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56 ChemQuest 18 Name: ____ Date: ____ Hour: ____ Information: Ions Figure 1: Below are four Bohr diagrams of atoms and ions. The two diagrams on the left are atoms; the two on the right are ions. 12 protons 12 protons Atom A Ion of Atom A 17 protons 17 protons Atom B Ion of Atom B Critical Thinking Questions 1.

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Answer the questions that follow based on this chemical equation: Na + MgCl 2 NaCl + Mg. a) Why can't NaMg be produced Na + cannot bond with Mg 2+ because they are both positive. b) Why can't NaCl 2 be produced <math>Na + only requires one Cl - I.

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48 18. There are three major divisions on the periodic table: the "s block", the "d block" and the "p block". Where are these blocks of elements located?

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18.9985 amu Ion 12 protons 12 neutrons 24 12 Mg 23.9968 amu 10 eutrons mass = 19 18.9980 amu Atom 12 protons 12 neutrons 12 Mg mass = 23.9978 amu Critical Thinking Questions 10. What is structurally different between an atom and an ion? Note: This is the ONLY structural difference between an atom and an ion. 11.

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Use the Guided Inquiry Chemistry

(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO2 are produced, so the number of moles of CO2 that can be produced when 4 moles of propane combusts = 4(3) = 12.

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