

Class:

G O S F R

(D)

Student Name:

Hiran Evans








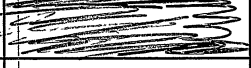








Part A /16

Part B / 27

TOTAL

/43

ANSWER SHEET for MULTIPLE CHOICE -Clearly mark 1 answer for each question.

QUESTION	A	B	C	D
1				
2				
3				
4				
5				
6				
7				
8				
9				
10				
11				
12				
13				
14				
15				
16				

Part II

27 marks

Attempt Questions 16-19.

Allow about 35 minutes for this section

Question 16 (15 marks)

Marks

The paragraph below is a student's write-up of an experiment.

1. I put 100 mL of water in a test tube and measured its temperature. It was 18°C . Then I put some of the crystals in it and stirred the mixture to dissolve the crystals. I kept stirring until some remained on the bottom of the tube no matter how much longer I stirred.
2. I filtered the mixture and then evaporated all the water from the solution. I weighed the amount of solid left behind and found that 6.0 g had been dissolved.
3. Then I did it again but this time I heated the water using a Bunsen burner, gauze mat and tripod while the thermometer was suspended from a retort stand using water at 29°C . I found that 8.0 g dissolved.
4. I repeated it at 40°C and at 47°C and got 10.0 g and 11.2 g as my results

- (a) Write an aim appropriate for the experiment.

1

Aim: To find the differences of the mass of crystal left behind after being put in water and evaporated.

- (b) Complete the table for the student's results.

Temperature ($^{\circ}\text{C}$)	Mass left behind (g)
18°C	6g
29°C	8g
40°C	10g
47°C	11.2g

- (c) Identify the independent and dependent variable for this experiment.

2

The independent variable is temperature at which the water starts at. The dependent variable is the mass of crystal left behind.

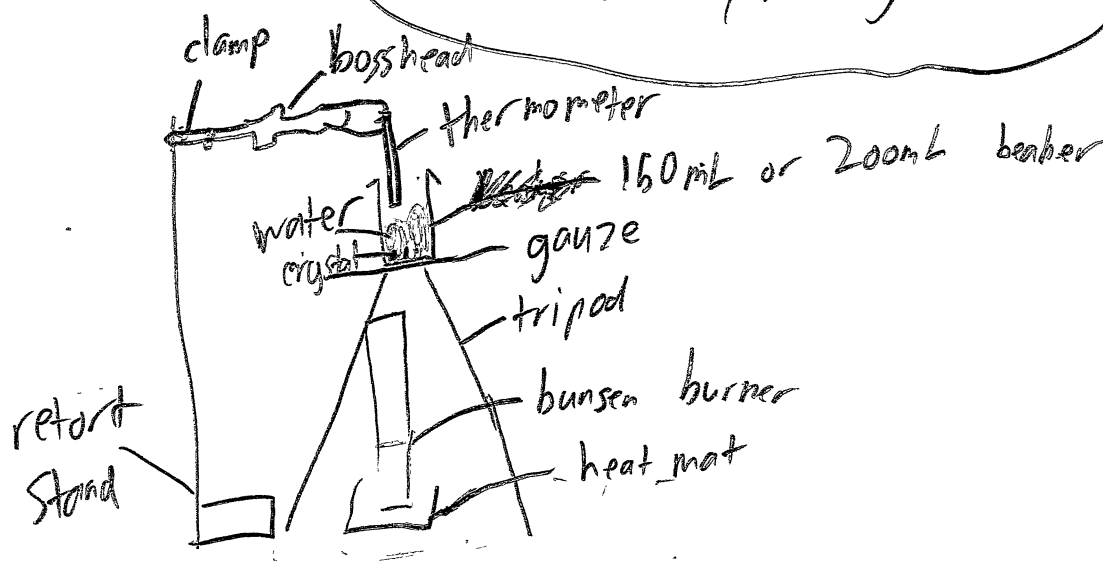
(d) Identify a variable that needs to be controlled during the experiment to make it a fair or valid test.

1

The amount of crystal added each time.

(e) Draw a labelled scientific diagram showing the equipment set up required to carry out step 3 as described above.

3

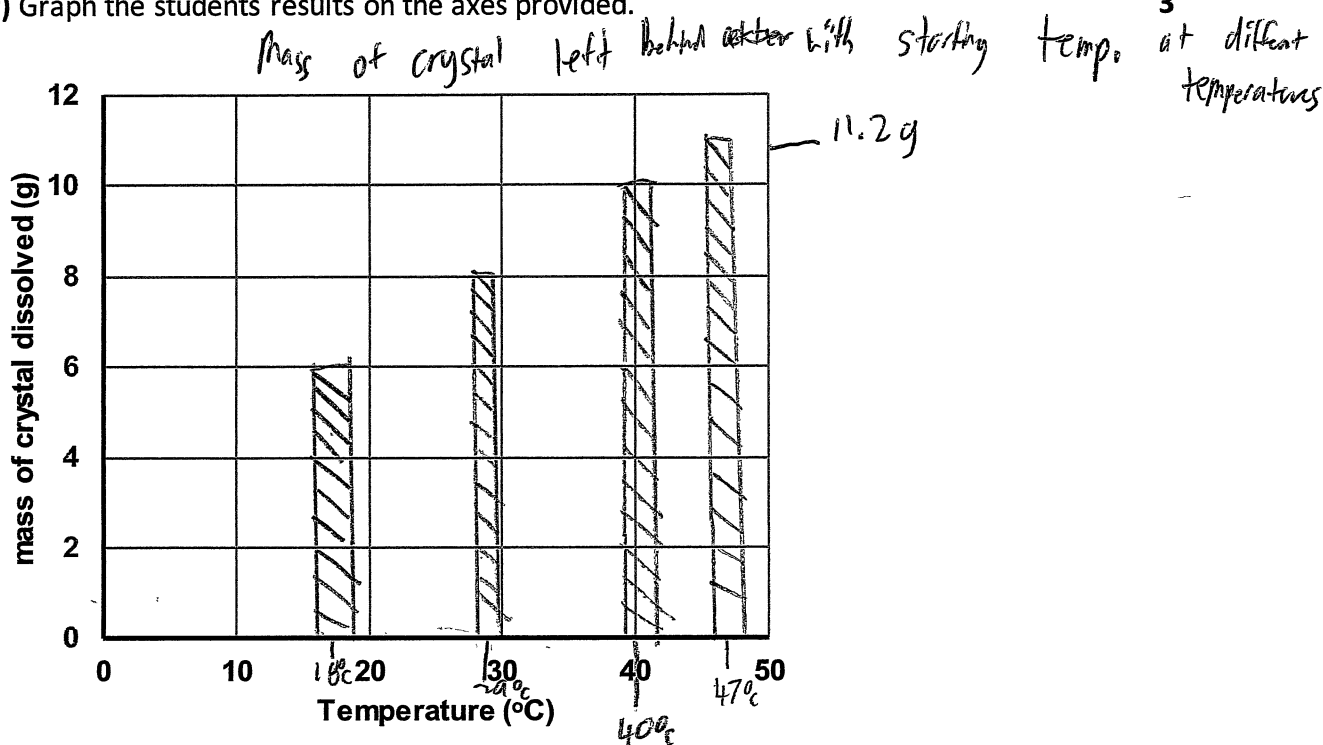


(f) Identify two safety issues the student will have to be concerned with through this experiment.

2

Safety issue 1: I believe the student has to be concerned about not getting a large blast of steam in his eyes. Safety issue 2: I think he has to be concerned about not getting any crystal in his mouth, eyes, or open lacerations because doing so could have adverse effects.

(g) Graph the students results on the axes provided.



(h) Write a conclusion for the experiment.

My conclusion is that changing the starting temperature of water affects the mass of crystal left behind. The higher the starting temperature, the more crystal left behind.

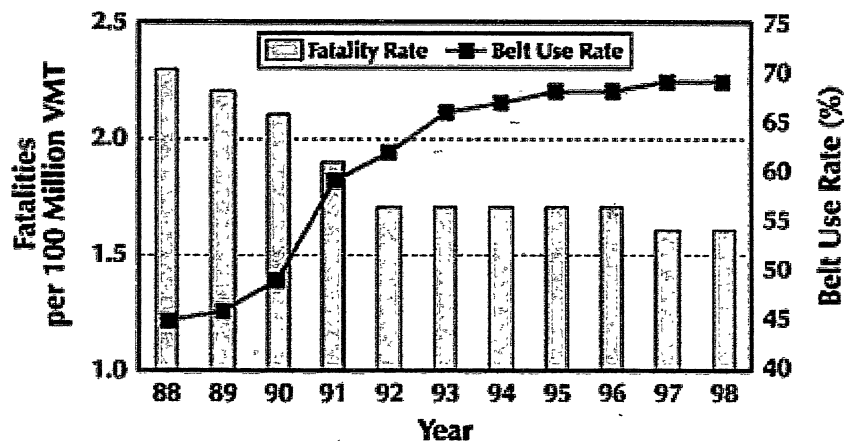
Question 17 (4 marks)

The following scientists are working in different branches or disciplines of science. Identify which branch each is working in:

Activity	Branch of Science
Paris is studying the crystals embedded in a rock.	geology
Beau is developing a new type of plastic	chemist
Shaun is investigating the eating habits of insects	entomologist ecologist zoologist
Angus is monitoring the movement of an asteroid	astronomer

Question 18. (4 marks).

The graph shows information about road fatalities and the use of seat belts in cars.



- (a) According to this data what is the trend shown in the number of fatalities between 1988 and 1996? Provide data to support your answer. 2

The trend is that the fatalities each year slowly go down, as seat belt usage goes up. In 1988,

- (b) Analyse the data presented and provide reasons for the conclusion you made. 2

I therefore conclude that from the data shown, higher seat belt use rates have been lowered the fatality rates per 100 million VMT. My reasoning is, if you analyse the graph above, you can see a clear trend that belt rates have increased, while at the same time, death rates decreased.

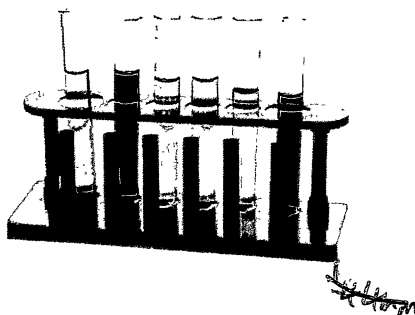
The fatality rate was about 2.3, and the seat-belt usage rate was about 45%. In 1996, the fatality rate had gone down to about 1.7, and the belt usage rate had gone up to 70%.

Question 19. (4 marks).

The drawing made by a scientist was twice as big as the real size of the object.

Determine the actual length of the whole piece of equipment. Show your working.

2



$$4.4\text{cm} \div 2 =$$

2.2cm, which is
the actual
length of the
object in real
life

$$4.4\text{mm} = 4.4\text{cm}$$

I measured the drawing to be 4.4cm. I then halved that
to get 2.2cm, since the drawing is 2x as large as the
object in real life. The answer is 2.2cm

b) There are some problems with the equipment diagram above. Identify two things that the scientist needs to change to accurately represent the equipment above.

2

I think that scientist has to make the
test tubes the same colour and some
of the test tube holes are bigger than others
to make them more scientifically accurate
by making them the same

END OF EXAM