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# Chemistry (12th Edition) Chapter 13 - States of Matter ...

The electron density around each nucleus is, for a moment, greater in one region of each cloud. Each molecule forms a temporary dipole. 394 Chapter 13 States of Matter. When temporary dipoles are close together, a weak dispersion force exists between oppositely charged regions of the dipoles, as shown in Figure 13-8.

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, The motion of the particles in this is rapid, constant, and random., The SI unit of pressure is this, abbreviated with the letters (Pa) or (kPa)., The average kinetic energy of a particle in a substance is directly related to the substance's this., At zero degrees Celsius and a pressure of 101.3 kPa (or 1 atmosphere) you have this term abbreviated with the letters (STP).

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still somewhat mysterious to humans, the plasma state of matter is the most common state in the universe. In this chapter, your exploration of the states of matter will go far beyond everyday, casual observations. You will explore a char-acteristic shared by liquids and gases, and investigate how these substances produce pressure.

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