

Chemlab 11 Hydrated Crystals Answers

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Pre-Lab: Hydrated Crystals Answer the following, where appropriate answer in complete sentences. If a calculation is required, you must show your work. 1. Define: a) Hydrated compound - A hydrated compound is a compound that is surrounded by water molecules that are held in place by hydrogen bonds.

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If you please take the time to read this.I'd be very thankful.and answer a few questions. Introduction:A hydrated crystal is a typically an ionic compound that has water molecules trapped within the crystal.This usually occurs because the compounds are hygroscopic;which means they absorb water from the air.Since the structure of most hydrates are unknown or verycomplex,a dot(.) is used in the ...

HYDRATED CRYSTALS CHEMISTRY LAB HELP - Yahoo Answers

CHEMISTRY LAB: HYDRATED CRYSTALS WHAT TO TURN IN: Hypothesis Data Table Calculations (5) Questions (4) OBJECTIVES To review and observe the characteristics of a hydrate To find the experimental and theoretical percentages of water in the hydrate

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Hydrated crystals. Hydrates are compounds that incorporate water molecules in their crystalline structures. The ratio of moles of water to one mole of that compounds is a small whole number. For example, in the hydrated compound copper (II) sulfate pentahydrate ($\text{CuSO}_4(5\text{H}_2\text{O})$), the ratio is 5:1.

CHEMLAB 11 - Yola

Questions i need to answer: 1.Compare the number of moles or anhydrous MgSO_4 , to the number of moles of water in the hydrate. Use the ratio of these two values to predict a formula for hydrated MgSO_4 . 2.The method used i this experiment to find the percentage of water in the hydrated crystals is not suitable for all hydrates.

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