## Lecture Notes: 20 Jan, 2012

## Gillespie Method

## Nearest neighbor: random walk on a lattice

We assume that the transition probability from site i to site j is given by the Poisson process

$$W_{ij}\Delta t$$

Master equation: We are interested in the probability p(x,t) of the particle being at site x at time t.

$$\begin{split} p(x,t+\Delta t) &= p(x,t)(1-W_{x,(x+1)}\Delta t - W_{x,(x-1)}\Delta t) + P(x-1,t)w_{x-1,x}\Delta t + P(x+1,x)W_{x+1,x}\Delta t \\ P(x,t+\Delta t) - P(x,t) &= -P(x,t)W_{x,x+1}\Delta t - P(x,t)W_{x,x-1}\Delta t + P(x+1,t)W_{x+1,x}\Delta t + P(x-1,t)W_{x-1,x}\Delta t \\ &\frac{P(x,t+\Delta t) - P(x,t)}{\Delta t} = -\sum_k P(x,t)W_{x,k} + \sum_k P(k,t)W_{k,x} \end{split}$$

If we let  $\Delta t \to 0$ , we have

$$\frac{d}{dt}P(x,t) = \sum_{k} (P(k,x)W_{k,x} - P(x,t)W_{x,k})$$

This is what is referred to as the Master equation.

## Stochastic chemical kinetics

In deterministic chemical kinetics we represent the state of the system by the concentrations of the various molecular species involved. In stochastic chemical kinetics, the state of the system is represented by a vector

$$X(t) = [x_1(t)x_2(t)...x_3(t)]^T$$

$$A + B \to C + D$$
 
$$X(t) = [A(t)B(t)C(t)D(t)]^T$$
 
$$[112000]^T \to [101911]^T \to [91822]^T$$