



UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

CANDIDATE NAME				
CENTRE NUMBER		CANDIDATE NUMBER		

Biology

0610/51

Paper 5 Practical Test

October/November 2012

1 hour 15 minutes

Candidates answer on the Question Paper

Additional Materials:

As listed in the Confidential Instructions

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a pencil for any diagrams or graphs.

Do not use staples, paper clips, highlighters, glue or correction fluid.

DO NOT WRITE IN ANY BARCODES.

Answer both questions.

Electronic calculators may be used.

You may lose marks if you do not show your working or if you do not use appropriate units.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.

For Exam	iner's Use
1	
2	
Total	

This document consists of 10 printed pages and 2 blank pages.



1 Read through all the questions on this paper carefully before starting work.

Examiner's Use

For

You are going to compare the metabolism of two yeast mixtures in test-tubes **W1** and **W2**.

Both mixtures contain the same concentration of sucrose.

(a) Set up and use the apparatus provided, as shown in Fig. 1.1, to obtain the results needed to complete Table 1.1.

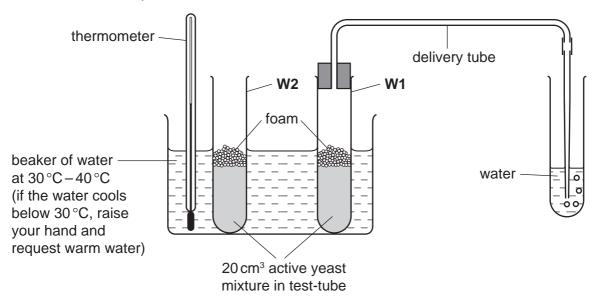


Fig. 1.1

- Connect the bung and delivery tube to test-tube **W1**, as shown in Fig. 1.1.
- Wait for two minutes before placing the open end of the delivery tube in the water, as shown in Fig. 1.1.
- Note the time and **immediately** start counting the number of gas bubbles. After two minutes, record the number of bubbles as **trial 1** in Table 1.1.
- Repeat the counting process for two minutes and record this number of bubbles as **trial 2** in Table 1.1.
- Repeat the counting process for two minutes and record this number of bubbles as **trial 3** in Table 1.1.
- Remove the bung and delivery tube from test-tube W1.
- Connect the bung and delivery tube to test-tube **W2** and repeat the above procedures.

Table 1.1

yeast mixture	number of bu	bbles of gas released ir	n two minutes
yeasi mixiure	trial 1	trial 2	trial 3
W1			
W2			

[4]

(b)	Gas	bubbles are produced in this experiment.
	(i)	State which metabolic process is being carried out by the yeast cells to produce this gas.
		[1]
	(ii)	Name this gas. [1]
	(iii)	Describe a test for this gas and the result that you would expect.
		[2]
(c)		gest why the test-tubes W1 and W2 were placed in a beaker of warm water during experiment.
		[2]
(d)		cribe and explain any differences observed in the number of bubbles of gas ased.
		[3]

(e)	State two sources of error in the method of this investigation. Suggest how to improve the method to reduce each source of error.
	source of error
	improvement
	source of error
	improvement
	[4]
	[Total: 17]

2 You are provided with two leaves, **W3** and **W4**.

(a) Make a large, labelled drawing of leaf W3.

For Examiner's Use

[4]

(b) Use the hand lens to carefully observe leaf W3 and leaf W4.

Describe one similarity and two differences that you can see. Do not include size in your comparison.

(i) similarity

[1]

(ii) differences

1

2

[2]

(iii) Leaf W3 is from a dicotyledon.

Give one visible feature to support this statement.

Fig. 2.1 shows a photomicrograph of a section of a leaf similar to **W3**.

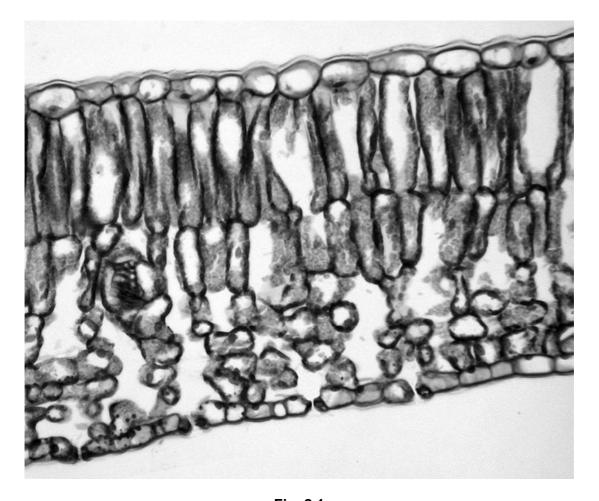


Fig. 2.1

- (c) (i) On Fig. 2.1, draw a line to label a photosynthetic cell in the palisade layer. [1]
 - (ii) Draw arrows on Fig. 2.1 to show the pathway that carbon dioxide gas must take to reach the photosynthetic cell labelled in (c)(i) from the air outside the leaf. [2]

When leaves die, they fall from the tree and are eventually decomposed.

For Examiner's Use

Some students investigated the decomposition of samples of leaves. They made drawings and weighed the samples at intervals over a period of two years.

Table 2.1 shows the results of this investigation.

Table 2.1

time / months	mass of leaves in sample / g	appearance of one leaf in the sample
0	42.5	
6	46.0	
12	32.5	
18	16.0	***************************************
24	7.5	

(d) (i)	Describe a years.	and ∈	explain	the	changes	in	appearance	of	the	leaves	during	the	two
								••••				•••••	
								••••					
								••••					
													[2]

	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
			cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
[4]	[4]		cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		
			cribe the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.	the results for the change in mass shown on the graph.		T41
							[4]
							[4]
scribe the results for the change in mass shown on the graph.						 	

[3]

(IV)	Suggest two factors that may change the rate of decomposition of the leaves.
	1
	2
	[2]

[Total: 23]

For Examiner's Use

BLANK PAGE

BLANK PAGE

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (UCLES) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

University of Cambridge International Examinations is part of the Cambridge Assessment Group. Cambridge Assessment is the brand name of University of Cambridge Local Examinations Syndicate (UCLES), which is itself a department of the University of Cambridge.