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Molecular refers to the mass ofs single molecule. but molar refers to the mass ol'a mole of something (or the mass Ol-filnl'fl 03' atoms or molecules), Information: Beginning Mole Conversions The mole (ofl'en abbreviated as mol) is the link bemecn the microscopic (atoms and molecules) and the macroscopic (Things measured in grams).

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This video is designed to help you complete the intro to moles chemquest.

Intro to Moles

16. Molecular refers to the mass of a single molecule, but molar refers to the mass of a mole of something (or the mass of 6.02×1023 atoms or molecules) . 17. 18. x atoms mol x atoms g mol g 23 23 3.82 10 6.02 10 197.0 125 x x x molecules mol x molecules g mol g 24 23 1.17 10 6.02 10 108 210 x x

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September 18, 2015 (Friday) - Sub Day. 1. Chemquest 30: Intro to Moles - page 1 by Monday

(screencast) There is not hand-out so use the copy on the site (either print it out or hand-write it). It goes on pg. 55 (Page 2-4 = due Tuesday at the end of class) 2. Complete study guide. - You will take the entire test on Monday.

Unit 4: Balancing Moles - Ms. Henderson's Chemistry ...

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