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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Label each of the following statements as True or False: True a. If the formula mass of one molecule is x amu, the molar mass is x g/mol. False b. Samples of equal numbers of moles of two different chemicals

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO 3 a. copper(II) carbonate Na 2SO 3 b. sodium sulfite (NH 4) 3PO 4 c. ammonium phosphate SnS 2 d. tin(IV) sulfide HNO 2 e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

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available: 1. How many elements of unsaturation are present for a molecule with formula C

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353 CHAPTER-BY-CHAPTER ANSWER KEY CHAPTER 1 ANSWERS FOR THE MULTIPLE CHOICE QUESTIONS 1. b The sociological perspective is an approach to understanding human behavior by placing it within its broader social context. (4) 2. d Sociologists consider occupation, income, education, gender, age, and race as dimensions of social location.(4)

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206 Chapter 7 • Ionic Compounds and Metals Section 77.1.1 Figure 7.1 As carbon dioxide dis-solves in ocean water, carbonate ions are produced. Coral polyps capture these car-bonate ions, producing crystals of calcium carbonate, which they secrete as an exo-skeleton. Over time, the coral reef forms. A coral reef is a complex habitat that

Chapter 7: Ionic Compounds and Metals

AP Chemistry Test (Chapter 7) Multiple Choice (50%) 1) Which electron would be most likely to absorb red light? A) n = 2 n = 3 B) n = 1 n = 7 C) n = 3 n = 2 D) n = 7 n = 1 2) Which statement about electromagnetic radiation is true? A) The frequency of radio waves is greater than the frequency of infrared waves.

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