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284 Study Guide for An Introduction to Chemistry Section Goals and Introductions Section 18.1 The Nucleus and Radioactivity Goals To introduce the new terms nucleon, nucleon number, and nuclide. To show the symbolism used to represent nuclides. To explain why some nuclei are stable and others not. To provide you with a way of predicting nuclear stability.

Chapter 18 Nuclear Chemistry

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Chapter 21

Nuclear Chemistry Kinetics of Radioactive Decay A wooden object from an archeological site is subjected to radiocarbon dating. The activity of the sample that is due to ${}^{14}\text{C}$ is measured to be 11.6 disintegrations per second.

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806 Chapter 25 Nuclear Chemistry Figure 25-2 Both Pierre and Marie Curie played important roles in founding the field of nuclear chemistry. Marie Curie went on to show that unlike chemical reactions, radioactivity is not affected by changes in physical conditions such as temperature and pressure. She is the only person in history to receive

Chapter 25: Nuclear Chemistry - Jayne Heier

Radioactivity •Radioactivity is the process by which nuclei emit particles and rays as they break down. •The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

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