# Chemquest 38 Answer Key

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#### **Chemquest 38 Answer Key**

Answer the questions that follow based on this chemical equation: Na + MgCl 2 NaCl + Mg. a) Why can't NaMg be produced Na + cannot bond with Mg 2 + because they are both positive. b) Why can't NaCl 2 be produced Na + only requires one Cl - a.

#### 82 ChemQuest 28 Name: - Yumpu

Information: What is a Mole Hopefully, you found that your answers to questions 2 and 4 were about the same. Both answers should be about  $6.02x10\ 23$ . The quantity,  $6.02x10\ 23$  is Avogadro's constant and we call it the "mole".

# 90 ChemQuest 30 Name: - Yumpu

116 Information: Dry Gas The gas that is collected is not "dry" because there is some water vapor left behind in the beaker. If you attempt to collect pure oxygen, you will actually get mostly pure oxygen with a little water

# **Information: Collecting Gas Over Water**

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#### Chemquest 26 Answer Key, Mar 15, 2015 chemquest 26 ...

(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO2 are produced, so the number of moles of CO2 that can be produced when 4 moles of propane combusts = 4(3) = 12.

#### ChemQuest 33 - baileykaze.webs.com

Answer the questions that follow based on this chemical equation: Na + MgCl2 ( NaCl + Mg. Why can't NaMg be produced? (Hint: Think about the charges) Why can't NaCl2 be produced? Are NaCl and Mg the only products that can be produced? ... ChemQuest 28 ...

#### ChemQuest 28 - AP Chemistry

#### O o o to to z CJQ o O o o O o o o o o o o cn O o o o ...

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38 electrons instead of 36. Strontium's electron configuration is 1s 2 2s 2 2p 6 3s 2 3p 6 4s 2 3d 10 4p 6 5s 2 . Strontium has the same electron configuration as krypton and then two additional electrons in the 5s

#### ChemQuest 15 - Yav Science

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