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A.P. Chemistry Practice Test: Ch. 11, Solutions Name_____ MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) Formation of solutions where the process is endothermic can be spontaneous provided that _____. A)the solvent is a gas and the solute is a solid

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warmer solution the particles of the liquid solvent are moving more rapidly. Because of this, there is an increased rate in solute-solvent interaction, which increases the rate of

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answer. $q = mc \Delta T = (110.0 \text{ g})(4.18 \text{ J/(g}\cdot\text{°C)})(35.6 \text{ °C} - 15.0 \text{ °C})$... Propose an experimental test the student could perform that could be used to support the hypothesis. Explain how the results of the test would support the hypothesis if the substance was ionic. ... AP Chemistry Scoring Guidelines, 2016

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