

BEBI5009 Homework3

Due 04/12/2018(Thur) before midnight

1. 3.7.9 Non-cooperative multi-site binding.

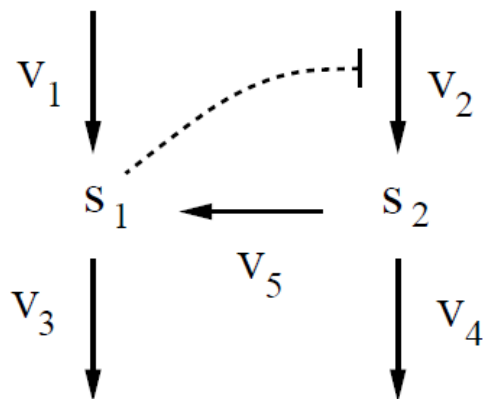
A protein with multiple binding sites that are independent (no cooperativity) cannot exhibit a sigmoidal binding curve, even when the binding sites have distinct affinities. Consider a protein with two ligand binding sites of different affinities. Show that in this case the fractional saturation is simply the sum of two hyperbolic relations:

$$Y = \frac{[X]/K_1}{2(1 + [X]/K_1)} + \frac{[X]/K_2}{2(1 + [X]/K_2)}$$

where K_1 and K_2 are the dissociation constants for the two sites. Plot this relation for various values of K_1 and K_2 to confirm that it describes a hyperbolic binding curve.

2. 4.8.7 Nullcline analysis.

Consider the network below.



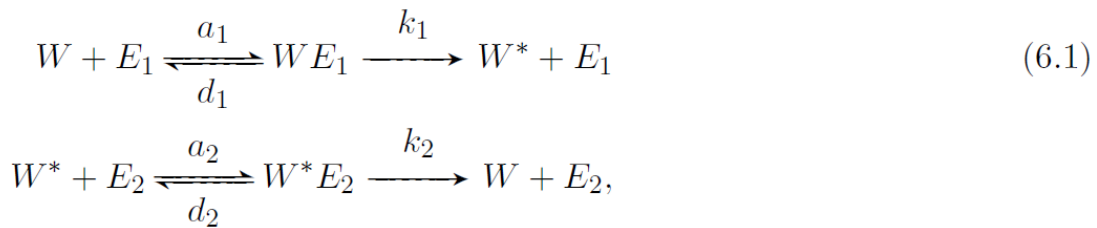
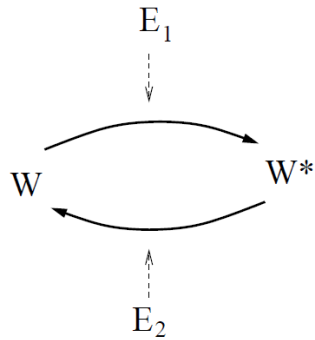
Suppose the reaction rates are given by

$$\begin{aligned} v_1 &= V & v_2 &= f(s_1) & v_3 &= k_3 s_1 \\ v_4 &= k_4 s_2 & v_5 &= k_5 s_2 \end{aligned}$$

Suppose that the parameters V , k_3 , k_4 , and k_5 are positive constants, and that $f(s_1)$ takes positive values and is a decreasing function of s_1 (i.e. as the values of s_1 increase, the values of $f(s_1)$ decrease). By sketching the nullclines and direction fields and demonstrate that this system cannot exhibit bistability. Please also sketch a few trajectories starting from different initial conditions in the S_1 - S_2 phase plane.

3. 6.8.6 Ultrasensitivity.

Consider the network below.



$$\frac{k_1 E_{1T}}{k_2 E_{2T}} = \frac{w^*(w + K_1)}{w(w^* + K_2)} = \frac{w^*(1 - w^* + K_1)}{(1 - w^*)(w^* + K_2)}, \tag{6.2}$$

where E_{1T} and E_{2T} are the total enzyme concentrations and

$$K_1 = \frac{1}{W_T} \frac{d_1 + k_1}{a_1}, \quad K_2 = \frac{1}{W_T} \frac{d_2 + k_2}{a_2}.$$

Derive equation (6.2) in Section 6.2.1, as follows. Begin by writing the steady-state conditions for each of the four species in network (6.1). Use the steady-state conditions for the complexes WE_1 and W^*E_2 to write the steady-state concentration $[WE_1]$ in terms of $[W]^{ss}$, E_{1T} , and the rate constants. Likewise, write $[W^*E_2]^{ss}$ in terms of $[W^*]^{ss}$, E_{2T} , and the rate constants. Finally, use the steady-state condition $k_1[WE_1] = k_2[W^*E_2]$ and the approximation $W_T = [W] + [W^*]$ to derive equation (6.2).

Please interpret the meaning of equation 6.2 as in figure 6.7.