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Answers to Chapter 12 Practice Problems Question 1. Provide reagents (by each arrow) that will accomplish the following transformations. Yes, this is mostly from Chapter 10 – but it's still good practice! $\text{H}_3\text{C}-\text{CH}_3 \rightarrow \text{H}_3\text{C}-\text{OH}$ $\text{H}_3\text{C}-\text{CH}_3 \rightarrow \text{H}_3\text{C}-\text{H}$ $\text{H}_3\text{C}-\text{H} \rightarrow \text{H}_3\text{C}-\text{Cl}$ $\text{HO}-\text{CH}_3 \rightarrow \text{O}=\text{C}-\text{H}$ 1. NaH (or equivalent strong base such as BuLi) 2. CH_3I H_2 , Lindlar's catalyst or Na ...

Answers to Chapter 12 Practice Problems - Colby College

AP Chemistry Chapter 12 Review Multiple-choice exercise. Choose the correct answer for each question. Show all questions $\leq \Rightarrow$ The rate law for a particular reaction is determined to be $R = k[\text{A}]^2[\text{B}]$ 2. What is the overall order for this reaction? ? zero ? first ?

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12.1 FOCUS Objectives 12.1.1 Explain how balanced equations apply to both chemistry and

everyday situations. 12.1.2 Interpret balanced chemical equations in terms of moles, representative particles, mass, and gas volume at STP. 12.1.3 Identify the quantities that are always conserved in chemical reactions. Guide for Reading Build Vocabulary ...

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