Definition Of Hypotonic Solution

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Definition Of Hypotonic Solution

Hypotonic Solution Definition. A hypotonic solution is a solution that has a lower solute concentration compared to another solution. A solution cannot be hypotonic, isotonic or hypertonic without a solution for comparison. Hypotonic is a description of the solute content of one solution in relation to another solution.

Hypotonic Solution - Definition and Examples | Biology ...

A hypotonic solution is any solution that has a lower osmotic pressure than another solution. In the biological fields, this generally refers to a solution that has less solute and more water than ...

Hypotonic Solution: Definition, Example & Diagram - Video ...

Hypotonic definition is - having deficient tone or tension. How to use hypotonic in a sentence.

Hypotonic | Definition of Hypotonic by Merriam-Webster

Tonicity is a measure of the effective osmotic pressure gradient, as defined by the water potential of two solutions separated by a semipermeable membrane. In other words, tonicity is the relative concentration of solutes dissolved in solution which determine the direction and extent of diffusion. It is commonly used when describing the response of cells immersed in an external solution.

Tonicity - Wikipedia

Tonicity Definition. Tonicity is the concentration of a solution as compared to another solution. Concentration describes the amount of solutes dissolved by a solution. If a solution has a higher concentration of solutes (less water) than another it is said to be hypertonic. A hypotonic solution has a lower concentration of solutes and more water than another solution.

Tonicity - Definition and Quiz | Biology Dictionary

Main Difference – Isotonic vs Hypotonic vs Hypertonic. A solution is a homogeneous liquid mixture of two or more components. A solution is made by dissolving a solute in a solvent. There are three types of solutions grouped based on their concentrations.

Difference Between Isotonic Hypotonic and Hypertonic ...

An isotonic solution refers to two solutions having the same osmotic pressure across a semipermeable membrane. This state allows for the free movement of water across the membrane without changing ...

Isotonic Solution: Definition & Example - Study.com

Memorizing the definition of hypertonic, hypotonic, and isotonic solutions is not necessary when you understand the root words that make up these descriptive words.

Difference between Hypertonic, Hypotonic, Isotonic ...

Isotonic definition, noting or pertaining to solutions characterized by equal osmotic pressure. See more.

Isotonic | Definition of Isotonic at Dictionary.com

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3 types of Osmotic solutions Flashcards | Quizlet

Osmosis (/ p z ' m ov. s I s /) is the spontaneous net movement of solvent molecules through a selectively permeable membrane into a region of higher solute concentration, in the direction that tends to equalize the solute concentrations on the two sides. It may also be used to describe a physical process in which any solvent moves across a selectively permeable membrane (permeable to the ...

Osmosis - Wikipedia

Hypertonic Solution. When a cell is placed in a hypertonic solution (in a relation to the cell), there will be a higher concentration of solute outside of the cell, so water will diffuse away from the cell, towards the hypertonic solution, in order to balance the solvent concentrations.

What is Osmosis: Definition, Diagram, Examples and Explanation

Name: KEY ACTIVE AND PASSIVE CELL TRANSPORT Match the definition on the left with the term on the right. 1. C release of wastes or cell products from inside to outside a cell

CELL TRANSPORT WORKSHEET - Mrs. Slovacek's Science

A tone is the kind of sound you hear in a musical note, or in a person's voice live or in writing. A newspaper article should be objective, but a poem can bring up all kinds of emotions, depending on the tone.

tone - Dictionary Definition : Vocabulary.com

Main Difference – Colloid vs Crystalloid. The main difference between colloid and crystalloid is their particle size. Colloidal systems have much larger particles compared to crystalloid systems. Hence, the permeability of colloidal systems is lower than that of crystalloid systems.

Difference Between Colloid and Crystalloid | Definition ...

hemolysis [he-mol´ĭ-sis] rupture of erythrocytes with release of hemoglobin into the plasma. Some microbes form substances called hemolysins that have the specific action of destroying red blood cells; the beta-hemolytic streptococcus is an example. Intravenous administration of a hypotonic solution or plain distilled water will also destroy red blood ...

Hemolysis | definition of hemolysis by Medical dictionary

Chapter 4 – Characteristics of Prokaryotic & Eukaryotic Cells. All cells have: . 1. Cell or plasma membrane (separates the cell from the outer environment)

Rohde - Bio

Hyponatremic Dehydration A. Signs of Hyponatremia B. Causes of Hyponatremia C. Diagnostic Algorithm D. Treatment Strategies A. Signs of Hyponatremia. Hyponatremic dehydration (dehydration with serum sodium < 140) may appear clinically similar to isotonic dehydration, but if the hyponatremia becomes pronounced, additional symptoms will become evident.

Hyponatremic Dehydration - NetScut Homepage

Human body cells are immersed in fluids both inside and outside the cell. The water that is inside accounts for about one-half to two-thirds of an average person's weight. Fluid inside the cells is called intracellular fluid (ICF), and fluid outside the cells is called extracellular fluid (ECF ...

10 Fluid And Electrolyte Imbalances Nursing Care Plans ...

pour definition: 1. to make a substance flow from a container, especially into another container, by raising just one side of the container that the substance is in: 2. to (cause to) flow quickly and in large amounts: 3. to make a substance flow, esp. out of a container and usually into.... Learn more.

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