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Molecular refers to the mass of single molecule. but molar refers to the mass of a mole of something (or the mass of  $6.02 \times 10^{23}$  atoms or molecules) , Information: Beginning Mole Conversions The mole (often abbreviated as mol) is the link between the microscopic (atoms and molecules) and the macroscopic (Things measured in grams).

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16. Molecular refers to the mass of a single molecule, but molar refers to the mass of a mole of something (or the mass of  $6.02 \times 10^{23}$  atoms or molecules) . 17.  $x \text{ atoms mol} \times \text{atoms g mol}^{-1} = 23$  23 3.82 10 6.02 10 197.0 125  $x \times x \text{ molecules mol} \times \text{molecules g mol}^{-1} = 24$  23 1.17 10 6.02 10 108 210  $x \times x$

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