

Fundamentals of Thermodynamics and Heat Transfer

Course Objectives:

After the completion of this course, students will be able to understand basic concepts, laws of thermodynamics and heat transfer and their applications as well.

1. Introduction (4 hours)

- a. Definition and Scope of Engineering Thermodynamics
- b. Value of energy to society
- c. Microscopic versus Macroscopic Viewpoint
- d. Concepts and Definitions
 - i. System, Surroundings, Boundary and Universe; Closed Systems, Open Systems, and Isolated Systems
 - ii. Thermodynamic Properties: Intensive, Extensive and Specific Property
 - iii. Thermodynamic Equilibrium
 - iv. State, Process, and Path, Cyclic Process, Quasi-equilibrium Process, Reversible and Irreversible Process
 - v. Common Properties: Pressure, Specific Volume, Temperature
- e. Zeroth Law of Thermodynamics, Equality of Temperature

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2. Energy and Energy Transfer (3 hours)

- a. Energy and its Meaning
- b. Stored Energy and Transient Energy: Total Energy
- c. Energy Transfer
 - i. Heat Transfer
 - ii. Work Transfer
- d. Expressions for displacement Work Transfer
- e. Power

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3. Properties of Common Substances (6 hours)

- a. Pure Substance and State Postulate
- b. Ideal Gas and Ideal Gas Relations
- c. Two Phase (Liquid and Vapor) Systems: Phase Change, Subcooled Liquid, Saturated Liquid, Wet Mixture, Critical Point, Quality, Moisture Content, Saturated Vapor and Superheated Vapor
- d. Properties of Two Phase Mixture
- e. Other Thermodynamic Properties: Internal Energy, Enthalpy and Specific Heats
- f. Development of Property Data: Graphical Data Presentation and Tabular Data Presentation

4. First Law of Thermodynamics (8 hours)

- a. First Law of Thermodynamics for Control Mass: First Law of Thermodynamics for Control Mass Undergoing Cyclic Process
- b. First Law of Thermodynamics for Control Volume
- c. Control Volume Analysis: Steady State Analysis and Unsteady State Analysis
- d. Control Volume Application: Steady and Unsteady Work Applications and Steady and Unsteady Flow Applications
- e. Other Statements of the First Law

5. Second Law of Thermodynamics(8 hours)

- a.Necessity of Formulation of Second Law
- b.Entropy and Second Law of Thermodynamics for an Isolated System
- c.Reversible and Irreversible Processes
- d.Entropy and Process Relation for an Ideal Gases and Incompressible Substances
- e.Control Mass and Control Volume Formulation of Second Law
- f.Isentropic Process for an Ideal Gas and for an Incompressible Substances
- g.Carnot Cycle, Carnot Efficiency, Heat Engine and Thermal Efficiency, Heat Pump, Refrigerator and coefficient of Performance(COP)
- h.Kelvin-Planck and Clausius Statements of the Second Law of Thermodynamics and their Equivalence

6. Thermodynamic Cycles(8 hours)

- a.Classification of Cycles
- b.Air Standard Analysis
 - i.Otto Cycle
 - ii.Diesel Cycle
 - iii.Brayton Cycle
 - iv.Rankine Cycle
 - v.Vapor Compression Refrigeration Cycle

7. Introduction to Heat Transfer (8 hours)

- a.Basic Concepts and Modes of Heat Transfer
- b.One dimensional steady state heat conduction through a plane wall
- c.Radial steady state heat conduction through a hollow cylinder
- d.Heat flow through composite structures
 - i.Composite Plane Wall
 - ii.Multilayer Tubes
- e.Electrical Analogy for Thermal Resistance
- f.Combined Heat Transfer and Overall Heat Transfer Coefficient for Plane Wall and Tube
- g.Nature of Convection: Free and Forced Convection
- h.Heat Radiation, Stefan's Law, Absorptivity, Reflectivity and Transmissivity; Black Body, White Body and Gray Body

References:

- 1."Engineering Thermodynamics", E. Rathakrishnan, Tata Mc Graw Hill.
- 2."Fundamentals of Engineering Thermodynamics", J. R. Howell & R. O. Buckius, McGraw Hill Publishers
- 3."Fundamentals of Thermodynamics", V. Wylen, Sonntag & Borgnakke, 6th Edition, Wiley
- 4."Fundamentals of Engineering Thermodynamics", M. J. Moran & H. N. Shapiro, 5th Edition, John Wiley & Sons, Inc.
- 5."Thermodynamics: An Engineering Approach", Y. A. Cengel & M.A. Boles, 5th Edition, McGraw-Hill, 2006
- 6."Heat Transfer", J. P. Holman, McGraw-Hill
- 7."Heat Transfer: A Practical Approach", Y. A. Cengel, 2nd Edition, McGraw-Hill

Evaluation Scheme:

The questions will cover all the chapters in the syllabus. The evaluation scheme will be as indicated in the table below:

Chapter	Hours	Marks distribution *
1	4	10
2	4	4
3	6	12
4	8	14
5	9	14
6	8	14
7	6	12
Total	45	80

***Note:** There may be minor deviation in marks distribution.