

Chemfax Leftover Aluminum Wire Answer

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Chemfax Leftover Aluminum Wire Answer

Chemfax Leftover Aluminum Wire Answer.pdf LEFTOVER ALUMINUM WIRE – A Stoichiometry Lab Sun, 21 Apr 2019 19:54:00 GMT Rinse the aluminum wire with tap water and dry the wire with a paper towel. Once dry, measure and record the actual “leftover” mass of the aluminum wire in the data table. Dispose the used aluminum wire by rolling into a ...

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So we did a lab in class and I need help finding the final answers. The data table says: the initial mass of $\text{CuCl}_2(\text{g})$ was .52g Initial mass of aluminum wire (g) was .74 The mass of "leftover" aluminum wire (g) was .68 The actual mass of aluminum reacted (g) was .6 The actual mass of recovered dry copper (g) .22 I need help finding : A) The moles of CuCl_2 B) Moles of aluminum that should react ...

Chemistry lab help please please help D;? | Yahoo Answers

Rinse the aluminum wire with tap water and dry the wire with a paper towel. Once dry, measure and record the actual “leftover” mass of the aluminum wire in the data table. Dispose the used aluminum wire by rolling into a ball (to eliminate sharp ends) and place in the trash. Rinse the graduated cylinder and test tube with tap water.

LEFTOVER ALUMINUM WIRE - A Stoichiometry Lab

With the Leftover Aluminum Wire Stoichiometry Laboratory Kit, students perform a reaction between aluminum wire and a copper salt. Students find the amount of aluminum that should react and compare to the amount of "leftover" aluminum.

Leftover Aluminum Wire—A Stoichiometry Super Value Kit

Leftover Aluminum Wire. CAUTION: CuCl_2 solution is toxic by ingestion and is a body tissue irritant. Avoid contact with body tissues. Wear safety goggles, apron, and gloves for this activity.

Leftover Aluminum Wire | Stoichiometry (1.3K views)

Leftover Aluminum Wire Pre-Lab. STUDY. PLAY. Terms in this set (...) What is the purpose of the lab? The purpose of the lab is to compare the actual mass of aluminum reacted and the actual mass of copper produced with that predicted from the balanced chemical equation using stoichiometric calculations.

Leftover Aluminum Wire Pre-Lab Flashcards | Quizlet

Lab 14 Aluminum Leftovers – A Quantitative Analysis. Honors Chemistry. ... Using your answer from b, calculate the number of moles of water can theoretically be produced from the number of moles of O_2 you have. ... Initial mass of aluminum wire (g) Mass of left over aluminum wire (g) Actual mass of aluminum reacted (g) Mass of watch glass and ...

Lab format - Hudson City Schools / Homepage

Moles of Aluminum and Copper Moles of Al Moles of Cu The materials for Aluminum and Copper are available as a traditional Student Laboratory Kit from Flinn Scientific, Inc. Catalog No. Description AP4678 Leftover Aluminum Wire—Student Laboratory Kit Consult your Flinn Scientific Catalog/Reference Manual for current prices.

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