

Chemteam Molarity Worksheet Answers

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A teacher might teach problems where the molarity is calculated but ask for the volume on a test question. Note: Make sure you pay close attention to multiply and divide. For example, look at answer #8. Note that the 58.443 is in the denominator on the right side and you generate the final answer by doing 0.200 times 0.100 times 58.443.

ChemTeam: Molarity Problems #1 - 10

Worksheet - Osmosis Problems - AP level. ... molarity (M) = $1.0 \times 10^{-3} \text{ mol/L} \times 65,047 \text{ g/mol}$ (Yes, this is a correct answer.) Problem #5: Osmosis is the process responsible for carrying nutrients and water from groundwater supplies to the upper parts of trees. The osmotic pressures required for this process can be as high as 18.6 atm.

Worksheet - Osmosis Problems - AP level - t.chemteam.info

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ChemTeam: Molarity Worksheet Molarity Worksheet Return to Solutions Menu Go to Molarity Go to answers Page 1 of 2 1. Sea water contains roughly 28.0 g of NaCl per liter. What is the molarity of sodium chloride in sea water? What is the molarity of 245.0 g of H₂SO₄ dissolved in 1.00 L of solution? What is the molarity of 5.30 g of Na₂CO₃ ...

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Molarity Worksheet W 331 Everett Community College Student Support Services Program What is the molarity of the following solutions given that: 1) 1.0 moles of potassium fluoride is dissolved to make 0.10 L of solution.

Molarity Worksheet W 331 - Everett Community College

3. What is the molarity of a solution of HNO₃ that contains 12.6 grams HNO₃ in 1.0 L of solution? ?
 $\text{mol HNO}_3 = 12.6 \text{ g HNO}_3 \times \frac{1 \text{ mol HNO}_3}{63.0 \text{ g HNO}_3} = 0.200 \text{ mol HNO}_3$
 $M = \frac{0.200 \text{ mol HNO}_3}{1.0 \text{ L}} = 0.200 \text{ M}$
4. How many grams of potassium nitrate are required to prepare 0.250 L of a 0.700 M solution? 0.700 M = moles of solute / 0.250 L moles of ...

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