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Chapter 11 Chemical Reactions113 SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329) This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329)

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11.1 Describing Chemical Reactions 11

Study Guide: Chapter 11—Chemical Reactions Key Concepts 11.1—Describing Chemical Reactions *How do you write a word equation? To write a word equation, write the names of the reactants t

the left of the arrow separated by a plus sign; write the names of the products to the right of the arrow, also separated by plus signs.

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11.2 Types of Chemical Reactions> - Useful Advice

11.2 Types of Chemical Reactions> 29 Whether one metal will displace another metal from a compound depends upon the relative reactivities of the

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Chapter 11 - Chemical Reactions - 11.1 Describing Chemical Reactions - 11.1 Lesson Check - Page 354: 7 Answer Write the chemical formulas of the reactants on the left of the yield sign and the chemical formulas of the products on the right of the yield sign.

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In general, how are word equations written to describe chemical reactions? 68. Describe how to write the skeleton equation for the following reaction: iron + oxygen iron(III) oxide. 69. Consider the description: copper(II) oxide reacts with sulfuric acid to produce copper(II) sulfate and water. ... Chapter 11 Practice Test ...

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