

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

Specimen for 2007

GCE A/AS LEVEL

MARK SCHEME
MAXIMUM MARK: 40
SYLLABUS/COMPONENT: 9701/31  ADVANCED PRACTICAL SKILLS

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Skill	Total marks	Breakdown of marks/expectations		Question 1	Question 2
Manipulation, measurement and observation	16 marks	Successful collection of data and observations	8 marks	2	6
		Decisions relating to measurements or observations	8 marks	5	3
Presentation of data and observations	12 marks	Recording data and observations	5 marks	3	1
		Display of calculation and reasoning	3 marks	3	0
		Data layout	4 marks	4	0
Analysis, conclusions and evaluation	12 marks	Interpretation of data or observations and identifying sources of error	6 marks	2	4
		Drawing conclusions	5 marks	3	1
		Suggesting improvements	3 marks	1	1

MMO = Manipulation, measurement and observation

Collection = Successful collection of data and observations

Decisions = Decisions relating to measurements or observations

PDO = Presentation of data and observations

Recording = Recording data and observations

Display = Display of calculation and reasoning

Layout = Data layout

ACE = Analysis, conclusions and evaluation

Interpretation = Interpretation of data or observations and identifying sources of error

Conclusions = Drawing conclusions

Improvements = Suggesting Improvements

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Question		Sections	Learning outcomes	Indicative material	mark
1	(a)	PDO Display	<ul style="list-style-type: none"><li>show their working in calculations, and the key steps in their reasoning</li></ul>	correct working for volume of H <sub>2</sub> SO <sub>4</sub>	1
	(b)	MMO decisions	<ul style="list-style-type: none"><li>decide how many tests or observations to perform</li></ul>	appropriate volume of acid added each time (between 2 and 4 cm <sup>3</sup> ) volumes spanning a sufficient range each side of calculated end point (between 20 and 30 cm <sup>3</sup> below end point and 10 and 20 cm <sup>3</sup> above end point)	1 1
	(c)	PDO Recording	<ul style="list-style-type: none"><li>draw up table in advance of taking readings so that they do not have to copy results</li><li>use column headings that include both the quantity and the unit and that conform to accepted scientific conventions</li><li>record raw readings of a quantity to the same degree of precision</li></ul>	no evidence on script of table having been produced or added to after measurements made;	1
				volume, temperature and ΔT columns correctly labelled	1
				volumes and temperatures recorded to consistent significant figures	1
		MMO collection	<ul style="list-style-type: none"><li>making measurements using burettes and thermometers</li></ul>	all volumes recorded to 0.05 cm <sup>3</sup> ; all temperatures recorded to 0.5 °C;	1 1
		MMO decisions	<ul style="list-style-type: none"><li>make and record sufficient, accurate measurements</li></ul>	volume at which max temp rise recorded within 5 cm <sup>3</sup> of Supervisor; ΔT for highest temp within 1 °C of that obtained by Supervisor (1 of these two marks if in range +1 °C to 3 °C)	1 2

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	(d)	PDO Layout	<ul style="list-style-type: none"> <li>plot appropriate variables on clearly labelled x- and y-axes</li> <li>choose suitable scales for graph axes</li> <li>plot all points to an appropriate accuracy.</li> </ul> <p>follow the ASE recommendations for putting lines on graphs</p>	<p><math>\Delta T</math> plotted on y-axis and volume of acid on x-axis, correctly labelled including units;</p> <p>suitable scales selected;</p> <p>points plotted as fine cross or encircled dot within <math>\frac{1}{2}</math> small square in either direction;</p> <p>two smooth intersecting curves drawn</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p>
	(e)	ACE Interpretation	<ul style="list-style-type: none"> <li>find an unknown value by using intercept on a graph</li> </ul>	reading the volume of $H_2SO_4$ at the end-point from the intercept of the graph	1
	(f)	PDO Display	<ul style="list-style-type: none"> <li>show working in calculations, and the key steps in reasoning</li> <li>use the correct number of significant figures for calculated quantities</li> </ul>	<p>shows working and explains the steps in the calculation;</p> <p>calculates concentration to same sf as titre/volume information recorded</p>	<p>1</p> <p>1</p>
	(g)	ACE Conclusions	<ul style="list-style-type: none"> <li>draw conclusions from an experiment, giving an outline description of the main features of the data, considering whether experimental data supports a given hypothesis.</li> </ul>	<p>first part of hypothesis not supported as the graph is not a straight line. (hypothesis supported is acceptable if the graph is a straight line)</p> <p>shape of graph described</p> <p>second part of hypothesis is supported as temperature falls after the end-point</p>	<p>1</p> <p>1</p> <p>1</p>
	(h)	ACE Interpretation	<ul style="list-style-type: none"> <li>identify the most significant sources of error in an experiment</li> </ul>	<p>comments on the closer spacing of temperatures at higher values or curve with decreasing gradient;</p> <p>explains that heat loss is greater/more rapid at higher temperatures</p>	<p>1</p> <p>1</p>
	(i)	ACE Interpretation	<ul style="list-style-type: none"> <li>estimate, quantitatively, the uncertainty in quantitative measurements</li> <li>express such uncertainty as an actual or percentage error</li> </ul>	calculates 0.05 or 0.10 as a % of the end-point volume	1
	(j)	ACE Improvements	<ul style="list-style-type: none"> <li>suggest modifications that will improve the accuracy of the experiment</li> </ul>	calculates (total volume x $\Delta T$ x 4.3)	1

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2	(a)	MMO Decisions	<ul style="list-style-type: none"> <li>selecting a suitable reagent</li> </ul>	use of $\text{Pb}(\text{NO}_3)_2$ or $\text{AgNO}_3/\text{NH}_3(\text{aq})$ as reagent;	1
		MMO Collection	<ul style="list-style-type: none"> <li>use apparatus to collect an appropriate quantity of data or observations, including subtle differences in colour, solubility or quantity of materials</li> </ul>	records appropriate observation for selected reagent	1
	(b)	MMO Decisions	<ul style="list-style-type: none"> <li>selecting a suitable reagent</li> </ul>	use of $\text{Pb}(\text{NO}_3)_2$ or $\text{AgNO}_3/\text{NH}_3(\text{aq})$ as reagent;	1
		MMO Collection	<ul style="list-style-type: none"> <li>use apparatus to collect an appropriate quantity of data or observations, including subtle differences in colour, solubility or quantity of materials</li> </ul>	records appropriate observation for selected reagent	1
		ACE conclusions	<ul style="list-style-type: none"> <li>draw conclusions from interpretations of observations</li> </ul>	draws a conclusion appropriate to the observations in (a) and (b)	1
	(c)-(f)	MMO collection	<p>follow instructions given in the form of written instructions</p> <p>use apparatus to collect an appropriate quantity of data or observations, including subtle differences in colour, solubility or quantity of materials</p>	<p>all tests attempted and some observation recorded</p> <p>at least three initial precipitates correctly recorded</p> <p>colours of precipitates correctly described</p> <p>solubility of precipitates in excess <math>\text{NaOH}/\text{NH}_3</math> correctly described</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p>
		MMO decisions	make appropriate qualitative observations	appropriate test for ammonia gas recorded	1
		PDO recording	record observations to the same level of detail	consistent standard in recording observations i.e. all precipitates and their solubilities in excess recorded	1

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	(g)	ACE Interpretation	<ul style="list-style-type: none"> <li>describes and summarises the key points of a set of observations.</li> </ul>	<p>explains how the observations identify and confirm the presence of <math>\text{Ba}^{2+}</math>.</p> <p>explains how the reaction with sodium hydroxide and ammonia identifies <math>\text{Al}^{3+}</math> or <math>\text{Pb}^{2+}</math> as the unknown cation</p> <p>explains which tests eliminate <math>\text{Pb}^{2+}</math></p>	<p>1</p> <p>1</p> <p>1</p>
	(h)	ACE Improvements	<ul style="list-style-type: none"> <li>suggest ways in which to extend the investigation</li> </ul>	suggests dilute acid to liberate NO	1