

## *Chemthink Gases Answers*

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**Chemthink Gases Answers**

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questions; you must answer 15 questions correctly in order to receive credit. Questions for the Tutorial: 1. Describe the effect of increasing temperature upon the speed with which gas particles move. 2. Consider a sample of gas with particles of helium (He), neon (Ne) and argon (Ar). Compare the relative speeds of these three types of particles. 3.

**ChemThink - Behavior of Gases <http://www.chemthink>**

How to do the gases tutorial and questions.

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Due on Tuesday 02/20 \*Now complete the quiz questions. You must answer 15 questions correctly before missing 3 in order to receive credit. You may take the quiz more than once if needed. ChemThink Guide sheet for Behavior of Gases Name \_\_\_\_\_

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Great news! "The Behavior of Gases" has been ported from Flash to HTML5. This ChemThink tutorial took a lot of time to make, and it looks great! All of the original controls are here, and the question set is as challenging as ever.

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In answering the question below, please be sure to include BOTH your work and your answer. When 3.56 grams of a compound is vaporized at 245°C and 1.20 atm, the resulting gas occupies a volume of 344 mL. Based on this result, what is the molar mass of the compound?

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