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### **Chemquest 45 Intro To Equilibrium**

Taking into account your answer to question 5, rearrange the equation that you wrote in number four and get  $K_f$  and  $K_r$  on the left side of the reaction and everything else on the right side.  $K_f K_r = [H_2O][CH_4][CO][H_2]^3$  Information The Equilibrium Constant The equilibrium constant is a constant that allows us to compare the ...

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Big Idea: Reactions quantitatively describe the changing interactions between atoms. NOTE - there are three parts to this unit Part 1: Be able to write and balance chemical equations

### **Unit 5: Reactions - Creekside Science**

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