

## *Chemistry 12 Intext Questions With Solution*

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NCERT TEXTBOOK QUESTIONS SOLVED. 3.1. How would you determine the standard electrode potential of the system Mg<sup>2+</sup>/Mg? Ans: A cell will be set up consisting of Mg/MgSO<sub>4</sub> (1 M) as one electrode and standard hydrogen electrode Pt, H<sub>2</sub> (1 atm)/H<sup>+</sup> (1 M) as second electrode, measure the EMF of the cell and also note the direction of deflection in the voltmeter.

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