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What is the difference between qualitative and quantitative observations (Your answers should reveal an understanding of the definitions for qualitative and quantitative.) Qualitative refers to the general quality of a characteristic (tall, short, long, wide, etc.) whereas quantitative includes the quantity of the characteristic (how tall, how ...

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1. Write the electron configurations for Li, Na and K. (Remember for electron configurations, arrows are not necessary.) Li 2= 1s 2s1 Na 2= 21s 2s 2p63s1 K 2= 21s 2s 2p63s23p64s1 2. What is similar about the electron configurations of all of the elements in question 1? Look at the very end of their configurations. Each of them ends with s1. 3.

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Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% 1271 17% 1261, and 3%1281. Calculate the average atomic mass of iodine. 2. Calculate the average atomic mass of lithium, which occurs as two isotopes that have the following

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For every 5 moles of oxygen, 1 mole of propane can be combusted, so the number of moles of propane that can be combusted with 20 moles of oxygen is $20/5 = 4$ moles. b) How many moles of carbon dioxide will be produced? (Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.)

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1. Which container—#1 or #2—should have the most collisions between particles? 2. Which container—#1 or #2—do you think will have the faster rate of reaction? Explain why . 3. Which of the below scenarios will have the fastest rate of reaction? Why? Recall that Molarity (M) is a unit of concentration. 4.

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Daniel S. Murad, President and CEO, The ChemQuest Group, Inc. Specialty Chemicals represent a \$730 billion industry. While fragmented, the industry comprises five major strategic markets—plus “Other”—as illustrated in Figure 1 above.. Within this landscape, there exists a common source of raw materials known as chemical intermediates.

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