UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Advanced Subsidiary Level and Advanced Level

CHEMISTRY 9701/05

Paper 5 Practical Test

May/June 2005

CONFIDENTIAL INSTRUCTIONS

Great care should be taken that any confidential information given does not reach the candidates either directly or indirectly.

ERRATUM

Page 4, Detailed Identities of Chemicals Required

Solutions, Question 1 and Question 2

For Centres using approximately 36% hydrochloric acid (11.7 M), the following instructions should be used to prepare solutions **FB 1** and **FB 4**:

FB 1 is 2.0 mol dm⁻³ hydrochloric acid, prepared by diluting 172 cm³ of concentrated hydrochloric acid to 1 dm³.

FB 4 is 2.2 mol dm⁻³ hydrochloric acid, *prepared by diluting 189 cm*³ of concentrated hydrochloric acid to 1 dm³.

Titrate 25 cm³ portions of **FB 4** against **FB 3** and adjust the concentration of **FB 4** to give a titre of 36.7 ± 0.2 cm³.

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