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What is the enthalpy (heat) of neutralization? Neutralisation is the reaction between an acid and a base to form a salt and water. Some examples of neutralisation reaction are as follows. During neutralisation reaction, hydrogen ions from acid react with hydroxide ions from alkali to form water. $\text{H}^+(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{aq})$ Since water [...]

What is the enthalpy of neutralization? - A Plus Topper

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Stoichiometry - Mass/Mass, Mass/Volume, Volume/Volume ...

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Content in this section of Wyzant Lessons has been curated by tutors, as well as by Chemtutor.com creator David Wilner. David Wilner is a native of Georgia, USA, born just before the end of WWII. Mr. Wilner is happiest when he is explaining things in science (pontificating), so it is natural that he has been a science teacher in high school and a teacher of chemistry at Georgia Perimeter College.

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The dynamics of surfactant interfacial layers was first discussed more than a century ago. In 1946 the most important work by Ward and Tordai was published which is still the theoretical basis of all new models to describe the time dependence of interfacial properties.

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