

Chem Lab 11hydrated Crystals Answers

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Chem Lab 11hydrated Crystals Answers

If you please take the time to read this.I'd be very thankful.and answer a few questions.

Introduction:A hydrated crystal is a typically an ionic compound that has water molecules trapped within the crystal.This usually occurs because the compounds are hygroscopic;which means they absorb water from the air.Since the structure of most hydrates are unknown or verycomplex,a dot(.) is used in the ...

HYDRATED CRYSTALS CHEMISTRY LAB HELP - Yahoo Answers

Pre-Lab: Hydrated Crystals Answer the following, where appropriate answer in complete sentences. If a calculation is required, you must show your work. 1. Define: a) Hydrated compound - A hydrated compound is a compound that is surrounded by water molecules that are held in place by hydrogen bonds.

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Hydrated crystals. Hydrates are compounds that incorporate water molecules in their crystalline structures. The ratio of moles of water to one mole of that compounds is a small whole number. For example, in the hydrated compound copper (II) sulfate pentahydrate ($\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$), the ratio is 5:1.

CHEMLAB 11 - Yola

CHEMISTRY LAB: HYDRATED CRYSTALS WHAT TO TURN IN: Hypothesis Data Table Calculations (5) Questions (4) OBJECTIVES To review and observe the characteristics of a hydrate To find the experimental and theoretical percentages of water in the hydrate

CHEMISTRY LAB: HYDRATED CRYSTALS - Kwanga.net

anyone who can answer the questions using the data table will get best answer and points. im really lost in this chapter and unsure of what to do. please and thank you -----
Mass of Evaporating dish = 46.64g
Mass of Evaporating dish + hydrated MgSO_4 = 49.46
Mass of Hydrated MgSO_4 = 2.82g
Mass of Evaporating dish + anhydrous MgSO_4 = 48.47g
Mass of anhydrous MgSO_4 = 1.83g
mass of H_2O in the ...

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