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Chapter 14 The Behavior of Gases147 SECTION 14.1 PROPERTIES OF GASES(pages 413–417) This section uses kinetic theory to explain the properties of gases. This section also explains how gas pressure is affected by the amount of gas, its volume, and its temperature. Compressibility (pages 413–414) 1. Look at Figure 14.1 on page 413.

SECTION 14.1 PROPERTIES OF GASES(pages 413-417)

Behavior of Gases Answer Key. In answering the question below, please be sure to include BOTH your work and your answer. When 3.56 grams of a compound is vaporized at 245°C and 1.20 atm, the resulting gas occupies a volume of 344 mL.

Behavior of Gases Answer Key - HelpTeaching.com

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a measure of how much the volume of matter decreases under pressure. Gases are easily compressed because. of the space between the particles in a gas. The amount of gas, volume, and temperature are factors that. affect gas pressure. If the temperature is constant, as the pressure of a gas increases, the volume decreases.

Chapter 14 The Behavior of Gases Flashcards | Quizlet

14.1 Properties of Gases 14.2 The Gas Laws 14.3 Ideal Gases 14.4 Gases: Mixtures and Movements Chapter 14: The Behavior of Gases study guide by dhrutip41 includes 16 questions covering vocabulary, terms and more. Quizlet flashcards, activities and games help you improve your grades.

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This section explains how materials are classified as solids, liquids, or gases. It also describes the behavior of these three states of matter. Comparing and Contrasting As you read about the states of matter, replace each letter in the diagram below with one of these phrases: definite volume, definite shape, variable volume, or variable shape.

Chapter 3 States of Matter Section 3.1 Solids, Liquids ...

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Gases are easily, or squeezed into a smaller volume 1. because of the between particles in a gas. The four variables 2. used to describe a gas are pressure, (P), (V), (T), 3. and number of (n). 4. You can use theory to predict and explain how gases 5. will respond to a change in conditions. Doubling the amount of 6. gas in a rigid container the pressure.

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Chapter 14

Behavior Of Gases Workbook Answers Behavior of Gases Answer Key. In answering the question below, please be sure to include BOTH your work and your answer. When 3.56 grams of a compound is vaporized at 245° C and 1.20 atm, the resulting gas occupies a volume of 344 mL. Behavior of Gases Answer Key - HelpTeaching.com Chapter 14 The Behavior of ...

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Chapter 14

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Section 3: Behavior of Gases Gas Laws The volume of a gas is the same as the volume of its container but there are other factors, such as pressure, to consider The gas law describes how the behavior of gases are affected by pressure and temperature

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