

Define A Saturated Solution

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Define A Saturated Solution

The definition of a supersaturated solution is one which contains more dissolved solute than could ordinarily dissolve into the solvent. A minor disturbance of the solution or introduction of a "seed" or tiny crystal of solute will force crystallization of excess solute.

Saturated Solution Definition and Examples - ThoughtCo

A saturated solution is one in which any additional solute added to the solution is no longer dissolved. Solutions are combinations of solvents – most commonly liquids – and solutes, which are typically solids. Keep Learning.

What Is the Definition of a "saturated Solution ...

Definition of Saturated Solution. A saturated solution is one where there are about equal amounts of particles or solutes and solvent in the solution. If you live on one of the coasts, you've probably gone to the beach and played in the water.

Saturated Solution: Definition & Examples - Study.com

Understanding Saturated Solutions . There are many different factors that can affect whether something is a saturated solution. For example, saturation is affected by: The solution's temperature ; The solution's pressure ; Chemical makeup of substances involved ; Ways to make a saturated solution include: Add solute to liquid until dissolving stops.

Examples of Saturated Solution

When water, or any solvent, has dissolved as much of any substance as it can, it is a saturated solution. A saturated solution is a solution in which there is so much solute that if there was any more, it would not dissolve. Collins English Dictionary.

Saturated solution definition and meaning | Collins ...

saturated solution. a solution in which the solvent contains the maximum amount of solute it can dissolve at a particular temperature. It is often expressed as the number of grams of solute that can dissolve in 100 mL of solution. See also solute, solvent.

Saturated solution | definition of saturated solution by ...

A solution in which the maximum amount of solvent has been dissolved. Any more solute added will sit as crystals on the bottom of the container.

saturated solution definition - Northwestern University

A saturated solution is a solution in which you can no longer dissolve any liquid, gas or solids. For example, take a cup of water. Put some salt in it and mix it. the salt will dissolve.

Define saturated solution - answers.com

(of a solution) containing the maximum amount of solute capable of being dissolved under given conditions. (of an organic compound) containing no double or triple bonds; having each single bond attached to an atom or group. (of an inorganic compound) having no free valence electrons.

Saturated | Definition of Saturated at Dictionary.com

A supersaturated solution is a solution with more dissolved solute than the solvent would normally dissolve in its current conditions. Supersaturation is achieved by dissolving a solute in one set of conditions, then transferring it to other conditions without triggering any release of the solute.

What Is a Supersaturated Solution? | Reference.com

A solution which is saturated with respect to double salt alone will be supersaturated with respect to potassium chloride. The Phase Rule and Its Applications | Alexander Findlay The alkaline liquor must be supersaturated with muriatic acid, and evaporated to dryness.

Supersaturated | Definition of Supersaturated at ...

Supersaturation is a state of a solution that contains more of the dissolved material than could be dissolved by the solvent under normal circumstances. It can also refer to a vapor of a compound that has a higher (partial) pressure than the vapor pressure of that compound. Supersaturation is used in cloud chambers.

Supersaturation - Wikipedia

If a solution has the maximum amount of solute that can be dissolved, it is said to be saturated. Adding any more solute would cause it to simply fall out of solution. Adding any more solute would ...

Supersaturated Solution: Definition & Example - Study.com

Solutions. In physical chemistry, saturation is the point at which the solute of a substance can dissolve no more of that substance and additional amounts of it will appear as a separate phase (as a precipitate, if solid, or as effervescence or inclusion, if gaseous).

Saturation (chemistry) - Wikipedia

Supersaturated solutions are solution which can't dissolve solute further. Example when common salt is dissolved in water upto some extent it becomes saturated normally we can not dissolve more salt but on heating we can dissolve more salt upto certain limit this is supersaturated condition.

What is a supersaturated solution? What are some examples ...

Define supersaturated. supersaturated synonyms, supersaturated pronunciation, supersaturated translation, English dictionary definition of supersaturated. tr.v. su·per·sat·u·rat·ed , su·per·sat·u·rat·ing , su·per·sat·u·rates 1. To cause to contain more dissolved solute than is normally possible under given...

Supersaturated - definition of supersaturated by The Free ...

Unsaturated Solution Definition. An unsaturated solution is a chemical solution in which the solute concentration is lower than its equilibrium solubility. All of the solute dissolves in the solvent. When a solute (often a solid) is added to a solvent (often a liquid), two processes occur simultaneous.

Unsaturated Solution Definition - ThoughtCo

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Saturated solution - Biology-Online Dictionary | Biology ...

Saturated solution @ chemistry dictionary & glossary. When saturated, the rate of dissolving solid and that aug 23, 2017 a statement to effect solvent holds all solute it can define saturation is ...

What Is The Definition Of Saturated In Science

Supersaturated definition is - containing an amount of a substance greater than that required for saturation as a result of having been cooled from a higher temperature to a temperature below that at which saturation occurs. How to use supersaturated in a sentence.

Define A Saturated Solution

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