

Chapter 7 Modern Chemistry Review Answers

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Chemistry > Modern Chemistry > Pg. 251 Modern Chemistry Modern Chemistry ISBN: 9780030367861 / 0030367867. Section PG Go. Chapter Review ... Chapter Review. 1. See explanation and examples 2 answ. 2. see explanation 1 answ. 3. see explanation 2 answ. 4. 2 answ. 5. a)Potassium ion b)Magnesium ion ...

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

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Modern Chemistry 57 Chemical Bonding CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Label each of the following statements as True or False: _____ a. If the formula mass of one molecule is x u, the molar mass is x g/mol. _____ b.

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds

Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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7. What is the relationship between vaporization and evaporation? Evaporation is a form of vaporization. It occurs only in nonboiling liquids when some liquid particles enter the gas state. Vaporization is a more general term that refers to either a liquid or a solid changing to a gas. 84 STATES OF MATTER MODERN CHEMISTRY

10 States of Matter - Ms. Agostine's Chemistry Page

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Modern Chemistry 7 Solutions CHAPTER 12 REVIEW Solutions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

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