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Chapter 13 - Gases - An Introduction to Chemistry

Chemistry Chapter 13 study guide questions. what is pressure? a property of all gas. It exerts pressure on (pushes against) the sides. what are the instruments that are used to measure pressure? barometer, what are the units used to describe pressure? atm, mmHg, torr, psi, pascals.

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Chemistry Chapter 13- Study Guide *Predict the "like-dissolves-like" rule of solubility in terms of intermolecular forces *Predict relative solubilities of substances from intermolecular forces. *Explain why gases have relatively low solubilities in water. General rule: like-dissolves-like—the simplest way is to determine the intermolecular forces in the molecule.

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Diffusion movement of one substance through another elastic ability of a substance or object to readily spring back to its original size or shape after being squeezed, hit, or subjected to similar disturbances kinetic-molecular theory model that explains the physical properties of gases based on the submicroscopic behavior of gas particles; in mathematical form, it [...]

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13. 16. In your textbook, read about using the ideal gas law to solve for molar mass, mass, or density. Use the following terms below to complete the statements. Each term may be used more than once. mass molar mass Volume per unit (20) divided by the and the of the gas must be known Study Guide The number of moles of a gas is equal to the (17 ...

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able to... Describe a . system in equilibrium . Find the reaction . quotient, Q , (K_{eq}) for a reaction equation . Solve for Q given . initial concentrations or pressures. Using . LeChatlier's principle

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Glencoe Chemistry - Matter And Change Chapter 13: Gases ...

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Chapter 13 - States of Matter Study Guide - Bossong ...

Study Guide for Content Mastery Answer Key Chemistry: Matter and Change T195 Name Date Class 76 Chemistry: Matter and Change • Chapter 13 Study Guide for Content Mastery Section 13.3 Liquids and Solids In your textbook, read about liquids and solids. In the space at the left, write true if the statement is true; if the statement is false,

13 STUDY GUIDE FOR CONTENT MASTERY 13 ... - HONORS CHEMISTRY

Chapter 13 Study Guide Chapter 13 Section 1 salinity: amount of salt dissolved in water thermocline: middle layer in ocean, does not receive a lot of sunlight so temperature drops drastically photosynthesis: process which plants make food, they use CO_2 and sunlight to produce a type of sugar density: when two items take up the same amount of ...

Chapter 13 Study Guide- Answer Key

Chemistry 11 Final Exam Study Guide Chemistry 11 - Final Exam Study Guide Page 15 When Electronegativities of bonding atoms are the same (as they are in diatomic molecules) or close to the same, they SHARE electrons. Bonds formed when atoms share electrons are called Covalent Bonds .

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