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CHEMISTRY DIAGNOSTIC TEST TOPIC 7 ANSWERS

7. For a chemical reaction it is usually found that the reaction rate is faster at higher temperature. The rate increases because (a) the concentrations of reactants increase (b) more reactants collide with energy equal to or greater than the activation energy (c) the concentrations of products increase

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7. For a chemical reaction it is usually found that the reaction rate is faster at higher temperature. The rate increases because a. the concentration of reactants increase b. more reactants collide with energy equal to or greater than the activation energy c. the concentrations of products increase d.

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The molar mass of a substance is the mass of one mole of the substance. These test questions deal with calculating and using molar masses. An example of a molar mass might be: GMM O₂ = 32.0 g or KMM O₂ = 0.032 kg.

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