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Balancing Chemical Equations -Answer Key Balance the equations below: 1)  $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$  2)  $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$  3)  $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$  4)  $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$  5)  $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$  6)  $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$

### Balancing Chemical Equations Answer Key

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### Instructional Fair Inc Chemistry If8766 27

r i c Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name \_ Ionicbondingoccurswhenametcl hcnstersonemoreelectrons10anonmelolln on ...

### trom .PO.' NO, - d2ct263enury6r.cloudfront.net

Naming Ionic Compounds Directions: Name the following compounds using the Stock Naming System. 1.  $\text{CaCO}_3$  calcium carbonate\_\_\_\_ 2. KCl potassium chloride\_\_\_\_ 3.  $\text{FeSO}_4$  iron (II ...

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### **Naming Ionic Compounds - Ms. Agostine's Chemistry Page**

Word Equations Answer Key 1. Zinc and Lead (II) nitrate react to form Zinc Nitrate and Lead.  $\text{Zn} + \text{Pb}(\text{NO}_3)_2 \rightarrow \text{Zn}(\text{NO}_3)_2 + \text{Pb}$  Single Replacement 2. Aluminum Bromide and Chlorine gas react to form Aluminum Chloride and Bromine gas.  $2\text{AlBr}_3 + 3\text{Cl}_2 \rightarrow 2\text{AlCl}_3 + 3\text{Br}_2$  Single Replacement 3.

### **Word Equations Answer Key Zinc and Lead (II) nitrate react ...**

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### **Chemistry If8766 Instructional Fair Inc - WordPress.com**

Writing Formulas From Names Directions: Write the formulas of the following compounds. 1. ammonium phosphate  $(\text{NH}_4)_3\text{PO}_4$  2. iron (II) oxide  $\text{FeO}$  3. iron (III) oxide  $\text{Fe}_2\text{O}_3$  ... Chemistry IF8766 Page 48 Instructional Fair, Inc. Title: Name:

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