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Covalent Bonding Covalent Bonding Solutions Manual Chemistry: Matter and Change • Chapter 8 121 Section 8.1 The Covalent Bond pages 240–247 Practice Problems page 244 Draw the Lewis structure for each molecule. 1. PH_3 $\text{H}-\text{H}-\text{H}$ $\text{H}-\text{H}-\text{P}$ respectively, for single, double, and triple P — 2. H_2S $\text{H}-\text{H}-\text{H}$ $\text{H}-\text{S}-\text{S}-$ 3. HCl $\text{H}-\text{Cl}-\text{H}$ $\text{H}-\text{Cl}-$ 4 ...

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The MoleThe Mole Solutions Manual Chemistry: Matter and Change • Chapter 10 161 Section 10.1 Measuring Matter page 320–324 Practice Problems pages 323–324 1. Zinc (Zn) is used to form a corrosion-inhibiting surface on galvanized steel. Determine the number of Zn atoms in 2.50 mol of Zn. 2.50 mol Zn 6.02 23 10 atoms Zn ___ 1 mol Zn

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