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Chapter 9 Chemical Calculations and Chemical Formulas

e. molar mass of $\text{Fe}(\text{NO}_3)_3 = 241.9 \text{ g}$ 272 g $\text{Fe}(\text{NO}_3)_3$ 1 mol 241.9 g = 1.12 mol $\text{Fe}(\text{NO}_3)_3$ f. molar mass of $\text{FeSO}_4 = 151.9 \text{ g}$ 12.7 mg FeSO_4 1 g 1000 mg 1 mol 151.9 g = 8.36 10^{-5} mol FeSO_4 4 g. molar mass of $\text{LiOH} = 23.95 \text{ g}$ 6.91 103 g LiOH 1 mol 23.95 g = 288.5 = 289 mol LiOH 14. a. molar mass of $\text{CaCO}_3 = 100.1 \text{ g}$ 2.21 10^{-4} mol CaCO_3 3 100.1 g 1 mol = 0 ...

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Chemical ReactionsChemical Reactions

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"Inner energy levels tend to block the attraction of the nucleus for the valence electrons; reason why electronegativity goes down in value in a group/family."- is called the ____.

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