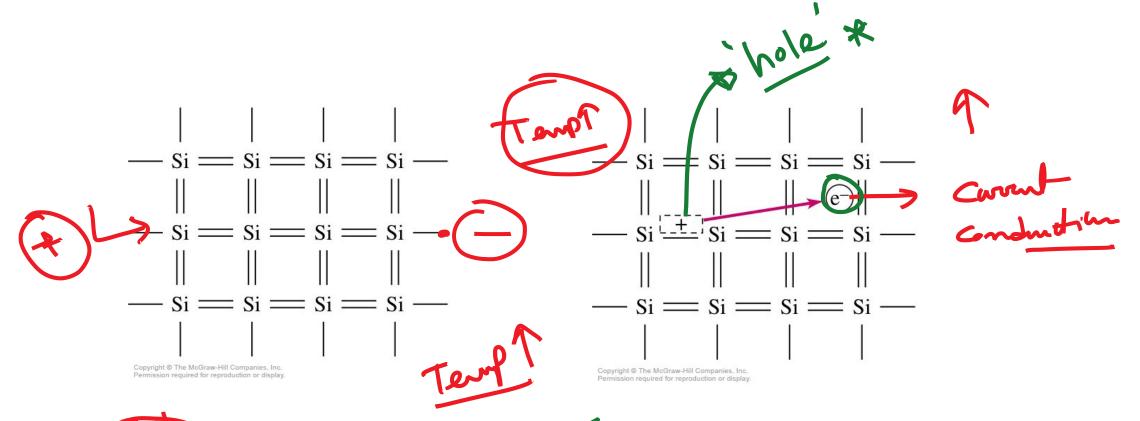
Semiconductor: Silicon Covalant bonds (c) (a)

Covalent bonding of one Si atom with four other Si atoms to form tetrahedral unit cell.

Valence electrons available at edge of crystal to bond to additional Si atoms.

Silicon: Effect of Temperature

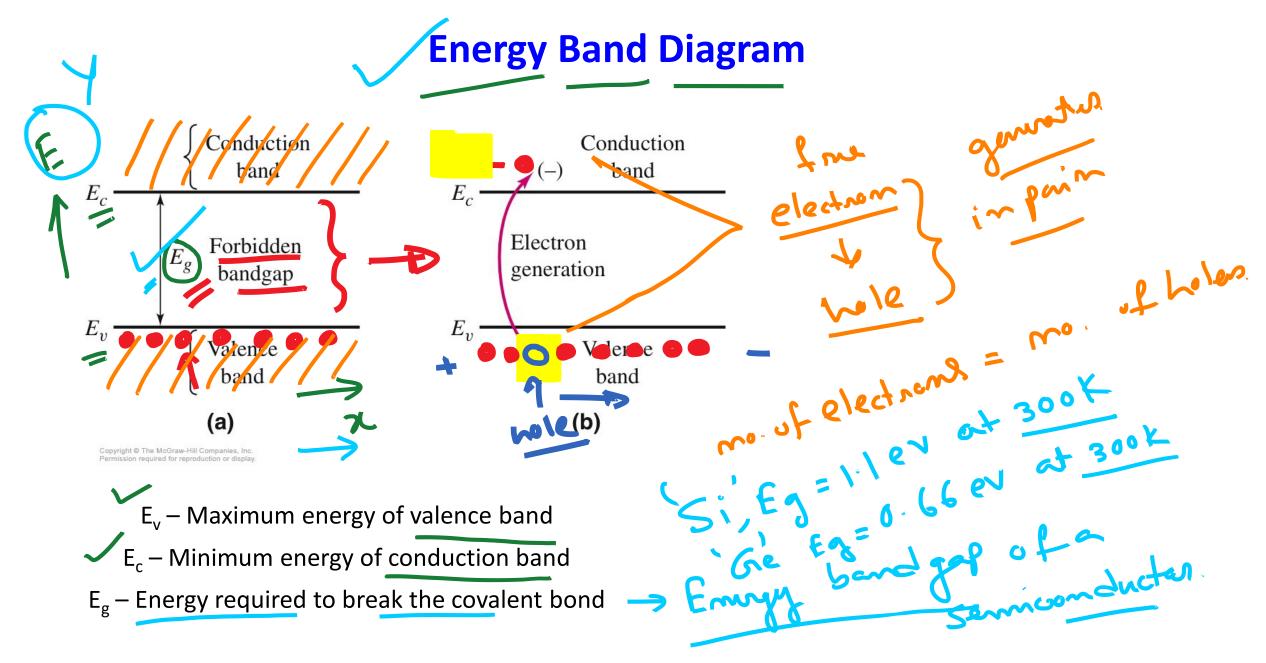


At 0 K, ho bonds are broken.

Si is an insulator.

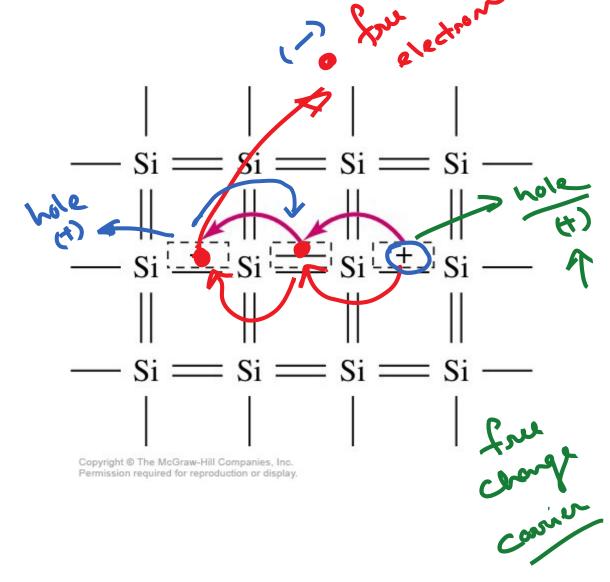
As temperature increases, a bond can break, releasing a valence electron and leaving a broken bond (hole).

Current can flow.



How does Hole moves??

A valence electron in a nearby bond can move to fill the broken bond, making it appear as if the 'hole' shifted locations.

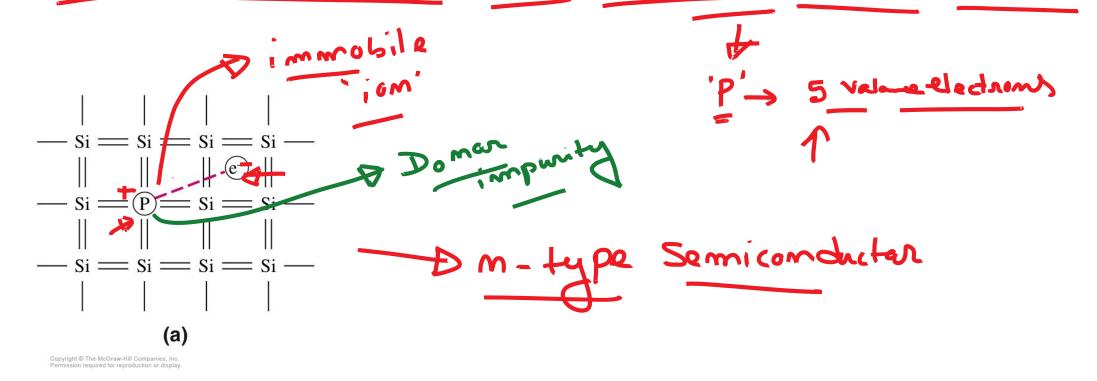


Carrier concentration in pure Silicon Intrinsic concentration (mi) is given as follows: $m_{i} = B T^{3/2} eRP(-\frac{E_3}{2kT}) Bandger(eV) 1.1eV$ Tour (k) = 300kCoefficient

Intrinsic carrier concentration in other semiconductors

Source: Electronic circuits: Analysis and Design -by D. A. Neamen, McGraw-Hill

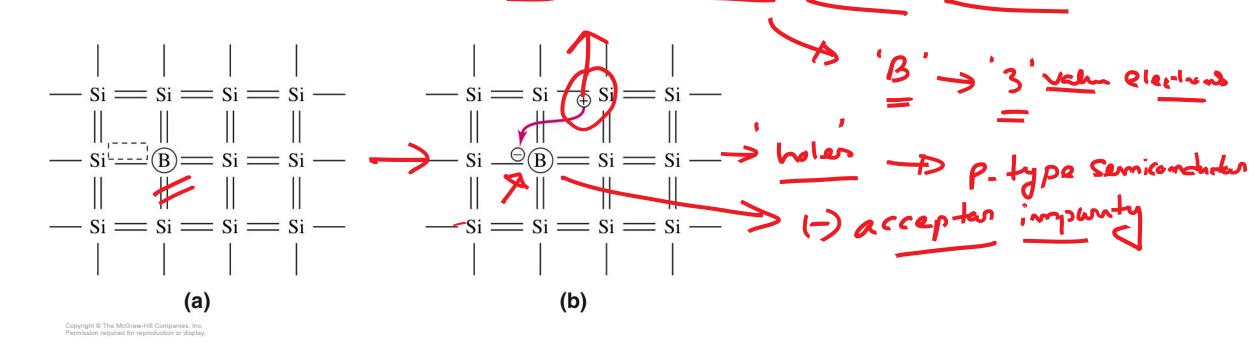
Extrinsic semiconductor (Si): when Gr-V impurity is added



Phosphorous (P) replaces a Silicon atom and forms four covalent bonds with surrounding four Si atoms.

The **<u>fifth</u>** outer shell electron of P is free to become a conduction band electron.

Extrinsic semiconductor (Si): when Gr-III impurity is added



Boron (B) replaces a Si atom and forms only **three** covalent bonds with other Si atoms.

The missing covalent bond is a hole, which can begin to move through the crystal when a valence electron from another Si atom is taken to form the fourth B-Si bond.

Carrier Concentration in Extrinsic Semiconductor

Carrier Concentration in Extrinsic Semiconductor

$$T = 300 \text{ K}, \quad \text{Si-ban}, \quad P - \text{conc} = (N_1) = 10^{16} \text{ cm}^3, \quad \text{hole conc} ??$$

$$m_1 = 1.5 \times 10^{10} \text{ cm}^3, \quad \text{ct} \quad 300 \text{ K}, \quad N_1 >> m_1, \quad m_0 P_0 = m_1^2 \\ P_0 = \frac{(1.5 \times 10^{10})^2}{10^{16}} = \frac{225 \times 10^6}{10^{16}}$$

$$P - \text{type} \qquad N_0 << m_1$$