

## *Chemquest 36 Answer Key*

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**Chemquest 36 Answer Key**

86 ChemQuest 29 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Introduction Questions 1. Write the equation for the reaction of sodium and chlorine to form sodium chloride.  $\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$  2. Write the equation for the reaction of calcium nitride and sodium chloride to produce calcium chloride and sodium nitride.

**86 ChemQuest 29 Name: - Yumpu**

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of  $\text{CO}_2$  are produced, so the number of moles of  $\text{CO}_2$  that can be produced when 4 moles of propane combusts =  $4(3) = 12$ .

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**Chem Quest 32 Answer Key**

a) Given your answers to questions 4-6, what happens to the rate of the reaction if you increase the temperature of the reactants? b) Try to think of two reasons WHY and write them down.

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**22 ChemQuest 7 Name: - Yumpu**

Title: ChemQuest 8: Answer Key Author: Kosiorowski Last modified by: Kosiorowski Created Date: 8/28/2008 11:49:00 PM Company: Sony Electronics, Inc.

**ChemQuest 8: Answer Key - St. Francis Preparatory School**

Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% 1271 17% 1261, and 3%1281. Calculate the average atomic mass of iodine. 2. Calculate the average atomic mass of lithium, which occurs as two isotopes that have the following

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