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The explanation and guided practice will all be done on the same Atomic math worksheet. I begin by providing student with a few examples on the worksheet: One that shows them how to calculate mass, one that shows them how to determine neutrons and one that shows them how to find the element name using protons.

### **Ninth grade Lesson Atomic Math | BetterLesson**

Chemistry--Unit 1: Atomic Structure and the Periodic Table Practice Problems I. Atoms 1)

100,000,000 copper atoms would form a line 1 cm long. How long would a line formed by  $1 \times 10^7$

copper atoms be? Express your answer in millimeters. II. Structure of the Atom 2) A sulfur-32 atom contains 16 protons, 16 neutrons, and 16 electrons.

### Chemistry--Chapter 5: Atomic Structure and the Periodic Table

CHEMISTRY 12. Determine the total number of electrons that are used in bonding.  $20\text{ e} - 14\text{ e} = 6\text{ e}^-$ , or 3 covalent bonds. Step 3 Determine the number of non-bonding electrons.  $14\text{ valence e}^- - 6\text{ bonding e}^- = 8\text{ non-bonding e}^-$ , or 4 lone pairs Hydrogen has no lone pair, and each oxygen atom has 2 lone pairs.

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