

Nuclear Plant Engineering



Stu Blair
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...PRINCIPLES ARE MORE IMPORTANT THAN FACTS...

H.G. RICKOVER

...ONE OF THE PREREQUISITES OF ORIGINALITY IS THE ART OF FORGETTING, AT THE PROPER MOMENT, WHAT WE KNOW. WITHOUT THIS ART THE MIND REMAINS CLUTTERED WITH READY-MADE ANSWERS AND IS NOT FORCED TO ASK THE PROPER QUESTIONS.

H.G. RICKOVER

UNITED STATES NAVAL ACADEMY

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Preface

The goal of this book is to teach a selection of topics in energy conversion and thermal-hydraulics. The book is organized into 30 lectures and is derived from notes developed for ER468 Nuclear Plant Engineering taught at the United States Naval Academy over a period spanning roughly from spring 2015 through the fall of 2022.

The textbook used while these lecture notes were prepared is Nuclear Systems by Neil Todreas and Mujid Kazimi.¹ Material from that excellent text is incorporated throughout these notes.

Students are all expected to have taken classes in engineering thermodynamics and fluid dynamics; most students also take heat transfer concurrently with ER468. Consequently several introductory topics are skipped altogether. The energy conversion topics—comprising extensive treatment of Rankine, Brayton, and combined cycles—omit treatment of the most fundamental topics of thermodynamics; the thermal-hydraulics lectures cover nuclear-specific extensions of fluid dynamics and heat transfer with only the briefest mention of introductory material.

One unique aspect of this text, at least relative to any other textbook on nuclear energy conversion or thermal-hydraulics, is the extensive use of MATLAB² for carrying out calculations. Students are expected to be familiar with the MATLAB programming environment and language features. Thermal and transport properties of fluids are made available with the help of EasyProp,³ which is a Python interface to the free and open-source library CoolProp and its associated Python wrapper.⁴ All of the examples provided in this text use MATLAB but the underlying Python libraries could, of course, be used in Python. The goal of this approach is to make these material properties available in a computing environment that the students could easily use in conjunction with their other engineering analysis workflows; I specifically wanted to avoid a special-purpose tool that could only be employed for this course.

As with any textbook the problem sets comprise an essential element of the material. The assignments presented in this text were delivered in a “workshop” format; the class met in a computer lab-

¹ Neil E Todreas and Mujid Kazimi. *Nuclear Systems Volume I: Thermal hydraulic fundamentals*, volume 1. CRC press, 2011

² Inc. The Math Works. Matlab, v2022a, 2022. URL <https://www.mathworks.com/>

³ Stu R Blair. Easyprop, v0.10.0, 2022. URL <https://github.com/stu314159/EasyProp>

⁴ Bell et al. Coolprop, v6.4.1, 2022. URL <http://www.coolprop.org/>

oratory and students spent a class period working the problems. This format provides ample opportunities for the students to ask questions—directed at the instructor or to each other—resulting in a very efficient learning environment. However you go about it, you will miss out on a lot of material if you choose to skip the assignments.

Part I

Nuclear Energy Conversion

Lecture 1 - Course Introduction, Rankine Cycle Review

Objectives

The objectives of this lecture are:

- Introduce the course including basic topics covered, syllabus, and course policy
- Review the basic Rankine Cycle that you learned in EM319 Thermodynamics I

Course Introduction

The basic topics that we will cover include:

1. Rankine Cycle
2. Brayton Cycles
3. Combined Cycles
4. Hydraulic analysis of nuclear systems
5. Thermal analysis of nuclear systems
6. Non-steady-state analysis

THE FIRST THREE TOPICS COMPRIZE a thorough introduction to nuclear energy conversion. The cycles that we analyze will range in complexity to allow introductory concepts to be reviewed while also addressing energy conversion cycles of practical relevance in modern nuclear systems; the latter are necessarily complex.

The last three topics are core elements of a typical “thermal-hydraulics” course and extend concepts taught in more general fluid dynamics and heat transfer classes with concepts and correlations that are relevant to nuclear applications.

Rankine Cycle Review

THE OVERWHELMING MAJORITY OF the energy conversion cycles for nuclear power plants are based on the Rankine Cycle. Features employed in various implementations of the Rankine cycle range from the compact, reliable, albeit *simple* steam plants used on nuclear submarines to complex but highly efficient—and correspondingly difficult to analyze—systems used in commercial nuclear power plants. In this course we will study Rankine cycles deeply. As a starting point, however, we will use a simplified and idealized representation of a basic Rankine cycle based on the Westinghouse AP1000.⁵ A schematic of this cycle is given in Figure 1.

THE SIMPLE IDEAL Rankine cycle comprises four basic processes:

1. (1 → 2) **isentropic compression** A feed pump is used to increase the pressure (flow energy) of the saturated liquid that exits the condenser.
2. (2 → 3) **isobaric heat addition** Feed enters the steam generator and, at constant pressure, is heated until it changes phase, exiting the steam generator as a saturated vapor.
3. (3 → 4) **isentropic expansion** High pressure saturated vapor is expanded in a turbine; converting its thermal energy into mechanical energy. The exhaust gas is (usually) a saturated mixture.
4. (4 → 1) **isobaric heat rejection** The saturated mixture leaving the turbine exhausts into a condenser where, again at constant pressure, heat is rejected to condenser cooling water until the fluid is in a liquid state.

The numbers in parenthesis correspond to state point labels provided in Figure 1.

It is also customary to illustrate the Rankine cycle on a Temperature-Entropy plot as shown in Figure 2. The numbers correspond to state-points, the curve is representative of the saturation curve for water. The lines connecting the state point numbers represent the constant pressure or constant entropy contours, and the arrows and labels are indicating the transfer of heat and work to and from the cycle.

Rankine cycle is another term for a "Steam Cycle" comprising a boiler or steam generator to make steam that is directed through a turbine that converts the thermal energy of the steam to mechanical work. The steam exiting the turbine is condensed to water and pumped back to the boiler or steam generator to complete the cycle.

⁵ LLC Westinghouse. Westinghouse ap1000 design control document rev. 19. *Westinghouse Electric Company LLC*, 2011

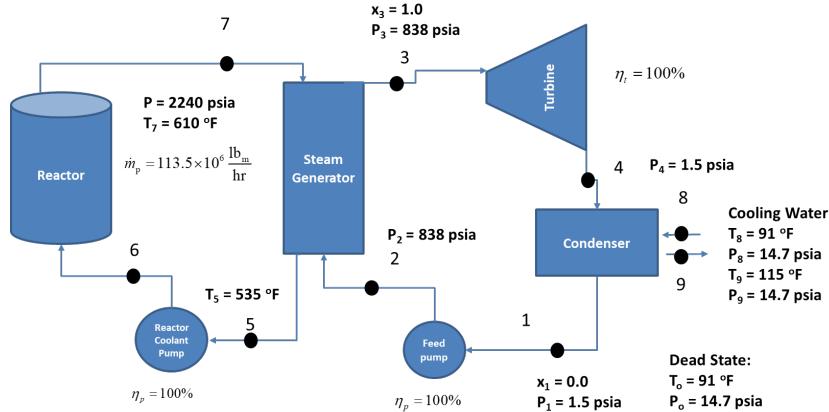


Figure 1: Simple ideal Rankine cycle modeled on the Westinghouse AP1000.

WE WILL FIRST ANALYZE this cycle from the perspective of conservation of energy. Equation 1 captures the energy terms of relevance for this analysis.

$$q_{i \rightarrow f} - w_{i \rightarrow f} = \underbrace{\frac{g}{g_c} (z_f - z_i)}_{\Delta PE} + \underbrace{\frac{1}{2g_c} (v_f^2 - v_i^2)}_{\Delta KE} + \underbrace{(h_f - h_i)}_{\Delta \text{enthalpy}} \quad (1)$$

For processes comprising energy conversion cycles in this course, we will assume that ΔPE and ΔKE are negligible unless specified otherwise. Consequently we can generally simplify the conservation of energy equation as in Equation 2.

$$q_{i \rightarrow f} - w_{i \rightarrow f} = h_f - h_i \quad (2)$$

ANALYZING EACH PROCESS OF the simple ideal Rankine cycle for conservation of energy can be carried out for each process as:

- $(1 \rightarrow 2)$, isentropic compression $\rightarrow q_{1 \rightarrow 2} = 0$. $w_{1 \rightarrow 2} = h_1 - h_2$
- $(2 \rightarrow 3)$, isobaric heat addition $\rightarrow w_{2 \rightarrow 3} = 0$. $q_{2 \rightarrow 3} = h_3 - h_2$
- $(3 \rightarrow 4)$, isentropic expansion $\rightarrow w_{3 \rightarrow 4} = h_3 - h_4$, $q_{3 \rightarrow 4} = 0$.
- $(4 \rightarrow 1)$, isobaric heat rejection $\rightarrow q_{4 \rightarrow 1} = h_1 - h_4$, $w_{4 \rightarrow 1} = 0$.

We can compute the net specific work (w_{net}) and net specific heat (q_{net}) by summing the corresponding terms from each process in the cycle.

$$q_{net} = q_{2 \rightarrow 3} + q_{4 \rightarrow 1} = h_3 - h_2 + h_1 - h_4 = h_1 + h_3 - (h_2 + h_4)$$

$$w_{net} = w_{1 \rightarrow 2} + w_{3 \rightarrow 4} = h_1 - h_2 + h_3 - h_4 = h_1 + h_3 - (h_2 + h_4)$$

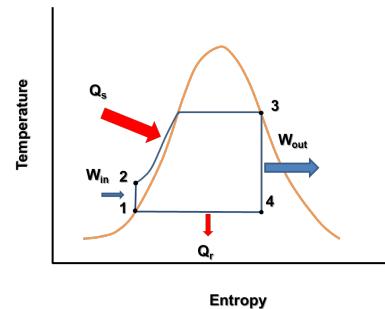


Figure 2: Temperature-Entropy plot of a simple, ideal, Rankine cycle.

Notes on Equation 1:

- PE = potential energy
- KE = kinetic energy
- enthalpy is the sum of flow energy (Pv) and internal energy (u).

Reminder: *isentropic* processes are *reversible* and *adiabatic*. When we say a process is *adiabatic* that means no heat in or out of the process, and thus $q_{1 \rightarrow 2} = 0$

Question: do you expect $w_{1 \rightarrow 2}$ to be positive or negative?

Net specific work is the amount of work produced per unit mass; *net specific heat* is similarly defined. In SI units both are typically given in kJ/kg; in USCS units: BTU/lb_m.

Comparing the equations above we can see that all of the same terms are listed in the right-most equality. From this we get a fundamental result stemming from conservation of energy:

$$w_{\text{net}} = q_{\text{net}}$$

When analyzing thermodynamic cycles you should⁶ always compute w_{net} and q_{net} and confirm that they are the same. If they are not the same, your answer cannot be correct.⁷

IN ADDITION TO COMPUTING NET SPECIFIC WORK and net specific heat, we customarily compute the thermal efficiency (η_{TH}). As you doubtless remember from your introductory thermodynamics class, the thermal efficiency is given by:

$$\eta_{\text{TH}} = \frac{w_{\text{net}}}{q_s} \quad (3)$$

Where q_s is the specific heat supplied and, for this cycle, is equal to $q_{2 \rightarrow 3}$.

Simple Ideal Rankine Cycle: Consider a Rankine Cycle used for energy conversion for a nuclear power plant. Saturated Steam is produced at a pressure of 838 psia and the condenser pressure is 1.5 psia. Assume all components are ideal. Using steam tables or a Mollier chart, estimate net work and thermal efficiency for this cycle.

Answers: $w_{\text{net}}=390.3 \text{ BTU/lbm}$, $\eta_{\text{TH}} = 35.1\%$

⁶ You *must*!!

⁷ Sadly, the converse is not true; if $w_{\text{net}} = q_{\text{net}}$, it is still possible that your answer is wrong.

Reminder: the Rankine Cycle is an example of a *heat engine*. For heat engines, the basic deal is: you provide the heat, the heat engine converts the heat to work. At best, you might hope that a really good heat engine would convert all of the heat to work; this would correspond to a thermal efficiency of 100 percent. In other words: the thermal efficiency is the measure that tells you what fraction of the input heat gets converted into output work. Unfortunately it is a practical fact, which we will discuss in future lectures, that the maximum thermal efficiency is limited to far less than 100 percent.

Lecture 2 - Thermodynamic Analysis with EasyProp

Objectives

The objectives of this lecture are:

- Describe the EasyProp tools
- Provide options for access to EasyProp
- Do an Example Rankine cycle calculation with EasyProp

The EasyProp Tools

What is EasyProp?

- A library written in the Python language based on the free and open-source library CoolProp. A schematic representation of the relationship between MATLAB, Python and the EasyProp and CoolProp Python libraries is given in Figure 3.
- Accessible in a MATLAB environment with a properly initialized call to `pyversion`⁸

How do I access EasyProp?

1. Login to ESSEX via remote desktop and use any version of MATLAB installed
 - (a) All software tools and libraries needed to use EasyProp have already been installed on this platform.
 - (b) Users *may* need to run `pyversion_fixer.m` to invoke the MATLAB built-in function `pyversion` with the path to the local Python installation on ESSEX.

Or

2. Install the required Python and MATLAB tools on your own laptop.



Figure 3: Schematic relationship between MATLAB and EasyProp

⁸ The MATLAB function `pyversion` sets several MATLAB environment variables that have the primary effect of “telling” MATLAB the path to the Python implementation that should be used when executing MATLAB scripts that use Python objects.

The script `pyversion_fixer.m` will be made available to students as necessary.

- (a) Download Anaconda (Anaconda3, Windows 64-bit, Python version 3.8)
- (b) From Anaconda prompt: conda install -c conda-forge coolprop
- (c) Locate Python executable. One way to do this is from Anaconda prompt use the command: where python
- (d) In MATLAB call *pyversion* with path to Python executable. E.g.
»pyversion('[path to python executable]')

EasyProp Interface

A SUCCINCT INTRODUCTION to the EasyProp interface is available in the appendices. For the remainder of this lecture I will try to exemplify the *structure* of a MATLAB script in which EasyProp is being used. I will follow this with a worked example where EasyProp is used to analyze a simple ideal Rankine Cycle.

A BASIC ANALYSIS SCRIPT using EasyProp will flow as follows:

1. Ensure the Python system path includes the current directory

This is accomplished with the following code:

```

1 %% Prepare the environment
2 clear;
3 clc;
4 close 'all';
5
6 EasyProp_path = ' '; %<- Path if EasyProp.py is in your current
7      directory
8 if count(py.sys.path,EasyProp_path) == 0 %<- see if desired
9      directory is on path
    insert(py.sys.path,int32(0),EasyProp_path); %<- if not; add
10     it.
11 end

```

Listing 1: Add the current directory to the Python system path

2. Construct a simple fluid object.

```

1 units = 'USCS';
2 fluid = 'Water';
3 water = py.EasyProp.simpleFluid(fluid, units);

```

Listing 2: Construct a *simpleFluid* object

A list of common fluids that we will use in this course is provided in Table 1.

3. Establish problem parameters

Note: If you opt for this method, you will have Python installed on your own laptop. This is worthwhile. When you graduate from USNA you will lose access to MATLAB along with all of the other non-free software packages that you may have become accustomed to using as a midshipman. Python is free and you can take it with you and it can become the most powerful tool in your computational toolbox.

Notice that we use the MATLAB commands *clear*, *clc*, and *close 'all'*. This is done in observation of the MATLAB Style requirement that we start every script with these three commands.

MATLAB Style requirements are specified in the Appendix however I will use repetition in the lecture notes in an attempt to provide added emphasis.

The leading *py* tells MATLAB that what follows must be executed by the Python interpreter. The name *EasyProp* points to a particular Python library, and *simpleFluid* is a function call within the EasyProp library. The function *simpleFluid* is a special type of function called a *constructor*.

Fluid	EasyProp Name
water	'Water'
air	'Air'
Carbon Dioxide	'CO ₂ '
Nitrogen	'N ₂ '
Helium	'He'
Refrigerant R245fa	'R245fa'

Table 1: Commonly used EasyProp fluid designations

```

1 numSp = 4; % number of state points
2 eta_turbine = 1.0; % isentropic efficiency of the turbine and
3   pump
4 eta_pump = 1.0;
5 Pmax = 838; % psia, max pressure
6 Pmin = 1.5; % psia, min pressure

```

Listing 3: Set problem parameters

4. Allocate fluid property arrays

```

1 %% declare fluid property arrays
2 h = nan(numSp,1);
3 h_s = nan(numSp,1);
4 s = nan(numSp,1);
5 s_s = nan(numSp,1);
6 P = nan(numSp,1);
7 T = nan(numSp,1);
8 T_s = nan(numSp,1);

```

Listing 4: Allocate arrays to hold fluid property data.

5. Determine fluid properties by using the functions associated with the EasyProp simpleFluid object.

An example function call as follows: water.h_pT(14.7,25)

water .
h_pT (14.7,25)

object function arguments

The "object" is the name of the Python object constructed in the step above. The function name is given by convention; in this case the property to be returned is the fluid enthalpy as a function of pressure and temperature. The arguments—14.7 and 25—are the corresponding pressure and temperature; in this unit system they are given in units of psia and degrees Fahrenheit respectively.

A list of fluid property abbreviations is given in Table 2. A list of units for each property used in the SI and USCS unit systems is given in Table 3.

THE REMAINING FUNCTIONS needed to determine fluid properties at the state points are given in the listing below.

```

%% calculate properties at state points
1 % state point 1 – condenser outlet condensate pump inlet
2 P(1) = Pmin;
3 h(1) = myFluid.hL_p(P(1));
4 s(1) = myFluid.sL_p(P(1));
5
6 % compression from state point 1 to state point 2
7 P(2) = Pmax;
8 s_s(2) = s(1);

```

Pre-allocation of arrays is mandated in the ER468 style rules. In this example we construct the arrays using the *nan* function which, besides allocating memory for the arrays, initializes each value to "Not-a-number" (or *nan*). "Not-a-number" is preferred because it can never be mistaken for a number and thus can never be mistaken for a valid property value. This avoids the common logical error of using an array value with some valid-but-incorrect initial value (like zero).

Note: Function name convention: [output property]_[input property][[input property](Arg 1, Arg 2)]. In most cases the output thermodynamic property is determined based on two other known properties. In the case given, enthalpy (*h*) is determined as a function of pressure (*p*) and temperature (*T*)

Property	EasyProp Abbreviation
Pressure	p
Temperature	T
enthalpy	h
internal energy	u
entropy	s
specific volume	v
quality	x
viscosity	mu

Table 2: Property abbreviations

Property	SI	USCS
Temperature	C	F
Pressure	kPa	psia
enthalpy	kJ/kg	BTU/lb _m

Table 3: EasyProp units

```

10 h_s(2) = myFluid.h_ps(P(2),s_s(2));
11 h(2) = h(1) - (h(1) - h_s(2))./eta_pump;
12 w_pump = h(1) - h(2);
13
14 % isobaric heat addition in the boiler from state point 2 to
15 % state point 3
16 P(3) = P(2);
17 h(3) = myFluid.hV_p(P(3));
18 s(3) = myFluid.sV_p(P(3));
19
20 q_s = h(3) - h(2);
21
22 % expansion in turbine from state point 3 to state point 4
23 P(4) = P(1);
24 s_s(4) = s(3);
25 h_s(4) = myFluid.h_ps(P(4),s_s(4));
26 h(4) = h(3) - eta_turbine*(h(3) - h_s(4));
27
28 w_turbine = h(3) - h(4);
29
30 % isobaric heat rejection in the condenser from sp 4 to sp 1
31 q_r = h(1) - h(4);

```

Listing 5: Find fluid properties for each state point in a simple Rankine cycle.

6. Complete analysis to find thermal efficiency and net work

```

1 %% First Law Check
2 w_net = w_pump + w_turbine;
3 q_net = q_s + q_r;
4
5 fprintf('Net work = %g kJ/kg \n',w_net);
6 fprintf('Net heat = %g kJ/kg \n',q_net);
7
8 %% Compute Thermal Efficiency
9 eta_th = w_net/q_s;
10
11 fprintf('Thermal efficieency = %4.1f percent \n',eta_th*100);

```

Listing 6: Analyze for net work and thermal efficiency

THERE IS MORE TO ANALYZE than simply obtaining values for net specific work and thermal efficiency. Some fundamental things to think about:

1. Are my results correct? In general it is difficult to know; most relevant problems do not come with a solution key. You should take steps to increase your confidence that your answer is sensible. A few essential steps include:
 - (a) If you have computed net specific work, you should also compute net specific heat (q_{net}) and verify that they are, to within reasonable limits of precision, equal to each other.⁹

Note: When writing your own MATLAB scripts for thermodynamic cycle analysis, you are *strongly encouraged* to write the necessary code to find one or two state point properties, run your script and examine the state point properties you have just found. This will allow you to:

1. Immediately identify the line responsible for the error in cases where you made a mistake in typing in a MATLAB or EasyProp function; and
2. Allow you to sanity-check the thermodynamic property you just found.

Observance of this practice will save you *many* hours of debugging effort. The example codes in the appendices illustrate a convenient way to have a state-point property table output in your MATLAB script.

⁹ When using a tool like EasyProp, q_{net} and w_{net} should be well within 1 BTU/lb_m or 1 kJ/kg of each other. If not, you probably have a bug somewhere.

- (b) Is your computed thermal efficiency less than the Carnot efficiency? If not: it is definitely wrong.
 - (c) How does this answer compare to your results for a similar cycle? Do the differences make sense? or
 - (d) If you make small changes to the problem parameters—e.g. you slightly change the maximum/minimum system temperature/pressure—does that result in the expected change to net work and thermal efficiency?
2. What are the strengths and weaknesses of the cycle that I have just analyzed? What could I change to make it better?

Answers to the question above cannot be given in general but you will become better at answering them with practice. This course is designed to give you plenty of practice.

Assignment #1: Basic Rankine Cycles

Use MATLAB and EasyProp to answer the following questions

1. Steam flowing at 15.9 lb_m/s enters the turbine of a simple Rankine cycle power plant at 1000 psia, 800°F and exits at 2 psia. Saturated liquid exits the condenser. Assume that the turbine and pump are isentropic. (**Note:** 1 kW = 3412 BTU/hr, 1 hp = 2545 BTU/hr)
 - (a) How many degrees superheat (degrees above saturation temperature) is the steam exiting the boiler?
 - (b) What is the quality of the steam exiting the turbine?
 - (c) Determine the power output of the turbine in kW and hp.
 - (d) Determine the power input to the pump in hp.
 - (e) Determine the cycle thermal efficiency.
2. In a 2-MW (turbine power output) Rankine cycle, saturated vapor leaves the steam generator at 2 MPa and expands in the turbine to an outlet condition of 15 kPa, 94% quality. Saturated liquid leaves the condenser. Assume the pump is isentropic.
 - (a) What is the isentropic efficiency of the turbine?
 - (b) Determine the flow rate of the steam [kg/s].
 - (c) Determine the cycle thermal efficiency.
3. Investigate the effect of the condenser pressure on the performance of a simple ideal Rankine cycle. Turbine inlet conditions of the steam are maintained constant at 5 MPa and 500°C while the condenser pressure is varied from 5 to 100 kPa. Determine the thermal efficiency of the cycle and plot it against the condenser pressure. Write a short essay (no need to exceed 500 words) discussing the results. Be sure to include a discussion on the extent to which the designer of an energy conversion system based on the Rankine Cycle has control of condenser pressure.

Lecture 3 - Carnot Cycle and Exergy Overview

Objectives

The objectives of this lecture are:

- Describe the Carnot Cycle
- Review the Simple Ideal Rankine Cycle in context of a Carnot Cycle
- Introduce exergy and provide related definitions.

The Carnot Cycle

THE CARNOT CYCLE IS an idealized heat engine that you should have learned about in a basic thermodynamics class or from a reference text.¹⁰ It is idealized in the sense that it represents the most efficient possible heat engine working between two (high and low temperature) thermal reservoirs. A temperature-entropy plot of the cycle is shown in figure 4 and consists of four processes:

1. isentropic compression 1 → 2
2. isothermal heat addition 2 → 3
3. isentropic expansion 3 → 4
4. isothermal heat rejection 4 → 1

The thermal efficiency of this cycle can be shown to be a function of the absolute temperature of the high temperature reservoir (T_h) and the absolute temperature of the low temperature reservoir (T_c).

$$\eta_{\text{Carnot}} = 1 - \frac{T_c}{T_h}$$

Note that as T_c decreases or as T_h increases, the Carnot efficiency increases. If we assume both reservoirs are non-negative and that T_h

¹⁰ Michael J Moran, Howard N Shapiro, Daisie D Boettner, and Margaret B Bailey. *Fundamentals of Engineering Thermodynamics*. John Wiley & Sons, seventh edition, 2010

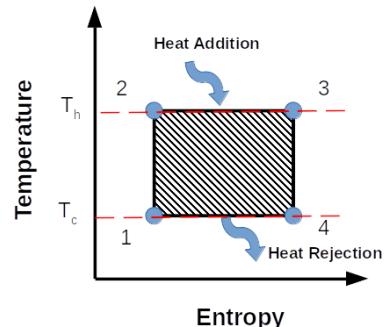


Figure 4: Temperature-Entropy plot of a Carnot cycle.

$> T_c$ (implying, $T_h > 0$) then the Carnot efficiency can be no greater than 1 and achieves that only for $T_c = 0$ or as $T_h \rightarrow \infty$.

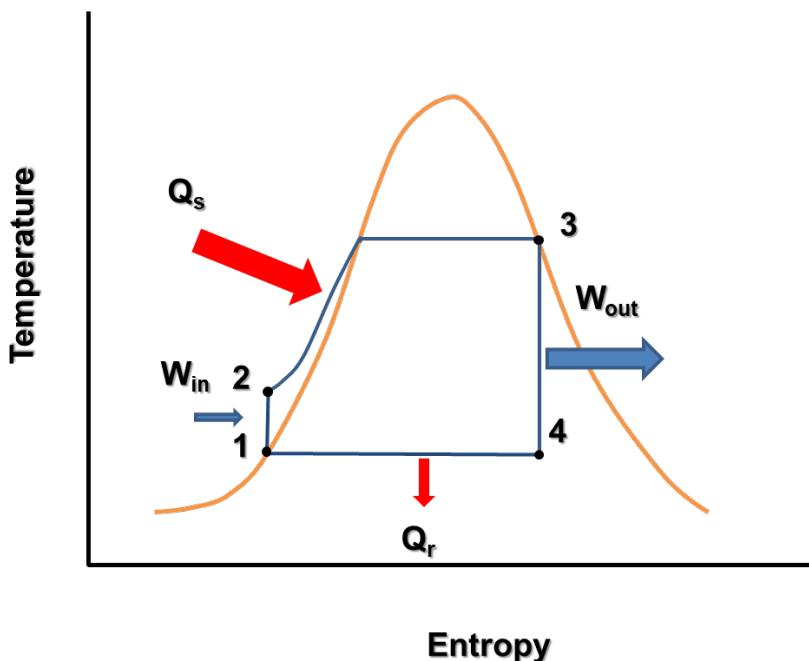
THE FOLLOWING CONCEPTS should be emphasized:

1. No heat engine can convert all heat input to work; and
2. A Carnot cycle is a heat engine that produces the maximum possible efficiency for any heat engine operating between a given fixed hot and cold temperature reservoir.

All of the processes of the Carnot Cycle are reversible and the work-in and work-out of process 1 and 3 are isentropic. The heat-in and heat-out of process 2 and 4 are reversible because the heat is transferred isothermally.

Simple Ideal Rankine Cycle - Revisited

CONSIDER AGAIN THE Simple Ideal Rankine cycle operating between the maximum pressure of 838 psia and minimum pressure of 1.5 psia. A schematic temperature-entropy diagram of this process is shown below.



Note: It is not evident from the temperature-entropy diagram, but the isothermal heat transfer processes in a Carnot cycle also must take place across an infinitesimal temperature gradient. We know from the laws of heat transfer that heat flows from high temperature to low temperature and the rate of heat transfer is proportional to:

1. the size of the heat transfer surface,
2. the temperature difference, and
3. some material property characterizing how "well" heat is transferred.

In order to transfer energy across an infinitesimal temperature gradient, the material must have arbitrarily good heat transfer properties, or the heat transfer surface must be arbitrarily large. We will dismiss these practical complications for the time being and simply move forward.

Figure 5: Temperature-Entropy plot of a simple, ideal, Rankine cycle.

From the example in Lecture 1, we know that the thermal efficiency is 35.1 percent and the net specific work is 390.3 BTU/lbm.

What is the Carnot efficiency of this cycle?

- To calculate the Carnot efficiency, we need the temperature of the hot and cold reservoirs respectively.
- If we take T_h to be the saturation temperature at 838 psia, we find that it is approximately 524 °F; this corresponds to 984 R.¹¹
- If we take T_c to be the condenser temperature (for heat rejection from state point 4 to 1), which is equal to the saturation pressure at 1.5 psia, we find that it is approximately 116 °F; this corresponds to 576 R.¹²
- Thus the Carnot efficiency is: $\eta_C = 1 - \frac{576}{984} = 0.415$; or 41.5 percent.

Consider the heat addition (Q_s) from state points 2 to 3; unlike the Carnot cycle, the heat transfer does not occur isothermally. Significant irreversibilities exist in this process. Although the heat rejected in the process from state point 4 to 1 does occur at constant temperature, in practice the heat is rejected to an external heat sink across a finite temperature difference. This also is a source of irreversibility although it is not evident from the temperature-entropy diagram.

SUPPOSE WE MODIFIED the Rankine Cycle above to make it *more like* a Carnot cycle. We could reject heat only to state point 1' in Figure 6 and apply work to isentropically compress the mixture until it became a saturated liquid at 838 psia, state point 2'. What would the thermal efficiency be for that cycle?

The answer is 41.5 percent just like the Carnot cycle. You would also find that the net specific work for the cycle is somewhat reduced. Is that what you expect? Again, the answer should be "yes"; by changing state points 1 and 2 we reduced the area contained within the T-S diagram and that corresponds with reduced net specific work.

Exergy Analysis - Basic Definitions

In order to analyze quantitatively the extent to which a cycle has reversibility we will need to perform an *Exergy Analysis*.

Definitions

The following definitions are provided below:

- Exergy* is the maximum theoretically obtainable work that can be extracted from a working fluid as it comes into equilibrium with the environment.

¹¹ If we're analyzing an AP1000 PWR energy conversion cycle, a better choice for T_h would be the hot-leg temperature which is approximately 610 °F. We will come back to this in a future lecture.

¹² Again, a better choice for T_c for the AP1000 would be the inlet temperature of cooling water to the condenser which we might conservatively take to be 91 °F.

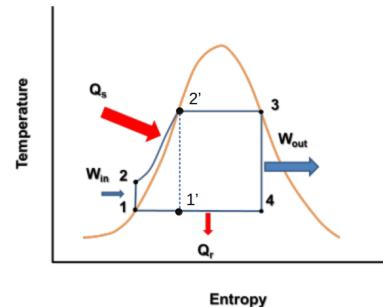


Figure 6: Temperature-Entropy plot of a modified Rankine cycle.

The course textbook does an *irreversibility* analysis which is equivalent.

Note: exergy is defined for a working fluid only *relative* to the environment. Working fluid of a given state will have a different exergy if environmental temperature or pressure goes up or down.

- *Dead state* is a state in which a working fluid is at rest relative to the environment. The dead state will normally be characterized by dead state temperature (T_o) and dead state pressure (P_o).
- *Specific Flow Exergy* (e_f) is an intrinsic thermodynamic variable quantifying the ability of a working fluid to do work. It is calculated relative to the dead state as follows:

$$e_f = h - h_o - T_o(s - s_o) + \frac{V^2}{2} + gz$$

where h and s are the enthalpy and entropy of the working fluid; h_o and s_o are the enthalpy and entropy at the dead state temperature and pressure; T_o is the dead state temperature; and $V^2/2$ and gz represent the kinetic and potential energy respectively of the fluid.¹³

Exergy Balance

UNLIKE ENERGY, exergy is not conserved. One can think of exergy like energy: we pass exergy in from a high energy source like the primary coolant of a PWR; we convert some of that exergy to work and some that is left over we have to reject to the environment. But unlike energy, exergy can be destroyed. Exergy can also be lost through other means like heat transfer to the surrounding medium.¹⁴

We want to account for exergy in much the same way that we account for energy. In words, this accounting balance looks like this:

$$\text{Exergy in} - \text{Exergy out} = \text{Exergy xfer by heat} + \text{Exergy xfer by work} + \text{Exergy Destroyed}$$

In equation form this translates as follows:

$$\sum_i \dot{m}_i e_{f_i} - \sum_e \dot{m}_e e_{f_e} = \sum_j \left(1 - \frac{T_o}{T_j}\right) \dot{Q}_j + \dot{W} + \dot{E}_d$$

The terms are interpreted as follows:

- $\sum_i \dot{m}_i e_{f_i}$ is the sum of all flow exergy going *in* to a particular process. The variable i corresponds to all of the inflows; \dot{m}_i is the mass flow rate for each in-flow and e_{f_i} is the specific flow exergy for each inflow.
- $\sum_e \dot{m}_e e_{f_e}$ has the same meaning but for all flows *exiting* a particular process.

¹³ Unless otherwise noted, changes in kinetic and potential energy will be neglected in this class so, for calculations of e_f , those terms are excluded.

¹⁴ For example: steam piping passing through the engineeroom of a submarine is hot; despite its lagging, it transfers some energy to the surrounding environment. This is a loss of exergy that can be taken into account.

- $\sum_j (1 - T_o/T_j) \dot{Q}_j$ is exergy transfer through heat. The variable j enumerates all of the exit points for exergy transfer through heat for the system. \dot{Q}_j is the corresponding rate of heat loss, T_j is the temperature at which this heat is lost, and T_o is the dead state temperature.
- \dot{W} is the rate of exergy transfer through work. Any process that does work (e.g. turbine, pump, compressor) transfers exergy out of (turbine) or into (pump or compressor) the working fluid.
- \dot{E}_d is the rate of exergy destruction. This should be interpreted as the rate at which exergy is lost due to irreversibilities. Any non-reversible process will have positive exergy destruction.¹⁵

¹⁵ Exergy destruction rate for a process should *never* be negative. If your calculations show a negative exergy destruction rate (other than rounding errors near zero) for any process you have somehow violated the Second Law of Thermodynamics. You need to stop and find the error before moving forward.

Lecture 4 - Rankine Cycle Exergy Accounting

Objectives

The objectives of this lecture are:

- Illustrate exergy accounting with a simple Rankine cycle
- Show how to carry out the calculations in a MATLAB environment using EasyProp.

Simple Ideal Rankine Cycle - Exergy Accounting

CONSIDER AGAIN the Rankine cycle with system parameters specified to resemble a highly simplified version of a Westinghouse AP1000. A schematic of this system is given in Figure 7. Our goal

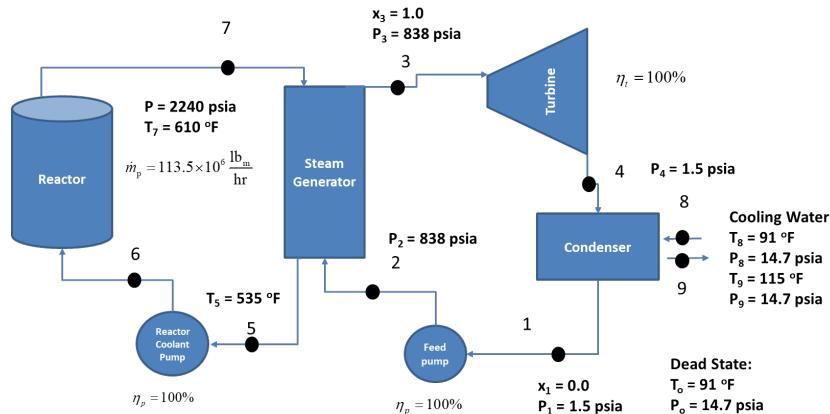


Figure 7: Simple ideal Rankine cycle modeled on the Westinghouse AP1000.

is to carry out the exergy accounting as described in Lecture 3 for this system.

WE WILL FOLLOW a step-by-step process to complete this analysis.

- Carry out the basic “First Law” of thermodynamics analysis for this system. This was done in Lecture 2; the details will not be repeated here except to remind you that we have instantiated an object to calculate thermodynamic properties of the working fluid, water:

```

1 units = 'USCS';
2 fluid = 'Water';
3 water = py.EasyProp.simpleFluid(fluid, units);

```

Listing 7: Construct a *simpleFluid* object

- Determine dead state enthalpy (h_o) and entropy (s_o). With the given dead state temperature and pressure, we can find this as follows:

```

1 To_F = 91; % F, dead state temp
2 Po = 14.7; % psia, dead state pressure
3 ho = water.h_pT(Po,To_F);
4 so = water.s_pT(Po,To_F);
5 To = To_F + 460; % R, dead state absolute temperature

```

Listing 8: Find dead state enthalpy and entropy

Note that EasyProp functions require the temperature in either C or F, but the T_o appearing in the specific flow exergy equations must be in absolute units: K or R.

- Calculate the specific flow exergy for all state points. You should calculate all state point enthalpy and entropy values as part of your “First Law” analysis. In MATLAB one can then create an in-line function to quickly calculate the specific flow exergy for all state points.

```

1 ef_fun = @(h,s) (h - ho) - To*(s - so);
2 ef = ef_fun(h,s); % BTU/lbm, specific flow exergy

```

Listing 9: Compute specific flow exergy for all state points

Note that the arrays for h and s should be populated with the correct enthalpy and entropy for all state points before this calculation is made.

- Calculate the exergy destruction rate for each process. In each process this is calculated from a re-arrangement of the exergy balance equation given in Lecture 3:

$$\dot{E}_d = \sum_i \dot{m}_i e_{f_i} - \sum_e \dot{m}_e e_{f_e} - \dot{W} \quad (4)$$

In this cycle there are four processes to consider:

- (a) **Exergy destruction rate in the steam generator.** Focus on the relevant process showing in Figure 8. The steam generator has two in-flows and two exits; one each from the primary and secondary side. For this particular problem we are only given the mass flow rate of the primary (\dot{m}_p) so we first need to get the mass flow rate in the secondary. This can be obtained from simple conservation of energy principles:

$$\dot{m}_s(h_3 - h_2) = \dot{m}_p(h_7 - h_5)$$

$$\text{m_s} = \text{m_p} * (\text{h}_7 - \text{h}_5) / (\text{h}_3 - \text{h}_2); \text{ % lbm/hr, mass flow rate through S/G.}$$

Now, using Equation 4, we compute the exergy destruction rate:

$$1 \quad \text{Ex_d_sg} = \text{m_p} * \text{ef}(7) + \text{m_s} * \text{ef}(2) - (\text{m_p} * \text{ef}(5) + \text{m_s} * \text{ef}(3)); \text{ % BTU/hr}$$

Note that since there is no work done in the steam generator, there is no exergy transfer through work.

- (b) **Exergy destruction rate in the turbine.** The relevant process here is shown in Figure 9. Note that since there is work being done in the turbine, there is exergy transfer through work that we need to take into account.

$$1 \quad \text{Ex_d_turb} = \text{m_s} * (\text{ef}(3) - \text{ef}(4)) - \text{m_s} * w_{\text{turb}}; \text{ % BTU/hr}$$

Here we expect w_{turb} , the specific work of the turbine, to have previously been calculated in the "First Law" analysis as $w_{\text{turb}} = h(3) - h(4)$.¹⁶

- (c) **Exergy destruction rate in the condenser.** Analysis of the condenser is much like the steam generator: two in-flows and two exit-flows and no work done in the process. Also in this case, like the steam generator, we need to use conservation of energy to determine the required flow rate of the condenser cooling water \dot{m}_c . The relevant process schematic is illustrated in Figure 10.

$$1 \quad \text{m_c} = \text{m_s} * (\text{h}(4) - \text{h}(1)) / (\text{h}(9) - \text{h}(8)); \text{ % lbm/hr, condenser cooling water flow}$$

$$3 \quad \text{Ex_d_cnd} = \text{m_c} * \text{ef}(8) + \text{m_s} * \text{ef}(4) - (\text{m_c} * \text{ef}(9) + \text{m_s} * \text{ef}(1)); \text{ % BTU/hr}$$

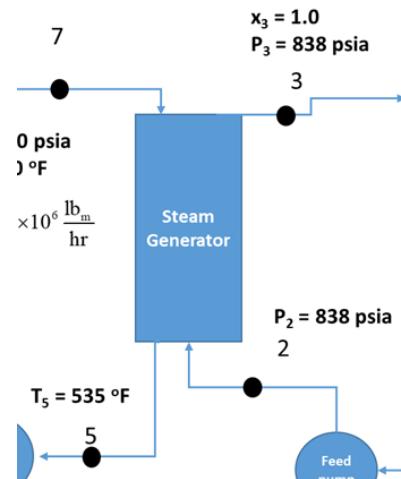


Figure 8: Constant pressure heat transfer process in the steam generator

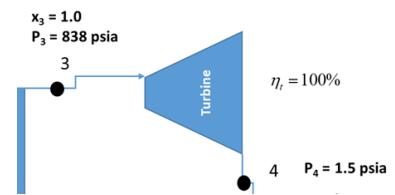


Figure 9: Expansion of the working fluid in the turbine.

¹⁶ Note: for this simple ideal cycle, you should find that the exergy destruction rate for the ideal turbine to be zero!

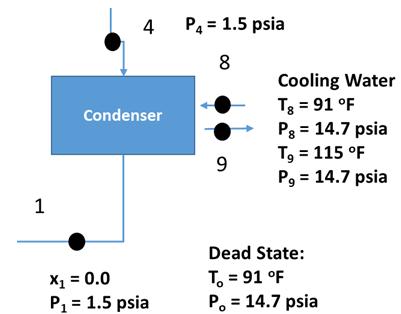


Figure 10: Constant pressure heat rejection in the condenser.

- (d) Exergy destruction rate in the feed pump. Like the turbine, the feed pump has one inlet and one exit and there is also exergy transfer through work. The relevant process schematic is illustrated in Figure 11. The MATLAB code would be:

```
1 Ex_d_fp = m_s*(ef(1) - ef(2)) - m_s*w_fp; % BTU/hr
```

where w_{fp} would be calculated simply as $w_{fp} = h(1) - h(2)$.

5. Calculate the “Exergy In” to the system from the primary coolant and “Exergy Out” from the system to the environment via the condenser. For the Exergy In we will simply use the mass flow rate of primary coolant times the change in specific flow exergy of the primary coolant through the steam generator; for Exergy Out we will use the mass flow rate of condenser cooling water times the increase in specific flow exergy of the water.

```
1 Ex_in = m_p*(ef(7) - ef(6)); % BTU/lbm
2 Ex_out = m_c*(ef(9) - ef(8)); % BTU/lbm
```

6. Verify the overall exergy balance. The balance should look something like:

```
balance = Ex_in - Ex_out - m_s*w_net - ...
2 (Ex_d_sg + Ex_d_cnd + Ex_d_turb + Ex_d_fp);
```

If everything is correct and if you have not violated the Second Law of Thermodynamics the computed balance should be zero. In exactly the same way that you should verify that net specific work is the same as net specific heat transferred ($w_{net} = q_{net}$) — thereby satisfying conservation of energy (First Law of Thermodynamics) — you should check the balance on the exergy analysis to ensure you have satisfied the Second Law of Thermodynamics.

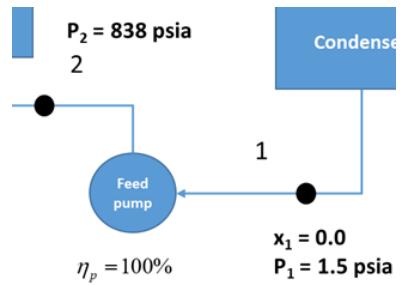


Figure 11: Work in to the working fluid from the feed pump.

Assignment #2: Rankine Cycle 2nd Law Analysis

Background

Russia has recently placed a nuclear powered barge into operation named the *Akademik Lomonosov*. This vessel is not self-propelled but is equipped with two pressurized water reactors designated as KLT-40S for providing up to 35 MW of electrical power (each).

We will model the power conversion system for the KLT-40S as a simple Rankine Cycle with a Steam Generator (SG), a Turbine Generator (TG), a condenser and a main feed pump (MFP). Thermal energy is supplied by primary coolant of a pressurized water reactor (PWR); waste heat is rejected to seawater which is modeled as ordinary water. Relevant design parameters are given in Table 4

Parameter	Value
Primary coolant pressure	12.7 MPa
Primary coolant SG inlet temperature	317°C
Primary coolant SG outlet temperature	280°C
SG pressure	3.8 MPa
SG steam outlet temperature	290°C
Steam mass flow rate	67 kg/s
Dead state temperature and pressure	5°C, 101 kPa
Condenser pressure	5 kPa
Condenser seawater inlet temperature	5°C
Seawater cooling temperature rise	10°C
MFP isentropic efficiency	65%
TG steam outlet quality	80%

Table 4: *Akademik Lomonosov* Nuclear Steam Supply System key parameters.

ANALYZE THE Rankine cycle described and answer the following questions:

1. What is the isentropic efficiency of the TG?
2. What is the net specific work of the Rankine cycle? [kJ/kg]

3. Verify that the net specific heat is the same as the net specific work.

4. What is the mass flow rate of the primary coolant? [kg/s]

5. What is the mass flow rate of the condenser cooling water? [kg/s]

6. What is the rate of exergy input (to the Rankine cycle) from the primary coolant? [kW]

7. What is the rate of exergy rejection to the environment via condenser cooling water? [kW]

8. Compute the rate of exergy destruction in the Rankine cycle components; fill in the table below:

Component	Rate of Exergy Destruction [kW]
Steam generator	
Turbine generator	
Condenser	
Main feed pump	
Total Exergy Destruction Rate	

9. What is the rate of exergy transfer through work? [kW]

10. Verify the 2nd law of thermodynamic balance:

$$\text{Exergy in} - \text{Exergy out} = \dot{W} + \dot{E}_d$$

Lecture 5 - Improving Rankine Cycle Performance

Objectives

The objectives of this lecture are:

- Describe performance parameters for Rankine Cycles
- Illustrate benefits of Rankine Cycle with a Moisture Separator and Open Feedwater Heater
- Show how adding a Reheater improves Performance
- Include modeling details essential for cycle analysis

Performance Changes with Increasing Steam Generator Pressure

BASED ON our discussion from last lecture, it seems obvious enough that one way to improve Carnot efficiency and thus (hopefully!) increasing Rankine cycle performance would be to increase the steam generator temperature. Since the steam generator is a saturated system, such a change implies an increase to steam generator pressure. Let us assume for the time being that nothing limits us from increasing steam generator temperature and pressure; what are the thermodynamic consequences? To start, let us examine the change on a temperature-entropy plot as is shown in Figure 12.

A FEW THINGS are apparent from this picture:

1. Since the temperature at which the majority of heat has been added to the system is increased, Carnot efficiency will increase and, all things being equal, the thermal efficiency that we actually achieve will also increase.
2. While the schematic is not drawn to scale, it suggests that net specific work will also increase at least a little.
3. The turbine exit quality decreases after we increase the steam generator pressure.

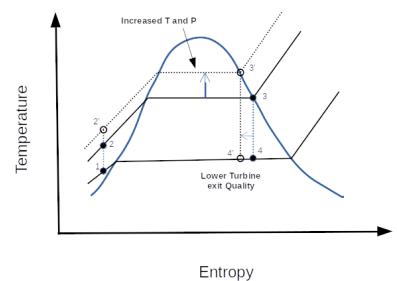


Figure 12: Temperature Entropy plot showing an increase in steam generator pressure.

The last two points on the above list might be clarified with a temperature-entropy plot that is drawn to scale. Such a plot can be made with EasyProp tools and is shown in Figure 13.

INCREASING EFFICIENCY IS a good goal but, in this case, the attendant reduction in turbine exit quality is a significant issue. Droplets of saturated vapor in the steam can result in excessive turbine wear and reduces turbine isentropic efficiency. The good news is that we can make changes to the cycle that address the issue with turbine exit quality while further increasing cycle thermal efficiency.

Rankine Cycle with Moisture Separator and Open Feedwater Heater

We can increase steam quality at the low-pressure stages of the turbines if, part way through the expansion process, we include a component that separates saturated liquid from the working fluid. Such an apparatus is called a Moisture Separator (M/S) and is shown in the cycle schematic in Figure 14.

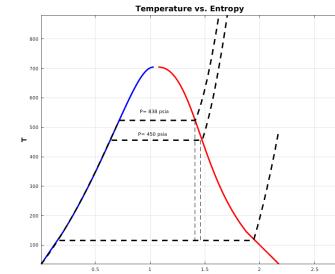
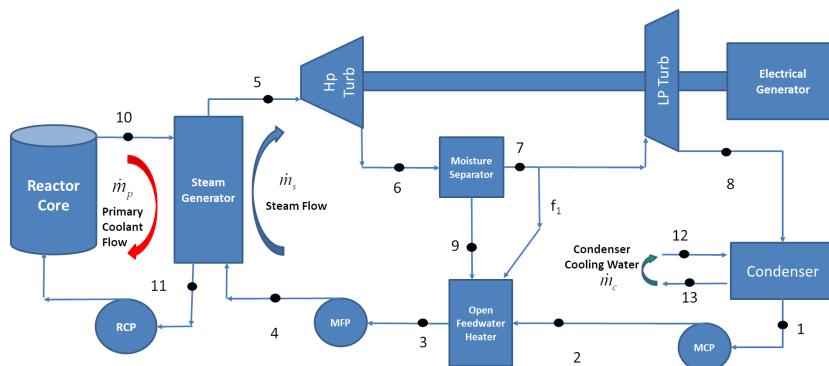


Figure 13: Temperature Entropy plot for steam pressure increase drawn to scale with EasyProp.

Figure 14: Rankine Cycle with moisture separator and open feedwater heater.

The moisture separator takes steam exiting the high pressure (HP) turbine at state point 6 which is a saturated mixture. The portion of the steam that is saturated vapor¹⁷ exits the moisture separator to state point 7. The portion of the steam that is saturated liquid¹⁸ is drained to the Open Feedwater Heater (OFWH).

THERE ARE THREE streams going into the OFWH: discharge of the main condensate pump (state point 2); saturated liquid draining from the M/S (state point 9); and additional saturated vapor extracted from the steam flow path at the steam exit from the M/S (state point 7). This last flow extraction is used to further pre-heat the feedwater in the OFWH. Conventionally we set the extraction steam flow fraction (f) such that the fluid exiting the OFWH at state point 3 is near

¹⁷ This fraction is easily obtained as the quality at state point 6, $x(6)$, using EasyProp.

¹⁸ Again, this can be found as the complement of quality at state point 6.

Note: The OFWH is “open” because all fluids flowing into the tank mix freely. This has the added consequence that all fluid streams going into or out of the OFWH must be maintained at the same pressure.

its saturation temperature.¹⁹

THE TEMPERATURE-ENTROPY PLOT for this modified cycle is shown in Figure 15. This illustrates how, at least, the problem of turbine exit quality is addressed.

One thing the figure does not show quantitatively is that the regeneration effect of pre-heating the feedwater in the OFWH has the result of increasing plant thermal efficiency. This benefit comes at the cost of another effect that the temperature-entropy plot fails to highlight; the mass flow rate of steam through the low-pressure (LP) turbine is reduced thereby lowering specific work output from the cycle.

"First Law" Analysis of Modified Cycle

THE ANALYSIS OF THIS CYCLE requires some modifications. The reason for this is, owing to the separation of vapor and liquid in the M/S, the working fluid has different mass flow rates at different portions of the system. For example, for a given mass flow rate through the S/G (\dot{m}_s), then the mass flow rate of steam exiting the M/S at state point 7 is $\dot{m}_s x(6)$, where $x(6)$ is the quality at state point 6.²⁰ Before the steam enters the LP turbine, the flow rate is further reduced since a fraction f is directed to the OFWH for regeneration. Thus the mass flow rate of steam entering the LP turbine is: $\dot{m}_s x(6)(1 - f)$.

AS A CONVENTION we will re-define "specific work" of a component (w) to be equal to the power from a component (\dot{W}) divided by the mass flow rate through the steam generator (\dot{m}_s). See the examples below.

- **Example:** the specific work of the HP turbine (w_{HP}) is given by:

$$w_{HP} = \frac{\dot{m}_s(h(5) - h(6))}{\dot{m}_s} = h(5) - h(6)$$

which happens to be no different than what we would have calculated before.

- **Example:** the specific work of the LP turbine (w_{LP}) is given by:

$$w_{LP} = \frac{\dot{m}_s x(6)(1 - f)(h(7) - h(8))}{\dot{m}_s} = x(6)(1 - f)(h(7) - h(8))$$

This expression takes into account the reduced mass flow rate through the LP turbine due to the moisture separator and extraction steam.

¹⁹ We will adopt the modeling convention that the fraction f extracted from the steam flow is set so the outflow from the OFWH is a saturated liquid.

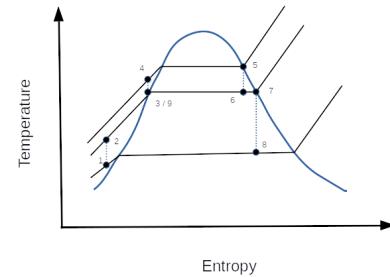


Figure 15: Temperature-entropy plot of Rankine cycle with M/S and OFWH.

²⁰ For this cycle where saturated vapor ($x=1$) is introduced into the HP turbine, $x(6)$ is generally less than 1.

- **Example:** the specific work of the main condensate pump (MCP) is, for this cycle, similar to the LP turbine:

$$w_{\text{MCP}} = \frac{\dot{m}_s x(6)(1-f)(h(1)-h(2))}{\dot{m}_s} = x(6)(1-f)(h(1)-h(2))$$

- **Example:** the specific work of the main feed pump (MFP) is:

$$w_{\text{MFP}} = \frac{\dot{m}_s(h(3)-h(4))}{\dot{m}_s} = h(3)-h(4)$$

Use of this convention makes calculation of net specific work easy: we simply add up the specific work of each component in the system: $w_{\text{net}} = w_{\text{HP}} + w_{\text{LP}} + w_{\text{MCP}} + w_{\text{MFP}}$. Calculation of specific heat added is done the same way.

CARE MUST ALSO be taken for calculation of energy balances in the system. When analyzing the simple ideal Rankine cycle, we used an energy balance on the condenser in order to determine condenser cooling water mass flow rate (\dot{m}_c). For this cycle an equivalent analysis gives:

$$\dot{m}_c(h(13)-h(12)) = \dot{m}_s x(6)(1-f)(h(8)-h(1))$$

Similarly, in order to deduce what the extraction steam flow fraction (f) must be, we will need to do an energy balance on the OFWH. For this we will equate the rate of energy exiting the tank to the rate of energy entering the tank:

$$\Sigma \dot{E}_{\text{out}} = \Sigma \dot{E}_{\text{in}}$$

$$\dot{m}_s h(3) = \dot{m}_s(1-x(6))h(9) + \dot{m}_s x(6)f h(7) + \dot{m}_s x(6)(1-f)h(2)$$

Note: Since \dot{m}_s appears in every term of the equations, it is convenient to eliminate that term from the equation.

Rankine Cycle with Reheating

We can further reduce moisture in the low-pressure sections of turbines by re-heating the steam after it exits the moisture separators.²¹ A schematic of such a system is shown in Figure 16.

The corresponding temperature-entropy sketch is provided in Figure 17. Note that the quality of the steam exiting the LP turbine at state point 9 has increased.

²¹ In typical energy conversion cycles for commercial nuclear power plants, moisture separators and reheaters are combined into a single component called a "moisture-separator-reheater".

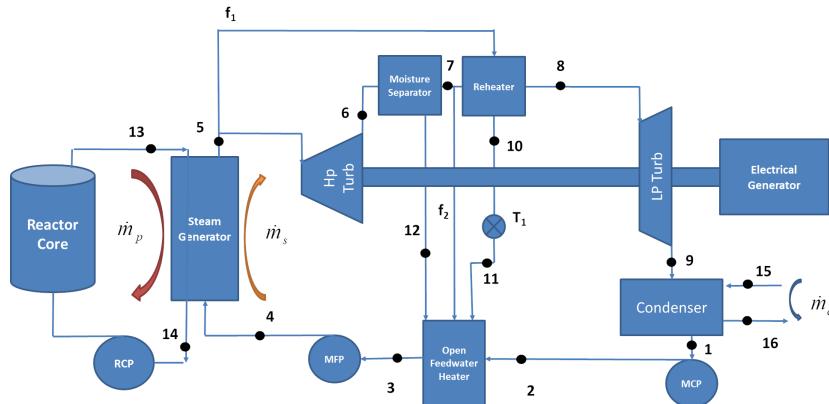


Figure 16: Rankine cycle with MS, OFWH and Reheat.

THERE ARE A COUPLE of notable features of this cycle:

1. We use steam extracted upstream of the HP turbine as the heat source to re-heat steam exiting the moisture separator.
2. A new type of system component has been used between state points 10 and 11. This device, which is referred to as a “trap” (T_1), allows for isenthalpic expansion of a working fluid; this is needed since the fluid at state point 10—which we will conventionally model to be in the saturated liquid state—is at higher pressure than the OFWH. As a result of this expansion, the fluid at state point 11 will typically be a saturated mixture.

“First-Law” Analysis of Modified System

This cycle is more complicated than the last one. We have two flow extractions (f_1 and f_2) that we need to solve for. Consequently, we will need two equations to find a solution. These equations will come from the energy balance for the open feedwater heater and the re-heater.

USING THE CONVENTIONS for calculating flow fractions established in the last example, the energy balance for the OFWH is given as follows:

$$\begin{aligned} \Sigma \dot{E}_{\text{out}} &= \Sigma \dot{E}_{\text{in}} \\ \dot{m}_s h(3) &= \dot{m}_s [(1 - f(1))(1 - x(6))h(12) + (1 - f(1))x(6)f(2)h(7) \dots \\ &\quad + f(1)h(11) + (1 - f(1))(x(6))(1 - f(2))h(2)] \end{aligned}$$

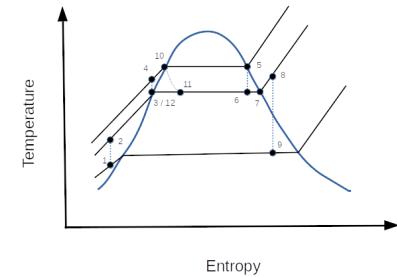


Figure 17: Temperature-entropy plot of modified cycle.

Note: Re-heat steam mass flow rate is typically around 5 percent of \dot{m}_s .

Isenthalpic means “constant enthalpy.”

HEALTH WARNING: you should take some time to study this equation; make sure every term makes sense and build confidence that you could derive this yourself.

The energy balance for the re-heater reads as follows:

$$\Sigma \dot{E}_{\text{out}} = \Sigma \dot{E}_{\text{in}}$$
$$\dot{m}_s(1 - f(1))x(6)(1 - f(2))(h(8) - h(7)) = \dot{m}_s f(1)(h(5) - h(10))$$

Both of these equations must be solved simultaneously to find $f(1)$ and $f(2)$. These are both non-linear equations and, for general cycle analyses, can be quite difficult to find without computational help. In the next lecture we will discuss MATLAB tools for carrying out the required calculations.

Lecture 6 - Solving Multi-Variable Non-linear Equations in MATLAB

Objectives

The objectives of this lecture are:

- Describe the types of multi-variable non-linear equations that arise in analyzing complex thermodynamic cycles; and
- Describe in detail how to use the built-in MATLAB function FMINCON

Multi-variable Non-linear Equation

In the last lecture, we analyzed a Rankine cycle with a moisture separator, re-heater, and an open feedwater heat exchanger. In performing the analysis of the cycle, we obtained two non-linear equations expressing conservation of energy for the re-heater and the open feedwater heater.

For the OFWH:

$$\dot{m}_s h(3) = \dot{m}_s [(1 - f(1))(1 - x(6))h(12) + (1 - f(1))x(6)f(2)h(7) + \dots f(1)h(11) + (1 - f(1))(x(6))(1 - f(2))h(2)] \quad (5)$$

For the re-heater:

$$\dot{m}_s (1 - f(1))x(6)(1 - f(2))(h(8) - h(7)) = \dot{m}_s f(1)(h(5) - h(10)) \quad (6)$$

THESE EQUATIONS WILL BE presented in MATLAB format to facilitate the discussion. As the equations are re-written in MATLAB form, we will also re-formulate the equations so they are amenable to solution with a tool like FMINCON. Specifically, we will put all terms of the equation on one side.

Note: For this example, assume that all state point property values and mass flow rates such as h , x , and \dot{m}_s are known.

This will be the case for most cycles you analyze; you will be given enough information to find most/all state point properties but you will need to solve the non-linear equations to get the flow rates.

```

2 % f = [f1 ,f2 ]
3 OFWH_heatBalance = @(f) (1-f(1))*(1-f(2))*x(6)*(h2) + ...
4 f(1)*h(11) + (1-f(1))*x(6)*f(2)*h(7) + ...
5 (1-f(1))*(1-x(6))*h(12) - h(3);
6 ReHeater_heatBalance = @(f) f(1)*h(5) + ...
7 (1-f(1))*x(6)*(1-f(2))*h(7) - ...
8 ( f(1)*h(10) + (1-f(1))*x(6)*(1-f(2))*h(8) );

```

Listing 10: Energy balance equations in MATLAB format

Now we have two in-line functions that, given the correct values for f_1 and f_2 , will be equal to zero. We have re-cast our problem into a “root-finding” problem; one that many computational tools have been developed to solve.

THE WAY we are going to solve this is using the built-in MATLAB function FMINCON. FMINCON is a function for solving a constrained minimization problem for a non-linear function with multiple variables. FMINCON expects only one equation to solve, so we will need to combine the two equations above into a single equation. We will do this as follows:

```

balance = @(f) abs(OFWH_heatBalance(f)) + ...
2 abs(ReHeater_heatBalance(f));

```

Listing 11: Combine two heat balance equations into one.

We want to find the values of $f(1)$ and $f(2)$ such that “balance” is minimized. It should be clear that:

1. minimum possible value of “balance” is zero; and
2. “balance = 0” corresponds to satisfying both of the energy conservation equations for the OFWH and the RH.²²
3. We need to find this minimum value while constraining $f(1)$ and $f(2)$ to the interval $[0, 1]$.²³

THE MATLAB FUNCTION FMINCON attempts to find a constrained minimum of a function of several variables. The function signature is:

```
X = fmincon(FUN,Xo,A,B,Aeq,Beq,LB,UB,NONLCON,OPTIONS);
```

The arguments for FMINCON are as follows:

- FUN: this is the function that you hope to minimize; as illustrated here, this is the function “balance.”
- Xo: this is a vector providing the initial guess for each variable in the function. For this example Xo must have two entries, one for

²² This is only true because we use the absolute value function on each heat balance equation individually. If we omitted the absolute value function then the minimization algorithm would drive the balance function towards negative infinity; if we used the absolute value function on the sum of the functions then positive errors for the OFWH could be offset by negative errors for the re-heater.

²³ Referencing the schematic from the previous lecture, $f(1)$ is the fraction of steam drawn off for re-heat and $f(2)$ is the fraction of steam flowing out of the moisture separator extracted and sent to the OFWH. Since both are defined as flow fractions, they cannot be negative and they cannot be greater than 1, corresponding to 100 percent.

the initial guess for $f(1)$ and $f(2)$. A reasonable value for X_0 in this case would be: [0.5 0.5]

- A and B. These arguments allow the user to express linear inequality constraints between the non-linear variables. The inequalities are expressed as: $AX \leq B$ so if, for example, we want to enforce an inequality such as $f(1) > f(2)$, this could be written mathematically as $-f(1) + f(2) \leq 0$; we would set $A = [-1, 1]$; and $B = [0]$. For the example we are using for this lecture, no such inequality constraint exists so we would set $A = []$ and $B = []$.
- Aeq and Beq. This is the same as above but are equality constraints instead of inequality constraints. If, for example, we want to enforce the constraint that $f(1) + f(2) = 1$, we would set $A = [1, 1]$ and $B = [1]$. For this problem we have no such constraints so we should set $Aeq = []$ and $Beq = []$.
- LB and UB: these arguments are where you provide the lower bound and upper bound for each of your variables. For multi-variable problems, these should be entered as vectors.
- NLCON: this argument allows for you to provide non-linear constraints between the dependent variables. We will not be using this functionality and set this argument to $NLCON = []$.
- OPTIONS. This argument allows you to provide keyword-value pairs for various options. The argument must have a specific form and is most easily set using another built-in function “optimoptions”. Use of this feature will be illustrated in the next code listing.

Using this information, we can provide values for the function arguments and find the constrained minimum for our equations as follows:

```

1 xo = [0.5 0.5]; % a reasonable guess within the lower and upper
2   bounds of the variables
3 A = []; % no linear inequality constraints
4 B = [];
5 Aeq = []; % no linear equality constraints
6 Beq = [];
7 lb = [0 0];
8 ub = [1 1];
9 nlcon = []; % no nonlinear constraints
10 options = optimoptions('fmincon','Display','none'); % suppress
11   extensive output
12 % other commonly used option sets:
13 % options = optimoptions('fmincon','Display','iter');
14 % options = optimoptions('fmincon','Algorithm','sqp');
15 % select constrained minimization algorithm. Default is '
16   interior-point'
17 % other options: 'sqp','sqp-legacy','active-set',

```

```
15 % and 'trust-region-reflective'. See documentation.  
16 %  
17 [f ,fval ,exitflag ] = fmincon(balance ,xo ,A,B,Aeq,Beq,lb ,ub ,nlcon ,  
18 options );  
19 % report the results  
20 fprintf('Flow fraction for reheat steam = %5.4f \n' ,f(1));  
21 fprintf('Flow fraction to OFWH from M/S exit = %5.4f \n' ,f(2));
```

Listing 12: Set arguments and call FMINCON to find constrained minimum

Lecture 7 - Rankine Cycle with Closed Feedwater Heaters

Objectives

The objectives of this lecture are:

- Describe Closed Feedwater Heaters (CFWH)
- Discuss the “Cascade Back” and “Pump Forward” arrangements for dealing with CFWH drains.
- Provide some design and modeling notes.

Closed Feedwater Heaters

IN A PREVIOUS LECTURE we learned about open feedwater heaters (OFGH). OFGHs are used in the Rankine Cycle to pre-heat feedwater before entering the Steam Generator. In workshop exercises you could see how use of OFGHs reduce overall exergy destruction (irreversibility) and improve cycle thermal efficiency.

Closed feedwater heaters (CFWHs) perform the same role. The differences is that in CFWHs the fluids do not mix. A simple schematic is given in Figure 18.

THE HIGH-TEMPERATURE FLUID is normally steam, extracted from a selected turbine stage, and the low-temperature fluid being heated is normally feedwater. The low-temperature water is, for all of our applications, a subcooled liquid. The high-temperature fluid is, for most nuclear applications, a saturated mixture of relatively high quality ($x \geq 80$ percent) although in principle it could be superheated. The outlet for high temperature fluid is normally a saturated or slightly subcooled liquid. For CFWHs the outlet of the low temperature fluid is usually still a subcooled liquid albeit at a higher temperature than it was at the inlet.

Although it is beyond the scope of this class, it is worthwhile to consider the added complexity of modeling—and at some point

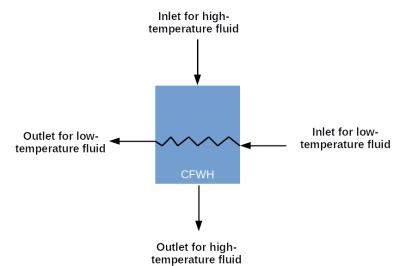


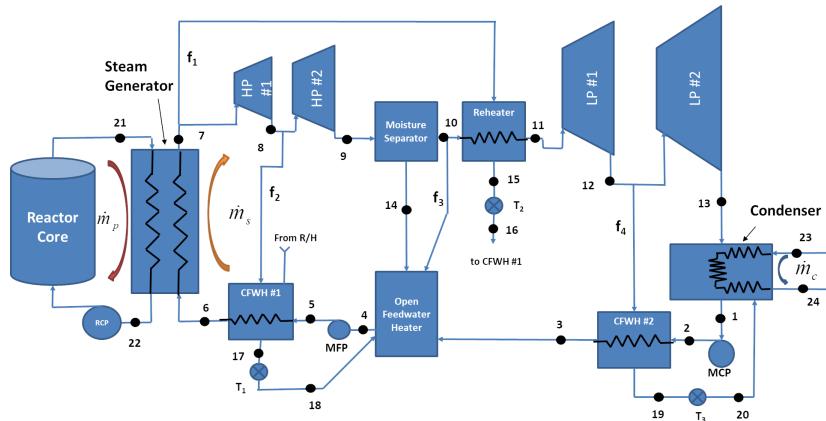
Figure 18: Simple CFWH schematic.

in our career designing—a heat exchanger where one of the fluids undergoes phase changes. A somewhat more realistic schematic representation of a CFWH is showing in Figure 19. Here, in the case that the extraction steam is above saturation temperature, the steam is reduced to saturation temperature in the “de-superheater” region on the heat exchanger. Heat transfer correlations used for single-phase convective heat transfer from the steam to the heat exchanger tubes would be used to model heat exchanger performance. In the “condenser” region, the complicated phase change process occurs where steam is converted to saturated liquid. In the “condensate cooler” region, once again single-phase heat transfer prevails as the liquid water gives up its energy to the incoming feed flow.

ONCE THE HEATING FLUID exits the CFWH, it has to go somewhere. In OFWHs, the heating fluid was simply carried away with the (now preheated) feedwater. In a CFWH, the high-temperature fluid exit needs to be directed somewhere else to be integrated into the feed-flow stream. There are two conventional choices for doing this:

1. “Drain Cascade Back”—type 1
2. “Drain Pump Forward”—type 2

The Rankine cycle schematic shown in Figure 20 illustrates a type 1 arrangement. CFWH #1 has two high-temperature fluid inputs:



one is extraction steam from the turbine labeled “HP #1”; the other comes from the re-heater exit. The heating fluid in the CFWH all exits through a common drain. This drain is directed to the OFWH. Since all fluids entering the OFWH must be at the same pressure, this fluid is expanded through a trap (labeled T_1) entering the OFWH at state point 18.

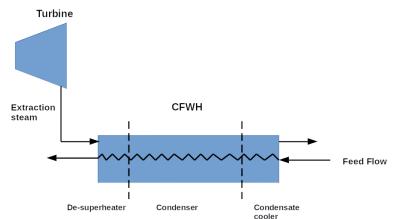
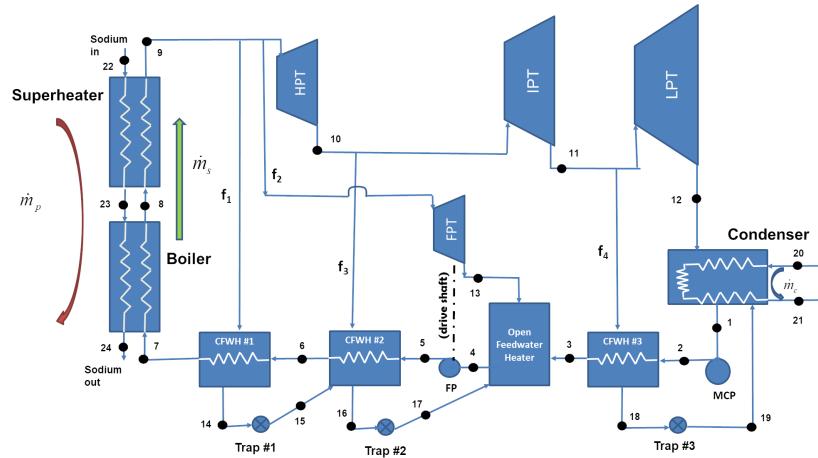


Figure 19: Conceptual schematic of a CFWH.

Figure 20: Complex Rankine cycle with type 1 “Drain Cascade Back” CFWH arrangement.

Question: Why is it probably better to direct the fluid at state point 16 to CFWH #1 rather than to the OFWH, CFWH #2, or the Condenser?

A CYCLE WITH MULTIPLE CFWHs will normally cascade the heat exchanger drains in order to extract maximum reheating benefit from the extraction steam. A simplified schematic of Rankine cycle employed with the Experimental Breeder Reactor II is presented in Figure 21.²⁴



²⁴ Leonard J Koch. *Experimental Breeder Reactor-II (EBR-II): An Integrated Experimental Fast Reactor Nuclear Power Station*. Amer Nuclear Society, 2008

Figure 21: Simplified schematic of the Experimental Breeder Reactor II energy conversion cycle with cascaded CFWH drains.

A TYPE 2 ARRANGEMENT for a CFWH is illustrated in Figure 22. Notice the extra state point required at the exit of the CFWH after the point where the drain pump discharge is mixed in to the feedwater flow.

Design and Modeling Notes

There are too many details on the variations of Rankine cycle designs in order to hope to cover them all in one course, much less one lecture. Here I provide a set of bullet-points to try and summarize some general principles that one can get by examining modern and relevant Rankine cycles for nuclear energy conversion:

- Rankine cycles typically utilize a mixture of OFWHs and CFWHs (e.g. 5-6 CFWHs + 1 or 2 OFWHs).
- BWRs generally will not use OFWHs due to the need to contain gaseous radionuclides in the coolant.²⁵
- High pressure / low temperature outlet from the CFWH is usually modeled as having the same (or nearly the same) temperature as the incoming heating steam.
- High temperature / low pressure outlet from CFWH is usually modeled as a saturated liquid (or slightly sub-cooled).

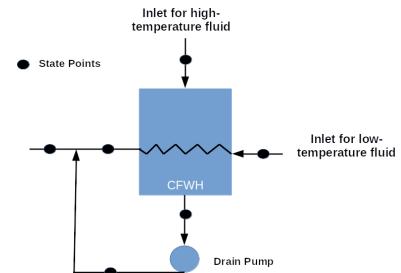


Figure 22: Schematic of CFWH with type 2 drain pump arrangement.

²⁵ OFWHs usually include a deaerator function where air and other non-condensable gasses are vented from the feedwater and ejected to the atmosphere. In a BWR some of those gasses may be radioactive.

- Extraction steam pressures are often chosen such that the steam temperature (in each extraction stream) is reduced in roughly equal increments.
- “Drain Pumped Forward” CFWHs result in slightly higher thermal efficiency but:
 1. Requires another pump (that might break); and
 2. Combinations of drain-forward and drain cascade back possible (of course).
- How much is it worth to increase thermal efficiency a little bit by adding all of this complexity:
 - For a navy nuclear powered warship?
 - for a commercial nuclear power plant operator?

Assignment #3: Rankine Cycle, Enhanced Regeneration

CONSIDER THE Rankine cycle depicted in the figure below:

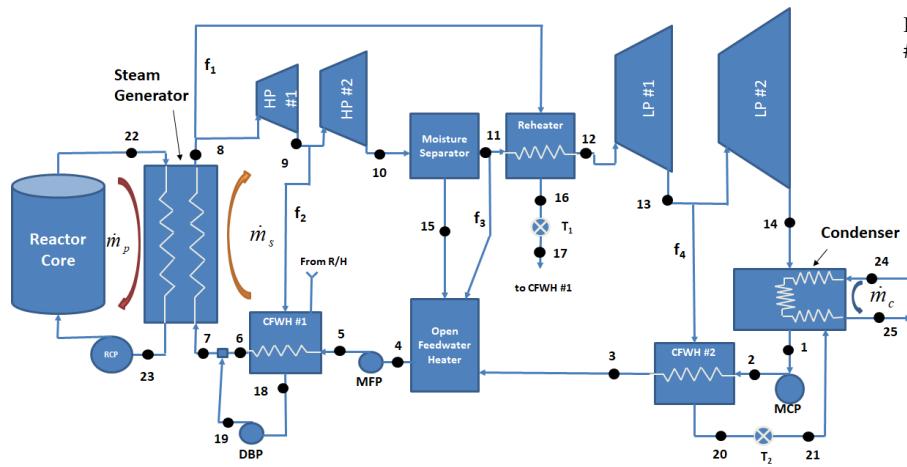


Figure 23: Power Cycle for Assignment #3.

State Point	Notes	State Point	Notes
1	$P = 1.5 \text{ psia}, x = 0.0$	2	$P = 164 \text{ psia}$
3	$T = T_{\text{sat}} @ 62 \text{ psia}, P = 164 \text{ psia}$	4	$P = 164 \text{ psia}$
5	$P = 838 \text{ psia}, \eta_{\text{MFP}} = 0.88$	6	$P = 838 \text{ psia}, T = T_{\text{sat}} @ 410 \text{ psia}$
7	$P = 838 \text{ psia}$	8	$P = 838 \text{ psia}, x = 1.0$
9	$P = 410 \text{ psia}$	10	$P = 164 \text{ psia}$
11	$P = 164 \text{ psia}$	12	$P = 164 \text{ psia}, T = 490^\circ F$
13	$P = 62 \text{ psia}$	14	$P = 1.5 \text{ psia}$
15	$P = 164 \text{ psia}$	16	$P = 838 \text{ psia}, x = 0.0$
17	$P = 410 \text{ psia}, \text{ isenthalpic expansion through } T_1$	18	$P = 410 \text{ psia}, x = 0.0$
19	$P = 838 \text{ psia}$	20	$P = 62 \text{ psia}, x = 0.0$
21	$P = 1.5 \text{ psia}, \text{ isenthalpic expansion through } T_2$	22	$P = 2240 \text{ psia}, T = 610^\circ F$
23	$P = 2240 \text{ psia}, T = 535^\circ F$	24	$P = 14.7 \text{ psia}, T = 91^\circ F$
25	$P = 14.7 \text{ psia}, T = 115^\circ F$	Dead State	$T_o = 551^\circ R, P_o = 14.7 \text{ psia}$
\dot{m}_p	$113.5 \times 10^6 \text{ lb}_m/\text{hr}$		

First Law Analysis

1. Calculate pressure, temperature, enthalpy, entropy, quality, and specific flow exergy for each state point. [use USCS units throughout]

2. Form energy-balance equations for CFWH#1 and CFWH#2. Use *fmincon* to solve for the flow fraction components of f_1, f_2, f_3, f_4 , and the enthalpy at state point 7. [flow fractions are unitless, h_7 is in BTU/lb_m]

3. Find the specific work for all pumps and turbines.[BTU/lb_m]

4. Find the specific heat added in the SG and specific heat rejected in the condenser. [BTU/lb_m]

5. Calculate the mass flow rate of steam from the steam generator (\dot{m}_s) and mass flow rate of condenser cooling water (\dot{m}_c).

6. Verify that $w_{\text{net}} = q_{\text{net}}$.

7. What is the thermal efficiency for this cycle?

Second Law Analysis

8. Calculate the total exergy transfer by work for the cycle. [BTU/hr]

9. Calculate the rate of exergy destruction in each component of the cycle. [BTU/hr]

10. Verify that the second law energy balance is satisfied:

$$\text{Exergy in} - \text{Exergy out} = \text{Exergy Destroyed} + \text{Exergy Transfer Through Work}$$

11. **Question for discussion:** What components of the NSSS are most responsible for exergy destruction? What design modifications would you consider to improve thermal efficiency?

Lecture 8 - Boiling Water Reactors and Supercritical Water Nuclear Reactors

Objectives

The objectives of this lecture are:

- Qualitative comparison between BWR and SCWR technologies
- Quantitative example of SCWR energy conversion cycle
- Practice analysis of complex Rankine cycles

BWRs and Direct Cycles

Early BWRs were built with the hopes of simplifying the system by having a direct steam cycle. A schematic is shown in Figure 24.

APART FROM THE NEED to contain radioactivity in the working fluid, this cycle is not, in principle, much different than the Rankine energy conversion cycle associated with a PWR. Some of the advantages of a “more simple” direct cycle are offset by the lower power density achievable in the core of a BWR. Most notably, the overall system for a given power output, is roughly as large as a PWR resulting in comparable construction costs; thus BWRs share one of the most significant problems with PWRs: high capital costs.

Supercritical Water Nuclear Reactor

ONE GEN IV CONCEPT that has been developed with the aim of addressing the power-density issue with direct-cycle light water reactors is the Supercritical Water Nuclear reactor.²⁶ For this cycle, the water is maintained above the critical pressure throughout the heat-addition phase. Good heat transfer can still be achieved without

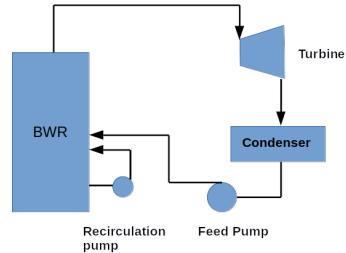


Figure 24: Simplified BWR schematic.

Note: Typical core average power density for a BWR is 50 kW/l; roughly one half that achievable with a PWR.

²⁶ Georgi V Tsiklauri, Robert J Talbert, Bruce E Schmitt, Gennady A Filippov, Roald G Bogojavlenksy, and Evgeny I Grishanin. Supercritical steam cycle for nuclear power plant. *Nuclear Engineering and Design*, 235(PNNL-SA-39686), 2005

phase change and the prevention of bulk boiling prevents hydraulic instabilities that can be associated with BWR cores.

ADDITIONAL NOTES ARE provided in bullet-format:

- Elimination of phase change in the core eliminates the need for complex moisture separation equipment for fluid exiting the reactor core.
- Lack of phase change in the core also shifts the power distribution higher in the core relative to a typical BWR; the combination of this effect along with elimination of the moisture separation equipment favors locating the control rods at the top of the core.
- Overall result is a smaller reactor pressure vessel and reduced containment size.

A schematic comparison of the size of a SCWR relative to competing PWR and BWR technologies is presented in Figure 25.

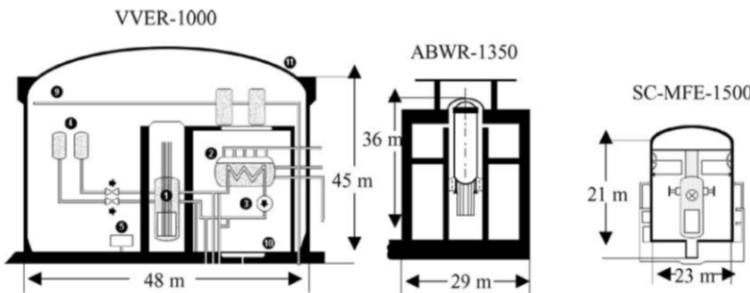


Figure 25: Comparison of the size of containments for SC-MFE-1500, ABWR-1350 and VVER-1000.

EVEN THOUGH MOISTURE SEPARATION equipment is not needed in the pressure vessel, there is a need for moisture separation equipment in the energy conversion cycle. A temperature-entropy plot is drawn to scale in Figure 26. Even with very high core outlet temperature, when the supercritical fluid is expanded to low pressure moisture content of the working fluid becomes unacceptably high. Typical coal-fired steam plants operating at supercritical pressures deal with this issue by using one or more stages of reheat. For nuclear powered systems, the steam piping arrangement necessary to use nuclear power for reheat can be impractical. For this cycle we will use additional moisture separation stages.

A schematic of a potential SCWR energy conversion cycle is shown in Figure 27. Analysis of this cycle involves some complexity insofar as the flow fractions for each branch of the cycle needs to be determined but the details of such calculations are straight-forward. Sig-

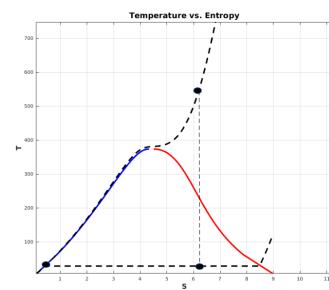
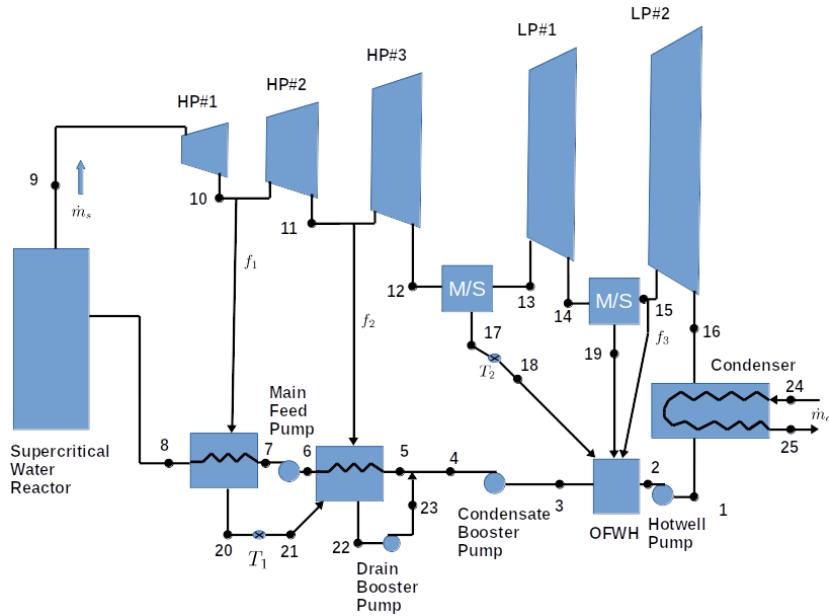


Figure 26: Temperature-entropy plot for SCWR drawn to scale.

nificant plant parameters are listed in Table 5. Note the high thermal efficiency and high net specific work; with the moisture separators in use, the minimum turbine exhaust quality is more than 85 percent.



Parameter	Value
T_9, P_9	550°C, 24 MPa
HP Turbine Outlet Pressures	8.9, 6, and 1 MPa
LP Turbine Outlet Pressures	200 kPa, 4 kPa
η_T, η_P	0.9 and 0.88
w_{net}	948 kJ/kg
Thermal Efficiency	47.3 percent

Table 5: SCWR Parameters.

Figure 27: SCWR power cycle with moisture separators. MATLAB code for a first-law analysis of this cycle is provided in the appendices.

Example calculations:

1. Write an expression for the specific work of Low Pressure Turbine #2:

Answer: $(h_{15} - h_{16}) [(1 - f_1)(1 - f_2)x_{12}x_{14}(1 - f_3)]$

2. Write an expression for the specific work of the Drain Booster Pump:

Answer: $(h_{22} - h_{23}) [f_1 + (1 - f_1)f_2]$

3. Write an expression for the specific work of the Condensate Booster Pump:

Answer: $(h_3 - h_4)[1 - (f_1 + (1 - f_1)f_2)]$

4. Write an expression for the energy balance at the mixing point between the Condensate Booster Pump and CFWH #2:

Answer: $h_5 = [f_1 + (1 - f_1)f_2]h_{23} + [1 - (f_1 + (1 - f_1)f_2)]h_4$

Lecture 9 - Pinch-Point Temperature Difference

Objectives

The objectives of this lecture are:

- Describe pinch-point temperature difference (ΔT_p) and show how to calculate it.
- Show relationship between ΔT_p and exergy destruction rate in the Steam Generator (SG).
- Compute ΔT_p for a few modern commercial PWRs.
- Provide thumb rules/guidance for concept design use.

Description of ΔT_p

THE PINCH-POINT TEMPERATURE DIFFERENCE is the minimum difference between the primary coolant and the secondary coolant in a PWR. This is shown schematically in the temperature-entropy diagram in Figure 28. An alternative representation of the process is shown in Figure 29.

LET US TAKE A MOMENT to discuss these representations:

- These representations are not intended to be taken literally. The SG for most PWRs is, of course, not a once-through counter-flow heat exchanger as Figure 29 indicates.²⁷
- Consequently, one shouldn't expect to be able to point to the spot on a SG where ΔT_p is realized.
- Nonetheless, ΔT_p can still be used as a useful engineering parameter because it relates to the relative temperature of the primary coolant and the steam generator.

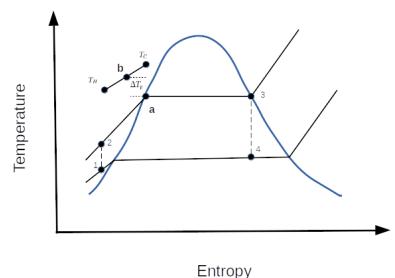


Figure 28: Schematic illustration of ΔT_p .

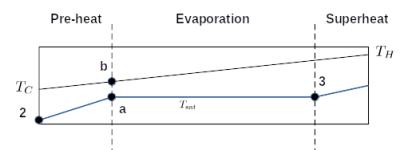


Figure 29: Illustration of ΔT_p .

²⁷ Although in fairness when the concept of ΔT_p was formulated, such SGs were in use and it probably seemed less crazy.

IN HEAT TRANSFER CLASS you will learn one method for analyzing heat exchangers, like the SG, using the log-mean temperature difference. Equation 7 says that:

- The rate of heat transfer (\dot{Q}_{SG}) is equal to the product of the overall-heat transfer coefficient (U), the heat transfer surface area (A) and the log-mean temperature difference (ΔT_{LM}); and
- That heat transfer rate is also equal to the rate of heat transfer into the secondary system water and heat transfer from the primary coolant.

$$\dot{Q}_{SG} = UA\Delta T_{LM} = \dot{m}_s(h_3 - h_2) = \dot{m}_p(h_H - h_C) \quad (7)$$

THIS IS THE FIRST TIME so far in this course where we have attempted to relate the performance of a heat exchanger to any real-world parameters of the heat exchanger. In past lectures, heat exchangers like feedwater heaters, SGs, and condensers were represented only as rectangles on a schematic with in-flows and out-flows and some expectation as to what the temperatures of the respective fluids would be. No consideration was given as to what a heat exchanger would look like; how big it would be, what materials it would be made of, and if the expected performance was in any way realistic.

TO THE EXTENT that this shortcoming is addressed at all, we do it with the “Second Law” exergy analysis. If the exergy destruction rate is calculated correctly and if the result is negative that is a clear sign that you’ve laid out impossible expectations for heat exchanger performance. With this lecture we will establish an additional safeguard against wishful thinking. We will calculate ΔT_p and compare it to “typical” values achieved for existing designs. If it is comparable to what currently exists: we can have faith that the specified performance is not crazy. If the ΔT_p is significantly lower than typical: it is a sign that we will have to do additional analysis to prove that our hopes can realistically be achieved.

Calculating ΔT_p

For the evaluation of ΔT_p we will assume that the following parameters are known: steam or primary coolant mass flow rate; hot- and cold-leg temperature of the primary; temperature of feedwater entering the SG; and the SG pressure or temperature.²⁸ The basic energy balance for the SG, using the cycle depicted in Figure 28 is given as:

$$\dot{m}_p(h_H - h_C) = \dot{m}_s(h_3 - h_2)$$

Note: The overall heat transfer coefficient combines the effects conductive heat transfer through heat exchanger surfaces as well as convective heat transfer to/from the heated/cooled fluids.

Note: Log-mean temperature difference is computed based on the difference in temperatures at the heat exchanger inlets and outlets. Specifically, $\Delta T_{LM} = \ln \frac{\Delta T_{side-a}}{\Delta T_{side-b}}$. Referencing Figure 29, the log mean temperature difference is calculated as: $\Delta T_{LM} = \ln \frac{T_H - T_3}{T_C - T_2}$

²⁸ For purposes of this discussion, we will assume that the steam leaving the SG is saturated vapor.

If we account for the enthalpy of the saturated liquid in the SG (h_a) and the corresponding enthalpy of the primary coolant at the pinch-point (h_b) we can break the energy balance down further:

$$\begin{aligned}\dot{m}_p(h_b - h_C) &= \dot{m}_s(h_a - h_2) \\ \dot{m}_p(h_H - h_b) &= \dot{m}_s(h_3 - h_a)\end{aligned}$$

Using the first of those equations, we derive an expression for h_b :

$$h_b = \frac{\dot{m}_s}{\dot{m}_p}(h_a - h_2) + h_C$$

If we know h_b and the primary system pressure (P_{pri}) we can find T_b . Using the steam generator pressure and the fact that h_a corresponds to the enthalpy of the saturated liquid, we can find T_a as the saturation temperature for the given SG pressure. Thus:

$$\Delta T_p = T_b - T_a$$

PINCH-POINT TEMPERATURE DIFFERENCE is directly related to exergy destruction rate in the SG. Higher ΔT_p corresponds to higher exergy destruction, more irreversibility, and lower efficiency; lower ΔT_p corresponds to lower exergy destruction, less irreversibility, and higher plant efficiency. On the other hand, SGs with lower ΔT_p , all other things being equal, must be larger or made with materials having improved thermal conductivity or have enhanced convective heat transfer performance (or both!).

THE FOLLOWING GUIDANCE is offered to you for consideration when developing a concept design for a PWR:

- A reasonable range of ΔT_p based on historical experience is 20 - 30 degrees Fahrenheit.²⁹
- Related to the ΔT_p calculation: the temperature of feedwater for the SG should be approximately 100 degrees F less than saturation temperature in the SG.³⁰
- An analysis of more contemporary PWR designs yield values of ΔT_p in the range of 18-22 degrees Fahrenheit.
- In short: around 20 degrees F is a good starting point.
- If you must make your SG more compact—e.g. developing a PWR-based micro reactor—it is likely that you will need to accept higher values for ΔT_p .

²⁹ James H Rust. *Nuclear power plant engineering*. Haralson Publishing Company, Buchanan, Ga, 1979

³⁰ Mohamed Mohamed El-Wakil. *Nuclear energy conversion*. American Nuclear Society, La Grange Park, IL, 1982

- If you can afford a very large SG then you can entertain a ΔT_p lower than 20 degrees F.
- A reasonable choice of ΔT_p should give you confidence that a detailed heat exchanger design analysis will show that the heat transfer performance can be achieved with a SG of reasonable size.

Lecture 10 - Brayton Cycle Review

Objectives

The objectives of this lecture are:

- Review the processes for a simple Brayton cycle
- Carry out example analysis using:
 - a constant specific heat approach; and
 - using EasyProp

Brayton Cycle Review

The Brayton cycle is a fundamental heat-engine thermodynamic cycle that is widely used to model gas turbine engines used for helicopters, tanks, and of course power plants. A schematic of a simple direct variant of the Brayton cycle for nuclear energy conversion is shown in Figure 30. Similar to a Rankine cycle the simplest version has the following processes:

1. isentropic compression ($1 \rightarrow 2$)
2. isobaric heat addition ($2 \rightarrow 3$)
3. isentropic expansion ($3 \rightarrow 4$)
4. isobaric heat rejection ($4 \rightarrow 1$)

These are the same processes as a Rankine cycle; the difference is that the working fluid does not undergo a phase change during either heat addition or heat rejection; the working fluid is modeled as a compressible gas. The temperature-entropy diagram for a non-ideal variation of the cycle is shown in Figure 31. For this, the compressor and turbine both are non-isentropic.

A more accurate representation of the cycle would also account for inevitable hydraulic losses as the working fluid flows through

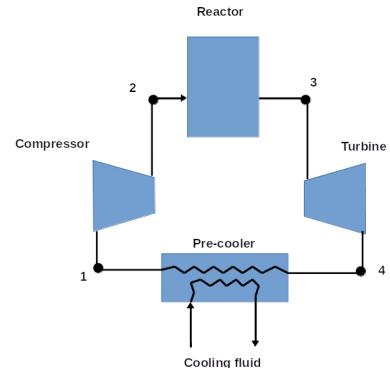


Figure 30: A simple direct Brayton cycle for nuclear energy conversion.

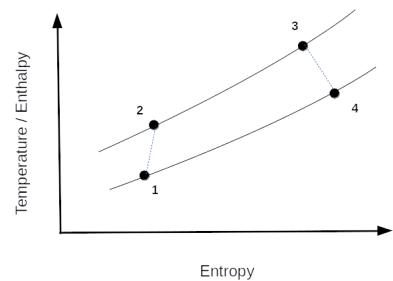


Figure 31: Temperature-Entropy diagram for a simple non-ideal Brayton cycle.

ducting and heat exchangers associated with the cycle. For a Rankine cycle using water as the working fluid, these loss mechanisms are comparatively small but for Brayton cycle these losses are important and have a large impact on overall system thermodynamic performance.

Analysis of Ideal Brayton Cycle - Constant Specific Heat

If we assume the working fluid is an ideal gas in which the specific heat at constant pressure (C_p) and constant volume (C_v) are both assumed to be constant, an analytic solution method is feasible. We will model changes in enthalpy using the relationship given in Equation 8.

$$\Delta h = C_p \Delta T \quad (8)$$

For isentropic processes a relationship exists between initial and final temperature of the working fluid, and the initial and final pressure of the working fluid. This relationship is given in Equation 9.

$$\frac{T_f}{T_i} = \left(\frac{P_f}{P_i} \right)^{\frac{\gamma-1}{\gamma}} = r_p^{\frac{P_f}{P_i}} \quad (9)$$

where γ is the ratio of specific heats and r_p is referred to as the "pressure ratio".

Note: $\gamma = C_p/C_v$, and $r_p = P_f/P_i$

Example: Consider the a simple, ideal, closed, direct nuclear Brayton cycle. Assuming the coolant is helium with $c_p = 5.23 \text{ kJ/kg}\cdot\text{K}$ and $\gamma = 1.658$. The maximum cycle temperature is 972 K and minimum temperature is 278 K. The compressor pressure ratio is 4.0. Find:

1. net specific work
2. thermal efficiency; and
3. back work ratio

Solution: Work of the isentropic compressor and turbine can be found by combining Equation 8 and Equation 9.

For the compressor we get:

$$w_c = h_1 - h_2 = C_p(T_1 - T_2) = C_p(T_1 - T_1 r_p^{\gamma-1/\gamma}) = C_p T_1 (1 - r_p^{\gamma-1/\gamma})$$

Applying the given numbers gives us:

$$w_c = (5.23 \text{ kJ/kg}\cdot\text{K})(273 \text{ K}) (1 - 4^{0.658/1.658}) = 1066.5 \text{ kJ/kg}$$

For the turbine we get:

$$W_t = h_3 - h_4 = C_p(T_3 - T_4) = C_p(T_3 - T_3(\frac{1}{r_p})^{\frac{\gamma-1}{\gamma}}) = C_p T_3 [1 - (\frac{1}{r_p})^{\gamma-1/\gamma}]$$

which comes to:

$$w_t = (5.23 \text{ kJ/kg-K})(972 \text{ K}) (1 - 0.25^{\frac{0.658}{1.658}}) = 2151.1 \text{ kJ/kg}$$

net specific work is thus: $w_{\text{net}} = -1066.5 + 2151.1 = 1084.3 \text{ kJ/kg}$.

We should calculate net specific heat added in order to compare with net work; for this we need values for the temperature at state points 2 and 4. Using Equation 9 we get $T_2 = T_1 r_p^{\gamma-1/\gamma} = 278(4)^{0.658/1.658} = 481.9 \text{ K}$ and $T_4 = T_3(1/r_p)^{\gamma-1/\gamma} = 972(0.25)^{0.658/1.658} = 560.7 \text{ K}$. Using these numbers with Equation 8 we get: $q_r = h_1 - h_4 = C_p(T_1 - T_4) = 5.23 \text{ kJ/kg-K}(278 - 560.7) = -1478.5 \text{ kJ/kg}$ and $q_s = h_3 - h_2 = C_p(T_3 - T_2) = 5.23 \text{ kJ/kg-K}(972 - 481.9) = 2563.2 \text{ kJ/kg}$. Net specific heat added is, therefore: $q_{\text{net}} = q_s + q_r = 2563.1 - 1478.5 = 1084.7 \text{ kJ/kg}$ which is very close to our calculated value for w_{net} .

Thermal efficiency is given as $\eta_{th} = w_{\text{net}}/q_s = 1084.7/2563.2 = 0.423$ or 42.3 percent.

The back work ratio is, by definition, the ratio of the magnitude of work supplied to the working fluid and the work extracted from the working fluid: $BWR = w_{\text{in}}/w_{\text{out}} = w_c/w_t = 1066.5/2151.1 = 0.496$.

Discussion of Example

A couple of points are worth noting on this example:

1. The analysis reveals a relatively high thermal efficiency. This is owing in part to the ideal work processes and the neglect of hydraulic losses in heat exchanger components.
2. This result also reflects the good thermodynamic properties of helium as the working fluid for a Brayton cycle.

In future problems we will examine the performance of alternative working fluids as well as the impact of non-ideal components and hydraulic pressure losses on overall thermodynamic performance.

Analysis with EasyProp

When air, or any other polyatomic gas, is the working fluid it is common to take into account the variability of the specific heats as a function of temperature. The thermodynamic properties available through EasyProp take such variability into account.

THIS EXAMPLE CONSIDERS an indirect, open Brayton cycle that uses air for the working fluid. The solution will be found using EasyProp with, in this case, the entirety of the analysis code included. A state point table summarizing the thermodynamic properties after each process in the cycle and numeric results are provided in Figure 33.

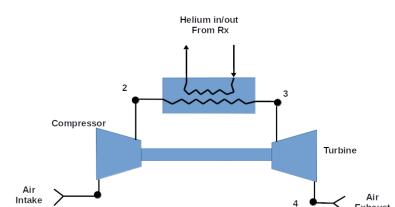


Figure 32: Open, indirect Brayton cycle.

Example: A small modular reactor concept is proposed in which a helium-cooled, graphite moderated reactor transfers energy in a heat exchanger to an open Brayton cycle as depicted below. A schematic is shown in Figure 32. Air at 14.7 psia and 60°F enters the compressor, passes through the heat exchanger at constant pressure and expands through a turbine that exhausts to the atmosphere. The compressor has a pressure ratio of 14 and the turbine inlet temperature is 1600°F. Compressor isentropic efficiency is 87% and the turbine isentropic efficiency is 90%.

Find:

1. net specific work
2. cycle thermal efficiency; and
3. back work ratio

Solution: Full solution code is provided in the listing below.

```
%>>> %% Open, Indirect, Air Brayton Cycle Example
2 clear
3 clc
4 close 'all'

6 %% Put EasyProp on Python path
% put location of EasyProp.py module on the python search path
8 if count(py.sys.path,' ') == 0 %-- see if desired directory
    is on path
    insert(py.sys.path,int32(0),' '); %-- if not; add it.
10 end

12 %% Initialize Fluid Property object
fluid = 'Air';
14 units = 'USCS';
air = py.EasyProp.simpleFluid(fluid,units);

16 %% Initialize state point arrays for Brayton Cycle
18 numSP = 4;
h = nan(numSP,1);
20 hs = nan(numSP,1);
s = nan(numSP,1);
22 ss = nan(numSP,1);
T = nan(numSP,1);
P = nan(numSP,1);

26 %% Problem Parameters
Pmin = 14.7; % psia, atmospheric pressure
28 Tin = 60; % F, inlet air temperature
r_p = 14;
30 r_e = 1/r_p;
T_turb_inlet = 1600; % F
32 eta_c = 0.87;
```

	P	T	h	s
1	14.7	60	178.39	0.91908
2	205.8	716.01	339.5	0.93755
3	205.8	1600	576.05	1.087
4	14.7	684.45	331.48	1.1119

Net specific work: 83.463 BTU/lbm
 Net specific heat added: 83.463 BTU/lbm
 Thermal Efficiency: 35.28 percent
 Back Work Ratio: 65.87 percent

Figure 33: Numeric results

```

34 eta_t = 0.9;

36 %% Compute State Point Property Data
% state point 1
38 P(1) = Pmin;
T(1) = Tin;
40 h(1) = air.h_pT(P(1),T(1));
s(1) = air.s_pT(P(1),T(1));

42 % state point 2
44 P(2) = P(1)*r_p;
ss(2) = s(1);
46 hs(2) = air.h_ps(P(2),ss(2));
h(2) = h(1) - (h(1)-hs(2))./eta_c;
48 T(2) = air.T_ph(P(2),h(2));
s(2) = air.s_ph(P(2),h(2));

50 % state point 3
52 P(3) = P(2); %isobaric heat addition
T(3) = T_turb_inlet; % F, given
54 h(3) = air.h_pT(P(3),T(3));
s(3) = air.s_pT(P(3),T(3));

56 % state point 4
58 P(4) = P(3)*r_e;
ss(4) = s(3);
60 hs(4) = air.h_ps(P(4),ss(4));
h(4) = h(3)-(h(3)-hs(4))*eta_t;
62 T(4) = air.T_ph(P(4),h(4));
s(4) = air.s_ph(P(4),h(4));

64 % display state point data neatly
66 SP = {'1','2','3','4'};
SP_table = table(P,T,h,s,'RowName',SP);
68 disp(SP_table);

70 %% First Law Analysis
w_c = h(1) - h(2);
72 w_t = h(3) - h(4);
w_net = w_c + w_t;

74 q_s = h(3) - h(2);
76 q_r = h(1) - h(4);
q_net = q_s + q_r;

78 assert(abs(w_net - q_net)<1,'Conservation of energy condition
    not met!');

80 fprintf('Net specific work: %g BTU/lbm \n',w_net);
82 fprintf('Net specific heat added: %g BTU/lbm \n',q_net);

84 eta_th = w_net/q_s;
86 fprintf('Thermal Efficiency: %5.2f percent \n',eta_th*100);
88 BWR = abs(w_c/w_t);
90 fprintf('Back Work Ratio: %5.2f percent \n',BWR*100);

```


Lecture 11 - More Complex Brayton Cycles

Objectives

The objectives of this lecture are:

1. Discuss impact of pressure ratio on cycle thermal efficiency
2. Regeneration and Regenerator effectiveness
3. intercooling and reheat

Pressure Ratio and Thermal Efficiency

For an ideal cycle (isentropic compressor/turbine and no hydraulic losses) thermal efficiency is related to compressor pressure ratio as follows:

$$\begin{aligned}
 \eta_{\text{TH}} &= \frac{w_{\text{net}}}{q_s} \\
 &= \frac{C_p(T_3 - T_4) - C_p(T_2 - T_1)}{C_p(T_3 - T_2)} \\
 &= \frac{(T_3 - T_2) - (T_4 - T_1)}{(T_3 - T_2)} \\
 &= 1 - \frac{T_4 - T_1}{T_3 - T_2} \\
 &= 1 - \frac{T_1(T_4/T_1 - 1)}{T_2(T_3/T_2 - 1)}
 \end{aligned}$$

If there are no pressure losses, then the pressure ratios for the compressor and turbine process are the same and $T_4/T_1 = T_3/T_2$ so the last equality reduces to:

$$\eta_{\text{TH}} = 1 - \frac{T_1}{T_2}$$

Using our ideal gas relation for isentropic processes, $T_2 = T_1(r_p)^{(\gamma-1)/\gamma}$ we obtain our final result in Equation 10.

$$\eta_{\text{TH}} = 1 - \frac{1}{r_p^{(\gamma-1)/\gamma}} \quad (10)$$

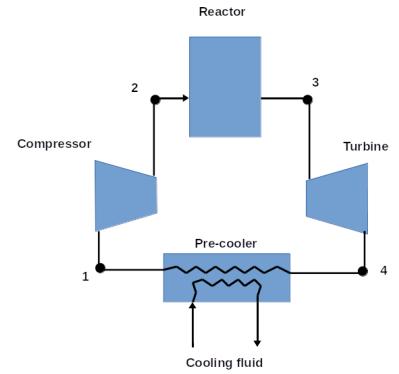


Figure 34: Simple Ideal Brayton Cycle.

For an ideal gas thermal efficiency increases for increasing pressure ratio. In principle this means we should be able to make thermal efficiency arbitrarily close to 100% by increasing pressure ratio. In practice there are other considerations that need to be taken into account.

Pressure Ratio and Net Specific Work

FOR MANY BRAYTON CYCLES the pressure ratio affects achievable net specific work. This is because:

1. Real systems have limitations to the maximum temperature of the working fluid. Some factors that lead to this limitation include:
 - (a) material limits of the system piping and structural components
 - (b) material limits of the heat source—such as nuclear reactor core outlet temperature limits—which, in order to add heat to the working fluid, are an upper bound to the maximum Brayton cycle system temperature.

As pressure ratio increases, so does the temperature at the compressor discharge (T_2), so these maximum temperature limits impose a limit to the compressor pressure ratio. Advances in material sciences offer the possibility of increasing these temperature limits over time but, for any concept, the designer has limited ability to influence this bound.

2. Real systems have limits to the minimum temperature of the working fluid. This is dictated by the temperature of the thermal reservoir to which the cycle must reject heat in the pre-cooler. This environmental temperature is generally outside of the control of the designer.

THESE LIMITS INFLUENCE the maximum net specific work that can be extracted from the working fluid for an ideal Brayton cycle. Figure 35 gives a qualitative feel for the influence of pressure ratio on net specific work for systems operating within material limits. Some key points:

- At low r_p , most of the added heat is simply rejected to the environment.
- At high r_p , the amount of heat that can be added to the working fluid is constrained by the temperature limit; thermal efficiency is high but, since w_{net} is limited, the mass flow rate of the coolant must be higher to achieve a desired power output.

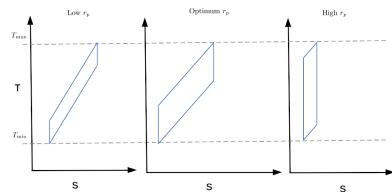


Figure 35: Effect of r_p on w_{net} for simple ideal Brayton cycles.

- The optimum r_p is a function of the working fluid. Figure 36 illustrates this for four typical Brayton cycle working fluids.
- Reduced w_{net} and increased \dot{m} of working fluid may seem like a small price to pay for increased thermal efficiency but:
 - Real systems will suffer hydraulic pressure losses that are proportional to \dot{m}
 - The thermal efficiency of a Brayton cycle is heavily impacted by those hydraulic pressure losses as we will see from our exercises.
 - Sadly a detailed treatment which connects all of the dots between pressure ratio, working fluid mass flow rate, and analysis of thermal efficiency including hydraulic pressure losses still needs to be brought into the scope of this class.

Brayton Cycle with Regeneration

ONE OBVIOUS WEAKNESS of the simple ideal Brayton cycle is the large amount of useful energy that gets rejected to the environment. The working fluid at the turbine exhaust (state point 4) is at low pressure but its temperature remains high. We can exploit the exergy contained in this high temperature fluid in a regenerator as illustrated in Figure 37. Use of the regenerator reduces the amount of energy that need be added in the heat addition process. This is illustrated in Figure 38.

WE CHARACTERIZE THE PERFORMANCE of the regenerator with a parameter called *regenerator effectiveness*. The regenerator effectiveness for this cycle can be expressed as in Equation 11.

$$\eta_{reg} = \frac{\Delta h_{actual}}{\Delta h_{max}} = \frac{h_3 - h_2}{h_5 - h_2} \quad (11)$$

Having a higher regenerator effectiveness is obviously desirable but is subject to limits. Qualitatively speaking one increases regenerator effectiveness by:

- Increasing heat transfer surface area in the regenerator—e.g. make the regenerator bigger.
- Enhancing heat transfer effectiveness in the regenerator. This might be done by:
 - use materials with better conductive heat transfer properties

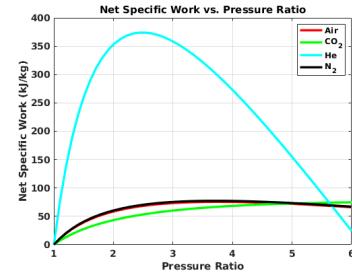


Figure 36: Net specific work vs pressure ratio for common fluids.

Note: looking at Figure 36, which fluid is different from the others?

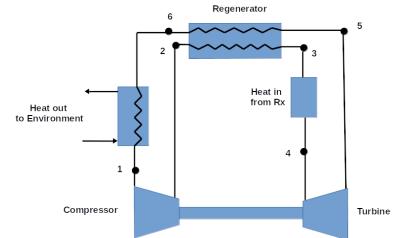


Figure 37: Schematic of a regenerative Brayton cycle.

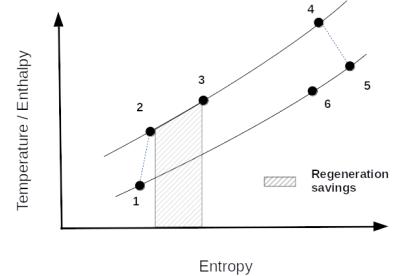


Figure 38: Reduced q_s due to regeneration savings

regenerator effectiveness: the ratio of the actual enthalpy increase of the fluid on the low temperature side to the maximum theoretical enthalpy increase.

- use convective heat transfer enhancements such as spiral grooves.
- use working fluids with better heat transfer properties.

With the exception of convective heat transfer enhancements which we will discuss in a later lecture, these considerations are beyond the scope of this course. Like isentropic efficiency data for pumps, compressors, and turbines, we will take regenerator effectiveness to be a given parameter. **A typical value will be around 60%.**

Inter-cooling

IN ADDITION TO using regeneration to reduce heat rejected, it is desirable to reduce the work required in the compression processes. As characterized by the back work ratio, as much as half (or more!) of the turbine work for a Brayton cycle may be needed to power the compressor. We can reduce the work required for compression by *cooling* the working fluid part-way through the compression. A Brayton cycle schematic with such an *inter-cooler* is shown in Figure 39. Heat extracted in the inter-cooler is normally rejected to the environment. In this sense the energy is lost but the savings on compressor work more than offsets this sacrifice. While it's traditional to show the temperature-entropy diagram—this is shown in Figure 40—the work savings from inter-cooling is more easily seen from the $P - v$ diagram in Figure 41.

A NATURAL QUESTION to ask is: How much of the compression should be done with the first compressor and how much with the second to get maximum work reduction? It is a straight-forward but tedious task to show that, for isentropically ideal compressors working on ideal gasses with constant specific heats, you *minimize overall compressor work if the pressure ratios for the compressors are the same*.

Brayton Cycle with Reheat

ANOTHER WAY TO INCREASE the net specific work and thermal efficiency of a Brayton cycle is to include one or more stages of reheat. A T-S diagram is shown in Figure 42 and a schematic representation of such a cycle is illustrated in Figure 43.

You should be asking the same question as with inter-cooling: how do I set the pressure ratios for the turbines such that I get maximum

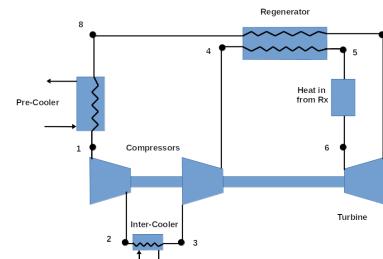


Figure 39: Brayton cycle with regeneration and inter-cooling.

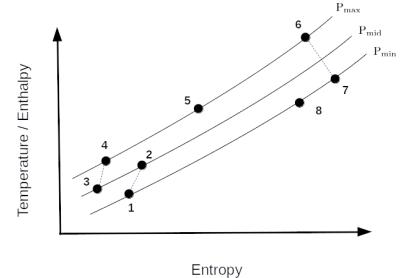


Figure 40: TS plot of a Brayton cycle with regeneration and inter-cooling.

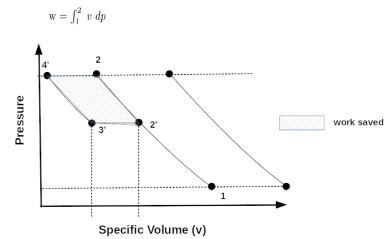


Figure 41: Work savings from inter-cooling.

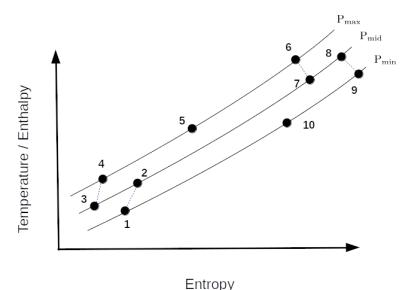


Figure 42: T-S diagram for Brayton cycle with reheat.

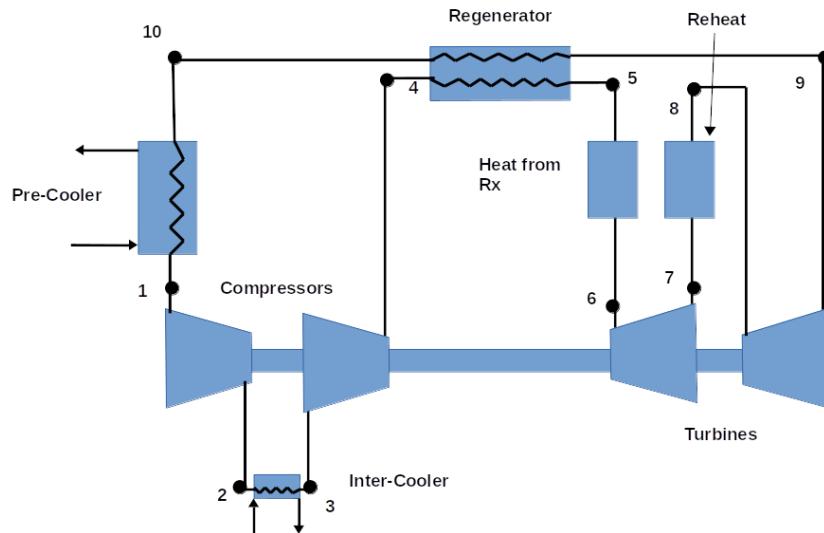


Figure 43: Brayton cycle with reheat.

benefit from reheating? Sadly I do not have a short answer to this question. To help you build some intuition on this, we will investigate this issue in our workshop assignments.

Assignment #4: Brayton Cycle Analysis

USE EASYPROP and MATLAB to analyze the following problems:

1. Air enters a simple open Brayton cycle at 14.7 psia and 85°F. Compressor pressure ratio is 5 and the isentropic efficiency of the compressor is 85%. Helium coolant from a Very High Temperature Reactor (VHTR) supplies heat to the air in a heat exchanger so that the inlet temperature to the turbine in the Brayton cycle is 1200°F. The turbine isentropic efficiency is 90%. Air exiting the turbine is discharged to the atmosphere at 14.7 psia. Calculate:
 - (a) Net specific work. [BTU/lb_m]
 - (b) Thermal efficiency.
 - (c) Mass flow rate needed for 5×10^6 BTU/hr power output. [lb_m/s]
2. Consider the cycle described in Problem #1. While maintaining constant maximum and minimum temperature of 85°F and 1200°F respectively, constant isentropic efficiency of the compressor and turbine of 85% and 90% respectively, vary the compressor pressure ratio between 2 and 15.
 - (a) Make plots of thermal efficiency and net specific work over that range of compressor pressure ratio.
 - (b) Determine the pressure ratio for maximum thermal efficiency.
 - (c) Determine the pressure ratio for maximum net work.Briefly explain your results in a short paragraph of approximately 250 - 500 words.
3. Consider again the cycle described in Problem #1. Instead of discharging the turbine exhaust directly to atmosphere, direct the turbine exhaust to a regenerator. Turbine exhaust gases preheat the air discharged from the compressor prior to having heat added by the VHTR coolant. Assume the regenerator effectiveness is 85%.
 - (a) Create a simple schematic of your cycle; label all state point numbers.
 - (b) Calculate net specific work for the cycle. [BTU/lb_m]
 - (c) Calculate the thermal efficiency of the cycle.

4. Consider the cycle you have created for Problem #3. Study the performance as regenerator effectiveness is varied from 0% to 100%; make a plot of thermal efficiency versus regenerator effectiveness. Briefly comment on your results (250 - 300 words) and verify that your results make sense.
5. Consider again the cycle described in Problem #1. Incorporate two-stage compression with an inter-cooler between stages. Assume that the working fluid is cooled to 85°F (at constant pressure) before entering the second compressor. Consider three different cases for the pressure ratios of the two compressors:
- Both compressors have the same compressor pressure ratio ($r_{p,1} = r_{p,2} = \sqrt{5}$).
 - $r_{p,1} = 1.5, r_{p,2} = 5/1.5$.
 - $r_{p,1} = 5/1.5, r_{p,2} = 1.5$.
- Calculate compressor work and thermal efficiency for each case. Briefly comment on your results. (**Note:** you need not do a second-law analysis on this cycle so there is no need to try and compute the state of the fluid that serves as the heat sink in the inter-cooler.)
6. Consider a Brayton cycle with inter-cooling, regeneration, and reheat. Air enters the cycle at 14.7 psia, 85°F. Two stages of compressors are used, each with equal compressor pressure ratio, so that the overall pressure ratio is 5. A regenerator is used with a regenerator effectiveness of 80%. After heat is added to a maximum temperature of 1200°F, two turbines are used with re-heating between the two stages; the reheat process is isobaric and the air temperature out of the reheat is 1200°F. Assume there are no hydraulic pressure losses within system components. **Note:** As with the last problem, you need not evaluate state point properties of the helium coolant from the VHTR, nor need you analyze the state point properties for the heat sink of the inter-cooler.

(a) Make a simple sketch of your system, being sure to identify all state points.

(b) Vary the expansion ratio of the turbines as follows:

i. $r_{e,1} = r_{e,2} = \sqrt{5}$

ii. $r_{e,1} = 1.5, r_{e,2} = 5/1.5$

iii. $r_{e,1} = 5/1.5, r_{e,2} = 1.5$

Calculate net specific work and thermal efficiency for all three cases. Briefly comment on your results (250 - 500 words).

Lecture 12 - Supercritical CO₂ Brayton Cycles

Objectives

The objectives of this lecture are:

1. Illustrate the benefits of S-CO₂ Brayton cycles.
2. Show why regeneration is so important for S-CO₂ cycles.
3. Provide an alternative definition of regenerator effectiveness that one should use for supercritical gas power cycles.

CO₂ Brayton Cycles

WHEN WE INTRODUCED THE BRAYTON CYCLE, we used helium as the working fluid. Indeed helium is a highly favored choice for numerous small modular reactor concept designs for high temperature reactors and we found that high cycle thermal efficiency could be obtained even with a simple (albeit ideal) Brayton cycle. Helium, however, has some problems; some of these include:³¹

- Helium is expensive.
- Power conversion systems using helium can expect significant helium leakage through seals of rotating components and even welded boundaries.
- Helium power cycle efficiency is more sensitive to regenerator effectiveness and fractional hydraulic pressure drops in plant components.
- There is less experience World-wide in the use of Helium as a working fluid compared to air and CO₂.

³¹ Vaclav Dostal, Pavel Hejzlar, and Michael J Driscoll. The supercritical carbon dioxide power cycle: comparison to other advanced power cycles. *Nuclear technology*, 154(3):283–301, 2006

For these reasons, for the time being, we will shift the focus of our attention to Brayton cycles using CO₂ as the working fluid.

LET US CONSIDER a simple, ideal, closed Brayton cycle using CO₂ as the working fluid. $P_{\max} = 2000$ kPa and $P_{\min} = 770$ kPa. A schematic of the system and performance results are shown in Figure 44.

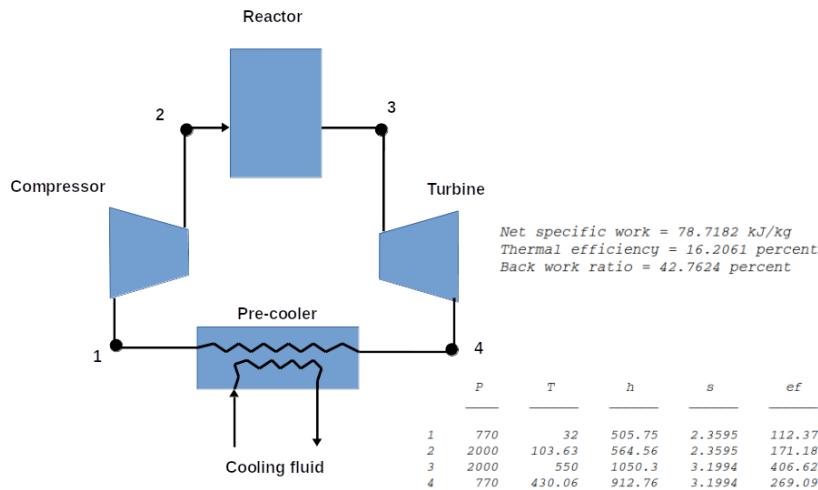


Figure 44: Simple ideal Brayton cycle with CO₂

FROM THE STATE POINT TABLE it is evident that the fluid is still at a high temperature at the turbine exhaust; a considerable fraction of the energy/exergy provided by the reactor is rejected to the atmosphere. We can also see that the back work ratio is quite high.

Simple Supercritical CO₂ Brayton Cycle

TAKING OUR LEAD from the title of this lecture, we will examine how system performance changes when we increase system pressure above the critical pressure for CO₂ which is approximately 7.4 MPa while maintaining the same maximum temperature and pressure ratio. The numerical results from the analysis are presented in Figure 45.

Note that the back work ratio has been drastically reduced. By pressurizing the CO₂ above the critical pressure, the specific volume is lower and the specific pump work has been reduced. Still, the thermal efficiency is very low; we're still throwing away too much energy in the pre-cooler. A regenerator is definitely needed.

Supercritical CO₂ Regenerative Brayton Cycle

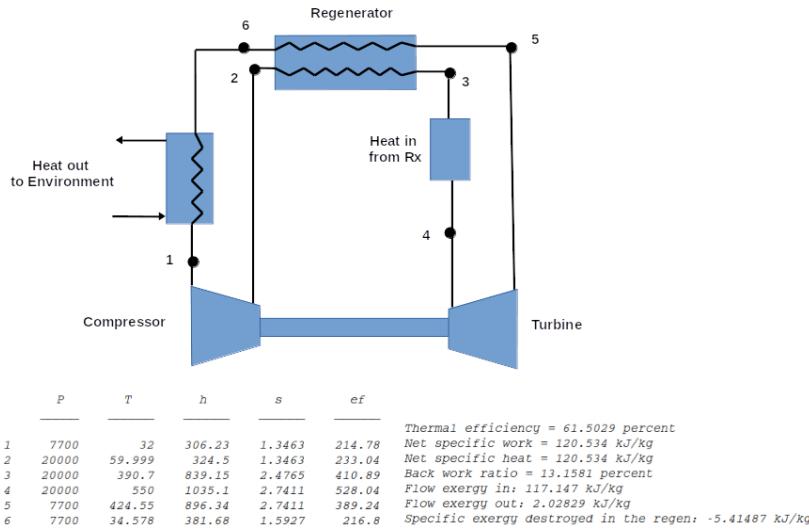
WE WANT TO reduce the amount of exergy rejected to the environment so we modify the cycle to add a regenerator. We will choose a

Net specific work = 120.534 kJ/kg
Thermal efficiency = 16.9614 percent
Back work ratio = 13.1581 percent

	P	T	<i>h</i>	s	ef
1	7700	32	306.23	1.3463	214.78
2	20000	59.999	324.5	1.3463	233.04
3	20000	550	1035.1	2.7411	528.04
4	7700	424.55	896.34	2.7411	389.24

Figure 45: Analysis of simple Supercritical CO₂ Brayton Cycle

very optimistic regenerator effectiveness of 90% while maintaining all other system parameters the same.³² A schematic of the cycle and analysis results are provided in Figure 46



³² We will justify this optimism by pointing to the expected heat transfer improvement with the CO₂ at high pressure and density. We will also take comfort in the fact that the textbook assumes similar regenerator effectiveness for supercritical CO₂ cycles.

Figure 46: Regenerating S-CO₂ cycle.

AS THE RESULTS INDICATE, adding regeneration did nothing to change net specific work, it's still at 120.5 kJ/kg, but recapturing that otherwise wasted energy, according to the analysis results in thermal efficiency increasing to over 61%.(!) Before popping open the Champagne we should review the results of the exergy analysis.³³ In round numbers, specific flow exergy passed in to the working fluid from the reactor is 117 kJ/kg, exergy rejected to the environment in the pre-cooler is a mere 2 kJ/kg; think of this as 115 kJ/kg of exergy being provided to the cycle for use. But somehow we extracted 120 kJ/kg of specific work?! The explanation, and the sore thumb that should stick out at you when you examine the results, is the negative specific exergy destruction rate calculated for the regenerator. What happened?

THE PROBLEM is contained in our definition of regenerator effectiveness; not that it is necessarily too high, but that it is inadequate for this problem involving a fluid above its critical pressure. Recalling from a previous lecture, we defined the regenerator effectiveness as:

$$\eta_{\text{reg}} = \frac{\Delta h_{\text{actual}}}{\Delta h_{\text{max}}} = \frac{h_3 - h_2}{h_5 - h_2} \quad (12)$$

but this formulation leaves out a key piece of physics: heat transfer, in a regenerator or elsewhere, is driven by temperature differences;

³³ Before looking at the numbers repeat to yourself: "Exergy in minus exergy out equals exergy transferred through work plus exergy destroyed."

the equation above only considers enthalpy values. Temperatures at the inlets and outlets of the regenerator for this analysis are shown in Figure 47. Note that the temperature at state point 6 (regenerator hot side outlet) is *lower* than state point 2 (cold side inlet). This is obviously a violation of the 2nd law of thermodynamics.

We can re-connect enthalpy with temperature by using the specific heat at constant pressure, C_p . Using this we can re-write Equation 12 as:

$$\eta_{\text{reg}} = \frac{C_p(T_3 - T_2)}{C_p(T_5 - T_2)} \quad (13)$$

This does not look like a great leap forward but we can now see that Equation 13 leaves another detail out: the specific heat is not constant. A plot of specific heat at constant pressure for the high pressure and low pressure side of the regenerator is shown in Figure 48.

An improvement can be made by refining our definition of regenerator effectiveness further³⁴ as follows:

$$\eta_{\text{reg}} = \frac{q}{q_{\max}} = \frac{C_c(T_{c,o} - T_{c,i})}{C_{\min}(T_{h,i} - T_{c,i})} = \frac{C_h(T_{h,i} - T_{h,o})}{C_{\min}(T_{h,i} - T_{c,i})} \quad (14)$$

where C_c and C_{\min} refer to the specific heat at constant pressure for the cold stream and the minimum specific heat at constant pressure respectively. But what is a good choice for either C_c or C_h ? —they both change significantly over the temperature range of interest.

THE METHOD WE WILL USE is based on techniques described in a technical report from MIT³⁵ and is given in Equation 15.

$$\eta_{\text{reg}} = \frac{h_3 - h_2}{h_5 - h_{P_6,T_2}} = \frac{h_5 - h_6}{h_5 - h_{P_6,T_2}} \quad (15)$$

where h_{P_6,T_2} is the enthalpy evaluated at the pressure of state point 6 (low pressure outlet), but the temperature at state point 2 (low temperature inlet). This ensures that the denominator of Equation 15 represents a maximum enthalpy change.

Supercritical CO₂ Regenerative Cycle Revisited

Using this new method the cycle will be re-analyzed with all of the same parameters as before. The results are presented in Figure 50. Once again, as expected, there is no change to the net specific work. For this analysis, the thermal efficiency is a respectable 37.7% and there is no negative exergy destruction.

YOU SHOULD NOTE the fairly high value of exergy destruction for the regenerator; particularly considering the optimistic regenerator

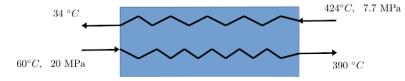


Figure 47: Regenerator temperatures.

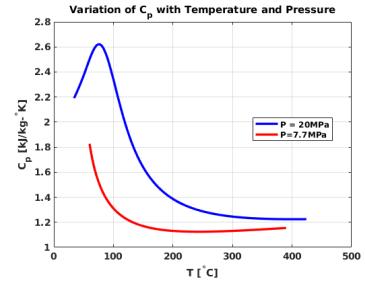


Figure 48: Variation of C_p within the regenerator.

³⁴ Theodore L Bergman, Adrienne S Lavine, Frank P Incropera, and David P DeWitt. *Introduction to Heat Transfer*. John Wiley & Sons, 2011

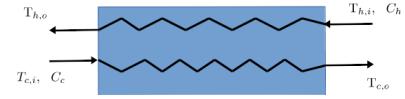


Figure 49: Regenerator schematic

³⁵ Vaclav Dostal, Michael J Driscoll, and Pavel Hejzlar. A supercritical carbon dioxide cycle for next generation nuclear reactors. 2004

	P	T	h	s	ef
1	7700	32	306.23	1.3463	214.78
2	20000	59.999	324.5	1.3463	233.04
3	20000	290.11	715.11	2.2738	347.25
4	20000	550	1035.1	2.7411	528.04
5	7700	424.55	896.34	2.7411	389.24
6	7700	87.807	505.73	1.9724	227.7

Thermal efficiency = 37.6636 percent
Back work ratio = 13.1581 percent
Flow exergy in: 180.793 kJ/kg
Flow exergy out: 12.9293 kJ/kg
Net specific work = 120.534 kJ/kg
Exergy destroyed in the regen: 47.3296 kJ/kg

Figure 50: Regenerated S-CO₂ results with revised η_{reg}

effectiveness that we assumed. Recall that exergy destruction is how we quantify irreversibility, and for heat transfer processes such as what occurs in the regenerator, irreversibility comes from transferring heat across temperature gradients.³⁶ Plotting, as we did before, the temperatures at the inlets and outlets of the regenerator in Figure 51, we get some idea as to the source of the extensive exergy destruction. Note the high differential temperature between the fluids at the right end of the regenerator as well as the significantly different change in temperature for the low- and high-pressure fluids respectively.

THE VARIATION IN TEMPERATURE change can be better understood in light of the variation of C_p as shown before in Figure 48. On the high-pressure side, especially at lower temperatures, C_p is much higher than it is at high temperature and low-pressure side. This means that, if we have the same mass flow rate of working fluid on both the low- and high-pressure side of the heat exchanger, we will not be able to avoid having large differences in the temperature change.

A resolution to this issue will be introduced in the next lecture. We will break the regenerator up into two stages: a low temperature regenerator and a high temperature regenerator. In the low temperature regenerator, where the variation in C_p between the fluids is greatest, we will reduce the mass flow rate on the high-pressure/low-temperature side. This will allow the temperature changes on each side of the regenerator to be better matched. Overall this scheme will result in effective regeneration with reduced exergy destruction and correspondingly improved efficiency.

³⁶ Of course you need at least an infinitesimal temperature gradient to transfer heat; reversible heat transfer must take place across such an infinitesimal temperature gradient.

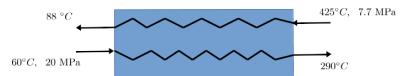


Figure 51: Regenerator temperatures with revised effectiveness.

Lecture 13 - Recompressing, Regenerating S-CO₂ Cycle

Objectives

The objectives of this lecture are:

1. Give a detailed description of the analysis method for the RRS-CO₂ Brayton cycle
2. Discuss the MATLAB implementation of the solution

Recompressing Regenerating S-CO₂ Brayton Cycle

A SCHEMATIC OF THIS CYCLE is given in Figure 52. In order to better match the temperature change on the cold- and hot- side of the regenerators, a portion of the working fluid bypasses the pre-cooler, compressor and low-temperature regenerator; instead it is recompressed and mixed with fluid leaving the low-temperature regenerator at state point 3. The combined flow is pre-heated in the high-temperature regenerator before heat addition from the reactor.

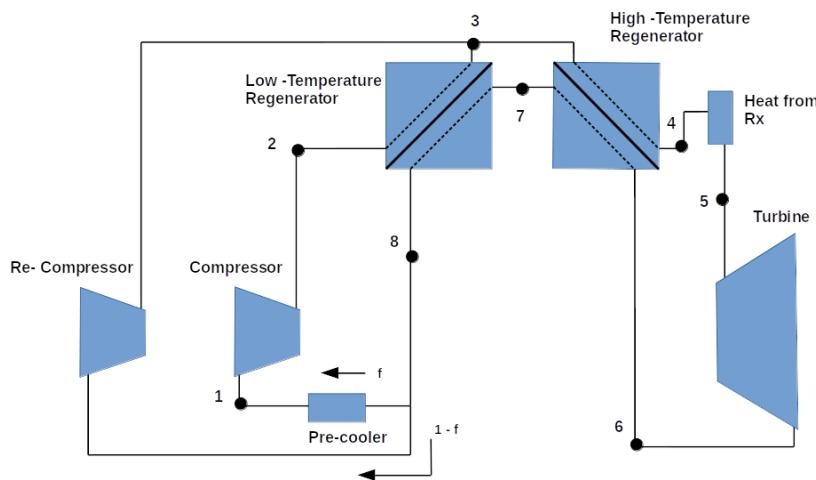


Figure 52: Schematic RR S-CO₂ Brayton cycle.

COMPARED TO some of the more complex Rankine cycles analyzed in earlier lectures, this cycle is relatively simple with only 8 state points. Still, there are some challenges to be overcome. Let us suppose that we know the minimum and maximum pressure and temperature for the working fluid; we know (or can assume appropriate values for) the isentropic efficiency of the turbine and both compressors and we can assume a value of regenerator effectiveness for both the low- and high-temperature regenerators.

- With the aforementioned given and assumed data, we should be able to calculate thermodynamic properties at state point 1 and 2 along with 5 and 6.
- If we do not know the flow fraction f of working fluid going through the pre-cooler then we cannot calculate the temperature or enthalpy of state point 3;
- Without knowing the temperature or enthalpy of state point 3, we cannot calculate properties of state point 4, 7, or 8 either.

We have a total of 5 unknown properties— f , and the enthalpy/temperature of state points 3, 4, 7, and 8—and thus need to develop 5 independent relationships. As with previous cycles, these equations in general will be non-linear and values of the dependent variables need to be constrained to ensure a physically meaningful solution. We will form these equations using the following relationships:

1. Low-temperature regenerator effectiveness:

$$\eta_{LTR} = \frac{h_7 - h_8}{h_7 - h_{(P_8, T_2)}}$$

```
%f = [flow fraction, h(3), h(4), h(7), h(8)]
2 gas = py.EasProp.simpleFluid('CO2', 'SI');
LTR_eff_eq = @(f) eta_LTR * (f(4) - ...
4 gas.h_pT(P(8), T(z))) - (f(4) - f(5));
```

Note: The equations are all re-arranged in the MATLAB code to facilitate use with FMINCON.

Partial MATLAB code listing provided for clarity/completeness. A full MATLAB script to carry out this analysis is provided in the appendices.

2. High-temperature regenerator effectiveness:

$$\eta_{HTR} = \frac{h_6 - h_7}{h_6 - h_{(P_7, T_2)}}$$

```
%f = [flow fraction, h(3), h(4), h(7), h(8)]
2 HTR_eff_eq = @(f) eta_HTR * (h(6) - ...
gas.h_pt(P(7), gas.T_ph(P(3), f(z)))) - (h(6) - f(4));
```

3. Low-temperature regenerator energy balance:

$$f(h_3 - h_2) = h_7 - h_8$$

```

1 %f = [flow fraction , h(3) , h(4) , h(7) , h(8)]
2 LTR_ebal = @(f) f(1)*(f(2)-h(2))-(f(4)-f(5));

```

4. High-temperature regenerator energy balance:

$$h_4 - h_3 = h_6 - h_7$$

```

1 %f = [flow fraction , h(3) , h(4) , h(7) , h(8)]
2 HTR_ebal = @(f) (f(3)-f(2) - (h(6)-f(4));

```

5. The last relationship will be that the state of the fluid at exit from the low-temperature regenerator is the same as the state of the fluid discharged from the re-compressor. On the schematic, both of these flows correspond to state point 3:

$$h_3 = h_8 - \frac{h_8 - h_{3s}}{\eta_{RC}}$$

where the left hand side is derived from a standard analysis of compressor work for the re-compressor. Note for this equation, we do not know h_{3s} and we do not want to add it to our list of unknown variables. Instead we will re-express h_{3s} in terms of what we do know along with existing unknowns:

$$h_{3s} = h_8 - \frac{h_8 - (h_8 - \overbrace{h(P_3, s(P_8, h_8))}^{s_8})}{\eta_{RC}}$$

```

1 %f = [flow fraction , h(3) , h(4) , h(7) , h(8)]
2 sp3_ident = @(f) f(5) - ...
    (f(5) - gas.h_ps(P(3),gas.s_ph(P(8),f(5))))/eta_rc - f(2);

```

The five equations are then combined into a single balance equation for use with FMINCON.

```

1 balance = @(f) abs(LTR_eff_eq(f)) + abs(HTR_eff_eq(f)) + ...
    abs(LTR_ebal(f)) + abs(HTR_ebal(f)) + abs(sp3_ident(f));

```

WE ALSO NEED to provide constraints to help ensure that FMINCON arrives at a physically meaningful solution. A table of possible constraints and starting values are provided in Table 6. These are encoded in MATLAB as follows:

```
%f = [flow fraction, h(3), h(4), h(7), h(8)]
1 lbound = [0 h(2) h(2) h(1) h(2)];
ubound = [1 h(6) h(5) h(6) h(6)];
4 Xo = [0.5 h(2) h(5) h(6)-1 h(2)];
```

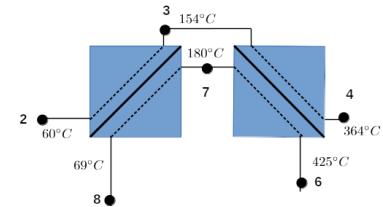
Results

AS A REMINDER, our goal was to reduce exergy destruction in the regenerator for a supercritical CO₂ Brayton cycle and thereby increase cycle thermal efficiency. Results of the analysis are shown in Figure 53.

We can see that the thermal efficiency is increased to 48.3% and that the temperature differences at each side of the regenerators is reduced. The largest differential temperature is at the high-temperature regenerator with state points 4 and 6 roughly 60 degrees different. It corresponds, therefore that the higher rate of exergy destruction is in the high-temperature regenerator. The net specific work (and heat) is reduced because the specific work of the re-compressor is elevated since it is compressing working fluid that has not been pre-cooled.

Constraint	Starting Value
$0 \leq f \leq 1$	0.5
$h_2 \leq h_3 \leq h_6$	h_2
$h_2 \leq h_4 \leq h_5$	h_5
$h_1 \leq h_7 \leq h_6$	$h_6 - 1$
$h_2 \leq h_8 \leq h_6$	h_2

Table 6: Constraints and starting values for cycle analysis.



Net specific heat = 110.282 kJ/kg
Thermal efficiency = 48.31 percent
Back work ratio = 20.5446 percent
Specific flow exergy in: 134.653 kJ/kg
Specific flow exergy out: 5.9878 kJ/kg
Specific flow exergy destroyed LTR: 5.29151 kJ/kg
Specific flow exergy destroyed HTR: 13.0917 kJ/kg

	P	T	h	s	ef
1	7700	32	306.23	1.3463	214.78
2	20000	59.999	324.5	1.3463	233.04
3	20000	153.56	528.98	1.8933	274.53
4	20000	364.35	806.85	2.4269	393.39
5	20000	550	1035.1	2.7411	528.04
6	7700	424.55	896.34	2.7411	389.24
7	7700	180.37	618.47	2.2514	257.29
8	7700	69.126	477.94	1.8933	223.49

Figure 53: Calculated results for RR S-CO₂ cycle.

Lecture 14 - Combined Cycles

Objectives

The objectives of this lecture are:

1. Describe an example combined cycle and illustrate key analysis methods
2. Illustrate incorporation of hydraulic pressure drops in a Brayton cycle.

Combined Cycle by Example

A DIFFERENT APPROACH TO improving the thermal efficiency of the nuclear energy conversion process is to use a *combined cycle*. In this concept, we use thermal energy remaining in the working fluid after expansion in a turbine as input into *another* thermodynamic cycle; a portion of that heat is thus converted to work. Since this was done with no additional heat being added, thermal efficiency is improved by increasing the net specific work. Compare this with regenerating Brayton cycles that increase thermal efficiency by reducing the amount of heat that must be added per unit mass of working fluid with no change in specific work output. In principle there are many different types of combined cycles one might explore but in this course we will focus on Brayton-Rankine combined cycles where waste heat from a Brayton cycle is used as heat input for a Rankine “bottoming cycle.” One such cycle³⁷ is illustrated in the schematic in Figure 54. Key system parameters are summarized in Table 7. State point property data are tabulated in Figure 55.

ONE NOTABLE FEATURE of this analysis is the inclusion of hydraulic pressure losses. In previous cycle analyses, it was assumed either explicitly or implicitly that heat transfer processes in a Brayton cycle would take place at constant pressure. In reality, viscous flow

³⁷ Charles Forsberg and Daniel Curtis. Meeting the needs of a nuclear-renewables electrical grid with a fluoride-salt-cooled high-temperature reactor coupled to a nuclear air-brayton combined-cycle power system. *Nuclear Technology*, 185(3):281–295, 2014

through complex heat exchanger geometry results in frictional pressure drops.

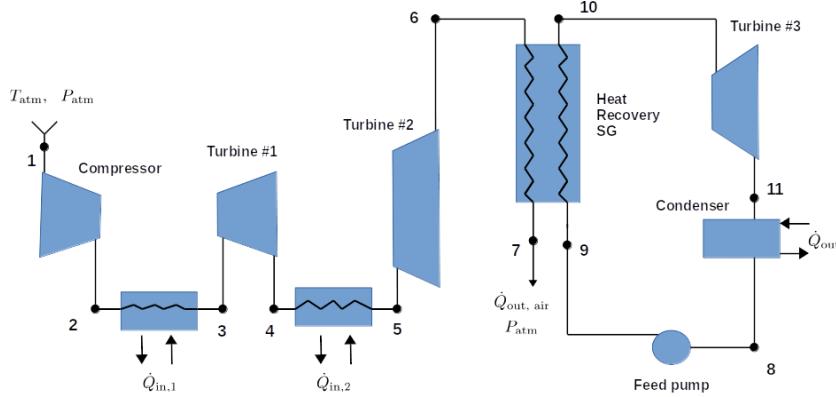


Figure 54: Schematic of a FHR combined cycle

Property	Value
T_{atm}	15°C
P_{atm}	101.3 kPa
T_{max}	770°C
$T_{exh,air}$	250°C
r_p	17
r_{e1}	4
η_T	0.92
η_c	0.89
$\eta_{ta,p}$	0.80

Table 7: FHR Parameters

	<i>T</i>	<i>P</i>	<i>h</i>	<i>s</i>
1	15	100.3	414.38	3.8491
2	407.56	1705.1	819.2	3.9166
3	770	1695.1	1222.9	4.3926
4	482.65	423.77	900.16	4.4304
5	770	413.77	1222.1	4.7977
6	503.72	116.3	923.06	4.8317
7	250	106.3	653.29	4.4379
8	41.509	8	173.84	0.59249
9	41.825	3500	178.24	0.59529
10	242.56	3500	2802.6	6.1243
11	41.509	8	1985.5	6.3501

Figure 55: State point property data for FHR.

Take for example air flow into the compressor. Air is drawn from the atmosphere, here assumed to be at standard temperature and pressure, through inlet ducting which, besides directing air flow to the compressor inlet would perform other essential functions such as preventing the introduction of harmful foreign objects into the machine. It should be clear that the fluid arriving at state point 1 will have lost some energy in the process of traversing the dampers, filters, and various duct work. That energy loss will be manifest as a reduction in static pressure. As is evident from the state point table, the pressure at state point 1 is listed as 100.3 kPa—1 kPa lower than atmospheric pressure.

Similarly, air flowing from the compressor discharge through the heat exchanger between state points 2 and 3 cannot be expected to pass through the complex heat exchanger geometry without some flow-energy losses. For this problem, the pressure loss $\Delta P_{HX} = P_3 - P_2 = 10\text{kPa}$. In upcoming lectures we will examine techniques for quantifying hydraulic pressure losses in the context of diabatic flows such as this one.

For the purposes of lectures and associated homework problems, we will take hydraulic pressure losses to be given parameters. The losses will be expressed either as:

1. fixed given ΔP as with the two cases just described; or
2. a fractional pressure drop where the ΔP for a given process will be given as a fraction of the state point pressure at the process inlet.

With this methodology it is often possible and preferable to compute all state point pressures first before determining all other state point properties.

THE HYDRAULIC PRESSURE DROPS have a significant effect on net specific work and thermal efficiency. If, as we have done up until now, we assumed no hydraulic pressure losses throughout the system, the net specific work and thermal efficiency would be 324.4 kJ/kg and 44.7%. Including pressure losses, net specific work is 300.5 kJ/kg and thermal efficiency is only 41.4%.

Analysis of Combined Cycle

There are a few details to the analysis of a combined cycle that are best communicated simply by doing the calculations.

1. What is the ratio of the air and steam mass flow rates?

This is determined through an energy balance for the heat recovery steam generator expressed in Equation 16

$$\dot{m}_a(h_6 - h_7) = \dot{m}_s(h_{10} - h_9) \quad (16)$$

For this problem, therefore:

$$\frac{\dot{m}_s}{\dot{m}_a} = \frac{h_6 - h_7}{h_{10} - h_9}$$

2. What is the net specific work for this cycle?

In this case, we need to be a bit more specific about what we mean by "specific work." Normally the specific work is the power divided by the mass flow rate of the working fluid; but in this cycle

we have two different working fluids with generally different flow rates. In order to perform a consistent analysis of the cycle we should define specific work to be power per unit mass flow rate of *one* of the working fluids; for this cycle we will choose the air in the Brayton cycle as the reference working fluid.

$$\begin{aligned} w_{\text{net}} &= \frac{\dot{W}}{\dot{m}_a} \\ &= w_c + w_{T_1} + w_{T_2} + w_{T_3} + w_p \\ &= (h_1 - h_2) + (h_3 - h_4) + (h_5 - h_6) + \frac{\dot{m}_s}{\dot{m}_a}(h_{10} - h_{11}) + \frac{\dot{m}_s}{\dot{m}_a}(h_8 - h_9) \end{aligned}$$

3. What is the net specific heat added?

For this cycle, heat is added only in the processes from state points 2 to 3 and 4 to 5.³⁸ Since air is the only working fluid involved in these processes, this calculation is carried out as before:

³⁸ **Question:** Why do you not count the heat exchanged in the heat recovery SG?

$$q_s = (h_3 - h_2) + (h_5 - h_4)$$

4. What is the cycle thermal efficiency?

Since we have taken care to consistently calculate both the net specific heat added and the net specific work in terms of the mass flow rate of air, we can compute thermal efficiency in the usual way:

$$\eta_{\text{TH}} = \frac{w_{\text{net}}}{q_s}$$

5. What would the thermal efficiency be without the Rankine "Bottoming" cycle?

For this calculation we would simply eliminate the specific work of the Rankine cycle turbine and pump from the net work calculation.

Assignment #5: Advanced Brayton Cycle & Combined Cycle Analysis

Use EASYPROP and MATLAB to solve the following problems:

1. The Marine Gas Cooled Reactor was proposed by General Atomics to the Atomic Energy Commission in 1961. The design concept used a closed, direct, Brayton cycle to produce power. The Brayton cycle includes two compressors (with inter-cooling), a regenerator, and one turbine. The working fluid is helium, the maximum helium temperature is 1300°F, the minimum temperature is 100°F, and the minimum pressure is 450 psia. The isentropic efficiency of the compressors are each 85% and the isentropic efficiency of the turbine is 90%. The regenerator effectiveness is assumed to be 90%. The inlet temperature to both compressors is 100°F and the pressure ratios of the compressors are equal.
 - (a) Make a simple sketch of the cycle labeling all state points and given information.
 - (b) Find the thermal efficiency if $r_{p,1} = r_{p,2} = 2.0$
 - (c) Find the pressure ratio that gives the maximum thermal efficiency along with the thermal efficiency thus obtained; and
 - (d) Using the pressure ratio that gives maximum thermal efficiency, find the helium flow rate [lb_m/s] and net power output [MW] if the thermal output of the reactor is 100 MW. (**Note:** 1 MW = 3.41×10^6 BTU/hr)
2. Consider the cycle described in problem #1. Assume that there are hydraulic pressure losses for each heat exchanger equal to 2% of the incoming pressure (e.g. $P_{\text{out}} = 0.98 \times P_{\text{in}}$); pressure losses across the reactor is 5% of the incoming pressure. What is the maximum thermal efficiency, helium flow rate, and net power output for 100 MW (thermal) reactor power output if hydraulic losses are included? Investigate the individual impact of pressure loss in each component and write a short discussion of your findings. (**Note:** In your analysis you should consider determining all state point pressures first since there will be, in general, no simple relationship between the compressor pressure ratios and the turbine expansion ratio. This is owing to the pressure drops in the regenerator (both low temperature and high temperature side), reactor and pre-cooler. Assume that, under all conditions, minimum system pressure is constant at 450 psia)

3. Consider a closed, direct Brayton cycle. Helium at 900 psia at 1500°F leaves a High Temperature Gas-cooled Reactor (HTGR) and enters a turbine where the helium is expanded down to 1000 psia. The helium exiting the turbine is used as the heat source in a Rankine cycle; the helium exits the boiler at 300°F and then enters a compressor. The steam produced in the boiler is expanded in a turbine to 1 psia and exhausted into a condenser which rejects wasted heat and delivers saturated liquid to a pump. The pressure loss of the helium in the boiler and HTGR is 10 psia in each component. For the Rankine cycle, assume that isobaric heat transfer occurs in the boiler and condenser. The isentropic efficiency of all pumps, compressors, and turbines is 100%. (**Hint:** consider using the EasyProp function "P_sT" to get the turbine exhaust pressure.)
- (a) Draw a simple sketch of the system indicating given temperature and pressure data.
 - (b) What is the ratio of helium mass flow rate and steam mass flow rate: $\frac{\dot{m}_{\text{steam}}}{\dot{m}_{\text{helium}}}$
 - (c) What is the compressor pressure ratio for the compressor in the Brayton cycle?
 - (d) What is the net work from the Rankine cycle and Brayton cycle per unit mass flow rate of helium in BTU/lb_m?
 - (e) What is the thermal efficiency of the combined cycle?
 - (f) What is the net specific work and thermal efficiency of the combined cycle if CO₂ is used (same temperatures and pressures) instead of helium? Briefly discuss the relative performance between CO₂ and helium (e.g. efficiency, net specific work, mass flow rates of gas for a given power output, etc ...)

Part II

Hydraulics

Lecture 15 - Hydraulics I, Internal Flow Review

Objectives

The objectives of this lecture are:

1. Introduce the goals of hydraulic analysis
2. Briefly review some of the theory of hydraulic analysis
3. Do a simple example problem

Introduction to Hydraulic Analysis

Purpose:

1. Estimate pressure losses in fluid systems
 - (a) quantify impact on energy conversion systems—especially gas power cycles
 - (b) estimate pumping power for primary coolant—i.e. low for PWRs, high for GCRs
 - (c) evaluate pressure distribution, especially in the core to evaluate effects on convective heat transfer.
2. Quantitatively evaluate engineering trade-offs

As an example to try and illustrate what is meant by the last bullet, consider some of the following example questions:

1. What should the diameter of the hot-leg piping be for an AP1000 reactor? This may seem quite arbitrary but there are important considerations to take into account.

	Bigger Piping	Smaller Piping
Pumping Power Required	↓	↑
Material Costs for Piping	↑	↓

As you can probably intuitively guess, larger diameter piping results in less hydraulic losses and thus reduced pumping power

hot-leg is the section of piping at the outlet of the reactor core. For a PWR this piping spans from the reactor vessel to the steam generator.

with the opposite effect for smaller piping. The hot leg piping is also an important pressure boundary and must withstand primary system pressure. If the pipe is larger diameter then, for a given allowed stress intensity, the wall thickness must be increased.³⁹ Thicker walls for a larger diameter pipe implies greatly increased material costs.

2. Consider a shell-and-tube heat exchanger intended for use as a regenerator for a Brayton cycle energy conversion system.
 - (a) Which fluid should flow on the shell-side and which on the tube side?
 - (b) How many tubes should there be? What are the dimensions? (length, diameter, wall thickness) From what material should they be constructed?
 - (c) What impact do the above decisions have on the effectiveness of the regenerator?
 - (d) What is the impact on hydraulic losses in the regenerator and consequently to the performance of the energy conversion cycle?

Evaluating Pressure Drop

From basic fluid dynamics class, we derived the Bernoulli Equation based on conservation of energy to quantify the fluid state for flow along a streamline as schematically illustrated in Figure 56.

$$\left(\frac{P}{\gamma} + \frac{v^2}{2g} + z \right)_{\text{out}} = \left(\frac{P}{\gamma} + \frac{v^2}{2g} + z \right)_{\text{in}} + h_s - h_L \quad (17)$$

where P is fluid static pressure, γ is the specific weight, v is fluid velocity, g is the gravitational acceleration, z is fluid elevation⁴⁰, h_s is specific shaft work, and h_L is fluid head losses.⁴¹

Head Loss

The head loss (h_L) is a function of:

1. fluid properties—density and viscosity
2. fluid flow rate; and
3. piping system properties
 - (a) pipe diameter
 - (b) pipe material surface roughness; and

³⁹ Recall for cylindrical pressure vessels:
 $S_m = \frac{\sigma_y}{\text{FOS}} = \frac{Pr}{t} + \frac{P}{2}$ where P is the interior pressure, r is the vessel radius, and t is the wall thickness. To satisfy a given S_m and factor of safety (FOS) allowance: $t_{\min} \geq \frac{Pr}{S_m - 0.5P}$

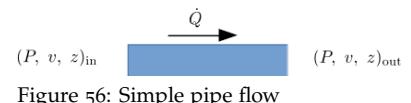


Figure 56: Simple pipe flow

⁴⁰ Elevation relative to an arbitrary reference.

⁴¹ Both h_s and h_L need to be specified in units of *length* to be used in this equation.

- (c) features of the piping system like bends, turns, elbows, flow-obstructions, etc...

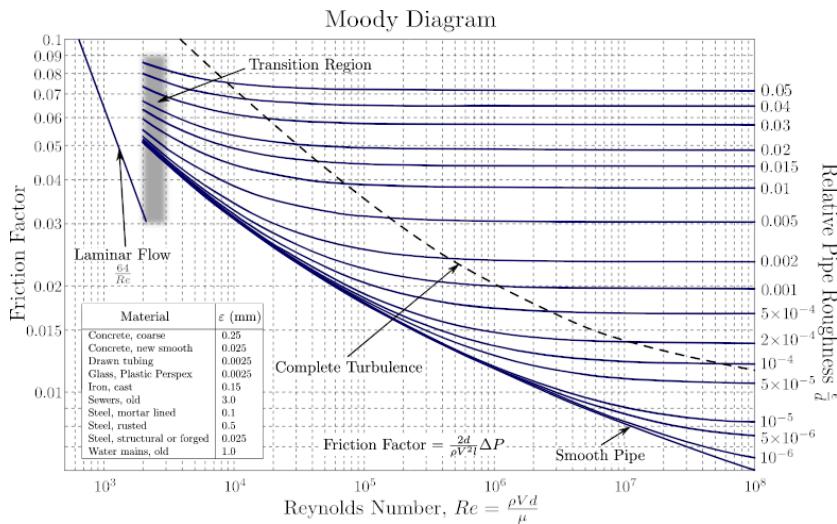
and will be expressed as shown in Equation 18

$$h_L = \left(\underbrace{\frac{fL}{D}}_{\text{Major/friction}} + \underbrace{\sum k_L}_{\text{Minor/form}} \right) \frac{v^2}{2g} \quad (18)$$

where f is the Darcy friction factor, L is the length of the piping segment between input/output, D is the pipe diameter, and k_L are coefficients characterizing minor head losses.

Major Head Losses

In order to calculate major head losses, one needs to get a value for the Darcy friction factor f . The friction factor is characterized by two dimensionless numbers: the Reynolds number and the relative roughness. Over a wide range of Reynolds numbers and relative roughness values the friction factor has been quantified and is traditionally displayed in graphical format in the Moody diagram as in Figure 57.



Reynolds number is given by: $Re = \frac{\rho v L}{\mu}$ where ρ is the fluid density, v is the average fluid velocity, L is a characteristic length which, for pipe-flow problems is the pipe diameter, and μ is the fluid viscosity.

Relative roughness is the feature size on the surface of the pipe (ϵ) divided by the diameter of the pipe D

Figure 57: The Moody Diagram for Darcy friction factor.

THIS REPRESENTATION is excellent for developing intuition on how the friction factor depends on Reynolds number and relative roughness. In the laminar flow regime— $0 \leq Re \leq 2000$ —friction factor is independent of the roughness and inversely proportional to Reynolds number:

$$f = \frac{64}{Re}$$

At high Reynolds number—In the region to the right of the “Complete Turbulence” dashed line in Figure 57—friction factor is independent of Reynolds number and only a function of relative roughness. Between these two regions is an area where friction factor is significantly impacted by both Reynolds number and relative roughness.

HOWEVER IMPORTANT such intuition may be, computationally it may be inconvenient to *require* the analyst to manually read the diagram to obtain friction factor. An alternative approach is to use the Colebrook Equation provided in Equation 19.

$$\frac{1}{\sqrt{f}} = -2.0 \log \left(\frac{\epsilon/D}{3.7} + \frac{2.51}{Re \sqrt{f}} \right) \quad (19)$$

This equation is easily incorporated into MATLAB or Python scripts that have built-in root-finding tools allowing for parametric searches or iterative-algorithms that may be necessary for solving some hydraulic analysis problems.

Minor Head Losses

MINOR HEAD LOSSES account for energy losses that occur when a viscous fluid flows around corners, through pipe contractions or expansions, through open or partially open valves, or any other piping feature that disrupts flow. Values for minor loss coefficients have been tabulated for many common piping system arrangements and, for the typical student of ER468, has already been studied in EM324 Fluid Dynamics.

In ER468 we seek to extend that training to include nuclear-specific design features that contribute to minor head losses. These include minor losses due to:

- flow past nuclear fuel rod bundles in either rectangular or hexagonal arrangements, including those fitted with mixing vanes and structural spacer grids; and
- flow through heat exchangers fitted with enhanced surface features intended for improvement of convective heat transfer

Fluid Flow Problem Types

In EM324 we discussed three hydraulic problem types:

- Type I—volumetric flow rate and pipe geometry given: find head loss;

- Type II—pipe geometry and head loss given: find volumetric flow rate; and
- Type III—head loss and volumetric flow rate given: find pipe geometry (e.g. diameter)

Most of the problems that we will pose in this class will fall into the Type I category and this category is in many ways the easiest. Due to the non-linear nature of Equation 19 problems of Type II or Type III generally require an iterative solution approach. Nonetheless, problems of these types are definitely relevant for the engineering analysis of a nuclear reactor. A thermal analysis of the core may dictate a given mass or volumetric flow rate; other limitations may impose some constraint on allowable head loss and your goal is to find an acceptable geometry. A well-prepared nuclear engineer should be able to meet any of these challenges.

Example: The hot leg of an AP1000 reactor is 16 ft long and has an inside diameter of 31 inches. The piping material has a relative roughness of 0.001. The fluid properties of water in the hot leg are constant and given in Table 8. Assume that a) the hot leg piping is of constant diameter; b) that minor losses are negligible, and c) that the hot leg piping is on roughly the same elevation from the reactor vessel outlet to the steam generator inlet. **Find:** the pressure drop (psid) as coolant flows through the hot leg.

Parameter	Value
Density	42.282 lb _m /ft ³
Viscosity	1.67 × 10 ⁻⁶ lb _m /ft-s
Mass flow rate	16,744 lb _m /s

Table 8: Fluid properties for example problem

Solution: From the Bernoulli equation under given assumptions:

$$\left(\frac{P}{\gamma} + \frac{v^2}{2g} + z \right)_{\text{out}} = \left(\frac{P}{\gamma} + \frac{v^2}{2g} + z \right)_{\text{in}} + h_s^0 - h_L$$

where the kinetic and potential energy terms are eliminated since they do not change over the length of the hot leg piping. This leaves:

$$\frac{P_{\text{in}} - P_{\text{out}}}{\gamma} = h_L = \frac{fL}{D} \frac{v^2}{2g}$$

The velocity is obtained from the continuity equation: $\dot{m} = \rho v A$ where A is the cross-sectional area of the piping. Thus:

$$v = \frac{\dot{m}}{\rho A} = 75.6 \text{ ft/s}$$

With the velocity, Reynolds number can be found:

$$\text{Re} = \frac{\rho v D}{\mu} = 4.9 \times 10^9$$

which is sadly off the scale of Figure 57 but definitely falls into the “Complete Turbulence” region so the friction factor can be obtained from the relative roughness alone. Since $\epsilon/D = 0.001$, reading the Moody chart gives us a Darcy Friction factor: $f \approx 0.02$.

Applying these calculated values with appropriate conversion factors gives us our result:

$$\Delta P = 3.2 \text{ psid}$$

Note: γ is the specific weight, but the density was given. In the USCS unit system, $\gamma = \rho \frac{g_c}{g}$ where $g_c = 32.2 \frac{\text{lb}_f}{\text{lb}_m \cdot \text{s}^2}$ and $g = 32.2 \frac{\text{ft}}{\text{s}^2}$ thus $\gamma = 42.282 \frac{\text{lb}_f}{\text{ft}^3}$.

Lecture 16 - Hydraulics II, Friction Factor and Rod-Bundle Flow

Objectives

The objectives of this lecture are:

1. List friction factor correlations for adiabatic and non-adiabatic flow
2. Provide a friction factor correlation to use with nuclear rod bundles.

Friction factor for Non-adiabatic and Adiabatic Flow

IN GENERAL, the friction factor that we use to calculate major head losses for internal flows depends not only on the geometry but whether or not the fluid is being heated, cooled, or neither. Most experimentally-based correlations are for either flow in a circular tube or are developed in reference to flow in a circular tube; that's the geometric connection. If a fluid is being heated or cooled (i.e. non-adiabatic) then a temperature profile is developed in the fluid at the thermal boundary layer that is different than the constant temperature profile—and consequent profile for temperature-dependent fluid properties—for adiabatic flows. All of these have non-negligible impact on viscous energy loss and related hydraulic pressure drop.

FOR ADIABATIC FLOW in a smooth tube, we use the Karman-Nikuradse equation:

$$\frac{1}{f} = -0.8 + 0.87 \ln (\text{Re} \sqrt{f})$$

Like the Colebrook equation which is used for adiabatic flow in non-smooth pipes, the Karman-Nikuradse equation requires an iterative solution that may be inconvenient. Two frequently used approximations include the equation due to Blasius:

$$f = 0.316 \text{Re}^{-0.25}, \quad 4000 < \text{Re} < 10^5$$

or that due to McAdams:

$$f = 0.184 \text{Re}^{-0.20}, \quad 10^4 < \text{Re} < 10^6$$

Although it was mentioned in the previous lecture, for non-smooth walls we use the Colebrook equation:

$$\frac{1}{\sqrt{f}} = -2.0 \log \left(\frac{\epsilon/D}{3.7} + \frac{2.51}{\text{Re}\sqrt{f}} \right)$$

A convenient approximation to the Colebrook equation is available using an equation due to Haaland⁴²:

$$\frac{1}{\sqrt{f_{\text{Haaland}}}} = -1.8 \log_{10} \left[\frac{6.9}{\text{Re}} + \left(\frac{\epsilon/D}{3.7} \right)^{1.11} \right]$$

or another equation provided by Churchill⁴³:

$$f_{\text{Churchill}} = \frac{8}{6.0516} \left\{ \ln \left[\frac{\epsilon/D}{3.7} + \left(\frac{7}{\text{Re}} \right)^{0.9} \right] \right\}^{-2}$$

In all cases above, the Reynolds number is based on the diameter of the circular channel. In cases where the channel is not circular and where you would like to use the correlations anyway, you should calculate the Reynolds number relative to the *hydraulic diameter*.

Diabatic Flow

IF HEAT IS BEING ADDED or removed from the fluid then there is a temperature gradient within the fluid. Temperature-dependent properties like density and viscosity vary in the direction along which heat is moving. For example, when heat is transferred from primary to secondary coolant in the steam generator of a PWR, primary coolant next to the U-tube wall is at a lower temperature than coolant near the center of the tube. Consequently the viscosity of the fluid—which for liquids tends to vary inversely with temperature—will be higher. This changes the friction factor and that change should be taken into account.

For this class we will use the correlation due to Petukhov⁴⁴. This correlation includes a basic friction factor:

$$f_{T_b} = (1.82 \log_{10} \text{Re}_D - 1.64)^{-2}, \quad 3000 < \text{Re} < 5 \times 10^6$$

where Re_D is the Reynolds number based on tube diameter (or hydraulic diameter) with fluid properties evaluated at their bulk average (i.e. not along the tube wall) temperature. The final friction factor

⁴² Skjalg E Haaland. Simple and explicit formulas for the friction factor in turbulent pipe flow. *J. Fluids Eng.*, 105: 89–90, 1983

⁴³ Stuart W Churchill. Empirical expressions for the shear stress in turbulent flow in commercial pipe. *AIChE J.*, 19(2):375–376, 1973

Note: If you have multiple approximations, it does not hurt to try both! Pick whichever one is the most accurate or conservative in accordance with your willingness to accept risk.

hydraulic diameter is computed as: $D_h = \frac{4A_{\text{flow}}}{P_{\text{wetted}}}$, where A_{flow} is the cross sectional flow area and P_{wetted} is the wetted perimeter.

⁴⁴ Boris Sergeevich Petukhov. Heat transfer and friction in turbulent pipe flow with variable physical properties. In *Advances in heat transfer*, volume 6, pages 503–564. Elsevier, 1970

Note: The term T_b , as part of f_{T_b} , should be read as "...at bulk temperature."

is then found using the following formula depending on whether the fluid is a gas or liquid and whether the fluid is being heated or cooled. For liquids:

$$\frac{f}{f_{T_b}} = g\left(\frac{\mu_b}{\mu_w}\right), \quad 0.5 \leq \mu_b/\mu_w \leq 3$$

where μ_b is the bulk viscosity and μ_w is the viscosity of the fluid adjacent to the wall. The function $g(\mu_b/\mu_w)$ is different for heating and cooling. For heating:

$$g(\mu_b/\mu_w) = \frac{1}{6} \left(7 - \frac{\mu_b}{\mu_w}\right)$$

For cooling:

$$g(\mu_b/\mu_w) = \left(\frac{\mu_b}{\mu_w}\right)^{-0.24}$$

For gasses, the relation is:

$$\frac{f}{f_{T_b}} = \left(\frac{T_b}{T_w}\right)^{0.23}, \quad 0.14 \leq T_b/T_w \leq 3.3$$

This relation is used for both heating and cooling.

Pressure Drop in Rod Bundles

THE FRICTION FACTOR for fluid flow in a rod bundle depends on a number of factors. The factors that we will take into account with the correlation we will use include:

- whether the flow is laminar or turbulent;
- whether the rod bundle is a rectangular or hexagonal array;
- channel position within a fuel assembly; and
- how closely packed the rod bundles are

the last term we will parameterize by the pitch-to-diameter ratio of the rod bundle. The channel position may correspond to an interior, edge, or corner channel as illustrated by Figure 58.

All of these factors are included in the Cheng-Todreas correlation⁴⁵ that we will use. The general form of the correlation is given by:

$$f = \frac{C}{Re^n}$$

The numerator C is a quadratic function of pitch-to-diameter ratio:

$$C = a + b_1 \left(\frac{P}{D} - 1\right) + b_2 \left(\frac{P}{D} - 1\right)^2$$

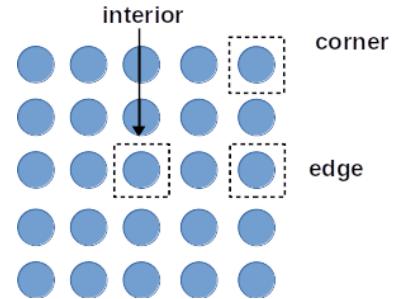


Figure 58: Rod assembly channel positions for a rectangular bundle.

Rod pitch is the distance between the center-line of adjacent rods. The **pitch-to-diameter ratio** (P/D) non-dimensionalizes this distance. The minimum possible value of P/D is 1 when the rods are touching. Typical P/D for a LWR is approximately 1.4.

⁴⁵ Shih-Kuei Cheng and Neil E Todreas. Hydrodynamic models and correlations for bare and wire-wrapped hexagonal rod bundles—bundle friction factors, subchannel friction factors and mixing parameters. *Nuclear engineering and design*, 92(2):227–251, 1986

in which the coefficients depend on whether the flow is laminar or turbulent, whether the rod bundle is rectangular or hexagonal, whether the subchannel is an edge, corner, or interior channel and the pitch-to-diameter ratio. Coefficient values for rectangular channels are provided in Table 9; values for hexagonal channels are provided in Table 10. For laminar flow, $n = 1$. For turbulent flow,

	$1.0 \leq P/D < 1.1$			$1.1 \leq P/D \leq 1.5$		
	a	b_1	b_2	a	b_1	b_2
Laminar Flow						
interior	26.37	374.2	-493.9	35.55	263.7	-190.2
edge	26.18	554.5	-1480	44.40	256.7	-267.6
corner	28.62	715.9	-2807	58.83	160.7	-203.5
Turbulent Flow						
interior	0.09423	0.5806	-1.239	0.1339	0.09059	-0.09926
edge	0.09377	0.8732	-3.341	0.1430	0.04199	-0.04428
corner	0.09755	1.127	-6.304	0.1452	0.02681	-0.03411

Table 9: Coefficients for the Cheng-Todreas correlation for a square array.

	$1.0 \leq P/D < 1.1$			$1.1 \leq P/D \leq 1.5$		
	a	b_1	b_2	a	b_1	b_2
Laminar Flow						
interior	26.00	888.2	-3334	62.97	216.9	-190.2
edge	26.18	554.5	-1480	44.40	256.7	-267.6
corner	28.98	1636	-10050	87.26	38.59	-55.12
Turbulent Flow						
interior	0.09378	1.398	-8.664	0.1458	0.03632	-0.03333
edge	0.09377	0.8732	-3.341	0.1430	0.04199	-0.04428
corner	0.1004	1.625	-11.85	0.1499	0.006706	-0.009567

Table 10: Coefficients for the Cheng-Todreas correlation for a hexagonal array.

$$n = 0.18.$$

In all cases, the Reynolds number is calculated relative to the hydraulic diameter. Calculation of flow area (A_{flow}) and wetted perimeter (P_{wetted}) for rectangular and hexagonal arrays are as shown in Figure 59.

Parting Note

This lecture provides a simplified but reasonably complete presentation of what I judge to be the most important hydraulic correlations for single-phase viscous fluid flow. One important phenomena that we have not discussed at all is the friction factor near the entrance of a circular tube or fuel assembly. This region of “developing flow”

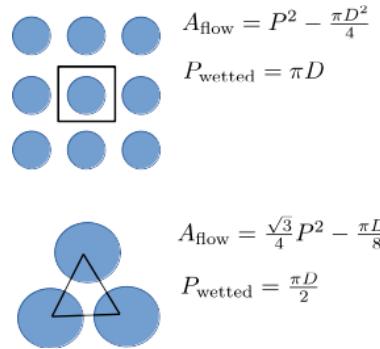


Figure 59: Calculation of flow area and wetted perimeter for flow channel within a rod bundle.

is where the fluid adjusts to the presence of channel walls. As you might expect, there is considerable “action” in the fluid flow field at such a point. For hydraulic analysis this means that, near the entrance of a channel you can expect the *actual* friction factor to be higher than what we have calculated for the fully-developed flow. While we will consider this detail to be outside the scope of this class, state-of-the-art hydraulic analysis codes take this effect into account.

Assignment #6: Applications of Hydraulics

Part I - AP1000 Primary Plant Hydraulics

Use the parameters given below in answering the following questions:

Cold Leg	
Flow rate	$30 \times 10^6 \text{ lb}_m/\text{hr}$ per leg
Length	20 ft
Inside diameter	22 in
Relative roughness	0.0035
Temperature	540°F
Hot Leg	
Flow rate	$60 \times 10^6 \text{ lb}_m/\text{hr}$ per leg
Length	16 ft
Inside diameter	31 in
Relative roughness	0.003
Temperature	610°F
Steam Generator	
Flow rate (primary side)	$60 \times 10^6 \text{ lb}_m/\text{hr}$ per leg
Number of tubes	10,025
Tube inside diameter	0.608 in
Average tube length	884 in
Minor loss coefficients (total)	1.8
Secondary-side Pressure	836
Core	
Nozzle-to-Nozzle pressure drop	62.3 psid
System nominal pressure	2250 psia

For all primary system fluid property evaluations, assume the pressure is equal to the system nominal pressure.

1. Calculate the cold leg friction factor and pressure drop using the Karman-Nikuradse equation for turbulent adiabatic flow in smooth pipes.

2. Calculate the cold leg friction factor and pressure drop using the Colebrook equation for turbulent adiabatic flow in non-smooth pipes.
3. Calculate the hot leg friction factor and pressure drop using the Colebrook equation for turbulent adiabatic flow in non-smooth pipes.
4. Calculate the friction factor and pressure drop for a typical tube in a Steam Generator using the Petukhov correlation for diabatic flow. Assume that the primary coolant temperature at the tube wall is equal to the temperature on the secondary side of the steam generator. Use the average coolant temperature on the primary side of the Steam Generator to establish bulk primary coolant water properties. Verify that the viscosity ratio used in the Petukhov correlation is within prescribed limits.
5. Assuming a pump efficiency of 80%, calculate the rated power [hp] for a single reactor coolant pump.

Part II - Pressurizer Spray Flow

6. The pressure at the cold-leg tap of the pressurizer spray line is approximately 2310 psia. The pressure of the steam volume of the pressurizer is 2235 psia. The spray nozzle of the AP1000 pressurizer is elevated approximately 60 ft above the cold leg tap and the overall pressurizer spray piping length (from the cold leg tap to the spray flow nozzle) is approximately 94 ft. If the design spray flow rate is 700 gal/min and the spray line is of uniform diameter of 4 inches with a relative roughness of 0.0004, estimate the pressure drop due to:
 - (a) Elevation change
 - (b) Major head losses
 - (c) Minor head losses

For all calculations, neglect kinetic energy effects. **Note:** there are approximately 7.48 gallons in a cubic foot.

Part III - Frictional Pressure Drop in the AP1000 core

Consider the AP1000 core configuration. Relevant design information is provided in the table below.

Total core flow rate	$51.4 \times 10^6 \text{ kg/hr}$
Number of fuel assemblies	157
Flow channels per assembly	289
Coolant inlet temperature	281°C
Coolant outlet temperature	321°C
Nominal pressure	15.51 MPa
Average coolant viscosity	$9.061 \times 10^{-5} \text{ Pa-s}$
Fuel pin outside diameter	9.5 mm
Fuel pin pitch	12.6 mm
Core height	4.8 m

7. Calculate the average velocity [m/s] through a typical flow channel in the interior of an assembly.
8. Compute the Reynolds number for a typical flow channel on the interior of a fuel assembly.
9. Calculate the equivalent circular-channel friction factor using the McAdams equation.
10. Calculate the friction factor for a typical interior channel using the Cheng-Todreas correlation.
11. Based on the friction factor calculated for the core, and accounting for potential energy effects (the core is oriented vertically), estimate the pressure drop [psia] between the core inlet and outlet.

Lecture 17 - Extended Surfaces Heat Exchanger

Objectives

The objectives of this lecture are:

1. Discuss motivation for heat exchangers with extended surfaces.
2. Describe a correlation for estimating the friction factor for such a heat exchanger

Extended Surfaces Heat Exchangers

IN PREVIOUS LECTURES we have discussed a range of different heat exchangers for energy conversion applications.

- For PWRs a recirculating u-tube steam generator provides heat transfer from the subcooled liquid water of the primary to a low-quality saturated mixture of the secondary. Heat transfer performance on both the primary and secondary side of the heat exchanger is excellent.
- The condenser of Rankine cycles provides heat transfer between the high-quality saturated mixture of turbine exhaust and sub-cooled liquid of the condenser cooling water. Closed feedwater heaters in Rankine power conversion cycles work with similar fluids.
- A reheater employed in a Rankine cycle is used to transfer heat from a saturated steam to a lower-temperature and pressure saturated steam.
- An inter-cooler or pre-cooler used in a Brayton power cycle rejects heat from a gas working fluid to some cooling medium.
- A regenerator used in a Brayton cycle has gas on both the high- and low-temperature side of the heat exchanger.

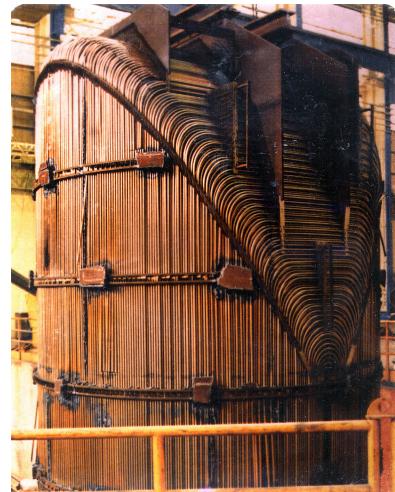


Figure 60: Steam generator internals.



Figure 61: Closed feedwater heater.

What we have not done is investigate in any level of detail the thermal or hydraulic performance of these devices; we simply drew them as boxes on our schematics and assumed that they would function as specified. For some applications, especially those for which at least one side of the heat exchanger is a gas, special measures are called for to enhance convective heat transfer performance. One such measure is the use of “extended surfaces.”

AN EARLY APPLICATION of extended surfaces for enhanced heat transfer was with the MAGNOX reactor.⁴⁶ This reactor comprised natural uranium fuel, graphite moderator and carbon dioxide as the coolant. In order to enhance convective heat transfer to the gas coolant, spiral ribs and fins were incorporated into the clad structure as illustrated in Figure 62.

Heat exchangers with extended surfaces are envisioned for use in Gen IV reactor concepts in which heat transfer occurs between highly conductive fluids—such as liquid lead or sodium—and fluids that conduct heat poorly such as carbon dioxide, air, or helium. The idea is to disrupt the thermal boundary layer on the side where heat transfer would otherwise be poor by using surface features that force complex flow behavior near the heat transfer surface. The benefit is improved convective heat transfer; the price we pay is increased hydraulic pressure losses.

A desirable extended surface design is one where the convective enhancement outweighs the hydraulic penalty. If we wanted to quantify this trade-off, a thermal performance factor,⁴⁷ such as η shown in Equation 20 might be defined.

$$\eta = \frac{\text{Nu}/\text{Nu}_0}{(f/f_0)^{1/3}} \quad (20)$$

Sadly, this sort of an analysis is beyond the scope of this course.

IN THIS LECTURE we will describe hydraulic correlations for two extended surface concepts:

- Helical rib or fin; and
- twisted tape insert

Both correlations provide a way to calculate a Fanning friction factor, f' , from which we can calculate hydraulic pressure losses for flow along a specified length (L) of enhanced pipe of a given diameter (D) using the equation below:

$$\Delta p = 4f' \left(\frac{L}{D} \right) \rho \left(\frac{v^2}{2} \right)$$

⁴⁶ MAGNOX = “magnesium non-oxidizing.”

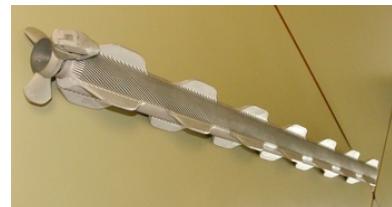


Figure 62: Magnox fuel can.

Nusselt number is a non-dimensional parameter that characterizes the convective heat transfer capability of a fluid relative to thermal conductivity.

$$\text{Nu} = \frac{hL}{k}$$

where h is the convective heat transfer coefficient, k is the thermal conductivity and L is some characteristic length, such as a pipe diameter.

⁴⁷ Chirag Maradiya, Jeetendra Vadher, and Ramesh Agarwal. The heat transfer enhancement techniques and their thermal performance factor. *Beni-Suef University Journal of Basic and Applied Sciences*, 7(1):1–21, 2018

Fanning friction factor, f' is related to the Darcy friction factor, f , by a factor of 4

Important note: when using USCS units, the equation for pressure drop must be modified to include g_c :

$$\Delta p = 4f' \left(\frac{L}{D} \right) \rho \left(\frac{v^2}{2g_c} \right)$$

where ρ is given in lb_m/ft^3 and p is in lb_f/ft^2 and $g_c = 32.2 \text{ lb}_m/\text{lb}_f \text{ ft/s}^2$.

In a homework exercise later in the course we will describe similar correlations for Nusselt number to allow us to estimate changes in thermal performance.

Students are strongly encouraged to work through the units of these equations for Δp to make sure it is clear why g_c is needed in USCS units.

Helical Rib or Fin

This correlation is for circular tubes enhanced with helical ribs or fins. The correlation⁴⁸ we will use estimates the Fanning friction factor relative to the friction factor of a smooth tube as a function of the following parameters:

- Reynolds number (Re).

$$Y1 = 0.67 - 0.06 \frac{p}{D} - 0.49 \frac{\alpha}{90}$$

where p is the rib separation distance, α is the helix angle measured in degrees, and D is the maximum tube inside diameter.

- Rib height (ϵ).

$$Y2 = 1.37 - 0.157 \frac{p}{D}$$

- Rib pitch (p)—distance of rib separation:

$$\left(\frac{\epsilon}{D} \right)^{Y2}$$

$$Y3 = -1.66 \times 10^{-6} Re - 0.33 \frac{\alpha}{90}$$

- Helix angle (α)

$$\left(\frac{\alpha}{90} \right)^{Y4}$$

$$Y4 = 4.59 + 4.11 \times 10^{-6} Re - 0.15 \frac{p}{D}$$

- Contact angle (β) and number (n) of sharp corners

$$\left(1 + \frac{2.94}{n} \right) \sin \beta$$

where β and n are as shown in Figure 63.

⁴⁸ TS Ravigururajan. A comparative study of thermal design correlations for turbulent flow in helical-enhanced tubes. *Heat transfer engineering*, 20(1): 54–70, 1999

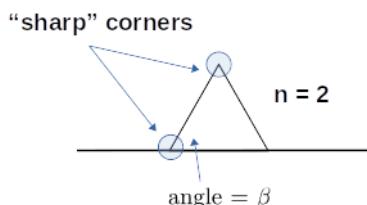


Figure 63: Contact angle and sharp corners for a rib.

The reference smooth-tube friction factor is given as:

$$f'_{sm} = (1.58 \ln (\text{Re}) - 3.28)^{-2}$$

and the complete correlation is formed in Equation 21.

$$\frac{f'_a}{f'_{sm}} = \left\{ 1 + \left[29.1 \text{Re}^{Y1} \left(\frac{e}{D} \right)^{Y2} \left(\frac{p}{D} \right)^{Y3} \left(\frac{\alpha}{90} \right)^{Y4} \left(1 + \frac{2.94}{n} \right) \sin \beta \right]^{15/16} \right\}^{16/15} \quad (21)$$

Quoting from the course textbook, "...this correlation predicts 96% of the database to within 50% and 77% [of the database] to within 20%." from which you should understand that high precision should not be expected.

Twisted Tape Insert

TWISTED TAPE INSERTS are a widely used class of swirl-flow inducing devices. Their advantages over helical ribs include:

- relative ease of manufacture and installation;
- allows for retrofitting by swapping out inserts

The course textbook provides multiple correlations for laminar and turbulent flow. In these notes I will briefly present only the turbulent flow correlation.⁴⁹

This friction factor is presented as a function of:

- Reynolds number
- Tape thickness (δ)
- Tape half-pitch (H)
- "Twist Ratio" (y) which is just the half-pitch divided by the pipe diameter:

$$y = \frac{H}{D}$$

The friction factor correlation is given by Equation 22.

$$f' = \frac{0.0791}{\text{Re}^{0.25}} \left[1 + \frac{2.752}{y^{1.29}} \right] \left[\frac{\pi}{\pi - (4\delta/D)} \right]^{1.75} \left[\frac{\pi + 2 - (2\delta/D)}{\pi - (4\delta/D)} \right]^{1.25} \quad (22)$$

where f' is again the Fanning friction factor for the following range of conditions:

$$\text{Re} \geq 10^4, \quad 2 \leq y \leq \infty, \quad 0.03 \leq \delta/D \leq 0.83$$

⁴⁹ Raj M Manglik and Arthur E Bergles. Heat transfer and pressure drop correlations for twisted-tape inserts in isothermal tubes: Part ii—transition and turbulent flows. *J. Heat Transfer*, 115: 890–896, 1993

tape half-pitch is the axial length for the tape to take a 180 degree twist.

Summary

In this lecture we discussed two correlations for hydraulic pressure drop for tubes with convective heat transfer enhancements. We did not discuss the actual convective enhancements, we only discussed the correlation to predict the attendant hydraulic pressure losses. The corresponding correlations for convective heat transfer will be treated in a set of homework problems.

Lecture 18 - Pressure Losses at Fuel Assembly Grids

Objectives

The objectives of this lecture are:

1. Discuss the role of spacer grids in a PWR core
2. Describe three correlations that may be used to estimate hydraulic losses at spacer grids.

Spacer Grids and Mixing Grids

MANY NUCLEAR POWER reactor cores consist of a cylindrical arrangement of long, thin, fuel pins. These pins are about the diameter of your pinky finger; may be more than 4 meters in length and, in the case of a PWR, may have liquid water coolant flowing by at 10 - 15 m/s. The fuel pins themselves need to be structurally supported in the core. This is done by placing the pins into square or hexagonal grid-like fuel assemblies; a bottom and top nozzle provide support on their respective ends and grid assemblies are placed between. The functional requirements of the grid assemblies are:

1. Structurally support the fuel pins. The fuel pins must be restrained against large amplitude vibrations due to the turbulent flow of coolant along the length of the pins. Vibration, over the long term, results in fuel clad degradation and potential failure.
2. Enhance convective heat transfer using mixing vanes. Some of the grid assemblies have vanes that are specifically designed to disrupt the flow field to promote enhanced convective heat transfer along a rod as well as to promote mixing with coolant flowing along adjacent rods.

Several of these grids are commonly used both for support and flow mixing. A schematic of a fuel assembly for the AP1000 is shown in Figure 64 with a top view in Figure 65.

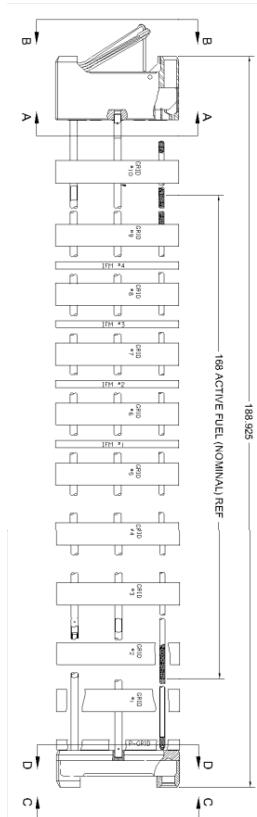


Figure 64: Schematic of AP1000 fuel assembly with 10 spacer grids and 4 mixing grids.

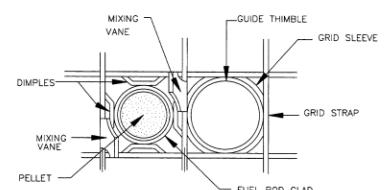


Figure 65: Top view of AP1000 grid.

BOTH THE GRIDS USED for structural support and those for enhanced flow mixing will result in hydraulic pressure losses that should be taken into account for the engineering analysis of the core.⁵⁰ The correlations we will use are all designed to provide a reasonable value for a *minor loss coefficient* (k) to capture the hydraulic losses for each spacer or mixing grid. Generically, pressure drop due to these minor losses are calculated as:

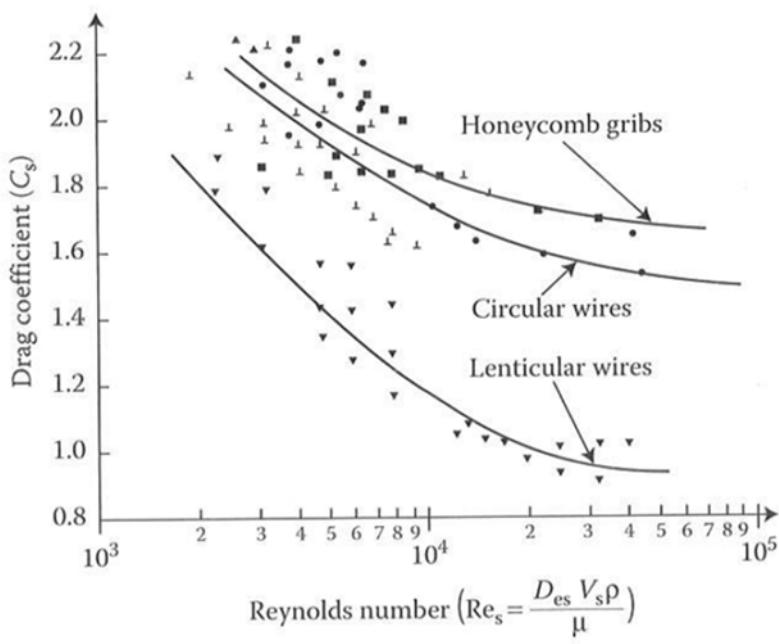
$$\Delta P_{\text{grid}} = k \rho \frac{v^2}{2}$$

or, if USCS units are used:

$$\Delta P_{\text{grid}} = k \rho \frac{v^2}{2 g_c}$$

De Stordeur Correlation

De Stordeur developed a drag coefficient correlation for a variety of wire spacers and spacer grids.⁵¹ The results are shown in Figure 66. The pressure drop is calculated from the drag coefficient C_s as shown in Equation 23



⁵⁰ Especially for light water reactors, local pressure conditions in the core can have a big impact on thermal performance.

⁵¹ A N de Stordeur. Drag coefficients for fuel-element spacers. *Nucleonics (US)* Ceased publication, 19(6), 1961

Figure 66: Drag coefficient results for different spacer wires or mixing grids per de Stordeur.

Note: Take a moment to reflect on the data on which this correlation is based. Notice the scatter and how the curves provided only approximately capture the general data trends. Notice how there are more data points at some Re_s regions and fewer in others. You should keep these observations in mind when you are presenting results of your hydraulic analysis and recognize that precision is limited.

Re_s is the Reynolds number in the region of the spacer grid. Note that the characteristic length used, D_{es} , is the **grid strap thickness** t .

$$\Delta p = C_s \left(\frac{\rho v_s^2}{2} \right) \left(\frac{A_s}{A_v} \right) \quad (23)$$

where ρ is the fluid density, v_s is the velocity of the coolant *in the spacer region*, A_v is the unrestricted flow area away from the grid or spacer, and A_s is the projected frontal area of the spacer. Figure 67 gives a schematic top-view of a grid spacer. The projected area of the spacer (A_s) can be calculated as:

$$A_s = 2Pt - t^2$$

where P is the rod pitch and t is the grid spacer strap thickness. The flow area in the channel (A_v) is given by:

$$A_v = P^2 - \frac{\pi}{4}D^2$$

were D is the fuel rod diameter. Note that v_s is somewhat **higher** than the velocity in the channel away from the grid spacer due to the flow constriction at the location of the spacer. For reactors cooled with incompressible fluids we can estimate v_s as follows:

$$v_s = v \left(\frac{A_v}{A_v - A_s} \right)$$

where v is the fluid velocity in the unobstructed channel.

ONE BOTHERSOME DETAIL is that, for a hydraulic analysis, it would be nice if we could combine the minor losses due to grid spacers with other minor loss coefficients such as core entrance or exit effects. We want a minor loss coefficient k to use. Such an expression is given in Equation 24.

$$K_{\text{grid}}^{\text{deStordeur}} = \left(C_s \frac{A_s}{A_v} \right) \left(\frac{A_v}{A_v - A_s} \right)^2 \quad (24)$$

Once we calculate A_v , A_s , and v_s , we determine Re_s and read a value of C_s off the graph, we compute $K_{\text{grid}}^{\text{deStordeur}}$ and find its contribution to pressure drop from $k\rho\frac{v^2}{2}$ like any other component of minor head loss.

Rehme Correlation

A different research group led by Rehme determined, after testing several grid spacers, that the ratio A_s/A_v had a more pronounced effect on the pressure drop than indicated by de Stordeur. They developed a different correlation⁵² of the form given in Equation 25.

$$\Delta p_{\text{spacer}} = C_v \frac{\rho v^2}{2} \left(\frac{A_s}{A_v} \right)^2 \quad (25)$$

Graphs to determine C_v for triangular and rectangular fuel array spacers are provided in Figure 68 and Figure 69 respectively.

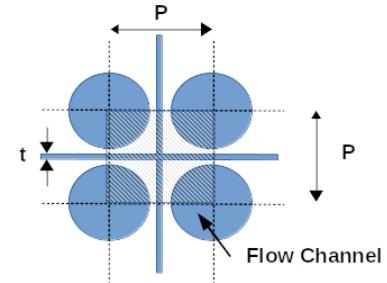


Figure 67: Schematic of flow channel at grid spacer.

⁵² Klaus Rehme. Pressure drop correlations for fuel element spacers. *Nuclear technology*, 17(1):15–23, 1973

Note: Re_B for this correlation is the Reynolds number in bulk coolant channels. The characteristic length is the rod diameter D .

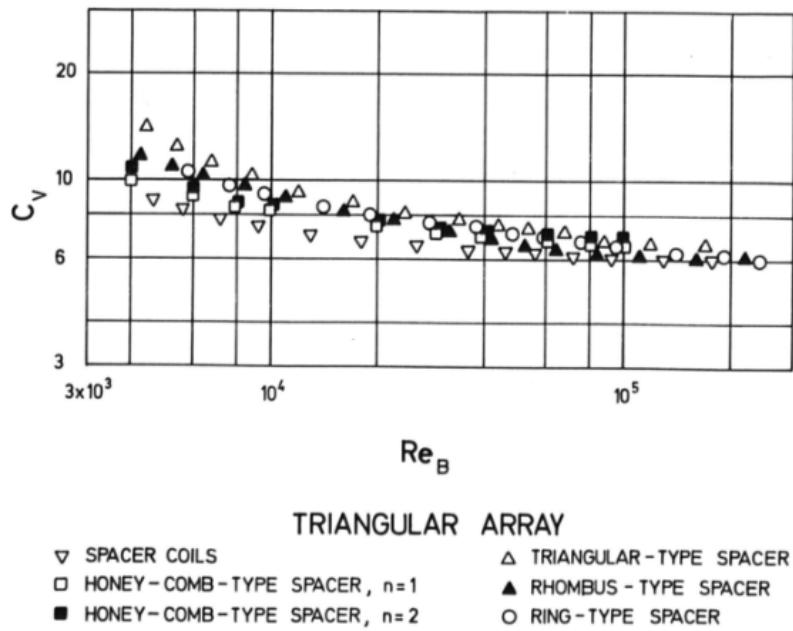


Figure 68: Graph to find C_v for triangular grids.

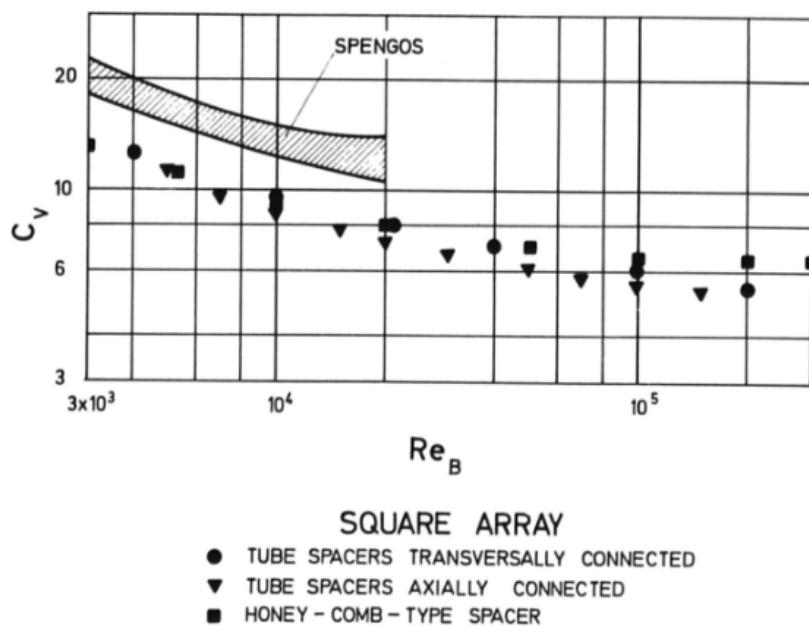


Figure 69: Graph to find C_v for rectangular grids.

The area labeled “Spengos” in Figure 69 corresponds to the data range in de Stordeur’s study. It should be clear that: a) Rehme created and used data from a wider range of configurations; and b) Rehme will predict a smaller drag coefficient than de Stordeur and thus be expected to predict a lower Δp .

In Correlation for PWR Grid Spacers

A more recently published correlation⁵³ for minor head losses due to grid spacers is due to In and his group. The equations that comprise the formulation are relatively complex but one compensating benefit is that there is no requirement to read a graph to obtain values for any coefficients. This makes use of this correlation particularly amenable to incorporation into your own hydraulic analysis code.⁵⁴ The output is a minor loss coefficient, $K_{\text{grid}}^{\text{In}}$, that you use in a standard minor head loss calculation:

$$\Delta p_{\text{grid}}^{\text{In}} = K_{\text{grid}}^{\text{In}} \frac{\rho v^2}{2}$$

The formula for $K_{\text{grid}}^{\text{In}}$ is given in Equation 26.

$$K_{\text{grid}}^{\text{In}} = \underbrace{\left[C_{\text{grid}}^{\text{form}} \frac{\epsilon}{(1-\epsilon)^2} \right]}_{\text{Term A}} + \underbrace{\left[C_{\text{grid}}^{\text{fric}} \frac{A_{\text{grid,wetted}}}{A_f} \frac{1}{(1-\epsilon)^2} \right]}_{\text{Term B}} + \underbrace{\left[C_{\text{rod}}^{\text{fric}} \frac{A_{\text{rods,wetted @ grid}}}{A_f} \frac{1}{(1-\epsilon)^2} \right]}_{\text{Term C}} + \underbrace{\left[C_{\text{d,mv}} \frac{\epsilon_{\text{mv}}}{(1-\epsilon_{\text{mv}})^2} \right]}_{\text{Term D}} \quad (26)$$

We will discuss each term individually.

Term A

This term tries to capture the effect of form losses for the grid geometry. The coefficient is given by:

$$C_{\text{grid}}^{\text{form}} = 2.75 - 0.27 \log_{10} (\text{Re})$$

Where Re is the Reynolds number evaluated in the flow channel away from the grid. The parameter ϵ , which is also used in Terms B and C, is really the same as the ratio A_s/A_v used in the de Stordeur and Rehme correlations. It captures the fraction of flow area occluded by the mixing grid. The nomenclature we will use here is adapted to match the textbook.

$$\epsilon = \frac{A_{\text{grid,frontal}}}{A_f} = \frac{A_s}{A_v}$$

⁵³ Wang Kee In, Dong Seok Oh, and Tae Hyun Chun. Empirical and computational pressure drop correlations for pressurized water reactor fuel spacer grids. *Nuclear technology*, 139(1):72–79, 2002

⁵⁴ For those so inclined to make such a tool.

where A_f is just the flow area in the channel away from the grid and the second equality is meant to make the comparison in nomenclature explicit.

Term B

This term tries to capture the pressure losses due to fluid friction with the surface area of the grid strap of height H . For this term we will define the term *mass flux* (G).

If the strap height, $H \geq 3 \times 10^4 \mu_{\text{avg}} / G_{\text{@ grid}}$ then: The coefficient $C_{\text{grid}}^{\text{fric}}$ is given by:

$$C_{\text{grid}}^{\text{fric}} = C_{\text{grid,laminar}}^{\text{fric}} \frac{3 \times 10^4 \mu_{\text{avg}}}{G_{\text{@ grid}} H} + C_{\text{grid,turbulent}}^{\text{fric}} \frac{[H - (3 \times 10^4 \mu_{\text{avg}} / G_{\text{@ grid}})]}{H}$$

Otherwise if $H < 3 \times 10^4 \mu_{\text{avg}} / G_{\text{@ grid}}$ then:

$$C_{\text{grid}}^{\text{fric}} = C_{\text{grid,laminar}}^{\text{fric}} \frac{3 \times 10^4 \mu_{\text{avg}}}{G_{\text{@ grid}} H}$$

where $C_{\text{grid,laminar}}^{\text{fric}}$ is given by:

$$C_{\text{grid,laminar}}^{\text{fric}} = 1.328 \left\{ G_{\text{@ grid}} \frac{[H - (3 \times 10^4 \mu_{\text{avg}} / G_{\text{@ grid}})]}{\mu_{\text{avg}}} \right\}^{-0.5}$$

and $C_{\text{grid,turbulent}}^{\text{fric}}$ is given by:

$$C_{\text{grid,turbulent}}^{\text{fric}} = 0.523 \left\{ \ln \left[0.06 \times G_{\text{@ grid}} \frac{[H - (3 \times 10^4 \mu_{\text{avg}} / G_{\text{@ grid}})]}{\mu_{\text{avg}}} \right] \right\}^{-2}$$

This term also needs $A_{\text{grid,wetted}}$ which, by inspection of Figure 70 is given by: $A_{\text{grid,wetted}} = 4(P - t)H$.

Term C

This term attempts to capture the pressure losses due to fluid friction with the surface area of the fuel rod in the vicinity of the grid spacer. The formula for $C_{\text{rod}}^{\text{fric}}$ is relatively simple, based on McAdams relation for viscous flow through smooth circular tubes.

$$C_{\text{rod}}^{\text{fric}} = 0.184 \text{Re}_{\text{@ grid}}^{-0.2}$$

where $\text{Re}_{\text{@ grid}}$ is the Reynolds number of flow at the grid spacer;

$\text{Re}_{\text{@ grid}} = \frac{\rho v_{\text{@ grid}} D_{\text{e,@ grid}}}{\mu}$. Calculation of the hydraulic diameter

Mass flux (G) is simply the mass flow rate divided by the cross sectional area: $G = \dot{m}/A$.

Note: You should take a few minutes, review the units of dynamic viscosity, (μ), write down the units of mass flux, (G), and prove to yourself that the ratio of viscosity and mass flux in $3 \times 10^4 \mu_{\text{avg}} / G_{\text{@ grid}}$ has units of length.

Note: The term $G_{\text{@ grid}}$ should be read "mass flux at the grid spacer." This is different than the mass flux in the rest of the channel around the fuel rod because the grid spacer is obstructing part of the flow area.

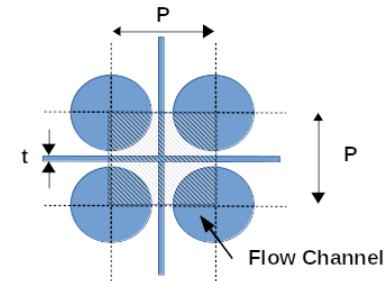


Figure 70: Schematic flow channel in vicinity of spacer grid.

requires, in addition to the flow area at the grid spacer, which you already need for this correlation, but also the wetted perimeter.

$$D_{e,\text{@ grid}} = \frac{4A_{\text{flow @ grid}}}{P_{w,\text{@ grid}}} = \frac{4(A_f - A_{\text{grid,frontal}})}{P_{w,\text{@ grid}}}$$

Recalling a simplified schematic of a flow channel in the vicinity of a spacer grid shown in Figure 70, you should be able to show that the wetted perimeter is: $P_{w,\text{@ grid}} = \pi D + 4(P - t)$.

This term also requires $A_{\text{rod,wetted}}$ which is: $A_{\text{rod,wetted}} = \pi DH$

Term D

This term attempts to capture the hydraulic effect of any mixing vanes incorporated into the grid. The coefficient $C_{d,mv}$ is empirically based and expected (by In) to be in the range of 0.6 to 0.8. They recommend in their paper to use the value: $C_{d,mv} = 0.72$. The parameter ϵ_{mv} is the ratio of the total “plugging area” of the mixing vanes to the bundle flow area away from the grid; it is analogous to the ϵ used to characterize the extent to which the support grid obstructs the flow channel although there is no reason they should be expected to be the same. Figure 71 illustrates a spacer grid with mixing vane on the top side.⁵⁵ Note that the mixing vanes necessarily “obstruct” the channel more than the spacer grid itself; this is to allow for the vanes to induce vigorous mixing as is desired for improved convective heat transfer performance.

Summary

Despite the complexity of the In correlation, if you want to incorporate an engineering estimate of minor head losses due to spacers and mixing grids, its use is recommended. The equations can be (carefully!) encoded using the computing tools of your choice and used without manual reading of graphs. If a more rigorous analysis is needed, one should commit to detailed computational fluid dynamics and/or experimental studies.

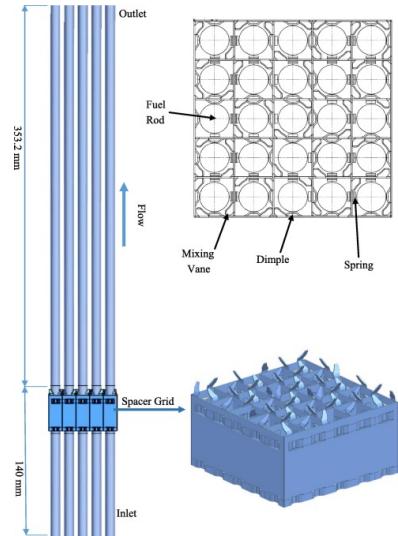


Figure 71: Grid spacer with mixing vanes.

⁵⁵ Deqi Chen, Yi Xiao, Shijie Xie, Dewen Yuan, Xuemei Lang, Zile Yang, Yunke Zhong, and Qi Lu. Thermal-hydraulic performance of a 5×5 rod bundle with spacer grid in a nuclear reactor. *Applied Thermal Engineering*, 103:1416–1426, 2016

Lecture 19 - Natural Circulation

Objectives

The objectives of this lecture are:

1. Describe natural circulation
2. Demonstrate an analysis of natural circulation flow for the Nuc-Scale SMR
3. Discuss the advantages and disadvantages of natural circulation for nuclear power reactors.

Motivation

The possibility that, at any given time for any of a number of reasons, pumps providing cooling flow to the core may lose power is a major contributor to the overall probability of core damage for nuclear reactors. If electrical power is lost to cooling pumps for a PWR, for instance, an operator may take action to restore flow; or they may not. Wouldn't it be a lot better and more safe if cooling flow circulation would take place *without any pumps at all* and there would be no need for operators to take actions in the event of loss of power; indeed, there may not be a need for any *operators* or *electrical power* at all! This is the motivation for natural circulation for primary coolant flow.

Natural Circulation

In principle, only three conditions are required to achieve natural circulation:

1. A heat sink and a heat source
2. The heat source must be “below” the heat sink with respect to some gravitational reference; and

3. flow-paths must exist for the fluid to flow and carry energy to and from the heat source and heat sink.

A schematic representation is provided in Figure 72. The hydraulic pressure, providing the driving force for natural circulation is given by:

$$\text{Driving Force} = (\rho_c - \rho_h) \frac{g}{g_c} h$$

where ρ_c and ρ_h are the density of the cold and hot fluids respectively; g is the acceleration due to gravity; g_c is that annoying constant we have to include from time-to-time when using USCS units; and h is the height of the heat sink relative to the heat source.

THE DRIVING FORCE will establish natural circulation flow. The hydraulic pressure drop from major and minor head losses constitute the resistance to flow:

$$\text{Flow Resistance} = \left(\frac{fL}{D} + \Sigma K \right) \rho \frac{g}{g_c} \frac{v^2}{2g}$$

If velocity (v) is zero, there is no resistance and flow will commence. As velocity goes up, flow resistance increases. Velocity will continue to increase until the flow resistance is in equilibrium to the driving force. Since the driving force is created by the density, and thus temperature difference between the hot and cold columns of water, the equilibrium velocity is a function of the thermal power output of the heat source.

NuScale Natural Circulation Demonstration

A current reactor concept design that depends upon natural circulation for primary coolant flow is the NuScale reactor. The NuScale concept includes a collection of power modules contained within an underground pool-type containment building as in Figure 73. Each individual power module is a pressurized water reactor of *integral* design. Primary coolant flow within each power module is driven by natural circulation; a simplified schematic to illustrate the flow-paths is shown in Figure 74.

To enable a quantitative analysis, we will estimate some of the dimensions:

- height of “downcomer” - 46 ft
- height of the “lower riser” - 9.4 ft
- height of the core - 7.9 ft

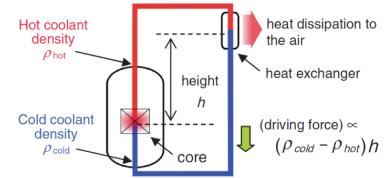


Figure 72: Schematic representation of a minimal natural circulation arrangement

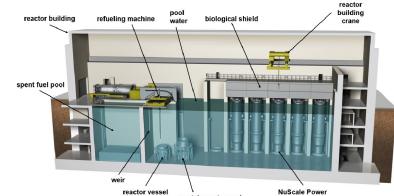


Figure 73: NuScale reactor containment building.

Integral pressurized water reactors combine the reactor core, steam generator, and pressurizer within a single pressure vessel.

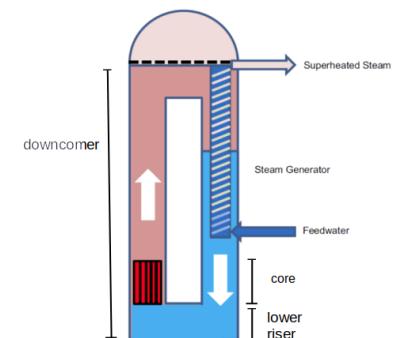


Figure 74: Schematic of natural circulation in a NuScale power module.

We combine these to estimate the height of the thermal driving head (H):

$$H = L_{\text{downcomer}} - L_{\text{lower riser}} - L_{\text{core}}$$

For the thermal model of the core, we will make the following simplifications:

- The core ΔT , which is $T_H - T_C$ is dictated by core power and coolant mass flow rate.
- We will assume average coolant temperature remains constant

The thermal driving head will thus be computed as:

$$\Delta P_{\text{thermal}} = (\rho_C - \rho_H) \frac{g}{g_c} H$$

Rather than explicitly creating a hydraulic model of the core and natural circulation flow path, we will assume that the coefficients comprising major and minor head losses, $\frac{fL}{D} + \sum K$, are constant (C).

Therefore the hydraulic flow resistance will be given by:

$$\Delta P_{\text{hydraulic}} = C \times \rho \frac{v^2}{2g_c}$$

where ρ is the nominal density at the average coolant temperature.

The mass flow rate from natural circulation will thus be:

$$\dot{m} = \rho v A_{\text{flow}}$$

where A_{flow} is the total core flow area. A complete MATLAB code for this analysis is provided in the appendices. The resulting mass flow rate is shown in Figure 75.

Advantages and Disadvantages of Natural Circulation

Some advantages of natural circulation include:

1. Natural circulation eliminates dependence on pumps for coolant circulation. Pumps are problematic in that they:
 - (a) require electrical power (that may fail);
 - (b) are subject to mechanical failure; and
 - (c) require space along with mechanical and electrical connections.

This is particularly a disadvantage for integral reactor designs
2. does not require human action to maintain flow; and
3. for some applications—such as propulsion for a military submarine—natural circulation is generally more quiet than forced circulation with a centrifugal pump

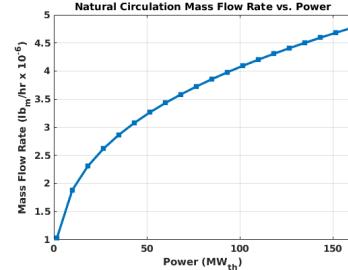


Figure 75: Natural Circulation mass flow rate as a function of reactor power for a simplified NuScale power module.

Some disadvantages of natural circulation include:

1. limited control over flow rate. The thermal/hydraulic design is constrained and dependence on flow from natural circulation may limit the thermal power density that can safely be achieved.
2. the thermal driving head for natural circulation depends upon coolant density differences which, in turn, is a function of coolant temperature difference across the core. A need to increase thermal driving head may result in unacceptably high differential temperatures and thermal stress for mechanical components in the core.
3. the procedures for establishing natural circulation flow may involve complexities and there may be no simple way to rapidly change coolant flow rate should the need arise.

The potential for safety and reliability that can come from natural circulation is the main source of interest.

Summary

Students are strongly encouraged to carefully work through the MATLAB demonstration code provided in the appendix. Analyze the parameters and make changes to the model to be sure you understand what is going on. One thing you may notice is the importance of the *distance* between the hot- and cold-temperature reservoirs. The greater that distance, the greater thermal driving head will be for a given temperature differential. This is one reason why the NuScale modules are relatively “tall and skinny” compared to a typical PWR pressure vessel—beyond the need to simply make space for the steam generator and pressurizer within the integral design.

The Economically Simplified Boiling Water Reactor (ESBWR) also relies on natural circulation in lieu of recirculation pumps for reactivity control.⁵⁶ The pressure vessel for the ESBWR is correspondingly much taller than that of a typical BWR, in part, to provide greater driving head for natural coolant circulation.

⁵⁶ David Hinds and Chris Maslak. Next-generation nuclear energy: The esbwr. *Nuclear News*, 49(1):35–40, 2006

Part III

Thermal Analysis

Lecture 20 - Thermal Analysis Introduction/Review

Objectives

The objectives of this lecture are:

1. Briefly review some relevant thermal analysis topics from previous coursework
2. Outline the elaborations to convective and conductive heat transfer topics appropriate for nuclear applications.

Thermal Analysis Review

In previous courses you learned to solve the heat equation to determine temperature within a specified domain using a form of the heat equation as in Equation 27.

$$\rho c_p \frac{\partial T}{\partial t} = \nabla \cdot k \nabla T + S(\bar{x}, t) \quad (27)$$

where ρ is the material density, c_p is the specific heat, T is temperature, k is the thermal conductivity, and $S(\bar{x}, t)$ is a heat source that in general depends on position and time.

THE HEAT EQUATION is derived based on conservation of energy and represents physical behavior for simple heat conduction problems. We found solutions assuming uniform and constant thermal diffusivity. For the context of conduction in a nuclear fuel pin the temperature profile was found to be:⁵⁷

$$T(r) = T(r_o) + \frac{q'''}{4k} \left(r_o^2 - r^2 \right) = T(r_o) + \frac{q'}{4\pi k_{\text{fuel}}} \left(1 - \frac{r^2}{r_o^2} \right) \quad 0 < r < r_o$$

where r_o is the fuel pellet outer radius, q''' is the heat production rate from fission, and q' is the linear heat generation rate. This can be rearranged to solve for the temperature change between the fuel pellet center line and outer surface:

$$T(0) - T(r_o) = \Delta T_{\text{fuel}} = \frac{q'}{4\pi k_{\text{fuel}}}$$

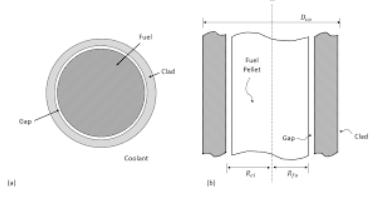


Figure 76: Schematic of fuel pin geometry.

Thermal diffusivity for a material is defined as $\alpha = \frac{k}{\rho c_p}$. By assuming constant thermal diffusivity we assumed that k , ρ and c_p were constant.

⁵⁷ For this solution, in addition to constant thermal diffusivity, the heat source represented by q''' was assumed to be constant. The equation was cast in polar coordinates with symmetry with respect to angular position and the temperature on the outside surface of the fuel pellet, $T(r_o)$, must be provided as a boundary condition.

Note: recall that by definition, $q''' \pi r^2 L = q' L$.

HEAT CONDUCTION FROM the fuel to the cladding across the gas-filled gap is modeled as a convective process, where the heat flux (q'') is proportional to the difference in temperature between the fuel outer radius, $T(r_o)$, and the clad inner radius, $T(r_{ci})$:

$$q'' = h_{\text{gap}} [T(r_o) - T(r_{ci})] = h_{\text{gap}} \Delta T_{\text{gap}}$$

where h_{gap} is called the “gap conductance” and represents the convective heat transfer across the pellet-to-clad gap. Re-arranging the above equation to solve for ΔT_{gap} we get:

$$\Delta T_{\text{gap}} = \frac{q''}{h_{\text{gap}}} = \frac{q'}{2\pi r_o h_{\text{gap}}}$$

where we take r_o as the reference radius for the heat transfer process.⁵⁸

HEAT CONDUCTION THROUGH the cladding is modeled using, once again, Equation 27. For the cladding analysis we assumed constant thermal diffusivity and no heat generation in the cladding.⁵⁹

$$T(r) = T(r_{co}) + \frac{q'}{2\pi k_{\text{clad}}} \ln \left(\frac{r_{co}}{r} \right), \quad r_{ci} < r < r_{co}$$

We can, once again, re-arrange this equation to get an expression for ΔT_{clad} :

$$T(r_{ci}) - T(r_{co}) = \Delta T_{\text{clad}} = \frac{q'}{2\pi k_{\text{clad}}} \ln \left(\frac{r_{co}}{r_{ci}} \right)$$

CONVECTION TO THE COOLANT on the outer surface of the fuel pin was handled in a similar fashion as convection across the fuel-clad gap:

$$\Delta T_{\text{cool}} = T(r_{co}) - T_{\text{coolant}} = \frac{q'}{2\pi r_{co} h_{\text{cool}}}$$

where h_{cool} is the convective heat transfer coefficient for the coolant flowing on the outer surface of the rod. In previous classes, except possibly your heat transfer course, the value of h_{cool} was taken to be a given parameter.

WE CAN COMBINE the ΔT 's computed above to get an overall temperature change between the center line of a fuel pellet and the surrounding bulk coolant:

$$\begin{aligned} \Delta T_{\text{total}} &= \Delta T_{\text{fuel}} + \Delta T_{\text{gap}} + \Delta T_{\text{clad}} + \Delta T_{\text{cool}} \\ &= \frac{q'}{2\pi} \left[\frac{1}{r_{co} h_{\text{cool}}} + \frac{1}{k_{\text{clad}}} \ln \left(\frac{r_{co}}{r_{ci}} \right) + \frac{1}{r_o h_{\text{gap}}} + \frac{1}{2k_{\text{fuel}}} \right] \end{aligned}$$

Note: by definition the following relation also holds: $q'' 2\pi r L = q' L$.

⁵⁸ This is a small approximation since, for relevant fuel pins on which this analysis is applied, r_o is very close to r_{ci} .

⁵⁹ This is not quite true: gamma radiation from the fuel deposits a small fraction of its energy in the cladding materials. We ignored that effect in previous courses and we will ignore it in this course also, I am sad to say. Current computer models used for fuel performance analysis do take non-local gamma energy deposition into account.

Preview for Thermal Section of ER468

In this course we will elaborate on those previously covered topics. These “elaborations” will capture the following additional physical phenomena:

- Fuel thermal conductivity is not constant. In fact, fuel thermal conductivity is a function of, among other things:
 1. **Temperature.** As $T_{\text{fuel}} \uparrow$, $k(T) \downarrow$. If we re-write the heat equation, making this temperature dependence explicit:

$$\rho c_p \frac{\partial T}{\partial t} = \nabla \cdot k(T) \nabla T + S(\bar{x}, t)$$
 it becomes apparent that, accounting for this fact, makes the heat equation non-linear.⁶⁰
 2. **Burn-up.** Burn-up changes the material composition of the fuel through the fission process. As burn-up proceeds, radiation and long-term exposure to high temperature changes the structural make-up of the fuel.
 3. **Enrichment.** This is a small effect, but modern models take it into account.
 4. **Presence of gadolinium.** Adding gadolinium for reactivity control generally results in somewhat reduced thermal conductivity.
 5. **Presence of Plutonium.** Plutonium oxide has many different properties than uranium oxide, including thermal conductivity.
- Gap conductance is not constant. Not to spoil the upcoming lecture, but gap conductance is primarily a function of the gap width. Gap width changes based on pellet and clad sizing, of course, but also due to pellet dimensional changes during operation and over core life. As linear power (q') increases, the pellet-to-clad gap is reduced and h_{gap} increases; as the pellet spends time in the core, a variety of changes occur to the oxide structure that result in changes to the gap size.
- Thermal conductivity of the cladding is not constant. Like the fuel material, clad material is generally a function of temperature. Since the clad material does not undergo fission, the thermal conductivity behavior is less complex than for fuel. Still, since fuel cladding serves as a containment boundary for the fuel material, it is important to accurately model clad behavior and that means temperature-dependent conductivity should be included.

⁶⁰ **Question:** Why does temperature dependent thermal conductivity make the heat equation non-linear?

Note: For an authoritative description of the chemical, mechanical, and thermal properties of Plutonium, I cannot help but point you to the Plutonium Handbook (7-volume set) available from the American Nuclear Society for \$2400.

- Convective heat transfer for fuel assemblies is not a constant.

Modeling convective heat transfer in nuclear reactor fuel assemblies, if all relevant phenomena are taken into account, is very complex. Fuel assemblies with mixing grids presents some geometric complexity; for light water reactors, the multi-phase flow commonly prevailing in the hottest fuel rod channels is practically impossible to model from first principles. We will discuss important and relevant correlations available to make use of experimental data that is useful for thermal/hydraulic design.

Lecture 21 - Fuel Thermal Conductivity Model

Objectives

The objectives of this lecture are:

1. Analyze the heat equation taking into account temperature-dependent thermal conductivity.
2. Present a current fuel thermal conductivity model.
3. Carry out an example calculation using a graphical technique

The Heat Equation, Revisited

This being a lecture about thermal conductivity models, we should first start with the heat equation.

$$\rho c_p \frac{\partial T}{\partial t} = S + \nabla \cdot k \nabla T \quad (28)$$

where, as a reminder, ρ is the mass density, c_p is the material specific heat, T is temperature, t is time, S is a volumetric heat source term, and k is the material thermal conductivity.

This model will be applied to cylindrical fuel pins that are typical for a nuclear reactor core; accordingly we will specify the cylindrical coordinate system for the ∇ operator. If we further assume steady state, Equation 28 becomes:

$$\begin{aligned} \rho c_p \frac{\partial T}{\partial t} &= q''' + \frac{1}{r} \frac{\partial}{\partial r} \left(kr \frac{\partial T}{\partial r} \right) + \frac{1}{r^2} \frac{\partial}{\partial \phi} \left(k \frac{\partial T}{\partial \phi} \right) + \frac{\partial}{\partial z} \left(k \frac{\partial T}{\partial z} \right) \\ 0 &= q''' + \frac{1}{r} \frac{\partial}{\partial r} \left(kr \frac{\partial T}{\partial r} \right) + \frac{1}{r^2} \frac{\partial}{\partial \phi} \left(k \frac{\partial T}{\partial \phi} \right) + \frac{\partial}{\partial z} \left(k \frac{\partial T}{\partial z} \right) \\ 0 &= q'''r + \frac{\partial}{\partial r} \left(kr \frac{\partial T}{\partial r} \right) \end{aligned} \quad (29)$$

where derivatives in ϕ are assumed zero due to radial symmetry and the derivative of temperature with respect to z is assumed to be

zero since most of the heat is conducted in the radial direction. The equation is now an second-order *ordinary* differential equation⁶¹ with a non-homogeneous source term that we will, for the purposes of this lecture, take to be a constant.

If we integrate both sides of Equation 29 we get:

$$\begin{aligned} 0 &= q''' \frac{r^2}{2} + kr \frac{dT}{dr} + c_1 \\ 0 &= q''' \frac{r}{2} + k \frac{dT}{dr} + \frac{c_1}{r} \end{aligned} \quad (30)$$

where c_1 is a constant that must be found from the boundary conditions.

At this point in the analysis we will introduce an additional complexity that we have been able to safely ignore up until now; namely we will assume that k is a function of material temperature, $k(T)$. Inserting this new notation into our equation and re-arranging gives us:

$$-k(T) \frac{dT}{dr} = \frac{q'''}{2} r + \frac{c_1}{r} \quad (31)$$

This equation is now non-linear but, thankfully is separable; we can formally move ∂r from the “denominator” on the right hand side to get:

$$-\int_{T(r=0)}^{T(r=R)} k(T) dT = \int_{r=0}^{r=R} \frac{q'''}{2} r + \frac{c_1}{r} dr = \frac{q''' R^2}{4} + c_1 \ln(r) \Big|_0^R \quad (32)$$

where, in order for the right hand side of Equation 32 to remain bounded as $r \rightarrow 0$, c_1 must be set equal to zero. We've kept the left hand side as it is, for the time being, until we can specify the form of $k(T)$.

The last manipulations we will do are cosmetic in nature: on the left hand side we will flip the interval of integration to remove the minus sign; on the right hand side we observe that $q''' \pi R^2 L = q' L$ to give us:

$$\int_{T(R)}^{T(0)} k(T) dT = \frac{q'}{4\pi} \quad (33)$$

where it is expected that $T(R)$ would be given as a boundary condition.

It would be a stretch to call this a “solution” to Equation 28; we have not derived an expression for temperature as a function of radius. What we have is a way to determine the temperature at the center-line of the fuel—the maximum fuel temperature T_{\max} —given a constant linear heat rate, q' , and a formula for $k(T)$. What we hope to do is find the value of $T(0)$ so that the integral on the left of Equation

Note: This last assumption is quite true near the axial center of the fuel pin, far from the top or bottom core boundary. Near the boundary there is some axial heat conduction and this assumption is not quite as good.

⁶¹ We have just assumed-away any dependence on independent variables other than r .

In Equation 30 we divided through by r to put the equation into a more standard form. We might also have divided through by k so that the coefficient for the highest derivative of r is equal to 1 - that would have really been the standard form - but for purposes of this lecture we will leave the k where it is.

33 is equal to the constant computed on the right. Alternatively we might assume a maximum permissible fuel center line temperature and solve for the corresponding linear heat rate, fuel surface heat flux, or volumetric heat generation rate.⁶²

FRAPCON Fuel Thermal Conductivity Model

The standard model to use for uranium oxide and mixed oxide nuclear fuel is the FRAPCON-3 model established from empirical data and maintained by Pacific Northwest National Laboratory.⁶³ The model captures changes to thermal conductivity due to the following factors:

- **Temperature.** For many materials, increase in temperature results in a decrease in thermal conductivity. As can be seen in Figure 77, this is true for oxide fuels below 1750°C but above that temperature thermal conductivity increases slightly.
- **UO₂ vs Mixed Oxides.** Adding a mixture of plutonium oxides along with uranium oxides results in lower thermal conductivity over most temperatures. This behavior is shown in Figure 78. Generally speaking the more complex behavior of thermal conductivity of MOX fuel adds to the list of issues to address when transitioning to a closed fuel cycle where plutonium produced in reactors is consumed as fuel.
- **Burn-up.** The overall effect on thermal conductivity is due to a number of phenomena that all occur during core life including changes in the porosity, composition and stoichiometry of the fuel. Generally, as burn-up increases, the thermal conductivity of the fuel decreases. This is shown for UO₂ fuel in Figure 79.
- **Density.** Generally, increasing UO₂ density increases thermal conductivity with the converse effect for reduced density. The effect of density on thermal conductivity is shown quantitatively in Figure 80.

All of these effects are captured in the FRAPCON model for thermal conductivity. The model is as shown in Equation 34

$$k_{0.95 \text{ TD}} = \left\{ A + BT + a \cdot \text{gad} + f(Bu) + \left[1 - 0.9e^{-0.04Bu} \right] g(Bu)h(T) \right\}^{-1} + \frac{E}{T^2} e^{-F/T} \quad (34)$$

where the following constants and functions are defined:

⁶² These last two would also require that you specify the fuel pin outer radius.

⁶³ DD Lanning, CE Beyer, and KJ Gelhood. Frapcon-3: Updates, including mixed-oxide fuel properties. nureg/cr-6534, vol. 4, pnll-11513. In *Technical report*. Pacific Northwest National Laboratory, Richmond, USA, 2005

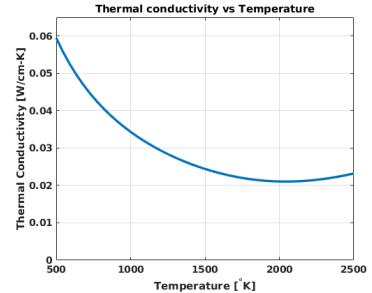


Figure 77: Thermal conductivity for unirradiated UO₂ at 95% theoretical density using the FRAPCON model.

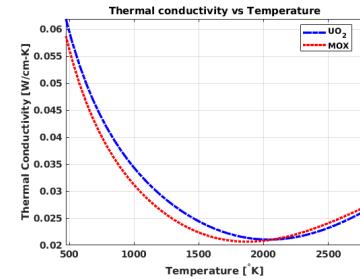


Figure 78: Comparison of UO₂ and MOX thermal conductivity.

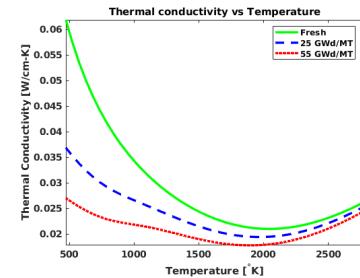


Figure 79: Impact of burn-up on k for UO₂.

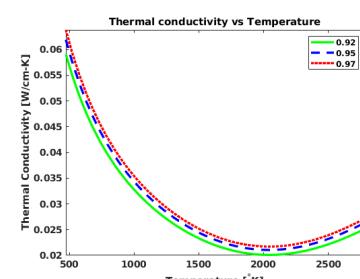


Figure 80: Impact of density on thermal conductivity for UO₂.

- $a = 115.99 \text{ [cm-K/W]}$
- Bu is burn-up given in MWd/kg or GWd/MT
- $f(Bu) = 0.187 \times Bu \text{ [cm-K/W]}$. This is intended to capture the effect of fission products in the fuel crystal matrix.
- gad = gadolinium weight percent
- $h(t) = [1 + 396e^{-Q/T}]^{-1}$. This term captures the temperature dependence of annealing of irradiation defects.
- $Q = 6380^\circ\text{K}$, temperature of annealing parameter.
- T = temperature, given in Kelvin.

The constants A , B , E , and F are given in Table 11; the value of x for the $(U,\text{Pu})\text{O}_2$ material correlations is defined by: $x = 2.0 - \text{O/M}$, where O/M is the oxygen-to-metal atomic ratio in the fuel material.

The resulting thermal conductivity $k_{0.95 \text{ TD}}$ is given in W/cm-K for oxide fuel at 95 percent of theoretical density. If the fuel density is something different than 95 percent of theoretical density, a correction factor must be applied; this is given by Equation 35.

$$\eta(\rho) = \frac{1.0789\rho/\rho_{\text{TD}}}{1 + 0.5(1 - \rho/\rho_{\text{TD}})} \quad (35)$$

so that $k(\rho) = \eta(\rho)k_{0.95 \text{ TD}}$. In total, therefore, thermal conductivity is a function of temperature, density, gadolinium concentration and burn-up for either UO_2 or $(U,\text{Pu})\text{O}_2$ fuel. We could re-write Equation 33 reflecting this as shown in the equation below:

$$\int_{T(R)}^{T(0)} k(T, gad, Bu) \eta(\rho) dT = \frac{q'}{4\pi} \quad (36)$$

Analytically solving this equation is not something we are interested in pursuing. In the next section, we will use a “graphical-method” to solve the problem; in the next lecture we will discuss how you can use MATLAB to solve Equation 36.

Example Calculation - Graphical Method

Consider a PWR core operating at full power with a fuel pin linear heat rate of 25 kW/m. Temperature of the fuel outer surface is 400°C . Assume the fuel pin was initially loaded with 5 w/o gadolinium for reactivity control. Using the FRAPCON-3 model incorporated into Figure 81, estimate the temperature at the center of the fuel pin at:

1. beginning of life;

	UO_2	$(U,\text{Pu})\text{O}_2$
$A \text{ [cm-K/W]}$	4.52	$285x + 3.5$
$B \text{ [cm/W]}$	2.46×10^{-2}	$(2.86 - 7.15x) \times 10^{-2}$
$E \text{ [W-K/cm]}$	3.5×10^7	1.5×10^7
$F \text{ [K]}$	16,361	13,520

Table 11: Values of constants in FRAPCON correlations for thermal conductivity.

2. burn-up of 25 MWd/kg; and
3. burn-up of 55 MWd/kg

To solve these problems we need to a) select the correct curve corresponding to burn-up and gadolinium concentration; b) use the given linear heat rate to calculate the right-hand-side of Equation 36; and c) use the curve as shown below.

Using the graph, we can evaluate:

$$\int_{T(R)}^{T(0)} k(T, gad, Bu) \eta(\rho) dT = K(T(0)) - K(T(R)) = \frac{q'}{4\pi} \quad (37)$$

where $K(T)$ the integral of the thermal conductivity evaluated to temperature T .

For the first problem we select the top dashed line corresponding to unirradiated UO_2 fuel with 5 wt% Gd; we evaluate the integral of the thermal conductivity at the fuel outer surface temperature of 400°C and get a result just under 20 W/cm. This is illustrated in Figure 82. We do not know $T(0)$ yet but we can calculate the right hand side of Equation 37 from the given linear heat rate; $\frac{q'}{4\pi} = \frac{250\text{W/cm}}{4\pi} = 19.9\text{W/cm}$ ⁶⁴ So even if we do not know $T(0)$, we know that $K(T(0)) - K(T(R)) \approx 20\text{W/cm}$, so we just need to find where the top dashed curve giving the integral of thermal conductivity intersects with, roughly, $20 + 20 = 40\text{W/cm}$. We read the corresponding temperature and its about 1100°C.

Parts 2 and 3 are done in the same way; we simply need to use different thermal conductivity curves corresponding to 25 MWd/kg and 55 MWd/kg burn-up. The graphical solutions to the second and third problems are shown in Figure 83 and Figure 84.

Note that as burn-up increases, fuel thermal conductivity decreases, and for the same linear heat rate and external fuel temperature, the center line temperature of the fuel increases. This should correspond with what you expect to see for system response to decreasing fuel thermal conductivity.

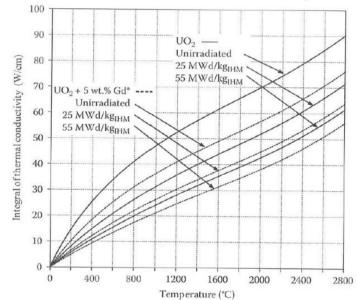


Figure 81: Integral of fuel thermal conductivity.

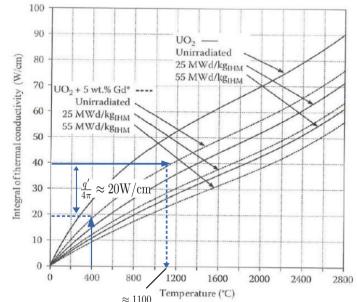


Figure 82: Using graphical method for fuel thermal conductivity; part 1.

⁶⁴ Note: Pay attention to units for the given linear heat rate and thermal conductivity.

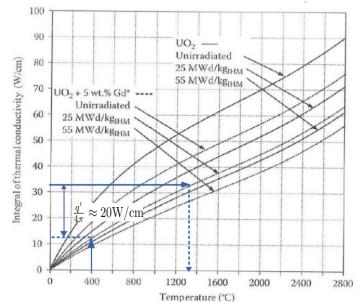


Figure 83: Using graphical method for fuel thermal conductivity; part 2.

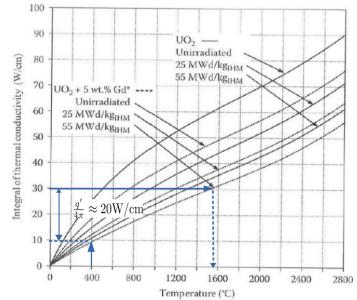


Figure 84: Using graphical method for fuel thermal conductivity; part 3.

Lecture 22 - Fuel Thermal Conductivity with MATLAB

Objectives

The objective of this lecture is:

1. Explain the implementation of the FRAPCON fuel thermal conductivity model in a MATLAB environment.

FRAPCON Model—restated

The FRAPCON Model was discussed in Lecture 22, but in brief, the equation is as follows:

$$k_{0.95 \text{ TD}} = \left\{ A + BT + a \cdot \text{gad} + f(Bu) + \left[1 - 0.9e^{-0.04Bu} \right] g(Bu)h(T) \right\}^{-1} + \frac{E}{T^2} e^{-F/T}$$

where the constants and functions included in this definition can be obtained from the previous lecture.

We adjust for density as follows:

$$\eta(\rho) = \frac{1.0789\rho/\rho_{\text{TD}}}{1 + 0.5(1 - \rho/\rho_{\text{TD}})}$$

so that $k(\rho) = \eta(\rho)k_{0.95 \text{ TD}}$.

MATLAB implementation

The first step will be to establish a MATLAB representation for the constants and functions that are used in the correlation:

```
% range of applicability
% 300K < T < 3000 K
% Bu < 62 Gwd/Mt
% Gad < 10 w.t. %
% 92% < TD < 97%
a_frapcon = 115.99; % cm-K/W
f_frapcon = @(Bu) 0.187*Bu; %cm-K/W
g_frapcon = @(Bu) 3.8*(Bu.^0.28); % cm-K/W
Q_frapcon = 6380; %K
h_frapcon = @(T) 1./(1 + 396*exp(-Q_frapcon./T)); % unitless
```

Listing 13: Step 1.

Since different types of oxide fuels are possible, we set up a selection structure to pick between UO₂ and MOX.

```

1 % set constants based on fuel type.
2 fuel_type = 'UO2'; % fuel type = 'UO2' or 'MOX'
3 switch fuel_type
4   case 'UO2'
5     A_frap = 4.52; %cm-K/W
6     B_frap = 2.46e-2; %cm/W
7     E_frap = 3.5e7; %W-K/cm
8     F_frap = 16361; %K
9
10  case 'MOX'
11    OM = 2.0; % set for MOX fuel - default 2.0
12    x_frap = 2.0 - OM;
13    A_frap = 285*x_frap + 3.5; %cm-K/W
14    B_frap = (2.86 - 7.15*x_frap)*1e-2; %cm/W
15    E_frap = 1.5e7; %W-K/cm
16    F_frap = 13520; %K
17  otherwise
18    % raise an error
19    error('Unknown fuel type for FRAPCON model\n');
20 end

```

Listing 14: Step 1 continued.

The next step will be to create a MATLAB representation of the FRAPCON thermal conductivity correlation. You need to include the density correction function $\eta(\rho)$ if you want to model fuels with density other than 95 percent theoretical density.

```

% equation 8.22a
2 k_op95_frap = @(T,gad,Bu) ...
3   1./(A_frap + B_frap*T+a_frapcon*gad+f_frapcon(Bu) +...
4   (1 - 0.9*exp(-0.04*Bu)).*g_frapcon(Bu).*h_frapcon(T)) +...
5   E_frap./(T.^2).*exp(-F_frap./T);
6 % (be careful with syntax to allow any argument to be a vector)
7
8 % correction factor for theoretical density
9 eta_frap = @(den) 1.0789*den/(1+0.5*(1-den));
10
11 % combined expressions
12 k_frap = @(T,gad,Bu,den) k_op95_frap(T,gad,Bu)*eta_frap(den);

```

Note: Be sure to write your local functions so they can accept vector arguments.

Listing 15: MATLAB function for the FRAPCON correlation.

Now we have the correlation available in our environment, it is time to apply problem specific parameters such as burn-up, density, and gadolinium concentration.

```

1 %% Given Parameters
2 T_fo = 400; % C
3 q_linear = 25; % kW/m
4 gad = 0.00; % mass fraction of gadolinia
5 Bu = 55; % GWd/MT_ihm

```

```

1 den = 0.92; % percent theoretical density
2
3 % Function that now includes
4 % problem-specific parameters.
5 k_frap_p = @(T) k_frap(T,gad,Bu,den);

```

Listing 16: Apply problem-specific parameters.

Recall that the equation we need to solve is:

$$\int_{T(R)}^{T(0)} k(T, gad, Bu) \eta(\rho) dT = \frac{q'}{4\pi} \quad (38)$$

where we need, for instance, to find the maximum center line temperature, $T(0)$, resulting from a given fuel outer temperature and linear heat rate. This is in general a non-linear equation. We can make use of MATLAB's powerful and reliable built-in non-linear root-finding function "fzero" if we re-arrange Equation 38 as follows:

$$\int_{T(R)}^{T(0)} k(T, gad, Bu) \eta(\rho) dT - \frac{q'}{4\pi} = 0 \quad (39)$$

We do this in the listing below where, please note, we have to do some unit conversions to make sure that q' is compatible with the units of thermal conductivity.

```

1 f_fuel = @(T) integral(k_frap_p ,T_fo+273.15,T) - ...
   (q_linear*1000/100)/(4*pi);

```

Listing 17: Set up non-linear equation for solver.

Lastly we will provide an initial guess for $T(0)$ and invoke "fzero" to get the actual fuel center line temperature.

```

T_est = 2250;
2 T_max_K = fzero(f_fuel ,T_est+273.15);
T_max = T_max_K - 273.15;
4
% report the result
6 fprintf('The fuel center line temperature is %5.4g degrees C.\n'
   ,T_max);

```

Listing 18: Solve for fuel centerline temperature.

You should compare the result you get to the result you obtain using the graphical method described in Lecture 21. Full code for this example is proved in the Appendices.

Note: The built-in function *integral* will call the in-line function *k_frap_p* with vectors of "integration points." If you failed to implement *k_frap_p* so that it could correctly accept vector arguments, *integral* will fail.

Assignment #7: Fuel Pin Thermal Analysis

CONSIDER A fuel pin with the following parameters:

Temperature on outer surface of fuel pin	630°C
Linear heat rate	33.0 kW/m
Fuel density	95% of theoretical density
Gadolinium weight fraction	5%
Burnup	25 GWD/MT _{ihm}

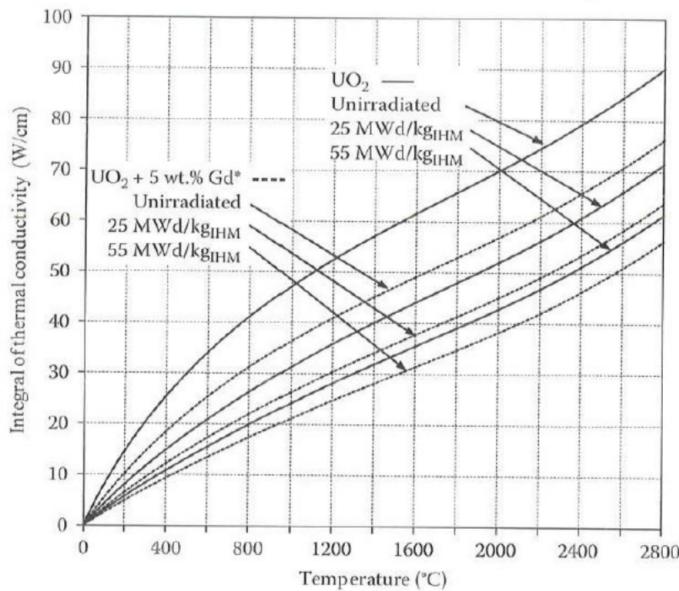


Figure 85: Integral of fuel thermal conductivity.

1. Use Figure 85 to estimate the fuel centerline temperature (°C) under the given conditions.
2. Implement the FRAPCON model in MATLAB to compute the fuel centerline temperature.

3. With your MATLAB implementation of the FRAPCON model, estimate fuel centerline temperature for 97% and 92% theoretical density.
4. Repeat your analysis using the FRAPCON model for the conditions given in the table above but with Gadolinium weight fraction changed to 0% and begining of life. (Burnup = 0)
5. Repeat the calculation of the previous problem but use the AREVA fuel thermal conductivity model (which you should implement in MATLAB) which gives thermal conductivity in units of W/m-K for UO₂ fuel at full theoretical density.

The AREVA model is as follows:

$$k_{TD} = \frac{1}{A_1 + A_2 T + A_3 BU + A_4 f(T)} + g(T)$$

$$A_1 = 0.0375 \text{ m-K/W}$$

$$A_2 = 2.165 \times 10^{-4} \text{ m/W}$$

$$A_3 = 1.70 \times 10^{-3} \text{ m-K/W}/(\text{GWd/MT}_{\text{IHM}})$$

$$A_4 = 0.058 \text{ m-K/W}$$

$$f(T) = \left[1 + \exp \left(\frac{T - 900}{80} \right) \right]^{-1}$$

$$g(T) = 4.715 \times 10^9 T^{-2} \exp \left(-\frac{16,361}{T} \right) \text{ W/m-K}$$

Lecture 23 - More Thermal Conductivity Models, Convection Intro & Review

Objectives

The objectives of this lecture are:

1. discuss gap conductance
2. present cladding and oxide layer thermal conductivity models
3. provide a summary/review of familiar convective heat transfer correlations

Gap Conductance

Typical PWR fuel engineering results in a fuel pin comprising a stack of pellets within cladding. The fuel pin is pressurized with an inert gas—typically helium—and pellets are held in place axially with springs as illustrated in Figure 86. Over time, multiple phenomena occur such as:⁶⁵

1. fill gas composition changes due to production of fission product gasses
2. structural changes to the fuel pellet such as central void formation and “hourglassing” due to sintering as illustrated in Figure 87
3. “creep-down” of cladding in which the cladding deforms slowly—possibly coming into contact with the pellets—due to external coolant pressure at high material temperature.

The void formation and complex structural re-alignment are most pronounced in fast reactor systems operating at very high temperature and high heat flux. The aforementioned FRAPCON model captures the most important features that we will focus on in this course. The gap conductance is the only other feature that we will include in this class.⁶⁶

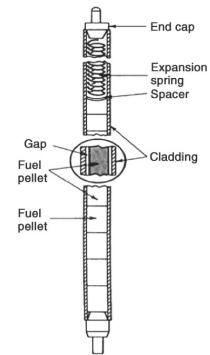


Figure 86: Typical PWR fuel pin schematic.

⁶⁵ TJ Bartel, R Dingreville, D Littlewood, V Tikare, M Bertolus, V Blanc, V Bouineau, G Carlot, C Desgranges, B Dorado, et al. State-of-the-art report on multi-scale modelling of nuclear fuels. Technical report, Organisation for Economic Co-Operation and Development, 2015

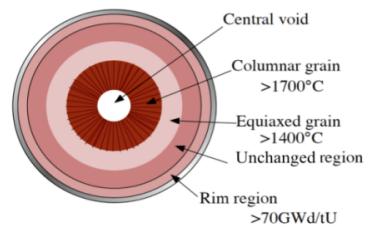


Figure 87: Schematic restructuring of nuclear fuel pellet after high temperature operation.

⁶⁶ In order to capture all effects of restructuring in the fuel pellet, different models would be needed for each region in the fuel.

For PWRs, the net dimensional changes can result in substantial contact between the fuel pellet and the cladding. This actually improves thermal performance at the cost of increased risk of fuel pin failure due to clad rupture. The mathematical model we have used for heat transfer across the fuel pellet to clad gap is:

$$\Delta T_{\text{gap}} = T_o - T_{ci} = \frac{q'}{2\pi r_g h_g} \quad (40)$$

where h_g is the gap conductance. But where does h_g come from and how is it modeled? For this course, we will consider gap conductance as a function of linear heat rate and of the initial cold diametrical gap. Such a relationship is shown graphically in Figure 88. Note that as initial gap is reduced and as linear heat rate increases, the gap conductance increases. That will be the extent of our modeling of h_g for this course.

Conduction through fuel cladding

For Zircaloy cladding, the thermal conductivity used by FRAPCON 3.5 is given by Equation 41.

$$k_{\text{clad}} = 7.51 + 2.09 \times 10^{-2}T - 1.45 \times 10^{-5}T^2 + 7.67 \times 10^{-9}T^3 \quad [\text{W/m-K}] \quad (41)$$

for temperature, T , given in Kelvin. Following a similar derivation as for conduction in the fuel, the heat equation in cylindrical coordinates⁶⁷ for conduction in the cladding with temperature-dependent thermal conductivity can be reduced to:

$$\int_{T(r_{co})}^{T(r_{ci})} k(T) dT = \frac{q'}{2\pi} \ln \left(\frac{r_{co}}{r_{ci}} \right) \quad (42)$$

There may be some circumstances where this problem can be posed and solved analytically; for this class, the default method of solving heat transfer problems of this type will be to use MATLAB. The method is completely analogous to fuel thermal conductivity and will be left for homework exercises.

Conduction through Clad Oxide Layer

At risk of being repetitive, we can apply the same mathematical tools to capture temperature-dependent conductivity effects in the oxide layer that forms on the exterior of metallic fuel cladding. Oxide layer thickness is a function of operating history and water chemistry. Typical thickness is near $20 \times 10^{-6}\text{m}$. For the model we will use,

Question: Why would h_g increase for higher values of q' ?

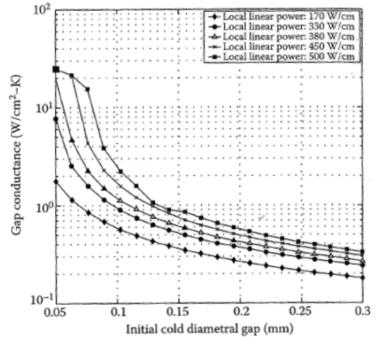


Figure 88: Gap conductance as a function of linear heat rate and initial cold diametrical gap.

⁶⁷ With the usual assumptions of symmetry with respect to ϕ and axial heat conduction being negligible.

thermal conductivity of this oxide layer is only weakly dependent on temperature. One conductivity model of the oxide layer on Zircaloy cladding reads as follows:

$$k_{\text{oxide}} = 0.835 - 1.81 \times 10^{-4}T \text{ W/m-K} \quad (43)$$

where, as usual, the temperature (T) is given in Kelvin. Similar to cladding, the steady-state heat equation can be transformed into Equation 44.

$$\int_{T(r_{oo})}^{T(r_{co})} k(T) dT = \frac{q'}{2\pi} \ln \left(\frac{r_{oo}}{r_{co}} \right) \quad (44)$$

Overall Conduction Through Fuel Pin

Cumulatively we have elaborated on the thermal models for the fuel pellet, pellet-to-clad gap, cladding, and the oxide layer outside the cladding. While will not take the time to generalize the mathematical analysis such that we could derive the full temperature profile such as is illustrated in Figure 89, we can solve for the temperature at the interfaces between each region and at the pellet center line.

Convective Heat Transfer Correlations

We have not yet elaborated on the analysis of heat transfer between the fuel clad (or fuel clad oxide) outer surface and the bulk coolant. In elementary courses it may have sufficed for us to simply provide a convective heat transfer coefficient, h_{cool} , and plug the number into an equation to find ΔT_{cool} . In this course we will pick up where your heat transfer course left off and discuss a variety of heat transfer correlations relevant to convective heat transfer in nuclear applications.

FOR NON-METALLIC FLUIDS in fully developed turbulent flow, correlations for convective heat transfer are presented in the context of the non-dimensional Nusselt number. The Nusselt number is a function of fluid material properties as well as the flow condition and is defined by Equation 45

$$\text{Nu} = \frac{hD}{k} \quad (45)$$

where h is the convective heat transfer coefficient, D is a characteristic length like pipe diameter or hydraulic diameter, and k is the fluid thermal conductivity. The correlations developed are usually of the

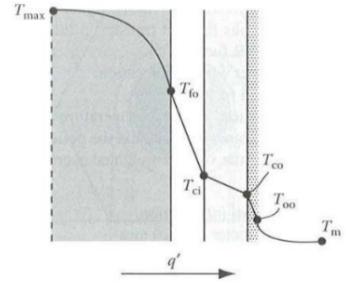


Figure 89: Fuel pin temperature profile.

Nusselt number: relates to the temperature gradient at the wall of a fluid-solid interface and the overall temperature difference between the wall and the bulk fluid.

form shown in Equation 46.

$$\text{Nu} = C \text{Re}^\alpha \text{Pr}^\beta \left(\frac{\mu_w}{\mu} \right)^\kappa \quad (46)$$

where Re is the Reynolds number, Pr is the Prandtl number of the fluid, μ and μ_w are the viscosity of the fluid in the bulk flow and along the wall where heat transfer takes place, respectively; and C, α, β , and κ are given constants.

Prandtl Number: a dimensionless fluid property that describes the ratio of viscous diffusivity to thermal diffusivity.

Correlations for Circular Tubes

Several correlations are available for convective heat transfer in circular tubes.

Dittus-Boelter Correlation

This is a widely used correlation⁶⁸ that we will present only for fully-developed turbulent flow.⁶⁹ The correlation is given by Equation 47.

$$\text{Nu}_\infty = 0.023 \text{Re}^{0.8} \text{Pr}^{0.4} \quad (47)$$

The following limits apply to the Dittus-Boelter correlation:

$$\text{Re} > 10,000, 0.7 < \text{Pr} < 100, \text{ and } L/D > 60;$$

where L corresponds to position along a circular tube and D is the tube diameter.

Colburn Correlation

For fluids with high viscosity consider the Colburn correlation.⁷⁰

$$\text{StPr} = 0.023 \text{Re}^{-0.2} \quad (48)$$

which, despite the note that this should be used with high viscosity fluids, has the same range of applicability as the Dittus-Boelter correlation. The Stanton Number (St) combines the Nusselt number, Reynolds number, and Prandtl number and is given by:

$$\text{St} = \frac{\text{Nu}}{\text{Re}\text{Pr}} \quad (49)$$

Applying Equation 49, then Equation 48 becomes:

$$\text{Nu}_\infty = 0.023 \text{Re}^{0.8} \text{Pr}^{1/3} \quad (50)$$

⁶⁸ FW Dittus and LMK Boelter. Heat transfer in automobile radiators of the tubular type. *International communications in heat and mass transfer*, 12(1):3–22, 1985

⁶⁹ As with the hydraulics analysis, we ignore heat transfer conditions near entrances or exits or, in short, anywhere other than “fully-developed” regions where the flow-field is undisturbed by such effects. Generally speaking convective heat transfer is enhanced in entrance regions owing to the increased “action” in the fluid flow field.

Note: The ∞ subscript in Nu_∞ is meant to denote that the correlation is valid for a location “infinitely far” from any entrance, exit, discontinuity, or other feature that would disrupt the fully developed turbulent flow. We will read ∞ simply to imply fully developed flow.

⁷⁰ Allan P Colburn. A method of correlating forced convection heat-transfer data and a comparison with fluid friction. *International Journal of Heat and Mass Transfer*, 7(12):1359–1384, 1964

Stanton Number: a dimensionless fluid property that characterizes the ratio of heat transfer at the wall to the energy transported by the fluid flow stream.

Seider and Tate Correlations

This correlation⁷¹ is suitable for diabatic flows of fluids including organic liquids, petroleum fractions, and air.

For organic liquids:

$$\text{Nu}_\infty = 0.015 \text{Re}^{0.85} \text{Pr}^{0.3} \quad (51)$$

For petroleum fractions:

$$\text{Nu}_\infty = 0.23 \text{Re}^{0.8} \text{Pr}^{1/3} \left(\frac{\mu_w}{\mu} \right)^{-0.14} \quad (52)$$

For air:

$$\text{Nu}_\infty = 0.027 \text{Re}^{0.8} \text{Pr}^{1/3} \left(\frac{\mu_w}{\mu} \right)^{-0.14} \quad (53)$$

where for Equation 52 and 53 all material properties should be evaluated at the bulk condition except, of course, μ_w which should be evaluated at the prevailing temperature and pressure along the wall. The range of applicability of the Seider and Tate correlations will be taken to be the same as those for the Dittus-Boelter correlation.

Gnielinski Correlations

A more recent correlation⁷² is available that extends previous work and covers transitionally turbulent flows.

For liquids:

$$\text{Nu}_\infty = 0.0120 \left(\text{Re}^{0.87} - 280 \right) \text{Pr}^{0.4} \quad (54)$$

where $0.4 < \text{Pr} < 1.5$, and $2300 < \text{Re} < 5 \times 10^6$.

For gases:

$$\text{Nu}_\infty = 0.0214 \left(\text{Re}^{0.80} - 100 \right) \text{Pr}^{0.4} \quad (55)$$

where $1.5 < \text{Pr} < 500$, and $2300 < \text{Re} < 5 \times 10^6$.

Petukhov

In cases where heating and cooling of the fluid result in significant variation in the physical properties, one should use the correlations due to Petukhov⁷³ which we first met in the hydraulics section of this course.

In this case, the basic Nusselt number will be calculated as:

$$\text{Nu}_D = \frac{(f/8)(\text{Re} - 1000)\text{Pr}}{1 + 12.7\sqrt{f/8}(\text{Pr}^{2/3} - 1)} \quad (56)$$

⁷¹ E Nn Sieder and Go E Tate. Heat transfer and pressure drop of liquids in tubes. *Industrial & Engineering Chemistry*, 28(12):1429–1435, 1936

⁷² Volker Gnielinski. New equations for heat and mass transfer in turbulent pipe and channel flow. *Int. Chem. Eng.*, 16(2): 359–368, 1976

⁷³ Boris Sergeevich Petukhov. Heat transfer and friction in turbulent pipe flow with variable physical properties. In *Advances in heat transfer*, volume 6, pages 503–564. Elsevier, 1970

where f is given by:

$$f = (1.82 \log_{10} \text{Re} - 1.64)^{-2} \quad (57)$$

Once this Nusselt number has been determined, as with friction factors in the hydraulics section of the course, we modify the Nusselt number to take into account the variability in fluid properties between the bulk conditions and those that prevail along the wall where heat transfer occurs.

For liquids:

$$\frac{\text{Nu}}{\text{Nu}_D} = \left(\frac{\mu_b}{\mu_w} \right)^m \quad (58)$$

For gases:

$$\frac{\text{Nu}}{\text{Nu}_D} = \left(\frac{T_b}{T_w} \right)^m \quad (59)$$

The coefficient m for corresponding conditions is provided in Table 12

	Liquids $0.025 \leq \mu_b/\mu_w \leq 12.5$	Gases $0.27 \leq T_b/T_w \leq 2.7$
Heating ($T_w > T_b$)	$m = 0.11$	$m = 0.47$
Cooling ($T_w < T_b$)	$m = 0.25$	$m = 0$

Table 12: Exponent for Petukhov correlation.

Lecture 24 - Convection in Rod Bundles

Objectives

The objectives of this lecture are:

1. Present relevant correlations to characterize convective heat transfer in arrays of cylindrical fuel pins.
2. Apply the correlation in a simple example problem

Convective Heat Transfer in Rod Bundles

As ONE MIGHT GUESS, rod bundles must be handled differently than circular tubes for heat transfer. Similar to hydraulic analysis, convective heat transfer performance around a fuel pin in a fuel bundle depends on a number of factors such as:

1. whether the fuel assembly is rectangular/square or triangular/hexagonal.
2. the spacing between adjacent fuel pins—non-dimensionalized, as usual, as the pitch-to-diameter ratio.
3. the position of the pin within the fuel assembly—i.e. whether the pin is on an assembly edge, corner, or somewhere in the middle.

Also, as was the case for hydraulic analysis, one must substitute characteristic dimensions such as pipe diameter, with a more generic hydraulic diameter calculated from:

$$D_h = \frac{4A_{\text{flow}}}{P_{\text{wetted}}} \quad (60)$$

where, as a reminder, flow area and wetted perimeter in coolant channels around fuel pins in rectangular and hexagonal arrays are illustrated in Figure 91.

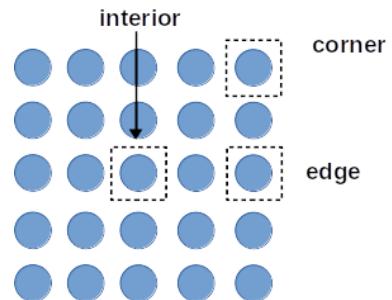


Figure 90: Schematic of fuel pin positions within a fuel assembly.

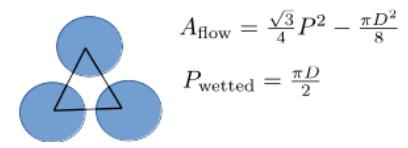
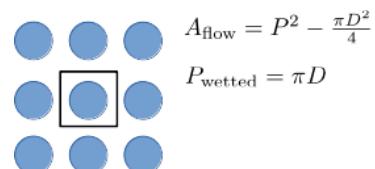


Figure 91: A_{flow} and P_{wetted} in fuel assemblies.

THE CUSTOMARY METHOD with correlations for convective heat transfer in a rod bundle will be to express the Nusselt number in terms of the Nusselt number for fully developed turbulent flow in a circular channel as shown in Equation 61

$$\text{Nu}_{\infty} = \psi (\text{Nu}_{\infty})_{\text{c.t.}} \quad (61)$$

where, of course, $(\text{Nu}_{\infty})_{\text{c.t.}}$ is the Nusselt Number for a circular tube which, unless otherwise stated, should be calculated using the Dittus-Boelter correlation. Consequently, the correlations to be presented are meant to specify a value for ψ .

Infinite Arrays

For an “infinite array” of rods, what we really mean is that all rods are treated as interior rods; there are no corners or boundaries in an infinite array and thus no corner or edge fuel cooling channels.

Presser Correlation

Presser and his colleagues provided correlations for both hexagonal and square arrays.

Hexagonal arrays:

$$\psi = 0.9090 + 0.0783 \frac{P}{D} - 0.1283 e^{-2.4(P/D-1)} \quad (62)$$

for $1.05 \leq P/D \leq 2.2$.

For square arrays:

$$\psi = 0.9217 + 0.1478 \frac{P}{D} - 0.1130 e^{-7.0(P/D-1)} \quad (63)$$

for $1.05 \leq P/D \leq 1.9$

Weisman Correlation

For the special case of water-cooled applications, Weisman provides a correlation.⁷⁴

For hexagonal arrays:

$$\psi = 1.130 \frac{P}{D} - 0.2609 \quad (64)$$

for $1.1 < P/D < 1.5$.

For square arrays:

$$\psi = 1.826 \frac{P}{D} - 1.0430 \quad (65)$$

for $1.1 \leq P/D \leq 1.3$.

⁷⁴ Joel Weisman. Heat transfer to water flowing parallel to tube bundles. *Nuclear Sci. and Eng.*, 6, 1959

Finite Arrays

Correlations for finite arrays generally incorporate some mechanism to account for the fuel rod location within the assembly. As previously discussed, a reasonably simple model will consider only three cases: a rod on the edge, corner, or interior of the assembly.

The one correlation of this type that we will consider is from Markoczy.⁷⁵ The basic correlation is given by Equation 66

$$\psi = 1 + 0.9120 \text{Re}^{-0.1} \text{Pr}^{0.4} \left(1 - 2.0043 e^{-B} \right) \quad (66)$$

where the parameter B is given by:

$$B = \frac{D_e}{D} \quad (67)$$

where, in turn, D_e is given by:

$$D_e = 4 \frac{\sum_j A_j}{\sum_j P_{\text{wetted},j}} \quad (68)$$

where A_j and $P_{\text{wetted},j}$ are the flow area and wetted perimeter of all flow channels adjacent to the rod. This is where the effect of a finite array can be seen.

In the case of a typical interior rod—where all of the adjacent channels are the same and which for this class is the only case we will consider—the equation for B reduces to:

$$B = \frac{2\sqrt{3}}{\pi} \left(\frac{P}{D} \right)^2 - 1 \quad (69)$$

for triangular arrays, and:

$$B = \frac{4}{\pi} \left(\frac{P}{D} \right)^2 - 1 \quad (70)$$

for square arrays.

Example

Consider a typical fuel assembly for an AP1000 core. The fuel pins are arranged in a 17x17 square array. Each fuel pin has an outside diameter of 0.374 inches and a rod pitch of 0.496 inches. Coolant flows through the average channel with a velocity of 15.9 ft/sec. Coolant properties are as given in Table 13

Find:

1. Nusselt number using Dittus-Boelter correlation.

⁷⁵ G Markoczy. Convective heat transfer in rod clusters with turbulent axial coolant flow. p. 1. *Waerme Stoffuebertrag*, 5(4):204–212, 1972

Property	Value
Average ρ [lb _m /ft ³]	45.2
Average μ [lb _m /ft-hr]	0.23
Average c_p [BTU/lb _m -R]	1.31
Average k [BTU/hr-ft-R]	0.32

Table 13: Fluid properties for Lecture 24 example.

2. Nusselt number and convective heat transfer coefficient using
 - (a) Presser correlation
 - (b) Markoczy correlation for an interior channel
3. The temperature at the clad surface at a location in the channel where the bulk coolant temperature is 580°F and the heat flux is 346,104 BTU/hr-ft². Use the convective heat transfer coefficient from the Markoczy correlation.

Solution

To use the correlations, we need to find the hydraulic diameter of the flow channel surrounding a typical interior fuel pin from the square array.

$$A_f = P^2 - \frac{\pi}{4}D^2 = 0.496^2 - \frac{\pi}{4}(0.374)^2 = 0.1362\text{in}^2 = 9.455 \times 10^{-4} \text{ ft}^2$$

$$P_w = \pi D = \pi \left(\frac{0.374}{12} \right) = 0.09791 \text{ ft}$$

$$D_h = \frac{4A_f}{P_w} = \frac{4(9.455 \times 10^{-4})}{0.09791} = 0.03863 \text{ ft}$$

With this we can calculate Reynolds number and, with given material properties Prandtl number

$$\text{Re} = \frac{\rho v D_h}{\mu} = \frac{(45.2 \text{ lb}_m/\text{ft}^3)(15.9 \text{ ft/s})(0.03863 \text{ ft})}{0.23 \text{ lb}_m/\text{ft}\cdot\text{hr}(1 \text{ hr}/3600 \text{ sec})} = 434,520$$

$$\text{Pr} = \frac{\mu c_p}{k} = \frac{(0.23 \text{ lb}_m/\text{ft}\cdot\text{hr})(1.31 \text{ BTU/lb}_m\cdot\text{R})}{0.32 \text{ BTU/hr}\cdot\text{ft}\cdot\text{R}} = 0.9416$$

Now we can use the computed parameters in the specified correlations.

1. Dittus Boelter: $0.023\text{Re}^{0.8}\text{Pr}^{0.4} = 0.023(434,545)^{0.8}(0.9416)^{0.4} = 727.3$

2.a Presser: $\phi()$ where

$$\phi = 0.9217 + 0.1478 \frac{P}{D} - 0.1130 e^{-7(P/D-1)}$$

where $P/D = 0.496/0.374 = 1.326$. This gives a Nusselt number of 804.5

2.b Markoczy: the equations will not be repeated here, the resulting Nusselt number is 801.5

3. To answer this question we need to convert the Nusselt number that we got with the Markoczy correlation into a convective heat transfer coefficient. From the definition of Nusselt number, we get:

$$h = \frac{\text{Nuk}}{D_h} = \frac{(801.5)(0.32 \text{ BTU/hr}\cdot\text{ft}\cdot\text{R})}{0.03863 \text{ ft}} = 6639 \text{ BTU/hr}\cdot\text{ft}^2\cdot\text{R}$$

$$\Delta T_{\text{cool}} = \frac{q''}{h} = \frac{364,104 \text{ BTU/hr}\cdot\text{ft}^2}{6639 \text{ BTU/hr}\cdot\text{ft}^2\cdot\text{R}} = 52.1^\circ\text{R}$$

So the clad surface temperature is 580°F + 52.1 ≈ 632°F.

Lecture 25 - Convection with Metallic Fluids

Objectives

The objectives of this lecture are:

1. Present correlations to characterize convective heat transfer for fully developed turbulent flow in smooth circular tubes with metallic fluids.
2. Present a correlation for convective heat transfer with metallic fluids between parallel flat plates and concentric annuli.
3. Present correlations for convective heat transfer with metallic fluids in rod bundles.

Liquid Metal Convection Correlations

WHEN IT COMES to convection, liquid metals are different. The balance between conduction and advection present in all convection problems is less heavily skewed towards advection, particularly for flow conditions prevalent in liquid metal reactors—or at least liquid metal reactor design concepts. One other difference between convection for common fluids like air or water and liquid metals is the vast disparity in operating experience and experimental results on which to base correlations.

The basic form of Nusselt number correlations for liquid metals is shown in Equation 71:

$$\text{Nu}_{\infty} = A + B(\text{Pe})^C \quad (71)$$

where A , B , and C are constants that depend on the geometry and boundary conditions— C being close to 0.8 and A reflecting the significance of conductive heat transfer even at low Reynolds number; and Pe is the Peclet number which is defined as the product of Reynolds number and Prandtl number.

Peclet number: a dimensionless number characterizing the ratio of heat transfer by convection to the heat transfer by conduction.

$$\text{Pe} = \text{Re} \cdot \text{Pr} \quad (72)$$

Circular Tube Correlations

FOR CIRCULAR TUBES with constant heat flux along the tube boundaries, Lyon provides a correlation⁷⁶ given by Equation 73:

$$\text{Nu}_{\infty} = 7 + 0.025\text{Pe}^{0.8} \quad (73)$$

For smooth circular tubes with uniform wall temperature, a correlation is provided by Seban and Shimazaki⁷⁷ given by Equation 74:

$$\text{Nu}_{\infty} = 5 + 0.025\text{Pe}^{0.8} \quad (74)$$

Flow Past Parallel Plates and Concentric Annuli

FOR FULLY DEVELOPED flow with constant heat flux through one wall only, Seban provides the following correlation⁷⁸ given by Equation 75:

$$\text{Nu}_{\infty} = 5.8 + 0.02\text{Pe}^{0.8} \quad (75)$$

Seban presented another correlation for fully developed flow through concentric annuli with uniform heat flux on the inner wall; this is given in Equation 76

$$\text{Nu}_{\infty} = 5.25 + 0.0188\text{Pe}^{0.8} \left(\frac{D_2}{D_1} \right)^{0.3} \quad (76)$$

where D_1 and D_2 are as shown in Figure 92 and $D_2/D_1 > 1.4$.

In the case that the concentric annuli are very closely fitted such that $D_2 \approx D_1$ then Seban recommends that Equation 75 be used instead.

Flow in Rod Bundles

THE COURSE TEXTBOOK lists a few correlations that can be used for flow in rod bundles. As one might expect from previous discussion on flow in rod bundles, it is essential that some dependence on pitch-to-diameter ratio is captured as this geometric parameter has great influence on flow behavior and resultant convective heat transfer performance.

⁷⁶ Richard N Lyon and H Poppendiek. Liquid-metal heat transfer. *Liquid-metals Handbook*, page 184, 1951

⁷⁷ Ralph Alois Seban and Trans Shimazaki. Heat transfer to a fluid flowing turbulently in a smooth pipe with walls at constant temperature. Technical report, CALIFORNIA UNIV BERKELEY INST OF ENGINEERING RESEARCH, 1949

⁷⁸ RA Seban. Heat transfer to a fluid flowing turbulently between parallel walls with asymmetric wall temperatures. In *ASME*, volume 72, page 789, 1950

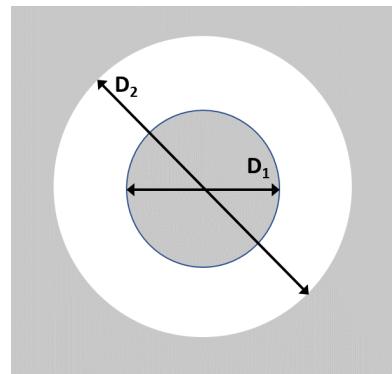


Figure 92: Concentric annuli geometry.

Two of the correlations presented by Kazimi and Carelli are repeated here⁷⁹; one we will refer to simply as the Kazimi and Carelli correlation given in Equation 77;

$$\text{Nu}_{\infty} = 4.0 + 0.33 \left(\frac{P}{D} \right)^{3.8} \left(\frac{\text{Pe}}{100} \right)^{0.86} + 0.16 \left(\frac{P}{D} \right)^{5.0} \quad (77)$$

for $1.1 \leq P/D \leq 1.4$ and $10 \leq \text{Pe} \leq 5000$. The other is the Schad-modified correlation given by Equation 78 and Equation 79:

$$\text{Nu}_{\infty} = \left[-16.15 + 24.96 \left(\frac{P}{D} \right) - 8.55 \left(\frac{P}{D} \right)^2 \right] \text{Pe}^{0.3} \quad (78)$$

for $1.1 \leq P/D \leq 1.5$ and $150 \leq \text{Pe} \leq 1000$

$$\text{Nu}_{\infty} = 4.496 \left[-16.15 + 24.96 \left(\frac{P}{D} \right) - 8.55 \left(\frac{P}{D} \right)^2 \right] \quad (79)$$

for $\text{Pe} \leq 150$.

Lastly, we include a correlation provided by Borishanskii et al⁸⁰ which covers a similar range of P/D and Peclet numbers, and is given in Equation 80:

$$\begin{aligned} \text{Nu}_{\infty} = 24.15 \ln & \left[-8.12 + 12.76 \left(\frac{P}{D} \right) - 3.65 \left(\frac{P}{D} \right)^2 \right] \\ & + 0.0174 \left[1 - e^{6-6^{P/D}} \right] [\text{Pe} - 200]^{0.9} \end{aligned} \quad (80)$$

for $1.1 \leq P/D \leq 1.5$ and $200 \leq \text{Pe} \leq 2000$ and if $\text{Pe} < 200$ you should omit the last part of Equation 80 and just use:

$$\text{Nu}_{\infty} = 24.15 \ln \left[-8.12 + 12.76 \left(\frac{P}{D} \right) - 3.65 \left(\frac{P}{D} \right)^2 \right] \quad (81)$$

RIGHT ABOUT NOW the reader may be wondering: "...out of all these correlations, which one should I choose?". A fair question, and my short answer is: all potentially applicable correlations. There may be some insight to be gained in comparing the results from different applicable correlations; perhaps some idea as to the uncertainty based on the range of results returned by different correlations. That would be something, but not anything you would hang your hat on. Whatever the result, from whichever correlation, you should understand that extensive testing—whether on prototypes or as pre-operational testing on the as-constructed system—will be required to prove that performance is within expected bounds.

⁷⁹ MS Kazami and MD Carelli. Heat transfer correlation for analysis of crbrp assemblies. westinghouse report. Technical report, CRBRP-ARD-0034, 1976

⁸⁰ VM Borishanskii, MA Gotovskii, and EV Firsova. Heat transfer to liquid metals in longitudinally wetted bundles of rods. *Soviet Atomic Energy*, 27(6): 1347–1350, 1969

Assignment #8: Convective Heat Transfer

A lead-cooled fast reactor with Supercritical Carbon Dioxide (S-CO₂) Brayton Cycle for energy conversion has been proposed for a transportable small modular reactor design. In order to transfer the thermal energy of the reactor to the S-CO₂ power conversion cycle, an intermediate heat exchanger (IHX) has been proposed. S-CO₂ will flow on the inside of the IHX tubes and liquid lead will flow over the outside. The convective heat transfer coefficient of liquid lead is an order of magnitude higher than that of S-CO₂. To compensate, the inside of the IHX tubes will be augmented with helical grooves. The goal of this workshop-style assignment is to compare the thermal and hydraulic consequence of these helical grooves.

RELEVANT DESIGN information is provided in the table below:

Basic Thermal Data	
Thermal Power	300 MW
\dot{m}_{lead}	21,700 kg/s
$\dot{m}_{\text{S-CO}_2}$	1,594 kg/s
$T_{\text{in,lead}}$	600°C
$T_{\text{out,lead}}$	503°C
$T_{\text{in,S-CO}_2}$	397°C
$T_{\text{out,S-CO}_2}$	550°C
Pressure S-CO ₂	19.7 MPa
IHX tube data	
Number of Tubes	10,000
Outer tube diameter	2 cm
Inner tube diameter	1.6 cm
Pitch-to-diameter ratio	1.25
Lattice type	Hexagonal
Tube wall thermal conductivity	0.026 kW/m·°C
Helical rib specification	
Rib height - e	0.35 mm
Helix angle - α	27 degrees
Contact angle profile - β	90 degrees
Rib separation - p or δ	0.02

THERMAL PROPERTY data is provided in the following table:

	Lead	S-CO₂
Average temperature (°C)	551.5	473.5
Average density (kg/m ³)	10,382	136.6
Average viscosity (Pa-s)	1.666×10^{-3}	3.519×10^{-5}
Average specific heat (kJ/kg-°C)	0.1443	1.228
Prandtl Number	0.01316	0.751

Compute Basic Parameters

1. Compute the total flow area for S-CO₂ (m²)
2. Compute the average velocity for S-CO₂ (m/s)
3. Compute the flow area (m²) and wetted perimeter (m) per tube for a typical interior subchannel for the lead; ignore the potential presence of spacer grids.
4. Compute the average velocity of the lead (m/s)
5. Compute the Reynolds number for the lead and S-CO₂ side of the IHX.
6. Compute the Peclet number for the lead.
7. Compute the log-mean temperature difference (ΔT_{LM}) (°C) for the heat exchanger.

Perform Analysis for Smooth IHX Tubes

8. Compute the Nusselt number on the lead side of the tubes using the Kazimi-Carelli correlation.
9. Compute the corresponding convective heat transfer coefficient on the lead-side of the IHX tube. (kW/m²-°C)

10. Compute the Fanning friction factor for a smooth tube for the S-CO₂ side. Use the smooth tube correlation associated with the Ravigururajan and Bergles correlation for heat exchangers with extended surfaces.
11. Compute the Nusselt number for the smooth-tube heat exchanger on the S-CO₂ side using the equation below.

$$\text{Nu}_{\text{sm}} = \frac{\left(f'_{\text{sm}}/2\right) \text{RePr}}{1 + 12.7 \sqrt{\frac{f'}{2}} \left(\text{Pr}^{2/3} - 1\right)}$$

12. Compute the corresponding convective heat transfer coefficient for the smooth tube for the S-CO₂ side. (kW/m²-°C)
13. Compute the overall heat transfer coefficient (kW/m²-°C) for the smooth-tube case using the equation below:

$$U = \frac{1}{\frac{1}{h_{\text{S-CO}_2}} + \frac{D_{\text{tube},i}}{2k_w} \ln \frac{D_{\text{tube},o}}{D_{\text{tube},i}} + \frac{D_{\text{tube},i}}{D_{\text{tube},o}} \frac{1}{h_{\text{lead}}}}$$

14. Compute total heat exchanger area (m²) required to transfer the design thermal power for the smooth tube case.
15. Compute the required length (m) of the heat exchanger (note that the area calculated in the previous question is in reference to the tube inner diameter, i.e. $A_{\text{total}} = \pi D_{\text{tube},i} L_{\text{tube}} N_{\text{tubes}}$).
16. Compute the differential pressure (kPa) due to major head losses for the smooth tube case recalling:

$$\Delta p = f \left(\frac{L}{D} \right) \frac{\rho V^2}{2}$$

where f is the Darcy friction factor. Recall also that the friction factor calculated in the analysis above is the *Fanning* friction factor. The relationship between the Fanning friction factor and the Darcy friction factor is:

$$f_{\text{Darcy}} = 4f'_{\text{Fanning}}$$

Perform Analysis for Enhanced IHX

17. Calculate the friction factors for the tubes with helical ribs using the Ravigururajan and Bergles correlation.

18. Calculate the Nusselt Number for the enhanced tubes using the correlation due to Bergles and Ravigururajan given in the equation below:

$$\frac{Nu_{augmented}}{Nu_{smooth}} = \left\{ 1 + \left[2.64Re^{0.036} \left(\frac{e}{D} \right)^{0.212} \left(\frac{\delta}{D} \right)^{-0.21} \left(\frac{\alpha}{90} \right)^{0.29} Pr^{-0.024} \right]^7 \right\}^{1/7}$$

19. Compute the corresponding convective heat transfer coefficient ($\text{kW}/\text{m}^2 \cdot ^\circ\text{C}$) for the augmented tubes.

20. Compute the overall heat transfer coefficient ($\text{kW}/\text{m}^2 \cdot ^\circ\text{C}$) for the augmented case.

21. Compute the overall area (m^2) required for the heat exchanger in the augmented case.

22. Compute the length (m) of the augmented heat exchanger.

23. Compute the pressure drop (kPa) for the augmented heat exchanger.

Lecture 26 - Introduction to Two-phase Flow

Objectives

The objectives of this lecture are:

1. Provide some basic definitions of two-phase flow relevant to nuclear engineering
2. Qualitatively describe two-phase flow regimes
3. Discuss relevance of two-phase flow for hydraulic and thermal phenomena important for nuclear engineers

Introduction and Disclaimer

FOR THE FIRST SIX YEARS this course was taught, no material was presented relating to two-phase flows; it was simply relegated to graduate school for any student who might want to specialize in thermal-hydraulics. During those six years there was an increasing sense that such an attitude is untenable; two-phase flows are an important element in the engineering of pressurized water reactors and two-phase flow phenomena—which play an essential role in understanding concepts such as “Departure from Nucleate Boiling” (DNB) and “Dry-out” as mechanisms of “Critical Heat Flux” (CHF)—have such relevance that I decided they no longer could be ignored completely.

THIS VERSION of the notes includes three lectures on two-phase flow and thermal limits, including this lecture. The treatment is still elementary but advances to the point where (what I think are) essential two-phase phenomena like sub-cooled nucleate boiling and DNB can be treated quantitatively. At least one correlation to predict DNB will be presented and, perhaps, that is a bare minimum of coverage. These three lectures are not intended as a rigorous treatment of two-phase thermal-hydraulics as, of course, they could never accomplish

such a feat in such a short time period. Nonetheless it is hoped that this so-called bare minimum suffices as an introduction.

Definitions

The following terms are needed for our short treatment of two-phase flow:

- **Void Fraction:** for two-phase flow in a channel, the void fraction is the fraction of the cross-sectional flow area that is filled with vapor as given in Equation 82:

$$\alpha = \frac{A_v}{A_v + A_\ell} \quad (82)$$

where A_v is the area with vapor and A_ℓ is the area with liquid.

- **Static quality:** is the same as what, up until now in these lecture notes, we have been calling “quality”. In a two-phase flow it is the fraction of the mass of fluid that is in the vapor form as given in Equation 83:

$$x_{st} = \frac{m_v}{m_v + m_\ell} \quad (83)$$

- **Flow quality:** this is similar to static quality but relates to mass flow rate instead; it is the fraction of the mass flow rate at a channel cross section that is vapor. This is shown in Equation 84:

$$x = \frac{\dot{m}_v}{\dot{m}_v + \dot{m}_\ell} \quad (84)$$

- **Equilibrium quality:** this is a generalization of static quality of flow quality in that it also extends to sub-cooled liquids and super-heated vapors. The definition is given in Equation 85:

$$x_{eq} = \frac{h - h_\ell}{h_{fg}} \quad (85)$$

where h is the enthalpy of the two-phase flow, h_ℓ is enthalpy of saturated liquid at the static pressure, and h_{fg} is the difference between the enthalpy of saturated vapor and saturated liquid at the static pressure.

- **Slip ratio:** is the ratio of velocity of the vapor phase and liquid phase in a two-phase flow. This is given in Equation 86:

$$S = \frac{v_v}{v_\ell} \quad (86)$$

- **Mixture density:** is a measure of the density of the two-phase flow where the density is weighted by the void fraction. This is shown in Equation 87:

$$\rho_m = \alpha \rho_v + (1 - \alpha) \rho_\ell \quad (87)$$

where α is the void fraction and, as you have probably guessed, ρ_v is the density of the vapor phase and ρ_ℓ is the density of the liquid phase.

A relation can be derived between flow quality, void fraction and slip ratio as shown in Equation 88:

$$\alpha = \frac{1}{1 + \frac{\rho_v}{\rho_\ell} S \frac{1-x}{x}} \quad (88)$$

FOR THIS CLASS, we will assume *homogeneous flow* which means that the vapor and fluid phases of the mixture are both moving at the same velocity⁸¹ and that flow quality equals static quality. This assumption is made so we can treat two-phase mixtures as a single fluid with properties that vary with flow quality.

The assumption of homogeneous flow is mostly valid in the *dispersed bubbly flow* regime which is widely prevalent in PWRs in which subcooled nucleate boiling is taking place. Homogeneous flow is also valid in light water reactors in a post-LOCA condition in which, if there is flow through the core, it may well be a *mist flow*.

Qualitative Description of Flow Regimes

What follows is a short and qualitative description of flow regimes. The map is shown in Figure 93

- For **low** values of flow quality and **low** flow rate, we get *bubbly flow*.
- For **low** values of flow quality and **high** flow rate, we get *dispersed bubbly flow*.
- For **intermediate** values of flow quality and **low** flow rate, we get *plug/slug flow*.
- For **intermediate** values of flow quality and **high** flow rate, we get *churn flow*.
- **High** flow quality and **low** flow rate results in *annular flow*; and
- **High** flow quality and **high** flow rate results in *mist flow*.

⁸¹ This means that slip ratio, S , equals unity.

Note: both *dispersed bubbly flow* and *mist flow* are regimes where one phase predominates; the other phase is sparsely dispersed within the (almost) single-phase medium.

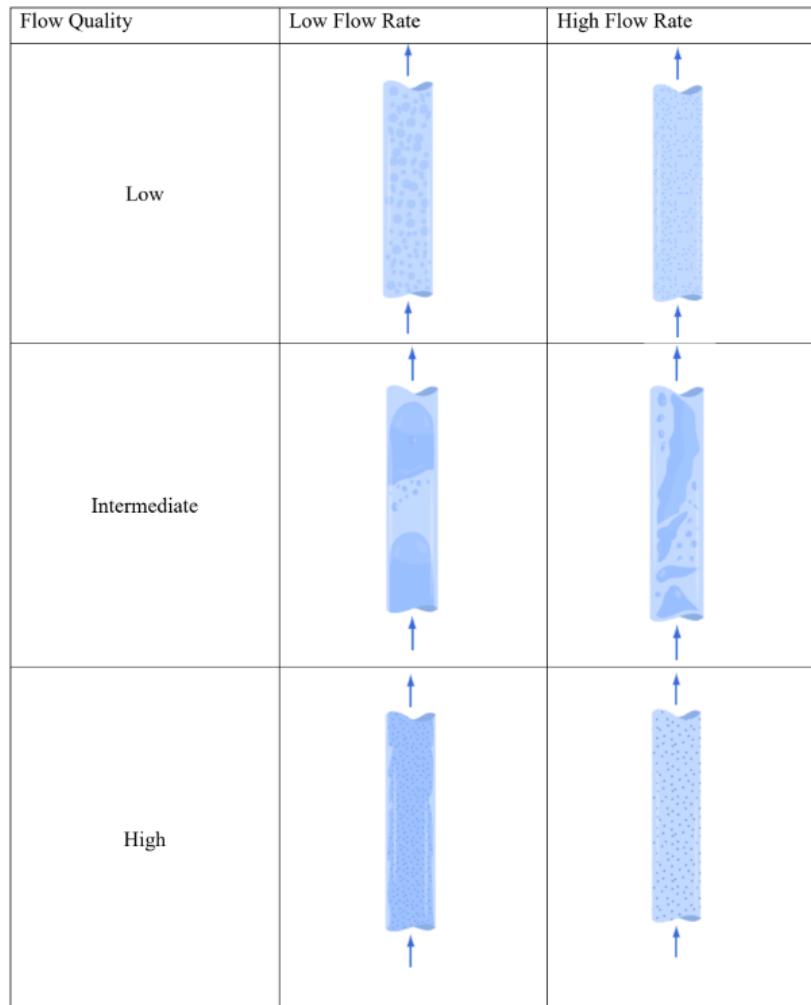


Figure 93: Flow Regime Map.

For systems where complex two-phase flow behavior is expected, like a boiling water reactor, a flow regime map can be made as shown in Figure 94 where an increase in the vertical axis is an increase in vapor mass flow rate; and an increase on the horizontal axis corresponds with increasing liquid mass flow rate.

A WIDE RANGE of heat transfer phenomena occur in two-phase systems over the range of two-phase flow regimes. These are illustrated graphically in Figure 95.

- Single-phase sub-cooled liquid enters at the bottom of the heated channel. For sub-cooled liquids, equilibrium quality is negative.
- As the fluid approaches saturation temperature and $-0.25 < x_{eq} < 0$ vapor bubbles begin to form at nucleation sites along the wall; these bubbles are swept away from the wall and ultimately collapse. This is referred to as *sub-cooled nucleate boiling*.
- Note that sub-cooled nucleate boiling does not occur until *after* the wall temperature has exceeded the fluid saturation temperature.
- As equilibrium quality increases above 0, void fraction increases and the two-phase flow regime begins to change also, transitioning from bubbly flow, to plug or slug flow depending on the mass flow rate. Boiling water reactors can enter into this phase of two-phase heat transfer.
- Mist flow begins at approximately $x_{eq} > 0.9$.

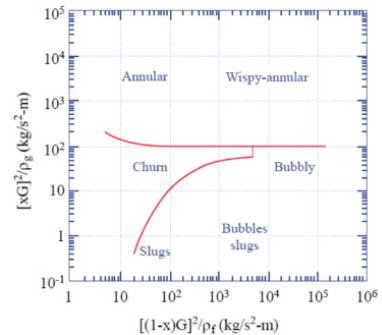


Figure 94: Flow regime map for a two-phase flow system.

Note: the bubbles in sub-cooled nucleate boiling carry a great deal of energy with them. This mode of heat transfer is both desirable and prevalent within pressurized water reactors. The ability to predict the onset of this behavior and quantitatively model the improvement in heat transfer is one of the main incentives for discussing two-phase flow in this course.

Applications for Nuclear Engineering

Some relevant applications of two-phase flow and heat transfer in the engineering of nuclear reactors for power production are listed below. This is by no means meant to be all-inclusive.

- **Boiling water reactor core.** Two-phase flow exists for BWRs in normal operation. Sub-cooled liquid ($x_{eq} < 0$) enters the bottom of the reactor core. The core exit condition is a saturated mixture with $x_{eq} \approx 12 - 15\%$.
- **Steam generators for a PWR.** Fluid conditions around the u-tubes for a recirculating SG of a PWR are similar to a BWR. Sub-cooled liquid comes in and a saturated mixture of low quality is fed into the moisture separator and steam dryer apparatus.

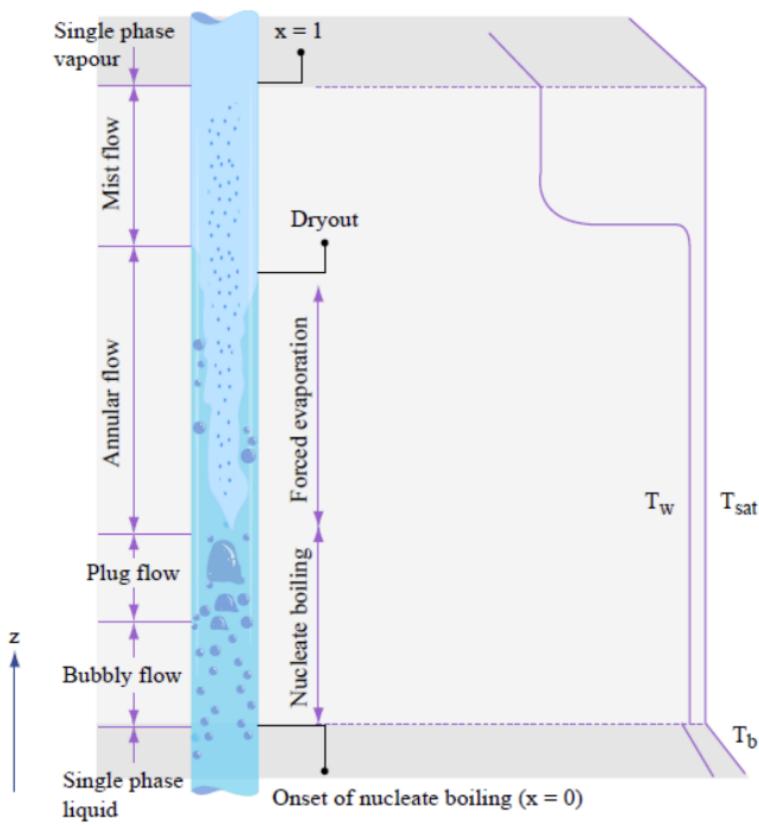


Figure 95: Two phase heat transfer regimes.

- **PWR core.** Sub-cooled liquid enters a PWR; the high temperature water exiting a PWR core is also sub-cooled but its void fraction is positive. As x_{eq} approaches zero and as the fuel clad outer temperatures exceed local saturation temperature, nucleate boiling begins. Nucleate boiling increases in intensity near the core exit.
- **Reactor accident analysis.** The transition from various regimes of two-phase heat transfer to critical heat flux is of great interest in the accident modeling community. For PWRs the relevant mechanism is DNB; for BWRs the relevant mechanism is Dry-out. These phenomena are a function of local heat flux as well as local hydraulic conditions. If we can predict the heat flux at which CHF is likely (q''_{CHF}) then we can measure the "Departure from Nucleate Boiling Ratio" (DNBR) as q''_{CHF}/q'' which, to the extent we can estimate its value accurately, we always want to maintain above unity.
- **NSSS Heat exchangers.** Closed Feedwater Heaters, Reheaters, and Condensers are all elements of the nuclear steam supply system for a PWR based on the Rankine cycle and all involve extensive two-phase heat transfer phenomena.
- **PWR Pressurizer.** The pressurizer in a PWR during normal operations maintains a two-phase system of saturated liquid and saturated vapor in a low/no-flow configuration.

NSSS stands for *nuclear steam supply system*.

Lecture 27 - Sub-cooled Nucleate Boiling

Objectives

The objectives of this lecture are:

1. Describe bulk coolant properties in a PWR fuel assembly in the context of two-phase flow.
2. Describe correlations to predict the onset of sub-cooled nucleate boiling and convective heat transfer coefficients for sub-cooled nucleate boiling.
3. Show an example calculation (in MATLAB) for an AP1000.

Heat Flux and Temperature in a PWR Fuel Channel

To simplify the discussion, we will consider a representative fuel channel in a PWR similar to what is depicted in Figure 96. If, for the time being, we simplify further by assuming that the power generated within the pin, and thus the heat flux at the pin boundary, is constant. We might expect a flow and temperature profile like what is depicted in Figure 97.

SOME OBSERVATIONS you might make about this flow configuration:

- Sub cooled liquid is assumed to enter the the channel and heat transfer initially is via sub cooled forced convection.
- In the sub cooled forced convection region, the ΔT between the coolant and the cladding outer surface $T_{co} - T_b$ is roughly constant since the heat flux and linear heat rate are constant.
- The temperature of the coolant increases due to the accumulated energy transferred from the fuel pin to the coolant, so both T_b and T_{co} increase more-or-less linearly.
- When T_{co} gets above T_{sat} , voids begin to form along the fuel clad outer surface. The process of this bubble formation will be discussed in the next section, but bubbles formed at *nucleation sites*

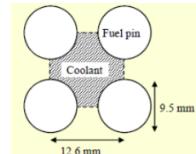


Figure 96: Representative channel in a PWR.

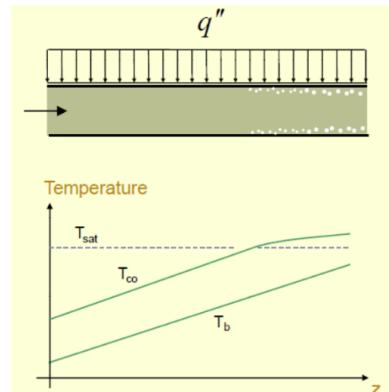


Figure 97: Temperature and heat flux profile for simplified channel.

result in enhanced convective heat transfer between the clad surface and the bulk coolant. When this occurs, ΔT between the clad and coolant starts to decrease.

- As sub cooled nucleate boiling gets more intense, the heat transfer between the clad and coolant continues to improve and the ΔT continues to get smaller. This explains the shape at the right-end of Figure 97.

OF COURSE a realistic heat flux profile would not be flat. A better approximation would be to use a cosine-shaped flux profile. If we assume this heat flux, we can, using tools already presented, estimate bulk coolant and wall temperature using the Presser correlation for convective heat transfer coefficient in a typical AP1000 fuel assembly. This is presented in Figure 98.

Sub-cooled Nucleate Boiling

A phenomena that we have not taken into account is the possibility of sub-cooled nucleate boiling. For the heat flux profile and convective heat transfer correlations that we have been using, the wall temperature eventually exceeds saturation temperature for the nominal pressure.⁸² If we superimpose the nominal saturation temperature on a plot of T_{wall} we get the result shown in Figure 99.

A schematic of the process of sub-cooled nucleate boiling is shown in Figure 100. When wall temperature exceeds saturation temperature, local voids of saturated vapor can form at nucleation sites, which are small pores, cracks or scratches in the clad surface. Since the wall temperature exceeds saturation temperature, the pressure of the vapor in the bubble exceeds nominal static pressure; that superheat is needed so the internal bubble pressure can overcome static pressure plus the surface tension of the bubble. As wall temperature increases locally, the vapor bubble can grow. Ultimately the vapor bubble is drawn away from the wall, at least in part, due to forces exerted by the sub-cooled liquid flowing along the surface.

Correlations to Predict Onset of Sub-cooled Nucleate Boiling

We will present two correlations to predict the onset of nucleate boiling. The Jens-Lottes correlation is given in USCS units in Equation 89.

$$T_{\text{wall,ONB}}(z) = T_{\text{sat}} + 1.897e^{-P/900} (q''(z))^{0.25} \quad (89)$$

where $T_{\text{wall,ONB}}$ and T_{sat} are in °F, P is in psia, $500 < P < 2250$, q'' is in BTU/hr-ft², and z is axial position in feet.

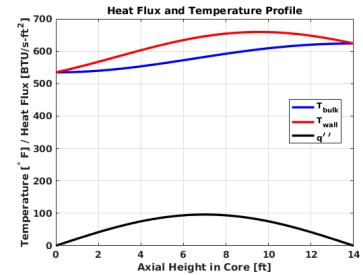


Figure 98: Temperature in a channel with cosine flux distribution using Presser correlation.

Note: see how the maximum clad temperature (T_{wall}) is reached in the upper half of, but not the top of, the core.

⁸² Please note that this demonstration has not taken into account axial variation in local pressure due to hydrostatic effects or due to major or minor head losses.

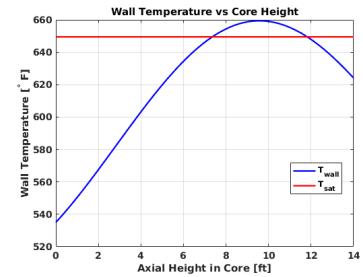


Figure 99: Wall temperature compared to saturation temperature in an AP1000 channel.

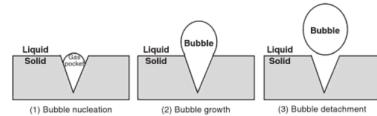


Figure 100: Schematic of the sub-cooled nucleation process.

Jens-Lottes in SI units is:

$$T_{\text{wall,ONB}}(z) = T_{\text{sat}} + 45e^{-P/62} (q''(z))^{0.25} \quad (90)$$

where T is in C° , P is in bar, q'' is in MW/m^2 , and z is axial position in meters.

The Thom correlation can also be used. In USCS units it is as shown in Equation 91.

$$T_{\text{wall,ONB}}(z) = T_{\text{sat}} + 0.072e^{-P/630} (q''(z))^{0.25} \quad (91)$$

USING BOTH the Thom and Jens-Lottes correlation for this example, the predicted temperature for onset of nucleate boiling is shown in Figure 101. Where the wall temperature is greater than $T_{\text{wall,ONB}}$, we predict nucleate boiling to take place.

Correlation for Sub-cooled Nucleate Boiling

Where we expect nucleate boiling to take place, we also expect enhanced convective heat transfer. This effect is not captured in any of the other correlations that we have presented. We will use the Bernath correlation for convective heat transfer in the sub-cooled nucleate boiling regime, which is given by Equation 92.

$$h_{\text{NB}} = 10,890 \left(\frac{D_e}{D'_e} \right) + \frac{48v}{D_e^{0.6}} \quad (92)$$

where D_e is the usual hydraulic diameter, $D_e = 4A_{\text{flow}}/P_{\text{wetted}}$; $D'_e = D_e + P_{\text{heated}}/\pi$, and v is average velocity in feet per second.

The complete code for calculations presented in this lecture is provided in the appendix.

Considerations for More Detailed Analysis

This lecture dealt with a highly simplified thermal model of a typical cooling channel in a PWR fuel assembly. Some thoughts are presented below as to how this analysis could be improved.

- *Obtain a more accurate axial power distribution.* The actual power distribution in a PWR core may not be cosine-shaped but instead will normally be skewed towards the bottom of the core due to coolant temperature/density variation and the feedback that has on core physics and power generation. For PWRs that use boric acid for reactivity control, boron may be preferentially deposited at sites where nucleate boiling occurs; this can result in the power distribution to be further shifted towards the lower portions of the

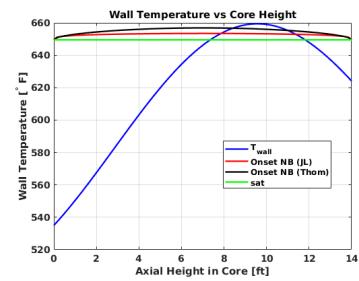


Figure 101: Prediction of onset of nucleate boiling using Jens-Lottes and Thom correlation.

core. These changes will impact the core height at which nucleate boiling can be expected to occur; they may also result in higher-than-expected fuel and clad temperatures than would otherwise occur.

- *Account for variation in cooling channel inlet pressure and velocity.* For typical PWRs operating today, the flow-path required for water coming in from the cold leg results in wide variation in the inlet pressure and velocity for cooling channels. This can result in some channels having lower inlet pressure and lower inlet mass flow rate than some other channels. While much of this effect is mitigated by mixing between adjacent channels it reduces the accuracy of simplified analysis of this sort.
- *A better analysis would take into account the variation of pressure between channel inlet and outlet.* For vertically oriented reactor cores, static pressure is reduced near the core outlet due to hydrostatic pressure variation. No matter how the core is oriented, major and minor head losses will result in a reduction in static pressure as the coolant flows through the core. Both of these effects would result in lower saturation temperature; nucleate boiling as well as other two-phase flow phenomena would begin lower in the core than this analysis would predict and overall the core would be closer to thermal limits than this analysis assumes.
- *Account for hydraulic and thermal communication between channels.* In a real LWR core, coolant mixes between adjacent channels carrying thermal and flow energy with it.
- *Use more correlations.* In this lecture I have presented only a subset of all available correlations for convective heat transfer and onset of nucleate boiling. For a novel reactor design it is almost sure to be the case that extensive experiments will need to be done to evaluate actual core thermal performance under safe conditions. Modern computer codes used in the nuclear industry capture all of these considerations and more. Sadly, their theory and use is currently beyond the scope of this course.

Lecture 28 - Departure from Nucleate Boiling in a PWR Core

Objectives

The objectives of this lecture are:

1. Qualitatively review DNB as a mechanism for CHF in a PWR core.
2. List some important correlations used to predict DNB.
3. Demonstrate the use of the Tong 68 correlation for DNB and illustrate variation with important parameters.

Departure from Nucleate Boiling

CONSIDER AGAIN the simplified channel with constant heat flux along the wall, this time, as pictured in Figure 102.

- Sub-cooled liquid enters the channel but as bulk and wall temperature increases, and wall temperature exceeds saturation temperature for the coolant, nucleate boiling begins.
- The vapor bubbles formed during nucleate boiling are swept away from the wall and collapse within the sub-cooled bulk coolant but, since coolant equilibrium quality continues to increase, the rate of bubble formation continues to increase.
- If heat flux is high enough and bulk fluid conditions permit, it may become the case that bubbles are formed at nucleation sites more rapidly than they are swept away in the coolant.
- Ultimately the wall becomes coated with a layer of vapor bubbles which impedes heat transfer; wall temperature increases rapidly which further contributes to vapor bubble formation along the wall and further degradation of convective heat transfer.

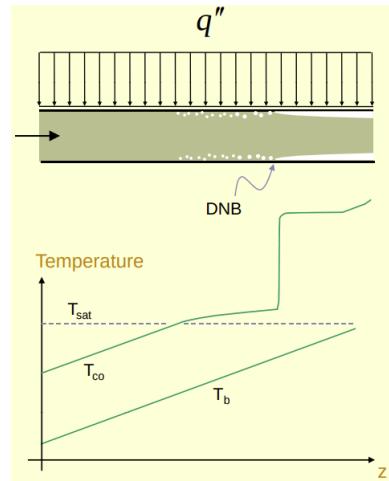


Figure 102: Simplified channel schematic illustrating departure from nucleate boiling.

- This last state is referred to as *departure from nucleate boiling*, or DNB.

THE PHENOMENA OF DNB depends on heat flux, local bulk temperature, pressure, and mass flux. For a more realistic cosine-shaped flux profile we might compare the local heat flux to the heat flux that would result in DNB as illustrated in Figure 103. As with many of the other thermal and hydraulic phenomena covered in this course, we do not have an equation or criteria based on first principles by which to judge the onset of DNB; we require a correlation. Thermal-hydraulic analysis codes incorporate several of these correlations; in this course I will present only one.

Tong 68 DNB Correlation

The Tong DNB correlation is given by Equations 93 and 94

$$k_{\text{Tong}} = 1.76 - 7.433x_{eq} + 12.222x_{eq}^2 \quad (93)$$

$$q''_{\text{DNB}} = k_{\text{Tong}} \frac{G^{0.4} \mu_f^{0.6} h_{fg}}{D_e^{0.6}} \quad (94)$$

where x_{eq} is equilibrium quality, G is mass flux, μ_f is the viscosity of saturated liquid, and h_{fg} is the difference in enthalpy between saturated vapor and saturated liquid at the local pressure. For a nominal cosine-shaped flux distribution, one can calculate channel temperature, enthalpy, and equilibrium quality within the channel and, using the Tong correlation, calculate q''_{DNB} . This is shown in Figure 104. If we further compare q''_{DNB} with the heat flux at that location in the channel, we can calculate the *departure from nucleate boiling ratio* (DNBR). This is shown for a portion of the channel in Figure 105. If $MDNBR \leq 1$, then the correlation predicts DNB will occur. If $MDNBR > 1$ then some margin of safety exists.

Variation with Plant Parameters

THE CHANGES OF DNB and DNBR with plant parameters may be intuitively clear to most readers. Nonetheless it is worthwhile, I think, to do some simple calculations to observe quantitatively how variations in plant parameters affect DNB and DNBR.

- Reduction in mass flow rate.* Reducing mass flow rate to a channel means that the channel temperature, enthalpy, and equilibrium

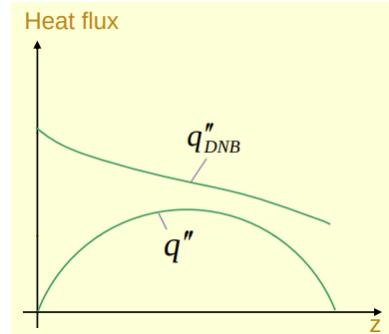


Figure 103: Simple schematic relating local heat flux to q''_{DNB} .

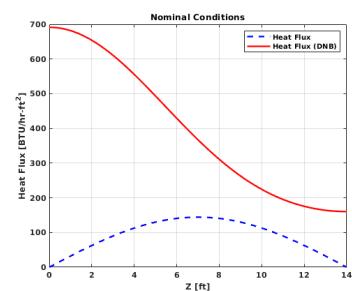


Figure 104: Calculation of DNB with Tong correlation.

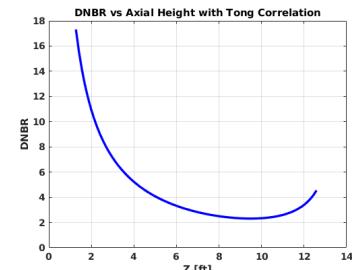


Figure 105: Calculation of DNBR with Tong correlation.

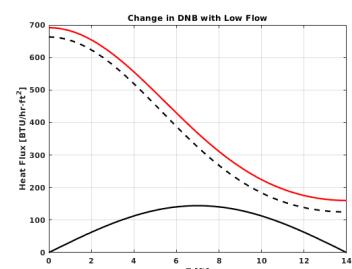


Figure 106: Effect of coolant flow rate reduction on DNB.

quality will increase more rapidly within the channel. Onset of nucleate boiling will occur lower in the channel and q''_{DNB} should be lower everywhere in the channel. Figure 106 illustrates this effect quantitatively.

- *Reduction in system pressure.* Reducing system pressure brings the coolant closer to saturation temperature and increases equilibrium quality everywhere. Figure 107 illustrates this.
- *Increase coolant inlet temperature.* Again, qualitatively this effect should be quite obvious. Figure 108 shows the result quantitatively.
- *Increase power.* Core power level has a strong influence on DNB and MDNBR; increasing power, because of its effect on temperature and equilibrium quality of the coolant, reduces q''_{DNB} while, since q'' is higher, further reduces DNBR. These effects are illustrated in Figure 109.

Application of Safety Factors for MDNBR

WHAT MAY NOT have yet been made explicitly clear, is that when DNB occurs, we expect fuel thermal limits to be exceeded and fuel failure to occur. As the thermal designer of a nuclear reactor core, it is your job to prevent this. You must ensure that DNB does not happen: not for the hottest channel in the core; despite uncertainties inherent in whatever DNB correlation you use; despite even uncertainties in plant parameters themselves; you must show that DNB does not occur during expected transients; if an accident happens, and such an accident is foreseeable, your design must make sure that DNB does not occur even under those accident conditions. In short: you need to add margins to your design to account for a great deal of uncertainty.

A SCHEMATIC of how you might account for all of the above variability and uncertainty is illustrated in Figure 110. I conclude this lecture with a brief description of these various safety factors that should be considered. For a more detailed discussion of this process, with application to an integral PWR, I invite you to peruse a relevant thesis.⁸³

- *Correlation limit.* In order to provide margin to DNB, we will demand that the MDNBR be greater than 1 by some amount. This is to account for uncertainties in the DNB correlation itself. The correlation may have been developed with fuel assemblies of a

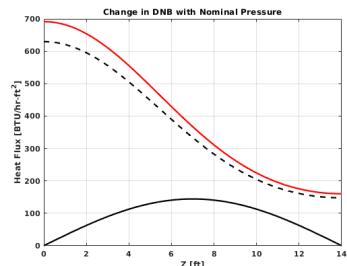


Figure 107: Effect of coolant pressure reduction on DNB.

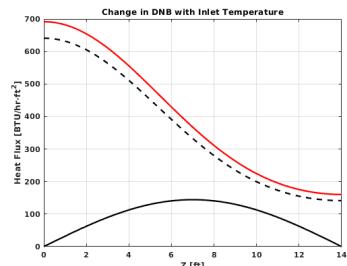


Figure 108: Effect of increasing coolant inlet temperature on DNB.

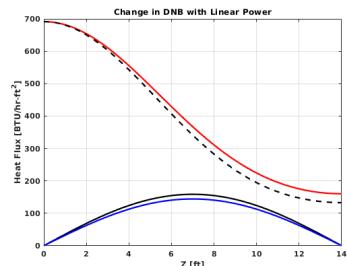


Figure 109: Effect of increasing core power on DNB.

⁸³ S R Blair. Thermal hydraulic performance analysis of a small integral pressurized water reactor core. Master's thesis, Massachusetts Institute of Technology, 2003

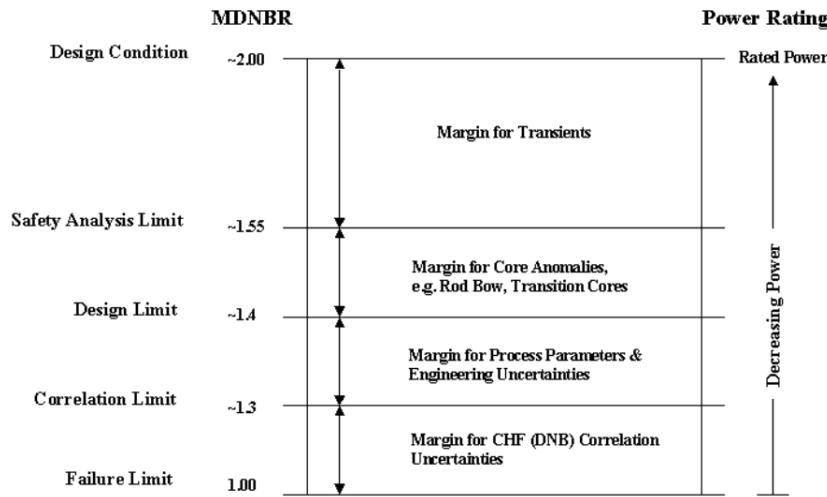


Figure 110: Safety factors to ensure DNB does not occur.

different pitch-to-diameter ratio in mind, or the correlation may have assumed that mixing grids of a certain design were employed which, in your design, perhaps they were not, or mixing grids of less effectiveness were used instead. For these reasons and more, margin should be provided just to account for correlation uncertainties.

- *Design limit.* The actual reactor that gets built will be only an approximation of the reactor that exists on design drawings and is represented in computer codes. Machine tolerances will impact actual fuel pellet diameter, cladding inside radius and thickness, reactor coolant pump volumetric flow rate, major and minor head losses: everything that you assume in your analysis can be expected to be somewhat different in practice due to manufacturing uncertainties. Margin should be provided to account for this.
- *Safety analysis limit.* During operations—both steady state and transient; normal operations and accident conditions—the plant will respond in ways that your analysis did not anticipate. Within reasonable limits, additional margin should be provided to account for this.
- *Design condition.* Besides the variations and transients that you *do not* expect, you must add margin to account for the transients you *do* expect. These margins can be quite large and rely extensively on detailed system analysis codes to allow you to estimate the impact of design-basis transients such as loss of cooling, loss of flow, uncontrolled rod withdrawal among many others.

In the end, the thermal-hydraulic designer will determine the power level, assumed power profile, and core geometry such that the core will be “safe” while accounting for all of the aforementioned uncertainties. If the *reactor physicist* ensures the reactor will go critical and produces the required amount of power for the desired length of time, the *reactor engineer* or *thermal-hydraulicist* will ensure ensure the core can be safely cooled and all materials will remain within their thermal limits under all conditions.

Part IV

Non-Steady-State Analysis

Lecture 29 - Loss of Coolant Accident

Objectives

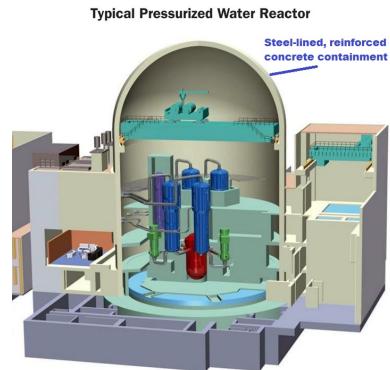
The objectives of this lecture are:

1. Review containment design requirements
2. Discuss factors to consider for coolant system ruptures as it applies to containment design
3. Describe the control volume approach to LOCA containment analysis

Containment Design Requirements

The containment for a nuclear reactor has several functions.

- *Public & Environmental Protection.* As the name “containment” implies, one of its chief functional requirements is to prevent release of radioactivity to the environment. In the event that a radiological release becomes inevitable, a secondary requirement is to prevent the *large early* release of radioactivity. If the release is not large or if it comes with ample warning, emergency planning measures can be put into effect to minimize damage and danger to people.
- *Protection of Plant Systems.* Nuclear power plant systems are not especially vulnerable and yet, like all industrial equipment, they must be protected from the elements under ordinary circumstances; protected from fires and floods in unusual circumstances. After the 9/11 attacks, additional emphasis was added to protection against human actions: crashing airplanes, explosives, and acts of sabotage. Whatever the threat, the stakes of damage to nuclear power plant systems is somewhat higher than a typical industrial building so additional design features—or at least additional design analysis—to ensure the nuclear systems have protection against foreseeable natural and man-made hazards are needed.



Source: U.S. Nuclear Regulatory Commission

Figure 111: Containment for typical PWR.

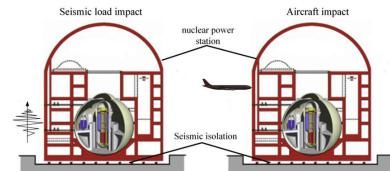


Figure 112: Containments must withstand seismic and impact loads.

- *Structural Support of Systems.* These loads include:
 1. **routine service loads:** Structural support of the reactor vessel and primary coolant systems. Lifting and handling equipment must be installed to support refueling and expected maintenance activities.
 2. **seismic loads:** Depending on geographic location, seismic loads may or may not qualify as “routine”, but earthquakes small as well as large must be accounted for in the design of the reactor containment as well as nuclear system support structures.
 3. **Internal loads during design basis accidents.** Perhaps this is a repeat of the “containment” function, but it is worth emphasizing that design basis accidents like loss of coolant accident (LOCA) and steam line rupture (SLR) accidents result in significant structural and thermal loads on the containment boundary that the containment must be designed to withstand.

Our current focus for this lecture is to quantify the consequences of a LOCA from a containment design perspective.

LOCA Containment Analysis

IN THE EVENT that there is a large loss of coolant accident, the engineer must ensure that the containment structure will be able to withstand the temperature, pressure, and, in the case of water-cooled reactor LOCA, humidity of the post-accident environment in the containment.

Some factors to consider in this analysis include:

- *Possible heat sources.* These heat sources include:
 1. **Stored heat in the reactor system fluids.** For a PWR the primary coolant system is at high temperature and pressure. After the LOCA, one can expect the mass/energy of the primary coolant to expand into the entire containment system.
 2. **Decay heat.** Even if control rods, or other reactivity control systems, are successfully employed to promptly shut down the reactor, decay heat will remain. The magnitude of the decay heat will depend on the operating history of the reactor but this significant energy source must be included in any realistic analysis.
 3. **Exothermic chemical reactions.** These reactions include:

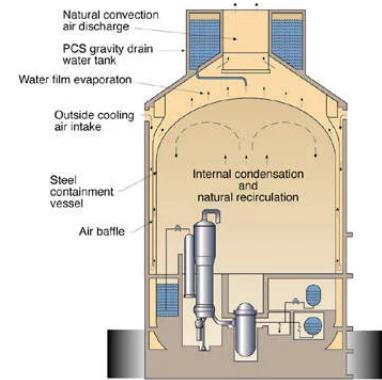


Figure 113: AP1000 containment schematic.

- (a) for LWRs, high-temperature oxidation of Zircaloy cladding in the presence of water. In addition to releasing heat, hydrogen is a product of this reaction.
 - (b) for sodium-cooled reactors: oxidation of the sodium coolant in the presence of air and/or water.
 - (c) for carbon-dioxide-cooled reactors: oxidation of the graphite moderator by the carbon dioxide coolant.
 - (d) for all reactors: oxidation of metals that may exist in molten core material ("corium") by water and carbon dioxide released from thermal decomposition of the concrete containment basemat on contact with the corium.
- *Possible heat sinks.* These include:
 1. Containment walls and other surfaces that are able to store energy.
 2. active containment heat/pressure mitigation systems such as:
 - (a) air coolers
 - (b) sprays
 - (c) endothermic chemical reactants
 - (d) built-in emergency cooling systems
 - *Fluids added from external sources.* Perhaps this can be categorized under "heat sinks," but the containment design analysis might consider the mitigating effects of providing additional cooling fluids into the containment environment after the accident has occurred. These might include additional core cooling water or, in the case of a BWR, additional feedwater.

Analysis Approach for LOCA

LET US CONSIDER the containment and everything inside of it as a fixed control volume. Upon initiation of the LOCA, the high pressure and temperature fluids inside the reactor coolant system interacts with and, eventually come into mechanical and thermal equilibrium with the air in the containment.

If we consider this as any other thermodynamic process between an initial and final state point we might first write down a statement of conservation of energy.

$$\begin{aligned} m_{\text{air}} u_{\text{air}_1} + m_w u_{w_1} + m_{w,\text{air}} u_{w,\text{air}_1} + Q_{1 \rightarrow 2} = \\ m_{\text{air}} u_{\text{air}_2} + m_{w,\text{air}} u_{w,\text{air}_2} + m_w u_{w_2} \quad (95) \end{aligned}$$

where m_{air} is the mass of air in the containment, m_w is the mass of water in the primary coolant system, $m_{w,\text{air}}$ is the mass of water in the air in the containment,⁸⁴ and $Q_{1 \rightarrow 2}$ is the heat that passes the control volume boundaries during the process. The temperature-entropy plot for the water in shown in Figure 114. The temperature-entropy plot for air is shown in Figure 115.

In addition to the energy equation provided in Equation 95, we will assume that the overall containment volume will remain fixed:

$$V_{\text{final}} = V_{\text{reactor coolant}} + V_{\text{containment}} \quad (96)$$

and we will assume the final static pressure to be equal to the partial pressure due to water (or water vapor) plus the pressure of the air:

$$P_{\text{final}} = P_{\text{air,final}} + P_{\text{water,final}} \quad (97)$$

Lastly we will assume that the air and water within the containment volume come to thermal equilibrium:

$$T_{\text{air,final}} = T_{\text{water,final}} \quad (98)$$

Basic Analysis Method

In this section, we will consider the case where humidity in the containment air is not considered. The following steps are taken:

1. Determine the initial mass of water in the primary and mass of air in the containment volume. There are a number of ways the problem can be posed, but let us assume that these values are either given or can be easily computed with given data.
2. Calculate the *final* specific volume of water in the containment volume. We will assume that mass does not cross the system boundaries—i.e. no water or air is allowed in or out of the containment boundary. We have the water mass from the previous step; we get the final volume from Equation 97, so we can compute the specific volume.
3. Begin an iterative process to find the final temperature:
 - (a) Guess final temperature.
 - (b) With this trial temperature and the (constant) specific volume of the water, we can determine the pressure and internal energy of the water.
 - (c) With the same trial temperature, compute the pressure and internal energy of the air.

⁸⁴ We assume that the air in the coolant initially has some humidity. Admittedly, this is a small detail and does not greatly impact the results of the analysis, but it is included in the treatment by Todreas, so I will include it here.

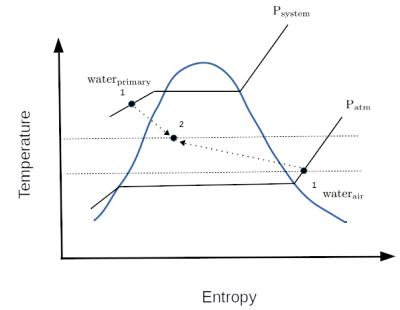


Figure 114: Temperature-Entropy plot for Water in a LOCA.

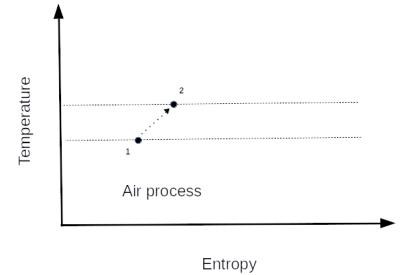


Figure 115: Temperature-Entropy plot for Air in a LOCA.

- (d) Check the energy balance given by Equation 95. Note that this steps requires that we specify how much energy is transferred in or out of the system. As a starting point, it is often useful/reasonable to assume no energy enters or leaves the system.⁸⁵
- (e) Adjust your temperature guess based on the energy balance and repeat until you converge to a temperature at which the energy balance is satisfied.

Complete MATLAB code to carry out this analysis is provided in the appendices.

Final Comments

THE POINT of this lecture is to help you answer, or at least partially answer, the questions: “*How big should my containment be?*” and “*How thick should my containment walls be?*” If we simply view the containment as a pressure vessel, sized so as to satisfy all of the non-accident functional requirements, we can use the final containment temperature and pressure to find necessary containment wall thickness. If we make the containment bigger, we expect a lower final pressure on a LOCA; if smaller: higher pressure and temperature after the LOCA. If we want to specify a maximum temperature or pressure post-LOCA, we can use the methods to determine how big the containment should be. This is a question that often gets left out of textbook treatments of this topic, but the tools to do at least a simple analysis are available to you; you should use them.

⁸⁵ A refinement would be to assume energy is added due to decay heat; one could then treat the problem as a quasi-steady-state system in which we find the “steady state” after, say 1 hour of decay heat addition; then another “steady state” after 2 hours (etc...). Another refinement would be to consider adding water from emergency core cooling systems to the system; yet another might consider energy dissipation due to endothermic chemical reactions or heat transfer through the containment boundary.

Assignment #9: PWR Loss of Coolant Containment Analysis

Background

To prevent the release of radioactive fission products to the environment, reactor systems are protected with multiple barriers. For light water reactors (LWRs) these barriers include the fuel material, the cladding around the fuel pins along with the reactor coolant system piping itself. Since, for LWRs, the reactor coolant system is highly pressurized, it is possible that radioactive materials may not be fully controlled due to failure of the reactor coolant pressure boundary. For this reason, the reactor coolant systems for LWRs are surrounded by a larger *containment building*.

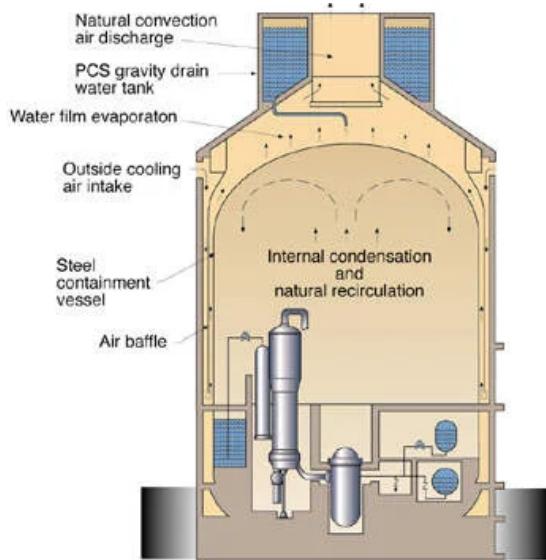


Figure 116: AP1000 containment building

WHAT IS NOT so clear from the figure above is that the containment vessel must serve purposes beyond that strictly required for exclusion of fission products from the environment; the containment must be large enough to accommodate refueling equipment and allow for accessibility of major reactor plant components among other design requirements.

IN THIS WORKSHOP we will practice the skills that will allow us to answer the question: if all of the pri-

mass of coolant were released from the reactor coolant system into the containment; what would the resulting temperature and pressure be? Of course to do this we will need to know some details about the design parameters of the primary coolant system and the containment; we will also need to make some reasonable assumptions to simplify the analysis.

Initially, we will only require knowledge of the temperature, pressure and volume of the reactor coolant system and containment vessel. We will assume that the air in the containment has negligible moisture content. The primary coolant escaping from the reactor coolant system will expand to fill the total volume—volume of reactor coolant system plus the volume of the containment—and come into thermal equilibrium with the air. We will assume that this process happens instantaneously and that there is no further heat input from the core (due, for example, to decay heat) or energy lost from the system due to heat transfer through the containment walls.

The conservation of energy equation for these assumptions reduces to the following:

$$m_w (u_{w2} - u_{w1}) + m_a (u_{a2} - u_{a1}) = Q_{1 \rightarrow 2}$$

We will assume that the total system volume remains constant:

$$V_{\text{final}} = V_{\text{containment}} + V_{\text{reactor coolant}}$$

where initially the air occupies $V_{\text{containment}}$ and the reactor coolant occupies $V_{\text{reactor coolant}}$. The total system pressure is the sum of the partial pressures of the air and the primary coolant at the final temperature

$$P_{\text{final}} = P_{\text{air,final}} + P_{\text{water,final}}$$

with the air and water in thermal equilibrium—i.e. $T_{\text{air,final}} = T_{\text{water,final}}$.

Part 1

We will carry out the analysis to determine final containment volume pressure and temperature following a loss of coolant accident. Relevant design information is provided in the table below.

Component	Volume (m ³)	Pressure (kPa)	Temperature (°C)
Reactor coolant system	354	15,500	340
Containment vessel air	50,970	101	27

CONSIDER A LOSS of coolant accident where there is no heat added (for example, due to decay heat) and no heat lost (for example, via the containment shell to the atmosphere). Answer the questions below:

1. What is the final containment temperature? [°C]

2. What is the partial pressure of water vapor in the containment? [kPa]

3. What is the partial pressure of air in the containment? [kPa]

Part 2

In the previous section we assumed that there would be no further heat input following the LOCA. Due to the presence of decay heat, this assumption is both incorrect and non-conservative. Over time, we can expect the pressure and temperature within the containment to increase due to heat input from the decay of fission products. We also assumed for simplicity that the structure and shielding components within the containment volume have negligible heat capacity and that there are no other sources of water to cool the core in the case of an accident. For this part of the assignment we will take into account heat input due to decay heat. We will also account for the presence of emergency water sources such as the core make-up tanks and the irradiated water storage tank.

Component	Volume (m ³)	Pressure (kPa)	Temperature (°C)
Irradiated water storage tank	2,070	101	27
Core make-up tanks	141	101	27

The heat input due to decay heat can be approximated as follows:

$$P(t) = 0.066 P_o \left[t^{-0.2} - (t + t_s)^{-2} \right]$$

where P_0 is the average power at which the reactor was operating prior to the accident, t is the time since shutdown (in seconds, assuming the reactor was shutdown at the beginning of the LOCA); and t_s is the duration of time that the reactor was previously operating before the shutdown (in seconds). For this problem assume that the reactor was operating at 3400 MW_{th} for a full year prior to the loss of coolant accident.

4. What is the final containment pressure and temperature at the end of one day?

 5. Make a plot of containment pressure and temperature from one second after the LOCA through one full day.

Lecture 30 - Pressurizer Sizing Analysis

Objectives

The objectives of this lecture are:

1. Discuss the design basis for a typical PWR Pressurizer
2. Describe an analysis method for pressurizer sizing

Pressurizer Design Basis

In a PWR, the pressurizer serves as an expansion volume for the primary coolant. This is required since the water is essentially incompressible, and yet its density changes as a function of temperature. Without the pressurizer, the primary coolant system would be subject to rapid and uncontrollable pressure fluctuations in response to small temperature variations. A simplified schematic of the primary coolant system is provided for reference in Figure 117.

The pressurizer is maintained as a saturated system with a saturated steam bubble over a volume of saturated liquid water. Electric heaters are installed in the lower portion of the pressurizer and used to maintain a steam bubble by evaporating primary coolant and generally increasing the temperature of the pressurizer. A spray system, which comprises relatively cool primary coolant piped in from the cold leg downstream of the reactor coolant pumps, is used to rapidly condense steam in the bubble to reduce system pressure.

As a REVIEW let us consider the change of water level in the pressurizer in response to routine power level changes:

- **Up-Power Maneuver:** In this case, the thermal energy extracted by the steam generator increases ($\dot{Q}_{S/G} \uparrow$) so that, temporarily $\dot{Q}_{S/G} > \dot{Q}_{Rx}$. Since more energy is being extracted from primary coolant than is being added, $T_{avg} \downarrow$. For water, this means the average coolant density goes down, so the coolant volume contracts. This results an *outsurge* of water from the pressurizer, and

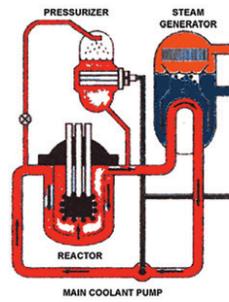


Figure 117: Simplified schematic of primary coolant system.

Question for discussion: How might you use the pressurizer system to perform a non-critical heat-up of the primary coolant system?

$L_{pqr} \downarrow$. The pressurizer is a fixed volume, however, if water level goes down the steam volume must expand; this lowers pressurizer pressure ($P_{pqr} \downarrow$). When this happens, saturated liquid in the pressurizer flashes to steam, mitigating the pressure drop. If pressure falls low enough, the heaters are energized to create still more steam and further mitigate the pressure drop.

- **Down-Power Maneuver:** For this case, the dynamics are the opposite. $\dot{Q}_{S/G} < \dot{Q}_{Rx}$ so $T_{avg} \uparrow$ resulting in an expansion of the primary coolant and $L_{pqr} \uparrow$. This *insurge* compresses the steam bubble resulting an increase in P_{pqr} . If pressure increases enough, spray flow will be engaged; relatively cool water partially condenses the vapor bubble thus reducing pressure. When pressure falls below a specified set-point, spray flow is shut off.

For both up-power and down-power maneuvers, temperature feedback mechanisms result—in the short term, anyway—in re-equilibrium of \dot{Q}_{Rx} and $\dot{Q}_{S/G}$ so that average coolant temperature and pressurizer water level ultimately return to their pre-transient level. Any water that is added or removed from the primary coolant—for example, by coolant discharge to remove water or by coolant charging to add coolant—would result in the pressurizer water level being increased or decreased respectively and this is undesirable. Ideally, the pressurizer would be large enough to accommodate these transients without the need to add or discharge coolant and needlessly change pressurizer water level.

THIS IS BASIC pressurizer dynamics. Further consideration reveals the following points:

1. It is important that the pressurizer heaters not be uncovered. If they are submerged in water, they can effectively transfer heat to water in the pressurizer and regulate pressure; if they are uncovered, their effectiveness is greatly reduced.
2. It is also important that the steam bubble never be fully collapsed. For very rapid pressure transients, relief valves (some located in the pressurizer steam space) would open. It is conceivable that a very rapid *insurge* could completely fill the pressurizer. When this happens, spray flow would no longer be effective in reducing system pressure.

THE OVERALL GOAL is to design the pressurizer so that, for “foreseeable transients,” the pressurizer is neither filled solid nor emptied to the point where the electric heaters would be uncovered. The design question is: *How big, then, should the pressurizer be?*

Clearly if the pressurizer is infinitely large, there will be no need to worry about filling solid or uncovering the heaters. On the other hand, infinitely large pressurizers are expensive so some reasonable compromise must be made.

AP1000 Pressurizer Design Basis

As an example, let us consider the design basis for the AP1000 pressurizer.⁸⁶

- The water volume is sufficient to prevent a reactor trip during a step-load increase of 10 percent of full power, with automatic reactor power.
- The water volume is sufficient to prevent uncovering the heaters following a reactor trip and turbine trip, with normal operation of control systems and no failures of nuclear steam supply systems.
- The steam volume is large enough to accommodate the surge resulting from a step load reduction from 100 percent power to house loads without reactor trip, assuming normal operation of control systems.
- The steam volume is large enough to prevent water relief through the safety valves following a complete loss of load with the high water level initiating a reactor trip, without steam dump.
- A low pressurizer pressure engineered safety features actuation signal will not be activated because of a reactor trip and turbine trip, assuming normal operation of control and makeup systems and no failures of the nuclear steam supply systems.

As a result of these design bases, the AP1000 pressurizer is roughly 40 percent larger than pressurizers installed on previous PWRs of similar power level.

Pressurizer Analysis Methodology

We will employ the simplified control volume approach described in the Todreas text section 7.3. The method calls for us to combine two separate analyses shown schematically in Figure 118.

1. **Insurge Analysis.** We will size the pressurizer such that there is sufficient *steam space* to accommodate a design-basis *insurge* without filling the pressurizer solid. This gives us a required volume of the steam space.

⁸⁶ This is extracted from the AP1000 Design Control Document, Revision 19, Tier 2, Chapter 5, section 5.4.5.1 available at: <https://www.nrc.gov/docs/ML1117/ML11171A454.pdf>

Question: What protective measure (influenced by the design of the pressurizer) would you expect to potentially cause a reactor trip due to the step-load increase?

Question: What protective measure (influenced by the design of the pressurizer) would you expect to potentially cause a reactor trip due to the step load decrease?

Question for discussion: What are the operational pros and cons of having a big or small pressurizer?

$$V_T = (V_{g_1})_{\text{insurge}} + (V_f)_{\text{outsurge}}$$

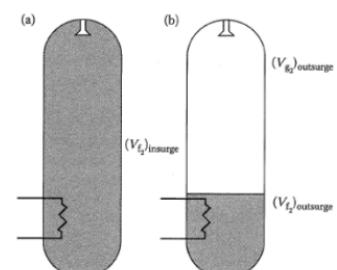


Figure 118: Schematic of insurge and outsurge final states for pressurizer size analysis.

2. **Outsurge Analysis.** We will find the initial water volume such that a design-basis outsurge can occur without uncovering the heaters. This gives us a required volume of the water-filled space.

The total pressurizer volume will be the sum of the steam volume and the water-filled volume obtained from those two separate analysis. The outsurge analysis will also provide an estimate of the required pressurizer heater capacity.

Disclaimer: the analysis method to follow contains significant, and in some ways unrealistic, assumptions as well as access to system response parameters—such as *mass of surged coolant* and *mass of sprayed coolant*—that would require detailed system modeling to obtain. Nonetheless, if one actually has access to such data, this method amounts to a way that you could use to obtain a preliminary estimate of pressurizer size. It is, perhaps, a little bit better than nothing.

Insurge

The design insurge will be modeled as a *constant pressure* process. A temperature-entropy plot is shown in Figure 119; a schematic of the control volume is given in Figure 120.

At the end of the transient, the entire pressurizer is filled with fluid, so the conservation of mass equation is given by:

$$m_{f_2} = m_{\text{surge}} + m_{\text{spray}} + m_{f_1} + m_{g_1} \quad (99)$$

where m_{f_2} and m_{f_1} is the mass of fluid at state point 1 and 2 respectively and m_{g_1} is the mass of vapor at state point 1. Let us define m_{spray} as equal to $f m_{\text{surge}}$ where f determined from system-wide modeling. Then Equation 99 becomes:

$$m_{f_2} = m_{\text{surge}}(1 + f) + m_{f_1} + m_{g_1} \quad (100)$$

THE PRESSURIZER will remain at constant volume throughout the transient, so we can apply a volume constraint as given by Equation 101.

$$m_{f_2} v_f = m_{g_1} v_g + m_{f_1} v_f = (m_{\text{surge}}(1 + f) + m_{f_1} + m_{g_1}) v_f \quad (101)$$

where v_f and v_g are the specific volume of liquid and vapor respectively. The right-hand side makes use of the value of m_{f_2} given by Equation 100.

We can combine the conservation of mass with the volume constraint to get an expression for m_{g_1} which is:

$$m_{g_1} = \frac{m_{\text{surge}}(1 + f)v_f}{v_g - v_f} \quad (102)$$

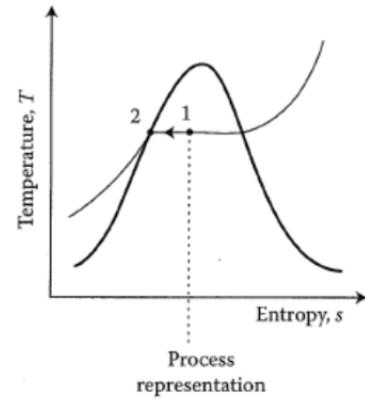


Figure 119: Temperature-entropy plot for design basis insurge.

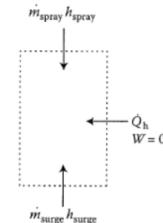


Figure 120: Control volume for design basis insurge.

With a value for m_{g_1} , we can directly find the required vapor volume for the design-basis insurge by applying the specific volume of vapor:

$$V_{g_1} = m_{g_1} v_g \quad (103)$$

SINCE ENERGY has to be conserved also, we need to calculate the required heater input to satisfy conservation of energy. The basic conservation of energy equation for this system reads:

$$m_{f_2} u_f = m_{\text{surge}}(h_{\text{surge}} + f h_{\text{spray}}) + m_{g_1} u_g + m_{f_1} u_f + Q_h \quad (104)$$

where Q_h is energy input from the heater.

Re-arranging Equation 104 to solve for m_{g_1} , combining with Equation 102, we can get an expression for \dot{Q}_h as given in Equation 105.

$$Q_h = \frac{m_{\text{surge}}(1+f) \left[v_g u_f - v_f u_g \right]}{v_g - v_f} - m_{\text{surge}}(h_{\text{surge}} + f h_{\text{spray}}) \quad (105)$$

Outsurge

The design basis insurge will also be modeled as a constant pressure process. A temperature-entropy plot is shown in Figure 121; a schematic of the control volume is given in Figure 122.

The conservation of mass equation for this outsurge is given as:

$$m_{f_2} + m_{g_2} = m_{f_1} + m_{g_1} - m_{\text{surge}} \quad (106)$$

where, even though it is an outsurge, m_{surge} is taken to be positive.

As with the insurge, we apply a constant volume constraint:

$$m_{f_2} v_f + m_{g_2} v_g = m_{f_1} v_f + m_{g_1} n u_g \quad (107)$$

Combining Equation 106 and Equation 107 and we get:

$$m_{f_1} = m_{f_2} + m_{\text{surge}} \frac{v_g}{v_g - v_f} \quad (108)$$

and, to get V_{f_1} :

$$V_{f_1} = m_{f_1} v_f \quad (109)$$

Considering conservation of energy, we get:

$$m_{f_2} u_f + m_{g_2} u_g - m_{f_1} u_f - m_{g_1} u_g = -m_{\text{surge}} h_{\text{surge}} + Q_h \quad (110)$$

Note: We will find that this last result of Q_h is not usually needed; it will provide the total heater power needed to ensure conservation of energy, but the corresponding result for the outsurge transient is expected to be much higher and set the real limit for what heater power capacity needs to be. Nonetheless it is worthwhile—particularly considering the gross assumptions needed such as coming up with a value of f , for example—to ensure, at least, that Q_h is a realistic non-negative number.

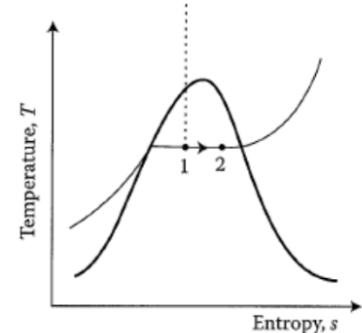
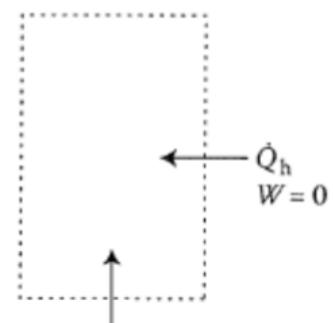


Figure 121: Temperature-entropy schematic for design-basis outsurge.



Through careful use of tedious algebra, we can solve for Q_h :

$$Q_h = m_{\text{surge}} h_{\text{surge}} - \left(u_f - \frac{v_f}{v_g} u_g \right) \left(m_{\text{surge}} \frac{v_g}{v_g - v_f} \right) \quad (111)$$

Summary

This analysis provided a method for estimating required pressurizer size. The initial vapor volume V_{g1} is obtained from Equation 103; the required initial liquid volume V_{f1} is obtained from Equation 109; the total pressurizer volume is their sum. The total energy input (during insurge or outsurge) is determined by the maximum value of Q_h from design-basis insurge or outsurge transients. For realistic system parameters, the Q_h from outsurge should be larger. Note, however, this is a heater energy input: we've said nothing about the time-scale for the insurge/outsurge transients so some additional analysis/estimates are needed to find required heater power. As mentioned in the disclaimer, it is hoped that this analysis is somewhat better than nothing and provides some insight into how one might go about determining a reasonable estimate of required pressurizer size.

Part V

Back Matter

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Appendices

Matlab Style Rules

1. **rule:** All scripts will start with the commands: `clear`, `clc`, and `close 'all'`
rationale: No script should depend upon any data visible in the MATLAB workspace when the script starts. By omitting these commands, residual data within the workspace may hide errors.
2. **rule:** Your code must be documented with enough details such that a reader unfamiliar with your work will know what you are doing.
rationale: Code documentation is a habit. For more significant projects readers may need help in deciding what the author of the code intended. For your own code, the most likely reader is you—a few months into the future.
3. **rule:** Function and variable names must be meaningful and reasonable in length.
rationale: Failing to do either make code harder to read and maintain.
4. **rule:** All outputs from the code must be meaningful; numbers should be formatted, part of a sentence, and include units. Graphs should be readable and axis labels should make sense and include units.
rationale: Code output is a form of communication. It is important that this communication be clear and unambiguous.
5. **rule:** Do not leave warnings from the Code Analyzer unaddressed.
rationale: Sometimes Code Analyzer warnings can be safely ignored. Most of the time the warning points to a stylistic error that would be unacceptable in software that you use. Occasionally these warnings are indicative of a hidden error.
6. **rule:** Use the “smart indentation tool” to format the indentation of your code.

rationale: This tool improves code readability. It will also occasionally point out errors that you did not see before.

7. **rule:** Pre-allocate arrays; if possible initialize with **NaN** values.

rationale: Pre-allocation improves performance and helps readability. Initialization with **NaN** helps avoid a range of potential logical errors.

8. **rule:** Avoid “magic numbers” — i.e. hard-coded constants.

rationale: Constants included in your code tend to hide your program logic. Also, “magic numbers” make code maintenance more difficult and error prone.

9. **rule:** Only write one statement per line.

rationale: Multi-statement-lines hurt code readability in almost all cases.

10. **rule:** Do not write excessively long lines of code; use the line continuation “...” and indentation to spread long expressions over several lines.

rationale: Following this rule improves code readability.

EasyProp Interface Description

Introduction

This appendix is provided as a overview of the most commonly used functions in the EasyProp interface.

In each section below, we will assume that the user has initialized an object in the MATLAB environment from which to obtain properties from EasyProp. As an example, the listing below initializes an object to obtain water properties in SI units.

To run this code, create a MATLAB script containing the code in the listing below. You should have a copy of the library EasyProp.py in the same folder.

```
%>> Prepare the Environment
2 clear
3 clc
4 close 'all'

6 EasyProp_path=''; %<-- enter relative or absolute path to folder containing EasyProp.py
if count(py.sys.path,EasyProp_path) == 0 %<-- see if desired directory is on path
    insert(py.sys.path,int32(o),EasyProp_path); %<-- if not; add it.
end

10 %% Initialize the fluid:
11 fluid = py.EasyProp.simpleFluid('Water','SI');
```

Thermodynamic Properties of Simple Fluids

Enthalpy

```
%>> Enthalpy - SI units: kJ/kg USCS units: BTU/lbm
2 h1 = fluid.hL_p(101.3); % saturated liquid enthalpy at 101.3 kPa
3 h2 = fluid.hV_p(8000.0); % saturated vapor enthalpy at 8000 kPa
4 h3 = fluid.h_ps(3500,1.25); % enthalpy at 3500 kPa, 1.25 kJ/kg-K
5 h4 = fluid.h_pT(101.3,200); % enthalpy at 101.3 kPa, 200 C
6 h5 = fluid.h_Tx(350,0.5); % enthalpy at 350 C, 50% quality
7 h6 = fluid.h_px(8000,0.25); % enthalpy at 8000 kPa, 25% quality
8 % also: h_pv(P,V)
```

Entropy

```

%% Entropy - SI units: kJ/kg-K USCS units: BTU/lbm-R
1 s1 = fluid.sL_p(101.3); % saturated liquid entropy at 101.3 kPa
2 s2 = fluid.sV_p(8000.0); % saturated vapor entropy at 8000 kPa
3 s3 = fluid.s_Tx(200,0.4); % saturated mixture entropy at 200 C, x=0.4
4 s4 = fluid.s_pT(101.3,200); % entropy at 101.3 kPa, 2000 C
5 s5 = fluid.s_ph(300.0,1234); % entropy at 300 kPa, h=1234 kJ/kg
6 s6 = fluid.s_px(300, 0.5); % entropy at 300 kPa, 50% quality
7 % also s_pu(P,U)
8

```

Temperature and Pressure

```

%% Temperature - SI units: C USCS units: F
1 T1 = fluid.T_ps(101.3,0.5); % temperature at 101.3 kPa, 0.5 kJ/kg-K
2 T2 = fluid.T_ph(101.3,1234); % temperature at 101.3 kPa, 1234 kJ/kg
3 T3 = fluid.T_pv(200.0,0.002); % temperature at 200 kPa, 0.002 m^3/kg
4 T4 = fluid.Tsat_p(8000); % saturation temperature at 8000 kPa
5 T5 = fluid.T_crit(); % critical temperature for the fluid (C for SI, F for USCS)
6 T6 = fluid.T_pu(101.3,200.); % temperature at 101.3 kPa, 200 kJ/kg
7 T7 = fluid.T_sv(1.5,3.5); % temperature at 1.5 kJ/kg-K and 3.5 m^3/kg
%% Pressure - SI units: kPa USCS units: psia
8
9 P1 = fluid.Psat_T(300); % saturation pressure at 300 C
10 P2 = fluid.P_vT(0.002,300.); % pressure at 0.002 m^3/kg, 300 C
11 P3 = fluid.P_crit(); % critical pressure for the fluid (kPa or psia)
12 % also: P_sT(S,T)
13 % P_sh(S,H)
14

```

Internal Energy

```

U1 = fluid.uL_p(101.3); % internal energy of saturated liquid at 101.3 kPa
2 U2 = fluid.uV_p(101.3); % internal energy of saturated vapor at 101.3 kPa
3 U3 = fluid.u_px(101.3,0.5); % internal energy of saturated mixture at 101.3 kPa, 50% quality
4 U4 = fluid.u_Tx(200,0.5); % internal energy of saturated mixture at 200 C, 50 % quality
5 U5 = fluid.u_pT(101.3,200); % internal energy of a fluid at 101.3 kPa, 200 C
6 U6 = fluid.u_ps(101.3,1.5); % internal energy of a fluid at 101.3 kPa, 1.5 kJ/kg-K
7 U7 = fluid.u_Tv(20,0.002); % internal energy of a fluid at 20 C, 0.002 m^3/kg

```

Specific Volume and Quality

```

%% specific volume - SI units: m^3/kg, USCS units: ft^3/lbm
1 V1 = fluid.v_ph(101.3,2300.); % specific volume of fluid at 101.3 kPa, 2300.0 kJ/kg
2 V2 = fluid.v_pu(101.3,500.); % specific volume of fluid at 101.3 kPa, 500. kJ/kg
3 V3 = fluid.v_Tx(200.,0.5); % specific volume for saturated mixture at 200 C, 50% quality
4 V4 = fluid.v_pT(101.3,200.); % specific volume for 101.3 kPa, 200 C
5
6 %% Quality
7

```

```

9 x1 = fluid.x_ph(100,700); % quality at 100 kPa, 700 kJ/kg enthalpy
x2 = fluid.x_pu(101,500); % quality at 101 kPa, 500 kJ/kg internal energy

```

Viscosity, Thermal Conductivity, and Prandtl Number

```

%% viscosity - SI units: Pa-s (kg/m-s) USCS units: lbm/ft-s
2 mu1 = fluid.mu_pT(101.3,200); % viscosity at 101.3 kPa, 200 C

%% thermal conductivity - SI units: kW/m-K USCS units: BTU/ft-sec-R
4 k1 = fluid.k_pT(101.3,200); % thermal conductivity at 101.3 kPa, 200 C

%% Prandtl Number
6 Pr1 = fluid.Prandtl_pT(101.3,200); % fluid Prandtl number at 101.3 kPa, 200 C
8

```

Specific Heat and Ideal Gas Constants

```

helium = py.EasyProp.simpleFluid('He', 'USCS');
%% Specific heats - SI units kJ/kg-K, USCS units: BTU/lbm-R
Cp = helium.Cp_pT(14.7,200); % specific heat at constant pressure for Helium at 14.7 psia, 200 F
4 Cv = helium.Cv_pT(14.7,200); % specific heat at constant volume
gamma = helium.gamma_pT(14.7,200); % ratio of: Cp/Cv for Helium
6
%% Gas Constant - SI units kJ/kg-K USCS units: BTU/lbm-R
8 R1 = helium.R_pT(14.7,200); % Gas constant for Helium at 14.7 psia, 200 F
10 %% Molecular weight - SI kg/kmol USCS units: lbm/mol
M = helium.M();

```

Humid Air Object and Functions

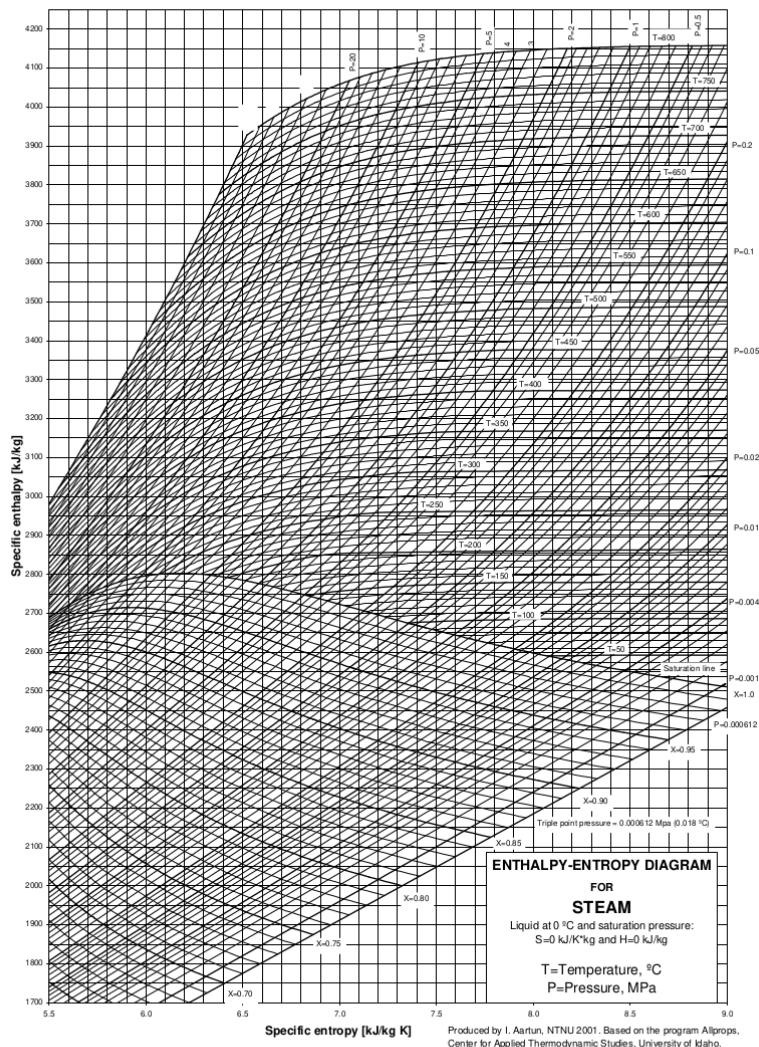
```

%% Humid Air Object and functions
2 humidAir = py.EasyProp.HumidAir('SI'); % option: 'SI' | 'USCS'

4 % humidity ratio as a function of temp, pressure, and relative humidity
w = humidAir.w_PTR(101.3,20,0.2); % no units
6
8 % enthalpy as a function of temp, pressure, and relative humidity
h = humidAir.h_PTR(101.3,20,0.2); % kJ/kg or BTU/lbm

```


Mollier Diagram



Example - Simple Rankine Cycle

Introduction

This MATLAB listing is provided as an example to help users better understand how to use EasyProp for thermodynamic analysis of cycles.

To run this code, create a MATLAB script containing the code in the listing below. You should have a copy of the library EasyProp.py in the same folder.

```
1 % EasyProp from matlab for Rankine Cycle Analysis
2 %
3 % Example problem using EasyProp for fluid properties.
4 % Problem is adapted from Example 8-5 of Introduction to Thermal and Fluids Engineering,
5 % 1st Edition , D. Kaminski and M. Jensen , John Wiley & Sons
6 %
7 % A Rankine cycle uses water as the working fluid. The boiler operates at 8 MPa
8 % and produces saturated vapor. Saturated liquid exits the condenser at 7.5 kPa.
9 % The pump has an isentropic efficiency of 82 percent, and the turbine has an
10 % isentropic efficiency of 88 percent. The steam flow rate is 2.8x10**4 kg/h.
11 % Cooling water enters the condenser at 20 C and leaves at 38 C.
12 %
13 % Find:
14 % a) the net power produced (kW)
15 % b) the net heat transfer rate into the boiler (kW)
16 % c) cycle thermal efficiency (%)
17 % d) cooling water flow rate (kg/h)
18 %
19 %% clear the environment
20 clear
21 clc
22 close 'all'
23 %
24 %% Add current directory to the Python Path
25 EasyProp_path = ' '; %<- Path if EasyProp.py is in your current directory
26 if count(py.sys.path,EasyProp_path) == 0 %<- see if desired directory is on path
27     insert(py.sys.path,int32(0),EasyProp_path); %<- if not; add it.
28 end
29 %
30 %% Establish working fluid and unit system
31 fluid = 'Water';
32 unitSystem = 'SI';
33 myFluid = py.EasyProp.simpleFluid(fluid,unitSystem);
34 %
35 %% Problem parameters
36 numSp = 4;
```

```

37 eta_turbine = 0.88;
eta_pump = 0.82;
39 m_dot_steam = 2.8e4; % kg/h
Pmax = 8000; % kPa
41 Pmin = 7.5; % kPa
Tcool_in = 20; % C
43 Tcool_out = 38; % C

45 %% declare fluid property arrays
h = nan(numSp,1);
47 h_s = nan(numSp,1);
s = nan(numSp,1);
49 s_s = nan(numSp,1);
P = nan(numSp,1);
51 T = nan(numSp,1);
T_s = nan(numSp,1);

53 %% calculate properties at state points
% state point 1 - condenser outlet condensate pump inlet
55 P(1) = Pmin;
h(1) = myFluid.hL_p(P(1));
s(1) = myFluid.sL_p(P(1));

59 % compression from state point 1 to state point 2
61 P(2) = Pmax;
s_s(2) = s(1);
63 h_s(2) = myFluid.h_ps(P(2),s_s(2));
h(2) = h(1) - (h(1) - h_s(2))./eta_pump;

65 w_pump = h(1) - h(2);

67 % isobaric heat addition in the boiler from state point 2 to state point 3
69 P(3) = P(2);
h(3) = myFluid.hV_p(P(3));
71 s(3) = myFluid.sV_p(P(3));

73 q_s = h(3) - h(2);

75 % expansion in turbine from state point 3 to state point 4
77 P(4) = P(1);
s_s(4) = s(3);
79 h_s(4) = myFluid.h_ps(P(4),s_s(4));
h(4) = h(3) - eta_turbine*(h(3) - h_s(4));

81 w_turbine = h(3) - h(4);

83 % isobaric heat rejection in the condenser from sp 4 to sp 1
q_r = h(1) - h(4);

85 %% First Law Check
87 w_net = w_pump + w_turbine;
q_net = q_s + q_r;

89 fprintf('Net work = %g kJ/kg \n',w_net);
91 fprintf('Net heat = %g kJ/kg \n',q_net);

93 %% Question a) Net power (kW)
net_power = w_net*m_dot_steam*(1/3600); % kW

95 fprintf('Net power = %g kW\n',net_power);
97

```

```
%% Question b) Net heat transfer rate into the boiler
99 net_Qdot_supplied = q_s*m_dot_steam*(1/3600); % kW
101 fprintf('Net heat transfer into the boiler = %g kW \n',net_Qdot_supplied);
103 %% Question c) cycle thermal efficiency
103 eta_th = w_net/q_s;
105 fprintf('Thermal efficieency = %4.1f percent \n',eta_th*100);
107 %% Question d) condenser cooling water flow rate
109 Po = 101.325; % kPa - assume Standard Pressure
111 h_cool_in = myFluid.h_pT(Po,Tcool_in);
111 h_cool_out = myFluid.h_pT(Po,Tcool_out);
113 m_dot_cool = m_dot_steam*(h(4) - h(1))/(h_cool_out - h_cool_in); % kg/h
115 fprintf('Condenser cooling water flow rate = %g kg/h \n',m_dot_cool);
```


Example - Supercritical Water Reactor Analysis

To run this code, create a MATLAB script containing the code in the listing below. You should have a copy of the library EasyProp.py in the same folder.

```
%>> SCWR Example Analysis
2 %>> Prepare the environment
3 clear
4 clc
5 close 'all'
6
7 %>> Add current directory to the Python Path
8 EasyProp_path = ' '; %<- Path if EasyProp.py is in your current directory
9 if count(py.sys.path,EasyProp_path) == 0 %<- see if desired directory is on path
10     insert(py.sys.path,int32(0),EasyProp_path); %<- if not; add it.
11 end
12
13 %>> Initialize Fluid Property object
14 fluid = 'Water';
15 units = 'SI';
16 water = py.EasyProp.EasyProp(fluid,units);
17
18 %>> Initialize State Point Data Arrays
19
20 numSP = 25;
21 h = nan(numSP,1);
22 h_s = nan(numSP,1);
23 s = nan(numSP,1);
24 s_s = nan(numSP,1);
25 T = nan(numSP,1);
26 P = nan(numSP,1);
27 x = nan(numSP,1);
28 ef = nan(numSP,1);
29
30 %>> Set parameter values from given data
31 Pmax = 24000; % kPa, reactor outlet pressure
32 Pm1 = 8900; % kPa, outlet pressure HP1
33 Pm2 = 6000; % kPa, outlet pressure HP2
34 Pm3 = 1000; % kPa, outlet pressure HP3
35 Pm4 = 200; % kPa, outlet pressure LP1
36 Pm5 = 4; % kPa, outlet pressure LP2
37
38 eta_t = 0.9;
39 eta_p = 0.88;
40
41 To_C = 16;
42 To = 16+273; % K, dead state temperature
43 Po = 101; % kPa, dead state pressure
```

```

44 ho = water.h_pT(Po,To_C);
46 so = water.s_pT(Po,To_C);

48 % function for specific flow exergy neglecting kinetic and potential energy
ef_fun = @(h_val,s_val) h_val - ho - To*(s_val - so);

50 %% Calculate Thermodynamic Properties of State Points

52 % state point 1 - condenser outlet, hotwell pump input
54 P(1) = Pm5;
55 x(1) = o.o;
56 T(1) = water.Tsat_p(P(1));
57 s(1) = water.sL_p(P(1));
58 h(1) = water.hL_p(P(1));
59 ef(1)= ef_fun(h(1),s(1));

60 % state point 2 - hotwell pump outlet / OFWH inlet
62 P(2) = Pm4;
63 s_s(2) = s(1);
64 h_s(2) = water.h_ps(P(2),s_s(2));
65 h(2) = h(1) - (h(1) - h_s(2))./eta_p;
66 s(2) = water.s_ph(P(2),h(2));
67 T(2) = water.T_ph(P(2),h(2));
68 ef(2) = ef_fun(h(2),s(2));

70 % state point 3 - OFWH outlet
72 P(3) = Pm4;
73 x(3) = o.o;
74 h(3) = water.hL_p(P(3));
75 T(3) = water.Tsat_p(P(2));
76 s(3) = water.sL_p(P(3));
77 ef(3) = ef_fun(h(3),s(3));

78 % state point 4 - CBP outlet
80 P(4) = Pm1;
81 s_s(4) = s(3);
82 h_s(4) = water.h_ps(P(4),s_s(4));
83 h(4) = h(3) - (h(3) - h_s(4))./eta_p;
84 s(4) = water.s_ph(P(4),h(4));
85 T(4) = water.T_ph(P(4),h(4));
86 ef(4) = ef_fun(h(4),s(4));

86 % state point 5 - outlet from mixing point / entry to CFWH 2
88 P(5) = P(4);
89 % remainder waits until after non-linear solver

90 % Skip to reactor outlet so I can get turbine exhaust conditions for CFWHs
91 % state point 9 - reactor outlet, HP1 inlet
92 P(9) = Pmax;
93 T(9) = 550; % C, reactor outlet temp (given) %— note: "saturation temperature" doesn't make sense
94 here.
95 h(9) = water.h_pT(P(9),T(9));
96 s(9) = water.s_pT(P(9),T(9));
97 ef(9) = ef_fun(h(9),s(9));

98 % state point 10 - HP1 outlet,
100 P(10) = Pm1;
101 s_s(10) = s(9);
102 h_s(10) = water.h_ps(P(10),s_s(10));
103 h(10) = h(9) - (h(9) - h_s(10))*eta_t;

```

```

104| s(10) = water.s_ph(P(10),h(10));
105| T(10) = water.T_ph(P(10),h(10)); %<-- a little superheated (good!!)
106| tsat1 = water.Tsat_p(P(10));
107| ef(10) = ef_fun(h(10),s(10));
108|
109| % state point 11 - HP2 outlet, HP3 inlet
110| P(11) = Pm2;
111| s_s(11) = s(10);
112| h_s(11) = water.h_ps(P(11),s_s(11));
113| h(11) = h(10) - (h(10) - h_s(11))*eta_t;
114| s(11) = water.s_ph(P(11),h(11));
115| T(11) = water.T_ph(P(11),h(11));
116| tsat2 = water.Tsat_p(P(11)); %<-- again, superheated
117| ef(11) = ef_fun(h(11),s(11));
118|
119| % state point 12 - HP3 outlet, M/S inlet
120| P(12) = Pm3;
121| s_s(12) = s(11);
122| h_s(12) = water.h_ps(P(12),s_s(12));
123| h(12) = h(11) - (h(11) - h_s(12))*eta_t;
124| s(12) = water.s_ph(P(12),h(12));
125| x(12) = water.x_ph(P(12),h(12));
126| T(12) = water.T_ph(P(12),h(12));
127| ef(12) = ef_fun(h(12),s(12));
128|
129| % state point 13 - M/S outlet, LP1 inlet
130| P(13) = P(12);
131| x(13) = 1.0;
132| h(13) = water.hV_p(P(13));
133| s(13) = water.sV_p(P(13));
134| T(13) = water.T_ph(P(13),h(13));
135| ef(13) = ef_fun(h(13),s(13));
136|
137| % state point 14 - LP1 outlet , MS2 inlet
138| P(14) = Pm4;
139| s_s(14) = s(13);
140| h_s(14) = water.h_ps(P(14),s_s(14));
141| h(14) = h(13) - (h(13) - h_s(14))*eta_t;
142| s(14) = water.s_ph(P(14),h(14));
143| T(14) = water.T_ph(P(14),h(14));
144| x(14) = water.x_ph(P(14),h(14));
145| ef(14) = ef_fun(h(14),s(14));
146|
147| % state point 15 - MS2 outlet , LP2 inlet
148| P(15) = P(14);
149| x(15) = 1.0;
150| s(15) = water.sV_p(P(15));
151| h(15) = water.hV_p(P(15));
152| T(15) = water.T_ph(P(15),h(15));
153| ef(15) = ef_fun(h(15),s(15));
154|
155| % state point 16 - LP2 outlet , condenser inlet
156| P(16) = Pm5;
157| s_s(16) = s(15);
158| h_s(16) = water.h_ps(P(16),s_s(16));
159| h(16) = h(15) - (h(15) - h_s(16))*eta_t;
160| s(16) = water.s_ph(P(16),h(16));
161| x(16) = water.x_ph(P(16),h(16));
162| T(16) = water.T_ph(P(16),h(16));
163| ef(16) = ef_fun(h(16),s(16));
164|

```

```

166 % state point 17 - MS1 liquid drain
167 P(17) = P(12);
168 x(17) = 0.0;
169 h(17) = water.hL_p(P(17));
170 s(17) = water.sL_p(P(17));
171 T(17) = water.T_ph(P(17),h(17));
172 ef(17) = ef_fun(h(17),s(17));

174 % state point 18 - Trap 2 outlet
175 P(18) = P(2);
176 h(18) = h(17); % isenthalpic
177 s(18) = water.s_ph(P(18),h(18));
178 T(18) = water.T_ph(P(18),h(18));
179 ef(18) = ef_fun(h(18),s(18));

180 % state point 19 - M/S #2 liquid drain
181 P(19) = P(14);
182 x(19) = 0.0;
183 h(19) = water.hL_p(P(19));
184 s(19) = water.sL_p(P(19));
185 T(19) = water.T_ph(P(19),h(19));
186 ef(19) = ef_fun(h(19),s(19));

188 % state point 20 - CFWH #1 drain / trap 1 inlet
189 P(20) = P(10);
190 T(20) = water.Tsat_p(P(20));
191 h(20) = water.hL_p(P(20));
192 s(20) = water.sL_p(P(20));
193 ef(20) = ef_fun(h(20),s(20));

196 % state point 21 - trap 1 outlet
197 P(21) = P(11);
198 h(21) = h(20); % isenthalpic
199 s(21) = water.s_ph(P(21),h(21));
200 T(21) = water.T_ph(P(21),h(21));
201 ef(21) = ef_fun(h(21),s(21));

202 % state point 22 - CFWH #2 drain, DBP suction
203 P(22) = P(11);
204 h(22) = water.hL_p(P(22));
205 s(22) = water.s_ph(P(22),h(22));
206 T(22) = water.T_ph(P(22),h(22));
207 ef(22) = ef_fun(h(22),s(22));

210 % state point 23 - dbp discharge
211 P(23) = P(4);
212 s_s(23) = s(22);
213 h_s(23) = water.h_ps(P(23),s_s(23));
214 h(23) = h(22) - (h(22) - h_s(23))/.eta_p;
215 s(23) = water.s_ph(P(23),h(23));
216 T(23) = water.T_ph(P(23),h(23));
217 ef(23) = ef_fun(h(23),s(23));

218 % state point 24 - condenser cooling water inlet
219 P(24) = Po;
220 T(24) = To_C;
221 h(24) = water.h_pT(P(24),T(24));
222 s(24) = water.s_pT(P(24),T(24));
223 ef(24) = ef_fun(h(24),s(24));

```

```

226 % state point 25 - condenser cooling water outlet
227 P(25) = Po;
228 T(25) = 27;
229 h(25) = water.h_pT(P(25),T(25));
230 s(25) = water.s_pT(P(25),T(25));
231 ef(25) = ef_fun(h(25),s(25));

232 % state point 6 - CFWH 2 outlet / main feedwater pump inlet
233 P(6) = P(5);
234 T(6) = 230; % C, given.
235 h(6) = water.h_pT(P(6),T(6));
236 s(6) = water.s_pT(P(6),T(6));
237 ef(6) = ef_fun(h(6),s(6));

240 % state point 7 - MFWP outlet / CFWH 1 inlet
241 P(7) = Pmax;
242 s_s(7) = s(6);
243 h_s(7) = water.h_ps(P(7),s_s(7));
244 h(7) = h(6) - (h(6) - h_s(7))./eta_p;
245 s(7) = water.s_ph(P(7),h(7));
246 ef(7) = ef_fun(h(7),s(7));
247 T(7) = water.T_ph(P(7),h(7));

248 % state point 8 - CFWH1 outlet Rx inlet
249 P(8) = P(7);
250 T(8) = water.Tsat_p(Pm1);
251 h(8) = water.h_pT(P(8),T(8));
252 s(8) = water.s_pT(P(8),T(8));
253 ef(8) = ef_fun(h(8),s(8));

256 %% Energy Balances and unknown calculation
258
259 % unknowns: f = [f1,f2,f3,h5]
260 OFWH_hb = @(f) (1-f(1))*(1-f(2))*x(12)*x(14)*(1-f(3))*h(2) + ...
261     (1-f(1))*(1-f(2))*x(12)*x(14)*f(3)*h(15) + ...
262     (1-f(1))*(1-f(2))*x(12)*(1-x(14))*h(19) + ...
263     (1-f(1))*(1-f(2))*(1-x(12))*h(18) - ...
264     (1-(f(1)+(1-f(1))*f(2)))*h(3);

266 MP1_hb = @(f) (1-(f(1)+(1-f(1))*f(2)))*h(4) + ...
267     (f(1)+(1-f(1))*f(2))*h(23) - ...
268     f(4); %% remember, f(4) = h(5)

270 CFWH2_hb = @(f) f(4)+ (1-f(1))*f(2)*h(11) + f(1)*h(21) - ...
271     (f(1)+(1-f(1))*f(2))*h(22) - h(6);

272 CFWH1_hb = @(f) f(1)*h(10) + h(7) - f(1)*h(20) - h(8);

274 balance = @(f) abs(OFWH_hb(f))+abs(MP1_hb(f))+abs(CFWH2_hb(f))+...
275     abs(CFWH1_hb(f));

278 %% provide inputs to fmincon to solve non-linear equations
279 xo = [0.5 0.5 0.5 h(4)];
280 A = []; % no linear inequality constraints
281 b = [];
282 Aeq = []; % no linear equality constraints
283 beq = [];
284 lb = [0 0 0 h(1)]; % all values must be non-negative
285 ub = [1 1 1 h(6)]; % flow fraction limited to 1
286 nlcon = [];

```

```

288 options = optimoptions('fmincon','Display','iter','Algorithm','sqp',...
    'StepTolerance',1e-14,'MaxFunctionEvaluations',1000);

290 [f,fval,exitflag] = fmincon(balance,xo,A,b,Aeq,beq,lb,ub,nlcon,options);
%
292 % % unpack and report the results
293 fprintf('balance function value at convergence: %g kJ/kg \n',fval);
294 fprintf('Flow fraction to CFWH1: %g \n',f(1));
295 fprintf('Flow fraction to CFWH2: %g \n',f(2));
296 fprintf('Flow fraction to OFWH: %g \n',f(3));
297 h(5) = f(4);
298 fprintf('Enthalpy of flow into CFWH2: %g kJ/kg \n',h(5));

300 % finish state point 5:
301 T(5) = water.T_ph(P(5),h(5));
302 s(5) = water.s_ph(P(5),h(5));
303 ef(5) = ef_fun(h(5),s(5));
%
304 % display state point data neatly
305 SP = {'1','2','3','4','5','6','7','8','9','10','11','12',...
    '13','14','15','16','17','18','19','20','21','22','23','24','25'};
306 SP_table = table(P,T,h,s,x,ef,'ColumnName',SP);
307 disp(SP_table);

310 %% First-law Analysis results

312 % pumps:
313 w_hwp = (h(1)-h(2))*(1-f(1))*(1-f(2))*x(12)*x(14)*(1-f(3));
314 fprintf('Specific work Hotwell Pump: %g kJ/kg \n',w_hwp);

316 w_cbp = (h(3)-h(4))*(1-(f(1)+(1-f(1))*f(2)));
317 fprintf('Specific work Condensate Booster Pump: %g kJ/kg \n',w_cbp);

319 w_dbp = (h(22)-h(23))*(f(1)+(1-f(1))*f(2));
320 fprintf('Specific work Drain Booster Pump: %g kJ/kg \n',w_dbp);

322 w_mfp = h(6) - h(7);
323 fprintf('Specific work main feed pump: %g kJ/kg \n',w_mfp);

325 w_p = w_hwp + w_cbp + w_dbp + w_mfp;

328 % turbines:
329 w_hp1 = h(9)-h(10);
330 fprintf('Specific work HP1: %g kJ/kg \n',w_hp1);

332 w_hp2 = (h(10)-h(11))*(1-f(1));
333 fprintf('Specific work HP2: %g kJ/kg \n',w_hp2);

335 w_hp3 = (h(11)-h(12))*(1-f(1))*(1-f(2));
336 fprintf('Specific work HP3: %g kJ/kg \n',w_hp3);

338 w_lp1 = (h(13)-h(14))*(1-f(1))*(1-f(2))*x(12);
339 fprintf('Specific work LP1: %g kJ/kg \n',w_lp1);

341 w_lp2 = (h(15)-h(16))*(1-f(1))*(1-f(2))*x(12)*x(14)*(1-f(3));
342 fprintf('Specific work LP2: %g kJ/kg \n',w_lp2);

344 w_t = w_hp1 + w_hp2 + w_hp3 + w_lp1 + w_lp2;

346 w_net = w_p + w_t;

```

```
348 fprintf('Net specific work: %g kJ/kg \n',w_net);  
350 % heat supplied  
351 q_s = h(9) - h(8);  
352 % heat rejected  
353 q_r = (h(1)-h(16))*(1-f(1))*(1-f(2))*x(12)*x(14)*(1-f(3));  
354 q_net = q_s+q_r;  
355 fprintf('Net specific heat: %g kJ/kg \n',q_net);  
356 eta_th = w_net/q_s;  
357 fprintf('Thermal efficiency: %5.2f percent \n',eta_th*100);  
358
```


Example - Recompressing Regenerating S-CO₂ Analysis

To run this code, create a MATLAB script containing the code in the listing below. You should have a copy of the library EasyProp.py in the same folder.

```
%>> %% SCO-2 energy conversion example
2 clear
3 clc
4 close 'all'
5
6 %% Add current directory to the Python Path
7 EasyProp_path = ' ';
8 if count(py.sys.path,EasyProp_path) == 0
9     insert(py.sys.path,int32(0),EasyProp_path);
10 end
11
12 %% Initialize Fluid Property object
13 fluid = 'CO2';
14 units = 'SI';
15 gas = py.EasyProp.EasyProp(fluid,units);
16
17 To_C = 25; % C
18 To = To_C + 273; % K
19 Po = 101.3; % kPa
20 ho = gas.h_pT(Po,To_C);
21 so = gas.s_pT(Po,To_C);
22
23
24 %% function for specific flow exergy
25 ef_fun = @(h_val,s_val) h_val - ho - To*(s_val - so);
26
27 %% Initialize State Point Data Arrays
28
29 numSP = 8;
30 h = nan(numSP,1);
31 h_s = nan(numSP,1);
32 s = nan(numSP,1);
33 s_s = nan(numSP,1);
34 T = nan(numSP,1);
35 P = nan(numSP,1);
36 x = nan(numSP,1);
37 ef = nan(numSP,1);
38
39 eta_turb = 1;
40 eta_c = 1;
41 eta_rc = eta_c;
42 xi_ltr = 0.90;
```

```

44 xi_htr = 0.90;

46 Pmax = 20000; % kPa
Pmin = 7700; % kPa

48 %% state point property calculations
50 P(1) = Pmin; % kPa
T(1) = 32; % C
52 h(1) = gas.h_pT(P(1),T(1));
s(1) = gas.s_pT(P(1),T(1));
ef(1) = ef_fun(h(1),s(1));

56 %% 1 -> 2 compression
P(2) = Pmax; % kPa
58 s_s(2) = s(1);
h_s(2) = gas.h_ps(P(2),s_s(2));
60 h(2) = h(1) - (h(1) - h_s(2))/eta_c;
T(2) = gas.T_ph(P(2),h(2));
62 s(2) = gas.s_ph(P(2),h(2));
ef(2) = ef_fun(h(2),s(2));

64 %% come back to sp 3
P(3) = P(2);

68 %% come back to sp 4
P(4) = P(3);

70 %% state point 5
72 P(5) = P(4);
T(5) = 550; %C
74 h(5) = gas.h_pT(P(5),T(5));
s(5) = gas.s_pT(P(5),T(5));
76 ef(5) = ef_fun(h(5),s(5));

78 %% expand in turbine for sp 6
80 P(6) = P(1); % kPa
s_s(6) = s(5);
82 h_s(6) = gas.h_ps(P(6),s_s(6));
h(6) = h(5) - eta_turb*(h(5) - h_s(6));
84 s(6) = gas.s_ph(P(6),h(6));
T(6) = gas.T_ph(P(6),h(6));
86 ef(6) = ef_fun(h(6),s(6));

88 %% state point 7 (what I know for now...)
P(7) = P(6);

90 %% state point 8 ( what I know for now...)
P(8) = P(7);

94 %% set up system of equations:
96 F = [f1, h(3), h(4), h(7), h(8)];

98 %% #1 - Low Temperature Regenerator Effectiveness
LTR_eff = @(F) xi_ltr*(F(4)-gas.h_pT(P(8),T(2))) - (F(4) - F(5));

100 %% #2 - High Temperature Regenerator Effectiveness
HTR_eff = @(F) xi_htr*(h(6)-gas.h_pT(P(7),gas.T_ph(P(3),F(2)))) - ...
(h(6) - F(4));

```

```

% #3 - Low Temperature Regenerator Energy Balance
106 LTR_ebal = @(F) F(1)*(F(2)-h(2)) - (F(4)-F(5));

% #4 - High Temperature Regenerator Energy Balance
108 HTR_ebal = @(F) (F(3) - F(2)) - (h(6)-F(4));

110 % #5 - State Point 3 "identity" <- any ideas for a better name?
112 SP3_id = @(F) F(5) - (F(5)-gas.h_ps(P(3),gas.s_ph(P(8),F(5)))./eta_rc - ...
    F(2));

114 % overall balance equation. Scalar output to vector input. Minimum value
116 % is zero.
117 balance = @(F) abs(LTR_eff(F))+abs(HTR_eff(F))+abs(LTR_ebal(F)) +...
    abs(HTR_ebal(F))+abs(SP3_id(F));

120 option = 1;
% 1 = no inequality constraints and option arguments
122 % 2 = additional option arguments and inequality constraints

124 Aeq = []; % no linear equalities to impose
Beq = [];
126 lb = [0 h(2) h(2) h(1) h(2)]; % lower bound
ub = [1 h(6) h(5) h(6) h(6)]; % upper bound
128 initial_guess = [0.5 h(2) h(5) h(6)-1 h(2)];

130 switch option
    case 1
        A = []; % no linear inequality constraints
        b = [];
        [F,fval,exitflag] = fmincon(balance,initial_guess,A,b,Aeq,Beq, ...
            lb,ub);
    case 2
        A = [0 1 -1 0 0; ...
            0 0 0 -1 1]; %< h(3) - h(4) <= 0, h(8)-h(7) <= 0
        b = [0;
            0];
        options = optimoptions('fmincon','FunctionTolerance',1e-6, ...
            'ConstraintTolerance',1e-9,'StepTolerance',1e-15, ...
            'Display','iter','Algorithm','sqp',...
            'MaxFunctionEvaluations',1500);
        [F,fval,exitflag] = fmincon(balance,initial_guess,A,b,Aeq,Beq, ...
            lb,ub,[],options);
    otherwise
        error('Invalid selection for option!');
    end

150 fprintf('fval = %g \n',fval);
152 fprintf('exit flag = %d \n',exitflag);

154 % unpack the results
ff = F(1);
156 h(3) = F(2);
h(4) = F(3);
158 h(7) = F(4);
h(8) = F(5);
160 % fill in missing details on state point data
162 T(3) = gas.T_ph(P(3),h(3));
s(3) = gas.s_pT(P(3),T(3));
164 ef(3) = ef_fun(h(3),s(3));

```

```

166 T(4) = gas.T_ph(P(4),h(4));
168 s(4) = gas.s_ph(P(4),h(4));
169 ef(4) = ef_fun(h(4),s(4));

170 T(7) = gas.T_ph(P(7),h(7));
171 s(7) = gas.s_ph(P(7),h(7));
172 ef(7) = ef_fun(h(7),s(7));

174 T(8) = gas.T_ph(P(8),h(8));
175 s(8) = gas.s_ph(P(8),h(8));
176 ef(8) = ef_fun(h(8),s(8));

178 fprintf('Flow fraction f = %g \n',ff);

180 w_comp = ff*(h(1) - h(2));
181 w_rc = (1-ff)*(h(8) - h(3));
182 w_turb = h(5) - h(6);
183 w_net = w_comp + w_rc + w_turb;

184 q_s = h(5) - h(4);
185 q_r = ff*(h(1) - h(8));
186 q_net = q_s+q_r;

188 fprintf('Net specific work = %g kJ/kg \n',w_net);
189 fprintf('Net specific heat = %g kj/kg \n',q_net);

192 eta_th = w_net/q_s;
193 fprintf('Thermal efficiency = %5.2f percent \n',eta_th*100);

194 bwr = abs(w_comp + w_rc)/w_turb;
195 fprintf('Back work ratio = %g percent \n',bwr*100);

198 ef_in = ef(5) - ef(4);
199 ef_out = ff*(ef(8) - ef(1));

200 Ex_d_ltr = ff*ef(2) + ef(7) - ef(8) - ff*ef(3);
201 Ex_d_htr = ef(3) + ef(6) - ef(4) - ef(7);

204 fprintf('Specific flow exergy in: %g kJ/kg \n',ef_in);
205 fprintf('Specific flow exergy out: %g kJ/kg \n',ef_out);
206 fprintf('Specific flow exergy destroyed LTR: %g kJ/kg \n',Ex_d_ltr);
207 fprintf('Specific flow exergy destroyed HTR: %g kJ/kg \n',Ex_d_htr);

208 Flow_ex_balance = (ef_in - ef_out) - (w_net + Ex_d_ltr + Ex_d_htr);
209 fprintf('Specific flow exergy balance: %g kJ/kg \n',Flow_ex_balance);

212 fprintf('\n\nState point data: \n\n');

214 % display state point data neatly
215 SP = {'1','2','3','4','5','6','7','8'};
216 SP_table = table(P,T,h,s,ef,'RowName',SP);
217 disp(SP_table);

```

Example - Combined Cycle Analysis

To run this code, create a MATLAB script containing the code in the listing below. You should have a copy of the library EasyProp.py in the same folder.

```
1 %% ER468 Combined Cycle code
2
3 clear
4 clc
5 close 'all'
6
7 %% Prepare the environment
8
9 EasyProp_path = ' '; %<-- Path if EasyProp.py is in your current directory
10 if count(py.sys.path,EasyProp_path) == 0 %<-- see if desired directory is on path
11     insert(py.sys.path,int32(0),EasyProp_path); %<-- if not; add it.
12 end
13
14 %% Establish working fluid and unit system
15 fluid = 'Air';
16 unitSystem = 'SI';
17 air = py.EasyProp.simpleFluid(fluid,unitSystem);
18 water = py.EasyProp.simpleFluid('Water',unitSystem);
19
20 %% Initialize state-point variables
21 numSp = 11;
22
23 h = nan(numSp,1);
24 h_s = nan(numSp,1);
25 s = nan(numSp,1);
26 s_s = nan(numSp,1);
27 P = nan(numSp,1);
28 T = nan(numSp,1);
29 x = nan(numSp,1);
30
31 %% Compute state point properties (incorporating given data)
32 Patm = 101.3; % kPa
33 Tatm = 15; % C
34 Tmax = 770;
35 Texh = 250; % C
36
37 rp = 17;
38 eta_c = 0.89;
39
40 r_e1 = 4;
41 eta_t = 0.92;
42
43 %% dP parameters
```

```

45 noDP = 0;
switch noDP
47   case 1
        % ignore hydraulic losses
        dpInlet = 0; % kPa
        dP_hx = 0;   % kPa
        dP_exh = 0;  % kPa
    case 0
53       dpInlet = 1; % kPa
55       dP_hx = 10; % kPa
56       dP_exh = 5;  % kPa
end

59 % compute pressures first
P(1) = Patm - dpInlet;
61 P(2) = rp * P(1);
P(3) = P(2) - dP_hx;
63 P(4) = P(3) / r_e1;
P(5) = P(4) - dP_hx;
65 P(7) = Patm + dP_exh;
P(6) = P(7) + dP_hx;

67 % compute remainin state point properties
T(1) = Tatm;
h(1) = air.h_pT(P(1), T(1));
71 s(1) = air.s_pT(P(1), T(1));

73 s_s(2) = s(1);
h_s(2) = air.h_ps(P(2), s_s(2));
75 h(2) = h(1) - (h(1) - h_s(2)) / eta_c;
s(2) = air.s_ph(P(2), h(2));
77 T(2) = air.T_ph(P(2), h(2));

79 T(3) = Tmax; % C, given
h(3) = air.h_pT(P(3), T(3));
81 s(3) = air.s_pT(P(3), T(3));

83 s_s(4) = s(3);
h_s(4) = air.h_ps(P(4), s_s(4));
85 h(4) = h(3) - (h(3) - h_s(4)) * eta_t;
s(4) = air.s_ph(P(4), h(4));
87 T(4) = air.T_ph(P(4), h(4));

89 T(5) = Tmax;
h(5) = air.h_pT(P(5), T(5));
91 s(5) = air.s_pT(P(5), T(5));

93 s_s(6) = s(5);
h_s(6) = air.h_ps(P(6), s_s(6));
95 h(6) = h(5) - (h(5) - h_s(6)) * eta_t;
s(6) = air.s_ph(P(6), h(6));
97 T(6) = air.T_ph(P(6), h(6));

99 T(7) = Texh;
h(7) = air.h_pT(P(7), T(7));
101 s(7) = air.s_pT(P(7), T(7));

103 % skip to state point 8;
x(8) = 0;

```

```

105 P(8) = 8; % kPa
106 T(8) = water.Tsat_p(P(8));
107 h(8) = water.hL_p(P(8));
108 s(8) = water.sL_p(P(8));
109 eta_p = 0.8;
110 P(9) = 3500; % kPa, given
111 s_s(9) = s(8);
112 h_s(9) = water.h_ps(P(9),s_s(9));
113 h(9) = h(8) - (h(8) - h_s(9))/eta_p;
114 s(9) = water.s_ph(P(9),h(9));
115 T(9) = water.T_ph(P(9),h(9));
116
117 P(10) = P(9); % isobaric heat addition
118 T(10) = water.Tsat_p(P(10));
119 h(10) = water.hV_p(P(10));
120 s(10) = water.sV_p(P(10));
121
122 P(11) = P(8);
123 s_s(11) = s(10);
124 h_s(11) = water.h_ps(P(11),s_s(11));
125 h(11) = h(10) - (h(10)-h_s(11))*eta_t;%<- assume same turbine efficiency
126 s(11) = water.s_ph(P(11),h(11));
127 x(11) = water.x_ph(P(11),h(11));
128 T(11) = water.Tsat_p(P(11));
129
130
131
132 SP = {'1','2','3','4','5','6','7','8','9','10','11',};
133 Tb = table(T,P,h,s,'RowName',SP);
134 fprintf('FHR Combined Cycle First-law analysis\n\n');
135 fprintf('State Point Property Data:\n');
136 disp(Tb);
137
138 %% Carry out first-law analysis and answer questions
139 m_a = 1;
140
141 m_s = m_a*(h(6)-h(7))/(h(10)-h(9));
142
143 q_s = h(3)-h(2) + h(5)-h(4);
144 q_r = h(1)-h(7) + (m_s/m_a)*(h(8)-h(11));
145 q_net = q_s + q_r;
146 fprintf('Net specific heat: %g kJ/kg \n',q_net);
147
148 w_c = h(1)-h(2);
149 w_t1 = h(3)-h(4);
150 w_t2 = h(5)-h(6);
151 w_t3 = (m_s/m_a)*(h(10)-h(11));
152 w_p = (m_s/m_a)*(h(8)-h(9));
153 w_net = w_c+w_t1+w_t2+w_t3+w_p;
154 fprintf('Net specific work: %g kJ/kg \n',w_net);
155
156 eta_th = w_net/q_s;
157 fprintf('Thermal Efficiency: %5.2f percent \n',eta_th*100);
158
159 eta_th_noCoGen = (w_c+w_t1+w_t2)/q_s;
160 fprintf('Thermal Efficiency without cogen: %5.2f percent \n',...
161 eta_th_noCoGen*100);

```


Example - Natural Circulation Demonstration

To run this code, create a MATLAB script containing the code in the listing below. You should have a copy of the library EasyProp.py in the same folder.

```
1 %% NuScale Natural Circulation Demonstration
3 % Purpose: Illustrate the relationship between mass flow rate and reactor
5 % power for a PWR utilizing natural circulation.
7 %% Establish Environment
9 clear
11 clc
11 close 'all'
13 %% Compute Mass Flow Rate vs. Power
15 Q = 160e3; % MW, Design thermal power
Qmin = 1.6e3; % 1.6 MW
Qmax = 160e3; % 160 MW
nQ = 20;
19 Qspace = linspace(Qmin,Qmax,nQ);
m_dot = nan(nQ,1);
21
23 for i = 1:nQ
    m_dot(i) = NuScale_model(Qspace(i));
end
25
27 %% Make a nice plot
27 figure(1)
28 plot(Qspace/1e3,m_dot/1e6,'-s','linewidth',3);
29 grid on
30 title('Natural Circulation Mass Flow Rate vs. Power','fontsize',16,...,
31 'fontweight','bold')
32 xlabel('Power (MW_{th})','fontsize',14,'fontweight','bold');
33 ylabel('Mass Flow Rate (lb_m/hr x 10^{-6})','fontsize',14,...,
34 'fontweight','bold');
35 set(gca,'fontsize',12,'fontweight','bold');
37 function m_dot = NuScale_model(Q_th)
%function m_dot = Nuscale_model(Q_th) - a simple natural circulation model
39 %of the NuScale PWR.
40 % Input:
41 % Q_th - thermal power in kW
42 %
43 % Output:
```

```

%  

45 % m_dot - mass flow rate (lbm/hr)  

47 %%  

48 fBal = @(m) abs(NuScale_Hydraulics(m) - NuScale_Thermal(m,Q_th));  

49  

50 initial_guess = 5e6;  

51 lb = 1e5;  

52 ub = 7e6;  

53 A = [];  

54 b = [];  

55 Aeq = [];  

56 beq = [];  

57 options = optimoptions('fmincon','FunctionTolerance',1e-6,...  

    'ConstraintTolerance',1e-9,'StepTolerance',1e-15,...  

    'Display','off',...  

    'MaxFunctionEvaluations',1500);  

58  

59 m_dot = fmincon(fBal,initial_guess,A,b,Aeq,beq,lb,ub,[],options);  

60  

61 end  

62  

63 function dP = NuScale_Thermal( m_dot, Q_th )  

64 %NuScale_Thermal Simplified thermal driving head model of the NuScale PWR.  

65 %thermal driving head (expressed as dP) given as a function of mass flow  

66 %rate and thermal power  

67 %  

68 % Input:  

69 % m_dot - mass flow rate lbm/hr  

70 % Q_th - thermal power kW  

71  

72 %% check the environment and instantiate fluid object  

73 EasyProp_path = ' '; %<- Path if EasyProp.py is in your current directory  

74 if count(py.sys.path,EasyProp_path) == 0 %<- see if desired directory is on path  

    insert(py.sys.path,int32(o),EasyProp_path); %<- if not; add it.  

75 end  

76  

77 water = py.EasyProp.simpleFluid('Water','USCS');  

78  

79 %% Model  

80  

81 P_nom = 1850; % psia, given, nominal pressure  

82 T_ave = 543; % F, given, core average temperature  

83  

84 % energy balance between core and coolant  

85 dH = (Q_th*3.41e3)/m_dot; % BTU/lbm  

86  

87 %% coolant properties  

88 h_nom = water.h_pT(P_nom,T_ave); % BTU/lbm  

89 h_h = h_nom+dH/2; % BTU/lbm, hot-leg enthalpy  

90 h_c = h_nom-dH/2; % BTU/lbm, cold-leg enthalpy  

91 rho_h = 1/water.v_ph(P_nom,h_h); % lbm/ft^3  

92 rho_c = 1/water.v_ph(P_nom,h_c); % lbm/ft^3  

93  

94 %% Reactor geometry data  

95 L_downcomer = 46.0; % ft, given  

96 L_lower_riser = 9.4; % ft, given  

97 L_core = 7.9; % ft, given  

98 %% Estimated height differenec between thermal source/sink  

99 dL_thermal = L_downcomer - L_lower_riser - L_core; % ft, estimated thermal elevation change
100
101
102
103

```

```

105 g = 32.2; % ft/sec^2
106 g_c = 32.2; % lbm/lbf * ft/sec^2
107 dP = (rho_c - rho_h)*(g/g_c)*dL_thermal; % lbf/ft^2
108 end

111 function dP = NuScale_Hydraulics( m_dot )
112 %NuScale_Hydraulics: provide a hydraulic model of the NuScale Reactor.
113 % input:
114 % m_dot - mass flow rate, lbm/hr
115 % output:
116 %
117 % dP - differential pressure in psia
118 %
119 % For the hydraulic model, we will assume that fluid density and
120 % viscosity are at their nominal conditions
121 %% check the environment and instantiate fluid object
122 EasyProp_path = ' '; %<- Path if EasyProp.py is in your current directory
123 if count(py.sys.path,EasyProp_path) == 0 %<- see if desired directory is on path
124     insert(py.sys.path,int32(o),EasyProp_path); %<- if not; add it.
125 end

127 water = py.EasyProp.simpleFluid('Water','USCS');

129 %% Model

131 P_nom = 1850; % psia, given, nominal pressure
132 T_ave = 543; % F, given, core average temperature
133 rho_nom = 1./water.v_pT(P_nom,T_ave);

135 % estimate core flow area:
136 N_assy = 37;
137 N_pins_per_assy = 17*17;
138 P_assy = 8.466/12; % ft, assembly pitch
139 D_pin = 0.374/12; % ft, pin outside diameter

141 % total core flow area taken as
142 % totality of flow areas for all pins and
143 % assemblies.
144 A_flow_core = N_assy*(P_assy^2 - (pi/4)*(D_pin^2)*N_pins_per_assy); %ft^2

145 % velocity of flow in core
146 v_core = (m_dot/3600)/(A_flow_core*rho_nom); % ft/sec

148 hyd_term = 31; % (fL/D + sum of K), unitless, assume constant.

151 g_c = 32.2; % lbm/lbf*ft/sec^2

153 dP = hyd_term*(v_core^2)*rho_nom/(2*g_c); %<- lbf/ft^2

155 end

```


Example - FRAPCON Fuel Thermal Conductivity with MATLAB

To run this code, create a MATLAB script containing the code in the listing below.

```
%>> %% Fuel Thermal Conductivity – Maximum Fuel Temperature (Frapcon Model)
2 clear
4 clc
close 'all'
6
8 %% Given Parameters
T_fo = 400; % C
q_linear = 25; % kW/m
10 gad = 0.00; % mass fraction of gadolinia
Bu = 55; % GWd/MT_ithm
12
14 %% Fuel Thermal Conductivity Model
% FRAPCON model
% accounts for
16 % temperature
% burnup
% gadolinia
18 % density other than 0.95% theoretical density is corrected with an
% additional factor.
20
22 % range of applicability
% 300K < T < 3000 K
24 % Bu < 62 GWd/Mt
% Gad < 10 w.t. %
26 % 92% < TD < 97%
28 a_frapcon = 115.99; % cm-K/W
f_frapcon = @(Bu) 0.187*Bu; %cm-K/W
30 g_frapcon = @(Bu) 3.8*(Bu.^0.28);% cm-K/W
Q_frapcon = 6380; %K
32 h_frapcon = @(T) 1./(1 + 396*exp(-Q_frapcon./T)); % unitless
34
36 % set constants from Table 8.4 based on fuel type.
fuel_type = 'UO2'; % fuel type = 'UO2' or 'MOX'
switch fuel_type
38 case 'UO2'
A_frap = 4.52; %cm-K/W
B_frap = 2.46e-2; %cm/W
40 E_frap = 3.5e7; %W-K/cm
```

```

42 F_frap = 16361; %K
43
44 case 'MOX'
45   O_M = 2.0; % set for MOX fuel - default 2.0
46   x_frap = 2.0 - O_M;
47   A_frap = 285*x_frap + 3.5; %cm-K/W
48   B_frap = (2.86 - 7.15*x_frap)*1e-2; %cm/W
49   E_frap = 1.5e7; %W-K/cm
50   F_frap = 13520; %K
51 otherwise
52   % raise an error
53   error('Unknown fuel type for FRAPCON model\n');
54 end
55
56 % equation 8.22a
57 k_op95_frap = @(T,gad,Bu) ...
58   1./ (A_frap + B_frap*T+a_frapcon*gad+f_frapcon(Bu) +...
59     (1 - 0.9*exp(-0.04*Bu)).*g_frapcon(Bu).*h_frapcon(T)) +...
60     E_frap ./ (T.^2).*exp(-F_frap./T));
61 % (be careful with syntax to allow any argument to be a vector)
62
63 % correction factor for theoretical density
64 eta_frap = @(den) 1.0789*den/(1+0.5*(1-den));
65
66 % combined expressions
67 k_frap = @(T,gad,Bu,den) k_op95_frap(T,gad,Bu)*eta_frap(den);
68
69 % set conditions for this problem.
70 den = 0.92; % percent theoretical density
71
72 k_frap_p = @(T) k_frap(T,gad,Bu,den);
73
74 f_fuel = @(T) integral(k_frap_p,T_fo+273.15,T) - ...
75   (q_linear*1000/100)/(4*pi);
76 T_est = 2245; %C - based on figure 8.2, fuel outer temperature and linear heat rate.
77 %fprintf('From figure 8.2 estimated fuel centerline temperature is %5.4g degrees C.\n',T_est);
78 T_max_K = fzero(f_fuel,T_est+273.15);
79 T_max = T_max_K - 273.15;
80
81 fprintf('The fuel centerline temperature is %5.4g degrees C.\n',T_max);

```

Example - Onset of Nucleate Boiling Demo with MATLAB

This appendix provides the complete code listing used in the lecture on nucleate boiling. To run this code, create a MATLAB script containing the code in the listing below.

```
%>> Onset of Nucleate Boiling Demo
2 clear
3 clc
4 close 'all'

6 %% Put EasyProp on Python path
% put location of EasyProp.py module on the python search path
8 if count(py.sys.path,'') == 0 %-- see if desired directory is on path
    insert(py.sys.path,int32(0),''); %-- if not; add it.
10 end

12 water = py.EasyProp.simpleFluid('Water','USCS');

14
15 %% AP1000 Given Data
16 Q_dot_total = 11.601e9*(1/3600); % BTU/s, total core heat output
P_nom = 2200; % psia, nominal system pressure
18 T_avg = 578.1; % F, average coolant temperature in the core (use for nominal coolant density)
mu_nom = 0.23*(1/3600); % lbm/ft-s
20 sigma_nom = 0.009743; % lbf/ft, surface tension
rho_nom = 1./water.v_pT(P_nom,T_avg); % lbm/ft^3, nominal density
22 Pr_nom = water.Prandtl_pT(P_nom,T_avg); % dimensionless, nominal Prandtl number
k_nom = 0.32*(1/3600); % BTU/s-ft-R
24 T_sat = water.Tsat_p(P_nom); % F, saturation temp in core
h_sat = water.hL_p(P_nom); % BTU/lbm, enthalpy of saturated liquid at nominal pressure
26 h_fg = water.hV_p(P_nom) - water.hL_p(P_nom); % BTU/lbm
m_dot_total = 113.5e6*(1/3600); % lbm/s, flow core flow rate
28 N_rads = 157*17*17;
D_co = 0.374/12; % ft, clad outside diameter
30 t_clad = 0.0225/12; % ft, clad thickness
D_gap = 6.5e-3/12; % ft, diametral gap
32 D_fuel = 0.3225/12; % ft, diameter of fuel (consistent with clad dimensions and diametral gap)
Pitch = 0.496/12; % ft, fuel rod pitch
34 H = 168/12; % ft, height of active fuel region
q_p_peak = (14.9/1.5)*3412*(1/3600); % kW/ft --> BTU/s-ft, axial peak linear heat rate
36 T_c = 535; % F, nominal core inlet temperature

38 %% calculated results
m_dot = m_dot_total/N_rads; % lbm/s, flow rate per fuel rod
40 A_flow = Pitch^2 - (pi/4)*D_fuel^2; % ft^2, flow area per fuel pin
```

```

G = m_dot/A_flow; %lbm/ft^2-s, average mass flux
42 v_avg = G/rho_nom; % ft/s

44 De = 4*A_flow/(pi*D_co); % ft, equivalent diameter

46 Re = G*De/mu_nom;

48 % inlet enthalpy
h_in = water.h_pT(P_nom,T_c);

50 % bulk coolant enthalpy as a function of core height
52 h_fun = @(z) h_in + (q_p_peak/m_dot)*(H/pi)*(sin(pi*z/H)+1); % BTU/lbm

54 core_z = @(z) z - H/2; % so core position can go from 0 to H rather than -H/2 to H/2

56 q_p = @(z) q_p_peak*cos(pi*z/H); % linear heat rate as a function of position -H/2 < z < H/2
q_pp = @(z) q_p(z)./(pi*D_co); % BTU/s-ft^2, heat flux

58 Nz = 100;
60 Z = linspace(0,H,Nz);

62 h_b = h_fun(core_z(Z)); % bulk coolant enthalpy
T_b = nan(1,Nz);
64 for z = 1:Nz
    T_b(z) = water.T_ph(P_nom,h_b(z)); % bulk coolant temperature
66 end

68 % plot bulk enthalpy
figure(1)
plot(Z,h_b,'-b',...
    Z,ones(1,Nz)*h_sat,'-r','linewidth',2)
title('Enthalpy vs Core Height','fontsize',16,'fontweight','bold');
xlabel('Axial Height in Core [ft]','fontsize',14,'fontweight','bold');
ylabel('Specific Enthalpy [BTU/lbm]','fontsize',14,'fontweight','bold');
grid on
legend('bulk','sat','location','best');
set(gca,'fontsize',12,'fontweight','bold');

78 % plot bulk temperature
figure(2)
plot(Z,T_b,'-b',...
    Z,ones(1,Nz)*T_sat,'-r','linewidth',2)
title('Temperature vs Core Height','fontsize',16,'fontweight','bold');
xlabel('Axial Height in Core [ft]','fontsize',14,'fontweight','bold');
ylabel('Temperature [^\circ F]','fontsize',14,'fontweight','bold');
grid on
legend('bulk','sat','location','best');
set(gca,'fontsize',12,'fontweight','bold');

90 % compute and plot equilibrium quality
x_eq = (h_b - h_fg)/h_fg; % equilibrium quality

92 figure(3)
plot(Z,x_eq,'-b','linewidth',2)
title('Equilibrium Quality vs Core Height','fontsize',16,'fontweight','bold');
xlabel('Axial Height in Core [ft]','fontsize',14,'fontweight','bold');
ylabel('x_{eq}','fontsize',14,'fontweight','bold');
grid on
set(gca,'fontsize',12,'fontweight','bold');

100 %% correlations

```

```

102 Nu_DB = @(Re,Pr) 0.023*(Re.^0.8).* (Pr.^0.4);
104 Nu = Nu_DB(Re,Pr_nom);
106 h_nom_DB = Nu*k_nom/De; % BTU/ft^2-sec-R, nominal convective heat transfer coefficient
108 % Presser correction factor
110 C_presser = @(P_D) 0.9217 + 0.1478*P_D - 0.1130*exp(-7*(P_D-1));
112 h_nom_FA = C_presser(Pitch/D_co)*h_nom_DB;
114 % Wall temperature using Presser correlation
116 T_wall = T_b + q_p(core_z(Z))/(2*pi*h_nom_FA*(D_co/2));
118 figure(4)
119 plot(Z,T_wall,'-b',...
120     Z,ones(1,Nz)*T_sat,'-r','linewidth',2)
121 title('Wall Temperature vs Core Height','fontsize',16,'fontweight','bold');
122 xlabel('Axial Height in Core [ft]','fontsize',14,'fontweight','bold');
123 ylabel('Wall Temperature [^\circ F]','fontsize',14,'fontweight','bold');
124 grid on
125 legend('T_{wall}', 'T_{sat}', 'location', 'best');
126 set(gca,'fontsize',12,'fontweight','bold');

128 figure(5)
129 plot(Z,q_p(core_z(Z))/(pi*D_co),'-b','linewidth',2)
130 title('Heat Flux vs Core Height','fontsize',16,'fontweight','bold');
131 xlabel('Axial Height in Core [ft]','fontsize',14,'fontweight','bold');
132 ylabel('Heat Flux [BTU/s-ft^2]','fontsize',14,'fontweight','bold');
133 grid on
134 set(gca,'fontsize',12,'fontweight','bold');

136 % try the Bernath correlation for h_NB
137 De_prime = De + D_co/pi;
138 h_NB = 10890*(De/De_prime) + 48*v_avg/(De^0.6);%BTU/hr-ft^2-F
139 h_NB = h_NB*(1/3600); %BTU/s-ft^2-F

140 % compare h_NB to h_nom_DB
141 fprintf('h_NB is equal to %5.2f percent of h_nom_DB \n',...
142     (h_NB/h_nom_DB)*100);

144 %% Use Jens-Lottes Correlation to estimate Twall where NB will start

146 T_wall_JL = @(z) T_sat + 1.897*exp(-P_nom/900).*(q_pp(core_z(z))*3600).^0.25;
147 T_wall_Thom = @(z) T_sat + 0.072*exp(-P_nom/1260).*(q_pp(core_z(z))*3600).^0.5;
148

150 figure(6)
151 plot(Z,T_wall,'-b',...
152     Z,T_wall_JL(Z),'-r',...
153     Z,T_wall_Thom(Z),'-k',...
154     Z,ones(1,Nz)*T_sat,'-g','linewidth',2)
155 title('Wall Temperature vs Core Height','fontsize',16,'fontweight','bold');
156 xlabel('Axial Height in Core [ft]','fontsize',14,'fontweight','bold');
157 ylabel('Wall Temperature [^\circ F]','fontsize',14,'fontweight','bold');
158 grid on
159 legend('T_{wall}', 'Onset NB (JL)', 'Onset NB (Thom)', 'sat', 'location', 'best');
160 set(gca,'fontsize',12,'fontweight','bold');

162

```

```
164 figure(7)
165 plot(Z,T_b,'-b',...
166     Z,T_wall,'-r',...
167     Z,q_p(core_z(Z))/(pi*D_co),'-k','linewidth',3);
168 grid on;
169 title('Heat Flux and Temperature Profile',...
170       'fontsize',16,'fontweight','bold')
171 legend('T_{bulk}', 'T_{wall}', 'q^{\prime\prime\prime}')
172 xlabel('Axial Height in Core [ft]', 'fontsize',14,'fontweight','bold')
173 ylabel('Temperature [\circ F] / Heat Flux [BTU/s-ft^2]',...
174       'fontsize',14,'fontweight','bold');
175 set(gca,'fontsize',12,'fontweight','bold');
```

Example - LOCA Analysis with MATLAB

Introduction

This appendix provides the full MATLAB code to carry out an analysis of a Loss of Coolant Accident (LOCA) for a PWR containment using MATLAB. The appendix includes two nearly identical analysis:

1. LOCA analysis without considering humidity in the air; and
2. LCOA analysis *with* considering humidity in the air.

The problem statement is, as close as possible, the same as given by Example 7.1 in the Todreas textbook. The results for both analysis are very similar; the final temperature matches to within a degree F, and a couple of kPa.

LOCA analysis without humid air

```
clear
1 clc
2 close 'all'

4 EasyProp_path = ' '; %<- Path if EasyProp.py is in your current directory
5 if count(py.sys.path,EasyProp_path) == 0 %<- see if desired directory is on path
6     insert(py.sys.path,int32(0),EasyProp_path); %<- if not; add it.
7 end

9
10 %% establish air and water objects
11
12 units = 'SI';
13 water = py.EasyProp.simpleFluid('Water',units);
14 air = py.EasyProp.simpleFluid('Air',units);

15 %% set initial temperature, pressure, and volumes
16 V_primary = 354; % m^3, given
17 V_containment = 50970; % m^3, given
18 V_total = V_primary + V_containment; % m^3

19 Po_primary = 15500; % kPa, initial primary pressure, given
20 To_primary = 344.75; % C, initial primary temperature, given
21
22 PoContainment = 101; % kPa, initial containment pressure, given
```

```

26 ToContainment = 26.85; % C, initial containment temperature, given
28 Q_total = 0; % kJ, total net energy exchange in/out of the containment during the transient.
30 %% compute the mass of air and water
32 rho_a_o = 1./air.v_pT(PoContainment,ToContainment); % kg/m^3
mass_air = VContainment*rho_a_o; % kg
34 rho_w_o = 1./water.v_pT(PoPrimary,ToPrimary); % kg/m^3
mass_water = VPrimary*rho_w_o; % kg
36
38 %% compute specific internal energy
40 u_air_o = air.u_pT(PoContainment,ToContainment); %kJ/kg
u_water_o = water.u_pT(PoPrimary,ToPrimary); % kJ/kg
42 %% set up system of equations to solve for the final state.
44 v_air_1 = V_total/mass_air; % m^3/kg
46 v_water_1 = V_total/mass_water; % m^3/kg
48 %% System of equations
% unknowns: T_1
50 Energy = @(T) mass_water*(water.u_Tv(T,v_water_1) - u_water_o) + ...
mass_air*(air.u_Tv(T,v_air_1) - u_air_o) - Q_total;
52
54 T_1 = fzero(Energy,0.5*(ToContainment + ToPrimary));
fprintf('Final system temperature = %g degrees C\n',T_1);
56 %% Determine final system pressure (no IRWST, no core make-up tanks)
58 P1_water = water.P_vT(v_water_1,T_1);
P1_air = air.P_vT(v_air_1,T_1);
60
P1_total = P1_water + P1_air;
62
% get final water quality
64 x_1 = water.x_pu(P1_water,water.u_Tv(T_1,v_water_1));
fprintf('Final water quality = %g \n',x_1);
66
fprintf('Partial pressure of water = %g kPa \n',P1_water);
68 fprintf('Partial pressure of air = %g kPa \n',P1_air);
fprintf('Total pressure = %g kPa \n',P1_total);

```

LOCA analysis with humid air

```

%% Todreas Example 7.1 Solution
2 clear
clc
4 close 'all'
6 EasyProp_path = ' '; %-- Path if EasyProp.py is in your current directory
if count(py.sys.path,EasyProp_path) == 0 %-- see if desired directory is on path
    insert(py.sys.path,int32(0),EasyProp_path); %-- if not; add it.
8 end
10

```

```

12 %% establish air and water objects
14 units = 'SI';
15 water = py.EasyProp.simpleFluid('Water',units);
16 air = py.EasyProp.simpleFluid('Air',units);
17 humidAir = py.EasyProp.HumidAir(units);
18
19 %% set initial temperature, pressure, and volumes
20 V_primary = 354; % m^3, given
21 V_containment = 50970; % m^3, given
22 V_total = V_primary + V_containment; % m^3
23
24 rel_humidity = 0.8;
25
26 Po_primary = 15500; % kPa, initial primary pressure, given
27 To_primary = 344.75; % C, initial primary temperature, given
28
29 Po_containment_total = 101; % kPa
30
31
32 To_containment = 26.85; % C, initial containment temperature, given
33
34 Pw_1 = rel_humidity*water.Psat_T(To_containment); % kPa (Step 1, pg 331 )
35 % resulting Pw_1 is slightly higher than the value given in the text
36
37 Po_containment = Po_containment_total - Pw_1; % kPa, initial containment pressure
38 % ^ Step 2 on page 332.
39
40 Q_total = 0; % kJ, total net energy exchange in/out of the containment during the transient.
41
42 %% compute the mass of air and water
43
44 rho_a_o = 1./air.v_pT(Po_containment,To_containment); % kg/m^3
45 mass_air = V_containment*rho_a_o; % kg
46
47 % get humidity ratio for temperature, pressure and relative humidity
48 w = humidAir.w_PTR(Po_containment,To_containment,rel_humidity); % use humidAir object
49 mass_water_in_air = w*mass_air; %<-- this is from step 3. humidAir gives a slightly higher result
50
51 rho_w_o = 1./water.v_pT(Po_primary,To_primary); % kg/m^3
52 mass_water = V_primary*rho_w_o + mass_water_in_air; % kg water in air and primary
53
54 %% compute specific internal energy
55
56 u_air_o = air.u_pT(Po_containment,To_containment); %kJ/kg
57 u_water_o = water.u_pT(Po_primary,To_primary); % kJ/kg
58
59 %% set up system of equations to solve for the final state.
60
61 v_air_1 = V_total/mass_air; % m^3/kg
62 v_water_1 = V_total/mass_water; % m^3/kg
63
64 %% System of equations
65 % unknowns: T_1
66
67 Energy = @(T) mass_water*(water.u_Tv(T,v_water_1) - u_water_o) + ...
68     mass_air*(air.u_Tv(T,v_air_1) - u_air_o) - Q_total;
69
70 T_1 = fzero(Energy,0.5*(To_containment + To_primary));
71 fprintf('Final system temperature = %g degrees C\n',T_1);

```

```
72
74 %% Determine final system pressure (no IRWST, no core make-up tanks)
76 P1_water = water.P_vT(v_water_1,T_1);
    P1_air = air.P_vT(v_air_1,T_1);
78 P1_total = P1_water + P1_air;
80
81 % get final water quality
82 x_1 = water.x_pu(P1_water,water.u_Tv(T_1,v_water_1));
    fprintf('Final water quality = %g \n',x_1);
83
84 fprintf('Partial pressure of water = %g kPa \n',P1_water);
85 fprintf('Partial pressure of air = %g kPa \n',P1_air);
86 fprintf('Total pressure = %g kPa \n',P1_total);
```

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