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Prentice hall chemistry book. "CORE TEACHING RESOURCES" also known as -CORE Teaching Resources Prentice Hall Chemistry - Spiral-bound - Teacher's Edition (2005) by Antony C. Wilbraham-

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34 Core Teaching Resources Part CMatching Match each description in Column B to the correct term in Column A. Part DQuestions and Problems Answer each of the following questions in the space provided. 19. State whether each of the following is a homogeneous or heterogeneous mixture. a. table salt dissolved in water a. b. carbon mixed with sand b.

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70 Core Teaching Resources Chapter Test A A. Matching Match each description in Column B with the correct term in Column A.Write the letter of the correct description on the line. B. Multiple Choice Choose the best answer and write its letter on the line. \_\_\_\_\_ 11. How many significant figures are in the measurement 2103.2 g? a. 2 c. 4 b. 3 d ...

### **Chapter Test A - Campbell County Schools**

High School Prentice Hall Chemistry (Wilbraham) © 2008, National Student Edition - South Carolina Teacher's Edition ... seek answers, and develop solutions. Indicators 1 SE = Student Edition - TE = Teacher Edition - TR = Teaching Resources - TECH = Technology ... Core Teaching Resource: 45; Small Scale Lab Manual: p.4-5, All 41

## **High School Prentice Hall - Instructional Resources**

Choose the best answer for each question and write its letter on the line. II. Another name for homogenous mixture is @solution. b. matter. 12. The chemical symbol for iron is: c. element. d. mass. 13. Which of the following is a physical property? a. color b. hardness 48 Core Teaching Resources c. freezing point . all of the above

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Core Teaching Resources . Name Date Class IONIC AND METALLIC BONDING Practice Problems In your notebook, answer the following SECTION 7.1 IONS For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most

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