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From your answers to questions 9 and 10, formulate a definition for boiling point in terms of atmospheric pressure and vapor pressure of the liquid. The boiling point is the temperature at which the vapor pressure of a liquid equals the atmospheric pressure.

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a) Given your answers to questions 4-6, what happens to the rate of the reaction if you increase the temperature of the reactants? b) Try to think of two reasons WHY and write them down.

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