

## *Chemquest 51 Answers*

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**Chemquest 51 Answers**

a) If this reaction took place in 2.0 L of solution, what is the concentration of leftover hydrochloric acid after the reaction  $\text{mol leftover} = 2.5 - 1.9 = 0.6 \text{ mol} \div 2.0\text{L} = 0.3 \text{ M}$  b) From your answer to part a, verify that the pH of the solution after the reaction is approximately 0.52.  $\text{pH} = -\log(0.3) = 0.52$ .

**154 ChemQuest 51 Name: - Yumpu**

chemquest 51 answers F195DFEBADE0405A69381F389A0CA9B0 Chemquest 51 Answers a) If this reaction took place in 2.0 L of solution, what is the concentration of leftover ...

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**HW Keys - Roosevelt High School AP Chemistry 2017-18**

Title: ChemQuest 8: Answer Key Author: Kosiorowski Last modified by: Kosiorowski Created Date: 8/28/2008 11:49:00 PM Company: Sony Electronics, Inc.

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a) Given your answers to questions 4-6, what happens to the rate of the reaction if you increase the temperature of the reactants? b) Try to think of two reasons WHY and write them down.

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**Chem Quest 32 Answer Key**

148 ChemQuest 49 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Measuring Acidity Many people have heard of "pH", and many people know that it probably refers to acidity, but not very many understand what exactly it measures. ... Use the information given and your answers to questions 1 and 2 to find the following: a) Divide the pH by the ...

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