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Chemistry Matter and Change: Chapter 12. Kinetic Molecular Theory. describes the behavior of matter in terms of particles in motion. Elastic Collision. collision in which no kinetic energy is lost. Temperature. measure of the average kinetic energy of the particles in a sample of matter.

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Section 12 .2 Forces of Attraction. 1. intermolecular forces. 2. intermolecular forces. 3. Dispersion forces are weak forces that result from temporary shifts in the density of electrons in electron clouds.

ch 12 Study guide TE - Mr. McKnight Clawson High School

Supplemental Problems Chemistry: Matter and Change • Chapter 2 1 Data Analysis Data Analysis 1. A sample of aluminum is placed in a 25-mL graduated cylinder containing 10.0 mL of water. The level of water rises to 18.0 mL. Aluminum has a density of 2.7 g/mL. Calculate the mass of the sample. 2. Saturn is about 1 429 000 km from the Sun.

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