

## *Chemistry Worksheet Isotope Notation Answers*

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Chemistry Worksheet Isotope Notation NAME: \_\_\_\_\_ Block: \_\_\_\_\_ Metal, non- metal, or metalloid? \_\_\_\_\_ Complete the table. Recall that A = mass number, Z = atomic number, and ions have charge!  $\pm$  Name 24 Mg 30 108Pd 13) 35 s2- 112 Cd2+ 52 2- 24 # of Atom or Atomic Mass # of 53 94 56 no 30 125 69 80 ion? Atom Atom Atom

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2. Define isotope: 3. The number of protons in an atom is known as the of that atom. 4. The number of determines the name of the atom. 5. The mass number of an atom is the number of plus the number of in the nucleus of the atom. 6. The isotope notation for nitrogen-15 is as follows: a.

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Chemistry Worksheet Isotope Notation NAME: Block: 1. Uranium-235 and uranium-238 are considered isotopes of one another. How are uranium-235 similar, and how are they different? 0-235 and have 92 protons 0-235 has 235-92 = 143 neutrons 0-238 has 92 protons 0-238 has 238-92 = 146 neutrons 2. Define isotope: atoms of the same element with different numbers of neutrons 3. The number of protons in an atom is ...

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Any given element can have more than one isotope. To distinguish between the different isotopes of an atom, the element is named with its mass number, for example lithium-7. ... Answer the following questions about atoms. 1. ... Isotope Notation Name \_\_\_\_ Chem Worksheet 4-2 7 Li 3  
EXAMPLES Lithium-7 3 protons

Isotopes Answer Key. Isotopes Answer Key - Displaying top 8 worksheets found for this concept.. Some of the worksheets for this concept are Isotopes, Atoms and isotopes work, H ws isotope abundance key, Atoms and isotopes work, Isotopes, Abundance of isotopes name chem work 4 3, Average atomic mass problems key 2013, Chemistry work isotope notation answer key.

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Isotope and Ion Notation Extra Practice Worksheet This is a worksheet of extra practice problems for students who struggled with the ions and ion notation worksheet, and/or the isotopes and isotope notation worksheet. Students are given a simple table that gives limited information about an isotope or ion, and they fill in the rest.

**Isotope and Ion Notation Extra Practice Worksheet ...**

Model 1 that supports your answer. NO, CARBON-12 AND CARBON-13 ARE ISOTOPES OF CARBON BUT HAVE DIFFERENT MASS NUMBERS. 12. Considering your answers to Question 11, work with your group to write a definition of isotope using a sentence. ISOTOPES ARE ATOMS OF THE SAME ELEMENT WITH DIFFERENT MASS NUMBERS. 13.

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