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To say that a solution has a concentration of 5 M means that in 1 L of solution (not solvent) there would be 5 mol of solute: to prepare such a solution one would place 5 mol of NaCl in a 1-L flask, and then add whatever amount of water is necessary so that the

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Solutions Manual Chemistry: Matter and Change • Chapter 13 255 CHAPTER 13 SOLUTIONS MANUAL Assume that the amount of gas is constant in the following problems. 11. A sample of air in a syringe exerts a pressure of 1.02 atm at 22.0°C. The syringe is placed in a boiling water-bath at 100.0°C. The pressure is

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1 Chapter 15 – Acid-Base Equilibria. Acid-Base Equilibria . 15.1 Solutions of Acids or Bases Containing a Common Ion . A. Common Ion 1. Ion provided in solution by an aqueous acid (or base) as well as a salt

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