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* applicable from Session 2012-13 onwards ** Syllabi of the subjects remain the same as in the previous scheme except for those given

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where y i represents the headspace mole fraction of the ith component of a mixture with N components, M i is the molecular mass of component i (g/mol), P is the total pressure in the gas phase (Pa), R is the ideal gas constant and T is the temperature (K).. The liquid phase consisting of the perfume mixture was considered as a non-ideal solution in equilibrium with the headspace concentration.

The diffusion of perfume mixtures and the odor performance ...

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