{N} + Cl^{N} + Na^{M}Cl^{N})$ $R_{MM} = g_{E}^{M} + g_{ENaCl}^{M}Na'^{M}Cl^{M} R_{NN} = g_{E}^{N} + g_{ENaCl}^{N}Na'^{N}Cl^{N}$	

Table 21: The corresponding equations describing the flux transported via sodium chloride symporters across the cell membrane