

## ***Chapter 20 Oxidation Reduction Reactions Answers Pearson Lesson Check***

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### Chapter 20 Oxidation Reduction Reactions

Chapter 20-Oxidation Reduction Reactions. because magnesium is a better reducing agent than iron and is more easily oxidized. so the magnesium immediately transfers electrons to the iron; preventing their oxidation to iron ions. So magnesium is sacrificed by oxidation and protects iron.

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636 Chapter 20 Redox Reactions Figure 20-1. The reaction of magnesium and oxygen involves a transfer of electrons from magnesium to oxygen. Therefore, this reaction is an oxidation-reduction reaction. Using the classifications given in Chapter 10, this redox reaction also is classified as a combustion reaction.

### Chapter 20: Redox Reactions - Neshaminy School District

Chapter 20: Oxidation-Reduction Reactions. The oxidation number of hydrogen in a compound is, except in metal hydrides, such as NaH, where it is -1.

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Chapter 20 Worksheet: Redox. I. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also. 1.  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$  2.  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$  3.  $2\text{Cs} + \text{Br}_2 \rightarrow 2\text{CsBr}$  4.  $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$  5.  $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$  6.  $\text{Cl}_2 + 2\text{NaBr} \rightarrow 2\text{NaCl} + \text{Br}_2$  7.  $\text{Si} + 2\text{F}_2 \rightarrow \text{SiF}_4$  8.  $2\text{Ca} + \text{O}_2 \rightarrow 2\text{CaO}$  9.

### Chapter 20 Worksheet Redox - bhhs.bhusd.org

Chapter 20: Oxidation -Reduction reactions Section 20.1 - The meaning of oxidation and reduction. Why does this happen to cars? Why is it less of an issue in Arizona compared to other states? Has the statue of liberty always been green? Oxidation reactions Oxidation reactions were originally

### Chapter 20: Oxidation -Reduction reactions

Chapter 20: Oxidation-Reduction Reactions Citations Oxidation-Reduction Reactions Here you will learn all about Redox Equations. Oxidation / Reduction Reactions ... You can tell a reaction is a redox reaction if the number of an element is different on both sides of the equation.

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