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Naming Ionic Compounds Directions: Name the following compounds using the Stock Naming System. 1.  $\text{CaCO}_3$  calcium carbonate\_\_\_\_ 2.  $\text{KCl}$  potassium chloride\_\_\_\_ 3.  $\text{FeSO}_4$  iron (II) ...  
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Writing Formulas From Names Directions: Write the formulas of the following compounds. 1. ammonium phosphate  $(\text{NH}_4)_3\text{PO}_4$  \_\_\_\_ 2. iron (II) oxide  $\text{FeO}$  \_\_\_\_ 3. iron (III) oxide  $\text{Fe}_2\text{O}_3$  ...  
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### **Writing Formulas From Names - Ms. Agostine's Chemistry Page**

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### **Answer Key Chemistry If8766 Moles And Mass**

Chemistry IF8766 OInstructional Fair, Inc. STOICHIOMETRY: MOLE-MOLE PROBLEMS I.  $N_2 + 3H_2$  Name How many moles of hydrogen are needed to completely react with two moles of nitrogen?  $2.0 + 302$  ... Chemistry 'Instructional Fair, tnc. ...

### **new.schoolnotes.com**

Stoichiometry: Mixed Problems (KEY) 1)  $N_2 + 3H_2 \rightarrow 2NH_3$  What volume of  $NH_3$  at STP is produced if 25.0 of  $N_2$  is reacted with an excess of  $H_2$ ? 3 3 3 2 3 2 2 40.0L  $NH_3$  1mol  $NH_3$  22.4L  $NH_3$  1mol N 2mol  $NH_3$  28.0g N 25.0g N 1mol N  $\times \times \times =$  2)  $2KClO_3 \rightarrow 2KCl + 3O_2$  If 5.0g of  $KClO_3$  is decomposed, what volume of  $O_2$  is produced at STP? 2

### **Stoichiometry: Mixed Problems (KEY)**

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