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2 Section-B 6. Calculate the freezing point of a solution containing 8.1 g of HBr in 100 g of water, assuming the acid to be 90 % ionized. [Given: Molar mass Br = 80 g/mol, Kf water = 1.86 K kg / mol]

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1. An object is placed between two plane mirrors inclined at an angle of 45° to each other. How many images do you expect to see? 2. A convex mirror is held in water.

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Let after a definite interval x mol/litre of B and y mol/litre of C are formed. i.e. This means that irrespective of how much time is elapsed, the ratio of concentration of B to that of C from the start (assuming no B and C in the beginning) is a constant equal to $k\ 1\ / k\ 2$.. Refer to the following video for parallel reactions:

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Jawahar Navodaya Vidyalaya - Wikipedia

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History of chemistry - Wikipedia

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