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Stoichiometry: Limiting Reagent 1. $N_2 + 3H_2 \rightarrow 2NH_3$ How many grams of NH_3 can be produced from the reaction of 28g of N_2 and 25 g of H_2 ? 2. How many grams of the excess reagent in Problem 1 is left over? 3. $Mg + 2HCl \rightarrow MgCl_2 + H_2$ What volume of hydrogen at STP is produced from the reaction of 50.0g of Mg and the equivalent of 75.0g ...

Stoichiometry: Limiting Reagent - Ms. Agostine's Chemistry ...

Name: _____ Stoichiometry: Mole-Mole Problems 1. $N_2 + 3H_2 \rightarrow 2NH_3$ How many moles of hydrogen are needed to completely react with 2.00 moles of nitrogen?

Stoichiometry: Mole-Mole Problems

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Gram Formula Mass Instructional Fair Answer Key

NAMING HYDROCARBONS Name Name the compounds below according to the IUPAC naming system 1. $H-H-H-I-I-I-H-C-C-C-H-I-I-I-H-H-H$ 5. $H-H-H-I-I-I-H-C-C-C-H$... Chemistry IF8766 96 ©instructional Fair, Inc. Name m, e, p, b STRUCTURE OF HYDROCARBONS

NAMING HYDROCARBONS Name - Ms Galloway

Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name _ ... Page 39 Chemistry IF8766 Page 38 TYPESOF CHEMICAL BONDS Name-----Cla ssify th ef ollowl ng comp u d aslonl (me tal+ meta l), cova (nonm + ... Page 40 112 ©Instructional Fair,Inc. ...

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ANSWER: 0.89 mole Chemistry IF8766 ... 8-14a,b Mixed Problems--Mole-Mole and Mole-Mass wkst-Key. KEY. Answer each of the following ... Gram Formula Mass Practice Worksheet - TMACHEMISTRY - home

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Stoichiometry Mixed Problems Instructional Fair Answers

COMPOSITION OF HYDRATES Name A hydrate is an ionic compound with water molecules loosely bonded to its crystal structure. The water is in a specific ratio to each formula unit of the salt.

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Stoichiometry: Mixed Problems (KEY) 1) $N_2 + 3H_2 \rightarrow 2NH_3$ What volume of NH_3 at STP is produced if 25.0 of N_2 is reacted with an excess of H_2 ? 3 3 3 2 3 2 2 2 40.0L NH_3 1mol NH_3 22.4L NH_3 1mol N 2mol NH_3 28.0g N 25.0g N 1mol N $\times \times \times =$ 2) $2KClO_3 \rightarrow 2KCl + 3O_2$ If 5.0g of $KClO_3$ is decomposed, what volume of O_2 is produced at STP? 2

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