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A. 25.00 mL aliquot of concentrated hydrochloric acid (11.7 M) is added to 175.00 mL of 3.25 M hydrochloric acid. a. is this a dilution? b. determine the number of moles of hydrochloric acid from the 175.00 mL of 3.25 M hydrochloric acid. c. determine the number of moles of hydrochloric acid in the solution after the addition of the concentrated hydrochloric acid. assume the volumes are additive.

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Dilutions Worksheet W 329 - Everett Community College

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