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Balancing Chemical Equations -Answer Key Balance the equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$ 6) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$

Balancing Chemical Equations Answer Key

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r i c Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name _ Ionicbondingoccurswhenametcl hcnstersonormoreelectrons10anonmelolln on ...

trom .PO.' NO, - d2ct263enury6r.cloudfront.net

Naming Ionic Compounds Directions: Name the following compounds using the Stock Naming System. 1. CaCO_3 calcium carbonate____ 2. KCl potassium chloride____ 3. FeSO_4 iron (II ...

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Naming Ionic Compounds - Ms. Agostine's Chemistry Page

Word Equations Answer Key 1. Zinc and Lead (II) nitrate react to form Zinc Nitrate and Lead. $\text{Zn} + \text{Pb}(\text{NO}_3)_2 \rightarrow \text{Zn}(\text{NO}_3)_2 + \text{Pb}$ Single Replacement 2. Aluminum Bromide and Chlorine gas react to form Aluminum Chloride and Bromine gas. $2\text{AlBr}_3 + 3\text{Cl}_2 \rightarrow 2\text{AlCl}_3 + 3\text{Br}_2$ Single Replacement 3.

Word Equations Answer Key Zinc and Lead (II) nitrate react ...

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Writing Formulas From Names Directions: Write the formulas of the following compounds. 1. ammonium phosphate $(\text{NH}_4)_3\text{PO}_4$ 2. iron (II) oxide FeO 3. iron (III) oxide Fe_2O_3 ... Chemistry IF8766 Page 48 Instructional Fair, Inc. Title: Name:

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