

## *Chemical Names Formulas Answers 221*

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### CHAPTER 9 CHEMICAL NAMES FORMULAS PAGE 221 ANSWER KEY PDF

Chapter 9 Chemical Names and Formulas 221 . 4. Write formulas for the ionic compounds.  
e. onium chloride f. potassium dichromate a. sodium phosphate b. magnesium sulfate c. sodium hydroxide d. potassium cyanide 5. Write formulas for compounds formed from these pairs of ions.  
a.  $\text{NH}_4^+$ .S042-

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### Chapter 9: Chemical Names and Formulas Flashcards | Quizlet

ANSWERS TO FINAL REVIEW: CHEMICAL NAMES AND FORMULAS 1. Name these compounds: LiCl lithium chloride K<sub>2</sub>S potassium sulfide AlI<sub>3</sub> aluminum iodide ZnS zinc sulfide BaF<sub>2</sub> barium fluoride Mg<sub>3</sub>N<sub>2</sub> magnesium nitride MgO magnesium oxide Na

### ANSWERS TO FINAL REVIEW: CHEMICAL NAMES AND FORMULAS - SNC 1PI

Here are printable chemistry worksheets in pdf format. The first worksheet in each set asks you to name the chemical formula based on the chemical name. The second worksheet asks for the chemical name based on the chemical formula. Answer keys are provided separately.

### Chemical Names and Formulas - Printable Worksheets

Chapter 9 Chemical Names and Formulas 85 Names and Formulas for Bases (page 273) 6. A base is a compound that produces \_\_\_\_\_ when dissolved in water. 7. How are bases named? SECTION 9.5 THE LAWS GOVERNING FORMULAS AND NAMES (pages 274-279) This section uses data to demonstrate that a compound obeys the law of definite proportions.

### Name Date Class CHEMICAL NAMES AND FORMULAS 9

Chapter 9 - Chemical Names and Formulas. Jennie L. Borders. Periodic Table for Naming. Section 9.1 Regular Metals Review. The regular metals include Group 1 (except H) and 2 and aluminum. When naming a compound that starts with a regular metal, name the first element and add -ide to the second element (except for polyatomic ions).

### Chapter 9 - Chemical Names and Formulas

Essay (10pts: 1 for spelling/grammar, 1 for having at least 3 sentences, 8 for correct answer) 28. Compare the compositions of binary ionic and binary molecular compounds. 29. Given the name of a binary molecular compound, describe how to write its formula. Use carbon tetrachloride as an example. 30. Name the compounds CuBr<sub>2</sub>, SiCl<sub>2</sub>, and BaF<sub>2</sub> ...

### Chemical Names and Formulas Test - Mr. Walk

Chemistry--Unit 2: Chemical Names and Formulas Test Review answers to questions to try 1. compound 2. molecular 3. nonmetals 4. molecule 5. ions 6. low 7. high 8. gain/lose 9. lose/gain 10.

compounds 11. chemical formula 12. molecular formula 13. subscript 14. formula unit 15. definite  
16. small whole number 17. monatomic 18. lose 19. +1 20. +2 ...

### **Chemistry--Chapter 6: Chemical Names and Formulas**

Prentice Hall Chemistry Chapter 9: Chemical Names and Formulas Chapter Exam Instructions.  
Choose your answers to the questions and click 'Next' to see the next set of questions.

### **Prentice Hall Chemistry Chapter 9: Chemical Names and ...**

Chapter 9 Practice Test - Naming and Writing Chemical Formulas Answer Section MATCHING 1. ANS:  
A PTS: 1 DIF: L1 REF: p. 253 OBJ: 9.1.1 Identify the charges of monatomic ions by using the periodic  
table, and name the ions.

### **Naming and Writing Chemical Formulas - wscacademy.org**

The name of a base is the name of the cation followed by the name of the anion (hydroxide). The  
formula of a base is written by showing the number of hydroxide ions needed to balance the  
positive charge on the cation. After reading Lesson 9.4, answer the following questions. Names and  
Formulas of Acids hydrogen1.

### **Chemical Names and Formulas - Weebly**

A B; Ammonia:  $\text{NH}_3$ : Carbon dioxide:  $\text{CO}_2$ : Carbon monoxide:  $\text{CO}$ : Chlorine:  $\text{Cl}_2$ : Hydrogen chloride:  
 $\text{HCl}$ : Hydrogen:  $\text{H}_2$ : Hydrogen sulfide:  $\text{H}_2\text{S}$ : Methane:  $\text{CH}_4$ : Nitrogen:  $\text{N}_2$  ...

### **Quia - Chemical Names and Formulas**

Molecular Formulas Write or recognize the definition of isotope. (Section 2.4) Describe the general  
structure of molecular and ionic compounds. (Sections 3.3 and 3.5) Convert between the names of  
compounds and their chemical formulas. (Section 5.3) Report the answers to calculations to the  
correct number of significant figures. (Section 8.2)

### **Chapter 9 ChemiCal C and C Formulas - Mark Bishop**

Chapter 9 Chemical Names and Formulas 221 Practice Problems In your notebook, solve the  
following problems. SECTION 9.1 NAMING IONS 1. What is the charge on the ion typically formed by  
each element?

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