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Section: Drawing and Naming Molecules Complete each statement below by choosing a term from the following list. Terms may be used more than once. ... Holt Chemistry 6 Covalent Compounds Name Class Date Concept Review continued 10. Draw the Lewis structure for water, H 20. 11. Draw the Lewis structure for carbon monoxide, CO.

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Drawing And Naming Molecules Concept Review Answers

Start studying Chemistry Chapter 6.2 Drawing and Naming Molecules. Learn vocabulary, terms, and more with flashcards, games, and other study tools. ... (Review Chapter on how to draw them!) ... what the structures are called when molecules that have two or more possible Lewis structures Examples are Ozone (O3) and Sulfur Dioxide (SO2) ...

Chemistry Chapter 6.2 Drawing and Naming Molecules ...

Drawing and Naming Molecules. STUDY. PLAY. valence electrons. an electron that is found in the outermost shell of an atom and that determines the atom's chemical properties. Lewis structures. a structural formula in which electrons are represented by dots; dot pairs or dashes between two atomic symbols represent pairs in covalent bonds.

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Name Class Date Concept Review continued e. What do the 4 and 2 signify in 4H 2? How many hydrogen atoms are there? ... Concept Review: Describing Chemical Reactions 1. A chemical reaction is the process by ... The 4 signifies how many molecules of hydrogen are produced in the reaction; the 2 signifies the number of hydrogen atoms present in a

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