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**Chemquest 29 Answers**

86 ChemQuest 29 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Introduction Questions 1. Write the equation for the reaction of sodium and chlorine to form sodium chloride.  $\text{Na} + \text{Cl}_2 \rightarrow \text{NaCl}$  2. Write the equation for the reaction of calcium nitride and sodium chloride to produce calcium chloride and sodium nitride.

**86 ChemQuest 29 Name: - Yumpu**

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Answer the questions that follow based on this chemical equation:  $\text{Na} + \text{MgCl}_2 \rightarrow \text{NaCl} + \text{Mg}$ . Why can't NaMg be produced? (Hint: Think about the charges) Why can't NaCl<sub>2</sub> be produced? Are NaCl and Mg the only products that can be produced? ... ChemQuest 28 ...

**ChemQuest 28 - AP Chemistry**

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**Chemistry homework: balancing equations worksheet ...**

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Company: New Castle Schools Other titles: ChemQuest 27 ...

**ChemQuest 27 - New Castle Community School Corporation**

ChemQuest 8 Information: Structure of the Atom Note the following symbols: (they are not to scale) ... 29 131 54 237 93 16 2- 70 31 52 Cr 24 5b Mass number 231 108 137 119 80 . ... Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% 1271 17% 1261, and 3%1281.

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**Chemquest 26 Answer Key, Mar 15, 2015 chemquest 26 ...**

a) Given your answers to questions 4-6, what happens to the rate of the reaction if you increase the temperature of the reactants? b) Try to think of two reasons WHY and write them down.

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO<sub>2</sub> are produced, so the number of moles of CO<sub>2</sub> that can be produced when 4 moles of propane combusts = 4(3) = 12.

**ChemQuest 33 - baileykaze.webs.com**

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