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Chapter 9 Practice Test - Naming and Writing Chemical Formulas

Chapter 9 – Chemical Calculations and Chemical Formulas 119 Chapter 9 Map Chapter Checklist Read the Review Skills section. If there is any skill mentioned that you have not yet mastered, review the material on that topic before reading this chapter. Read the chapter quickly before the lecture that describes it.

Chapter 9 Chemical Calculations and Chemical Formulas

MODERN CHEMISTRY STOICHIOMETRY 73 ... CHAPTER 9 REVIEW Stoichiometry SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. b The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products. ... mc06se_cFMsr_i-vi.qxd Author: williams

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e. molar mass of $\text{Fe}(\text{NO}_3)_3 = 241.9 \text{ g}$ 272 g $\text{Fe}(\text{NO}_3)_3$ 1 mol 241.9 g = 1.12 mol $\text{Fe}(\text{NO}_3)_3$ f. molar mass of $\text{FeSO}_4 = 151.9 \text{ g}$ 12.7 mg FeSO_4 1 g 1000 mg 1 mol 151.9 g = 8.36 10^{-5} mol FeSO_4 4 g. molar mass of $\text{LiOH} = 23.95 \text{ g}$ 6.91 103 g LiOH 1 mol 23.95 g = 288.5 = 289 mol LiOH 14. a. molar mass of $\text{CaCO}_3 = 100.1 \text{ g}$ 2.21 10^{-4} mol CaCO_3 3 100.1 g 1 mol = 0 ...

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CHAPTER 9, Chemical Names and Formulas (continued) 8. Use the periodic table to write the name and formula (including charge) for each ion in the table below. Polyatomic Ions (pages 257–258) 9. What is a polyatomic ion? 10. Is the following sentence true or false? The names of polyatomic anions always end in -ide. ____ 11.

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