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CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.

6 Chemical Bonding - Effingham County Schools / Overview

CHAPTER The Periodic Table and Periodic Law Section 6.1 Development of the Modern Periodic Table 1 session 1/ 2 block P LS 6 ... Math in Chemistry, p. 167 Key to Teaching Strategies Glencoe Technology Multiple Learning Styles ... students answer the accompanying questions using Section Focus Transparency Master 20.

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Rates. CHAPTER 6 ANSWERS AND EXPLANATIONS. Multiple-Choice 1. D The answer to (A) shows all of the reactants and products that are part of this reaction. However, species that appear on both sides need to cancel.

CHAPTER 6 ANSWERS AND EXPLANATIONS - Chemical Reactions ...

Chapter 6 Chemistry in Biology Neutrons and protons are located at the center of the atom. ...

Chapter 6. Reactants and Products A chemical reaction is the process by which atoms or groups of atoms in substances are reorganized into different substances. 6.2 Chemical Reactions

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f. mass of 6 mol C = 6 (12.01 g) = 72.06 g mass of 6 mol H = 6 (1.008 g) = 6.048 g mass of 1 mol O = 16.00 g = 16.00 g molar mass of C₆H₆O = 94.11 g 21. a. molar mass of SO₃ = 80.07 g 49.2 mg SO₃ 1 g 1000 mg 1 mol 80.07 g = 6.14 10⁻⁴ mol SO 3 b. molar mass of PbO₂ = 239.2 g 7.44 x 10⁴ kg PbO 2 1000 g 1 kg 1 mol 239.2 g = 3.11 10⁵ mol PbO 2

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