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Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO 3 a. copper(II) carbonate Na 2SO 3 b. sodium sulfite (NH 4) 3PO 4 c. ammonium phosphate SnS 2 d. tin(IV) sulfide HNO 2 e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

View Notes - Chapter 7 Answer Key from CHEM 007 at University of Iowa. Chapter 7: Chemical Reactions DISCUSSION PLAN Chapter Summary Chapter 7 reviews the information about chemical changes and

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Chapter 7 Ionic and Metallic Bonding 7.1 Ions 7.2 Ionic Bonds and ... CHEMISTRY & YOU Sodium chloride is found in underground rock deposits as a solid. Like most ionic compounds, sodium chloride has a high ... Key Concepts Although they are composed of ions, ionic compounds are electrically neutral.

7.2 Ionic Bonds and Ionic Compounds > CHEMISTRY YOU

Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. ... Modern Chemistry 224 Answer Key TEACHER RESOURCE PAGE.

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CHEMISTRY NOTES - Chapter 7 Chemical Quantities Goals: To gain an understanding of: 1. Problem solving in chemistry. 2. The use of dimensional analysis to solve problems. 3. The concept of the mole. 4. The relationship between masses of substances and moles of substances. 5. The relationship between moles and the volumes and densities of ...

CHEMISTRY NOTES - Chapter 7 Chemical Quantities

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Chemistry 101 ANSWER KEY REVIEW QUESTIONS Chapter 4 1. Identify each of the following substances as a non-electrolyte (NE), weak electrolyte (WE), or strong electrolyte (SE): a)

Chemistry 101 ANSWER KEY 1 REVIEW QUESTIONS Chapter 4 1 ...

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