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SUMMARY 1.1 Identity ...Chapter 10 Chemical Quantities Pearson Answers Chemical Quantities THE MOLE AND QUANTIFYING MATTER 10.1 The Mole: A Measurement of Matter Essential Understanding The mole represents a large number of very small particles.

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CHAPTER 10: Chemical Quantities BASICS: • The basic unit that is used to determine the amount of a chemical substance is called a mole • A mole(mol) of a substance is equivalent to 6.02×10^{23} particles of that substance • The mole was founded by a scientist named Avagadro, and he decided to use the

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10.3 Percent Composition and Chemical Formulas

Chapter 10 - Chemical Quantities 1. Chemical Quantities Or How I Learned to Love the Mole 3. The Mole How do you measure matter? If it is something large like the number of beans in a bag, you just count them. If, however, it is something exceedingly small like an atom or ion, how do you count it?

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05 CTR ch10 7/9/04 3:29 PM Page 256 Chapter 10 Review ...

Chapter 10 Chemical Quantities 243 Section Review Objectives • Convert the mass of a substance to the number of moles of a substance, and the number of moles of a substance to mass • Calculate the volume of a quantity of gas at STP Vocabulary • Avogadro's hypothesis • standard temperature and pressure (STP) • molar volume Key Equations • mass (grams) number of moles

Objectives Vocabulary Key Equations

10.1 The Mole: A Measure-ment of Matter 26 > Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved.. What are the different ways you can

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Chapter 10 Chemical Quantities91 SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287-296) This section defines the mole and explains how the mole is used to measure matter. It also teaches you how to calculate the mass of a mole of any substance. Measuring Matter (pages 287-289) 1.

SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287-296)

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its chemical formula and can be used to convert from mass to moles of that compound. ... 320 Chapter 10 • The Mole Section 110.10.1 Measuring Matter MAIN Idea Chemists use the mole to count atoms, molecules, ... mole quantities of water, copper, and salt are shown, each with a different

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