

Site: ACE ; Element: Zn

Zn clr [ICPMS]

$n = 302$

WLS: $y = -0.9 + 0.12 \cdot x$, $R^2 = 0.144$, $P = 9e-12$

OLS: $y = -0.78 + 0.15 \cdot x$, $R^2 = 0.213$, $P = 2.5e-17$

-2e+00

-1e+00

0e+00

1e+00

Zn clr [XRF-CS]

● BI10 ● KER1 ● PB1
● HER42PB ● KER3 ● POB4

