

Definition Solution Biology

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Definition Solution Biology

Solution. A type of homogenous mixture in which the particles of one or more substances (the solute) are distributed uniformly throughout another substance (the solvent).

Solution - Biology-Online Dictionary | Biology-Online ...

Solution Definition. A solution is a homogeneous mixture of solvent and solute molecules. A solvent is a substance that dissolves another substance by pulling the molecules apart through electrochemical interactions. The solute then diffuses through the solvent until the concentration is equal in all parts of the solution.

Solution - Definition, Types and Examples | Biology Dictionary

An isotonic solution is one that has the same osmolarity, or solute concentration, as another solution. If these two solutions are separated by a semipermeable membrane, water will flow in equal parts out of each solution and into the other.

Isotonic Solution - Definition and Examples | Biology ...

A hypertonic solution refers to a solution that has a greater concentration of solute than another solution. In the context of biology, when two aqueous solutions are separated by a cell membrane, if the concentration of solute is greater outside the cell than inside the membrane, the solution is called hypertonic.

Hypertonic Solution: Definition And Role In Cell Biology ...

Solution Definition A solution is a homogeneous mixture of solvent and solute molecules. A solvent is a substance that dissolves another substance by pulling the molecules apart through electrochemical interactions.

*** Solution (Biology) - Definition, meaning - Online ...**

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Biology I - Chapter 2-2. For BIO 1 class. This is Section 2 (Properties of Water) of Chapter 2 (The Chemistry of Life). Word Bank: cohesion, adhesion, mixture, solution, solute, solvent, suspension, pH scale, acid, base, buffer. cohesion. the attraction between molecules of the same substance. adhesion.

Biology I - Chapter 2-2 Flashcards | Quizlet

The air we breathe, what we drink, as well as products in our household - these are a few examples of solutions we encounter every day. In this lesson, we will discuss what a solution is, the ...

What is a Solution in Science? - Definition & Examples ...

A solution is a homogeneous mixture of two substances—that is, it has the same distribution of particles throughout. Technically speaking, a solution consists of a mixture of one or more solutes dissolved in a solvent. The particles of solute and solvent are molecules or ions, with one or more solvent molecules bound to each solute particle.

Solution - definition of solution by The Free Dictionary

Hypotonic Solution: Definition, Example & Diagram. ... This is basically what happens when two solutions meet in biology - the two solutions compare their respective crowdedness to each other, and ...

Hypotonic Solution: Definition, Example & Diagram - Video ...

A solution consists of a solute and a solvent. The solute is the substance that is dissolved in the

solvent. The amount of solute that can be dissolved in solvent is called its solubility. For example, in a saline solution, salt is the solute dissolved in water as the solvent.

Solution Definition in Chemistry - ThoughtCo

By definition, a buffer system is a solution that resists a change in pH when acids or bases are added. Acidic solutions contain high concentrations of hydrogen ions (H^+) and have pH values less than seven. Basic solutions contain high concentrations of hydroxide ions (OH^-) and have pH values greater than seven.

What is the definition of buffer in biology? - Quora

An example of a hypertonic solution is the interior of a red blood cell compared with the solute concentration of fresh water. When two solutions are in contact, solute or solvent moves until the solutions reach equilibrium and become isotonic with respect to each other.

What Is a Hypertonic Solution? - ThoughtCo

An isotonic solution has an equal amount of dissolved solute in it compared to the things around it. Typically in humans and most other mammals, the isotonic solution is 0.9 weight percent (9 g/L) salt in aqueous solution, this is also known as saline, which is generally administered via an intravenous drip.

What is a solution in biology? | Yahoo Answers

Biology is the natural science that studies life and living organisms, including their physical structure, chemical processes, molecular interactions, physiological mechanisms, development and evolution. Despite the complexity of the science, there are certain unifying concepts that consolidate it into a single, coherent field.

Biology - Wikipedia

noun, plural: solutes. (chemistry) 1. a component of a solution: in a solution, the dissolving substance is called a solvent whereas the dissolved substance is called a solute. 2. a substance (usually in lesser amount) dissolved in another substance.

Solute - Biology-Online Dictionary | Biology-Online Dictionary

When talking about solutions, there are three words used to compare one solution to another. These words are hypotonic, isotonic, and hypertonic. Hypotonic means a lower concentration of solute (dissolved substance) compared to another solution, i...

What does 'isotonic' mean in biology? What are some ...

A hypertonic solution (pronounced "HĪ-per-TAWN-ik") is a solution with a higher concentration of solute (dissolved substance) than some other, specified solution (and which therefore has a higher osmotic pressure than the other solution). The solution with the lower concentration is then termed hypotonic.

Hypertonic solution | Online Biology Dictionary

AP Biology Chapter 3 study guide by hrahman includes 39 questions covering vocabulary, terms and more. Quizlet flashcards, activities and games help you improve your grades.

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