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of study / pure vs applied / reasons to study Chapter 24 The Chemistry of Life 24-1 A. Chemistry Lessons, Notes, Charts, Worksheets, AP CHAPTER 8/9 Study Guide on Bonding ==\_ Here! Therefore, aquatic life can survive in winter. A model of hemoglobin in blood will be provided in chapters 24 and 25 (sections 24.2.

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Chapter 24 "The Chemistry of Life". chloroplast specialized organelle that contains a light-harvesting system to convert solar energy into chemical energy.

### **Quia - Chapter 24 "The Chemistry of Life"**

Chapter 24: The Chemistry of Life cell - tiny structures that are the fundamental units of life. 24.2 Carbohydrates Polymer - a very large molecule formed by covalent bonds of repeating small molecules, known as monomers monomer - a simple molecule that combines repeatedly to

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Chapter 2: Introduction to the Chemistry of Life Figure 2.1 Foods such as bread, fruit, and cheese are rich sources of biological macromolecules. (credit: modification of work by Bengt Nyman) The elements carbon, hydrogen, nitrogen, oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things.

### **Chapter 2: Introduction to the Chemistry of Life ...**

828 Chapter 23 • The Chemistry of Life The amide bond that joins two amino acids, shown in Figure 23.3, is referred to by biochemists as a peptide bond. A chain of two or more amino acids linked together by peptide bonds is called a peptide. A molecule that consists of two amino acids bound together by a peptide bond is called a dipeptide.

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