

## *Balancing Act Practice Answer Key*

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### **Balancing Act Practice Answer Key**

Balancing Act Answer Key: Page 1 Problems  $2\text{Ca} + \text{O}_2 \rightarrow 2\text{CaO}$   $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$   $2\text{Cu}_2\text{O} + \text{C} \rightarrow 4\text{Cu} + \text{CO}_2$   $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$  Hint: Add the O atoms on the product side together when doing the counts. Page 2 Practice Problems 1.  $2\text{Na} + \text{MgF}_2 \rightarrow 2\text{NaF} + \text{Mg}$  2.  $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$  3.  $\text{Cl}_2 + 2\text{KI} \rightarrow 2\text{KCl} + \text{I}_2$  4.

### **Name Atoms are not or during a chemical reaction. Mg + O**

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### **Balancing Chemical Equations Worksheet Answer Key ...**

Balance each equation. Be sure to show work. Remember you cannot add subscripts of place coefficients in the middle of a chemical formula.

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Balancing Equations Practice ANSWER KEY Part A: Identify the following parts of each chemical formula by circling the subscripts and drawing a square around the coefficients.  $\text{H}_2$  2  $\text{HCl}$  4  $\text{O}_2$   $\text{CH}_4$  3  $\text{CO}_3$  2  $\text{NaOH}$  Part B: List the symbols for the atoms in each formula and give the number of each.  $\text{C}_2\text{H}_6$  2 $\text{MgO}$  4 $\text{P}$  4 $\text{O}_{10}$

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