[6]

y (!)	Name <b>two</b> se	conually Si	ii uolui <del>U</del> S I	in proteins	·-			
I			II					
(ii)	Draw simplifie stabilised.	ed diagram	ns of each	structure,	showing	the bondir	ng by whic	ch each is
ı			II					
<b>.</b>	D : :	4		. 41		- <b>c</b>	! II-	
	P is an importa ite a simple eq						in cells.	
	P is an importa ite a simple equ						in cells.	
							in cells.	
Wri	ite a simple equ	uation to re	epresent t	he hydroly	sis of ATI	P		rate of
  <b>:)</b> The		uation to re	epresent t	he hydroly	sis of ATI	P		rate of
Wri  The hyd	ite a simple equestions in the perfect of the perfe	uation to re	epresent t	he hydroly	sis of ATI	P		rate of 0.300
Wri  Wri  And	ite a simple equestions in the parties in the parti	uation to re	epresent t btained a f the enzy	t different	sis of ATI	tions of A	TP for the	_
Wri	e following resurted the polysis in the polysis of dm <sup>-3</sup>	ults were o presence o	btained a f the enzy	t different me myosi 0.075 0.0140	concentra n. 0.100 0.0158	0.150 0.0176	0.200 0.0187	0.300
[ATF /mm/rate /mm/	e following result folysis in the polysis in the polysis ol dm <sup>-3</sup>	ults were opresence of 0.025  0.0065	btained a f the enzy 0.050 0.0115	t different me myosi 0.075 0.0140	concentra n. 0.100 0.0158	0.150 0.0176 on the x-a	0.200 0.0187 xis, and ra	0.300
(i)	e following result following result following resu	ults were opresence of 0.025  0.0065	btained a f the enzy 0.050 0.0115	t different me myosi 0.075 0.0140	concentra n. 0.100 0.0158	0.150 0.0176 on the x-a	0.200 0.0187 xis, and ra	0.300
[ATF /mm/rate /mm/	e following result following result following resu	ults were obresence of 0.025  0.0065  per to plot	btained a f the enzy 0.050 0.0115 these res	t different me myosi 0.075 0.0140 sults, plotti	concentran.  0.100  0.0158  ng [ATP] out	tions of A  0.150  0.0176  on the x-a  e of inhibi	TP for the  0.200  0.0187  xis, and rates tor it is.	0.300 0.0196 ate on the

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[4]

# **Applications of Analytical Chemistry**

2.	(a)	(i)	State what is meant by	the term	partition	coefficient.
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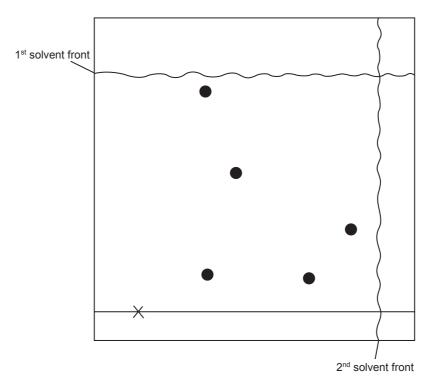
(ii)	A solution of 5.00 g of an organic compound X in 50 cm <sup>3</sup> of water was shaken with 1	00
	cm <sup>3</sup> of ether. After separation, the aqueous solution was found to contain 0.80 g of X	Χ.

Determine the partition coefficient of X between ether and water.

(b)	Explain briefly how the separation of different components in mixtures is achieved in each of the following chromatographic techniques.					
	(i) paper chromatography					
	(ii)	thin layer chromatography				
			[4]			

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(c) The diagram shows a two-way chromatogram carried out on the products of the hydrolysis of a polypeptide.



x = start point

(i)	How many different amino acids were present in the sample?
(ii)	What could have been use to make the spots visible?

(iii) Sketch the chromatogram you might expect if only the first solvent had been used.

[4]

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# **Materials and Design**

3.

(a)	Sta	te what you understand by the term <i>nanotechnology</i> .	
	••••		[2]
(b)		notubes' are rolled up cylinders of graphite with diameters of about 1 nanometre, and gths micrometres.	
	(i)	If a nanotube is 5 micrometres long, how many diameters does this represent?	
	(ii)	These tubes are believed to be stronger than steel. Suggest a possible use for nanotubes.	
(	(iii)	One problem in the synthesis of nanotubes is that a mixture of tubes of different length and orientations is produced. Suggest why this is a problem.	า
			[5]
(c)	dru tecl	e delivery of cancer-destroying drugs has, in the past, been by injection of the relevant g into the bloodstream, allowing it to be carried around the body to the tumour. New hniques have been developed which rely on binding the relevant drug molecule to an cyme.	
		ggest the advantages of this new technique, both in economic terms and in terms of the ect on the patient.	)
			[4]

