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Answer Questions, Exercises, and Problems 53–65. Check your answers with those at the end of the chapter. Workbook If your instructor recommends the Active Learning Workbook, do Questions, Exercises, and Problems 53–65. Chapter 16–Assignment E: Summary and Review Learn well the language of solutions and use it correctly.

Chapter 16

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Chapter 16 – The Process of Chemical Reactions 249 Exercise 16.5 – Predicting the Effect of Disruptions on Equilibrium: Nitric acid can be made from the exothermic reaction of nitrogen dioxide gas and water vapor in the presence of a rhodium and platinum catalyst at 700-900 C and 5-8 atm. Predict whether each of the

Chapter 16 - The Process of Chemical Reactions

Chapter 16 Solutions 167 SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471–477) This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves. Solution Formation (pages 471–472) Look at Figure 16.1 on page 471 to help you answer Questions 1 and 2.1.

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SAMPLE EXERCISE 16.3 Predicting the Position of a Proton-Transfer Equilibrium For the following proton-transfer reaction, use Figure 16.4 to predict whether the equilibrium lies predominantly to the left (that is, Kc < 1) or to the right (Kc > 1):

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