

## *Chapter 12 Stoichiometry Study Guide Answers*

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12 O 6 +6 O 2 • On a summer day, one acre of corn ... study the entire chapter online ... Stoichiometry The study of quantitative relationships between the amounts of reactants used and amounts of products formed by a chemical reaction is called stoichiometry. Stoichiometry is based on the law

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d. elastic; kinetic 3.  $KE = \frac{1}{2}mv^2$  4. Temperature 5. true 6. true 7. false 8. true 9. true

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Chapter 11 study guide True/False Indicate whether the statement is true or false. \_\_\_\_ 1. The  
actual yield is always lower than the theoretical yield. Matching Match each item with the correct  
statement below. a. stoichiometry b. mole ratio c. stoichiometric equivalent \_\_\_\_ 2.

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the goals of a stoichiometry calculation are. Interpret balanced equation coefficients as mole,  
particle, or liter (if all gases) ratios. Perform stoichiometry calculations - refer to the "roadmap"  
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