

## *Chemactivity 9 Shell Model Answers*

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is consistent with your answer to part c. 18. Construct a shell model diagram of F that is consistent with the information in Table 1. 30 ChemActivity 5 The Shell Model (II) 19. Locate F and Cl on the periodic table. a) Describe any relationship between the core charge of these atoms, the number of valence electrons, and their position in the ...

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4.The IE of  $\text{Li}^+$  should be larger than the IE of He because both atoms have 2 electrons in the 1st shell and  $\text{Li}^+$  has a core charge of +3 whereas He only has a core charge of +2. ... ChemActivity 9. 1.a) Two. b) Lower energy peak (1. s ... Answers Last modified by ...

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a shell model of the atom, and noted that the number of valence electrons in the ... In order to answer this question, we must consider ionization energies in greater detail. Model 1: Ionization Energies and Energy Levels ... ChemActivity 8 Photoelectron Spectroscopy 50 received. So, if IE is the ionization energy of the electron and KE is the ...

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