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MIXED MOLE PROBLEMS - KEY 1. a) How many grams are there in 1.5 x 1025 molecules of CO 2? 1.110g 1 mol 44.0 g 6.0210molecules 1.510molecules CO1mol 3 23 2 25!=!!!! b) What volume would the CO

#### **KEY - CP - Mixed Mole Problems**

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MIXED PROBLEMS Name What volume of NH3 at STP is produced if 25.0 g of N2 is reacted with an excess of ... Chemistry IF8766 Olnstructional Fair, Inc. STOICHIOMETRY: MOLE-MOLE PROBLEMS I. N2 +3H2 Name How many moles of hydrogen are needed to completely react with two moles

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Stoichiometry: Mixed Problems (KEY) 1) N2 + 3H2  $\rightarrow$  2NH3 What volume of NH3 at STP is produced if 25.0 of N2 is reacted with an excess of H2? 3 3 3 2 3 2 2 2 40.0L NH 1mol NH 22.4L NH 1mol N 2mol NH 28.0g N 25.0g N 1mol N × × × = 2) 2KClO3  $\rightarrow$  2KCl + 3O2 If 5.0g of KClO3 is decomposed, what volume of O2 is produced at STP? 2

# **Stoichiometry: Mixed Problems (KEY)**

ANSWER: 9.3 E 21 atoms 6) A 4.0g sample of O 2 is in a container at STP. What volume is the container? ANSWER: 2.8 L 7) How many molecules of O 2 are in the container in Problem 6? How many atoms of oxygen are in the container? ANSWER: 7.5 E 22 molecules Chemistry IF8766 Page 53 Instructional Fair, Inc.

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