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**NCERT Solutions For Class 12 Chemistry Chapter 1 The Solid ...**

Class 12 Chemistry. Solid State NCERT InText Solution. Question: 1.1 - Why are solids rigid? Answer: The particles of solids are close packed and can only oscillate about their fixed positions. These properties make solids rigid. Question: 1.2 - Why do solids have a definite volume?

**12 Chemistry Solid State NCERT In Text Solution part 1**

NCERT Solutions Class 12 Chemistry The Solid State PDF. In this the particles are of indefinite shape and have short range order. Example of amorphous solids are: Quartz glass. Amorphous solids are also called supercooled or pseudo solids. liquids or This was the introduction to the chapter 1 of chemistry class 12.

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NCERT Class XII Chemistry: Chapter 1 - The Solid State. National Council of Educational Research and Training (NCERT) Book for Class XII Subject: Chemistry Chapter: Chapter 1 - The Solid State. After studying this Unit, you will be able to. describe general characteristics of solid state; distinguish between amorphous and crystalline solids;

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CBSE & NCERT Chemistry Notes for Class 12 Solid State In this article, we have discussed the about the solid state of matter, and the two different types of solids. It covers the Unit 1: Solid State, of CBSE Class 12th Chemistry .

**CBSE & NCERT Chemistry Notes for Class 12 Solid State**

NCERT Solutions for Class 12 Chemistry Chapter 1 The Solid State. INTEXT Questions. Question 1. Why are solids rigid? Solution: Unlike gases and liquids in which the molecules are free to move about and hence constitute fluid state, in a solid, the constituent particles are not free to move but oscillate about their fixed positions providing them a rigid structure.

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www.ncerthelp.com (Visit for all ncert solutions in text and videos, CBSE syllabus, note and many more) Chemistry Notes for class 12 Chapter 1 The Solid State Solids Solids are the chemical substances which are characterised by definite shape and volume, rigidity, high density, low compressibility. The constituent particles (atoms, molecules or ions)

**Chemistry Notes for class 12 Chapter 1 The Solid State**

Solid State InText Solution 2 Question: 1.13 - Explain how much portion of an atom located at (i) corner and (ii) bodycentre of a cubic unit cell is part of its neighbouring unit cell.

**12 Chemistry Solid State NCERT In Text Solution part 2**

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