

Chemistry Of Life Chapter Test A Answers

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Chemistry Of Life Chapter Test

Chapter 2 Test B 17 Name _____ Class _____ Date _____ Chapter 2 The Chemistry of Life Chapter Test B Multiple Choice Write the letter that best answers the question or completes the statement on the line provided. _____ 1. The three particles that make up an atom are a. protons, neutrons, and isotopes.

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Biology 1 Chemistry of Life Chemistry of Life Chapter 2 Test DO NOT WRITE ON THE TEST!!!!!! MULTIPLE CHOICE Choose the letter of the best answer. 1. Which molecules are not carbohydrates? a. Simple sugars c. Polysaccharides b. Hydrolytic enzymes d. Disaccharides 2. Which represents a carbohydrate molecule? a. C 6 H 6 O 6 c. C 6 H 12 O 5 b. C 12 ...

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The Chemistry of Life Of the 92 naturally existing elements on the Earth, only 25 play a role in the chemical processes of life. Of these 25, four elements constitute more than 98 percent of all biological matter: carbon (C), oxygen (O), hydrogen (H), and nitrogen (N).

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An element is a substance made up of atoms that have the same number of protons. – For example, each atom of the element carbon has six protons. – Atoms of an element may have different numbers of neutrons. These atoms are called isotopes of elements. . CHEMICAL BONDS. Why do atoms form bonds?

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Chapter 2: The Chemistry of Life Review Questions. Section 2-1: The Nature of Matter: The basic unit of matter is called a(n) _____. Describe the nucleus of an atom. _____ Complete the table about subatomic particles. Particle Charge Location in the Atom Positive Neutral Negative Why are atoms neutral despite having charged particles?

Chapter 2: The Chemistry of Life Review Questions

774 Chapter 24 The Chemistry of Life CHAPTER 24 What You'll Learn You will learn the functions of the four major classes of biological molecules: proteins, carbohydrates, lipids, and nucleic acids. You will identify the building blocks that form the major biological molecules. You will compare and contrast the metabolic processes of ...

Chapter 24: The Chemistry of Life - Neshaminy School District

Chapter 2: Introduction to the Chemistry of Life Figure 2.1 Foods such as bread, fruit, and cheese are rich sources of biological macromolecules. (credit: modification of work by Bengt Nyman) The elements carbon, hydrogen, nitrogen, oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things.

Chapter 2: Introduction to the Chemistry of Life ...

Learn how chemistry makes life possible! From you, to your dog, to your dinner, to the global ecosystem, all living systems are made out of atoms that obey the basic rules of chemistry. Here, you can learn about the key properties of atoms, including what particles they contain, how they are organized, and how they form chemical bonds with one another.

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View Test Prep - Chemistry of Life packet 1 ANSWER KEY (1) from BIOLOGY 109 at Lee University. The Chemistry of Life ANSWER KEY Section 21 The Nature of Matter (pages 3539) This section identifies

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