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NCERT Solutions for Class 12 Chemistry Chapter 4 Chemical Kinetics. INTEXT Questions. Question 1. For the reaction $R \rightarrow P$, the concentration of a reactant changes from 0.03 M to 0.02 M in 25 minutes.

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Class 12 Chemistry. Solid State NCERT InText Solution. Question: 1.1 - Why are solids rigid? Answer: The particles of solids are close packed and can only oscillate about their fixed positions. These properties make solids rigid. Question: 1.2 - Why do solids have a definite volume?

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Class 12 Chemistry. Solution In Text Questions 2. Question – 2.5 - Calculate (a) molality, (b) molarity and (c) mole fraction of KI if the density of 20% (mass/mass) aqueous KI is 1.202 g mL⁻¹. Answer: Given, 20% (mass/mass) aqueous solution of KI. This means 20 gm of KI is dissolved in 80 gm of water.

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NCERT Solutions Class 12 Chemistry UNIT 1 THE SOLID STATE Intext Questions. NCERT Solutions, CBSE Sample paper, Latest Syllabus, NCERT Books, ... Question 1.12: Distinguish between (i) Hexagonal and monoclinic unit cells (ii) Face-centred and end-centred unit cells.

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