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Chemistry 1412 CHEM 1412 Final. 2 CHEM 1412 FINAL EXAM Version F-1/06A PART I (2 points each) Multiple choice - Please DO NOT write or mark on this paper. Write your correct answer on scantron. 1. In the reaction,  $\text{HSO}_4^- + \text{HS}^- \rightleftharpoons \text{H}_2\text{S} + \text{SO}_4^{2-}$ , which one of the sets below constitutes a conjugate (acid-

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CHEM 1412 - General Chemistry II (with lab) CHEM 1412 Practice Exams; CHEM 1412 Practice Exams Chapters covered vary slightly from semester to semester, and the actual exams may not have formulas included. CHEM 1412 Practice Exam 1, Chapters 11-13 Zumdahl These are the same chapters as 13, 14, and 15 in the current Brown textbook.

### **CHEM 1412 Practice Exams — HCC Learning Web**

CHEM 1412 Practice Exam 2, Chapters 15, 16, and 17 Chang \_\_\_\_\_  $\text{pH} = \text{pK}_a + \log \left( \frac{[\text{conj. base}]}{[\text{weak acid}]} \right)$   $G = H - TS$   $G = G^\circ + RT \ln Q$   $G^\circ = -RT \ln K$  eq  $R = 8.314 \text{ J/mole K}$  Circle your choice on multiple choice questions. Show your work for calculation problems. A periodic table is attached to the end of the exam. Good luck!

### **1412 Practice Exam 2 - HCC Learning Web**

Chemistry 1412. Dr. Pahlavan. CHEM 1412 Exam #1 Sample Exam # 1A Name: \_\_\_\_\_ ... PART I - ( 3 points each) - Please write your correct answer next to each question number. DO NOT CIRCLE \_\_\_\_ 1. In which colligative property(ies) does the value decrease as more solute is added? a. boiling point b. freezing point

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