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Use the following passage and table to answer questions 11–14. A student studying electron configurations in atoms created Table 1 in order to summarize the various facts she learned about principal energy levels, energy sublevels, and orbitals. 11. Based on Table 1 what is the relationship between the principal energy level number (n)

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Use the following passage and table to answer questions 11–13. A student researched the properties of both ionic and covalent compounds. To summarize his findings, he created a table comparing the characteristics of the two compounds. Table 2 is shown below. Table 2 Comparison of Ionic and Covalent Compounds

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