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10.5 10.4 0.819 1.48 2.22 2.63 3.61 5.45 The polarizability volumes are mean values over all orientations Of the molecule. Because it attracts the electrons more strongly, the more electronegative atom is usually the nega- tive end of the dipole. However, there are excep- tions, particularly when antibonding orbitals are occupied.

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