## **Kinetic Monte Carlo Notes**

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#### Monte Carlo Methods

- Solve complex problems using random sampling from a probability distribution (i.e. stochastic description).
- Useful to evolve a physical system to a new state from an esemble of potential future states.

### Integrating a function MC sampling

• If we want to evaluate the integral of a function over some domain we can numerically approximate this using the midpoint rule:

$$\int_{a}^{b} f(x)dx = \frac{b-a}{N} \sum_{i=1}^{N} f(x_{i})$$
 (0.1)

• There is an alternative way to do this using probablity theory to determine the expectation value of a function f(x) for random variable x:

$$\int_{a}^{b} p(x)f(x)dx = \frac{b-a}{N} \sum_{i=1}^{N} f(x_{i})$$
 (0.2)

where p(x) is a uniform probability distribution over the interval [a, b].

- The difference between numerically evaluating Equation 0.1 and Equation 0.2, is that Equation 0.1 is evaluated over a grid of points and Equation 0.2 is randomly sampled points.
- The error of MC integration is  $\propto \frac{1}{\sqrt{N}}$  as a result of central limit theorem

# Example integrating a function using MC sampling<sup>1</sup>

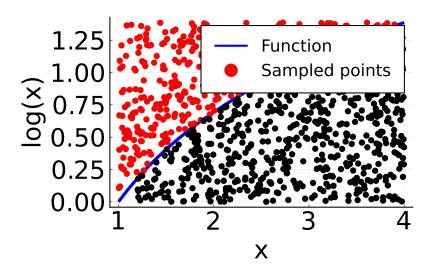


Figure 0.1: Random sampled points from uniform distribution over the interval [1, 4]. The black points are those that are accepted.

<sup>&</sup>lt;sup>1</sup>A more detailed notebook implementing the code can be viewed here

# Example integrating a function using MC sampling

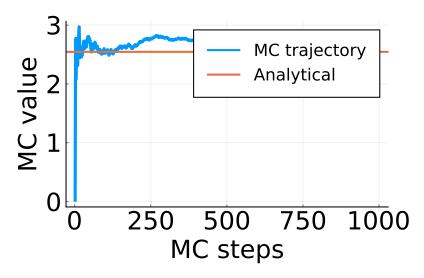


Figure 0.2: Integration of log(x) using MC.

# Statistical Thermodynamics & Ensemble Properties

- Microscopic  $\rightarrow$  Macroscopic description
  - How positions and momenta of  $10^{23}$  particles relates to bulk temperature, pressure, or volume.
- Ensembles use probability of specific microstate. Probability theory provides average of a function or variable, \( \lambda X \rangle :\)

$$\langle X \rangle = \frac{1}{N} \sum_{i=1}^{N} n_i X_i = \sum_{i=1}^{N} \underbrace{p_i}_{\text{PDF}} X_i \qquad (0.3)$$

- If  $\langle X \rangle$  is continous, Equation 0.3 is an integral.
- $p_i$  is the probability the system is in state i. The probability density function (PDF) has the property that its normalized, i.e.  $\sum_{i=1}^{N} p_i = 1$

# Statistical Thermodynamics & Ensemble Properties

- The consequence of Equation 0.3 is that microscopic collections (i.e. ensemble of systems) can be used to calculate macroscopic properties.
- Choice of  $p_i = \frac{z_i}{Z}$  depends on macroscopic conditions which manifest through the partition function:

$$Z = \sum_{i} e^{-\beta X_i} \tag{0.4}$$

• For a macroscopic system that has constant particles, volume and temperature, i.e., canonical.

$$-\beta=\frac{1}{k_bT}$$
 and  $X_i=E_i$  where Boltzmann factor is 
$$z_i=e^{-\frac{E_i}{k_bT}}$$

$$\langle E \rangle = \frac{1}{Z} \sum_{i} e^{-\frac{E_i}{k_b T}} E_i \tag{0.5}$$

# **Statistical Thermodynamics & Ensemble Properties**

- The biggest challenge in evaluating Equation 0.5 is it requires knowledge of all possible configurations.
- If Z is a configurational integral, e.g.,  $Z = \int e^{-U(\mathbf{r}^N)/k_B T} d\mathbf{r}^N$ , then there are 3N possible configs!
- The key insight is that most configurations are not probable:
  - If the two atoms are extremely close at moderate T, the term  $U(\mathbf{r})$  is large an hence the probability low.
- The question then becomes, can we determine  $p_i = \frac{1}{Z}e^{-\frac{E_i}{k_BT}}$  efficiently, that is the states with highest probablity centered around  $\langle E \rangle$  given that Z is not accessible.

 $U(\mathbf{r})$  is the potential energy between pairs of atoms.

### Metropolis Monte Carlo

- If we wanted to evaluate Equation 0.5 for an atomic system (i.e. the discrete states are replaced by continuous atomic configurations), we could use the MC sampling as in Equation 0.2.
- However we need to integrate over 3N dimensions!
- This eliminates the feasability for determining the partition function Z which is required to know the probability of any specific configuration  $p_i$

### **KMC CVD Example**

### KMC CVD Example height

#### **Backmatter**

#### Connect with me!



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#### Note

This presentation can be viewed online at https://stefan bringuier.github.io/KMCNotes. A report formated PDF of this presentation can be downloaded here.

**?** Tip

To export  $\tt revealjs$  presentations to pdf, press 'e' then 'ctrl-p' 'save as pdf'

### References

Andersen, Mie, Chiara Panosetti, and Karsten Reuter. 2019. "A Practical Guide to Surface Kinetic Monte Carlo Simulations." Frontiers in Chemistry 7. https://doi.org/10.3389/fchem.2019.00202.

LeSar, R. 2013. Introduction to Computational Materials Science: Fundamentals to Applications. Cambridge University Press. https://books.google.com/books?id=QzkhAwAAQBAJ

"Numerical Integration - Midpoint, Trapezoid, Simpson's Rule." 2021.