

Chemfax Thermodynamics Lab Answers

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Chemfax Thermodynamics Lab Answers

Thermodynamics: Enthalpy of Reaction and Hess's Law Judy Chen Partner: Mint Date: 13 Sept, 2011
Purpose: The purpose of this lab is verify Hess's law by finding the enthalpies of the reactions;
NaOH and HCl, NH₂Cl and NaOH, and NH₃ and HCl. The overall enthalpy should equals to the sum
of enthalpy of the three reactions in

Thermodynamics: Enthalpy of Reaction and Hess's Law

3.) The specific heat of a solution is 4.18J/(g°C °) and its density is 1.02g/mL. The solution is formed
by combing 25.0mL of solution A with 25.0mL of solution B, with each solution initially at 21.4 °C.
The final temperate of the combined solutions is 25.3 °C. Calculate the heat of reaction, q rxn,
assuming no heat loss to the calorimeter.

Pre-lab Questions - Thermodynamics-Enthalpy of Reaction ...

Chemfax is a Calgary, Alberta based company that has grown tremendously over the past decade.
Chemfax proudly operates out of a 60,000 sq. ft. State-of-the-Art facility which includes one of the
largest Class 1, Division 1, flammable liquid handling areas in Canada.

Chemfax: Manufacturer of Specialty Chemicals.

Thermodynamics - Enthalpy Of Reaction And Hess's Law. This lab demonstrates the principle of
Hess's law. Hess's law states that the change in enthalpy is the same whether a reaction takes
place in one step or in a series of steps.

Thermodynamics -Enthalpy Of Reaction And Hess's Law

Thermodynamics-Enthalpy of Reaction and Hers's Law Page 3 3. The specific heat of a solution is
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Thermodynamics-Enthalpy of Reaction and Hess's Law

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Chemistry - Flinn Scientific

Hess's Law Lab Demonstration with NaOH and HCl (Part 2: Data & Calculation) - Julia Le - Duration:
8:39. Phuong (Julia) Le 29,325 views

Hess Law Lab Calculations

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