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Stoichiometry of a Precipitation Reaction. weigh 1g of CaCl2 +2 H2O, then pour it into the 100mL glass beaker with 25mL of distilled water and then stir it to make the solution calcium chloride. The next step was to calculate the amount of Na2CO3 I needed for it to completely react with the CaCl2 +2H2O using stoichiometry. 1.0g CaCl2+2H2O= 1mol CaCl2+2H2O =0.006802mol CaCl2+ 2H2O 147.0156g ...

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Chemistry: Flame Test Lab Flame Test Lab Morgan Chemistry CP 5/30/14 Abstract- In the flame test lab, the flame test was performed to excite the electrons in the samples and observe the color of the flameThe flame emits a color because each element has an exactly defined emission spectrum, which one can use to identify them. For example, NaCl was highlighter yellow, Sr(NO3)2 was sun orange ...

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