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Chemfiesta Answers

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Balancing Chemical Equations Worksheet Answer Key Chemfiesta

Balancing Equations Worksheet – Answers Note to students: It is acceptable to leave spaces blank when balancing equations – blank spaces are interpreted as containing the number “1”.

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Answers – Naming Chemical Compounds . Name the following chemical compounds: 1) NaBr sodium bromide. 2) $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$ 2 calcium acetate. 3) P_2O_5 diphosphorus pentoxide. 4) $\text{Ti}(\text{SO}_4)_2$ 2 titanium(IV) sulfate. 5) FePO_4 iron(III) phosphate. 6) K_3N potassium nitride. 7) SO_2 sulfur dioxide. 8) CuOH copper(I) hydroxide. 9) $\text{Zn}(\text{NO}_3)_2$

Answers - Naming Chemical Compounds

Balancing Equations Worksheet – Answers Note to students: It is acceptable to leave spaces blank when balancing equations – blank spaces are interpreted as containing the number “1”.

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How many liters of a 0.88 M solution can be made with 25.5 grams of

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Types of Reactions Worksheet Indicate the type of reaction taking place:

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Solutions to the Molar Mass Practice Worksheet: Important note to students: All of the units given here are “grams per mole”, which may be abbreviated as “g/mol”, “grams/mol”, or “g . mol -1 ”, depending on

Molar Mass Practice Worksheet - My Chemistry Class

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5) Some students believe that teachers are full of hot air.

Charles' Law Worksheet - Jackson County School District

Naming Worksheets. Posted on January 13, 2015 by misterguch. If you're anything like me (and pray that you aren't), one of your favorite things in the whole world is to name chemical compounds. I just sit and name compounds all day long, happy in the knowledge that one day the world will need a compound naming guru to save our species. With ...

Naming Worksheets | The Cavalcade o' Chemistry

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Solutions to the Ideal gas law practice worksheet: The ideal gas law states that $PV=nRT$, where P is the pressure of a gas, V is the volume of the gas, n is the number of moles of gas present, R is the ideal gas

Ideal Gas Law Practice Worksheet - Jackson County Schools

big difference in the final answer). 3) If I leave 750 mL of 0.50 M sodium chloride solution uncovered on a windowsill and 150 mL of the solvent evaporates, what will the new concentration of the sodium chloride solution be? 0.63 M (this is the opposite of a dilutions problem – the V 2 value is smaller than V 1

Dilutions Worksheet - Florida State University

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Balancing Equations Race - Solutions 1) $1\text{ C}_3\text{H}_8 + 5\text{ O}_2 \rightarrow 3\text{ CO}_2 + 4\text{ H}_2\text{O}$ 2) $2\text{ Al} + 1\text{ Fe}_2\text{N}_3 \rightarrow 2\text{ AlN} + 3\text{ Fe}$ 3) $2\text{ Na} + \text{Cl}_2 \rightarrow 2\text{ NaCl}$ 4) $2\text{ H}_2\text{O}_2 \rightarrow 2\text{ H}_2\text{O} + 1\text{ O}_2$

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