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Isotope name atomic # mass # # of protons # of neutrons # of electrons 92 uranium-235 92 235 92 143 92 92 uranium-238 92 238 92 146 92 5 boron-10 5 10 5 5 5 5 boron-11 5 11 5 6 5

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The Atoms Family - Atomic Math Challenge • Atomic Number = the number of protons in the nucleus. • Atomic Mass = the average weight of the neutrons and protons in any isotope of the element. Ac Actinium Ag Silver Al Aluminum Am Americium Ar Argon As Arsenic At Astatine

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#### **AMC Answers**

answer key to the atoms family atomic math challenge 3AD262A648E237702AC92EE30FBA7737 things: electrons, protons and neutrons. Protons and neutrons are very similar ...

# **Answer Key To The Atoms Family Atomic Math Challenge**

As students complete the assignment I let them look at the Atomic math key at my desk. As students are practicing the problems I walk around answering questions students might have. Typically they struggle with how to calculate the number of neutrons in an atom using the atomic number and mass number.

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This The Atoms Family ~ Atomic Math Challenge Worksheet is suitable for 5th - 8th Grade. Bold boxes present parts of the periodic table, with some of the contents conveniently missing. Beginning chemists fill in the blanks, which may include the atomic number, atomic mass, atomic symbol, or element name.

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