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pH and pOH The pH of a solution indicates how acidic or basic that solution is. pH range of 0-7 acidic 7 neutral 7-14 basic Since ...  $[H^+]$  pH  $[OH^-]$  pOH Acidic or Basic 1.  $10^{-5} M$  5  $10^{-9} M$  9 Acidic 2.  $10^{-7} M$  7  $10^{-7} M$  7 Neutral 3.  $10^{-10} M$  10  $10^{-4} M$  4 Basic 4.  $10^{-2} M$  2  $10^{-12} M$  12 Acidic 5.

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