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Answer: a) Nucleons: The particles found inside the nucleus i.e., protons and neutrons are called nucleons. b) Atomic number: The number of protons present in the nucleus of an atom is called the atomic number. It is denoted as Z. The number of protons is equal to the number of electrons.

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These doors have a prompt or question on them and can be opened to reveal student's answers. ... The novel situation in this case is for students to explore topics that are even more complex than atomic structure: 1) Sub-Atomic Particles: Students explore the research question: Is there anything smaller than a proton or neutron in an atom ...

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Atomic Structure Review Multiple-choice exercise. Choose the correct answer for each question. Show all questions <= => The nucleus of most atoms is made up of: ? protons and electrons ? ... Of the basic atomic particles, the one that would be attracted to a negatively charged metallic plate is the: ?

Atomic Structure Review - ScienceGeek.net

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