

Chemquest 15 Answer Key

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15. Complete the following: Elements ending in d2 are in column number 4 or IVB and those ending in d4 are in column number 6 or VIB. Therefore, elements ending in d3 are in column number 5 or VB. 7 Elements ending in d must be in column number 9 or VIIIB. 16. Notice from question 6 that an element that ends in 3s is in the 3rd row. Also, from ...

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Title: ChemQuest 8: Answer Key Author: Kosiorowski Last modified by: Kosiorowski Created Date: 8/28/2008 11:49:00 PM Company: Sony Electronics, Inc.

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO₂ are produced, so the number of moles of CO₂ that can be produced when 4 moles of propane combusts = 4(3) = 12.

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Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% 1271 17% 1261, and 3%1281. Calculate the average atomic mass of iodine. 2. Calculate the average atomic mass of lithium, which occurs as two isotopes that have the following

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15. A catalyst is a substance that speeds up a reaction when it is added to the mixture of the reactants. A catalyst speeds up a reaction by doing something to the activation energy.

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