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when 2.00 moles of potassium phosphate react with 2.00 moles of aluminum nitrate? 8. sulfuric
acid + sodium hydroxide Chemistry IF8766 Page 61 Instructional Fair, Inc. A sample of argon gas is
cooled and its volume went from 380 mL to 250 mL, If its final Gas

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Name: _____ The Mole and Volume Directions: For gases at STP (273K and 1.00 atm pressure), one
mole occupies a volume of 22.4L. What volume will the following quantities of gasses occupy at

STP? 1) 5.00 mole of H

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Balancing Chemical Equations -Answer Key Balance the equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$ 6) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$

Balancing Chemical Equations Answer Key

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