

Chemistry 101 Section Review Answers

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Chemistry 101 ANSWER KEY 1 REVIEW QUESTIONS Chapter 9 1. Draw Lewis structures for each of the following structures and assign formal charges to each atom: a) SF₂ 20 electrons 0 0 0 b) NH₂ OH (N and O are bonded to one another) 14 electrons 0 0 0 0 0 c) PO₄³⁻

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Chemistry 101 ANSWER KEY REVIEW QUESTIONS Chapter 4 1. Identify each of the following substances as a non-electrolyte (NE), weak electrolyte (WE), or strong electrolyte (SE): a)

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Chemistry 101 1 REVIEW QUESTIONS Chapter 11 1. Use the phase diagram above to answer the following questions: A) What is the normal boiling point of the substance above? ____ B) What are the critical pressure and the critical temperatures for this substance? T c ...

Chemistry 101 REVIEW QUESTIONS - mymission.lamission.edu

Chemistry 101 ANSWER KEY 1 REVIEW QUESTIONS Chapter 10 1. Draw Lewis structures and determine the molecular geometry of each molecule or ion shown below: A) ClO 2 - 20 electrons Bent (from tetrahedral) B) ICl 3 28 electrons T-shaped (from trigonal bipyramidal) C) TeF 4 34 electrons See-saw (from trigonal bipyramidal) 2.

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The goal of chemistry is to accumulate knowledge. ____13. Biochemistry is the study of the chemistry of living organisms. 10 9 8 7 6 5 4 3 2 1 • chemistry • organic chemistry • inorganic chemistry • analytical chemistry • physical chemistry • biochemistry CHEMISTRY SECTION REVIEW 1.1

1.1 Chemistry Section Review - LPS

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