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chemical calculations, mole ratios are used to convert between moles of reac- ... 360 Chapter 12 Section 12.2 (continued) Writing and Using Mole Ratios Using Visuals ... (Acceptable answers include the number of customers, the number of cylinders per customer, the amount of ammonia per ...

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### **Chapter 12**

Chemistry Chapter 12.2. ... In chemical calculations, mole ratios are used to convert between a given number of moles of a reactant or product to moles of a different reactant or product. is a conversion factor derived from the coefficients of a balanced chemical equation interpreted in terms of moles.

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**Chapter 12 - Stoichiometry - 12.2 Chemical Calculations ...**

Chapter 11 Small-Scale Lab Section 11.3 Precipitation Reactions: Formation of Solids, page 345 Analysis 1. a.  $\text{Na}_2\text{CO}_3 + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3 + \text{Ag}_2\text{CO}_3(\text{s})$  b.  $2\text{Na}_3\text{PO}_4 + 3\text{Pb}(\text{NO}_3)_2 \rightarrow 6\text{NaNO}_3 + \text{Pb}_3(\text{PO}_4)_2(\text{s})$

**Chapter 11 Small-Scale Lab**

Chapter Test A, Chapter Test B Technology • Computer Test Bank, Chapter 12 Test • Interactive Textbook with ChemASAP, Chapter 12 • Virtual Chem Labs, Lab 14 378 Chapter 12 12 CHAPTER Study Guide Key Concepts 12.1 The Arithmetic of Equations • A balanced chemical equation provides the same kind of quantitative information that a recipe does.

**CHAPTER 12 Study Guide - Quia**

Chemical Kinetics Answers: (a)  $8.4 \times 10^{-7} \text{ M/s}$ , (b)  $2.1 \times 10^{-7} \text{ M/s}$  SAMPLE EXERCISE 14.3 continued The decomposition of  $\text{N}_2\text{O}_5$  proceeds according to the following equation: If the rate of decomposition of  $\text{N}_2\text{O}_5$  at a particular instant in a reaction vessel is  $4.2 \times 10^{-7} \text{ M/s}$ , what is the rate of appearance of (a)  $\text{NO}_2$ , (b)  $\text{O}_2$ ?

**Chapter 14 Chemical Kinetics - University of Massachusetts ...**

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