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Chapter 1 Introduction To Chemistry

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Chapter 1: Introduction to Chemistry

No embeds. No notes for slide. 1. Chapter 1 Introduction to Chemistry. 2. 1.1 Chemistry and Its Importance • Chemistry is the study of the composition, structure, properties and interactions of matter • In chemistry, we study:

What chemicals are made of and what are their properties.

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14 Chapter 1 • Introduction to Chemistry. Experiments A hypothesis is meaningless unless there are data to support it. Thus, forming a hypothesis helps the scientist focus on the next step in a scientific method—the experiment. An experiment is a set of controlled observations that test the hypothesis.

Chapter 1: Introduction to Chemistry

12 Chapter 1 An Introduction to Chemistry. SI Units Derived from Base Units. Many properties cannot be described directly with one of the seven SI base units. For example, chemists often need to measure volume (the amount of space that something occupies), and volume is not on the list of SI base units (Table 1.1).

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1.1 The subject of chemistry and what it can be used for In the most basic terms, chemistry is the study of matter and matter is anything that has mass and takes up space. Chemistry is related to the subject of physics because physics is the study of energy and energy affects matter so they are interrelated.

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Chemistry 101 Chapter 1 1 INTRODUCTION TO CHEM ISTRY • Chemistry is the science that deals with the materials of the universe, and the changes they undergo. • Materials of the universe can be of several forms: Gas: air, oxygen Liquid: water, gasoline, vinegar, orange juice,

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We assume that each reported number is ± 1 in the last decimal position reported. Therefore, we assume the mass of the graduated cylinder between 1124.1 g and 1124.3 g, the volume of the methanol is between 1.19 L and 1.21 L, and the total mass is between 2073.8 g and 2074.0 g. Chapter 2 Key Ideas

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