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Concept Review: Energy. 1. energy 2. physical 3. chemical 4. endothermic 5. exothermic 6. kinetic 7. transferred 8. In any chemical or physical change, the total quantity of energy remains constant. Energy cannot be created or destroyed. 9. Heat is the enegy transferred between objects that are at different tempera- tures.

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Concept Review: Trends in the Periodic Table. 1. ionization energy 2. bond radius 3. electron affinity 4. electronegativity 5. increases, decreasing 6. d 7. a 8. c 9. The electron cloud model is based on the probability of finding an electron at a specific location.

Skills Worksheet Concept Review

Holt Science Spectrum 2 Nuclear Changes Skills Worksheet Concept Review Section: Nuclear Fission and Fusion 1.Write in the blank the term that matches each definition. critical mass fusion reaction strong nuclear force _____ a. attractive force that acts between protons and neutrons at very short distances

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Concept Review continued 6. Determine the mass in grams of 7.20 mol of antimony. 7. Determine the mass in grams of 0.500 mol of uranium. 8. Determine the mass in grams of 0.750 mol of francium. 9. A sample of lead has a mass of 150.0 g. What amount of lead in moles does the sample contain? 10. A sample of gold has a mass of 5.00 10 3 g. What ...

Skills Worksheet Concept Review

Concept Review: Energy 1. energy 2. physical 3. chemical 4. endothermic 5. exothermic 6. kinetic 7. transferred 8. In any chemical or physical change, the total quantity of energy remains constant. Energy cannot be created or destroyed.

Skills Worksheet Concept Review

10. Accept any reasonable answer. Sample answer: Yes; the data so far indicate that the Arctic is already warming faster than the rest of Earth. The more these regions warm, the faster they will continue to warm, and Alaska and other polar regions will show a greater temperature change than temperate and tropical regions. AGREE OR DISAGREE 11.

Skills Worksheet Concept Review

Concept Review Skills Worksheet. Concept Reviews SECTION: THE NATURE OF CHEMICAL REACTIONS 1. a. reactants: Fe ... RATES OF CHANGE 1. If a change is made to a system in chemical equilibrium, the equilibrium shifts to oppose the change until a new equilibrium is reached. 2.

Answers may vary. The table salt can be ground up into small particles. The

Concept Review - Manchester High School

Review 1. top row: Uranium-235 nuclei undergo a chain reaction. third row: Water absorbs energy from the coolant and changes to steam. fourth row: The steam turns a turbine attached to a generator. 2. Less than; when a nucleus forms, a small amount of mass changes to energy. 3. Nuclear fission; a heavier nucleus is splitting into lighter nuclei.

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