



UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Advanced Level

CANDIDATE NAME					
CENTRE NUMBER			CANDIDATE NUMBER		



PHYSICS 9702/41

Paper 4 A2 Structured Questions

October/November 2013

2 hours

Candidates answer on the Question Paper.

No Additional Materials are required.

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a pencil for any diagrams, graphs or rough working.

Do not use staples, paper clips, highlighters, glue or correction fluid.

DO NOT WRITE IN ANY BARCODES.

Answer all questions.

Electronic calculators may be used.

You may lose marks if you do not show your working or if you do not use appropriate units.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.

For Examiner's Use		
1		
2		
3		
4		
5		
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7		
8		
9		
10		
11		
12		
Total		

This document consists of 22 printed pages and 2 blank pages.



Data

speed of light in free space,	$c = 3.00 \times 10^8 \mathrm{ms^{-1}}$
permeability of free space,	$\mu_0 = 4\pi \times 10^{-7} \mathrm{H} \mathrm{m}^{-1}$
permittivity of free space,	$\varepsilon_0 = 8.85 \times 10^{-12} \mathrm{F} \mathrm{m}^{-1}$
	$(\frac{1}{4\pi\varepsilon_0} = 8.99 \times 10^9 \mathrm{mF^{-1}})$
elementary charge,	$e = 1.60 \times 10^{-19} \mathrm{C}$
the Planck constant,	$h = 6.63 \times 10^{-34} \mathrm{Js}$
unified atomic mass constant,	$u = 1.66 \times 10^{-27} \text{ kg}$
rest mass of electron,	$m_{\rm e} = 9.11 \times 10^{-31} \rm kg$
rest mass of proton,	$m_{\rm p} = 1.67 \times 10^{-27} \rm kg$
molar gas constant,	$R = 8.31 \text{ JK}^{-1} \text{ mol}^{-1}$
the Avogadro constant,	$N_{\rm A} = 6.02 \times 10^{23} \rm mol^{-1}$
the Boltzmann constant,	$k = 1.38 \times 10^{-23} \mathrm{JK^{-1}}$
gravitational constant,	$G = 6.67 \times 10^{-11} \mathrm{N}\mathrm{m}^2\mathrm{kg}^{-2}$
acceleration of free fall,	$g = 9.81 \mathrm{ms^{-2}}$

Formulae

uniformly accelerated motion,	$s = ut + \frac{1}{2}at^2$
	$v^2 = u^2 + 2as$

work done on/by a gas,
$$W = p\Delta V$$

gravitational potential,
$$\phi = -\frac{Gm}{r}$$

hydrostatic pressure,
$$p = \rho gh$$

pressure of an ideal gas,
$$p = \frac{1}{3} \frac{Nm}{V} < c^2 >$$

simple harmonic motion,
$$a = -\omega^2 x$$

velocity of particle in s.h.m.,
$$v = v_0 \cos \omega t$$

$$v = \pm \omega \sqrt{(x_0^2 - x^2)}$$

electric potential,
$$V = \frac{Q}{4\pi\varepsilon_0 r}$$

capacitors in series,
$$1/C = 1/C_1 + 1/C_2 + \dots$$

capacitors in parallel,
$$C = C_1 + C_2 + \dots$$

energy of charged capacitor,
$$W = \frac{1}{2}QV$$

resistors in series,
$$R = R_1 + R_2 + \dots$$

resistors in parallel,
$$1/R = 1/R_1 + 1/R_2 + \dots$$

alternating current/voltage,
$$x = x_0 \sin \omega t$$

radioactive decay,
$$x = x_0 \exp(-\lambda t)$$

decay constant,
$$\lambda = \frac{0.693}{t_{\frac{1}{2}}}$$

Section A

For Examiner's Use

Answer **all** the questions in the spaces provided.

1	(a)	Def	ine gravitational potential at a point.
			[2]
	(b)		e Moon may be considered to be an isolated sphere of radius 1.74×10^3 km with its ss of 7.35×10^{22} kg concentrated at its centre.
		(i)	A rock of mass 4.50 kg is situated on the surface of the Moon. Show that the change in gravitational potential energy of the rock in moving it from the Moon's surface to infinity is 1.27×10^7 J.
			[1]
		(ii)	The escape speed of the rock is the minimum speed that the rock must be given when it is on the Moon's surface so that it can escape to infinity. Use the answer in (i) to determine the escape speed. Explain your working.
			speed = ms ⁻¹ [2]
	(c)		Moon in (b) is assumed to be isolated in space. The Moon does, in fact, orbit the
			th. te and explain whether the minimum speed for the rock to reach the Earth from the face of the Moon is different from the escape speed calculated in (b) .
			[2]

The	e pro	duct of the pressure p and the volume V of an ideal gas is given by the expression	F
		$\rho V = \frac{1}{3} Nm < c^2 >$	Exa
whe	ere n	n is the mass of one molecule of the gas.	
(a)	Sta	te the meaning of the symbol	
	(i)	N,	
		[1]	
	(ii)	$\langle c^2 \rangle$.	
		[1]	
(b)	The	e product pV is also given by the expression	
		pV = NkT.	
		duce an expression, in terms of the Boltzmann constant k and the thermodynamic operature T , for the mean kinetic energy of a molecule of the ideal gas.	
		[2]	
(c)	A c	ylinder contains 1.0 mol of an ideal gas.	
	(i)	The volume of the cylinder is constant.	
		Calculate the energy required to raise the temperature of the gas by 1.0 kelvin.	
	/** ·	energy =	
	(ii)	The volume of the cylinder is now allowed to increase so that the gas remains at constant pressure when it is heated. Explain whether the energy required to raise the temperature of the gas by 1.0 kelvin is now different from your answer in (i).	
		ro.	
		[2]	

3 A metal ball is suspended from a fixed point by means of a string, as illustrated in Fig. 3.1.



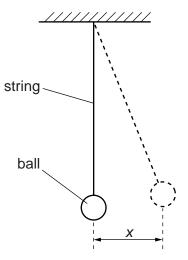


Fig. 3.1

The ball is given a small displacement and then released. The variation with time t of the displacement x of the ball is shown in Fig. 3.2.

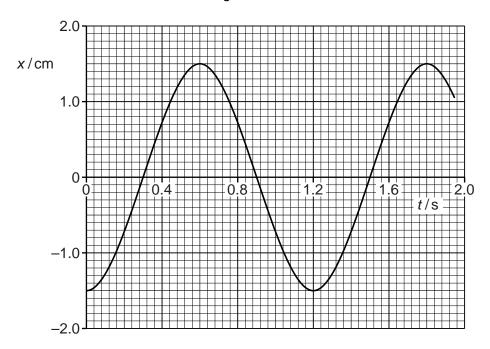
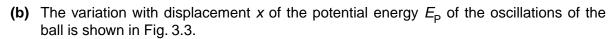


Fig. 3.2

(a) (i) State two times at which the speed of the ball is a maximum.

(ii) Show that the maximum speed of the ball is approximately $0.08 \,\mathrm{m\,s^{-1}}$.

[2]





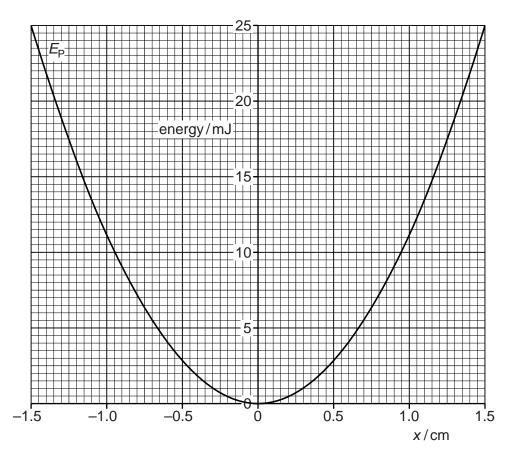


Fig. 3.3

- (i) On the axes of Fig. 3.3, sketch a graph to show the variation with displacement *x* of the kinetic energy of the ball. [2]
- (ii) The amplitude of the oscillations reduces over a long period of time. After many oscillations, the amplitude of the oscillations is 0.60 cm.

Use Fig. 3.3 to determine the total energy of the oscillations of the ball for oscillations of amplitude 0.60 cm. Explain your working.

8 An α -particle and a proton are at rest a distance 20 μm apart in a vacuum, as illustrated in Fig. 4.1. $20 \, \mu m$ proton α -particle Fig. 4.1 State Coulomb's law. (a) (i) The α -particle and the proton may be considered to be point charges. (ii) Calculate the electric force between the α -particle and the proton.

force = N [2]

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Define electric field strength.

(b) (i)

(ii) A point P is distance x from the α -particle along the line joining the α -particle to the proton (see Fig. 4.1). The variation with distance x of the electric field strength E_{α} due to the α -particle alone is shown in Fig. 4.2.



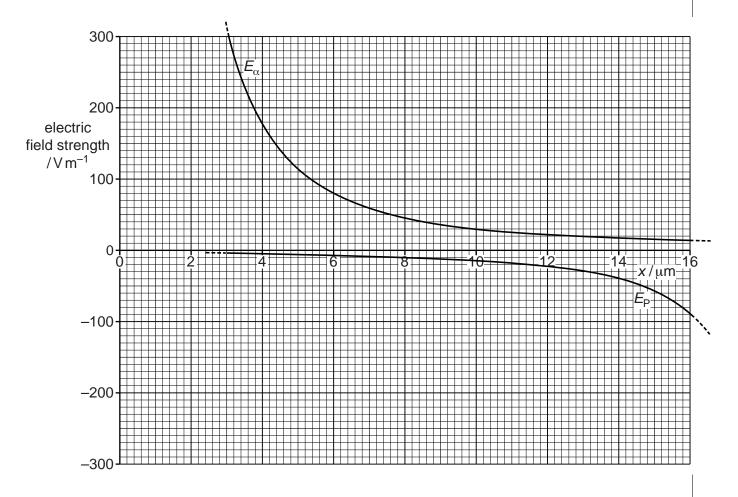


Fig. 4.2

The variation with distance x of the electric field strength $E_{\rm P}$ due to the proton alone is also shown in Fig. 4.2.

1.	Explain why the two separate electric fields have opposite signs.	
		21

2. On Fig. 4.2, sketch the variation with x of the combined electric field due to the α -particle and the proton for values of x from $4 \mu m$ to $16 \mu m$. [3]

5 (a) An incomplete diagram for the magnetic flux pattern due to a current-carrying solenoid is illustrated in Fig. 5.1.

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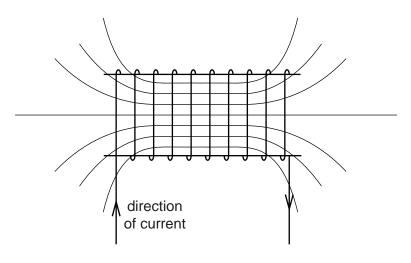


Fig. 5.1

- (i) On Fig. 5.1, draw arrows on the field lines to show the direction of the magnetic field. [1]
- (ii) State the feature of Fig. 5.1 that indicates that the magnetic field strength at each end of the solenoid is less than that at the centre.

.....[1]

(b) A Hall probe is placed near one end of the solenoid in (a), as shown in Fig. 5.2.

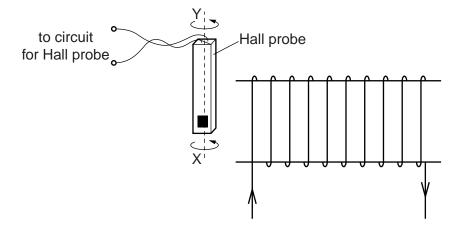


Fig. 5.2

The Hall probe is rotated about the axis XY. State and explain why the magnitude of the Hall voltage varies.

.....

.....[2

(c)	(i)	State Faraday's law of electromagnetic induction.	For
			Examiner's Use
		[2]	
	(ii)	The Hall probe in (b) is replaced by a small coil of wire connected to a sensitive voltmeter.	
		State three different ways in which an e.m.f. may be induced in the coil.	
		1	
		2	
		3	
		[3]	

spe It er	harged particle of mass m and charge $-q$ is travelling through a vacuum at constant ed v . Inters a uniform magnetic field of flux density B . The initial angle between the direction of	Examin Use
	ion of the particle and the direction of the magnetic field is 90°.	
(a)	Explain why the path of the particle in the magnetic field is the arc of a circle.	
	[3]	
(b)	The radius of the arc in (a) is r.	
	Show that the ratio $\frac{q}{m}$ for the particle is given by the expression	
	$\frac{q}{m} = \frac{v}{Br}$.	
	[1]	
(c)	The initial speed v of the particle is $2.0 \times 10^7 \mathrm{ms^{-1}}$. The magnetic flux density B is	
	2.5×10^{-3} T. The radius <i>r</i> of the arc in the magnetic field is 4.5 cm.	
	(i) Use these data to calculate the ratio $\frac{q}{m}$.	
	(i) Goo moss data to calculate me ratio m.	
	ratio = C kg ⁻¹ [2]	
	1000 Org [2]	

(ii) The path of the negatively-charged particle before it enters the magnetic field is shown in Fig. 6.1.

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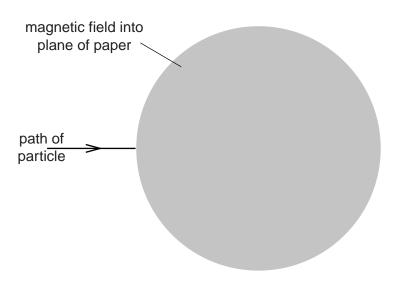


Fig. 6.1

The direction of the magnetic field is into the plane of the paper.

On Fig. 6.1, sketch the path of the particle in the magnetic field and as it emerges from the field. [2]

7 Electrons, travelling at speed v in a vacuum, are incident on a very thin carbon film, as illustrated in Fig. 7.1.

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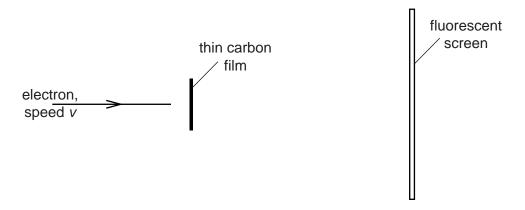


Fig. 7.1

The emergent electrons are incident on a fluorescent screen. A series of concentric rings is observed on the screen.

(a)	Suggest why the observed rings provide evidence for the wave nature of particles.
	[2]
(b)	The initial speed of the electrons is increased. State and explain the effect, if any, on the radii of the rings observed on the screen.
	[3]
	[-1

(c) A proton and an electron are each accelerated from rest through the same potential

difference.

For

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Determine the	TallO	
	de Broglie wavelength of the proton	
	de Broglie wavelength of the electron	
	ratio =	

8 (a) State what is meant by nuclear binding energy.

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(b) The variation with nucleon number A of the binding energy per nucleon $B_{\rm E}$ is shown in Fig. 8.1.

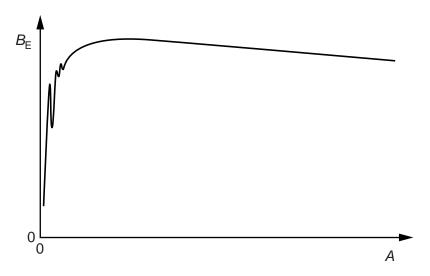


Fig. 8.1

When uranium-235 ($^{235}_{92}$ U) absorbs a slow-moving neutron, one possible nuclear reaction is

$$^{235}_{92}$$
U + $^{1}_{0}$ n \rightarrow $^{95}_{42}$ Mo + $^{139}_{57}$ La + $^{1}_{0}$ n + $^{1}_{-1}$ 0 β + energy.

(i) State the name of this type of nuclear reaction.

[1]

(ii) On Fig. 8.1, mark the position of

1. the uranium-235 nucleus (label this position U), [1]

2. the molybdenum-95 (95/42)Mo) nucleus (label this position Mo), [1]

3. the lanthanum-139 $\binom{139}{57}$ La) nucleus (label this position La). [1]

(iii) The masses of some particles and nuclei are given in Fig. 8.2.

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	mass/u
β-particle	5.5×10^{-4}
neutron	1.009
proton	1.007
uranium-235	235.123
molybdenum-95	94.945
lanthanum-139	138.955

Fig. 8.2

Calculate, for this reaction,

1. the change, in u, of the rest mass,

change in mass = u [2]

2. the energy released, in MeV, to three significant figures.

energy = MeV [3]

Section B

For Examiner's Use

Answer all the questions in the spaces provided.

9 An electronic sensor may be represented by the block diagram of Fig. 9.1.

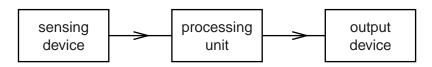


Fig. 9.1

State the function of the processing unit.	
	[2

(b) A student designs a sensing unit for temperature change. A 4V supply, a fixed resistor of resistance $2.5\,\mathrm{k}\Omega$ and a thermistor are available. The thermistor has resistance $3.0\,\mathrm{k}\Omega$ at $6\,^\circ\mathrm{C}$ and resistance $1.8\,\mathrm{k}\Omega$ at $20\,^\circ\mathrm{C}$.

Complete the circuit diagram of Fig. 9.2 to show how the resistor and the thermistor are connected to provide an output that is greater than 2V at 6 °C and less than 2V at 20 °C. Mark clearly the output $V_{\rm OUT}$



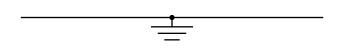


Fig. 9.2

[3]

(c) Suggest two uses of a relay as part of an output device.

1	 	
	 •	
2		

[2]

10	(a)	Explain the main principles behind the use of ultrasound to obtain diagnostic information about internal body structures.	For Examiner's Use
		[6]	
	(b)	State and explain one advantage of the use of high frequency ultrasound as compared with low frequency ultrasound for medical diagnosis.	
		[2]	
	(c)	The absorption (attenuation) coefficient for ultrasound in muscle is 23 m ⁻¹ . A parallel beam of ultrasound is passed through a muscle of thickness 6.4 cm.	
		(i) Calculate the ratio	
		intensity of transmitted beam	
		intensity of incident beam	
		ratio =[3]	

(11)	An ultrasound transmitter emits a pulse. Suggest why, when the signal from the pulse is processed, any signal received later at the detector is usually amplified more than that received at an earlier time.	For Examiner's Use
	[2]	

11 The variation with time t of the output V produced by a microphone is shown in Fig. 11.1.

For Examiner's Use

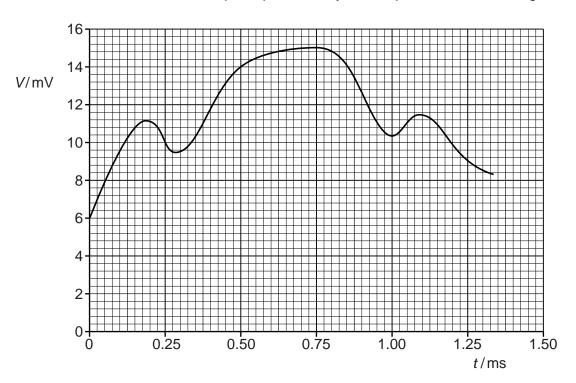


Fig. 11.1

The output is processed by a four-bit analogue-to-digital converter (ADC) that samples the output every 0.25 ms.

The first sample is taken at time t = 0 and is shown in Fig. 11.2.



Fig. 11.2

- (a) On Fig. 11.2, underline the most significant bit (MSB) of the sample shown. [1]
- **(b)** Complete Fig. 11.2 for the next five samples. [2]
- **(c)** Explain whether the sampling frequency is adequate to enable detail of the output *V* to be reproduced.

[0]

For Examiner's Use

12	(a)	_	gest why attenuation of a signal in channels of communication is usually measured a logarithmic rather than a linear scale.
			[1]
	(b)		a particular channel of communication having low attenuation, the input power is mW and the attenuation per unit length is 1.8 dB km ⁻¹ .
		(i)	Suggest the name of this channel of communication.
			[1]
		(ii)	Calculate the distance over which the power of the signal is reduced to $1.5 \times 10^{-15} \text{W}$.
			distance = km [3]

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