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answer with justification. (b) As the mass of the Sn electrode decreases, where does the mass go? The atoms on the Sn electrode are going into the solution as Sn^{2+} ions. 1 point is earned for the correct answer. (c) In the expanded view of the center portion of the salt bridge shown in the diagram below, draw and label

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DRAFT 5/7/2014 10:27:00 PM 4 Question 4 (a) $P V = n R T$ $(1.04)(1.00) = n (0.08206)(1100)$ $n = 0.0115$ moles (1) (b) Disagree. The reaction is at equilibrium so there must be some $\text{CaCO}_3(\text{s})$ left in the container. (It is entirely possible to answer this via stoichiometry, proving that if there was total decomposition, that the moles and

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