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What is the difference between qualitative and quantitative observations (Your answers should reveal an understanding of the definitions for qualitative and quantitative.) Qualitative refers to the general quality of a characteristic (tall, short, long, wide, etc.) whereas quantitative includes the quantity of the characteristic (how tall, how ...

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From your answers to questions 9 and 10, formulate a definition for boiling point in terms of atmospheric pressure and vapor pressure of the liquid. The boiling point is the temperature at which the vapor pressure of a liquid equals the atmospheric pressure.

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO2 are produced, so the number of moles of CO2 that can be produced when 4 moles of propane combusts = 4(3) = 12.

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ChemQuest 15 Name: ____ Date: ____ Hour: ____ Information: Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready. Remember that a row on the periodic table

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Using only your answers to questions 7-9, calculate how many Joules of energy it takes to change 12.7 g of water at 22 o C to steam at 159 o C. Questions 7-9 are each different steps for taking

12.7g of water from 22 o C to 159 o C so you can simply add the answers: $4140.7 + 28,702 + 1513.6 = 34,356.3 \mid 11$.

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Chem Quest 32 Answer Key

Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% 1271 17% 1261, and 3%1281. Calculate the average atomic mass of iodine. 2. Calculate the average atomic mass of lithium, which occurs as two isotopes that have the following

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