UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS
General Certificate of Education
Advanced Subsidiary Level and Advanced Level

CHEMISTRY **9701/33**

Paper 33 Practical Test **October/November 2009**

CONFIDENTIAL INSTRUCTIONS

Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.

The Supervisor's attention is drawn to the form on page 7 which must be completed and returned with the scripts.

If you have any problems or queries regarding these instructions, please contact CIE
by e-mail: International@cie.org.uk
by phone: +44 1223 553554
by fax: +44 1223 553558
stating the Centre number, the nature of the query and the syllabus number quoted above.

This document consists of **8** printed pages.

DC (SJF5338) 11758/3

UNIVERSITY of CAMBRIDGE International Examinations

[Turn over

Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution.

Only those tests described in the question paper should be attempted. Please also see under 'Apparatus' on the use of pipette fillers, safety goggles and plastic gloves.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out.

Attention is drawn in particular, to certain materials used in the examination. The following codes are used where relevant.

C	corrosive substance	**F**	highly flammable substance
H	harmful or irritating substance	**O**	oxidising substance
T	toxic substance	**N**	dangerous for the environment

The attention of Supervisors is drawn to any local regulations relating to safety and first-aid.

'Hazard Data Sheets', relating to materials used in this examination, should be available from your chemical supplier.

Before the Examination

1 Access to the question paper is NOT permitted in advance of the examination.

2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the question paper to be completed.

In preparing materials, the bulk quantity for each substance should be increased by 25% as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate to within one part in two hundred of those specified.

Supervisors are asked to carry out any confirmatory tests given on page 4 to ensure the materials supplied are appropriate.

If the concentrations differ slightly from those specified, the Examiners will make the necessary allowance. They should be informed of the exact concentrations.

3 Labelling of materials

Materials must be labelled as specified in these instructions. Materials with an **FA** code number should be so labelled **without** the identities being included on the label. Where appropriate the identity of an **FA** coded chemical is given in the question paper itself.

4 Identity of materials

It should be noted that descriptions of solutions given in the question paper may not correspond exactly with the specifications in these Instructions. **The candidates must assume the descriptions given in the question paper.**

5 Size of group

In view of the difficulty of the preparation of large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

Apparatus

1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.

2 Pipette fillers (or equivalent safety devices), safety goggles and disposable plastic gloves should be used where necessary.

3 *For each candidate*

1 × heat proof mat
1 × Bunsen burner
1 × tripod
1 × pipe clay triangle
1 × gauze
1 × crucible (at least 15 cm^3 capacity)
1 × crucible tongs
1 × 50 cm^3 burette
1 × stand and burette clamp
1 × funnel (for filling burette)
1 × white tile
1 × 25 cm^3 measuring cylinder
1 × 250 cm^3 conical flask
1 × 25 cm^3 bulb pipette
1 × pipette filler
1 × wash bottle of distilled water
8 × test-tubes *****
1 × boiling-tube
1 × 250 cm^3 beaker
1 × test-tube rack
2 × teat/squeeze pipettes
paper towels

Access to a balance weighing to 1 decimal place or better. (1 balance per 8-12 candidates)

Where access to the balance is limited – some candidates should be instructed to start the examination with the titration.

***** Candidates are expected to rinse and re-use test-tubes where possible.
Additional test-tubes should be available.

Chemicals Required

1 It is ***especially important*** that great care is taken that the confidential information given below does **not** reach the candidates either directly or indirectly.

2 Particular requirements

hazard	label	per candidate	identity	notes (hazards given in this column are for the raw materials)
[H]	**FA 1**	3 g	hydrated iron(II) sulfate $FeSO_4.7H_2O$	As iron(II) salts are readily oxidised, it is recommended that a **fresh supply be purchased** for the examination. Provide the solid in a stoppered container.
[C]	**FA 2**	120 cm^3	approximately 0.05 mol dm^{-3} ammonium iron(II) sulfate	Dissolve 19.3 – 19.9 g of **freshly purchased** $(NH_4)_2SO_4.FeSO_4.6H_2O$ **[H]** in 500 cm^3 of 1.0 mol dm^{-3} sulfuric acid, H_2SO_4 **[C]**. Make up the solution to 1 dm^3 with distilled water. Issue to candidates in a stoppered container or in a beaker covered with "clingfim" or "gladwrap".
[H] [N]	**FA 3**	150 cm^3	0.010 mol dm^{-3} potassium manganate(VII)	Dissolve 1.58 g of $KMnO_4$ **[O] [H] [N]** in each dm^3 of solution.
[C]	**FA 4**	100 cm^3	1.0 mol dm^{-3} sulfuric acid	See instructions for dilute sulfuric acid in **3** on page 5.
Check on suitability of reagents. *Pipette 25.0 cm^3 of* **FA 2** *into a conical flask and add 20 cm^3 of* **FA 4***. Titrate with* **FA 3** *– a titre in the region of 24.5 cm^3 to 25.5 cm^3 should be obtained. Adjust the concentration of one solution, if necessary, to come within this range.*				
[T] [N]	**FA 5**	20 cm^3	1.0 mol dm^{-3} sodium nitrite	Dissolve 69.0 g of **freshly purchased** $NaNO_2$ **[T] [O] [N]** in each dm^3 of solution.
	FA 6	20 cm^3	0.1 mol dm^{-3} sodium chloride	Dissolve 5.8 g of NaCl in each dm^3 of solution.
[H]	**FA 7**	20 cm^3	0.1 mol dm^{-3} potassium iodide	Dissolve 16.6 g of KI **[H]** in each dm^3 of solution.
	FA 8	20 cm^3	distilled water	
[T] [N]	**FA 9**	10 cm^3	0.1 mol dm^{-3} lead(II) nitrate	Dissolve 33.1 g of $Pb(NO_3)_2$ **[T] [O] [N]** in each dm^3 of solution.
[T] [N]	**FA 10**	10 cm^3	0.1mol dm^{-3} potassium chromate(VI)	Dissolve 19.4 g of K_2CrO_4 in each dm^3 of solution. Warming may be necessary. **Note:** A fume cupboard should be used when handling the solid if it is a fine powder. *Use of plastic gloves may be considered necessary for candidates when handling this solution.*
[F]	**ethanol**	10 cm^3	ethanol	Industrial Methylated Spirit (IMS) may be used.

NOTE: Heating of a solid during the practical examination may release a small amount of sulfur dioxide **[T] [C]** which can cause respiratory distress in some people. The laboratory should be well ventilated.

3 The standard bench reagents specifically required are set out below. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the fact that such an arrangement may enhance the opportunity for malpractice between candidates.

hazard	label	identity	notes (hazards given in this column are for the raw materials)
[H]	**dilute hydrochloric acid**	2.0 mol dm^{-3} HC*l*	Dilute 172 cm^3 of concentrated (35% w/w; approximately 11 mol dm^{-3}) acid **[C]** to 1 dm^3.
[C]	**dilute nitric acid**	2.0 mol dm^{-3} HNO_3	Dilute 128 cm^3 of concentrated (70% w/v) acid **[C][O]** to 1 dm^3.
[C]	**dilute sulfuric acid**	1.0 mol dm^{-3} H_2SO_4	Cautiously pour 55 cm^3 of concentrated (98%) sulfuric acid **[C]** into 500 cm^3 of distilled water with continuous stirring. Make the solution up to 1 dm^3 with distilled water. **Care** – *concentrated sulfuric acid is very corrosive.*
[H]	**aqueous ammonia**	2.0 mol dm^{-3} NH_3	Dilute 112 cm^3 of concentrated (35% w/w) ammonia **[C][N]** to 1 dm^3.
[T] **[H]**	**0.1 mol dm^{-3} barium chloride** **or** **0.1 mol dm^{-3} barium nitrate**	0.1 mol dm^{-3} $BaCl_2$ **or** 0.1 mol dm^{-3} $Ba(NO_3)_2$	Dissolve 24.4 g of $BaCl_2.2H_2O$ **[T]** in each dm^3 of solution. Dissolve 26.1 g of $Ba(NO_3)_2$ **[H]** in each dm^3 of solution.
[H]	**0.05 mol dm^{-3} silver nitrate**	0.05 mol dm^{-3} silver nitrate	Dissolve 8.5 g of $AgNO_3$ **[C][N]** in each dm^3 of solution.
[T] [N]	**0.1 mol dm^{-3} lead(II) nitrate**	0.1 mol dm^{-3} lead nitrate	Dissolve 33.1 g of $Pb(NO_3)_2$ **[T][O][N]** in each dm^3 of solution.

The reagents, materials and apparatus to test the gases listed in the syllabus must be available to candidates. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the fact that such an arrangement may enhance the opportunity for malpractice between candidates.

hazard	label	identity	notes
[H]	**limewater**	saturated aqueous calcium hydroxide, $Ca(OH)_2$	Prepare fresh limewater by leaving distilled water to stand over solid calcium hydroxide, **[H]**, for several days, shaking occasionally. Decant or filter the solution.
[T][N]	**acidified aqueous potassium dichromate(VI)**	0.05 mol dm^{-3} $K_2Cr_2O_7$ 0.05 mol dm^{-3} H_2SO_4	Dissolve 14.8 g of $K_2Cr_2O_7$ **[T][N]** in 50.0 cm^3 of 1.0 mol dm^{-3} sulfuric acid **[H]**. Make the solution up to 1 dm^3 with distilled water. *The use of plastic gloves may be considered to prevent contact with skin.*
red and blue litmus paper, plain filter paper strips for use with aqueous potassium dichromate(VI), aluminium foil for testing for nitrate/nitrite, wooden splints, the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide			

Responsibilities of the Supervisor during the Examination

1 The Supervisor, or other competent chemist **must carry out both experiments in question 1** and complete tables of readings on a spare copy of the question paper which should be labelled 'Supervisor's Results'.

This should be done for:
each session held and each laboratory used in that session, and each set of solutions supplied.

N.B. The question paper cover requests the candidate to fill in details of the examination session and the laboratory used for the examination.

It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.

2 The Supervisor must complete the Report Form on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Report Form. A copy of the Report Form must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

After the Examination

Each envelope returned to Cambridge must contain the following items.

1 The scripts of those candidates specified on the bar code label provided.

2 A copy of the Supervisor's Report relevant to the candidates in **1**.

3 A copy of the Report Form, including details of any difficulties experienced by candidates (see pages 7 and 8).

4 The Attendance Register.

5 **A Seating Plan for each session/laboratory**.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

COLOUR BLINDNESS

With regard to colour-blindness – a minor handicap, relatively common in males – it is permissible to advise candidates who request assistance on colours of, for example precipitates and solutions (especially titration end-points). Please include with the scripts a note of the index numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a 'Special Consideration' application for this handicap.

REPORT FORM

This form must be completed and sent to the Examiner in the envelope with the scripts.

Centre Number …… Name of Centre ……

1 Supervisor's Results

Please submit details of the readings obtained **in both experiments in Question 1** on a spare copy of the question paper clearly marked 'Supervisor's Results' **and showing the Centre number and appropriate session/laboratory number**.

2 The index numbers of candidates attending each session were:

First Session	*Second Session*

3 The Supervisor is required to give details overleaf of any difficulties experienced by particular candidates, giving names and index numbers. These should include reference to:

(a) any general difficulties encountered in making preparation;

(b) difficulties due to faulty apparatus or materials;

(c) accidents to apparatus or materials;

(d) assistance with respect to colour-blindness.

Other cases of hardship, e.g. illness, temporary disability, should be reported direct to CIE on the normal 'Application for Special Consideration' form.

4 A plan of work benches, giving details by index numbers of the places occupied by the candidates for each experiment for each session, must be enclosed with the scripts.

Report on any difficulties experienced by candidates.