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Modern Chemistry 109 Chapter Test Chapter: Solutions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ... 15. a 16. d 17. c 18. b 19. a 20. b 21. d 22. 23. c 24. b 25. a TEST B 1. b 2. c 3. a 4. 5. b 6. d 7. d 8. c 9. decreases 10. solute 11. colloid 12 ...

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Chapter 16

Chapter 16 Sample Test Instructions: You may use a "clean" periodic table. Sample test questions for the optional sections on molality, normality, and colligative properties follow this test. 1) A solution that can dissolve more of a substance than it currently contains is said to be ____.

Chapter 16

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b. a solution containing the maximum amount of. solute that can be dissolved at a given temperature. c. the number of moles of solute dissolved in. 1 L of solution. d. describes liquids that are insoluble in each other. e. ... Chapter 16 Test REVIEW ...

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Modern Chemistry 129 Chapter Test Chapter: Acids and Bases PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. Which of the following is an oxyacid? a. HCl b. H₂O c. H₂S d. H₂SO₄ ____ 2. Which acid is produced in the stomach? a. HNO₃ b. CH₃COOH c ...

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CHAPTER 16, Solutions (continued) 7. Look at Figure 16.2 on page 472. Circle the letter of each sentence that is true about a saturated solution. a. The total amount of dissolved solute remains constant. b. The total mass of undissolved crystals remains constant. c. When the rate of solvation equals the rate of crystallization, a state of

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A)43 B)0.60 C)160 D)50 E)28 26) A solution is prepared by dissolving 6.00 g of an unknown nonelectrolyte in enough water to make 1.00 L of solution. The osmotic pressure of this solution is 0.750 atm at 25.0°C. What is the molecular weight (g/mol) of the unknown solute? A)195 B)110 C)16.4 D)30.6 E)5.12 x 10⁻³ 4

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