

Chemistry 11 Describing Chemical Reactions Answer Key

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Chemistry 11 Describing Chemical Reactions

Chapter 11 - Chemical Reactions - 11.1 Describing Chemical Reactions - Sample Problem 11.1 - Page 349: 1 Answer The addition of solid sodium in water yields hydrogen gas and aqueous sodium hydroxide.

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In a chemical reaction, one or more reactants change into one or more products. Figure 11.1a shows the ingredients for making leavened bread— flour, salt, yeast, and water. Chemical reactions take place when the ingredients are mixed together and heated in the oven. Figure 11.1b shows the product—a loaf of bread.

11.1 Describing Chemical Reactions 11

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These are the reactants in a chemical reaction that takes place when the ingredients are mixed together and heated in the oven. Figure 11.1b shows the product—a loaf of bread. The four ingredients (reactants) were changed to bread (the product). Like all chemical reactions, baking bread involves changing substances.

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11.2 Types of Chemical Reactions> 32 Double-Replacement Reactions This is an example of a double-replacement reaction, which is a chemical change involving an exchange of positive ions between two compounds. •Double-replacement reactions are also referred to as double-displacement reactions.

11.2 Types of Chemical Reactions> - Quia

Chapter 11 Chemical Reactions113 SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329) This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329)

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Chapter 11

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Chapter 11: Chemical Reactions. Jennie L. Borders. Section 11.1 – Describing Chemical Reactions. In a chemical reaction, the reactants are written on the left and the products on the right. The arrow that separates them is called yield. Reactants Products. Symbols in Equations. Symbol.

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11.2 Types of Chemical Reactions> 13 A decomposition reaction is a chemical change in which a single compound breaks down into two or more simpler products. • Decomposition reactions involve only one reactant and two or more products. • The products can be any combination of elements and compounds. • Most decomposition reactions require ...

11.2 Types of Chemical Reactions> - Useful Advice

Unformatted text preview: Chapter 11 Matter and Change 11.1 Describing Chemical Reactions: Introduction to Chemical Equations: Chemical reactions involve changing substances (reactants change to one or more products). Chemists use a chemical equation to convey as much information as possible about what happens in a chemical reaction.

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