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18.3.3 The vapor pressure of a component in an ideal solution One of the properties of an ideal solution that can be easily calculated is the equilibrium - Notes on the Thermodynamics of Solids J.W.

Morris, Jr.; Fall, 2008

### Chapter 18: Solutions - Materials Science

18-1 CHAPTER 18 ACID-BASE EQUILIBRIA 18.1 The Arrhenius definition classified substances as being acids or bases by their behavior in the solvent water. 18.2 All Arrhenius acids contain hydrogen and produce hydronium ion ( $\text{H}_3\text{O}^+$ ) in aqueous solution. All Arrhenius bases contain an OH group and produce hydroxide ion ( $\text{OH}^-$ ) in aqueous solution ...

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Chapter 16 Solutions 167 SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471-477) This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves. Solution Formation (pages 471-472) Look at Figure 16.1 on page 471 to help you answer Questions 1 and 2. 1.

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