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In the scientific study of chemistry, the process of electrolysis involves using direct electric current to prompt a chemical reaction which otherwise would be non-spontaneous. Do you think you can answer all of the questions regarding electrolysis correctly in this quiz? Find out now!

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Electrolysis questions and answers, permanent, hair removal

Chemistry Class-X 8 Questions Bank 17. Complete the table given below : Ans. 18. (a) Describe what would you see at the anode and the cathode during electrolysis of copper (II) chloride, using platinum as cathode and carbon as anode. Write ionic equations for the

Question Bank Electrolysis - Testlabz

Each time, you will be given a different set of questions. Good luck! Show all questions <= =>
Which English chemist is associated with the discovery of sodium and potassium in 1807, using a new method of electrolysis? ? Humphry Davy ? John Dalton ? Joseph Priestley ? Michael Faraday;
Which one of the following statements is NOT correct? ...

Chemistry - Electrolysis

Best Answer: You're doing nothing wrong at all. In a salt solution, it's quite normal for the electrolysis of water to be the preferred reaction. That's why electrolysis processes like the formation of Na or Cl₂ from NaCl happen in molten salt, and specifically in the absence of water.

electrolysis question? | Yahoo Answers

Extraction of Metals. Revision Questions. The best way to remember the information in this chapter is to get a pen and paper and write down your answers before clicking on the Answer link which will take you to the correct page.. You may have to read through some of the page before you find the answer. If the answer you have written is not right, change it to the ...

GCSE CHEMISTRY - Revision Questions - Extraction of Metals ...

AP REVIEW QUESTIONS – Electrochemistry - Answers 2007 part A, question #3 An external direct-current power supply is connected to two platinum electrodes immersed in a beaker containing 1.0 M CuSO₄(aq) at 25 °C, as shown in the diagram above. As the cell operates, copper metal is

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