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a) The carbon and nitrogen atoms are both sp^2 hybridized. The carbon-nitrogen double bond is composed of a sigma bond formed from two sp^2 orbitals, and a pi bond formed from the side-by-side overlap of two unhybridized 2p orbitals. b) As shown in the figure above, the nitrogen lone pair electrons occupy one of the three sp^2 hybrid orbitals ...

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Solutions to Chapter 2 Exercise Problems Problem 2

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Chapter 2 Exercise Solutions EX 2.1. What is the difference between the literals 4, 4.0, '4', and "4"? The literal 4 is an integer literal, with no fractional part. The literal 4.0 is a floating-point literal. The literal '4' represents the digit as a character. Using double quotes in the literal "4" make it a String literal. The first three are ...

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Convex Optimization Theory Chapter 2 Exercises and ...

Exercise 2-4 This change simply adds 2 to each service time in Table 2-1, so that they become 4.90, 3.76, 5.39, 6.52, 6.46, 6.36, 4.07, 5.36, 4.37, and 7.38. The hand-simulation table then looks like this (entries whose values or positions are changed due to the new service times are shaded):

Solutions to Exercises in Chapter 2 - Rivier

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Chapter 3 Exercise Solutions - Cornell University

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