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CHAPTER 13: CHEMICAL KINETICS 343 From the first set of data: $3.20 \times 10^{-1} \text{ M/s} = k(1.50 \text{ M})$ $k = 0.213 \text{ s}^{-1}$ What would be the value of k if you had used the second or third set of data? Should k be constant? 13.18 Strategy: We are given a set of concentrations and rate data and asked to determine the order of the reaction and the initial rate for specific concentrations of X and Y.

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Modern Chemistry 7 Solutions CHAPTER 12 REVIEW Solutions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

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Laboratory Manual Chemistry: Matter and Change vii How to Use This Laboratory Manual Chemistry is the science of matter, its properties, and changes. In your classroom work in chemistry, you will learn a great deal of the information that has been gathered by scientists about matter. But, chemistry is not just information.

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