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This booklet has been produced to support chemistry teachers delivering the new GCE A level Chemistry specification (first assessment summer 2017). The booklet looks at questions from the Sample Assessment Materials.

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Converting Between Mass and Moles When converting between mass and moles it is important to understand what a mole is. A mole is a representation of how many particles a sample has. The number of moles can be expressed as a coefficient in front of the compound or element in a chemical equation. For example, the chemical

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Chapter 12

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1.1 The Scope of Chemistry > - stjoes.org

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Chapter 3 Chemical Reactions and Reaction Stoichiometry

The mole The mole, abbreviated mol, is the SI base unit used to measure the amount of a substance. A mole is defined as the number of carbon atoms in exactly 12 g of pure carbon-12. Through years of experimentation, it has been established that a mole of anything contains

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