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Answers will vary, but a typical answer might be 5.17 10^{-7} m. 8. Calculate the mass of the drop and record the value. You can use the answer from #7 for the radius (r) to calculate the mass of the drop given the density of the oil. The final equation to calculate the mass, in kg, is $m = \rho_{oil} \cdot \frac{4}{3} \pi r^3$ $\rho_{oil} = 821 \text{ kg m}^{-3}$ $m = 3439.0 \text{ kg m}^{-3} \cdot r^3$

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Read the following description of an experiment. Then answer the questions. All plants need water, minerals, carbon dioxide, sunlight, and living space. If these needs are not met, plants cannot grow properly. A scientist wanted to test the effectiveness of different fertilizers in supplying needed minerals to plants.

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Flame Test Lab Activity Key Note: If chloride compounds are not available, metal nitrate compounds may be substituted. Use dilute or approximately 0.1 M solutions Unknowns: Number the beakers 1-12. Add solutions as follows: 1. Strontium Chloride 7. Calcium Chloride 2. Calcium Chloride 8. Strontium Chloride 3. Potassium Chloride 9. Barium Chloride

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