



UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS  
General Certificate of Education  
Advanced Subsidiary Level and Advanced Level

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**CHEMISTRY**

**9701/36**

Advanced Practical Skills

**October/November 2010**

**CONFIDENTIAL INSTRUCTIONS**

**Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.**

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**The Supervisor's attention is drawn to the form on page 7 which must be completed and returned with the scripts.**

If you have any problems or queries regarding these instructions, please contact CIE

by e-mail: [international@cie.org.uk](mailto:international@cie.org.uk)

by phone: +44 1223 553554

by fax: +44 1223 553558

stating the Centre number, the nature of the query and the syllabus number quoted above.

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This document consists of **8** printed pages.



## Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution.

Only those tests described in the question paper should be attempted. Please also see under 'Apparatus' on the use of pipette fillers, safety goggles and plastic gloves.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out.

Attention is drawn in particular, to certain materials used in the examination. The following codes are used where relevant.

<b>C</b> corrosive substance	<b>F</b> highly flammable substance
<b>H</b> harmful or irritating substance	<b>O</b> oxidising substance
<b>T</b> toxic substance	<b>N</b> dangerous for the environment

The attention of Supervisors is drawn to any local regulations relating to safety and first-aid.

'Hazard Data Sheets', relating to materials used in this examination, should be available from your chemical supplier.

## Before the Examination

### 1 Access to the question paper is NOT permitted in advance of the examination.

### 2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the question paper to be completed.

**In preparing materials, the bulk quantity for each substance should be increased by 25%** as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate to within one part in two hundred of those specified.

**Supervisors are asked to carry out any confirmatory tests given on page 4 to ensure the materials supplied are appropriate.**

If the concentrations differ slightly from those specified, the Examiners will make the necessary allowance. They should be informed of the exact concentrations.

### 3 Labelling of materials

Materials must be labelled as specified in these instructions. Materials with an **FB** code number should be so labelled **without** the identities being included on the label. Where appropriate the identity of an **FB** coded chemical is given in the question paper itself.

### 4 Identity of materials

It should be noted that descriptions of solutions given in the question paper may not correspond exactly with the specifications in these Instructions. **The candidates must assume the descriptions given in the question paper.**

### 5 Size of group

In view of the difficulty of the preparation of large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

**Apparatus**

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- 2 Pipette fillers (or equivalent safety devices), safety goggles and disposable plastic gloves should be used where necessary.

3 *For each candidate*

1 × 25.0 cm<sup>3</sup> bulb form pipette  
1 × 10.0 cm<sup>3</sup> bulb form pipette  
1 × pipette filler  
1 × 250 cm<sup>3</sup> graduated (volumetric) flask  
2 × 250 cm<sup>3</sup> conical flask  
2 × 50 cm<sup>3</sup> burette  
2 × stand and burette clamp  
2 × funnel (for filling burette)  
1 × white tile  
2 × 50 cm<sup>3</sup> measuring cylinder  
1 × **stoppered conical flask** (to hold 150 cm<sup>3</sup> of solution) and **labelled Flask A**  
1 × **stoppered conical flask** (to hold 150 cm<sup>3</sup> of solution) and **labelled Flask B**  
1 × heat proof mat  
1 × Bunsen burner  
1 × test-tube holder  
4 × boiling tubes\* – *one of these should be suitable for heating a solid*  
9 × test-tubes\*  
1 × test-tube rack  
3 × teat/squeeze dropping pipettes  
1 × wash bottle of distilled water  
1 × marker pen (for labelling test-tubes)  
1 × spatula  
paper towels

\*Candidates are expected to rinse and re-use test-tubes and boiling tubes where possible. Additional tubes should be available.

## Chemicals Required

1 It is **especially important** that great care is taken that the confidential information given below does **not** reach the candidates either directly or indirectly.

2 Particular requirements

hazard	label	per candidate	identity	notes (hazards given in this column are for the raw materials)
	<b>FB 1</b>	60 cm <sup>3</sup>	0.125 mol dm <sup>-3</sup> sulfuric acid	Prepare 1 mol dm <sup>-3</sup> sulfuric acid (see p45 of 2010 Syllabus). Dilute 125 cm <sup>3</sup> of this 1 mol dm <sup>-3</sup> sulfuric acid <b>[H]</b> to 1 dm <sup>3</sup> of solution.
<b>[H]</b>	<b>FB 2</b>	150 cm <sup>3</sup>	0.050 mol dm <sup>-3</sup> sodium hydroxide	Dissolve 2.00 g of NaOH <b>[C]</b> in each dm <sup>3</sup> of solution.
<b>Check on suitability of reagents.</b> <i>Check, and adjust where necessary, the concentration of FB 2 by titrating 10.0 cm<sup>3</sup> of FB 1 with FB 2 using phenolphthalein as indicator. Adjust the concentration of FB 2 to give a titration reading between 49.20 and 49.60 cm<sup>3</sup>.</i>				
<b>[H]</b>	<b>FB 4</b>	200 cm <sup>3</sup>	0.050 mol dm <sup>-3</sup> sodium hydroxide	This is 200 cm <sup>3</sup> of <b>FB 2</b> following the adjustment of the concentration of <b>FB 2</b> .
	<b>FB 5</b>	150 cm <sup>3</sup>	0.200 mol dm <sup>-3</sup> propanoic acid	Dissolve 14.80 g of C <sub>2</sub> H <sub>5</sub> CO <sub>2</sub> H <b>[C]</b> in each dm <sup>3</sup> of solution. <i>Weigh the liquid by adding to a tared container. e.g. beaker.</i>
<b>[H][F]</b>	<b>FB 6</b>	150 cm <sup>3</sup>	2-methylpropan-1-ol CH <sub>3</sub> CH(CH <sub>3</sub> )CH <sub>2</sub> OH	
	<b>FB 7</b>	20 cm <sup>3</sup>	0.1 mol dm <sup>-3</sup> ammonium iron(II) sulfate	Dissolve 39.2 g (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> ·FeSO <sub>4</sub> ·6H <sub>2</sub> O <b>[H]</b> in each dm <sup>3</sup> of solution.
	<b>FB 8</b>	20 cm <sup>3</sup>	0.1 mol dm <sup>-3</sup> sodium sulfate	Dissolve 14.2 g of Na <sub>2</sub> SO <sub>4</sub> in each dm <sup>3</sup> of solution.
	<b>FB 9</b>	20 cm <sup>3</sup>	0.1 mol dm <sup>-3</sup> calcium chloride	Dissolve 21.9 g of CaCl <sub>2</sub> ·6H <sub>2</sub> O <b>[H]</b> in each dm <sup>3</sup> of solution.
<b>[H]</b>	<b>FB 10</b>	1 g	basic copper carbonate	<i>Any sample of copper(II) carbonate will be suitable. The most common formula is CuCO<sub>3</sub>·Cu(OH)<sub>2</sub>·H<sub>2</sub>O <b>[H]</b></i>
<b>[F]</b>	<b>Phenolphthalein indicator</b>	10 cm <sup>3</sup>	phenolphthalein indicator	see p45 of 2010 Syllabus

- 3 The standard bench reagents required are set out below. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the fact that such an arrangement may enhance the opportunity for malpractice between candidates.

hazard	label	notes
[H]	dilute hydrochloric acid	See identity details and preparation instructions on page 45 of the 2010 syllabus
[C]	dilute nitric acid	
[H]	dilute sulfuric acid	
[C]	aqueous sodium hydroxide	
[H]	aqueous ammonia	
[T]	0.1 mol dm <sup>-3</sup> barium chloride or	
[H]	0.1 mol dm <sup>-3</sup> barium nitrate	
[H][N]	0.05 mol dm <sup>-3</sup> silver nitrate	

The following materials and apparatus should be available.

red and blue litmus paper, plain filter paper strips for use with dichromate(VI), aluminium foil for testing for nitrate/nitrite, wooden splints, the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide

### Responsibilities of the Supervisor during the Examination

- 1 The Supervisor, or other competent chemist **must carry out the experiments in Question 1 and Question 2** and complete tables of readings on a spare copy of the question paper which should be labelled 'Supervisor's Results'.

**This should be done for:  
each session held and each laboratory used in that session, and each set of solutions supplied.**

**N.B. The question paper cover requests the candidate to fill in details of the examination session and the laboratory used for the examination.**

**It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.**

- 2 The Supervisor must complete the Report Form on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Report Form. A copy of the Report Form must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

### After the Examination

**Each envelope returned to Cambridge must contain the following items.**

- 1 The scripts of those candidates specified on the bar code label provided.
- 2 A copy of the Supervisor's Report relevant to the candidates in 1.
- 3 A copy of the Report Form, including details of any difficulties experienced by candidates (see pages 7 and 8).
- 4 The Attendance Register.
- 5 A Seating Plan for each session/laboratory.

**Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.**

### COLOUR-BLINDNESS

With regard to colour-blindness – a minor handicap, relatively common in males – it is permissible to advise candidates who request assistance on colours of, for example precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a 'Special Consideration' application for this handicap.

**REPORT FORM**

**This form must be completed and sent to the Examiner in the envelope with the scripts.**

Centre Number ..... Name of Centre .....

**1 Supervisor's Results**

Please submit details of the readings obtained in **Question 1 and Question 2** on a spare copy of the question paper clearly marked 'Supervisor's Results' and showing the **Centre number and appropriate session/laboratory number**.

**2 The candidate numbers of candidates attending each session were:**

*First Session*

*Second Session*

**3 The Supervisor is required to give details overleaf of any difficulties experienced by particular candidates, giving names and candidate numbers. These should include reference to:**

- (a)** any general difficulties encountered in making preparation;
- (b)** difficulties due to faulty apparatus or materials;
- (c)** accidents to apparatus or materials;
- (d)** assistance with respect to colour-blindness.

Other cases of hardship, e.g. illness, temporary disability, should be reported direct to CIE on the normal 'Application for Special Consideration' form.

**4 A plan of work benches, giving details by candidate numbers of the places occupied by the candidates for each experiment for each session, must be enclosed with the scripts.**





Report on any difficulties experienced by candidates.

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