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Chemistry: Form WS3.3.1A Name ____ PERIODIC TABLE Date ____ Period ____ What are the Trends in the Periodic Table? Below is a portion of the periodic table. In the answer spaces provided in the table, fill in the [1] atomic number, ... Answer the questions below by referring to the data on the table you filled in on the first page. 1.

What are the Trends in the Periodic Table?

Chemistry I Name ____ KEY ____ Unit 7 WS 3 Yet More Balancing Equations Write balanced chemical equations for the following reactions. 1. Ammonia (NH_3) reacts with aqueous hydrogen chloride to form ammonium chloride. $\text{NH}_3 + \text{HCl} \rightarrow \text{NH}_4\text{Cl}$ 2. Calcium carbonate decomposes upon heating to form calcium oxide and carbon dioxide.

7.05b WS3 Key - Chemistry I Name KEY Unit 7 WS 3 Yet More ...

Chemistry: Form WS3.2.1A Name ____ PERIODIC TABLE Date ____ Period ____ How is the Periodic Table Arranged? Below is a portion of the periodic table. In the answer spaces provided in the table, fill in the [1] atomic number, ... Answer the questions below by referring to the data on the table you filled in on the first page. 1.

How is the Periodic Table Arranged?

Chemistry Form 3 Answer all questions 1 (a) Name the elements present in the following compounds i. Calcium carbonate (1 1/2 mks) ii. Ammonium nitrate (1 1/2 mks) iii. Iron sulphate (1 mk) iv. Tri-iron tetra oxide (1 mk) b. Lead and copper have a valence of 2. Write the formula of i. Lead (ii) carbonate (1mk) ii. Copper (ii) chloride (1mk) iii.

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form two molecules of water, there are no atoms missing and ... Chemistry 1F8766 ©Instri Jctinnnl Fnir. Name Write the word equations below as chemical equations and balance, 1, zinc + lead (II) nitrate yield zinc nitrate + lead ... Answer the questions below by circling the number of the correct response 1. If 46 g of X combines with 16 g of ...

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