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Thermochemistry is the scientific study of the heat energy that is involved in or produced by chemical reactions and/or physical transformations. What can you tell us about how this topic is studied and what its uses are? Take the quiz and we'll find out together!

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Chapter 17 Thermochemistry187 10. Complete the enthalpy diagram for the combustion of natural

gas. Use the thermochemical equation in the first paragraph on page 517 as a guide. SECTION 17.3 HEAT IN CHANGES OF STATE (pages 520–526) This section explains heat transfers that occur during melting, freezing, boiling, and condensing.

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505–510)

Chapter 17 Thermochemistry 431 Section Review Objectives • Construct equations that show the enthalpy changes for chemical and physical processes • Calculate enthalpy changes in chemical and physical processes Vocabulary Key Equation • $q_{\text{sys}} = -q_{\text{surr}} = m C \Delta T$, where $T_f > T_i$ Part A Completion Use this completion exercise to check your understanding of the concepts and terms

05 CTR ch17 7/12/04 8:15 AM Page 429 THE FLOW OF ENERGY ...

Review Unit 4 Test IPOD #29 – Chapter 13 concepts Review – Phase Changes Slide 24: Describe terms for phase changes Worksheet: Chapter 17 – Thermochemistry Notes Chapter 17 Notes, Slides 1-3: What is thermochemistry? Chapter 17 Notes, Slides 4-11: Heat, energy, calorie/joule conversions Chapter 17 Notes, Slides 12-13: Endo vs. Exo

McLaughlin, Kimberly / Thermochemistry

thermochemistry chemical potential energy heat • system surroundings ... Compare your answer to the values listed in Table 17.2 of your text to identify the unknown metal? 430 Core Teaching Resources . Name ... Chapter 17 Thermochemistry formation of mole of a compound from its elements.

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