

Dilute Solution Definition

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Dilute Solution Definition

A: A dilute solution has a low concentration of the solute compared to the solvent. The opposite of a dilute solution is a concentrated solution, which has high levels of solute in the mixture. Continue Reading.

What Is a Dilute Solution? | Reference.com

Dilute Definition: A solution containing a relatively small quantity of solute as compared with the amount of solvent. This term is the opposite of 'concentrated'.

What Does Dilute Mean in Chemistry? - ThoughtCo

reduced in strength, as a chemical by admixture; weak: a dilute solution. RELATED CONTENT Excelsior, Stay In Your Lane, And Other Trending Words On Dictionary.com From the loss of one of the world's superheroes to the happenings on the world stage, take a look at the news stories that made people scurry to Dictionary.com to find meaning.

Dilute | Definition of Dilute at Dictionary.com

Dilute definition is - attenuate. How to use dilute in a sentence. Verb. You can dilute the medicine with water. The hiring of the new CEO diluted the power of the company's president. diluting the quality of our products . Adjective. a dilute solution of acid a dilute acid that's safe to handle in the classroom

Dilute | Definition of Dilute by Merriam-Webster

A dilute solution is a one that has only a little solute dissolved in a certain amount of solvent. In other words, a small amount of solute in between the solvent molecules. Do note that the solvent is not always water. Some examples of other solvents are alcohol and turpentine.

What is a dilute solution? What are some examples? - Quora

A dilute solution is one in which there is a relatively small amount of solute dissolved in the solution. A concentrated solution contains a relatively large amount of solute. These two terms do not provide any quantitative information (actual numbers) - but they are often useful in comparing solutions in a more general sense.

2.1 Dilute & Concentrated Solutions - Chemistry LibreTexts

Definition of 'dilute'. dilute. If a liquid is diluted or dilutes, it is added to or mixes with water or another liquid, and becomes weaker. If you give your baby juice, dilute it well with cooled, boiled water. The liquid is then diluted. The poisons seeping from contaminated land quickly dilute in the water. ...ditches dug for sewage dilution.

Dilute definition and meaning | Collins English Dictionary

For physical reasons, "infinite dilution" must be the same as pure solvent. However, when the concept is used, one most likely refers to a rate of change with respect to composition of the solution. For instance, the vapor pressure of a solution o...

What is the definition of an infinitely dilute solution ...

Definition of Dilution. Dilution is the process of reducing the concentration of a solute in solution, usually simply by mixing with more solvent Example 1: You can add water to concentrated orange juice to dilute it until it reaches a concentration that is pleasant to drink. Example 2: Pure oxygen is too rich for us to breathe...

Definition of dilution - Chemistry Dictionary

Define dilute. dilute synonyms, dilute pronunciation, dilute translation, English dictionary definition of dilute. tr.v. di-lut-ed , di-lut-ing , di-lutes 1. To make thinner or less concentrated by adding a liquid such as water.

Dilute - definition of dilute by The Free Dictionary

Dilution (equation) Dilution is the process of decreasing the concentration of a solute in a solution, usually simply by mixing with more solvent like adding more water to a solution. To dilute a solution means to add more solvent without the addition of more solute. The resulting solution is thoroughly mixed so as to ensure...

Dilution (equation) - Wikipedia

A concentrated solution is one in which there is a large amount of substance present in a mixture. The degree of concentration is measured in moles. An aqueous solution contains at least two substances: the solvent, water and a solute — the matter that will be dissolved in the water.

What Is a Concentrated Solution? | Reference.com

Biology-online is a completely free and open Biology dictionary with over 60,000 biology terms. It uses the wiki concept, so that anyone can make a contribution.

Dilute - Biology-Online Dictionary | Biology-Online Dictionary

dilute verb To weaken the potency of a factor or chemical in a solution by adding more solute, vehicle or diluent. dilute verb To weaken the potency of a thing by adding more vehicle or diluent. See Single molecular dilution.

Dilute | definition of dilute by Medical dictionary

But on the "dilution" of labour, the burning question of the hour, I shall have something to say in my next letter. The War on All Fronts: England's Effort | Mrs. Humphry Ward From these numbers we can, by help of the equation, calculate the conductivity of the acids for any dilution .

Dilution | Definition of Dilution at Dictionary.com

Dilution can also be achieved by mixing a solution of higher concentration with an identical solution of lesser concentration. Diluting solutions is a necessary process in the laboratory, as stock solutions are often purchased and stored in very concentrated forms.

Dilutions of Solutions | Introduction to Chemistry

Chapter 13 (Solutions) Definitions. Definitions from Chapter 13. STUDY. PLAY. Terms in this set (...) Soluble. capable of being dissolved. Solution. a homogeneous mixture of two or more substances in a single phase. ... Dilute. to make a liquid thinner by adding water or another solvent. Concentrated.

Chapter 13 (Solutions) Definitions Flashcards | Quizlet

'The new test uses a dilute solution of potassium permanganate.' 'The carbonates are washed repeatedly with a dilute solution of sodium carbonate, thus removing any coloration from the precipitate.' 'In dilute solution, starch molecules will precipitate, with the insoluble material being difficult to redissolve by heating.'

dilute | Definition of dilute in English by Oxford ...

Chapter 5 - Solutions study guide by lcroom includes 20 questions covering vocabulary, terms and more. Quizlet flashcards, activities and games help you improve your grades.

Chapter 5 - Solutions Flashcards | Quizlet

From the definition, we can calculate the number of moles of the solute, n ,: $n = M * V$ [2] Dilution. Dilution is the process where a solution is added more of the solvent to decrease the concentration of the solute. In dilution, the amount of solute does not change, the number of moles are the same before and after dilution.

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