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Koretsky***

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goal. The intended audience is sophomore/junior students in chemical engineering. The book is divided into two parts. Part I covers the laws of thermodynamics, with applications to pure fluids; Part II extends thermodynamics to mixtures, with emphasis on phase and chemical equilibrium. The

selection of topics was guided

Fundamentals of Chemical Engineering Thermodynamics

Solution Thermodynamics CH2351 Chemical Engineering Thermodynamics II Unit – I, II
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Condensation of a vapor stream 107

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Chapter 1 Solutions. Engineering and Chemical Thermodynamics. Wyatt Tenhaeff Milo Koretsky.
Department of Chemical Engineering Oregon State University. koretsm@enr.orst.edu 1.2 An
approximate solution can be found if we combine Equations 1.4 and 1.5: $1.2 \text{ mV} \text{ ekmolecular } 2.3$
 $\text{kT ekmolecular } 2.3 \text{ kT V m}$. Assume the temperature is 22 C.

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Chemical Engineering Thermodynamics A Brief Review SHR Chapter 2 ... dP Nonideal liquid
solutions at moderate pressures and below T_c Henry's law $K_i = H_i P$ Low to moderate pressures
for species above T_c ... M. L. McWilliams, Chemical Engineering, 80(25), 1973 p. 138. $\ln K = a_1 T^2$
 $+ a_2 T + a_3 + b$

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