

Colligative Properties Questions And Answers

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Colligative properties Quiz - MCQs Questions and Answers ...

Answer all non-integer questions to at least 3 significant figures. Correct answers MUST be within ± 1 unit of the third significant figure or they are scored as wrong. ... Colligative properties depend upon: The type of solute particles The number of solute particles

Colligative Properties Exercises

Colligative Properties Problems - Answers . Remember that Colligative properties depend only on the number or concentration of particles in a solution. The properties are, for ideal solutions, independent of the kind or size of the particles, whether ionic, large or small etc. As a consequence:

Colligative Properties Problems - Answers

Best Answer: "1. Solubility is a measure of the quantity of a solute that will dissolve in a given amount of solvent. The rate of solution measures the time it takes for a given amount of solute to dissolve in a given amount of solvent and at a specific temperature. 2. Colligative properties of solutions ...

TWO Chemistry questions! Solubility and Colligative ...

Best Answer: The solution in which the concentration of the solute particles is the largest would have the largest effect on colligative properties. The molarities of all solutions are the same , but the molarities of the solute particles may be different in some of them. C_6H_6 , CCl_4 and PCl_3 are covalent molecular substance and the do not produce ions.

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Solutions colligative properties questions for interview, competitive examination and entrance test. These are most frequently asked questions. Interview; Current affairs ... ANSWER : the number of moles of a solute present in one litre of the solution. 2) Which one of the following is not a colligative property? ...

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