## SHAHEED BHAGAT SINGH STATE TECHNICAL CAMPUS, FEROZEPUR

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| B.Tech.    CHE    8th Sem   7th scm   |        |
| Chemical process simulation Subject Code BTCH-801   | )      |
| Paper ID: M\18  |        |
|   | re: 60 |
| Time allowed: 3 Hrs  * (2011 belth orwards) Max Mark Important Instructions:  | .s. 00 |
| All questions are compulsory  |        |
| Assume any missing data   |        |
| PART A (2×10)   |        |
| Q. 1. Short-Answer Questions:  a) What do you understand by manipulated variables? b) What do you mean by Mathematical consistency of a model? c) What are the various uses of mathematical modeling in Research & Development? d) Give the iterative procedure steps for bubble-point calculation. e) Write steady state Model in reference to a Single Component Vaporizer. f) Write down general component continuity for j-th component & total energy equation for any g) What is equilibrium constant model for pH measurement? h) Write mathematical form of law of Mass action. i) What is deterministic model? j) What is lumped Parameter model?  PART B (8×5)  Q. 2. Define the following terms related to Mathematical Modeling: (a) Distributed parameter model.   |        |
| <ul><li>(b) Thermal Equilibrium model.</li><li>(c) Significant Metal wall capacitance.</li><li>(d) Slop</li></ul>   |        |
| OR  |        |
| Discuss principles of formulation of mathematical modeling  | COI    |
| <ul> <li>Q. 3. Write the Component continuity &amp; energy equations for a semi-batch reactor in which B is added to A.         <ol> <li>(i). Consecutive reaction occurring with exothermic heat of reaction λ<sub>1</sub> &amp; λ<sub>2</sub></li> <li>k<sub>1</sub> k<sub>2</sub></li> <li>A B C</li> <li>(ii). Reversible with forward reaction exothermic heat of reaction λ<sub>1</sub> and λ<sub>2</sub> is taking place, k<sub>1</sub> and k<sub>2</sub> are rate constants for forward and backward reactions.</li> </ol> </li> </ul>  | CO2    |
|   |        |

|           | OR   |     |
|-----------|--|-----|
|           | The liquid in a jacked, non isothermal CSTR is stirred by an agitator whose mass is significant as compared to reaction mass. The mass of reactor wall and   | CO2 |
|           | the mass of jacket wall are also significant. Write the energy equations for the system. Neglect the radial temperature gradient in agitator, reactor and jacket wall.   |     |
| Q. 4.     | Develop the equations describing an "inverted" batch distillation column. This system has a large reflux drum into which the feed is charged. The material is fed to the top of the column. Vapor is generated in a re-boiler in the base. Heavy material is | CO3 |
|           | withdrawn from the bottom of column. Derive a mathematical model for the case when tray hold up cannot be neglected  |     |
|           | OR   |     |
|           | In ideal binary distillation column, derive the modeling equations for   | CO3 |
|           | a) Condenser and reflux dram.  |     |
|           | b) Reboiler and column base.   |     |
|           | c) n <sup>th</sup> tray.   |     |
| Q. 5.     | Develop a Mathematical Model for heat conduction through a solid sphere of radius  | CO4 |
|           | R the center temperature in this sphere is maintained constant at T <sub>max</sub> .   |     |
|           | OR   |     |
|           | Rigorously model multicomponent flash drum.  | CO4 |
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| Q. 6.     | Derive the Mathematical Model for a Lumped Parameter Gas Absorber. List all the assumptions used in deriving the model.  | CO: |
| Property. | OR   |     |
|           | Assuming equilibrium state and uniform mixing develop model for continuous extraction process.   | CO  |