

# **Kinetic Monte Carlo Notes**

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## Monte Carlo Methods

- Solve complex problems using random sampling from a probability distribution (i.e. stochastic description).
- Useful to evolve a physical system to a new state from an ensemble of potential future states.

## Integrating a function MC sampling

- If we want to evaluate the integral of a function over some domain we can numerically approximate this using the [midpoint rule](#):

$$\int_a^b f(x)dx = \frac{b-a}{N} \sum_{i=1}^N f(x_i) \quad (0.1)$$

- There is an alternative way to do this using probability theory to determine the expectation value of a function  $f(x)$  for random variable  $x$ :

$$\int_a^b p(x)f(x)dx = \frac{b-a}{N} \sum_{i=1}^N f(x_i) \quad (0.2)$$

where  $p(x)$  is a uniform probability distribution over the interval  $[a, b]$ .

- The difference between numerically evaluating Equation 0.1 and Equation 0.2, is that Equation 0.1 is evaluated over a grid of points and Equation 0.2 is evaluated over randomly sampled points.
- The error of MC integration is  $\propto \frac{1}{\sqrt{N}}$  as a result of [central limit theorem](#)

## Example integrating a function using MC sampling<sup>1</sup>

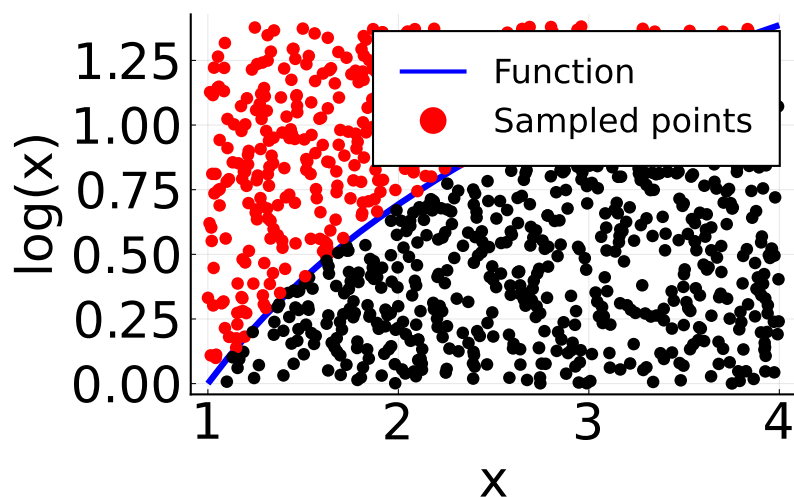


Figure 0.1: Random sampled points from uniform distribution over the interval  $[1, 4]$ . The black points are those that are accepted.

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<sup>1</sup>A more detailed notebook implementing the code can be viewed [here](#)

## Example integrating a function using MC sampling

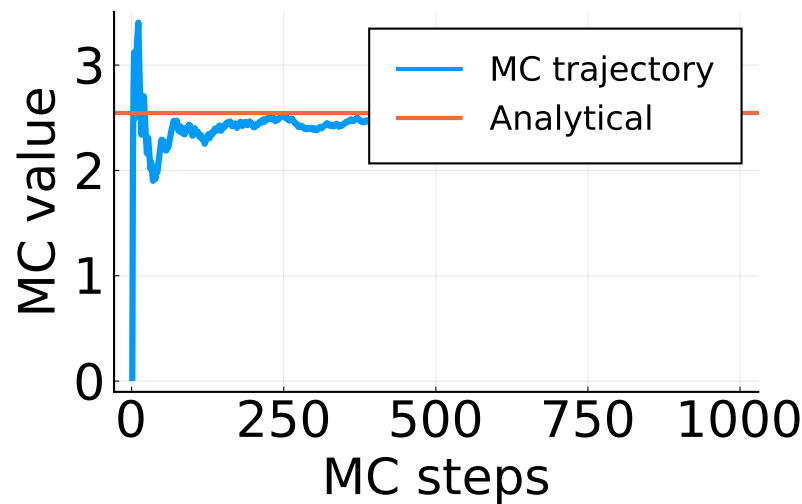


Figure 0.2: Integration of  $\log(x)$  using MC.

## Statistical Thermodynamics & Ensemble Properties

- Microscopic  $\rightarrow$  Macroscopic description
  - How positions and momenta of  $10^{23}$  particles relates to bulk temperature, pressure, or volume.
- Ensembles use probability of specific microstate. Probability theory provides average of a function or variable,  $\langle X \rangle$ :

$$\langle X \rangle = \frac{1}{N} \sum_{i=1}^N n_i X_i = \sum_{i=1}^N \underset{\text{PDF}}{p_i} X_i \quad (0.3)$$

- If  $\langle X \rangle$  is continuous, Equation 0.3 is an integral.
- $p_i$  is the probability the system is in state  $i$ . The probability density function (PDF) has the property that its normalized, i.e.  $\sum_{i=1}^N p_i = 1$

## Statistical Thermodynamics & Ensemble Properties

- The consequence of Equation 0.3 is that microscopic collections (i.e. ensemble of systems) can be used to calculate macroscopic properties.
- Choice of  $p_i = \frac{z_i}{Z}$  depends on macroscopic conditions which manifest through the partition function:

$$Z = \sum_i e^{-\beta X_i} \quad (0.4)$$

- For a macroscopic system that has constant particles, volume and temperature, i.e., [canonical](#).

$$- \beta = \frac{1}{k_b T} \text{ and } X_i = E_i \text{ where Boltzmann factor is } z_i = e^{-\frac{E_i}{k_b T}}$$

$$\langle E \rangle = \frac{1}{Z} \sum_i e^{-\frac{E_i}{k_b T}} E_i \quad (0.5)$$

## Statistical Thermodynamics & Ensemble Properties

- The biggest challenge in evaluating Equation 0.5 is it requires knowledge of all possible configurations.
- If  $Z$  is a configurational integral, e.g.,  $Z = \int e^{-U(\mathbf{r}^N)/k_B T} d\mathbf{r}^N$ , then there are  $3N$  possible configs!
- The key insight is that most configurations are not probable:
  - If the two atoms are extremely close at moderate  $T$ , the term  $U(\mathbf{r})$  is large and hence the probability low.
- The question then becomes, can we determine  $p_i = \frac{1}{Z} e^{-\frac{E_i}{k_B T}}$  efficiently, that is the states with highest probability centered around  $\langle E \rangle$  given that  $Z$  is not accessible.

$U(\mathbf{r})$  is the potential energy between pairs of atoms.

## Metropolis Monte Carlo

- If we wanted to evaluate Equation 0.5 for an atomic system (i.e. the discrete states are replaced by continuous atomic configurations), we could use the MC sampling as in Equation 0.2.
- However we need to integrate over  $3N$  dimensions!
- This eliminates the feasibility for determining the partition function  $Z$  which is required to know the probability of any specific configuration  $p_i$

## Metropolis Monte Carlo

- The [Metropolis algorithm](#) is a process to sample states  $i$  with probability  $p_i$
- This is achieved by using relative probabilities, i.e.,  $\frac{p_i}{p_j}$
- From this we get the correct average quantities.
- This works because, even though we don't know  $Z$  and can't determine  $p_i$ , the results of  $\frac{p_i}{p_j}$  gives the correct distribution
- Relative probabilities are given as:

$$\frac{p_i}{p_j} = \frac{e^{-E_i/k_B T}}{Z} \frac{Z}{e^{-E_j/k_B T}} = e^{-(E_i-E_j)/k_B T} \quad (0.6)$$

- Which only depends on energy difference between states as shown in graph

## Metropolis Monte Carlo

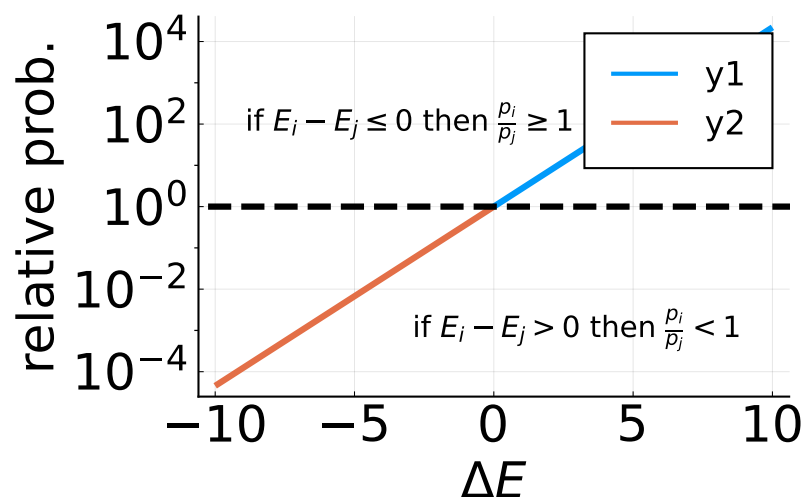


Figure 0.3: Relative probability for two states.

## Metropolis Monte Carlo

- In the metropolis MC approach we use the relations on Figure 0.3 to create a trajectory of states.
- The steps for MMC are:
  1. Generate configuration  $i$  with  $E_i$
  2. Randomly trial configuration,  $i + 1$ , and calculate  $E_{i+1}$
  3. Get relative probability via Equation 0.6.
  4. Use relations in Figure 0.3 to accept or accept with probability  $\frac{p_i}{p_j} < 1$  given a randomly generated number between  $(0, 1)$



## Backmatter

### Connect with me!



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#### Note

This presentation can be viewed online at <https://stefanbringuier.github.io/KMCNotes>. A report formatted PDF of this presentation can be downloaded [here](#).

#### Tip

To export `revealjs` presentations to pdf, press ‘e’ then ‘ctrl-p’ ‘save as pdf’

## References

Andersen, Mie, Chiara Panosetti, and Karsten Reuter. 2019. “A Practical Guide to Surface Kinetic Monte Carlo Simulations.” *Frontiers in Chemistry* 7. <https://doi.org/10.3389/fchem.2019.00202>.

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