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ATOMIC STRUCTURE SKILL PRACTICE 8 ANSWERS

Skill Practice 13 Name: _____ Date: _____ Hour: _____ 1. An atom has a mass number of 43 and it also has 21 electrons. a) How many protons does this atom have? b) What is the identity of this atom? c) How many neutrons does this atom have? 2. What is an isotope? Give an example. 3.

Skill Practice 13 - chemistryinquiry.com

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(6 protons, 6 neutrons) (6 protons, 7 neutrons) (6 protons, 8 neutrons) Notice the type of notation used for atoms: X = chemical symbol of the element. Z = "atomic number" A = "mass number", , and are notations that represent . isotopes. of carbon., and are notations that represent . isotopes. of hydrogen.

ChemQuest 8: Answer Key - St. Francis Preparatory School

3. A certain ion has an atomic number of 16, a mass number of 33, and 18 electrons. a) What is the

charge on the ion? -2 b) What is the identity of this ion? sulfur c) How many neutrons does the nucleus of this ion have? 17 4. Tritium (an isotope of hydrogen) has 2 neutrons. How many protons does it have? What is its mass number? 1; 3 5.

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Atomic Structure Questions for Tests and Worksheets

ChemQuest 8 Information: Structure of the Atom Note the following symbols: (they are not to scale) O^+ = proton (positive charge) e^- = electron (negative charge) n^0 = neutron (no charge) The following three diagrams are hydrogen atoms: The following three diagrams are carbon atoms: Name: 24 Date: Hour: 14 12 (6 protons, 6 neutrons) Notice the type of ...

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Chapter 8 7 Glencoe Algebra 1 Skills Practice Adding and Subtracting Polynomials Find each sum or difference. 1. $(2x + 3y) + (4x + 9y)$ $6x + 12y$ 2. $(6s + 5t) + (4t + 8s)$ $14s + 9t$ 3. ... 15. $5x^2 - 3x - 4$ 16. $2c + 8c + 9 - 3$ no yes; 2; trinomial Write each polynomial in standard form. Identify the leading coefficient.

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Unit 2 Atomic Theory Practice Packet Key

Chemistry--Unit 1: Atomic Structure and the Periodic Table Practice Problems I. Atoms 1) 100,000,000 copper atoms would form a line 1 cm long. How long would a line formed by 1×10^7 copper atoms be? Express your answer in millimeters. mm Cu atoms mm Cu atoms? 1 10 10 1 108 u 7 mm u? 1 II. Structure of the Atom

Chemistry--Chapter 5: Atomic Structure and the Periodic Table

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3. A certain ion has an atomic number of 16, a mass number of 33, and 18 electrons. a) What is the charge on the ion? b) What is the identity of this ion? c) How many neutrons does the nucleus of this ion have? 4. Tritium (an isotope of hydrogen) has 2 neutrons. How many protons does it have? What is its mass number? 5.

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