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Modern Chemistry 70 Chapter Test Name Class Date Chapter Test B, continued \_\_\_\_ 7. In a reaction, the ions of two compounds exchange places in aqueous solution to form two new compounds. This reaction is called a a. synthesis reaction. b. decomposition reaction. c. single-displacement reaction. d. double-displacement reaction. \_\_\_\_ 8.

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### **Assessment Chapter Test A - Kettering City School District**

Modern Chemistry 34 Chapter Test Name Class Date Chapter Test B, continued PART IV Write the answers to the following questions in the space provided. 27. What two early twentieth-century observations involving the interaction of light and matter could not be explained solely by the wave theory of light? 28.

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Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1.

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Modern Chemistry 85 Chapter Test Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_ Chapter Test A, continued \_\_\_\_\_ 6. Which factor is the most important in determining the average kinetic energy of gas particles? a. pressure b. temperature c. volume of the container d. mass of the container \_\_\_\_\_ 7. Which gas is most likely to deviate from ideal gas behavior? a. Ne b. CO<sub>2</sub> c. ...

**Assessment Chapter Test A - Kettering City School District**

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CHAPTER 8 REVIEW Class \_\_\_\_\_ Chemical Equations and Reactions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg. 2. Consider the metals iron and silver, both listed in Table 3 on page 286 of the text. Which one

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