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There were 2 steps to this question, and i got the first one doing $M_1V_1=M_2V_2$. I do not understand how to get the second part? any help appreciated thx (a) How many milliliters of a stock solution of 11.0 M HNO_3 would you have to use to prepare 0.450 L of 0.500 M HNO_3 ? Answer= 20.5 mL b) If you dilute 21.0 mL of the stock solution to a final volume of 0.500 L, what will be the concentration of ...

Chemistry solutions help please?? easy 10 points? | Yahoo ...

This is taking me all day to do. >.< Can someone please help me? Give me a website, provide the answers... i don't care as long as i can get this done in the next hour. THANK YOU. Traditional acids are _____ compounds that ionize in water to produce hydronium ions. One way to prepare a salt is by the reaction of a _____ with an acid.

Simple Chemistry "fill in the blank" questions ... help ...

Likewise, simple pictures of uninteresting and garden variety chemistry-related things are not appreciated. If a caption or explanation is included this helps, but please use your discretion. ... I want to make a caffeine solution. Little help please? (self.chemistry) submitted 5 years ago by zeedude.

I want to make a caffeine solution. Little help please ...

Making Solutions A simple solution is basically two substances that are evenly mixed together. One of them is called the solute and the other is the solvent. A solute is the substance to be dissolved (sugar). The solvent is the one doing the dissolving (water). As a rule of thumb, there is usually more solvent than solute.

Chem4Kids.com: Matter: Solutions - Chemistry basics for ...

This section will focus on what you need to know about solutions, solution concentrations, and colligative properties in order to be successful on the SAT II Chemistry test. This material is closely tied in with the material from the first half of this chapter and "The Structure of Matter."

SparkNotes: SAT Chemistry: Solutions

Chemistry Solutions Practice Problems 1. Molar solutions. a. Describe how you would prepare 1 L of a 1 M solution of sodium chloride. The gram formula weight of sodium chloride is 58.44 g/mol. Answer: To make a 1 M solution of sodium chloride, dissolve 58.44 g sodium chloride in 500 mL water in a 1000-mL volumetric flask. When all the solid is ...

Chemistry Solutions Practice Problems | Carolina.com

This is a collection of worked general chemistry and introductory chemistry problems, listed in alphabetical order. I have included printable pdf chemistry worksheets so you can practice problems and then check your answers. You may also browse chemistry problems according to type of problem.

Worked Chemistry Problems and Worksheets - ThoughtCo

I'm Tyler, and if you're having trouble with Chemistry, I want to help. I have tons of videos on almost every topic in Chemistry, and I'm making more all the time.

Tyler DeWitt - YouTube

This is the general knowledge questions and answers section on "Chemistry" with explanation for various interview, competitive examination and entrance test. Solved examples with detailed answer description, explanation are given and it would be easy to understand.

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