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#### **Average Atomic Mass Pogil Answer Key**

Use one of the methods in Model 3 that gave the correct answer for average atomic mass to calculate the average atomic mass for oxygen. Isotope information is provided below. Show all of your work and check your answer against the mass listed on the periodic table. Isoto e Natural Abundance on Earth (0/0) Atomic Mass  $(am\ u)$  - 16.00

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POGIL atomic mass answer key.pdf - Average Atomic Mass How... N or since Likere are 3 du¥¥erm+ mas; numbers "We/"Cr m 7') A'fiF'Fe m'l' mat—\$5 \, 4. For the sample of 20 atoms of magnesium shown in Model 1, draw a table indicating the mass numbers of the three isotopes and the number of atoms of each isotope present. nae-FM 5.

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correct answer for average atomic mass (the one on the periodic table). Alan's method did not include % abundance of each isotope c. For the methods that gave the correct answer for average atomic mass, sho w that they are ... POGIL 3 Average Atomic Mass Key Author:

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CP Chemistry 2015-16 Kaspriskie Average Atomic Mass POGIL Activity. Isotopes Review. Fill in the following table. Isotope Symbol Atomic Number Mass Number Protons Neutrons 1940 9 19 9 10 918 19 40 19 21 S-31 16 31 16 15 Do all isotopes of an element have the same atomic number? Explain your answer. YES.

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There are three isotopes of silicon. Si-28 has an atomic mass of 27.98 amu and an abundance of 92.21%, Si-29 has an atomic mass of 28.98 amu and an abundance of 4.70%, and Si-30 has an atomic mass of 29.97 amu and an abundance of 3.09%. Calculate the average atomic mass of silicon. Award credit for the following answer: • 28.09 amu

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15. Use one of the methods in Model 3 that gave the correct answer for average atomic mass to calculate the average atomic mass for oxygen. Isotope information is provided below. Show all of your work and check your answer against the mass listed on the periodic table. Isotope Natural Abundance on Earth (%) Atomic Mass (amu) 16O 99.76 15.9949

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