## **4.2 ANSWERS TO EXERCISES**

## 4.2 Exercise 1

 $[\underline{N_2O_4}] \ mol^{\text{-}1}dm^3$ [CH<sub>3</sub>CH<sub>2</sub>CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>][H<sub>2</sub>O] no units 1. i) ii) [CH<sub>3</sub>CH<sub>2</sub>CO<sub>2</sub>H][CH<sub>3</sub>CH<sub>2</sub>OH]  $[NO_2]^2$ 

 $[HI]^2$  no units  $[SO_3]^2$  $mol^{-1}dm^3$ iii) iv)  $\overline{[H_2]}[I_2]$  $[SO_2]^2[O_2]$ 

 $\frac{[NH_3]^2}{[N_2][H_2]^3} mol^{\text{-}2} dm^6$ v)

2. 0.042 moldm<sup>-3</sup> 3. 1.33 x 10<sup>-3</sup> moldm<sup>-3</sup> 4. 56.3

5. 0.2 moldm<sup>-3</sup> 6. 6.6 mol<sup>-2</sup>dm<sup>6</sup> 7. 1013 mol<sup>-1</sup>dm<sup>3</sup>

8. 3.2 x 10<sup>-4</sup> moldm<sup>-3</sup>
11. 12 moldm<sup>-3</sup> 9. 46.3 mol<sup>-2</sup>dm<sup>6</sup> 10.4.12