

## *Chemquest 15 Answers*

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**Chemquest 15 Answers**

15. Complete the following: Elements ending in d<sup>2</sup> are in column number 4 or IVB and those ending in d<sup>4</sup> are in column number 6 or VIB. Therefore, elements ending in d<sup>3</sup> are in column number 5 or VB. 7 Elements ending in d must be in column number 9 or VIIIB. 16. Notice from question 6 that an element that ends in 3s is in the 3rd row. Also, from ...

**ChemQuest 15 - Yav Science**

46 ChemQuest 15 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready.

**46 ChemQuest 15 Name: - Yumpu**

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ChemQuest 8 Information: Structure of the Atom Note the following symbols: (they are not to scale) ... 15 45 21 64 cu 29 131 54 237 93 16 2- 70 31 52 Cr 24 5b Mass number 231 108 137 119 80 . ... 21Ne (10% abundant). Calculate the average atomic mass of Ne. Include proper units with your answer. ( Average Atomic Mass Calculate the average ...

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15. The charge (in the upper right hand corner of the element symbol) is -1 for a fluorine ion. Why isn't it +1 or some other number? 16. What is the charge on every atom ? Why is this the charge? 17. How do you determine the charge on an ion? 18. An oxygen ion has a -2 charge. (Use your periodic table if necessary)

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**ChemQuest #16 Part 1 Ms Holder**

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