

## *Chapter 9 Review Stoichiometry Section 2 Answers*

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### **Chapter 9 Review Stoichiometry Section**

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Stoichiometry. SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_ The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products. (b) relative number of moles of reactants and products. ... CHAPTER 9 REVIEW ...

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**Chapter 9 Review Stoichiometry Worksheet Answers**

Chapter 9 Section 1 Intro to Stoichiometry including use of molar mass and BEMR (Balanced Equation Mole Ratio)

**9.1 Introduction to Stoichiometry**

Chapter 9 Chemical Quantities 1. Although we define mass as the “amount of matter in a substance,” the units in which we measure mass are a human invention. Atoms and molecules react on an individual particle-by-particle basis, and we have to count individual particles when doing chemical calculations. 2.

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