

Colligative Properties Of Solution Ppt

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Properties of Solutions - HUBBARD'S HONORS CHEMISTRY CLASS

Problem Example 1. The Normal Saline solution used in medicine for nasal irrigation, wound cleaning and intravenous drips is a 0.91% (w/v) solution of sodium chloride in water. How would you prepare 1.5 L of this solution? Solution: The solution will contain 0.91 g of NaCl in 100 mL of water, or 9.1 g in 1 L. Thus you will add $(1.5 \times 9.1\text{g}) = 13.6$ g of NaCl to 1.5 L of water.

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