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4. The bond energy (in MJ/mole) is the energy required to break one mole of the specified bonds. Triple bonds share six electrons between two atoms and are stronger (have greater bond energy) than double bonds, which share four electrons between two atoms.

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It's the energy of position/ stored energy between two stationary charged particles.  $q_1$  and  $q_2$  are the charges on the particles, d is the distance between them, and k is a positive-valued proportionality constant.

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