Lecture Notes: 1 Feb, 2012

Stoichiometric matrix

The stoichiometric matrix corresponds to adjacency matrix (graph theory). All kinetic laws can be expressed as

$$\frac{dx}{dt} = Sv$$

We have: m reactions and n molecular species

$$x = (x_1 x_2 ... x_n)^T \in R^n$$

 $v = (v_1 v_2 ... v_n)^T \in R^m$

typical example: $A + B \rightarrow C, C \rightarrow A + B$

$$\frac{d}{dt}[A] = k_2[C] - k_1[A][B]$$

$$\frac{d}{dt}[B] = k_2[C] - k_1[A][B]$$

$$\frac{d}{dt}[C] = k_1[A][B] - k_2[C]$$

Singular value decomposition

Let A be an mxn matrix. Then, there exist orthogonal matrices U and V and a matrix Σ such that $A = U\Sigma V^T$

Steady states

A steady state is a state x of the syste for which

$$\frac{dx}{dt} = Sv = 0$$

Solving for $v = (V_1 V_2 ... V_m)^T$ will give us the possible reaction rates at equilibrium

Matlab code

% Solving linear systems in Matlab
A = [1 2, 0 1]
Y = [1, 2]

% If we want to solve the equation Ax = y, we do... $x=A \setminus y$

 $S = [-1 \ 1 \ -1, \ 1 \ -1 \ 0, \ 0 \ 0 \ 1]$ v = $S \times (3,1) \%$ uh oh...singular matrix null(S)

We want to describe all possible solutions of Sv = 0 (equation *). It turns out that ever solution v of * can be written as

$$v = c_1 v_1 + c_2 v_2 + \dots + c_k v_k$$

where $v_1...v_k$ are some special solutions of *.

Lecture Notes: 1 Feb, 2012

Conservation relations

$$\frac{dx}{dt} = Sv$$

S is nxm

Consider a vector x such that $x^T S = 0$

$$y^T \frac{dx}{dt} = y^T (Sv) \rightarrow \frac{dy^T x}{dt} = y^T S_v = 0 \rightarrow \frac{d}{dt} y^T x = 0$$

This implies that $y^T x$ is constant. $y^T x = [y_1...y_n][x_1...x_n]^T = y_1 x_1 + y_2 x_2 + ... + y_n x_n = \text{constant}$

Example

glucose --(ATP -> ADP)--> gluc_6P <----> fruc_6P --(ATP -> ADP)--> fruc-1,6P

$$S_{-1}$$
 S_{-2} S_{-3} S_{-4} S_{-5} S_{-6} V_{-1} V_{-2} V_{-3}

Matlab code starts here

 $A = [-1 \ 1 \ 0 \ 0 \ -1 \ 1, \ 0 \ -1 \ 1 \ 0 \ 0, \ 0 \ 0 \ -1 \ 1 \ -1 \ 1]$ null(A)

% if you scale, you get this vector [2 1 1 0 0 1, 0 0 0 0 1 1, 1 1 1 1 0 0]

% first column y_1, second y_2, third y_3

% additional conservation relation

 $\% -y_1 + 3*y_2 + 2*y_3 = [0 1 1 2 3 2]^T$