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Complete Lab 14A "Oxidation-Reduction Reactions of Elements and their Ions" Page 230 in Essential Experiments for Chemistry. Make sure that you answer all of the questions in the "Analysis of Results" Section as well as all of the "Follow-up" Questions.

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CORRELATION: ESSENTIAL EXPERIMENTS FOR CHEMISTRY AND THE CHEMISTRY ADVANCED PLACEMENT CURRICULUM AP Expt. Number AP Title Essential Expt. No. Essential Experiment for Chemistry Experiment 1 Determination of the formula of a compound 4B Determining the Empirical Formula of a Compound 2 Determination of the percentage of water in a hydrate

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In all, Essential Experiments for Chemistry is an excellent laboratory reference. It is an ideal text for both Grades Eleven and Twelve University Chemistry including AP. The activities are significantly better suited, better annotated and more varied than those you will find presented in student textbooks, and

Off the Mark - University of Waterloo

Lab #3 ***as always, refer to your green sheet and the lab reports you have done previously to avoid losing any marks unnecessarily Title: Purpose: 3 purposes/objectives together in one sentence OR pick out the objective from the handout that is most important Procedure: for the first time, we are going to reference a textbook (that the handout came from).

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