

## *Chapter 8 Review Chemical Equations And Reactions Answers*

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CHAPTER 8 REVIEW Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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Modern Chemistry 2 Chemical Equations and Reactions 6. Gold has a low reactivity and therefore does not corrode over time. 7. In single-displacement reactions, if the activity of the free element is greater than that of the element in the compound, the reaction will take place. 8. Yes; because aluminum is above copper in

### CHAPTER 8 REVIEW Chemical Equations and Reactions

8-2 Balancing Chemical Equations Atoms rearrange in chemical reactions, but atoms cannot be created or destroyed (Law of Conservation of Matter). For this reason, the number of each kind of atom on the product and the reactant side must be the same.

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Chapter 8 Reactions and Equations Chapter 8-Assignment A: Balancing Chemical Equations In Chapter 2 you learned about the characteristics of a chemical change. Chemists describe chemical changes, or reactions, by writing chemical equations. In this assignment you'll learn to describe the chemical reactions of elements and compounds by using a ...

### Chapter 8

Modern Chemistry 5 Chemical Equations & Reactions CHAPTER 8 STUDY GUIDE Chemical Equations and Reactions SECTION 8-2 SHORT ANSWER Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. \_\_\_\_ synthesis

### CHAPTER 8 Chemical Equations and Reactions

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Chemical reactions are described by chemical equations. A chemical equation represents, with symbols and formulas, the identities and relative molecular or molar amounts of the reactants and

products in a chemical reaction. For example, the following chemical equation shows that the reactant ammonium dichromate yields the products nitrogen,

**CHAPTER 8 Chemical Equations and Reactions**

CHAPTER 8 REVIEW. Chemical Equations and Reactions. SECTION 2. SHORT ANSWER Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right.

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CHAPTER 8 REVIEW Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_ A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

**CHAPTER 8 REVIEW Chemical Equations and Reactions**

CHAPTER 8 REVIEW Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products. (c) identities of reactants and products in a chemical reaction.

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Chapter 8 Review and Study Guide Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. Knowledge about what products are produced in a chemical reaction is obtained by a. inspecting the chemical equation. c. laboratory analysis. b. balancing the chemical equation. d. writing a word equation. \_\_\_\_ 2.

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Chapter 8 - Chemical Equations and Reactions. Describing Chemical Reactions \_\_\_\_ - process in which one or more substances are changed into one or more different substances. Reactants - original substance ... Net Ionic Equation Practice - write complete ionic equation and the net ionic equation. Include coefficients, physical states and ...

**Chapter 8 - Chemical Equations and Reactions**

CHAPTER 8 REVIEW Class S E-ci7b IV Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. b d A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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Characteristics of Chemical Equations • The following requirements will aid you in writing and reading chemical equations correctly. Section 1 Describing Chemical Chapter 8 Reactions 1. The equation must represent known facts. 2. The equation must contain the correct formulas for the reactants and products. 3.

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