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Atomic Basics Answer Key Part A: Atomic Structure 1. Draw five protons in the nucleus of the atom. Label them with their charge. 2. Draw six neutrons in the nucleus of the atom. 3. Draw two electrons in the first energy level and label them with their charge. 4. Draw three electrons in the second energy level and label them with their charge.

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## **Basic Atomic Structure Worksheet ANSWERS - Course Hero**

3. The atomic number tells you the number of \_\_\_\_\_ in one atom of an element. It also tells you the number of \_\_\_\_\_ in a neutral atom of that element. The atomic number gives the "identity" of an element as well as its location on the periodic table. No two different elements will

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(variable form), account for all units, labels, significant figures, and work. Use dimensional analysis format whenever possible.

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Because the masses of atoms are so small, it is more convenient to give nuclide masses in atomic mass units, abbreviated amu or u (the latter is the official SI abbreviation), rather grams. The atomic mass unit is defined as follows: One atomic mass unit is defined as 1/12 of the mass of  $(ce^{12}_{6}C)$  atom.

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