

# ANALYTICAL CHEMISTRY X-ray Crystallography

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## X-ray Crystallography

- X-rays are very short (about 0.1 nm) wavelength electromagnetic rays.
- An X-ray spectrometer consists of three main parts:
  - an X-ray source
  - a stable support on which the crystal is mounted, and which can be rotated by a measured angle around all three axes
  - a system for detecting the reflected X-rays. This can be either a photographic film or a scintillation counter.



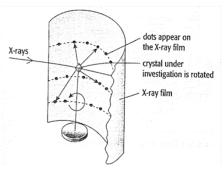
#### X-ray Crystallography

- Principle: X-rays are diffracted by the electron clouds in the atoms of the crystal.
- All atoms except hydrogen contain enough electrons to diffract the X-rays.
- The heavier the atom, has more electron → the more intense is the spot produced in its diffraction pattern.
- Diffraction of each atom is small but effect is magnified by regular repeating patterns of atoms and ions in crystal.



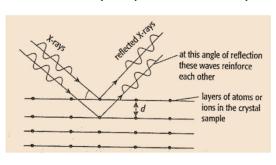
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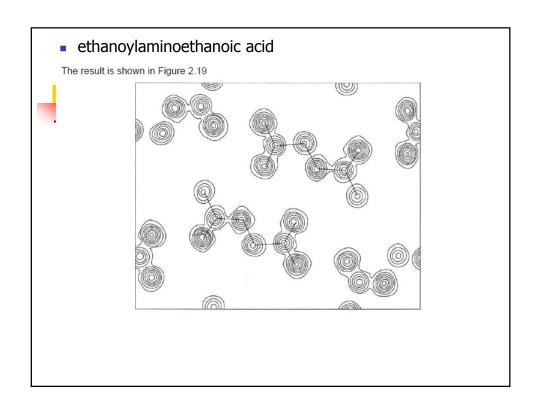
Process: crystal irradiated  $\rightarrow$  image of diffraction pattern recorded  $\rightarrow$  crystal rotated slightly (1°) and image recorded again. Process repeated until every angle of crystal is irradiated.





- By measuring the angles between the incident beam and the diffracted beams, and the relative intensities of those beams, we can piece together a picture of the electron density at all points in the unit cell.
- Electron density map constructed by computer.







## Information obtained

- Shows the **position** of most of the atoms in the structure.
- Allows measurement of bond length.
- Bond angle.
- **Shape** of molecule.
- Application : Analysis of large biological molecules (e.g DNA).
- **Disadvantage**: a very pure crystal of sample needed.

