

A mathematical modeling toolbox for ion channels and transporters across cell membranes

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1 The following supplementary material is from " [A mathematical modeling toolbox for ion channels](#)
2 [and transporters across cell membranes](#)" manuscript. It contains an overview of all equations
3 related to Ion channels, Pumps, Cotransporters, and Symporters, organized in a table form. The
4 detailed transporters along with the descriptions of their equations can be found from [here](#).

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37 4. Antiporters (Exchangers) model

38 4.1. Chloride Bicarbonate Antiporter (Cl/HCO₃)

Chloride Bicarbonate Antiporter (Cl/HCO3)		Ref
<div> $\mathbf{J}_{\text{Cl,BCE}} = E_t \frac{[g_{\text{EHC}O_3}^M g_{\text{ECI}}^N \text{HCO}_3^M(\text{Cl}^N) - g_{\text{EHC}O_3}^N g_{\text{ECI}}^M \text{HCO}_3^N(\text{Cl}^M)]}{R_M R_{00} + R_N R_{ee}}$ </div> <div> $\mathbf{J}_{\text{HCO}_3,\text{BCE}} = E_t \frac{[g_{\text{EHC}O_3}^M g_{\text{ECI}}^N (\text{Cl}^M \text{HCO}_3^N) - g_{\text{EHC}O_3}^N g_{\text{ECI}}^M (\text{Cl}^N \text{HCO}_3^M)]}{R_M R_{00} + R_N R_{ee}}$ </div> <div> <p>where</p> <p>$[E]_t = [E]_M + [\text{EHC}O_3]_M + [\text{ECI}]_M + [E]_N + [\text{EHC}O_3]_N + [\text{ECI}]_N$</p> <p>$\text{HCO}_3^M = \frac{[\text{HCO}_3]_M}{K_{\text{HCO}_3}^M}, \text{Cl}^M = \frac{[\text{Cl}]_M}{K_{\text{Cl}}^M} \mid \text{HCO}_3^N = \frac{[\text{HCO}_3]_N}{K_{\text{HCO}_3}^N}, \text{Cl}^N = \frac{[\text{Cl}]_N}{K_{\text{Cl}}^N}$</p> <p>$R_M = 1 + \text{HCO}_3^M + \text{Cl}^M \mid R_N = 1 + \text{HCO}_3^N + \text{Cl}^N$</p> <p>$R_{MM} = g_{\text{EHC}O_3}^M \text{HCO}_3^M + g_{\text{ECI}}^M \text{Cl}^M \mid R_{NN} = g_{\text{EHC}O_3}^N \text{HCO}_3^N + g_{\text{ECI}}^N \text{Cl}^N$</p> </div>		[34, 59]
<div> $J_{\text{BCE}} = P_{\text{BCE}} \frac{[\text{Cl}]_i [\text{HCO}_3]_c - [\text{Cl}]_c [\text{HCO}_3]_i}{K_{\text{Cl}} K_{\text{HCO}_3} \left(\left(1 + \frac{[\text{Cl}]_c}{K_{\text{Cl}}} + \frac{[\text{HCO}_3]_c}{K_{\text{HCO}_3}} \right) \left(\frac{[\text{Cl}]_i}{K_{\text{Cl}}} + \frac{[\text{HCO}_3]_i}{K_{\text{HCO}_3}} \right) \right.}$ $\left. \left. \left(1 + \frac{[\text{Cl}]_i}{K_{\text{Cl}}} + \frac{[\text{HCO}_3]_i}{K_{\text{HCO}_3}} \right) \left(\frac{[\text{Cl}]_c}{K_{\text{Cl}}} + \frac{[\text{HCO}_3]_c}{K_{\text{HCO}_3}} \right) \right) \right)$ </div>		[34, 38]
Basolateral		[6, 60]
<div> $J_{\text{BCE}} = n_{\text{BCE}}'' \frac{k_{\text{Cl}}^+ k_{\text{HCO}_3}^+ [\text{Cl}]_{\text{M}(e)} [\text{HCO}_3]_{\text{N}(i)} - k_{\text{Cl}}^- k_{\text{HCO}_3}^- [\text{Cl}]_{\text{N}(i)} [\text{HCO}_3]_{\text{M}(e)}}{k_{\text{Cl}}^+ [\text{Cl}]_{\text{M}(e)} + k_{\text{HCO}_3}^+ [\text{HCO}_3]_{\text{N}(i)} + k_{\text{HCO}_3}^- [\text{HCO}_3]_{\text{M}(e)} + k_{\text{Cl}}^- [\text{Cl}]_{\text{N}(i)}}$ </div>		(145)

Table 27: The corresponding equations describing the flux and current transported via chloride bicarbonate antiporter (exchangers) across the cell membrane