

# Fundamentals of Thermodynamics and Heat transfer



Year: I

Semester: II

Teaching Hours/week					Examination Scheme						Total Marks
					Internal		Final				
					Theory	Practical	Theory		Practical		
Cr	L	T	P			Duration (in hrs)	Marks	Duration	Marks		
2	2	1	2/2	20	25	1.5	30			75	

## Course Objectives:

To provide the students with a basic understanding and norms of Thermodynamics and Heat Transfer

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## Course Content:

### 1. Basic Concepts (3 hours)

1.1. Definition and Application areas of Thermodynamics

1.2. Concepts and Definitions

1.2.1. System, Surroundings, Boundary and Universe

1.2.2. Thermodynamic Properties: Intensive, Extensive and Specific Properties

1.2.3. Thermodynamic State and equilibrium and quasi-static process

1.2.4. Thermodynamic Processes and Cycles: Isobaric, isothermal and isochoric processes  
Cyclic Process, Quasi-equilibrium Process, Reversible and Irreversible Process

1.3. Common Properties: Pressure, Specific Volume, Temperature

1.4. Zeroth Law of Thermodynamics, Equality of Temperature

1.5. Value of energy to society

### 2. Energy and Energy Transfer (4 hours)

2.1. Energy and its Meaning

2.2. Stored Energy and Transient Energy; Total Energy

2.3. Energy Transfer as heat and work

2.4. Expressions for displacement work transfer (*Isobaric work, Isochoric Work, Isothermal and Polytropic work*)

2.5. Power



3. **Properties of Common Substances (4 hours)**
  - 3.1. Ideal Gas (*Boyles Law, Charles Law*) and *Combined Gas Equation*
  - 3.2. Saturation curves For Two- Phase Mixture (T-v and P-v) Diagrams
  - 3.3. Two Phase (Liquid and Vapor) Systems: Phase Change; Subcooled Liquid, Saturated Liquid, Wet Mixture, Critical Point, Quality, Moisture Content, Saturated Vapor and Superheated Vapor
  - 3.4. Other Thermodynamic Properties: Internal Energy, Enthalpy, *Specific heat Capacities*
  - 3.5. *Numerical related to Steam Table (Specific Properties related to quality)*
4. **First Law of Thermodynamics (6 hours)**
  - 4.1. Introduction and law of conservation of energy
  - 4.2. First Law of Thermodynamics for closed systems; First Law of Thermodynamics for closed system Undergoing Cyclic Process
  - 4.3. Application of the First law of Thermodynamics to Closed systems undergoing some common process: Constant Volume, Adiabatic, Constant Pressure, Constant Internal Energy.
  - 4.4. First Law of Thermodynamics for open systems (Control Volume)
    - 4.4.1. Conservation of mass, expression of mass flow rate, flow work and general energy equations
    - 4.4.2. Steady State Analysis and applications
  - 4.5. Other Statements of the First Law
  - 4.6. Perpetual Motion Machine of the kind PMM I
5. **Second Law of Thermodynamics (7 hours)**
  - 5.1. Necessity of Formulation of Second Law
  - 5.2. Kelvin-Planck and Clausius Statements of the Second Law of Thermodynamics
  - 5.3. Heat Engine and Thermal Efficiency, Heat Pump, Refrigerator and coefficient of Performance (COP) (*Theory and Numerical*)
  - 5.4. Entropy and entropy change (*introduction and definition*)
  - 5.5. Reversible and Irreversible Process
  - 5.6. Entropy and Process Relation for an Ideal Gases
  - 5.7. Isentropic Process for an Ideal Gas
  - 5.8. Air Standard Otto Cycle and Diesel Cycle (*Theory and Numerical*)
6. **Introduction to Engineering Heat Transfer: (6 hours)**
  - 6.1. Basic concepts and modes of heat transfer
  - 6.2. One Dimensional Steady State Heat Conduction through a Plane Wall/Flat Plate
  - 6.3. Radial Steady State Heat Conduction through a Hollow Cylinder
  - 6.4. Heat Flow through Composite Structures
    - a. One Dimensional Steady State Heat Conduction through a Composite Wall
    - b. Radial Steady State Heat Conduction through a Multilayer Tube
  - 6.5. Overall heat transfer for Plane composite wall
  - 6.6. Electrical Analogy for Thermal Resistance



### Laboratories:

- Temperature measurement
- Experiment related to heat pump or refrigerator
- Experiment related to heat transfer conduction, radiation and Convection

### Tutorials:

- Three assignments in each before first and second assessments.
- Quiz before first and second assessments.

### Final Examination Scheme:

Chapters	Marks*	Remarks
1	2	
2	4	
3 and 4	7	
5	9	
6	8	
Total	30	

\*There may be minor deviation in marks distribution.

### References: (in APA style)

1. C.P., G., & R., P. (1991). *Engineering Thermodynamics*. Roorkee: Nemchand & Broj.
2. Cengel, Y. A., Boles, M. A., & Kanoglu, M. (2019). *Thermodynamics: An Engineering Approach*. McGraw-Hill Education.
3. Howell, J. R., & Buckius, R. O. (1987). *Fundamental of Engineering Thermodynamics*. McGraw Hill Publishers.
4. P.K., N. (n.d.). *Engineering Thermodynamics*. New Delhi: Tata Mc Graw Hill.