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2 and 3). Follow the rules for writing ionic formulas listed in the background material. In our example, the formula is AlCl<sub>3</sub>. Record the chemical formula for the ionic compound in Column 6 of the data table. 9. Write the name of the ionic compound following the rules for naming ionic compounds discussed in the background information.

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The first time seeing formulas and names of ionic compounds can be confusing. However, upon further investigation, there are actually simplistic formulated steps to do so. Discover a systematic method of writing formulas for ionic compounds without the need to experimentally determine each formula using cation and anion cards.

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6. Write the formula for the ionic compound, using subscripts to indicate the number of each kind of ion used (from Columns 2 and 3). Follow the rules for writing ionic formulas listed in the background material. In our example, the formula is AlCl<sub>3</sub>. Record the chemical formula for the ionic compound in Column 6 of the data table. 7.

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Write the formula for each reactant on the left hand side of the equation and the formula for each product on the right side of the equation. 3. Count the number of atoms of each element on the reactants side and compare this to the number of atoms of each element on the products side.

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