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CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2. c For the expression V nR P T, which of the following will cause the volume to increase?

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Chpter 11 Review Gases Section 1 Answers

Modern Chemistry 95 Gases CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: _____ a. temperature increases, volume stays the same

CHAPTER 11 REVIEW Gases - Manasquan Public Schools

Chapter 11 Map Chapter Checklist Read the Review Skills section. If there is any skill mentioned that you have not yet mastered, review the material on that topic before reading this chapter. Read the chapter quickly before the lecture that describes it. Attend class meetings, take notes, and participate in class discussions.

Chapter 11 - Gases - Mark Bishop

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Section 3 Gas Volumes and the Ideal Gas Law What is the law of combining volumes of gases? What is Avogadro's law and what is its significance? What is the ideal gas law and how is it used? Section 4 Diffusion and Effusion How are molecular velocities and molar masses related? Review Previous Concepts Gases CHAPTER 11 1. What is a mole?

CHAPTER 11 Gases - St. Charles Parish

CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L.

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