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Example 13. Show that the reaction.  $\text{Fe}_2\text{O}_3(\text{s}) + 3\text{SO}_3(\text{g}) \rightarrow \text{Fe}_2(\text{SO}_4)_3(\text{s})$ . can be written as a combination of formation reactions. Solution. There will be three formation reactions. The one for the products will be written as a formation reaction, while the ones for the reactants will be written in reverse.

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### **pH and Color Change | Chapter 6: Chemical Change | Middle ...**

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### **Introductory Chemistry - Open Textbook Library**

Chapter 6 – Oxidation-Reduction Reactions 67 thus creating a voltaic cell, which is often called a

battery. This section describes the fundamental components of voltaic cells and describes several different types.

### **Chapter 6 Oxidation-Reduction Reactions - Mark Bishop**

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